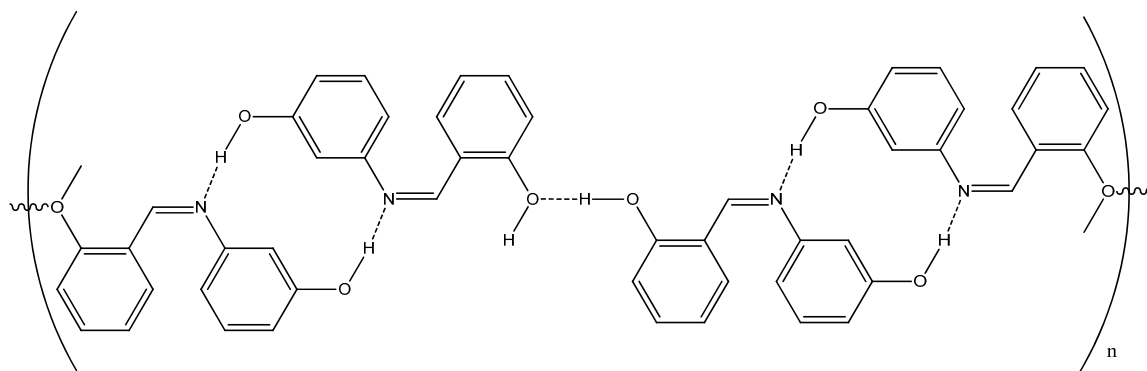


# Supplementary Materials: The Scavenging of DPPH, Galvinoxyl and ABTS Radicals by Imine Analogs of Resveratrol

Peter Kotora <sup>1,\*</sup>, František Šeršeň <sup>1</sup>, Juraj Filo <sup>1</sup>, Dušan Loos <sup>1</sup>, Juraj Gregáň <sup>2,3,\*</sup> and Frídrieh Gregáň <sup>4</sup>



**Scheme S1.** A possible structure of aggregates for compound 3.

**Table S1.** Proton dissociation energy of prepared (hydroxyphenyliminomethyl)phenols in methanol.

Compound	PDE (kJ/mol)					
	H <sup>+</sup> from R <sup>6-8</sup>	H <sup>+</sup> from R <sup>1</sup>	H <sup>+</sup> from R <sup>2</sup>	H <sup>+</sup> from R <sup>3</sup>	H <sup>+</sup> from R <sup>4</sup>	H <sup>+</sup> from R <sup>5</sup>
1	201.4	225.0				
2	151.9		163.8			
3	130.9	149.4				
4	170.7			170.2		
5	202.0	231.0				
6	167.9		183.1			
7	174.7			182.9		
8	220.2	214.1	215.2			
9	166.4	225.9		217.2		
10	220.4	215.2			219.3	
11	149.4		165.1		164.6	
12	149.0	120.7	123.4			
13	228.4	205.8			211.0	
14	177.9	189.8		192.4		
15	220.7	216.0			220.2	
16	167.5		190.6		185.8	
17	144.4	124.3	116.2	124.1		
18	152.4	112.1		181.0		196.0
19	162.8		141.9	127.6	142.1	
20	155.8	186.3		190.5		186.5
21	161.1		161.3	144.7	161.5	
resveratrol	143.7		163.1		163.3	

**Table S2.** Dissociation energy of hydrogen and electron of prepared (hydroxyphenyliminomethyl)phenols in methanol.

Compound	BDE (kJ/mol)						IP (kJ/mol)
	H <sup>+</sup> from R <sup>6-8</sup>	H <sup>+</sup> from R <sup>1</sup>	H <sup>+</sup> from R <sup>2</sup>	H <sup>+</sup> from R <sup>3</sup>	H <sup>+</sup> from R <sup>4</sup>	H <sup>+</sup> from R <sup>5</sup>	
1	316.7	340.3					859.6
2	320.8		332.6				874.4
3	317.6	336.1					888.0
4	335.9			335.3			886.9
5	317.0	345.9					850.0
6	320.0		335.2				864.8
7	317.3			325.6			850.4
8	313.1	306.9	308.0				871.8
9	290.7	350.2		341.6			853.0
10	314.8	309.6			313.7		844.5
11	324.7		340.5		340.0		882.5
12	337.3	309.0	311.7				880.7
13	335.3	312.6			317.9		860.4
14	317.4	329.3		332.0			848.9
15	314.9	310.2			314.4		845.4
16	324.8		347.9		343.1		873.5
17	335.6	315.4	307.3	315.3			887.4
18	336.0	295.7		364.6		379.6	889.6
19	336.2		315.3	301.0	315.5		890.1
20	318.3	348.7		352.9		348.9	855.0
21	314.7		314.9	298.3	315.1		858.3
resveratrol	318.5		337.8		338.0		868.3

**Table S3.** Proton affinity of prepared (hydroxyphenyliminomethyl)phenols in methanol.

Compound	PA (kJ/mol)					
	H <sup>+</sup> from R <sup>6-8</sup>	H <sup>+</sup> from R <sup>1</sup>	H <sup>+</sup> from R <sup>2</sup>	H <sup>+</sup> from R <sup>3</sup>	H <sup>+</sup> from R <sup>4</sup>	H <sup>+</sup> from R <sup>5</sup>
1	214.1	264.4				
2	270.2		262.8			
3	223.7	249.7				
4	261.5			255.4		
5	224.0	281.6				
6	263.1		262.3			
7	264.6			254.4		
8	205.2	262.3	257.0			
9	205.6	246.2		242.5		
10	216.4	280.1			266.8	
11	269.6		249.0		249.7	
12	218.6	254.1	265.0			
13	259.7	268.3			266.9	
14	266.1	237.9		240.4		
15	258.9	255.1			267.9	
16	264.2		252.9		250.2	
17	211.0	229.7	256.4	239.3		
18	212.1	228.8		235.7		253.8
19	260.0		239.2	246.8	239.3	
20	211.9	230.4		236.3		230.4
21	262.0		240.1	248.3	239.6	
resveratrol	261.5		253.0		253.1	

**Table S4.** Electron transfer enthalpy of prepared (hydroxyphenyliminomethyl)phenols in methanol.

Compound	ETE (kJ/mol)					
	H <sup>+</sup> from R <sup>6-8</sup>	H <sup>+</sup> from R <sup>1</sup>	H <sup>+</sup> from R <sup>2</sup>	H <sup>+</sup> from R <sup>3</sup>	H <sup>+</sup> from R <sup>4</sup>	H <sup>+</sup> from R <sup>5</sup>
1	332.7	306.0				
2	280.7		299.9			
3	324.0	316.6				
4	304.5			310.0		
5	323.1	294.5				
6	287.0		303.0			
7	282.8			301.3		
8	337.9	274.7	281.1			
9	315.2	334.2		329.2		
10	328.5	259.6			277.0	
11	284.6		321.0		319.8	
12	348.8	285.0	276.8			
13	305.7	274.4			281.1	
14	281.4	321.5		321.7		
15	286.1	285.2			276.6	
16	290.7		325.2		323.0	
17	354.7	315.8	281.0	306.0		
18	354.1	297.0		359.1		355.9
19	306.3		306.3	284.3	306.3	
20	336.5	348.4		346.7		348.6
21	282.9		304.9	280.2	305.6	
resveratrol	287.1		315.0		315.0	

**Table S5.** Proton dissociation energy of prepared (hydroxyphenyliminomethyl)phenols in water.

Compound	PDE (kJ/mol)					
	H <sup>+</sup> from R <sup>6-8</sup>	H <sup>+</sup> from R <sup>1</sup>	H <sup>+</sup> from R <sup>2</sup>	H <sup>+</sup> from R <sup>3</sup>	H <sup>+</sup> from R <sup>4</sup>	H <sup>+</sup> from R <sup>5</sup>
1	160.4	183.5				
2	111.1		87.4			
3	89.3	108.3				
4	129.3			128.8		
5	160.9	189.3				
6	126.8		142.0			
7	133.2			141.5		
8	179.5	173.0	173.7			
9	123.8	180.3		170.5		
10	179.2	173.6			178.0	
11	108.4		124.1		123.7	
12	108.2	80.0	82.7			
13	189.6	166.7			172.3	
14	136.0	148.1		150.7		
15	179.6	174.8			179.2	
16	126.0		145.0		144.4	
17	103.4	83.2	75.0	82.5		
18	111.0	70.5		139.5		153.9
19	121.6		101.0	86.8	101.1	
20	114.3	145.1		114.4		145.2
21	120.0		120.3	103.9	120.6	
resveratrol	102.9		122.4		122.6	

**Table S6.** Dissociation energy of hydrogen and electron of prepared (hydroxyphenyliminomethyl)phenols in water.

Compound	BDE (kJ/mol)						IP (kJ/mol)
	H <sup>+</sup> from R <sup>6-8</sup>	H <sup>+</sup> from R <sup>1</sup>	H <sup>+</sup> from R <sup>2</sup>	H <sup>+</sup> from R <sup>3</sup>	H <sup>+</sup> from R <sup>4</sup>	H <sup>+</sup> from R <sup>5</sup>	
1	320.6	331.9					860.5
2	316.8		340.0				875.2
3	316.9	335.9					888.7
4	335.8			335.4			887.7
5	316.8	345.1					850.6
6	319.1		334.3				864.5
7	317.3			325.6			851.2
8	313.0	306.6	307.2				872.2
9	289.3	345.8		336.6			853.7
10	314.9	309.3			313.7		845.6
11	324.6		340.1		339.9		883.4
12	337.0	308.7	311.4				881.4
13	335.1	312.3			317.8		861.0
14	316.5	328.6		331.1			850.0
15	314.7	310.0			314.4		846.1
16	324.8		343.8		343.2		874.2
17	335.8	315.6	307.4	314.9			887.9
18	335.9	295.4		364.3		378.7	890.4
19	336.0		315.4	301.2	315.5		890.6
20	317.7	348.4		317.7		348.6	855.8
21	314.7		315.0	298.6	315.2		858.9
resveratrol	318.2		337.6		337.8		868.7

**Table S7.** Proton affinity of prepared (hydroxyphenyliminomethyl)phenols in water.

Compound	PA (kJ/mol)					
	H <sup>+</sup> from R <sup>6-8</sup>	H <sup>+</sup> from R <sup>1</sup>	H <sup>+</sup> from R <sup>2</sup>	H <sup>+</sup> from R <sup>3</sup>	H <sup>+</sup> from R <sup>4</sup>	H <sup>+</sup> from R <sup>5</sup>
1	163.2	213.1				
2	217.7		210.7			
3	172.1	198.4				
4	209.2			204.3		
5	172.5	229.0				
6	210.8		209.1			
7	213.2			203.0		
8	154.6	211.2	212.4			
9	154.8	194.3		191.5		
10	165.7	211.6			214.9	
11	218.0		198.0		198.8	
12	167.1	202.2	213.0			
13	208.0	200.4			215.2	
14	213.4	186.1		188.5		
15	207.3	203.7			215.9	
16	213.5		197.6		198.5	
17	160.3	179.5	205.7	187.8		
18	160.4	178.3		184.8		201.4
19	208.8		188.1	196.2	187.7	
20	16.1	179.3		160.1		179.2
21	211.1		188.8	197.4	188.1	
resveratrol	210.4		201.0		201.1	

**Table S8.** Electron transfer enthalpy of prepared (hydroxyphenyliminomethyl)phenols in water.

Compound	ETE (kJ/mol)					
	H <sup>+</sup> from R <sup>6-8</sup>	H <sup>+</sup> from R <sup>1</sup>	H <sup>+</sup> from R <sup>2</sup>	H <sup>+</sup> from R <sup>3</sup>	H <sup>+</sup> from R <sup>4</sup>	H <sup>+</sup> from R <sup>5</sup>
1	399.0	372.2				
2	348.3		366.5			
3	390.1	382.9				
4	371.9			376.4		
5	389.6	361.5				
6	353.6		370.6			
7	349.5			367.9		
8	403.7	340.7	340.2			
9	379.9	396.9		389.9		
10	394.6	343.1			344.2	
11	351.9		387.7		386.5	
12	415.2	351.8	343.8			
13	372.5	357.2			348.0	
14	348.5	387.8		388.0		
15	352.8	351.7			343.8	
16	356.7		391.6		390.0	
17	420.9	381.5	347.0	372.4		
18	420.8	362.4		424.9		422.7
19	373.4		372.6	350.3	373.1	
20	403.0	414.5		403.0		414.7
21	348.9		371.6	346.6	372.5	
resveratrol	353.1		382.0		382.1	