

<b>Supplementary Information</b>
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**Discovery of flavonoids from *Scutellaria baicalensis* with inhibitory activity against PCSK 9 expression: Isolation, synthesis and their biological evaluation**

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## Figure legends

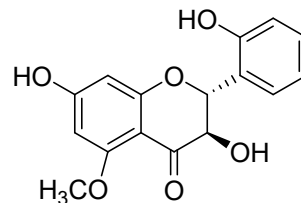
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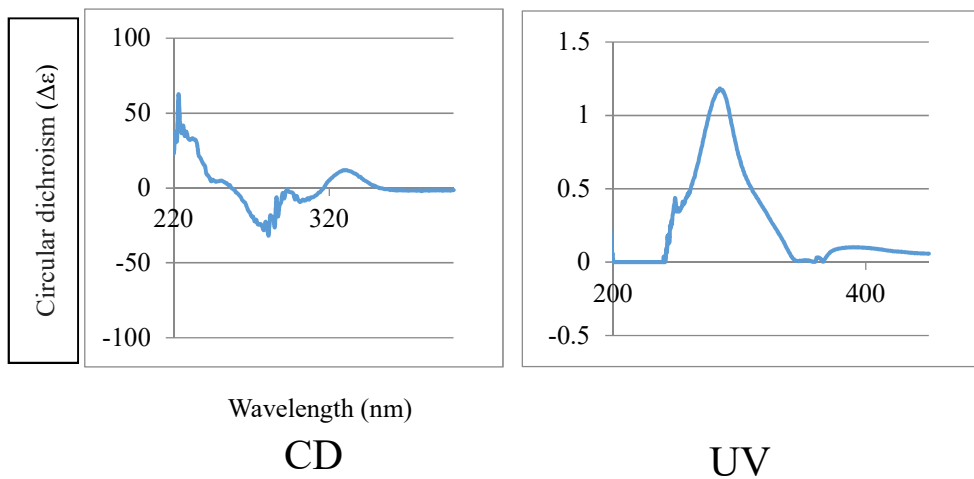
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**Table 1.**  $^1\text{H}$ -and $^{13}\text{C}$ -NMR spectroscopic (in  $\text{CD}_3\text{OD}$ ) of Compound **1** and **1a**-----15

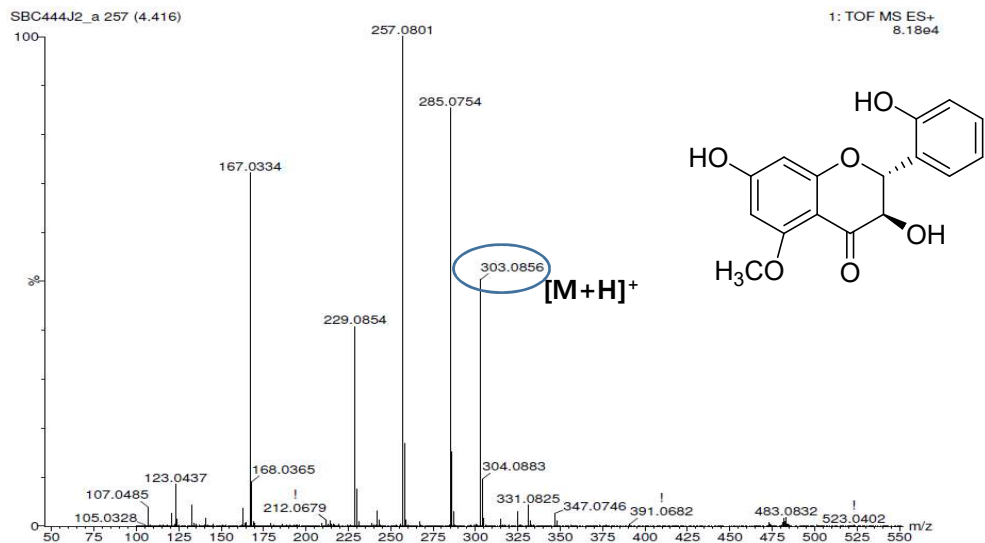
## Compound 1



- Appearance: yellow powder
  - $[\alpha]_D^{20}$  95.11° (*c* 0.31, CD<sub>3</sub>OD)
  - UV  $\lambda_{\max}^{\text{MeOH}}$  (log  $\epsilon$ ): 284.5 (4.55)
  - HRESIMS *m/z* [M+H]<sup>+</sup> 303.0856 (calcd for C<sub>16</sub>H<sub>15</sub>O<sub>6</sub> 303.0990)
- CD [MeOH, nm ( $\Delta\epsilon$ )] : 276 (-0.39), 299 (-0.13), 328 (+0.21).



**Fig. 1** CD and UV spectrum of compound 1



## Elemental Composition Report

Page 1

### Single Mass Analysis

Tolerance = 100.0 PPM / DBE: min = -1.5, max = 50.0

Element prediction: Off

Number of isotope peaks used for i-FIT = 3

Monoisotopic Mass, Even Electron Ions

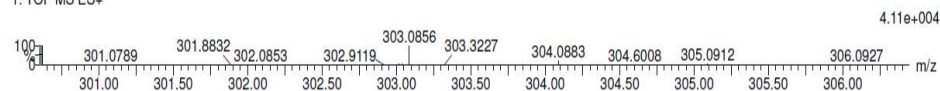
57 formula(e) evaluated with 8 results within limits (up to 100 closest results for each mass)

Elements Used:

C: 0-100 H: 0-100 O: 0-50

SBC444J2\_a 257 (4.416)

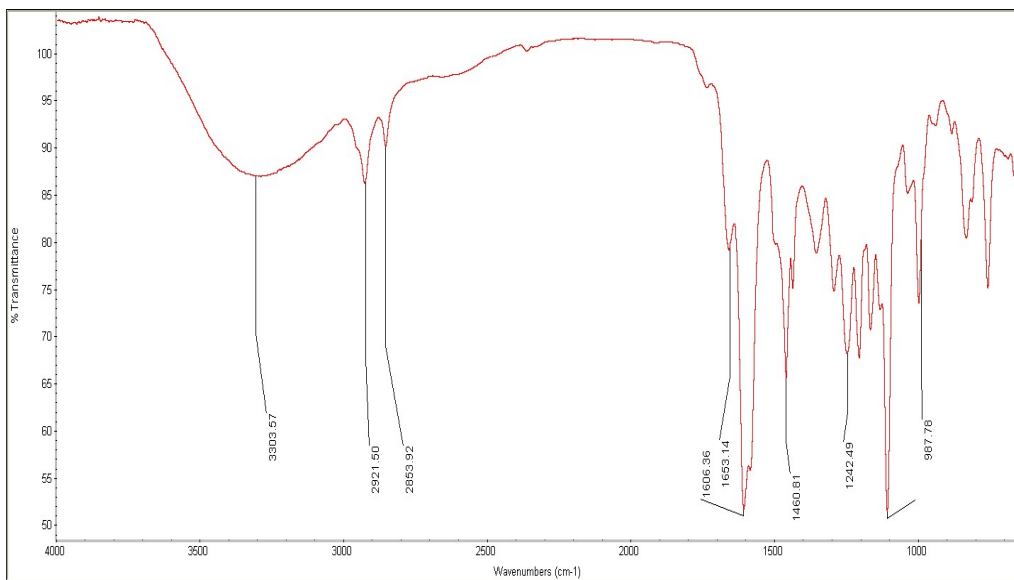
1: TOF MS ES+



Minimum: -1.5  
Maximum: 5.0 100.0 50.0

Mass	Calc. Mass	mDa	PPM	DBE	i-FIT	Norm	Conf (%)	Formula
303.0856	303.0869	-1.3	-4.3	9.5	187.4	0.147	86.32	C16 H15 O6
	303.0810	4.6	15.2	18.5	193.7	6.387	0.17	C23 H11 O
	303.0927	-7.1	-23.4	0.5	197.2	9.875	0.01	C9 H19 O11
	303.0716	14.0	46.2	5.5	194.5	7.242	0.07	C12 H15 O9
	303.1021	-16.5	-54.4	13.5	190.7	3.469	3.12	C20 H15 O3
	303.0657	19.9	65.7	14.5	189.6	2.282	10.20	C19 H11 O4
	303.1080	-22.4	-73.9	4.5	194.0	6.765	0.12	C13 H19 O8
	303.0564	29.2	96.3	1.5	198.5	11.253	0.00	C8 H15 O12

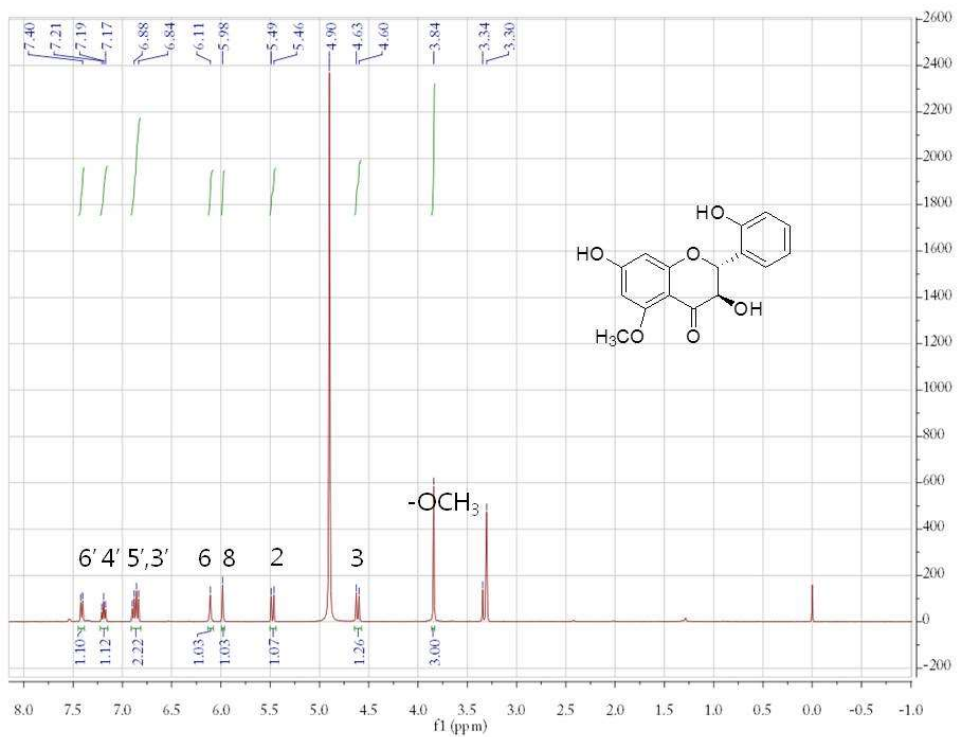
Fig. 2 Mass spectrum of compound 1



**Fig. 3** IR spectrum of compound **1**

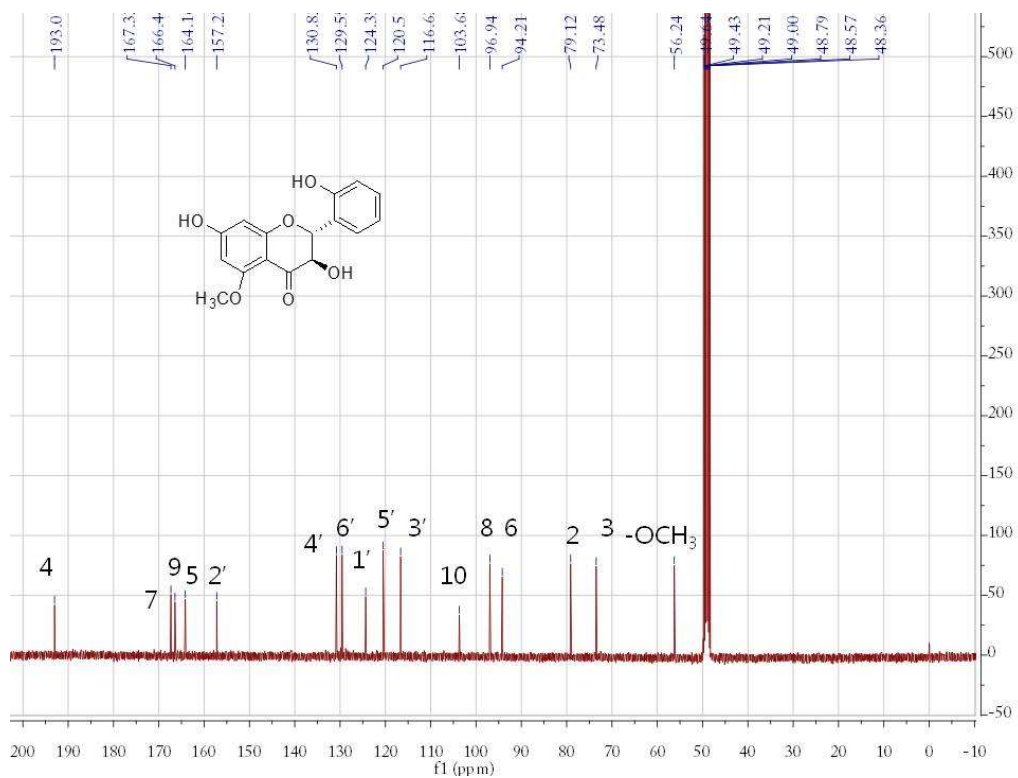
The IR spectrum of compound **1** was recorded by Attenuated Total Reflection (ATR) method in CD<sub>3</sub>OD.

Band assignment: IR spectrum of **1** exhibited typical absorption bands at 3303 cm<sup>-1</sup> due to the stretching vibration of hydroxyl group. The band at 1653 cm<sup>-1</sup> can be attributed to the stretching vibration of carbonyl group.

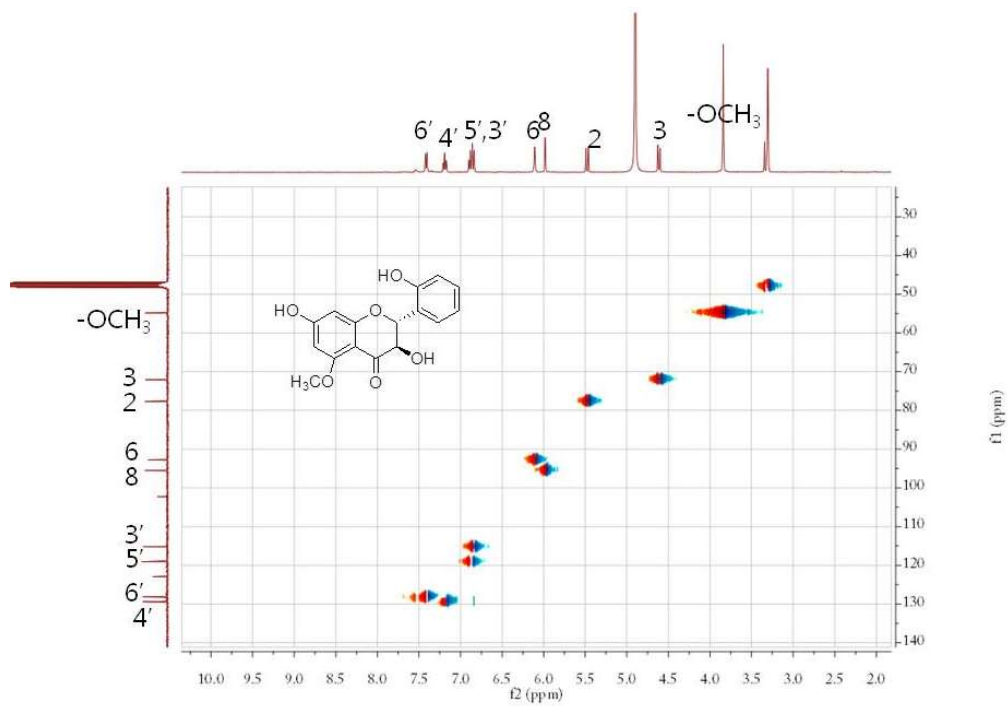


**Fig. 4** <sup>1</sup>H-NMR (400 MHz, CD<sub>3</sub>OD) spectrum of compound 1

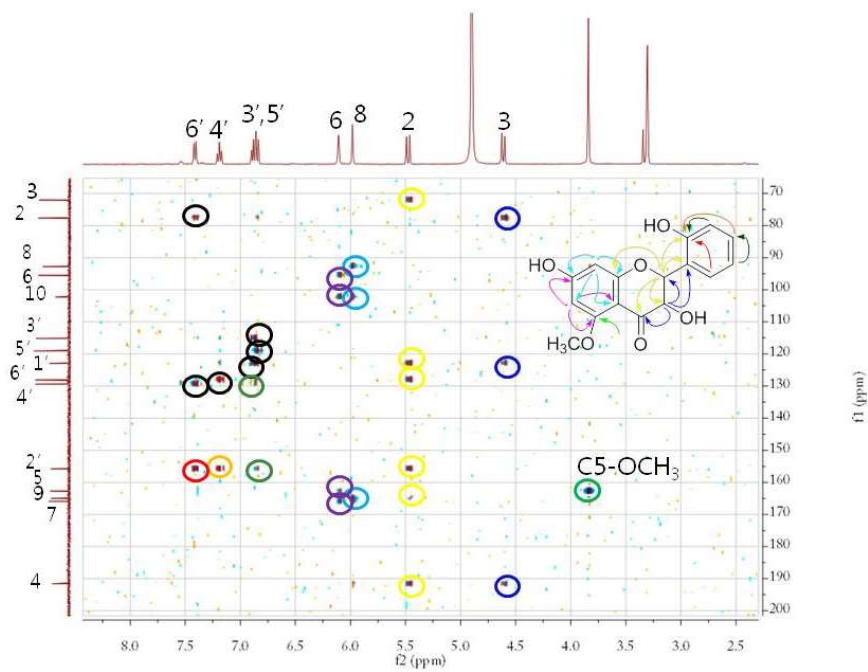




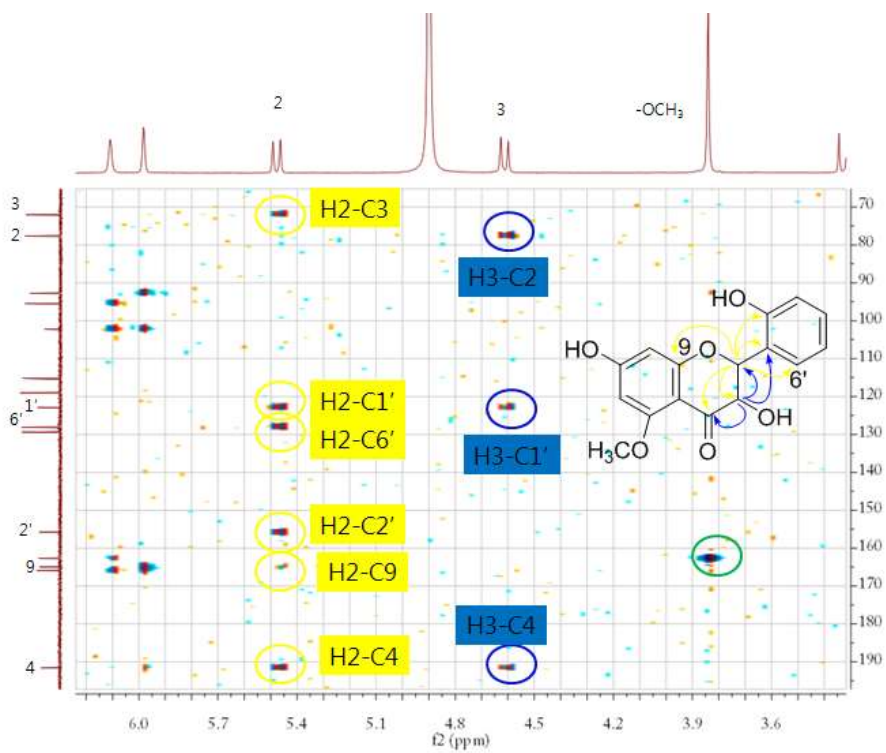
**Fig.5** <sup>13</sup>C-NMR (100 MHz, CD<sub>3</sub>OD) spectrum of compound 1



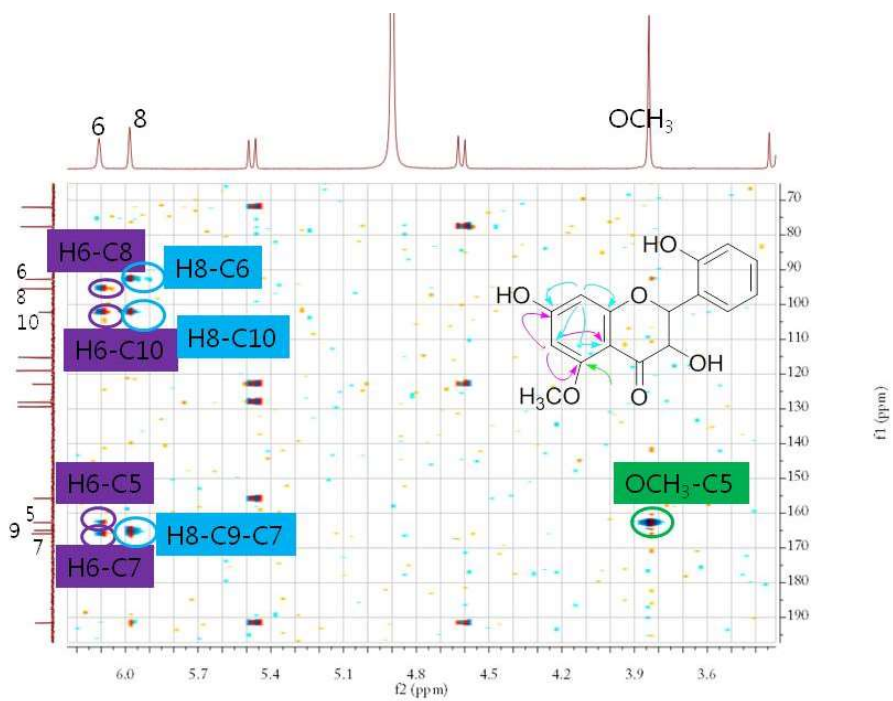
**Fig. 6**  $^1\text{H}$ - $^{13}\text{C}$ -HSQC spectrum of compound 1



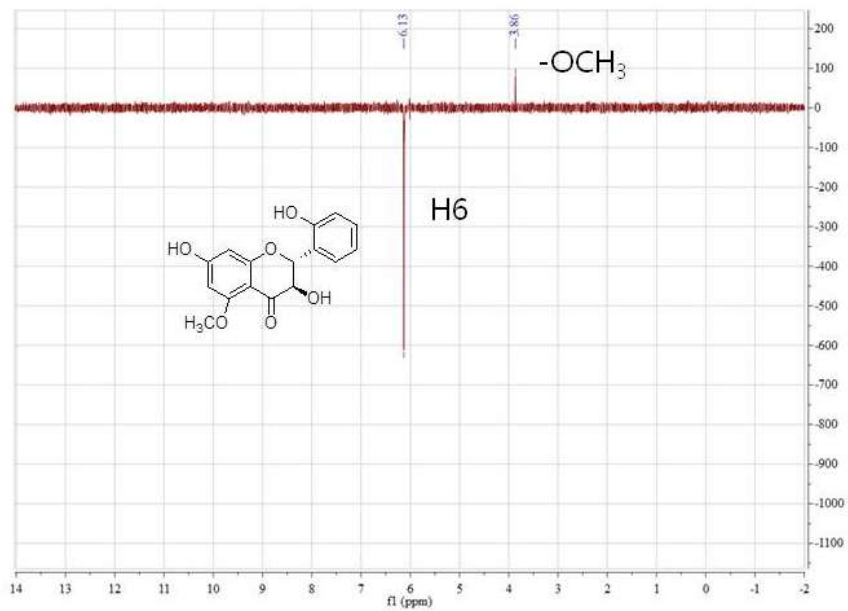
**Fig. 7**  $^1\text{H}$ - $^{13}\text{C}$ -HMBC spectrum of compound **1**



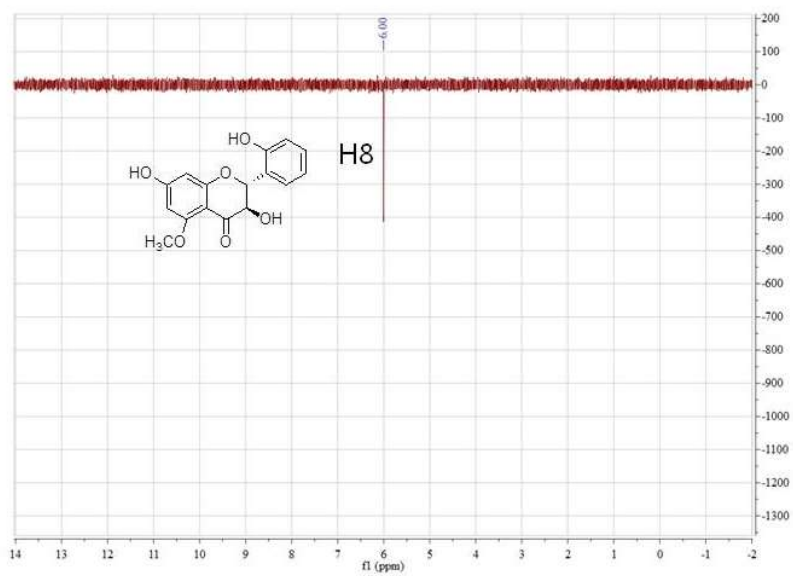
**Fig. 8**  $^1\text{H}$ - $^{13}\text{C}$ -HMBC spectrum of compound **1** (confirmation of correlations of H2 and H3)



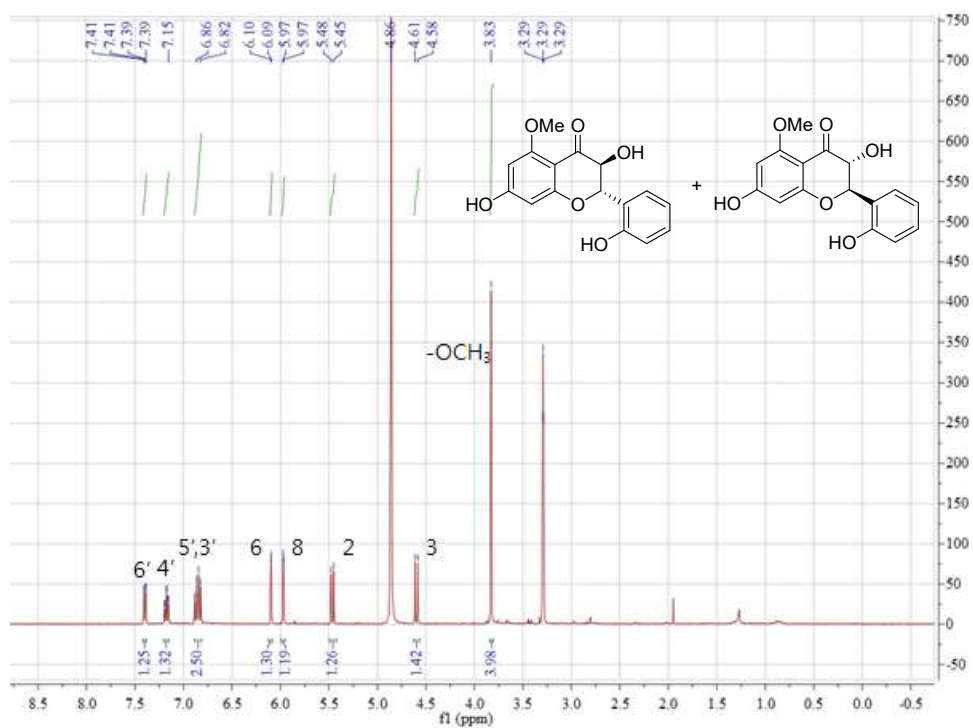
**Fig. 9**  $^1\text{H}$ - $^{13}\text{C}$ -HMBC spectrum of compound **1** (confirmation of correlations of H6 and H8 and methoxyl group)



**Fig. 10** NOESY difference spectrum for H6 irradiation of compound 1.

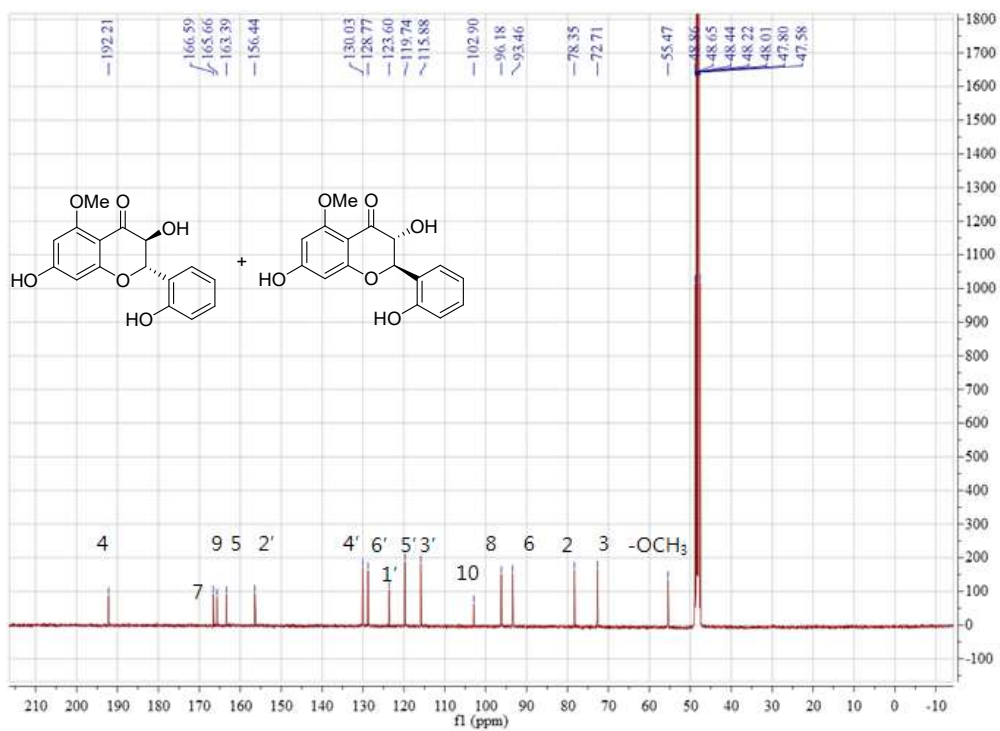


**Fig. 11** NOE difference spectrum for H8 irradiation of compound **1**.

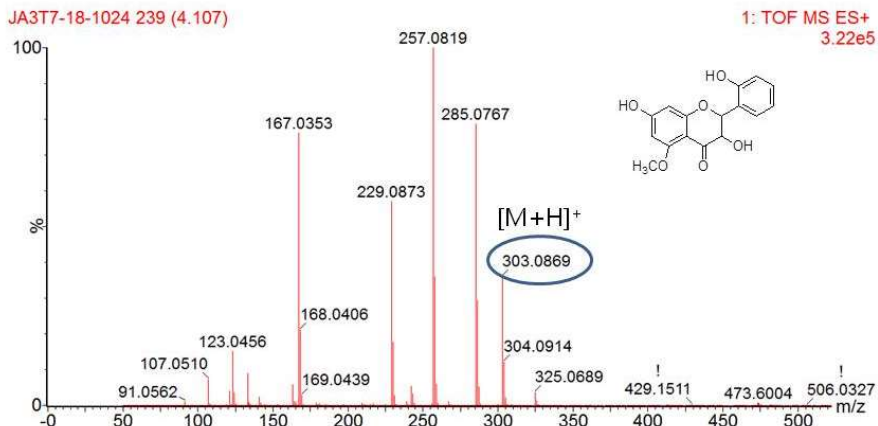


**Fig. 12** <sup>1</sup>H-NMR (400Hz, CD<sub>3</sub>OD) spectrum of compound **1a**





**Fig. 13** <sup>13</sup>C-NMR (100Hz, CD<sub>3</sub>OD) spectrum of compound **1a**



### Elemental Composition Report

Page 1

#### Single Mass Analysis

Tolerance = 100.0 PPM / DBE: min = -1.5, max = 50.0

Element prediction: Off

Number of isotope peaks used for i-FIT = 3

Monoisotopic Mass, Even Electron Ions

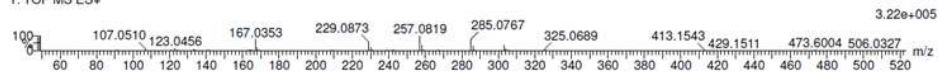
57 formula(e) evaluated with 7 results within limits (up to 100 closest results for each mass)

Elements Used:

C: 0-100 H: 0-100 O: 0-50

JA3T7-18-1024 239 (4.107)

1: TOF MS ES+



Minimum: -1.5  
Maximum: 50.0

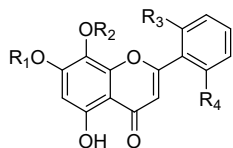
Mass	Calc. Mass	mDa	PPM	DBE	i-FIT	Norm	Conf(%)	Formula
303.0869	303.0869	0.0	0.0	9.5	458.3	3.849	2.13	C16 H15 O6
	303.0927	-5.8	-19.1	0.5	461.6	7.124	0.08	C9 H19 O11
	303.0810	5.9	19.5	18.5	454.7	0.238	78.80	C23 H11 O
	303.1021	-15.2	-50.2	13.5	456.6	2.203	11.05	C20 H15 O3
	303.0716	15.3	50.5	5.5	460.3	5.842	0.29	C12 H15 O9
	303.1080	-21.1	-69.6	4.5	460.3	5.864	0.28	C13 H19 O8
	303.0657	21.2	69.9	14.5	457.0	2.608	7.37	C19 H11 O4

Fig. 14 Mass spectrum of compound 1a

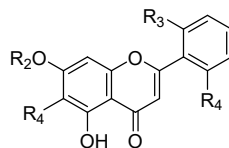
Table of  $^1\text{H}$ -and  $^{13}\text{C}$ -NMR spectroscopic (in  $\text{CD}_3\text{OD}$ ) of Compound **1** and **1a**

No.	<b>1</b>		<b>1a</b>	
	$\delta_{\text{H}}^{\text{a)}$	$\delta_{\text{C}}^{\text{b)}$	$\delta_{\text{H}}^{\text{a)}$	$\delta_{\text{C}}^{\text{b)}$
1	-	-	-	-
2	5.49(1H, d, $J=11.2$ Hz),	79.1	5.48 (d, $J = 11.2$ Hz)	78.3
3	4.62(1H, d, $J=11.2$ Hz)	73.5	4.61 (d, $J = 11.2$ Hz)	72.7
4	-	193.0		192.2
5	-	164.1		164.0
6	6.11 (1H, s)	94.2	6.09 (d, $J = 2.0$ Hz)	93.5
7	-	167.3		166.6
8	5.98 (1H, s)	96.9	5.97 (d, $J = 2.0$ Hz)	96.2
9	-	166.4		165.7
10	-	103.7		102.9
1'	-	124.3		123.6
2'	-	157.2		156.4
3'	6.84 (1H, d, $J=8.4$ Hz)	116.6	6.82 (1H, d, $J=8.5$ Hz)	115.9
4'	7.17 (1H, t, $J=7.6$ Hz)	130.8	7.17 (1H, td, $J=7.6, 1.5$ Hz)	130.3
5'	6.88 (1H, d, $J=7.6$ Hz)	120.5	6.88 (1H, d, $J=7.6$ Hz)	119.8
6'	7.41 (1H, d, $J=7.6$ Hz)	129.5	7.41 (1H, dd, $J=7.6, 1.5$ Hz)	128.8
$\text{OCH}_3$	3.84 (3H, s)	56.2	3.83 (3H s,)	55.5

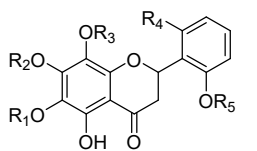
a) Recorded at 400 MHz. b) 100 MHz.



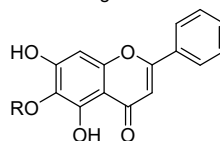
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<b>1</b>	H	H	H	H
<b>2</b>	CH <sub>3</sub>	CH <sub>3</sub>	H	H
<b>3</b>	CH <sub>3</sub>	CH <sub>3</sub>	OH	OH
<b>4</b>	H	CH <sub>3</sub>	OH	H



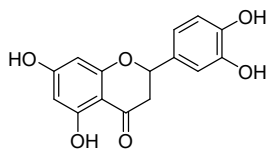
	R <sub>1</sub>	R <sub>2</sub>	R <sub>3</sub>	R <sub>4</sub>
<b>5</b>	H	H	H	H
<b>6</b>	CH <sub>3</sub>	H	H	H
<b>7</b>	H	H	OH	OH
<b>8</b>	CH <sub>3</sub>	CH <sub>3</sub>	OH	OH
<b>9</b>	H	glcA	H	H



	R <sub>1</sub>	R <sub>2</sub>	R <sub>3</sub>	R <sub>4</sub>	R <sub>5</sub>
<b>10</b>	CH <sub>3</sub>	H	H	H	H
<b>11</b>	CH <sub>3</sub>	CH <sub>3</sub>	CH <sub>3</sub>	H	H



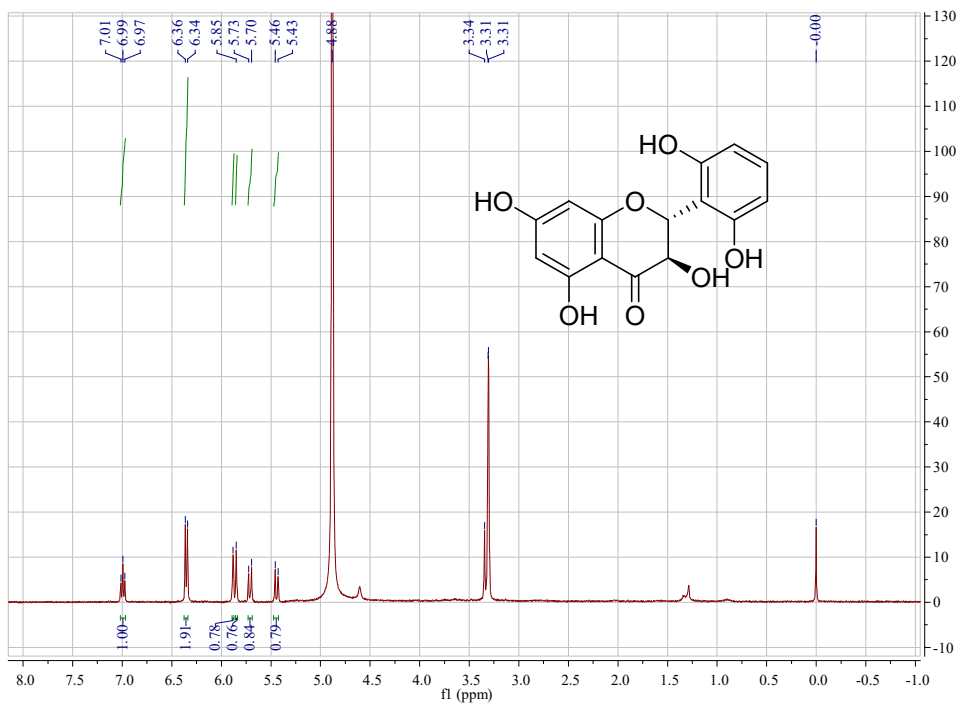
	R
<b>12</b>	H
<b>13</b>	CH <sub>3</sub>



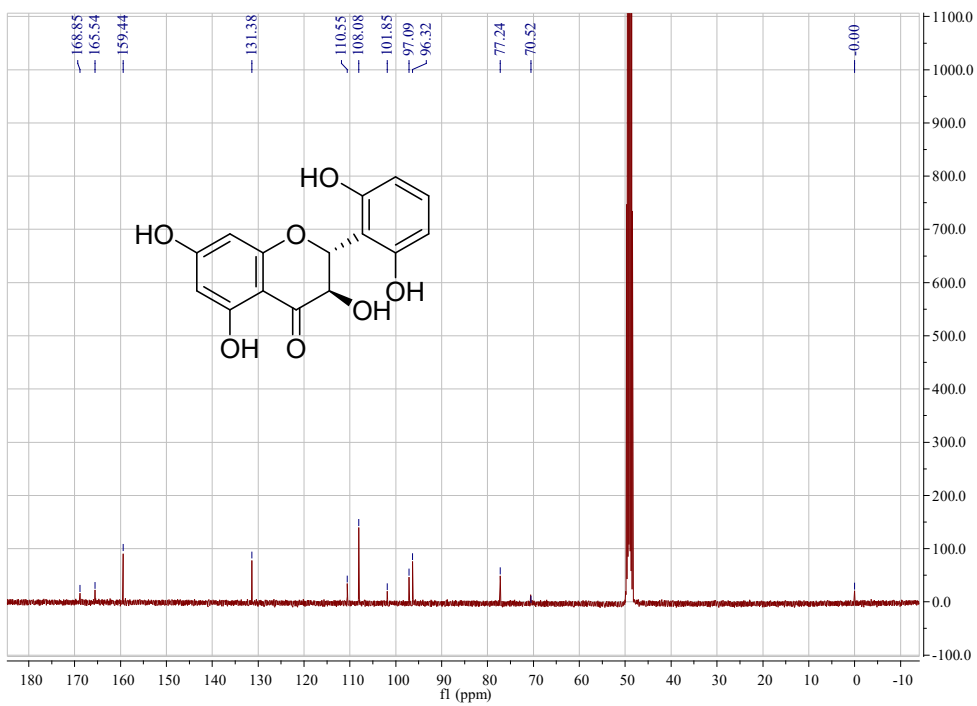
**14**

No.	Compounds
1	norwogonin
2	moslosooflavone
3	viscidulin II
4	scutevulin
5	baicalein
6	oroxylin A
7	2-(2,6-dihydroxyl phenyl)-5,6,7-trihydroxy-flavone
8	mosloflavone
9	baicalin
10	5,7,8-trihydroxy-6-methoxyflavone
11	alnetin
12	dihydrobaicalein
13	dihydrooroxylin A
14	5,7,30,40-tetrahydroxy flavanone

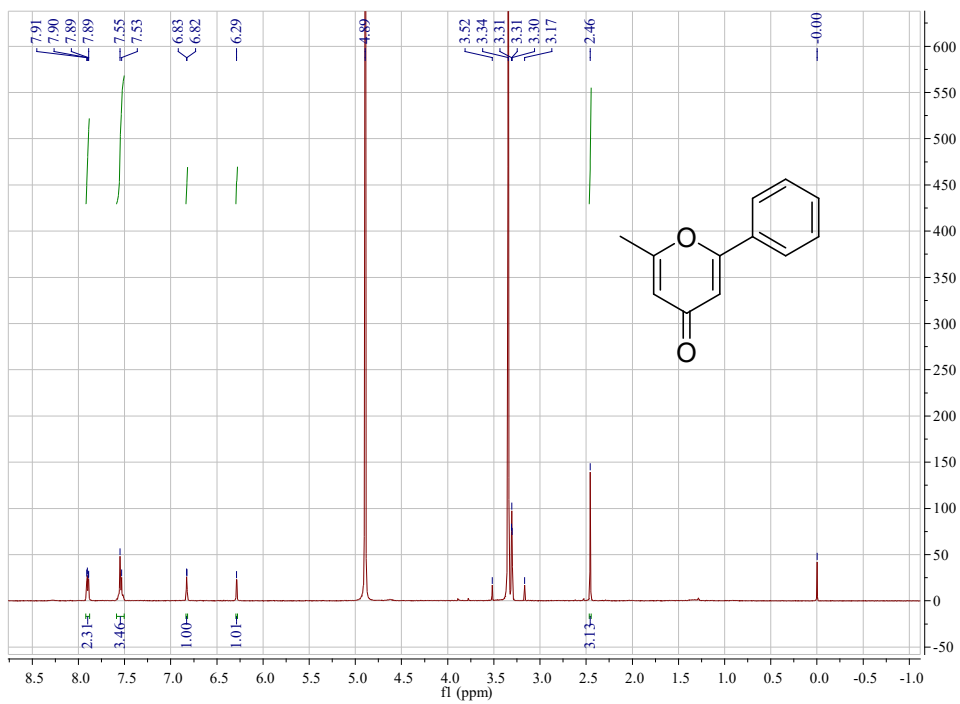
**Fig. 15** Structures of isolated compounds from *S. baicalensis* in the same laboratory



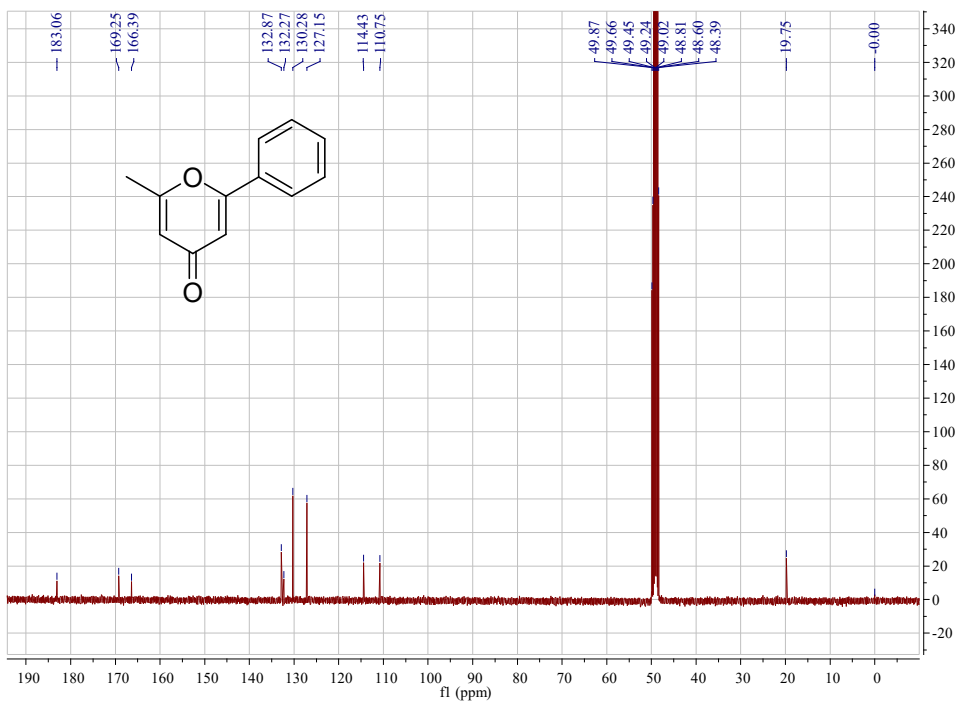
**Fig. 16** <sup>1</sup>H-NMR (400Hz, CD<sub>3</sub>OD) spectrum of compound 2



**Fig. 17**  $^{13}\text{C}$ -NMR (100Hz,  $\text{CD}_3\text{OD}$ ) spectrum of compound **2**

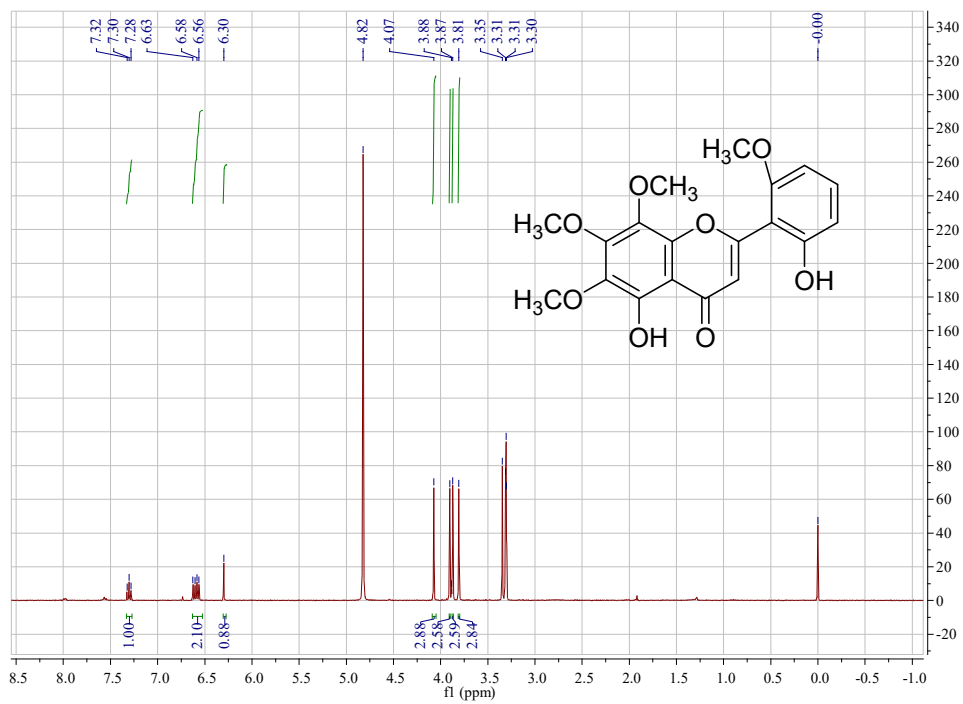


**Fig. 18**  $^1\text{H-NMR}$  (400Hz,  $\text{CD}_3\text{OD}$ ) spectrum of compound **3**

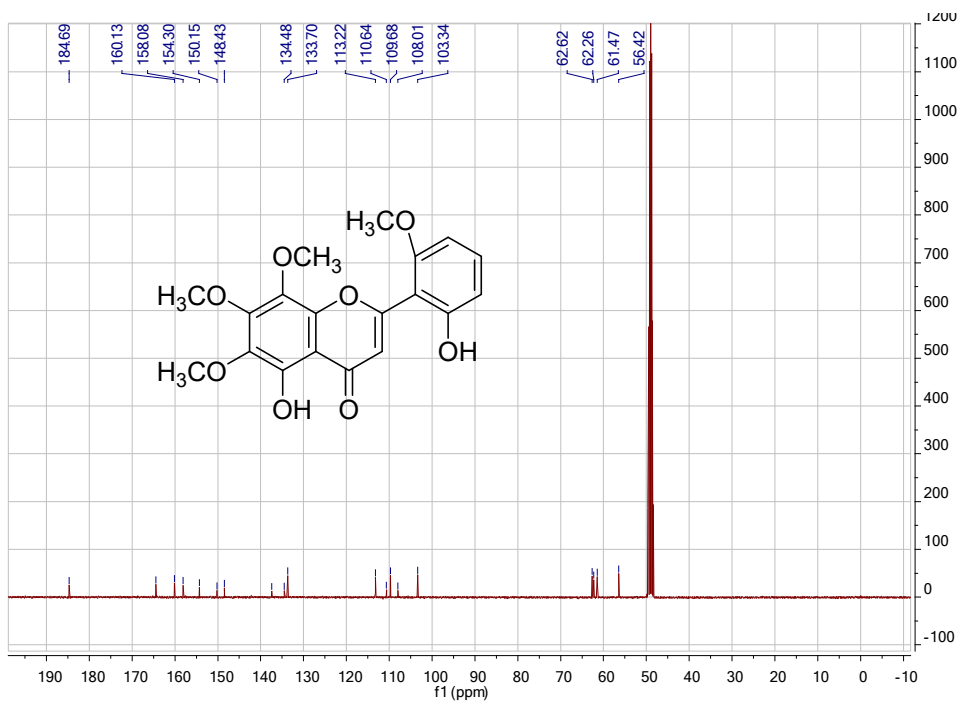


**Fig. 19** <sup>13</sup>C-NMR (100Hz, CD<sub>3</sub>OD) spectrum of compound 3

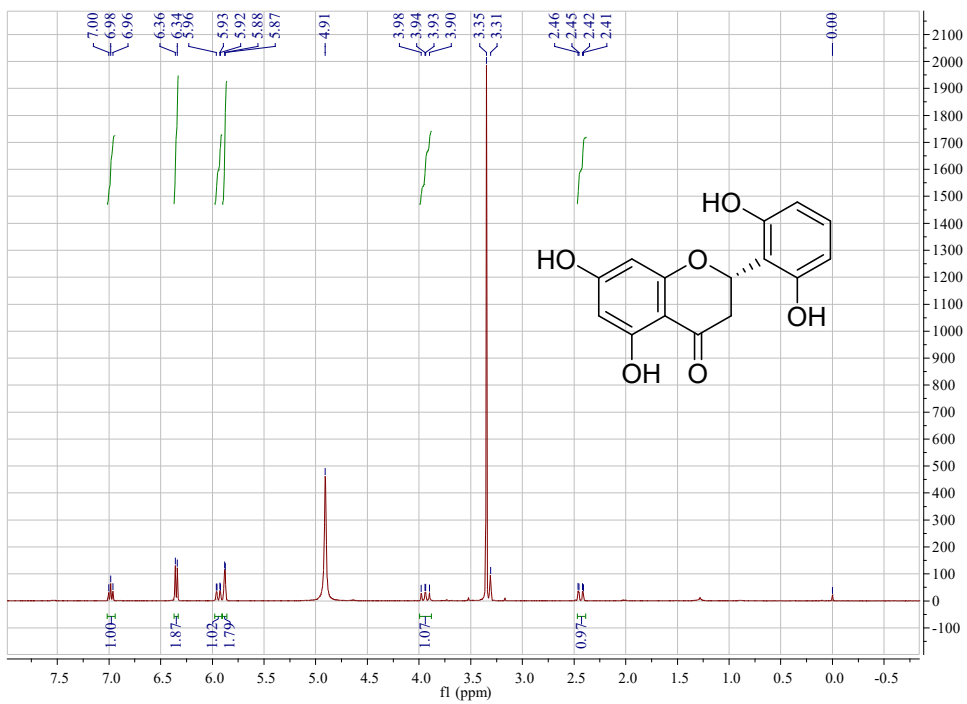




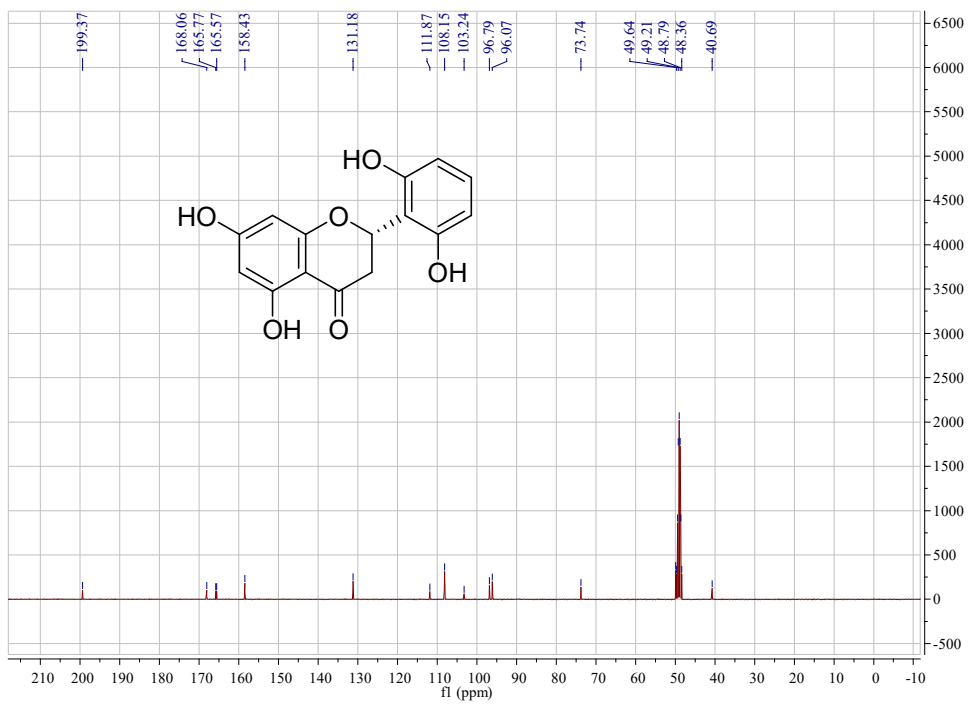
**Fig. 20**  $^1\text{H-NMR}$  (400Hz,  $\text{CD}_3\text{OD}$ ) spectrum of compound 4



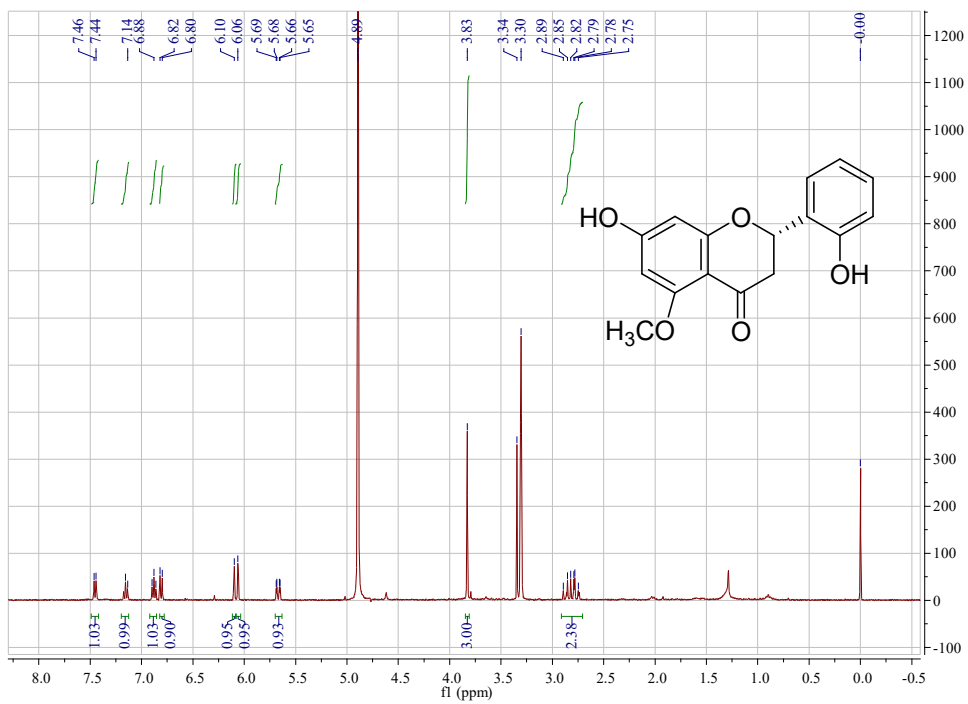
**Fig. 21** <sup>13</sup>C-NMR (100Hz, CD<sub>3</sub>OD) spectrum of compound 4



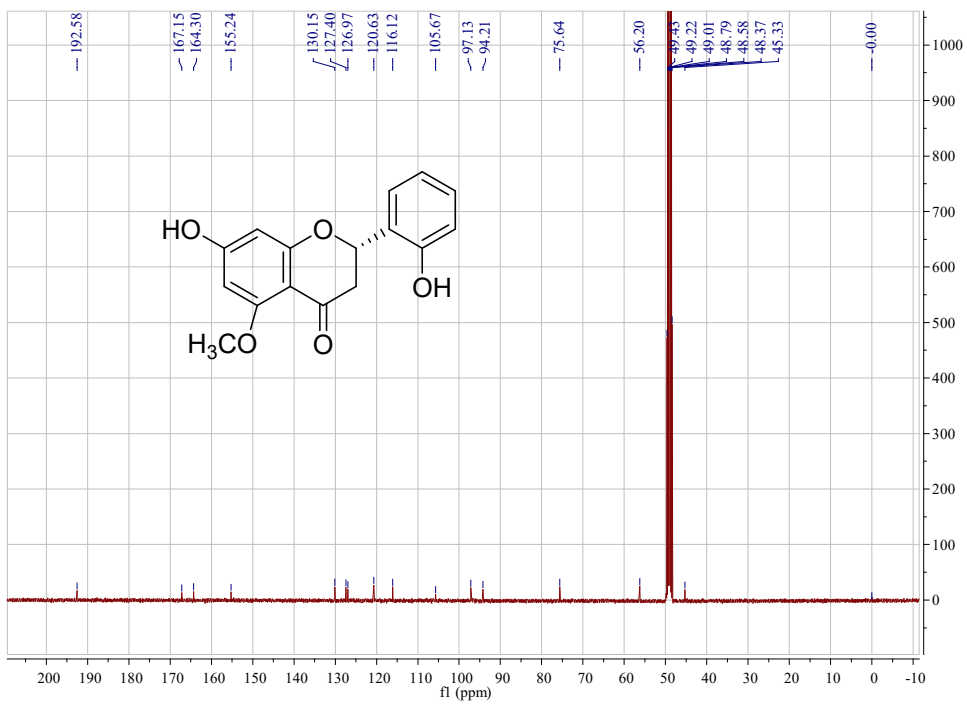
**Fig. 22**  $^1\text{H-NMR}$  (400Hz,  $\text{CD}_3\text{OD}$ ) spectrum of compound **5**



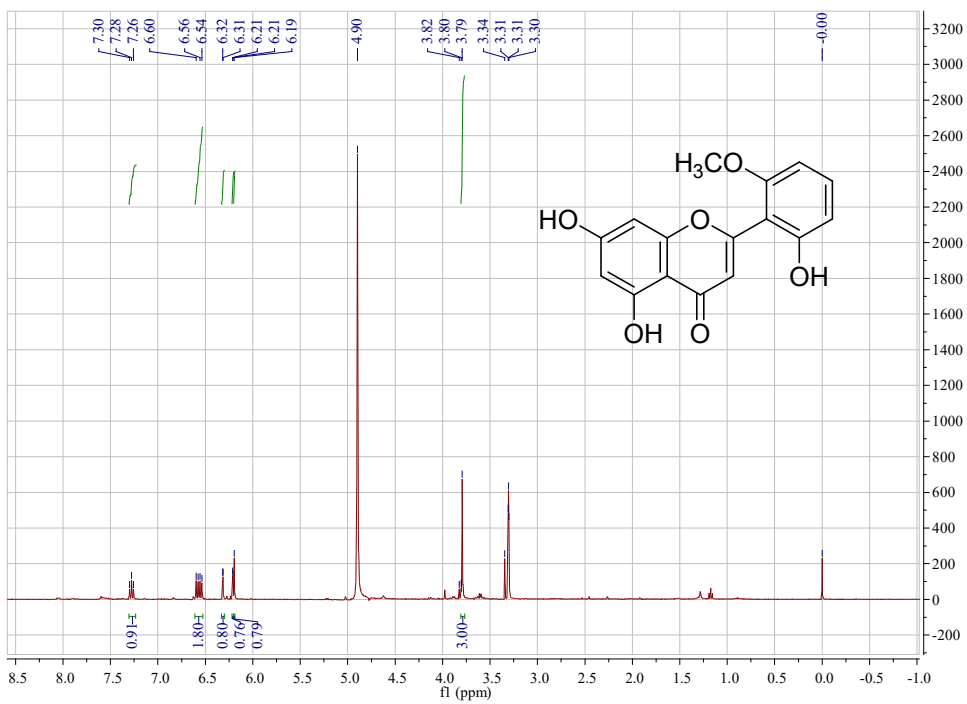
**Fig. 23**  $^{13}\text{C}$ -NMR (100Hz,  $\text{CD}_3\text{OD}$ ) spectrum of compound **5**



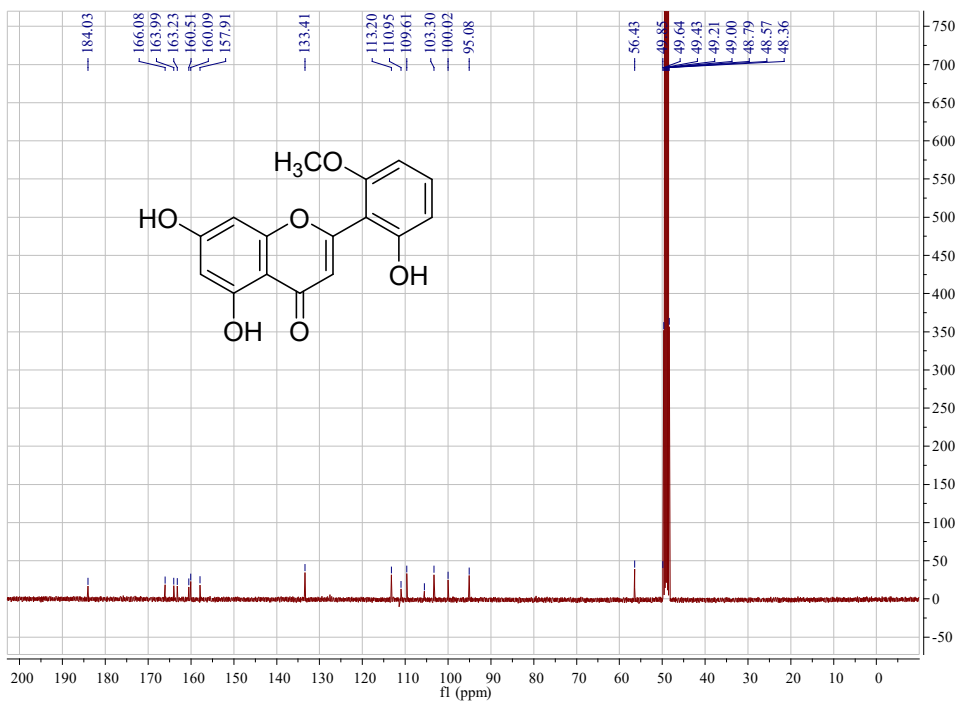
**Fig. 24**  $^1\text{H-NMR}$  (400Hz,  $\text{CD}_3\text{OD}$ ) spectrum of compound **6**



**Fig. 25**  $^{13}\text{C}$ -NMR (100Hz,  $\text{CD}_3\text{OD}$ ) spectrum of compound 6

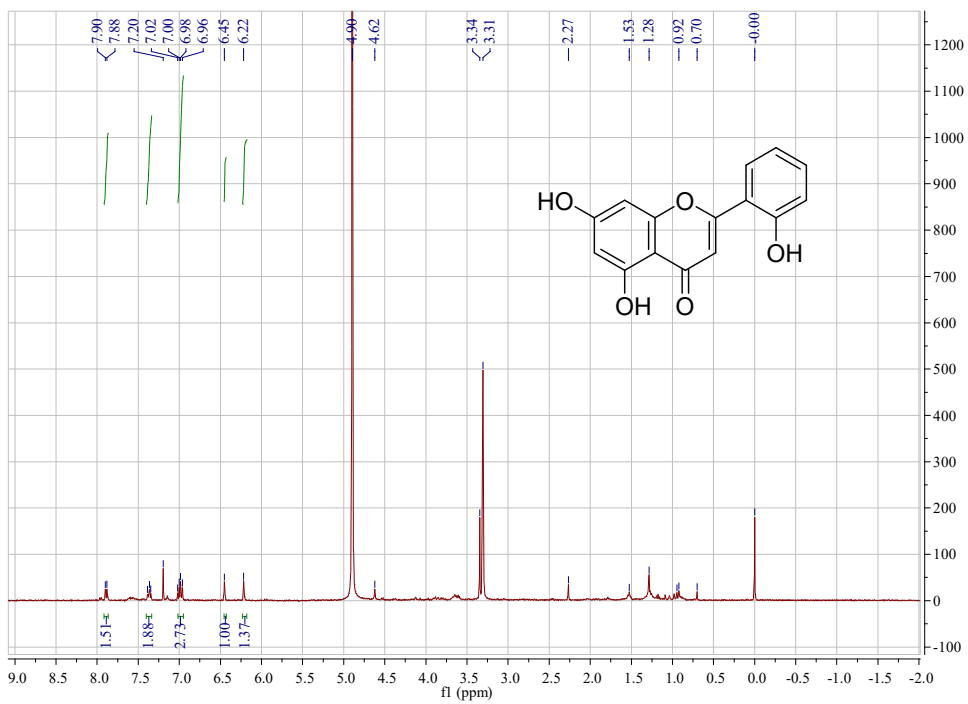


**Fig. 26**  $^1\text{H-NMR}$  (400Hz,  $\text{CD}_3\text{OD}$ ) spectrum of compound 7

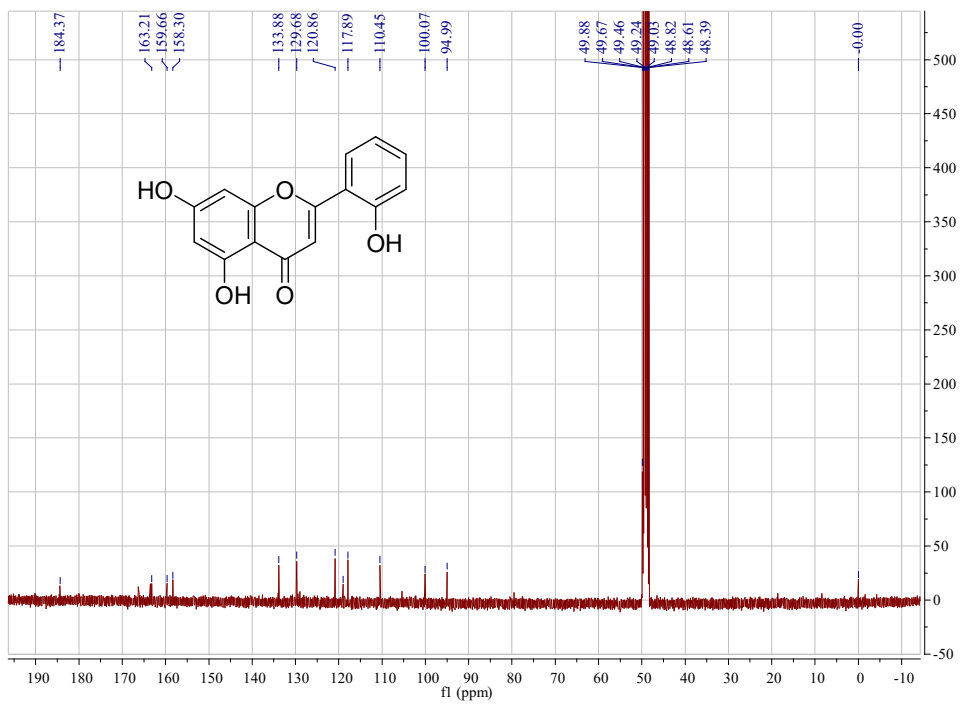


**Fig. 27** <sup>13</sup>C-NMR (100Hz, CD<sub>3</sub>OD) spectrum of compound 7

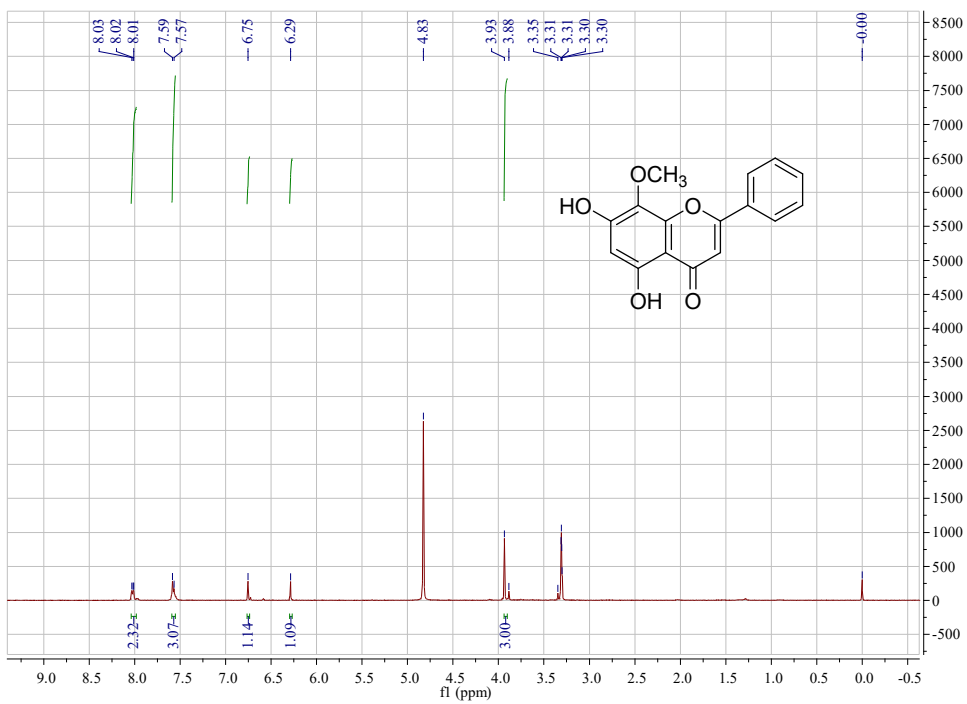




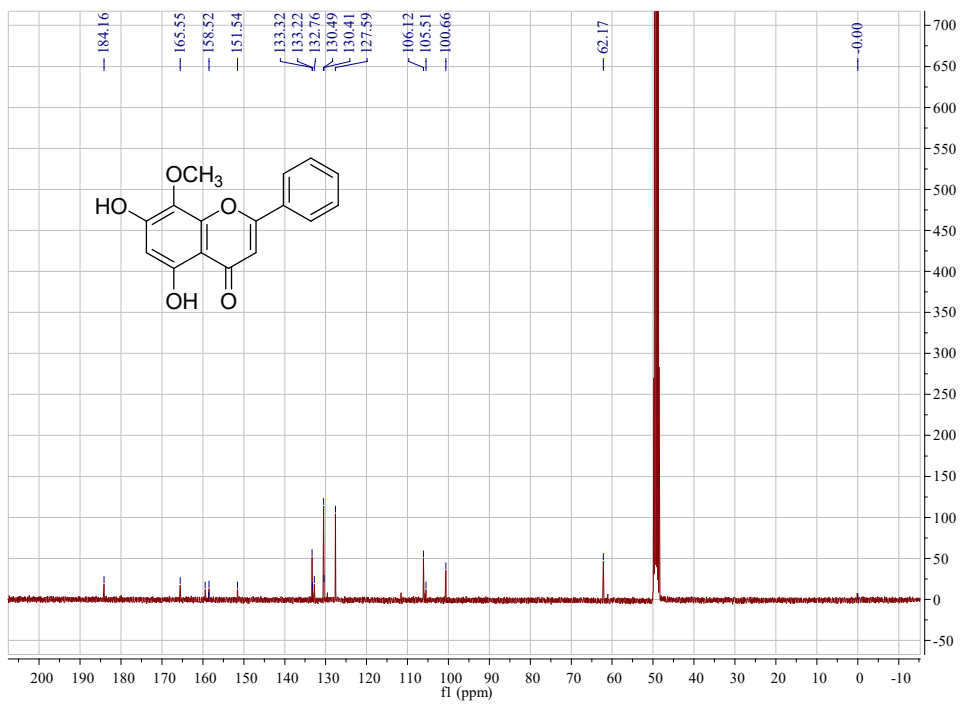
**Fig. 28**  $^1\text{H-NMR}$  (400Hz,  $\text{CD}_3\text{OD}$ ) spectrum of compound **8**



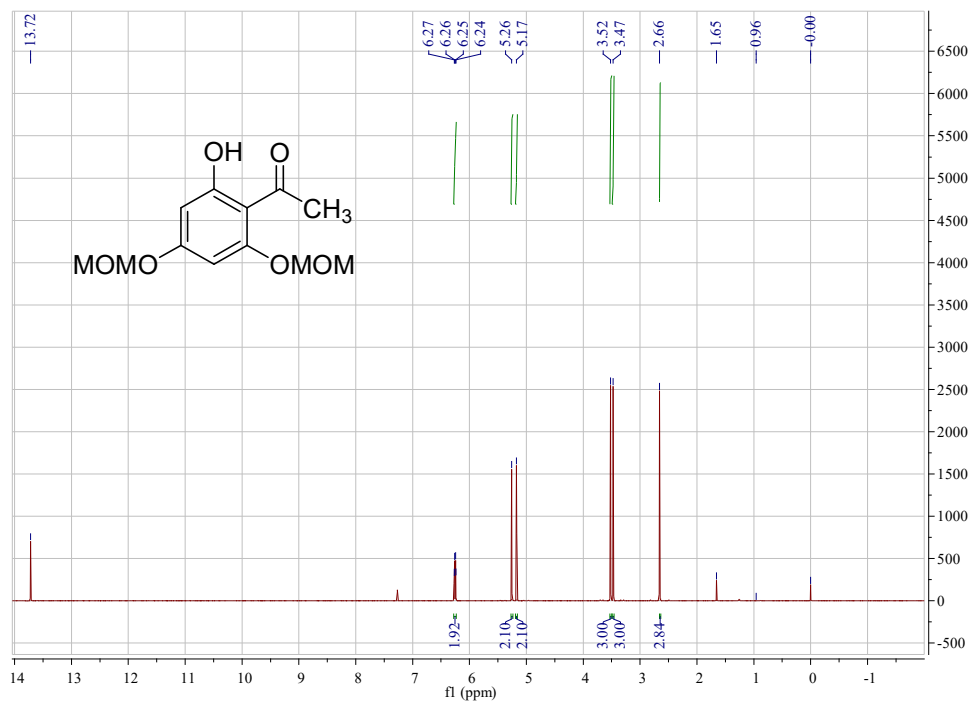
**Fig. 29**  $^{13}\text{C}$ -NMR (100Hz,  $\text{CD}_3\text{OD}$ ) spectrum of compound **8**



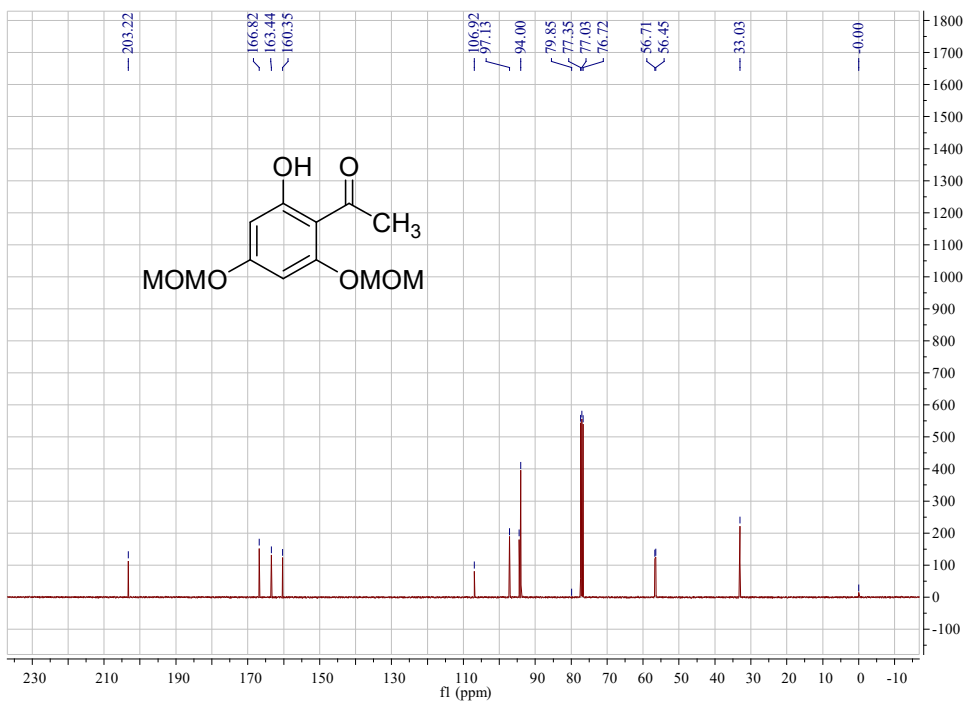
**Fig. 30**  $^1\text{H-NMR}$  (400Hz,  $\text{CD}_3\text{OD}$ ) spectrum of compound **9**



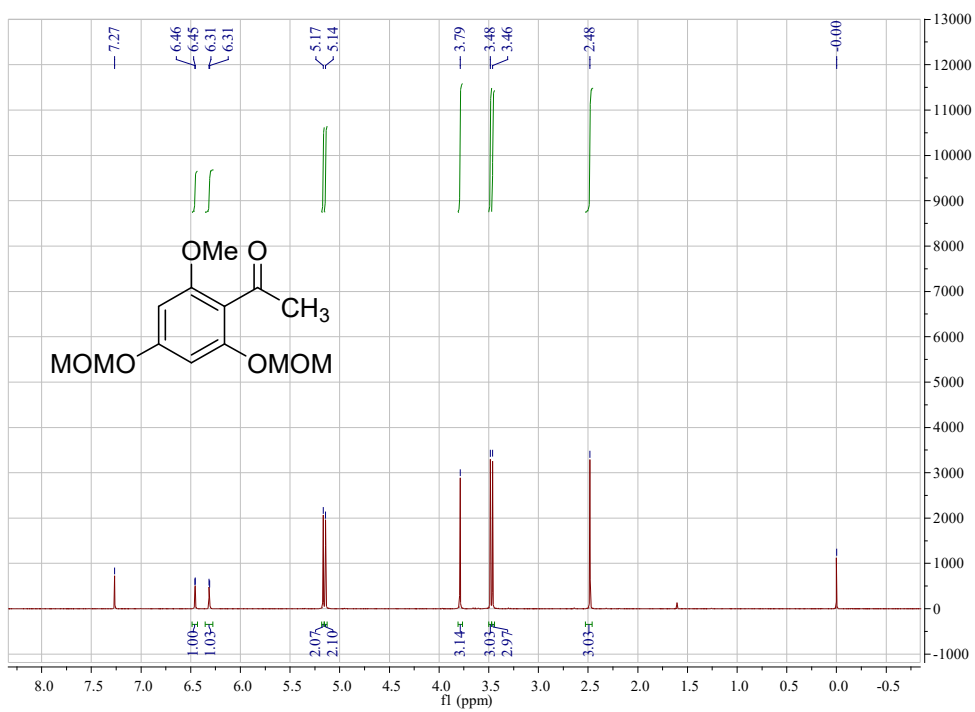
**Fig. 31**  $^{13}\text{C-NMR}$  (100Hz,  $\text{CD}_3\text{OD}$ ) spectrum of compound **9**



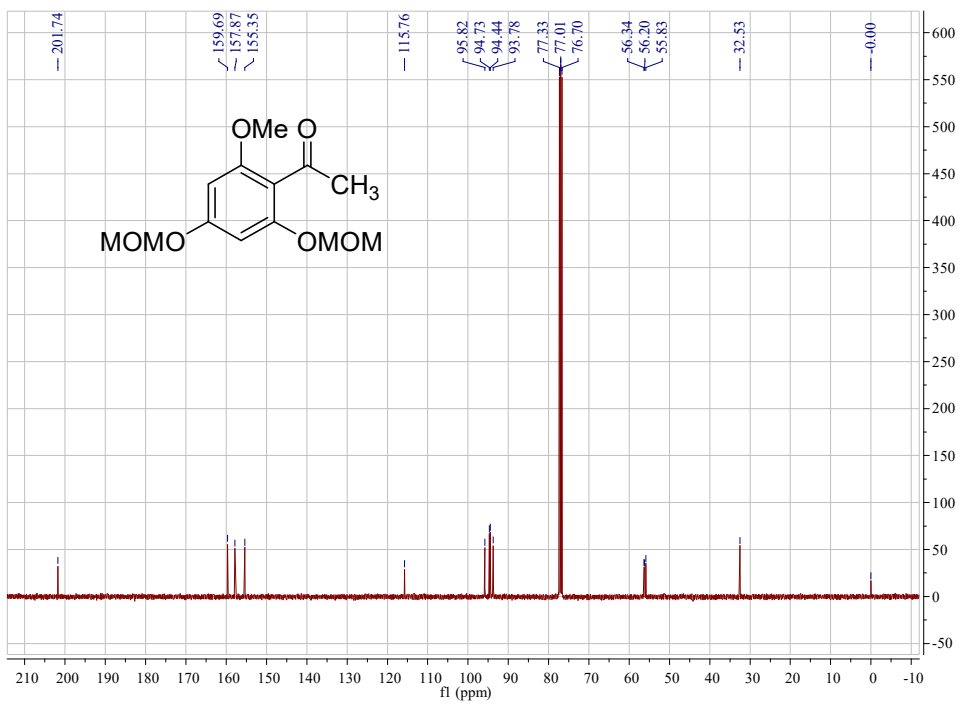
**Fig. 32** <sup>1</sup>H-NMR (400Hz, CD<sub>3</sub>OD) spectrum of compound **11**



**Fig. 33**  $^{13}\text{C}$ -NMR (100Hz,  $\text{CD}_3\text{OD}$ ) spectrum of compound **11**

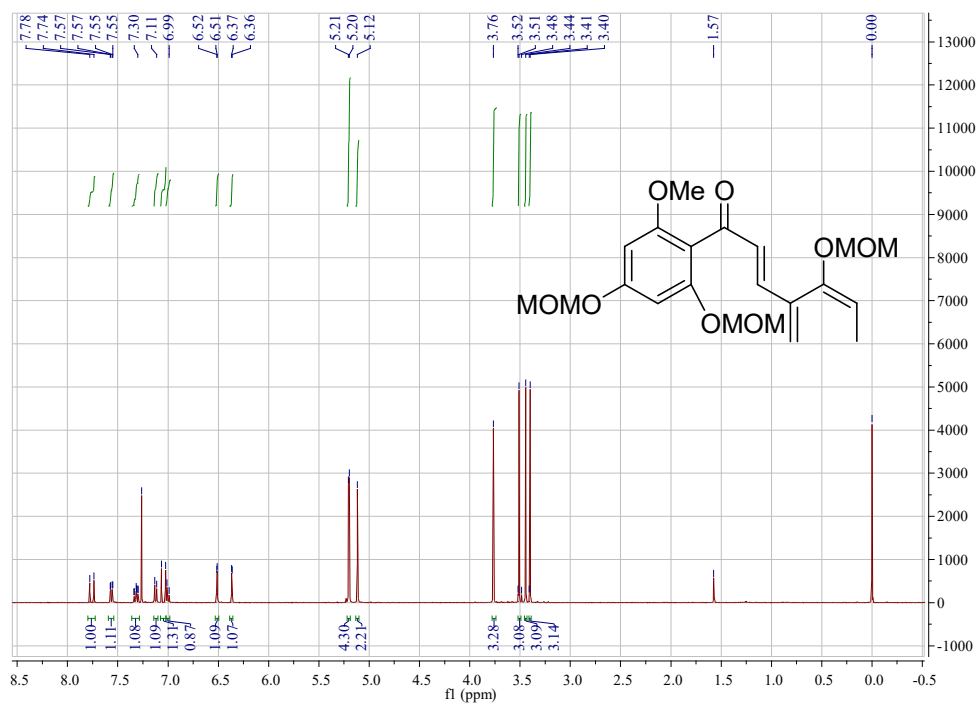


**Fig. 34**  $^1\text{H-NMR}$  (400Hz,  $\text{CD}_3\text{OD}$ ) spectrum of compound **12**

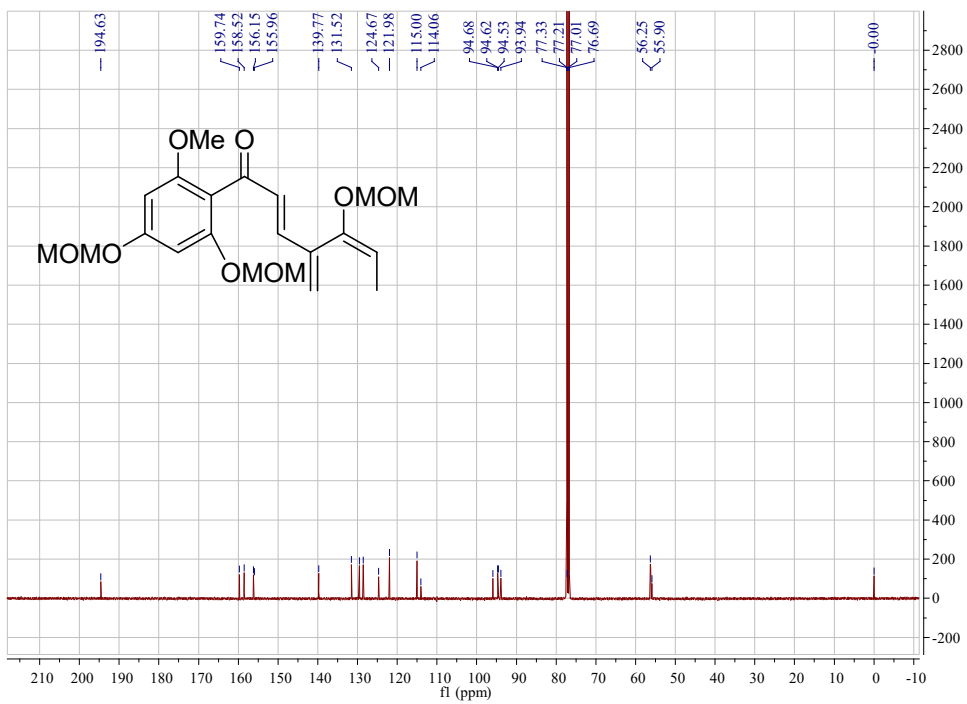


**Fig. 35**  $^{13}\text{C}$ -NMR (100Hz,  $\text{CD}_3\text{OD}$ ) spectrum of compound 12

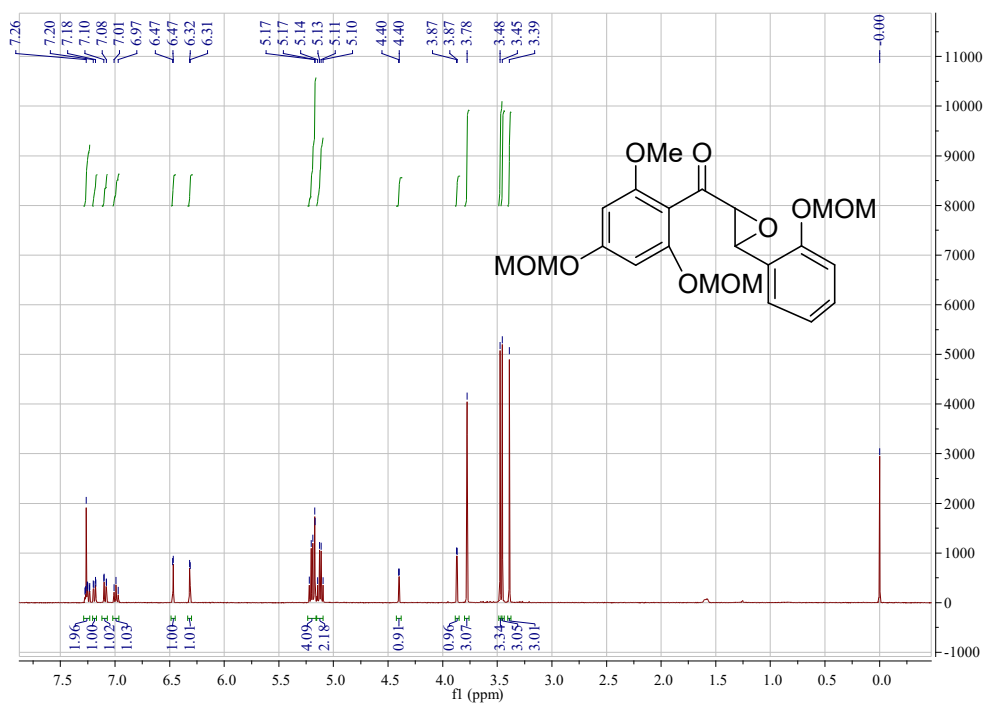




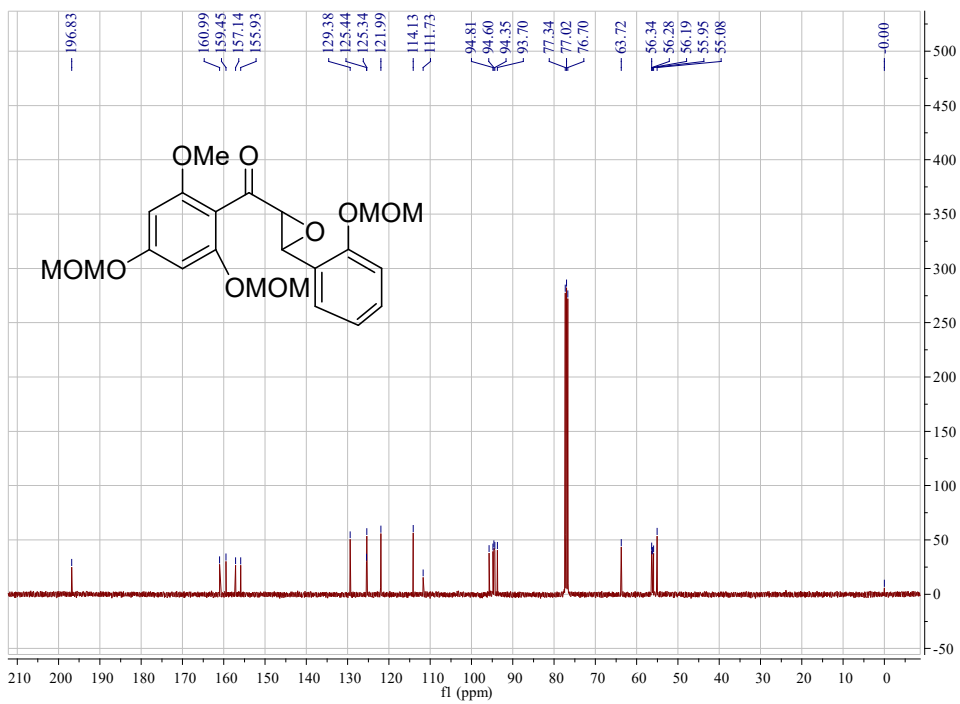
**Fig. 36**  $^1\text{H-NMR}$  (400Hz,  $\text{CD}_3\text{OD}$ ) spectrum of compound 14



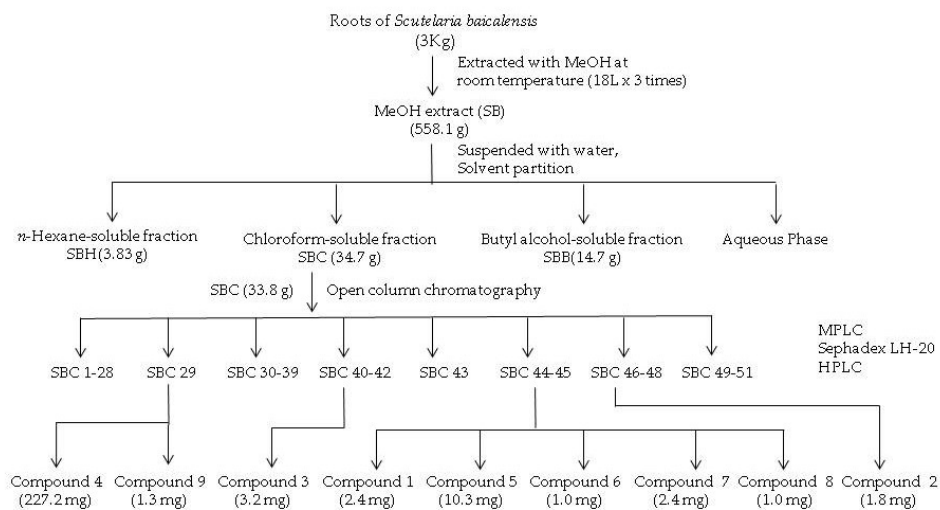
**Fig. 37** <sup>13</sup>C-NMR (100Hz, CD<sub>3</sub>OD) spectrum of compound 14



**Fig. 38** <sup>1</sup>H-NMR (400Hz, CD<sub>3</sub>OD) spectrum of compound **15**



**Fig. 39**  $^{13}\text{C-NMR}$  (100Hz,  $\text{CD}_3\text{OD}$ ) spectrum of compound 15



**Fig. 40** Isolation scheme of *S. baicalensis*.