

Supporting Information

Electrochemical Reduction of CO₂ to Formate on Easily Prepared Carbon-Supported Bi Nanoparticles

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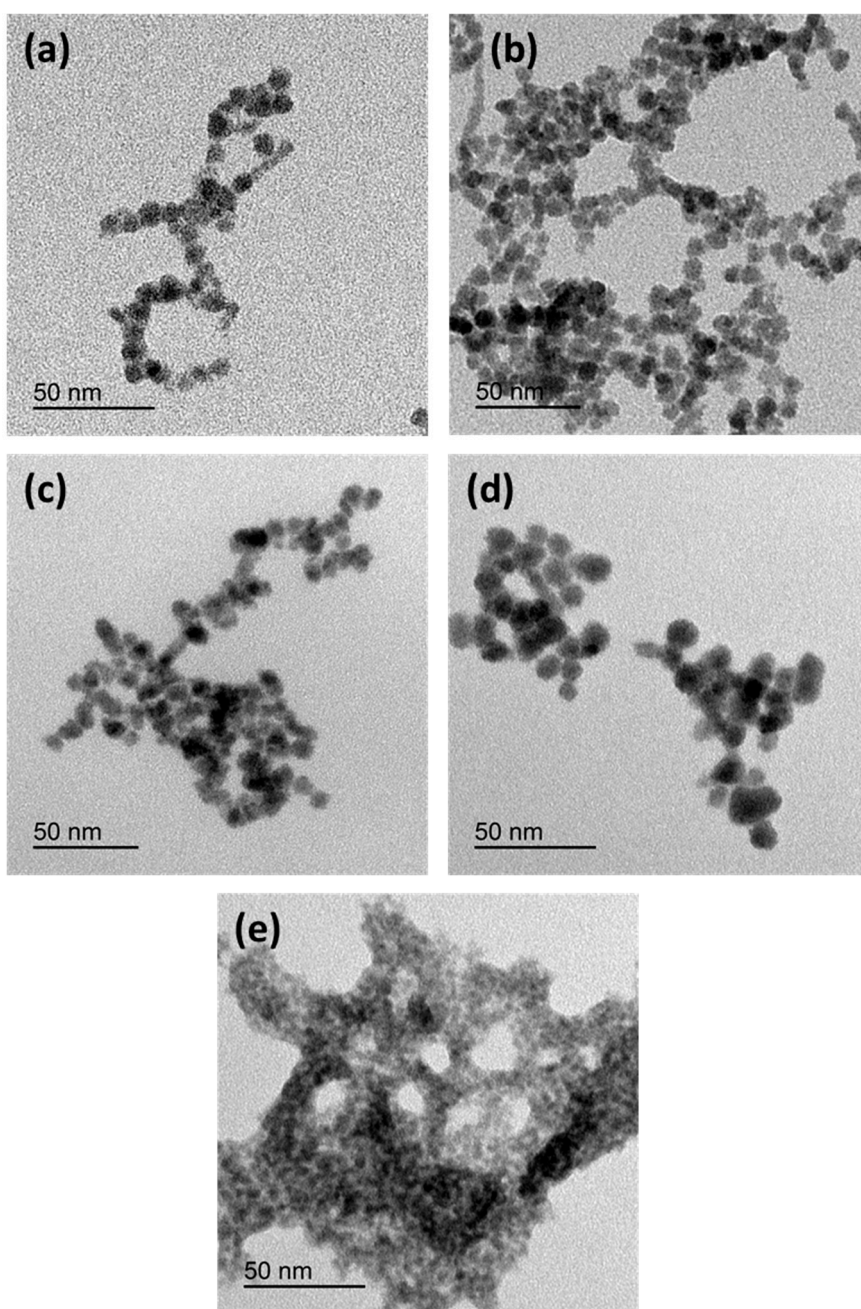


Figure S1. TEM images of unsupported BiNPs synthesized from different stoichiometric PVP to BiCl₃ ratios of (a) 10, (b) 5, (c) 2, (d) 1, and (e) 0.

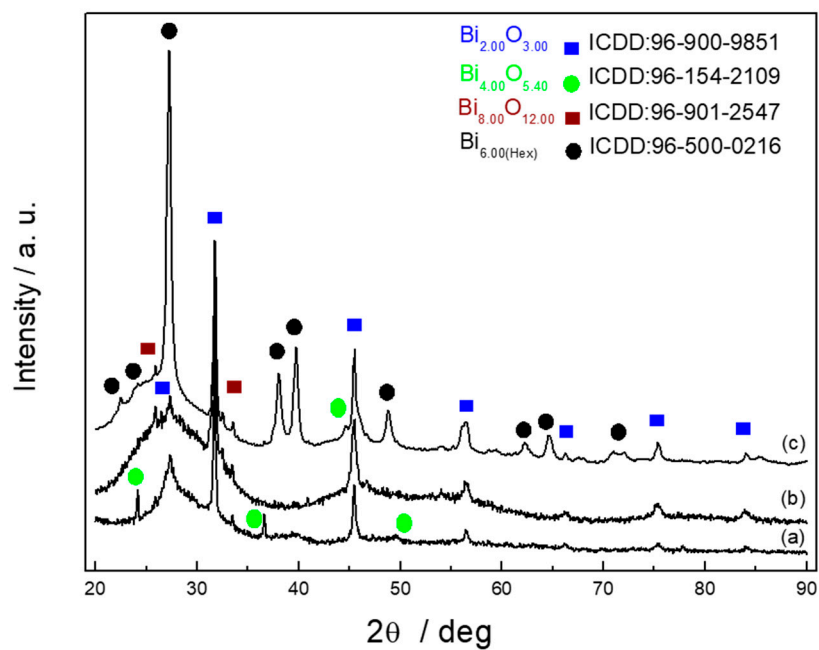


Figure S2. XRD diffractograms of BiNPs prepared with a PVP to Bi ratio of 1: (a) unsupported BiNPs and (b)Bi/C. (c) XRD diffractogram of unsupported BiNPs prepared with a PVP:Bi ratio of 0:1.

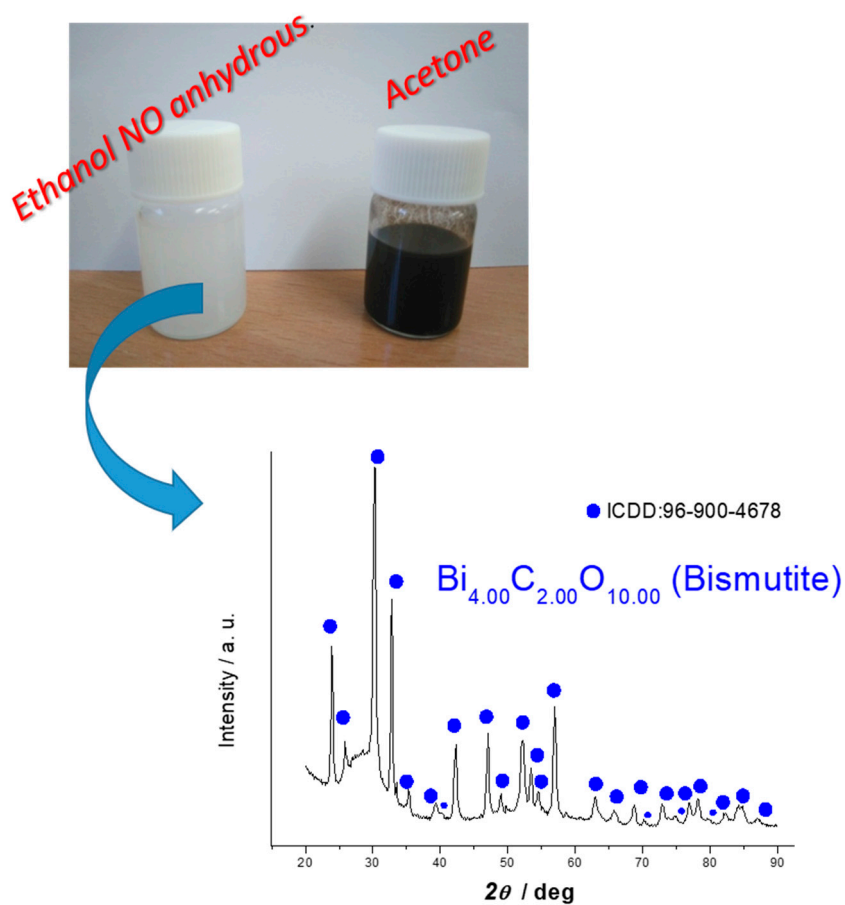


Figure S3. XRD diffractogram of BiNPs prepared with a PVP to Bi ratio of 1 and washed with non-anhydrous ethanol instead of acetone.

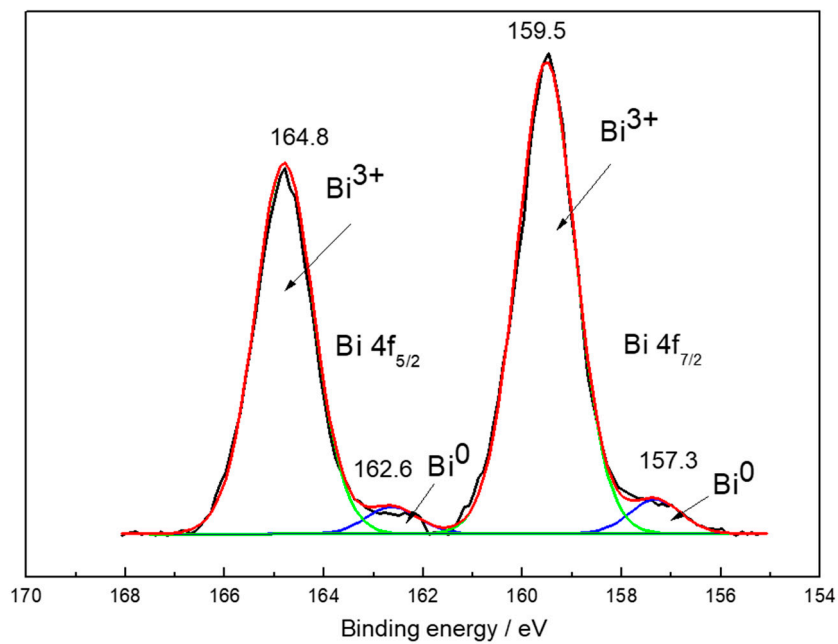


Figure S4. High resolution XPS spectra recorded of Bi 4f region of the Bi/C electrodes.

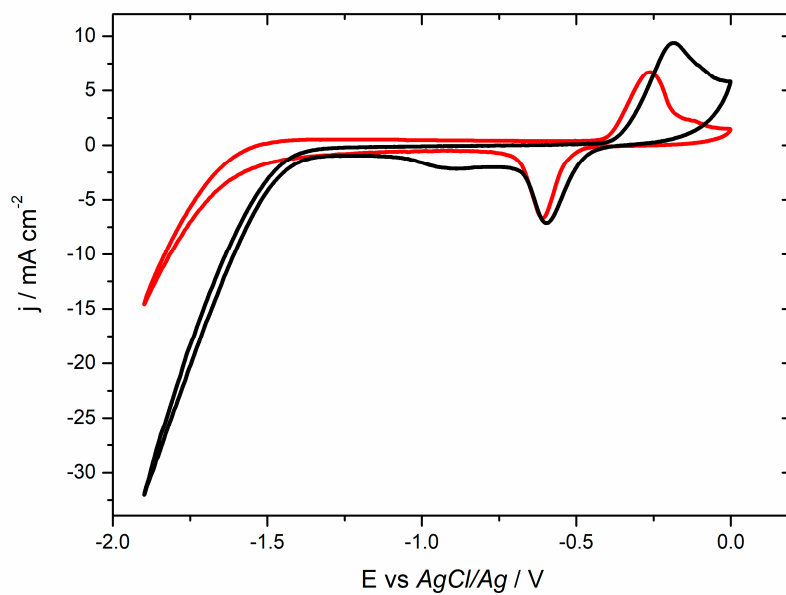


Figure S5. Cyclic voltammograms obtained in Ar-saturated 0.5 M KHCO_3 solution at a scan rate of 50 mV s^{-1} saturated with a (red) Bi/C electrode (Bi loading: 0.1 mg cm^{-2}) and with a (black) massive Bi rod.

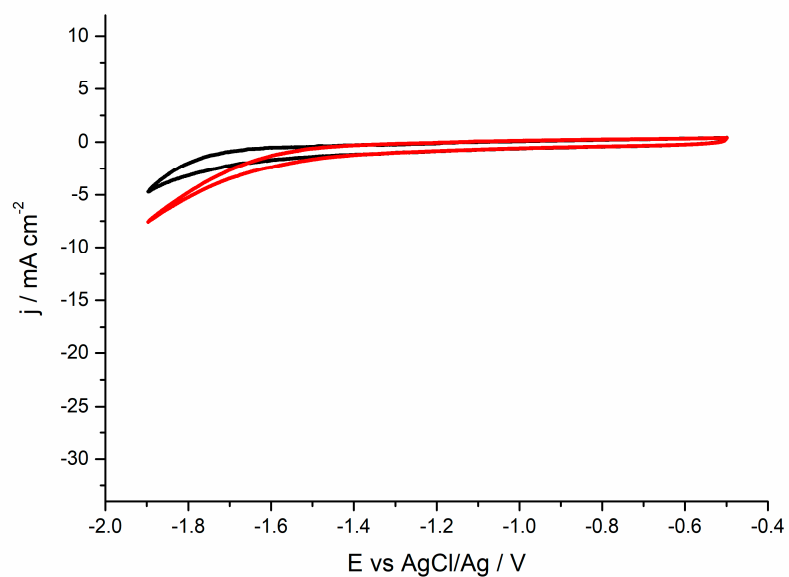


Figure S6. Cyclic voltammograms in Ar (black) and CO₂ (red) saturated 0.5 M KHCO₃ solution at a scan rate of 50 mV s⁻¹ with a Vulcan XC-72R carbon electrode (without Bi).

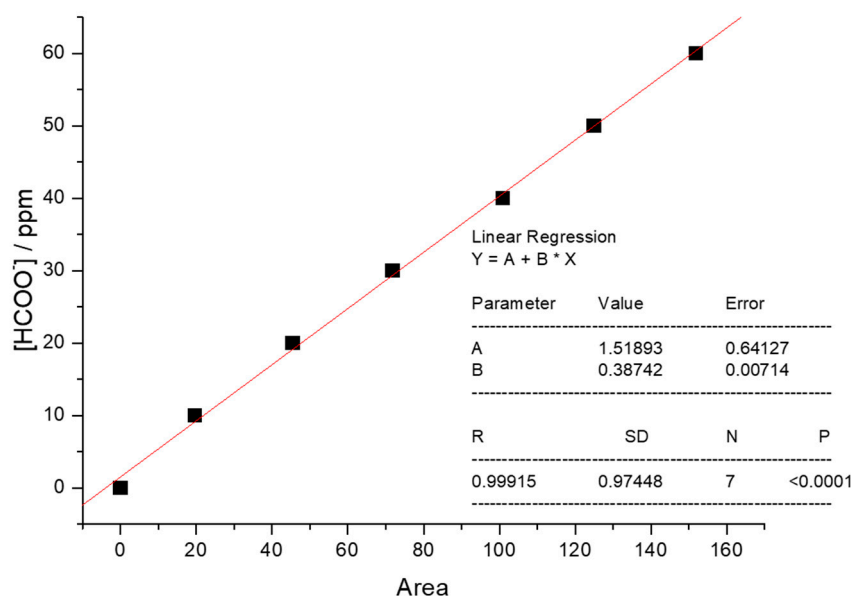


Figure S7. Formate calibration curve obtained from ion chromatography analysis.

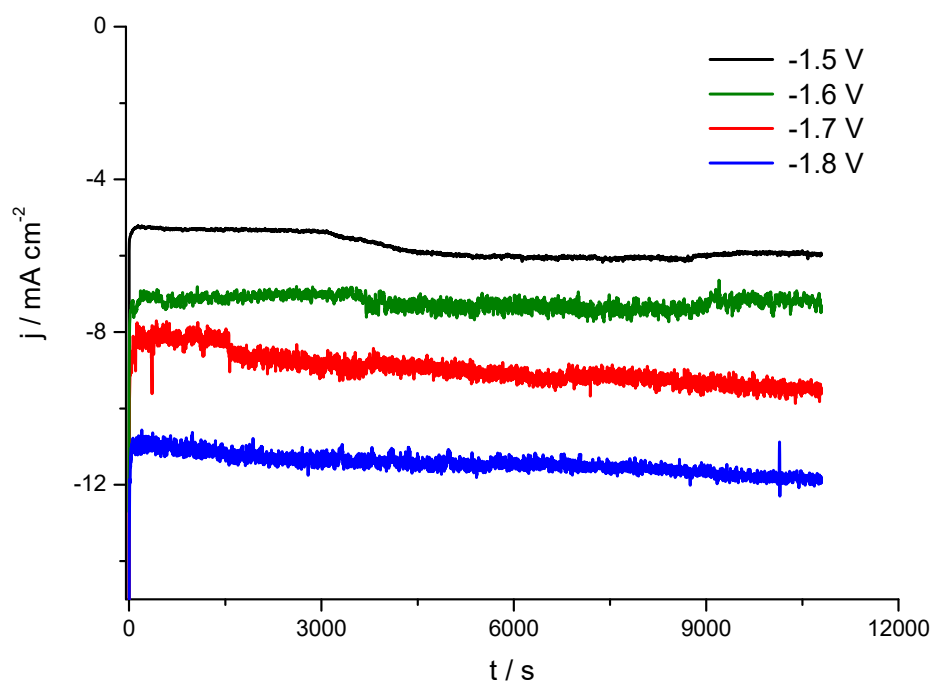


Figure S8. Chronoamperometric measurements at relevant potentials for 3 h.

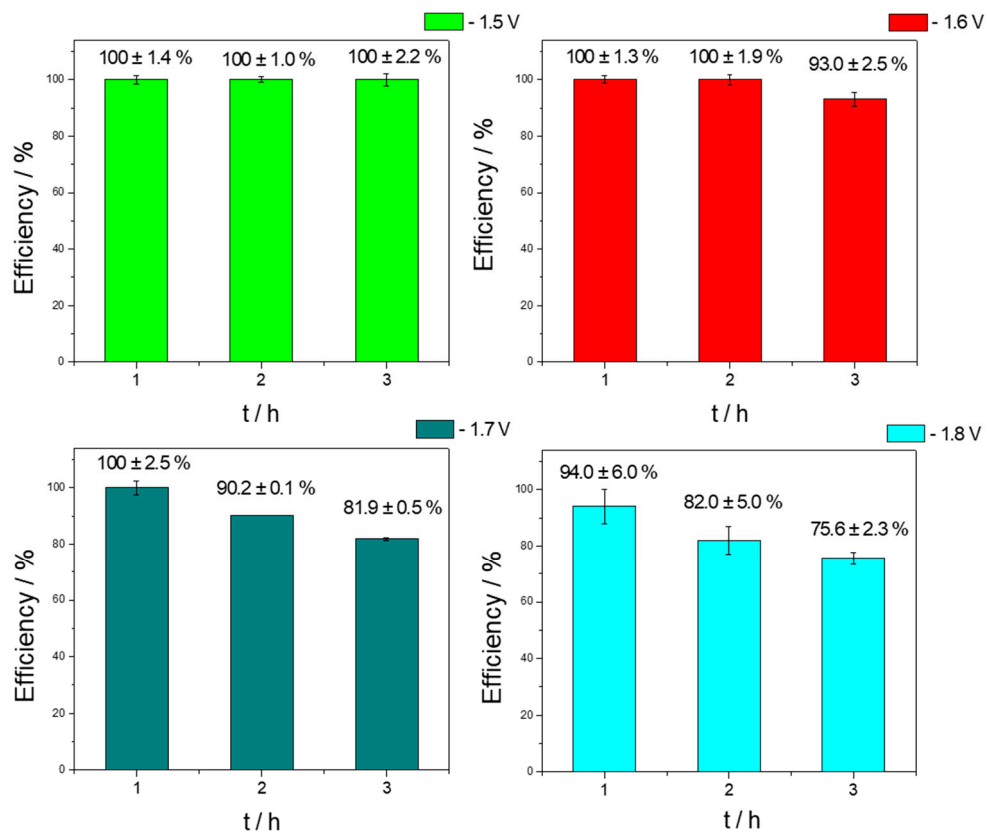


Figure S9. Faradaic efficiency for formate production at different controlled potential as a function of time.

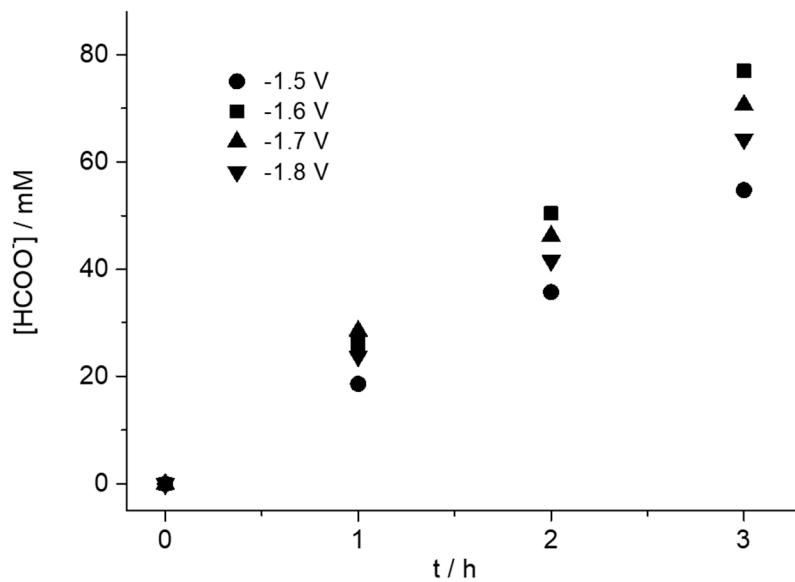


Figure S10. Formate concentration vs time at different potentials.

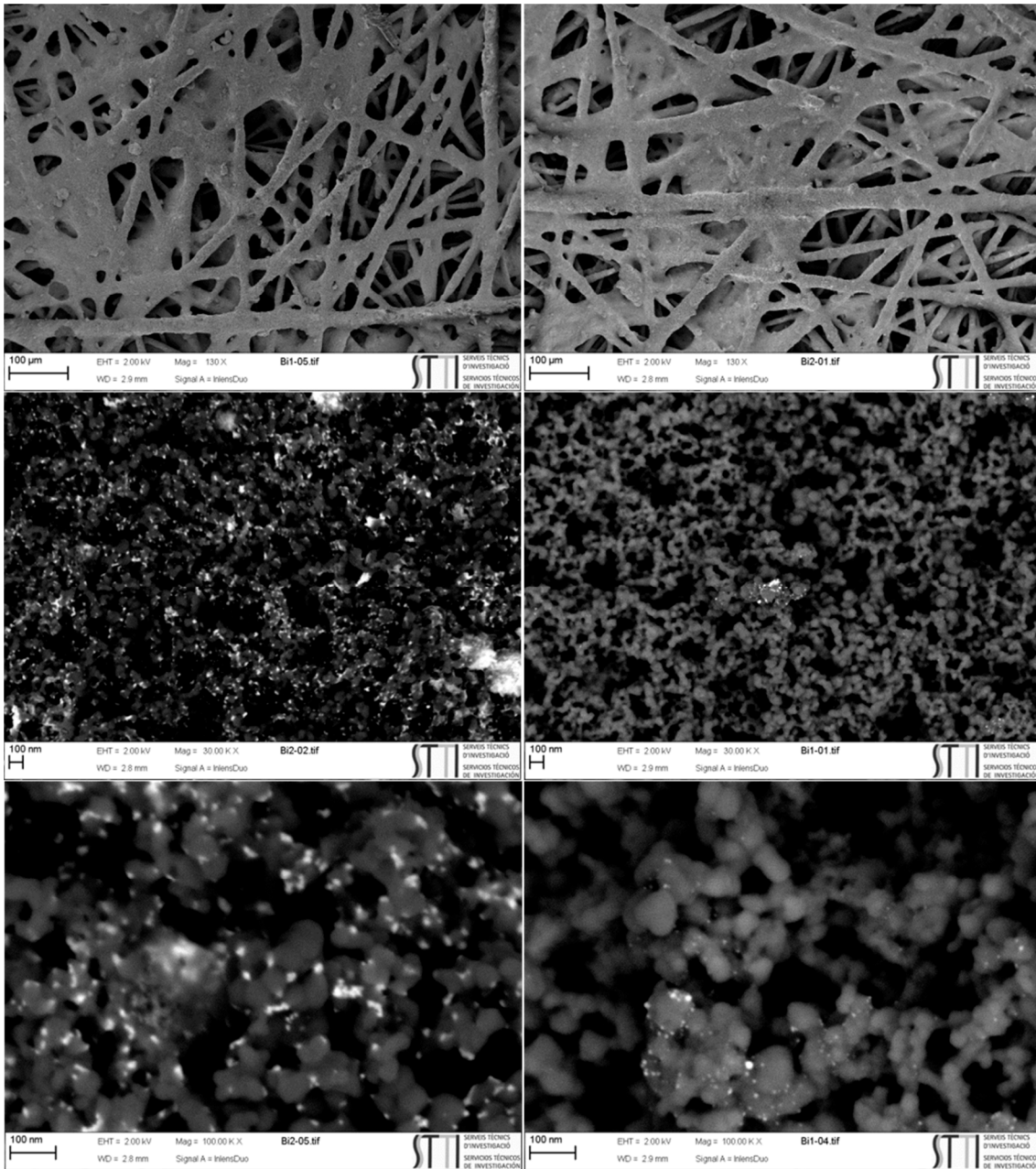


Figure S11. Back-scattered electrons field emission SEM images of the Bi-based electrodes with a Bi loading of 0.1 mg cm^{-2} , (left column) as-prepared, and (right column) after aprox. 70 hours in 3-hour CO₂ electrolyses at different controlled potentials.

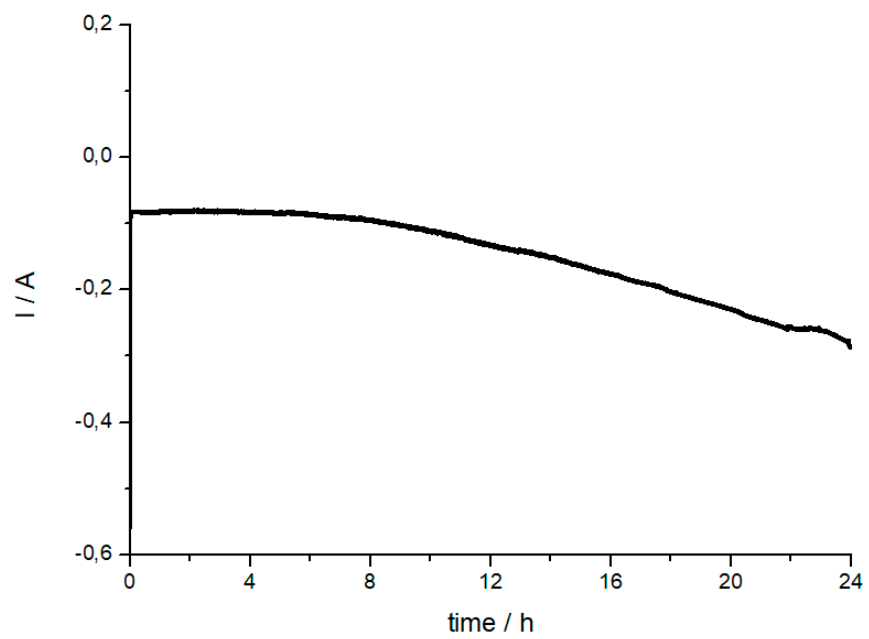


Figure S12. Chronoamperometric measurement at -1.6 V vs. *AgCl/Ag* during 24 h.