

Supplementary Materials

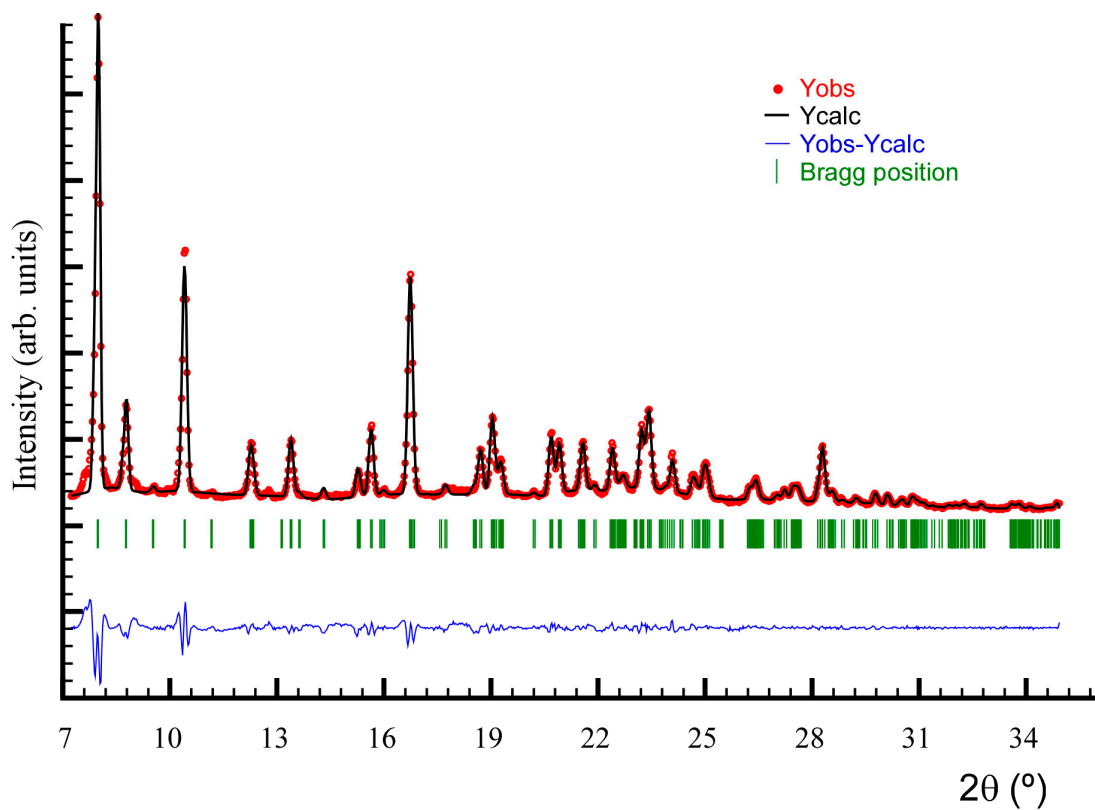


Figure S1. X-ray full profile analysis for compound 1.

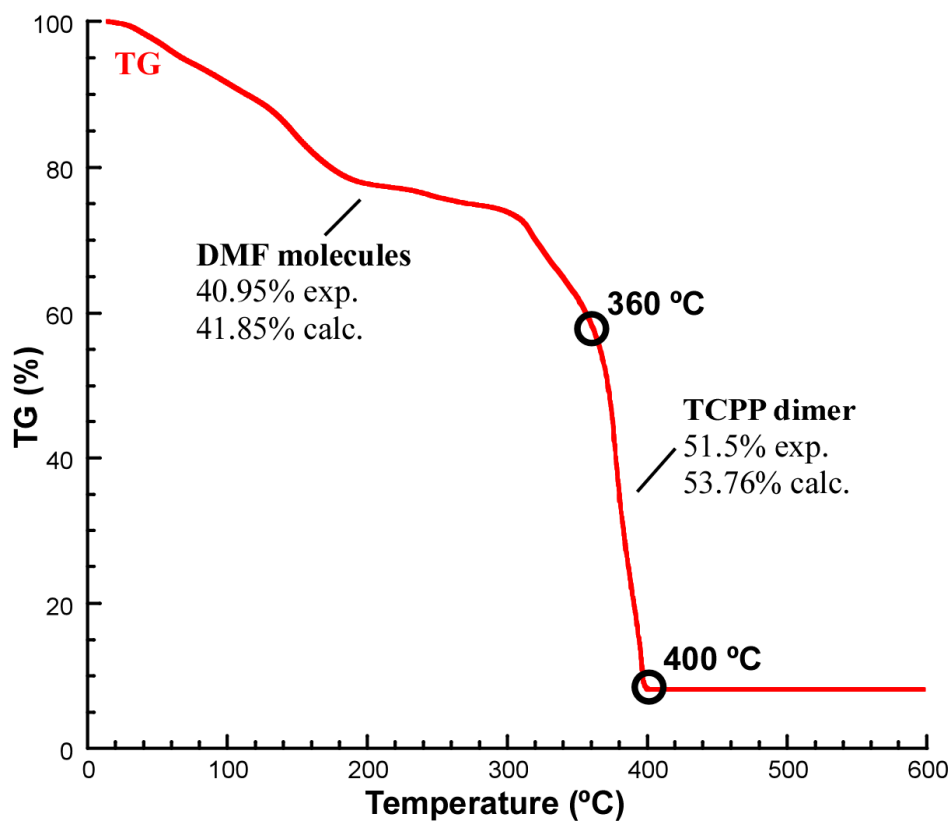


Figure S2. Thermogravimetric analysis for compound 2.

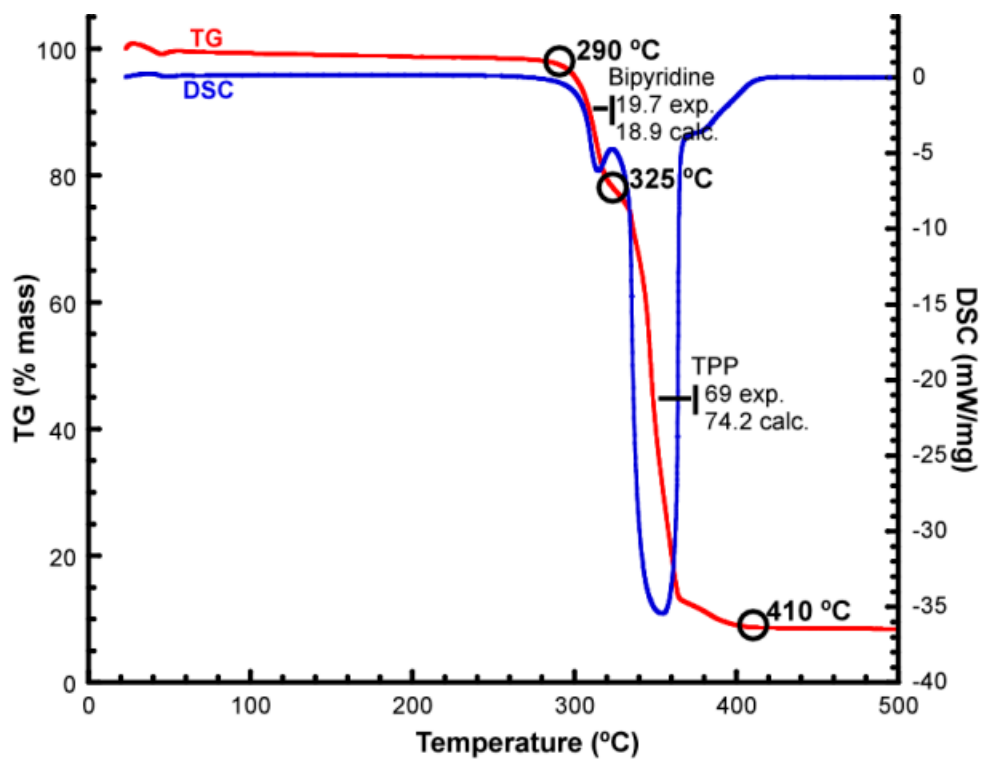


Figure S3. Thermogravimetric analysis for compound 3.

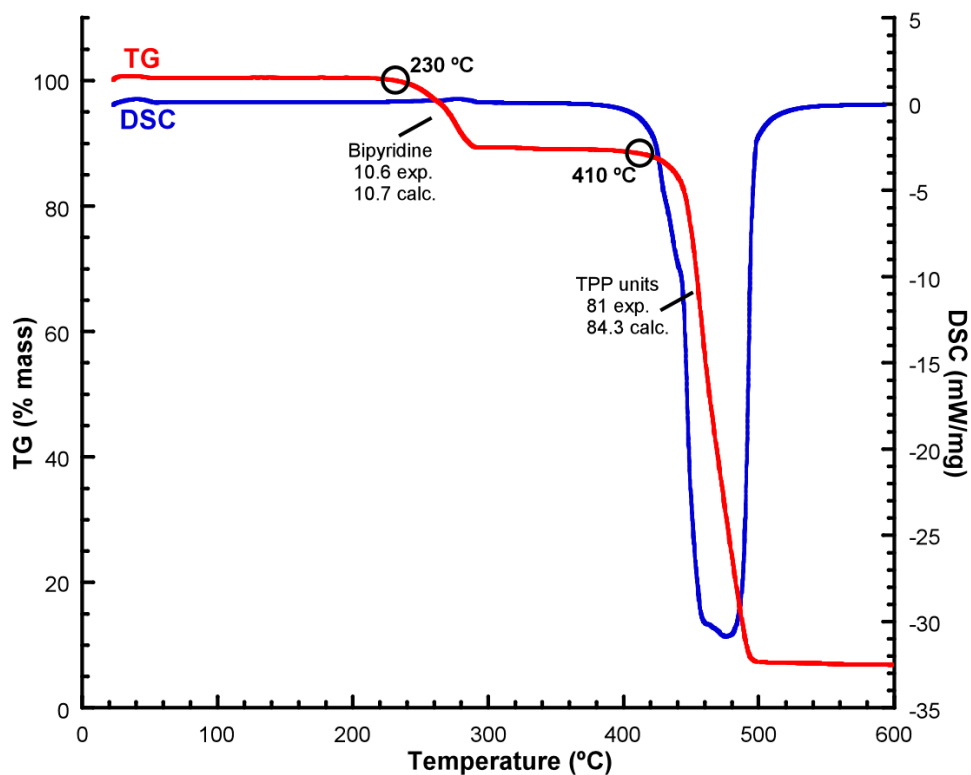


Figure S4. Thermogravimetric analysis for compound 4.

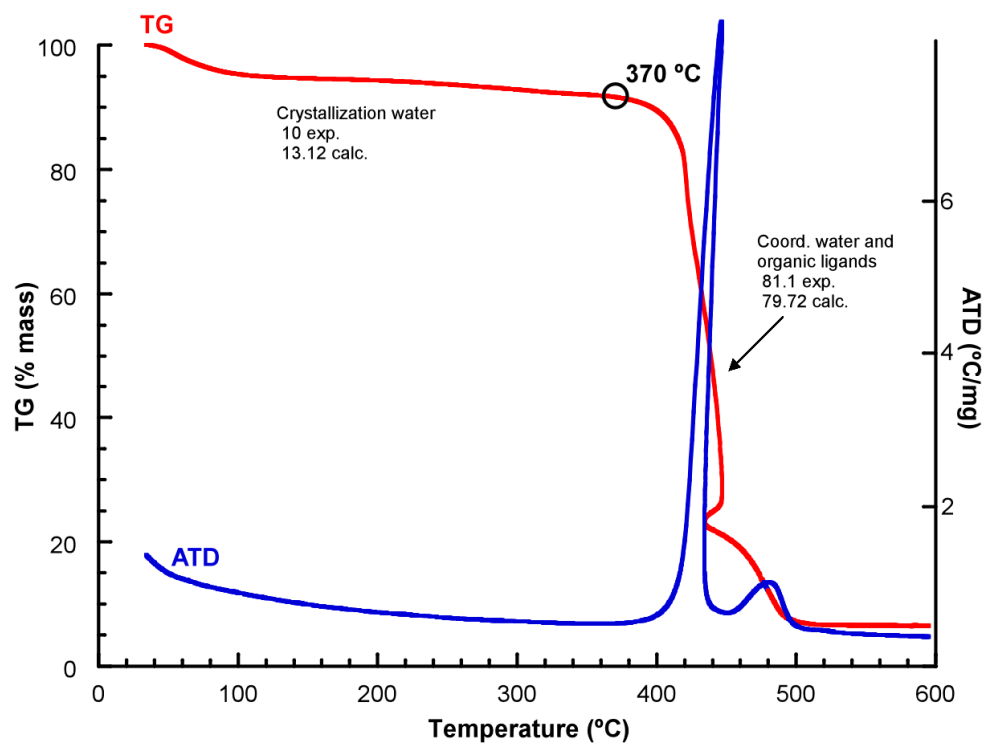


Figure S5. Thermogravimetric analysis for compound **5**.

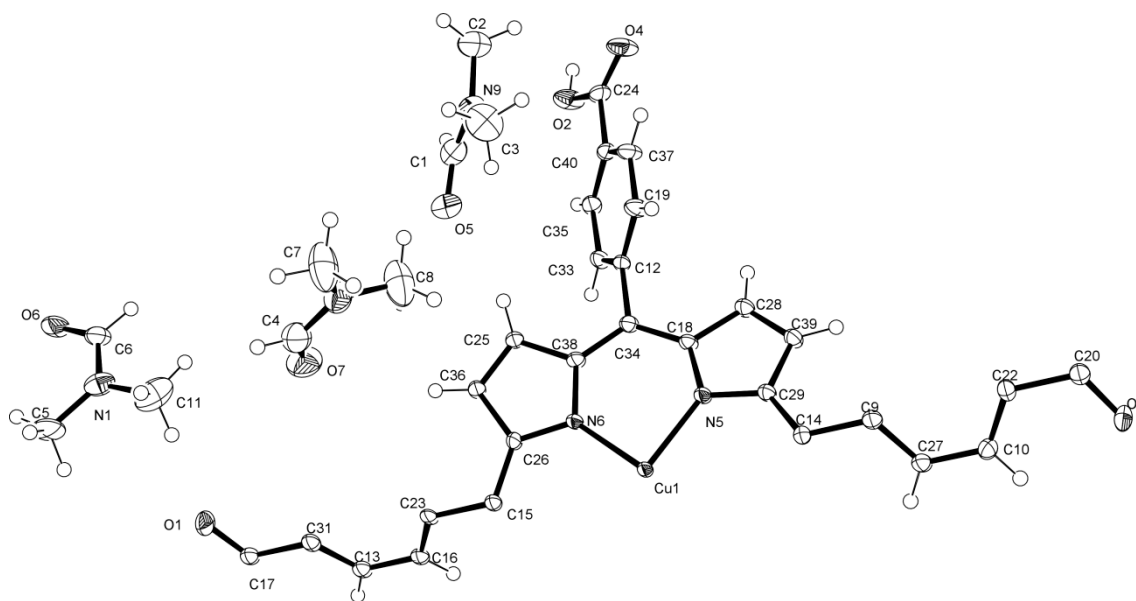


Figure S6. ORTEP detail for compound **1**. The thermal ellipsoids correspond to 50% probability.

Table S1. Most significant angles ($^{\circ}$) and distances (\AA) for compounds **4** and **5** (distances in bold).

		Co(1). CoN ₆ Octahedra				Co(2). CoN ₄ Square Planar			
		Co1	N1	N2	N3	N4	Co2	N5	N6
Compound 4	N4	89.84(4)	89.89(4)	180	2.296(2)	N6	90.38(5)	2.055(1)	
	N3	90.16(4)	90.11(4)	2.357(2)		N5	2.032(1)		
	N2	90.21(6)	1.962(1)						
	N1	1.966(1)							
		Co(1). CoN ₆ Octahedra				Co(2). CoN ₂ (H ₂ O) ₄ Octahedra			
		Co1	N1	N2		Co2	O4	N3	
Compound 5	N2	88.20(7)	1.976(4)			N3	90.76(6)	2.137(4)	
	N1	1.963(3)				O4	2.092(3)		

Table S2. Hydrogen bond parameters for compound **5**.

D-H	A	D-H (\AA)	H \cdots A (\AA)	O-H \cdots A ($^{\circ}$)
O(4)-H(20)	O(3) ⁱ	0.81(3)	1.93(3)	173(4)
O(4)-H(21)	O(1) ⁱⁱ	0.82(4)	1.89(4)	178(4)

Symmetry codes (i) $-1/2 + x, 1/2 + y, 1/2 + z$; (ii) $-1/4 - y, -1/4 + x, 3/4 - z$

Table S3. Most significant angles ($^{\circ}$) and distances (\AA) for compounds **6** (distances in bold).

		Fe(1). FeN ₄ O ₂ Octahedra						
		Fe	N1	N2	N3	N4	O1	O6
Compound 6	O6	85.44(10)	89.98(10)	86.91(11)	84.09(10)	178.34(11)	2.220(3)	
	O1	95.62(11)	88.74(11)	92.03(11)	97.18(11)	1.988(2)		
	N4	90.15(11)	174.07(13)	89.35(11)	2.060(3)			
	N3	172.34(13)	90.17(11)	2.039(3)				
	N2	89.54(11)	2.067(3)					
	N1	2.030(3)						

Table S4. Intra- and interlayer hydrogen bond parameters for compound **6**.

D-H	A	D-H (\AA)	H \cdots A (\AA)	O-H \cdots A ($^{\circ}$)
O(4)-H(40)	O(2) ⁱ	0.84	1.78(4)	171(4)
O(5)-H(50)	O(3) ⁱⁱ	0.84	2.00(4)	144(2)
O(8)-H(80)	O(2) ⁱⁱⁱ	0.84	1.91(5)	169(3)

Symmetry codes (i) $2 - x, -3/2 + y, -z$; (ii) $2 - x, 1/2 + y, 1 - z$; (iii) $1 - x, 1/2 + y, -z$

Table S5. Bond distances (Å) and angles (°) for compound **1**.

Distances			
Cu(1)-N(6)	1.998(1)	O(6)-C(6)	1.235(4)
Cu(1)-N(6) ⁱ	1.998(1)	N(1)-C(11)	1.450(4)
Cu(1)-N(5) ⁱ	2.003(1)	C(5)-H(5A)	0.9600
Cu(1)-N(5)	2.003(1)	C(5)-H(5C)	0.9600
O(1)-C(17)	1.268(2)	C(11)-H(11A)	0.9600
O(2)-C(24)	1.317(3)	C(11)-H(11C)	0.9600
O(2)-H(2)	0.8199	N(7)-C(4)	1.337(4)
O(3)-C(20)	1.272(2)	N(7)-C(7)	1.460(5)
O(4)-C(24)	1.207(3)	C(7)-H(7A)	0.9600
N(5)-C(29)	1.369(3)	C(7)-H(7C)	0.9600
N(5)-C(18)	1.371(3)	C(8)-H(8B)	0.9600
N(6)-C(26)	1.372(3)	C(13)-C(16)	1.382(3)
N(6)-C(38)	1.380(3)	C(13)-H(13)	0.9300
C(9)-C(27) ⁱ	1.400(3)	C(14)-C(29) ⁱ	1.404(3)
C(9)-C(27)	1.400(3)	C(15)-C(26)	1.400(3)
C(9)-C(14)	1.491(4)	C(16)-C(23)	1.403(3)
C(10)-C(27)	1.378(3)	C(17)-O(1) ⁱ	1.268(2)
C(10)-C(22)	1.392(3)	C(18)-C(28)	1.447(3)
C(10)-H(10)	0.9300	C(19)-H(19)	0.9300
C(12)-C(19)	1.397(3)	C(20)-C(22)	1.484(4)
C(12)-C(33)	1.398(3)	C(23)-C(16) ⁱ	1.403(3)
C(12)-C(34)	1.487(3)	C(25)-C(36)	1.354(3)
C(12)-C(34)	1.487(3)	C(25)-H(25)	0.9300
C(13)-C(31)	1.396(3)	C(27)-H(27)	0.9300
C(14)-C(29)	1.404(3)	C(28)-H(28)	0.9300
C(15)-C(26) ⁱ	1.400(3)	C(31)-C(13) ⁱ	1.396(3)
C(15)-C(23)	1.488(4)	C(33)-H(33)	0.9300
C(16)-H(16)	0.9300	C(35)-C(40)	1.390(3)
C(17)-C(31)	1.488(4)	C(36)-H(36)	0.9300
C(19)-C(37)	1.388(3)	C(37)-H(37)	0.9300
C(20)-O(3) ⁱ	1.272(2)	O(5)-C(1)	1.230(4)
C(22)-C(10) ⁱ	1.392(3)	N(9)-C(3)	1.439(5)
C(24)-C(40)	1.497(3)	C(1)-H(1)	0.9300
C(25)-C(38)	1.438(3)	C(2)-H(2B)	0.9600
C(26)-C(36)	1.444(3)	C(3)-H(3A)	0.9600
C(28)-C(39)	1.358(3)	C(3)-H(3C)	0.9600
C(29)-C(39)	1.440(3)	N(1)-C(6)	1.320(4)
C(33)-C(35)	1.385(3)	N(1)-C(5)	1.452(4)
C(34)-C(38)	1.395(3)	C(5)-H(5B)	0.9600
C(35)-H(35)	0.9300	C(6)-H(6)	0.9300
C(37)-C(40)	1.394(3)	C(11)-H(11B)	0.9600
C(39)-H(39)	0.9300	O(7)-C(4)	1.204(4)
N(9)-C(1)	1.329(4)	N(7)-C(8)	1.435(5)
N(9)-C(2)	1.451(4)	C(4)-H(4)	0.9300
C(2)-H(2A)	0.9600	C(7)-H(7B)	0.9600
C(2)-H(2C)	0.9600	C(8)-H(8A)	0.9600
C(3)-H(3B)	0.9600	C(8)-H(8C)	0.9600

Table S5. *Cont.*

Angles			
N(6)-Cu(1)-N(6) ⁱ	89.69(1)	C(10) ⁱ -C(22)-C(20)	120.51(1)
N(6)-Cu(1)-N(5) ⁱ	175.59(7)	C(10)-C(22)-C(20)	120.51(1)
N(6) ⁱ -Cu(1)-N(5) ⁱ	90.44(8)	C(16)-C(23)-C(16) ⁱ	118.5(3)
N(6)-Cu(1)-N(5)	90.44(8)	C(16)-C(23)-C(15)	120.86(1)
N(6) ⁱ -Cu(1)-N(5)	175.59(7)	C(16) ⁱ -C(23)-C(15)	120.86(1)
N(5) ⁱ -Cu(1)-N(5)	89.96(1)	O(4)-C(24)-O(2)	124.0(2)
C(24)-O(2)-H(2)	109.4	O(4)-C(24)-C(40)	123.3(2)
C(29)-N(5)-C(18)	106.2(1)	O(2)-C(24)-C(40)	112.6(2)
C(29)-N(5)-Cu(1)	126.8(1)	C(36)-C(25)-C(38)	107.0(2)
C(18)-N(5)-Cu(1)	126.3(1)	C(36)-C(25)-H(25)	126.5
C(26)-N(6)-C(38)	106.1(1)	N(6)-C(26)-C(15)	125.2(2)
C(26)-N(6)-Cu(1)	127.1(1)	N(6)-C(26)-C(36)	109.9(1)
C(38)-N(6)-Cu(1)	126.3(1)	C(15)-C(26)-C(36)	124.7(2)
C(27) ⁱ -C(9)-C(27)	118.5(3)	C(10)-C(27)-C(9)	120.6(2)
C(27) ⁱ -C(9)-C(14)	120.8(1)	C(10)-C(27)-H(27)	119.7
C(27)-C(9)-C(14)	120.8(1)	C(9)-C(27)-H(27)	119.7
C(27)-C(10)-C(22)	120.5(2)	C(39)-C(28)-C(18)	106.9(1)
C(27)-C(10)-H(10)	119.7	C(39)-C(28)-H(28)	126.6
C(22)-C(10)-H(10)	119.7	C(18)-C(28)-H(28)	126.6
C(19)-C(12)-C(33)	118.5(2)	N(5)-C(29)-C(14)	125.2(2)
C(19)-C(12)-C(34)	120.9(2)	N(5)-C(29)-C(39)	110.4(1)
C(33)-C(12)-C(34)	120.7(2)	C(14)-C(29)-C(39)	124.3(2)
C(16)-C(13)-C(31)	120.0(2)	C(13)-C(31)-C(13) ⁱ	119.9(3)
C(16)-C(13)-H(13)	120.0	C(13)-C(31)-C(17)	120.1(1)
C(31)-C(13)-H(13)	120.0	C(13) ⁱ -C(31)-C(17)	120.1(1)
C(29)-C(14)-C(29) ⁱ	123.1(3)	C(35)-C(33)-C(12)	120.7(2)
C(29)-C(14)-C(9)	118.5(1)	C(35)-C(33)-H(33)	119.6
C(29) ⁱ -C(14)-C(9)	118.5(1)	C(12)-C(33)-H(33)	119.6
C(26) ⁱ -C(15)-C(26)	122.9(3)	C(38)-C(34)-C(18)	123.7(2)
C(26) ⁱ -C(15)-C(23)	118.6(1)	C(38)-C(34)-C(12)	117.6(1)
C(26)-C(15)-C(23)	118.6(1)	C(18)-C(34)-C(12)	118.8(1)
C(13)-C(16)-C(23)	120.8(2)	C(33)-C(35)-C(40)	120.4(2)
C(13)-C(16)-H(16)	119.6	C(33)-C(35)-H(35)	119.8
C(23)-C(16)-H(16)	119.6	C(40)-C(35)-H(35)	119.8
O(1) ⁱ -C(17)-O(1)	123.0(3)	C(25)-C(36)-C(26)	107.1(1)
O(1) ⁱ -C(17)-C(31)	118.6(1)	C(25)-C(36)-H(36)	126.5
O(1)-C(17)-C(31)	118.6(1)	C(26)-C(36)-H(36)	126.5
N(5)-C(18)-C(34)	125.6(2)	C(19)-C(37)-C(40)	120.0(2)
N(5)-C(18)-C(28)	109.8(1)	C(19)-C(37)-H(37)	120.0
C(34)-C(18)-C(28)	124.6(2)	C(40)-C(37)-H(37)	120.0
C(37)-C(19)-C(12)	120.9(2)	N(6)-C(38)-C(34)	125.4(2)
C(37)-C(19)-H(19)	119.5	N(6)-C(38)-C(25)	109.7(1)
C(12)-C(19)-H(19)	119.5	C(34)-C(38)-C(25)	124.6(2)
O(3) ⁱ -C(20)-O(3)	123.4(3)	C(28)-C(39)-C(29)	106.7(1)
O(3) ⁱ -C(20)-C(22)	118.3(1)	C(28)-C(39)-H(39)	126.7
O(3)-C(20)-C(22)	118.3(1)	C(29)-C(39)-H(39)	126.7
C(10) ⁱ -C(22)-C(10)	119.2(3)		

Symmetry code: (i) $-x, y, -z - 3/2$

Table S6. Fractional atomic coordinates ($\times 10^4$) and equivalent thermal factors ($\times 10^3$) for compound **1**.

Atoms	X	Y	Z	$U_{eq}, \text{\AA}^2$
Cu(1)	0	-625(1)	-7500	13(1)
O(1)	-273(1)	-5042(1)	-7001(2)	29(1)
O(2)	-2598(1)	-1086(1)	-15,336(2)	36(1)
O(3)	314(1)	3789(1)	-7663(2)	26(1)
O(4)	-2792(1)	-290(1)	-14,104(3)	36(1)
N(5)	-380(1)	13(1)	-8614(2)	15(1)
N(6)	-396(1)	-1263(1)	-8457(2)	15(1)
C(9)	0	1591(1)	-7500	18(1)
C(10)	324(1)	2533(1)	-7804(3)	19(1)
C(12)	-1351(1)	-630(1)	-11,114(3)	16(1)
C(13)	242(1)	3786(1)	-8333(3)	19(1)
C(14)	0	920(1)	-7500	16(1)
C(15)	0	-2171(1)	-7500	15(1)
C(16)	238(1)	-3164(1)	-8347(3)	18(1)
C(17)	0	-4770(1)	-7500	20(1)
C(18)	-736(1)	-79(1)	-9737(3)	16(1)
C(19)	-1660(1)	-301(1)	-10,708(3)	23(1)
C(20)	0	3518(1)	-7500	19(1)
C(22)	0	2850(1)	-7500	17(1)
C(23)	0	-2841(1)	-7500	16(1)
C(24)	-2542(1)	-656(1)	-14,223(3)	22(1)
C(25)	-1002(1)	-1729(1)	-9510(3)	17(1)
C(26)	-359(1)	-1870(1)	-8175(3)	15(1)
C(27)	324(1)	1913(1)	-7810(3)	18(1)
C(28)	-878(1)	482(1)	-10,544(3)	18(1)
C(29)	-309(1)	619(1)	-8613(3)	16(1)
C(31)	0	-4100(1)	-7500	19(1)
C(33)	1438(1)	-971(1)	-12,559(3)	18(1)
C(34)	-941(1)	-625(1)	-10,025(3)	16(1)
C(35)	-1822(1)	-990(1)	-13,550(3)	19(1)
C(36)	-738(1)	-2163(1)	-8824(3)	18(1)
C(37)	-2043(1)	-306(1)	-11,723(3)	24(1)
C(38)	-785(1)	-1170(1)	-9333(3)	16(1)
C(39)	613(1)	914(1)	-9845(3)	18(1)
C(40)	-2127(1)	-658(1)	-13,142(3)	19(1)
O(5)	-1963(1)	-2137(1)	-10,576(3)	46(1)
N(9)	-2521(1)	-1600(1)	-10,505(3)	40(1)
C(1)	-2309(1)	-1977(1)	-11,191(4)	37(1)
C(2)	-2937(1)	-1473(2)	-11,270(6)	62(1)
C(3)	2357(1)	-1346(2)	-8906(5)	59(1)
O(6)	-1743(1)	-6027(1)	-7477(2)	35(1)
N(1)	-1277(1)	-5462(1)	-5775(3)	33(1)
C(5)	-1059(1)	-5976(2)	-4941(4)	44(1)
C(6)	-1602(1)	-5533(1)	-6969(3)	30(1)
C(11)	-1124(1)	-4865(2)	-5287(5)	54(1)
O(7)	-1119(1)	-3430(1)	-6215(3)	54(1)
N(7)	-1424(1)	-2742(1)	-4923(3)	40(1)
C(4)	-1253(1)	-3272(2)	-5076(4)	42(1)
C(7)	-1583(2)	-2596(2)	-3500(5)	65(1)
C(8)	-1448(2)	-2281(2)	-6142(5)	66(1)

$$U_{eq} = 1/3[U_{11}(aa^*)^2 + U_{22}(bb^*)^2 + U_{33}(cc^*)^2 + 2U_{13}aca^*c^* \cos\beta].$$

Table S7. Anisotropic displacement parameters ($\text{Å}^2 \times 10^3$) for compound **1**.

Atoms	U11	U22	U33	U23	U13	U12
Cu(1)	11(1)	11(1)	16(1)	0	-2(1)	0
O(1)	29(1)	15(1)	46(1)	-1(1)	12(1)	-2(1)
O(2)	20(1)	45(1)	34(1)	-12(1)	-13(1)	7(1)
O(3)	26(1)	15(1)	35(1)	2(1)	2(1)	-5(1)
O(4)	17(1)	44(1)	41(1)	-8(1)	-5(1)	9(1)
N(5)	13(1)	15(1)	16(1)	-1(1)	-1(1)	-1(1)
N(6)	12(1)	15(1)	16(1)	1(1)	-2(1)	1(1)
C(9)	13(2)	20(2)	16(1)	0	-2(1)	0
C(10)	17(1)	18(1)	20(1)	1(1)	1(1)	-3(1)
C(12)	13(1)	15(1)	17(1)	4(1)	-3(1)	1(1)
C(13)	15(1)	17(1)	20(1)	-1(1)	-4(1)	1(1)
C(14)	15(1)	15(1)	16(1)	0	2(1)	0
C(15)	14(1)	16(1)	13(1)	0	-2(1)	0
C(16)	14(1)	18(1)	17(1)	1(1)	-3(1)	0(1)
C(17)	15(2)	16(2)	24(2)	0	-3(1)	0
C(18)	14(1)	16(1)	16(1)	-1(1)	0(1)	4(1)
C(19)	17(1)	24(1)	25(1)	-5(1)	-3(1)	3(1)
C(20)	20(2)	18(2)	16(1)	0	1(1)	0
C(22)	21(2)	15(1)	13(1)	0	0(1)	0
C(23)	10(1)	15(1)	18(1)	0	-7(1)	0
C(24)	15(1)	27(1)	23(1)	2(1)	-1(1)	-2(1)
C(25)	11(1)	17(1)	19(1)	1(1)	-3(1)	-2(1)
C(26)	15(1)	12(1)	17(1)	-1(1)	0(1)	-1(1)
C(27)	16(1)	18(1)	19(1)	-2(1)	1(1)	1(1)
C(28)	17(1)	19(1)	17(1)	2(1)	-2(1)	2(1)
C(29)	13(1)	15(1)	17(1)	1(1)	-1(1)	2(1)
C(31)	15(2)	16(2)	19(2)	0	-7(1)	0
C(33)	17(1)	18(1)	18(1)	2(1)	0(1)	2(1)
C(34)	14(1)	16(1)	17(1)	-1(1)	2(1)	1(1)
C(35)	19(1)	19(1)	17(1)	0(1)	-2(1)	0(1)
C(36)	16(1)	15(1)	19(1)	1(1)	-3(1)	-3(1)
C(37)	14(1)	28(1)	28(1)	-5(1)	-2(1)	5(1)
C(38)	14(1)	17(1)	15(1)	-1(1)	-2(1)	1(1)
C(39)	19(1)	15(1)	18(1)	2(1)	1(1)	3(1)
C(40)	15(1)	20(1)	19(1)	4(1)	-1(1)	-1(1)
O(5)	30(1)	49(1)	60(1)	2(1)	12(1)	0(1)
N(9)	30(1)	42(1)	49(2)	5(1)	13(1)	-2(1)
C(1)	33(2)	40(2)	38(2)	1(1)	9(1)	-7(1)
C(2)	36(2)	66(3)	85(3)	22(2)	16(2)	12(2)
C(3)	67(3)	60(2)	48(2)	-7(2)	15(2)	6(2)
O(6)	23(1)	39(1)	35(1)	2(1)	-10(1)	2(1)
N(1)	23(1)	45(1)	30(1)	-6(1)	4(1)	-4(1)
C(5)	22(1)	68(2)	32(2)	11(1)	-9(1)	0(1)
C(6)	20(1)	37(1)	31(1)	5(1)	3(1)	5(1)
C(11)	47(2)	58(2)	61(2)	-28(2)	20(2)	-18(2)
O(7)	45(1)	54(2)	71(2)	-10(1)	30(1)	4(1)
N(7)	45(2)	34(1)	44(1)	-3(1)	16(1)	-4(1)
C(4)	40(2)	37(2)	49(2)	-3(1)	11(2)	-1(1)
C(7)	110(4)	30(2)	66(2)	-2(2)	43(2)	3(2)
C(8)	107(4)	41(2)	65(2)	7(2)	50(2)	10(2)

$$U_{ij} = \exp(-2\pi^2[h^2(a^*)^2 U_{11} + k^2(b^*)^2 U_{22} + \dots + 2hka^*b^* U_{12}]).$$

Table S8. Fractional atomic coordinates ($\times 10^4$) and isotropic thermal factors ($\times 10^3$) of hydrogen atoms for compound **1**.

Atoms	X	Y	Z	$U_{iso} \text{ \AA}^2$
H(2)	-2808	-1026	-16,020	54
H(10)	541	2741	-8005	23
H(13)	406	-3995	-8879	22
H(16)	396	-2957	-8926	21
H(19)	-1608	-76	-9744	28
H(25)	-1272	-1782	-10,002	20
H(27)	541	1706	-8021	22
H(28)	-1107	536	-11,382	22
H(33)	-1235	-1188	-12,857	22
H(35)	-1877	-1227	-14,494	23
H(36)	-791	-2572	-8780	21
H(37)	-2243	-74	-11,457	29
H(39)	-625	1321	-10,110	22
H(1)	-2431	-2132	-12,221	44
H(2A)	-3104	-1642	-10,614	93
H(2B)	-3006	-1647	-12,350	93
H(2C)	-2977	-1045	-11,351	93
H(3A)	-2071	-1383	-8644	88
H(3B)	-2461	-1556	-8097	88
H(3C)	-2428	-928	-8909	88
H(5A)	-1192	-6340	-5383	65
H(5B)	-1048	-5953	-3786	65
H(5C)	-792	-5974	-5103	65
H(6)	-1733	-5189	-7459	36
H(11A)	-856	-4832	-5426	81
H(11B)	-1123	-4799	-4153	81
H(11C)	-1290	-4570	-5958	81
H(4)	-1234	-3545	-4220	50
H(7A)	-1422	-2287	-2864	97
H(7B)	-1577	-2949	-2834	97
H(7C)	-1853	-2457	-3868	97
H(8A)	-1356	-2437	-7055	99
H(8B)	-1284	-1946	-5671	99
H(8C)	-1721	-2151	-6516	99

$$U_{iso} = \exp[-8\pi^2 U(\sin\theta/\lambda)^2].$$