

# Supporting Information

for

## Effects of Temperature and Host Concentration on the Supramolecular Enantiodifferentiating [4+4] Photodimerization of <sup>2-</sup> Anthracenecarboxylate through Triplet-Triplet Annihilation Catalyzed by Pt-Modified Cyclodextrins

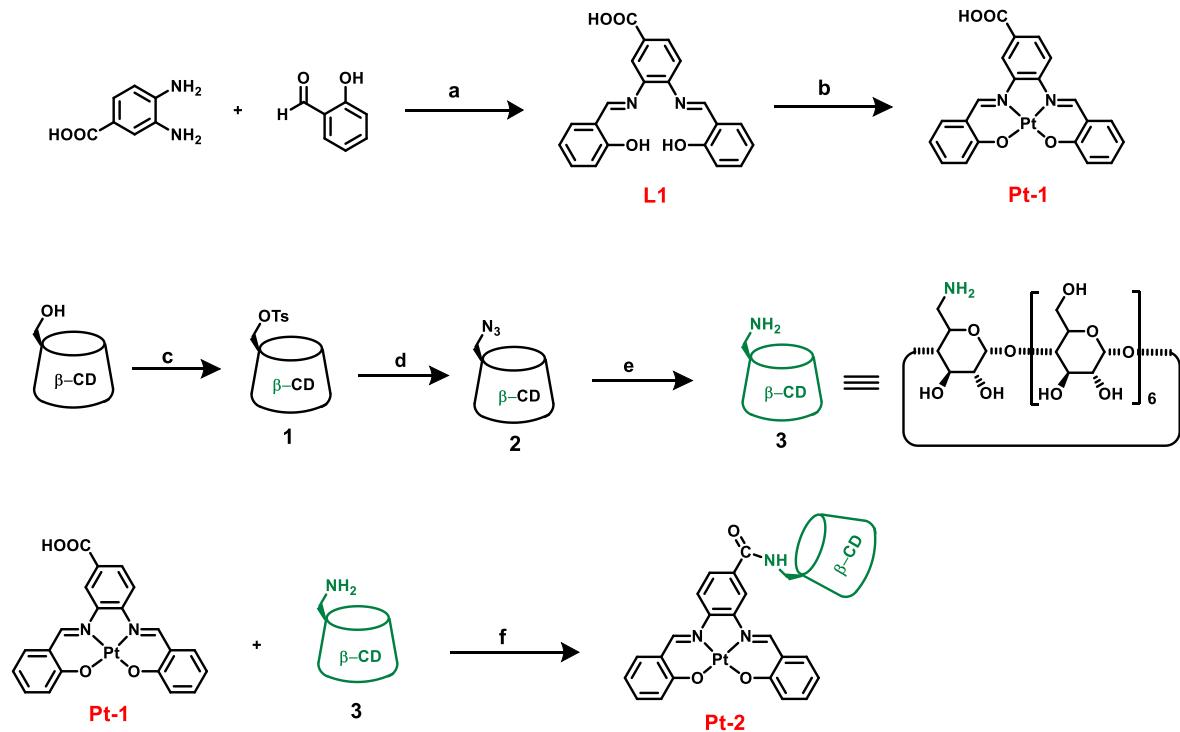
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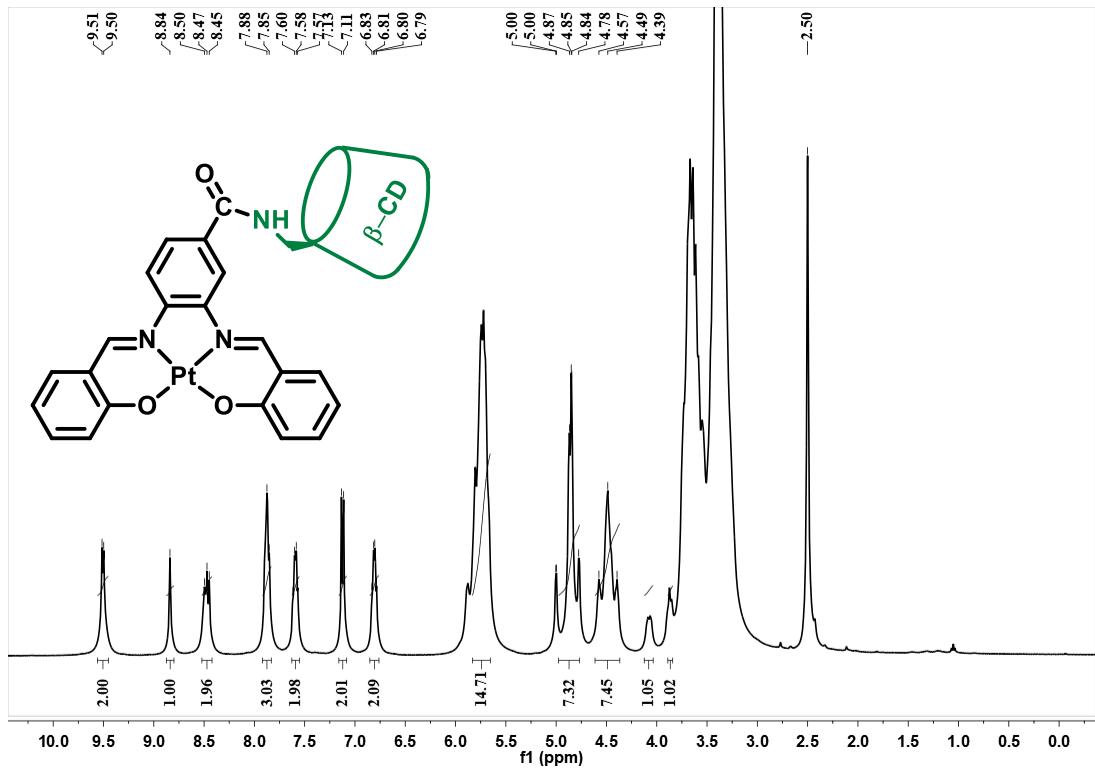
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## 1. Synthetic route

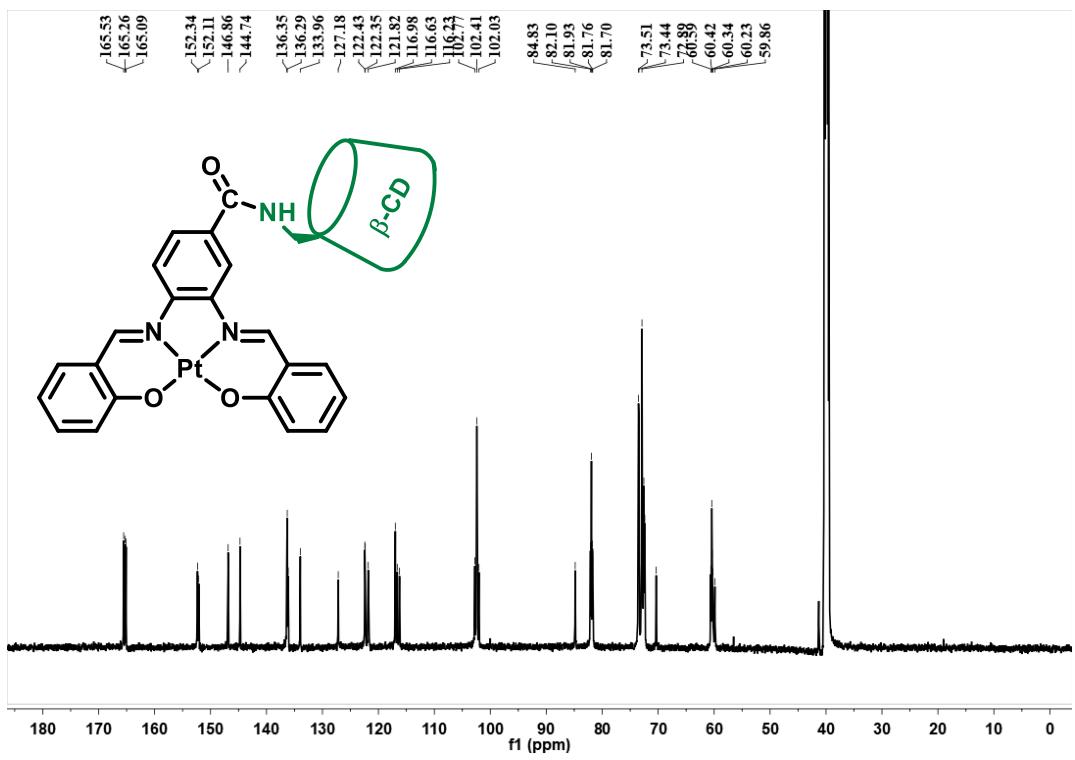


**Scheme S1.** Synthesis of **Pt-1**, **Pt-2**: a) EtOH, RT, 2 h; b)  $\text{K}_2\text{PtCl}_4$ ,  $\text{K}_2\text{CO}_3$ , DMSO,  $80^\circ\text{C}$ ; c) Pyridine, Paratoluensulfonyl Chloride, RT, 6 h; d)  $\text{NaN}_3$ , DMF/H<sub>2</sub>O (10/1, v/v),  $80^\circ\text{C}$ ; e) 1)  $\text{PPh}_3$ , DMF, RT, 6 h; 2)  $\text{NH}_3 \cdot \text{H}_2\text{O}$ , RT, 1 h; f) 1) EDC, HOBT,  $-20^\circ\text{C}$ , 2 h; 2) **3**, RT, 6 h;

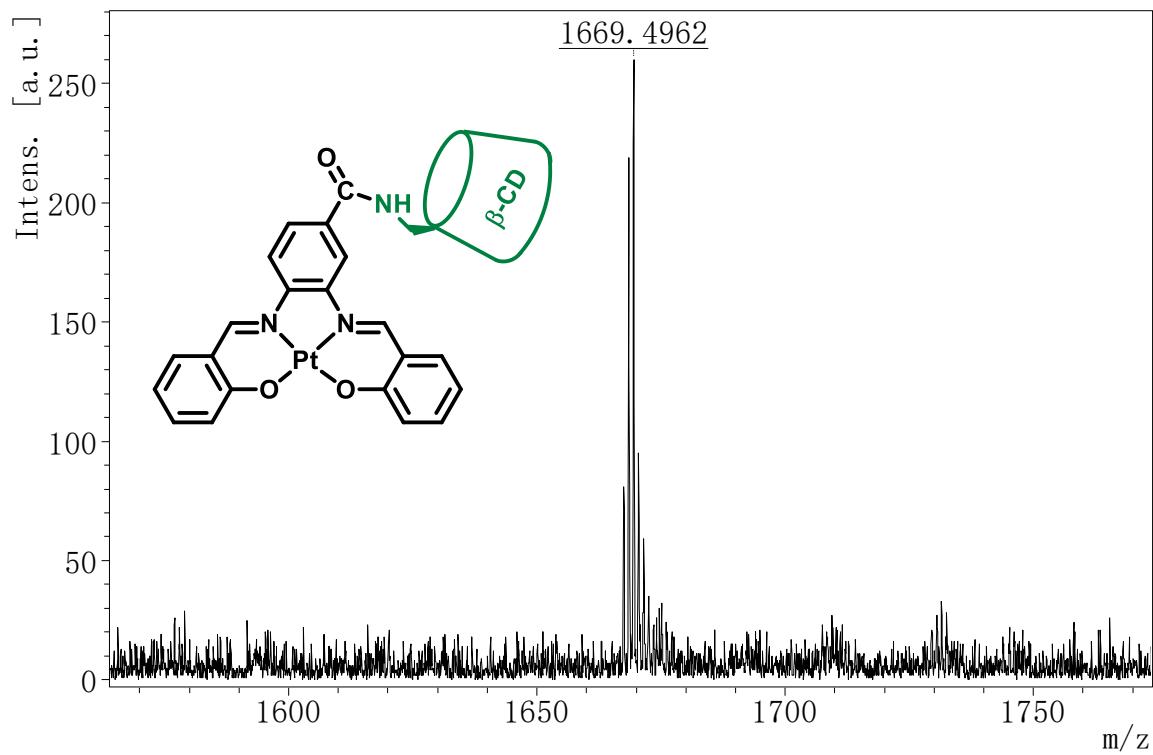
## 2. NMR and HRMS spectra of Pt-2 and AC



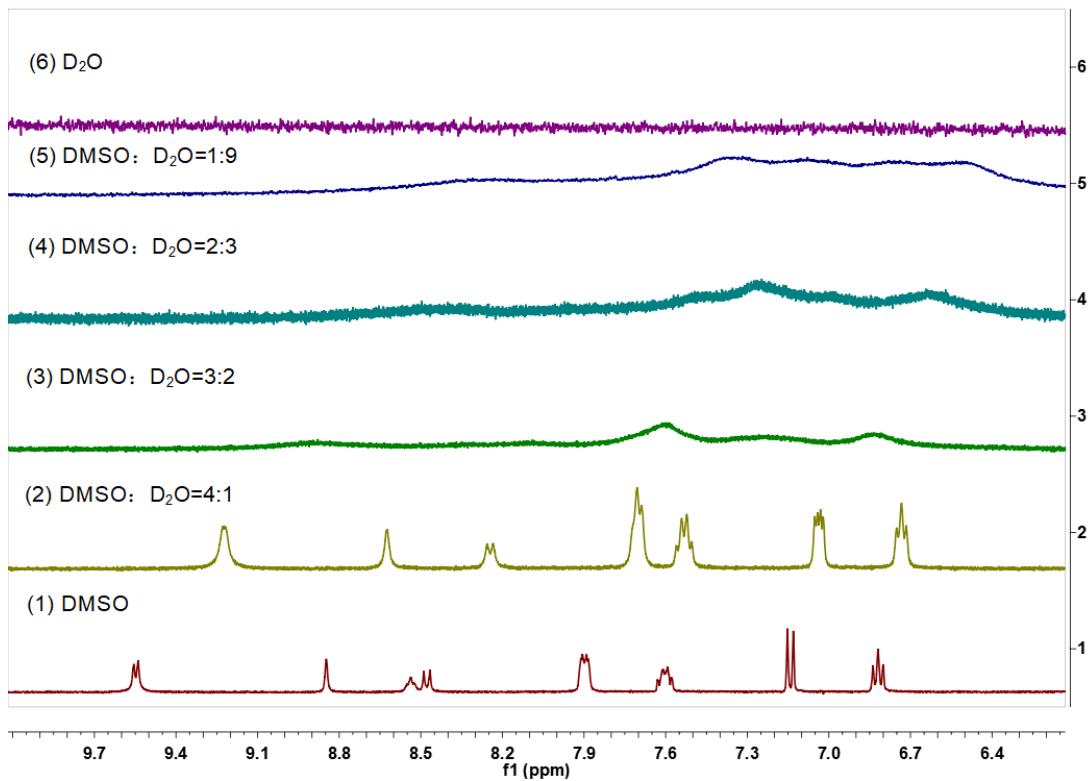
**Figure S1.** <sup>1</sup>H NMR spectrum (400 MHz, DMSO-*d*<sub>6</sub>, room temperature) of Pt-2.



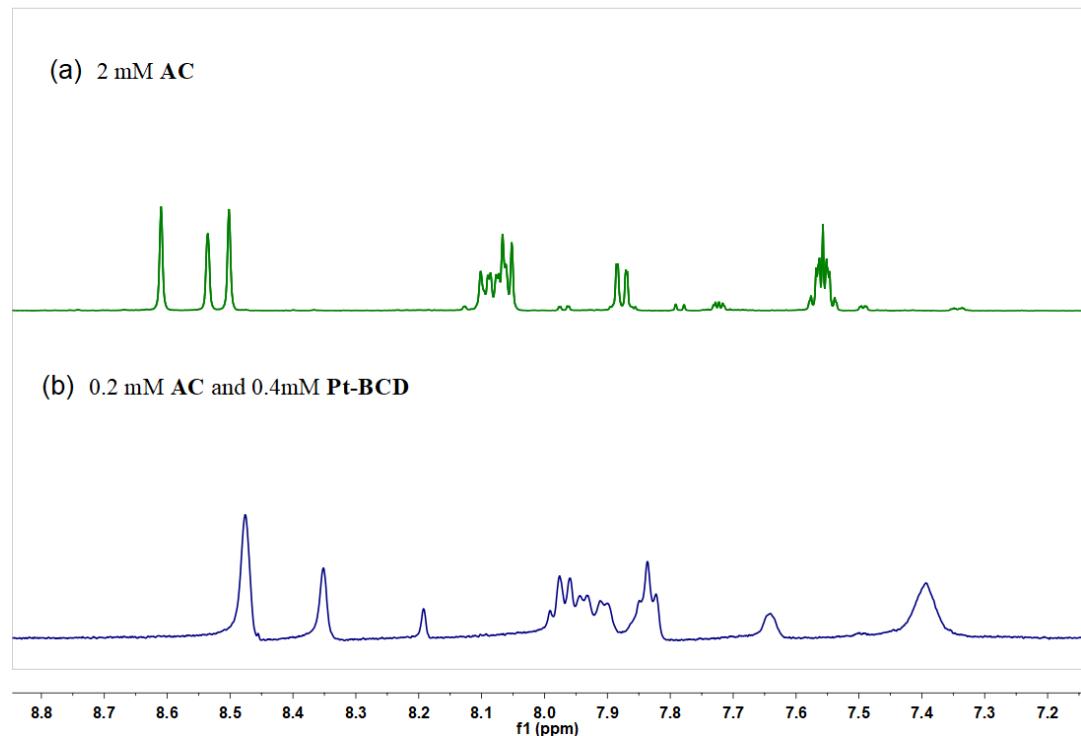
**Figure S2.**  $^{13}\text{C}$  NMR spectrum (151 MHz,  $\text{DMSO}-d_6$ , room temperature) of **Pt-2**.



**Figure S3.** MALDI-TOF-MS of **Pt-2**.

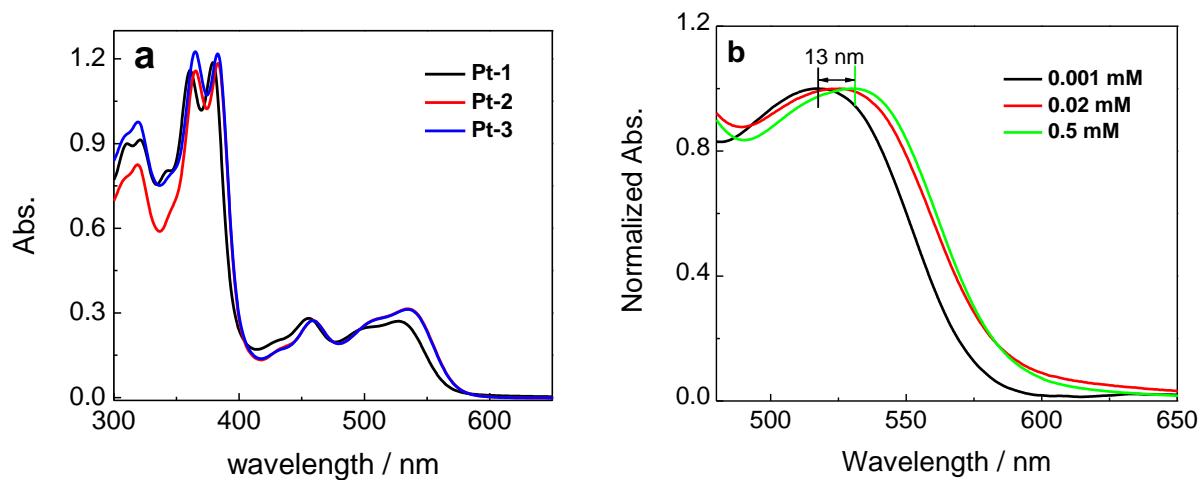


**Figure S4.**  $^1\text{H}$  NMR spectra (400 MHz, room temperature) of **Pt-2**. (1) DMSO- $d_6$ , (2) DMSO- $d_6$ : D<sub>2</sub>O=1:4 (v:v), (3) DMSO- $d_6$ : D<sub>2</sub>O=2:3 (v:v), (4) DMSO- $d_6$ : D<sub>2</sub>O=3:2 (v:v), (5) DMSO- $d_6$ : D<sub>2</sub>O=1:9 (v:v), (6) D<sub>2</sub>O.

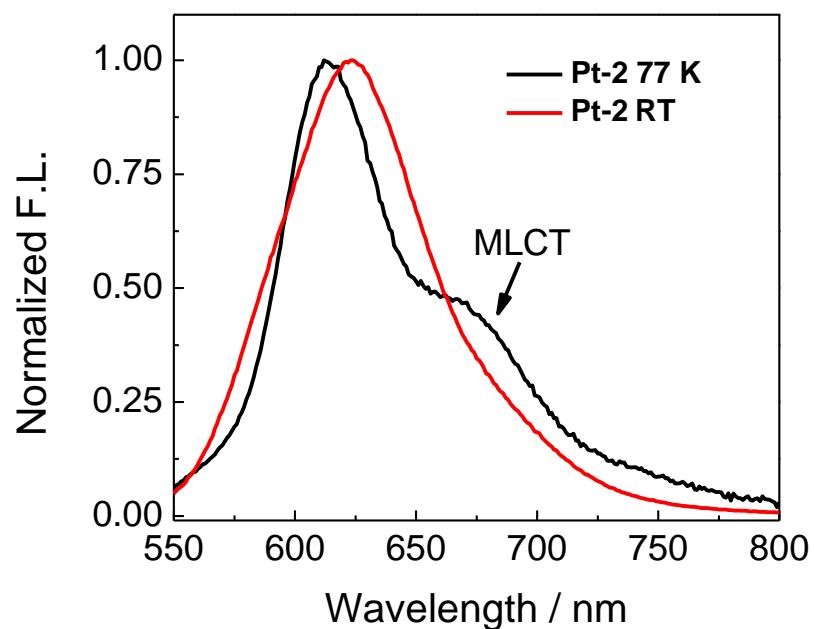


**Figure S5.**  $^1\text{H}$  NMR spectra (600 MHz, room temperature) of (a) 2 mM **AC**, (b) 0.2 mM **AC** and 0.4 mM **Pt-BCD**, measured in D<sub>2</sub>O.

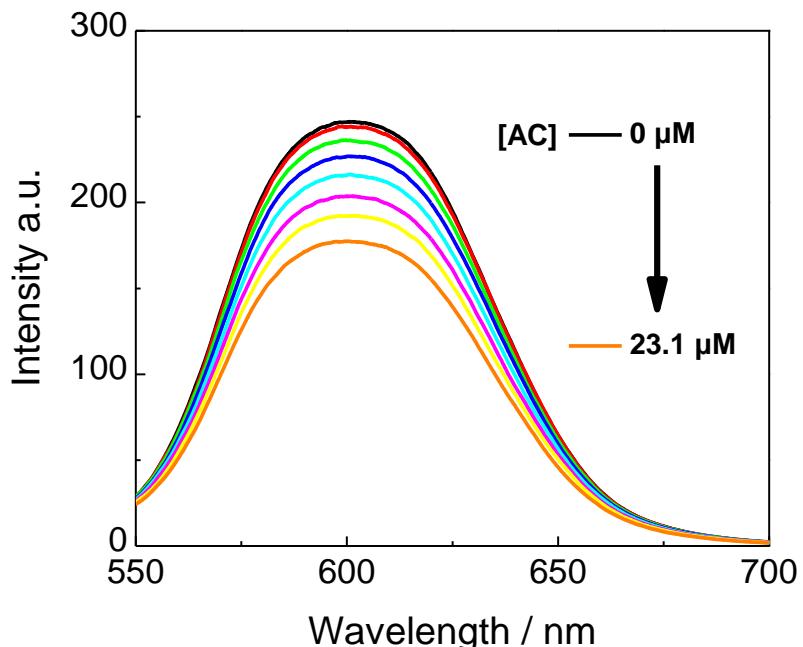
### 3. Absorption and emission spectra of Pt-1, Pt-2 and Pt-3



**Figure S6.** (a) UV-vis absorption spectra of **Pt-1**, **Pt-2**, and **Pt-3** in DMF solution,  $c = 5 \times 10^{-5}$  M, 20 °C. (b) Normalized UV-vis absorption spectra of **Pt-2** at different concentrations in aqueous solution, 20 °C.



**Figure S7.** Emission spectra of **Pt-2** measured in ethanol-methanol (4:1, v/v) glass at 77 K (black) and in aqueous solution at room temperature (red) (298 K),  $\lambda_{\text{ex}} = 532$  nm.



**Figure S8.** Phosphorescence spectral change of  $5 \times 10^{-5}$  M **Pt-2** upon increasing the concentration of **AC** ( $0 - 23.1 \times 10^{-6}$  M),  $\lambda_{\text{ex}} = 532$  nm,  $T = 20$  °C.

#### 4. Photophysical properties of the sensetizers/hosts **Pt-1**, **Pt-2**, and **Pt-3**

**Table S1.** Photophysical properties of the sensetizers/hosts **Pt-1**, **Pt-2**, and **Pt-3**, and Stern-Volmer quenching constant ( $K_{\text{SV}}$ ) between sensitizers/host and **AC** at 20 °C.

	Abs <sup>a</sup> /nm	$\mathcal{E}^b$	Em <sup>c</sup> /nm	$\Phi_F^d$	$\tau_p^e/\mu\text{s}$	$K_{\text{SV}}/10^4$ M	$K_q/10^9 \text{ M}^{-1}\text{s}^{-1}$
<b>Pt-1</b>	529	5396	599	0.13	2.0	1.28	6.46
<b>Pt-2</b>	533	6286	621	0.19	3.1	1.57	5.06
<b>Pt-3</b>	533	6267	611	0.20	4.1	2.29	5.65

<sup>a</sup> In DMF at  $5 \times 10^{-5}$  M. <sup>b</sup> Molar absorption coefficient at the absorption maxima. <sup>c</sup> Emission in water at RT. <sup>d</sup> Fluorescence quantum yield, rose bengal as standard ( $\Phi_F = 0.11$  in EtOH). <sup>e</sup> Phosphorescence lifetimes in deaerated water, 20°C.

## 5. Binding constants of Pt-3 and AC at different temperature

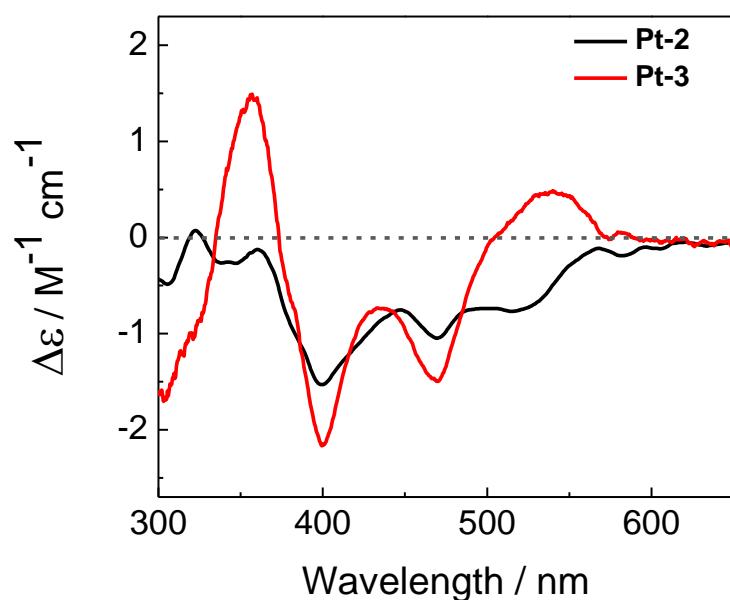
**Table S2.** Association Constants for the 1:1 and 1:2 Inclusion Complexes of **Pt-3** with AC Obtained by ITC at Various Temperature.

T/°C	Association constants <sup>a</sup> / M <sup>-1</sup>		$K_2 / K_1$	$K_1K_2 (10^6 \text{ M}^{-2})$
	$K_1^b$	$K_2^c$		
0	3810 (68)	5350 (88)	1.4	20.38 (0.006)
25	10400 (160)	719 (25)	0.07	7.48 (0.004)

<sup>a</sup>The values in parentheses the errors. <sup>b</sup> Association constant for **Pt-3** and **AC** making 1:1 complex. <sup>c</sup>

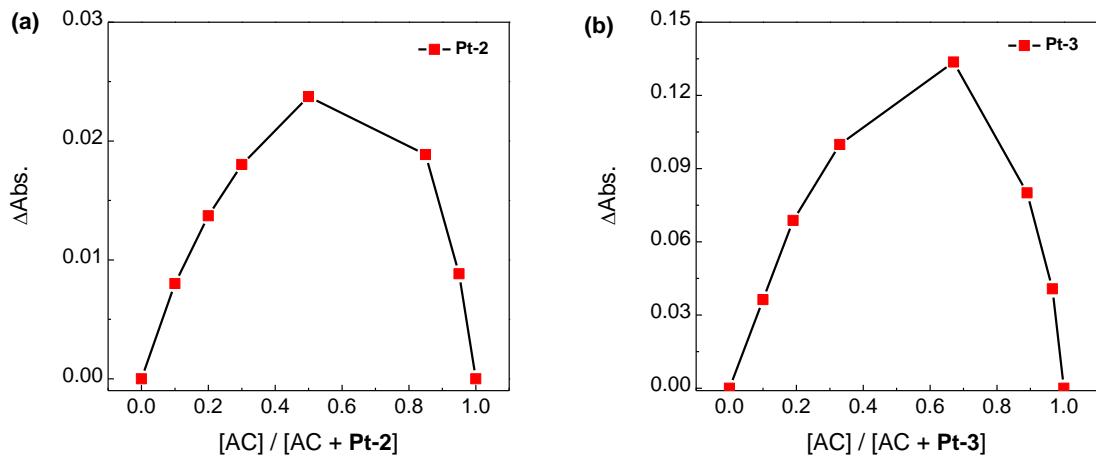
Association constant for 1:1 complex and **AC** making 1:2 complex

## 6. CD spectra of Pt-2 and Pt-3



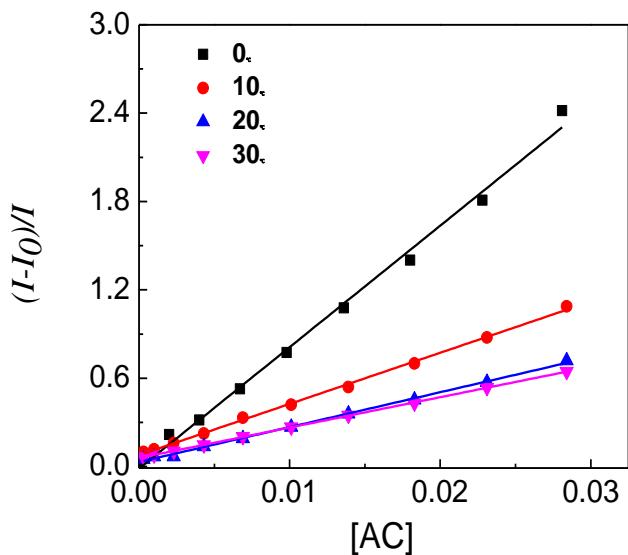
**Figure S9.** CD Spectra of **Pt-2**, **Pt-3** in aqueous solution, T=20 °C.

## 7. Job Plots of Pt-2 and Pt-3



**Figure S10.** Job plots of the absorption changes at 532 nm of (a) **Pt-2**, (b) **Pt-3** measured by holding the total concentration of  $[\text{AC}]+[\text{sensitizer}] = 0.2 \text{ mM}$

## 8. Stern-Volmer plots of Pt-3



**Figure S11.** Stern-Volmer plots of **Pt-3** of  $(I - I_0)/I$  versus  $[\text{AC}]$  in deareated water,  $C_{[\text{Pt-3}]} = 5 \times 10^{-5} \text{ M}$ ,  $C_{[\text{AC}]} = 0 - 2.8 \times 10^{-5} \text{ M}$ ,  $T = 0, 10, 20, 30 \text{ }^\circ\text{C}$ .