

## Supporting Information

### Solvent-free iron(III) chloride catalyzed direct amidation of esters

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## 1. Materials and Methods

All the solvents used were freshly distilled and dried by appropriate techniques. All reagents were purchased from Sigma Aldrich FeCl<sub>3</sub> reagent grade 97% was used as the catalyst. All reactions were monitored by thin layer chromatography (TLC) on aluminum-backed Merck silica gel 60 F254 plates using an ascending technique. The plates were visualized under UV light at 254 nm. Gravity column chromatography was done on Merck silica gel 60 (70–230 mesh). All proton nuclear magnetic resonance (<sup>1</sup>H NMR) spectra were recorded using deuterated chloroform solutions on a Bruker Ultrashield (400 or 500 MHz) spectrometer. Carbon-13 nuclear magnetic resonance (<sup>13</sup>C NMR) spectra were recorded on the same instruments at 100 MHz. All chemical shifts are reported in ppm. The chemical structures of the synthesized compounds were confirmed by comparison of their NMR data to literature reported data.

## 2. Optimization of reaction conditions

### 2.1 Catalyst Screening

Catalyst screening, reaction temperature as well as catalyst loading were optimized using ester **1a** and amine **2a** as model substrates. Reactions were carried out as reported in the manuscript using 10 mol% of each catalyst under solvent-free conditions. The reactions were left to stir for 24 hours at 50 °C. The formed products were isolated by column chromatographic techniques and the yields reported as shown in table S1. No product was observed upon TLC analysis of the control experiment.

Table S1

Entry	Catalyst	Temperature (°C)	Time (h)	Yield
1	AlCl <sub>3</sub>	50	24	91%
2	FeBr <sub>3</sub>	50	24	84%
3	FeCl <sub>3</sub>	50	24	98%
4	BiCl <sub>3</sub>	50	24	96%
5	FeCl <sub>2</sub>	50	24	81%
6	None	50	24	0%

### 2.2 Reaction Temperature

Amidation of ester **1a** with amine **2a** was conducted at 50 °C, 80 °C and 100 °C as shown in Table S2. The reaction time to reach completion was monitored by TLC. Lowering the reaction temperature resulted in prolonged reaction times and there was no difference between reactions at 80 and 100 °C.

**Table S2**

Entry	Temperature (°C)	Catalyst Loading (mol%)	Reaction Time (h)
1	100	10	6
2	80	10	6
3	50	10	24

### 2.3 Catalyst Loading

Amidation of ester **1a** with amine **2a** was conducted according to the procedure reported in the manuscript using 5 mol%, 10 mol% and 15 mol% FeCl<sub>3</sub>. The reaction progress was monitored by TLC plate and the reaction times are shown in Tables S3. The optimum catalyst loading was 15 mol% as the reaction was complete within 1.5 hours.

**Table S3**

Entry	Catalyst Loading (mol%)	Temperature (°C)	Reaction Time (h)
1	5	80	24
2	10	80	6
3	15	80	1.5

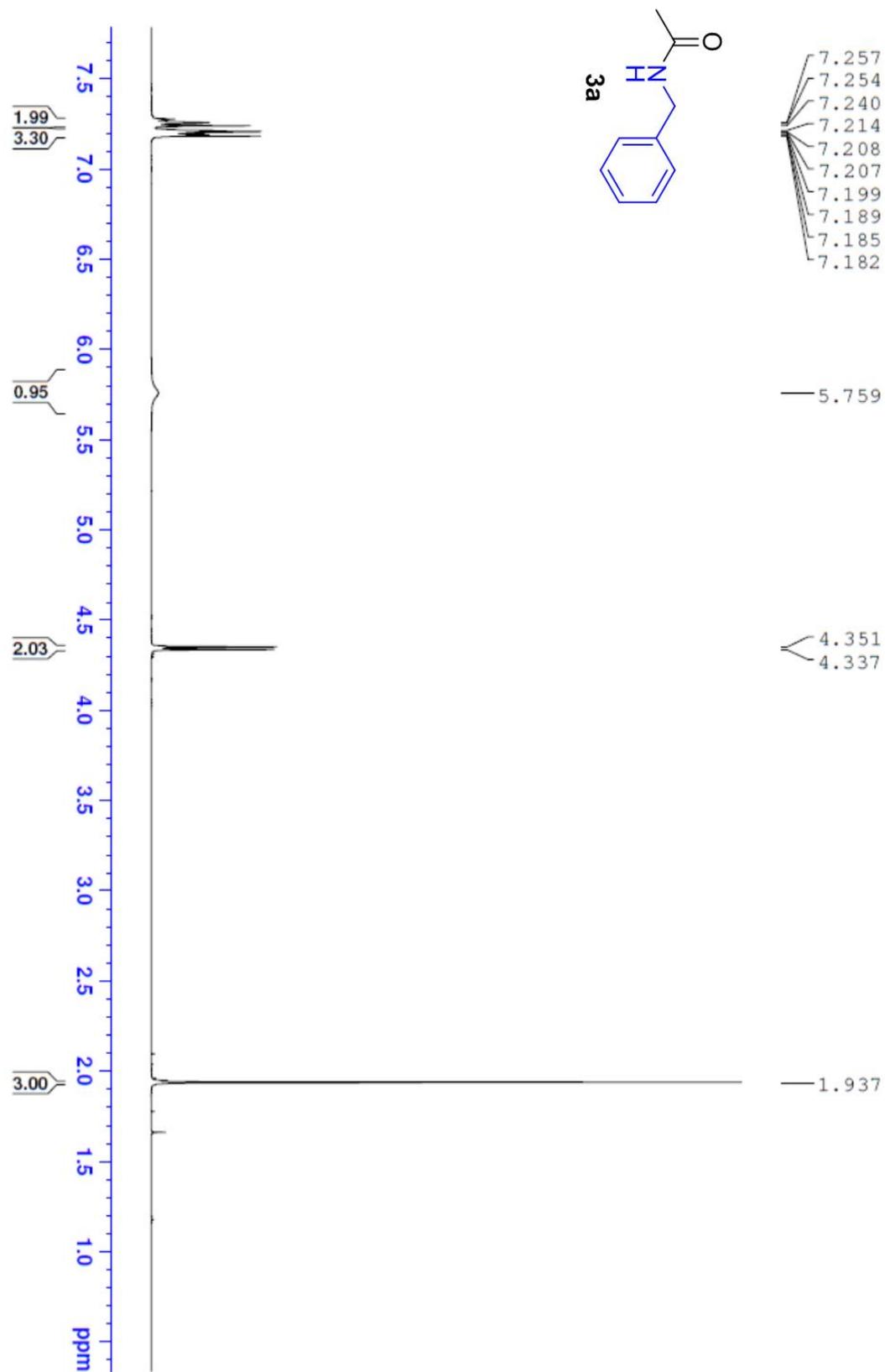
### 2.4 Solvent Screening

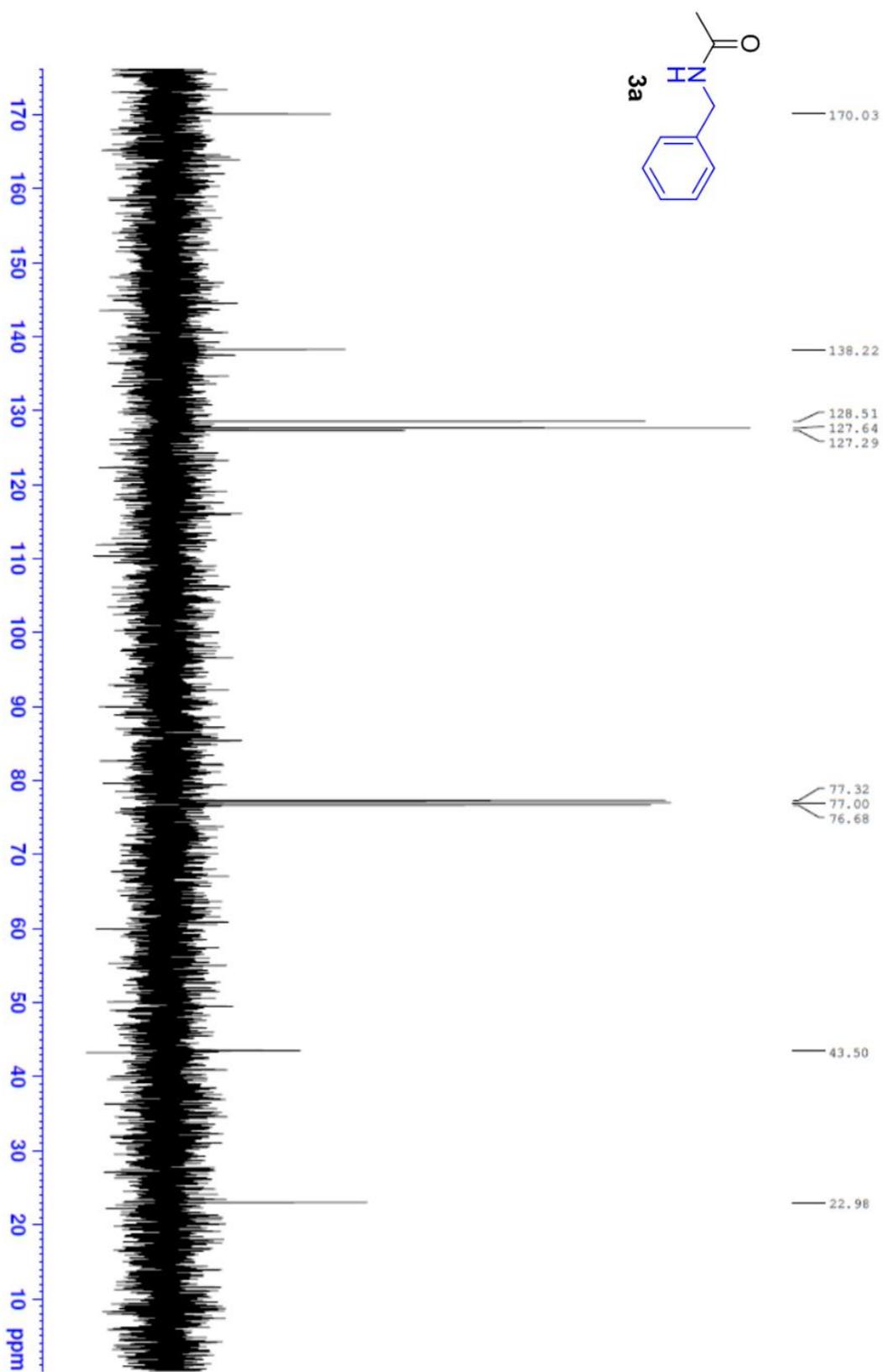
An oven-dried pressure tube equipped with a magnetic stirrer was evacuated with nitrogen. To this was added of ester **1a** (492  $\mu$ l, 5.04 mmol) followed by amine **2a** (0.5 ml, 4.58 mmol) and a 1 ml of solvent (Toluene, acetonitrile, tetrahydrofuran or 1,2-dichloroethane. Finally, FeCl<sub>3</sub> (111 mg, 0.684 mmol) was added and the reaction mixture was then sealed and stirred at 80 °C. The reaction was left to stir for 90 min and was diluted with EtOAc and washed once with saturated NaHCO<sub>3</sub> and once with distilled H<sub>2</sub>O. The combined aqueous layers were extracted once with ethyl acetate. The combined organic layers were then dried over MgSO<sub>4</sub>, filtered and solvents removed under reduced pressure. The crude product was purified by silica gel flash column chromatography using a combination of hexane and ethyl acetate (3:2). The isolated product yields are reported in Table 1 of the manuscript.

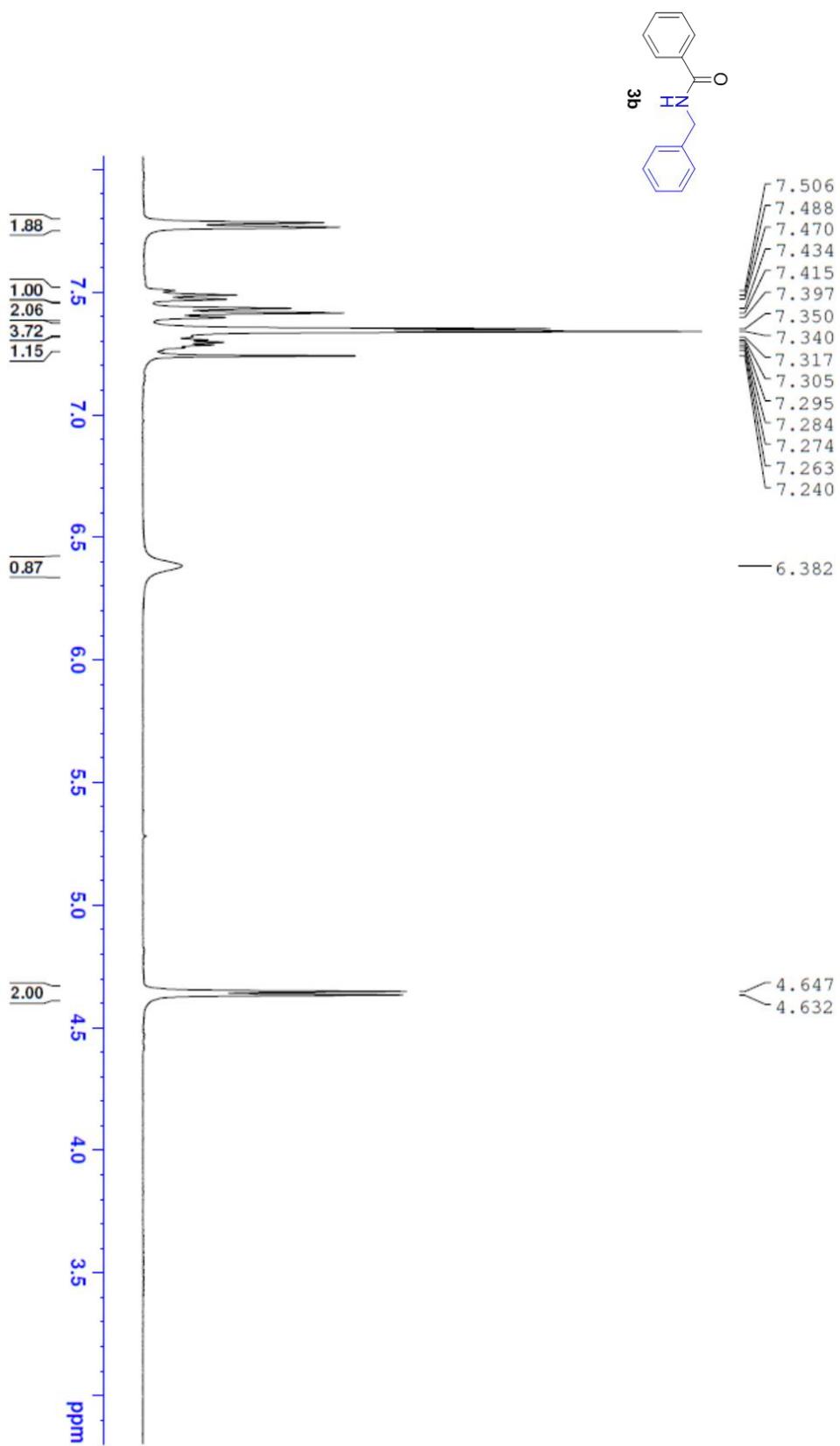
### 3. Typical experimental procedure for FeCl<sub>3</sub> catalyzed direct amidation of esters

An oven-dried pressure tube equipped with a magnetic stirrer was evacuated with nitrogen. To this was added of ester **1a** (492  $\mu$ l, 5.04 mmol) followed by amine **2a** (0.5 ml, 4.58 mmol) and finally FeCl<sub>3</sub> (111 mg, 0.684 mmol). The mixture was then sealed and stirred at 80 °C (0.5 ml of CH<sub>3</sub>CN was added if the reaction mixture solidified). The reaction was monitored by TLC until completion upon which it was diluted with EtOAc and washed once with saturated NaHCO<sub>3</sub> and once with distilled H<sub>2</sub>O. The combined aqueous layers were extracted once with ethyl acetate. The combined organic layers were then dried over MgSO<sub>4</sub>, filtered and solvents removed under reduced pressure. The crude product was purified by silica gel flash column chromatography using a combination of hexane and ethyl acetate (3:2).

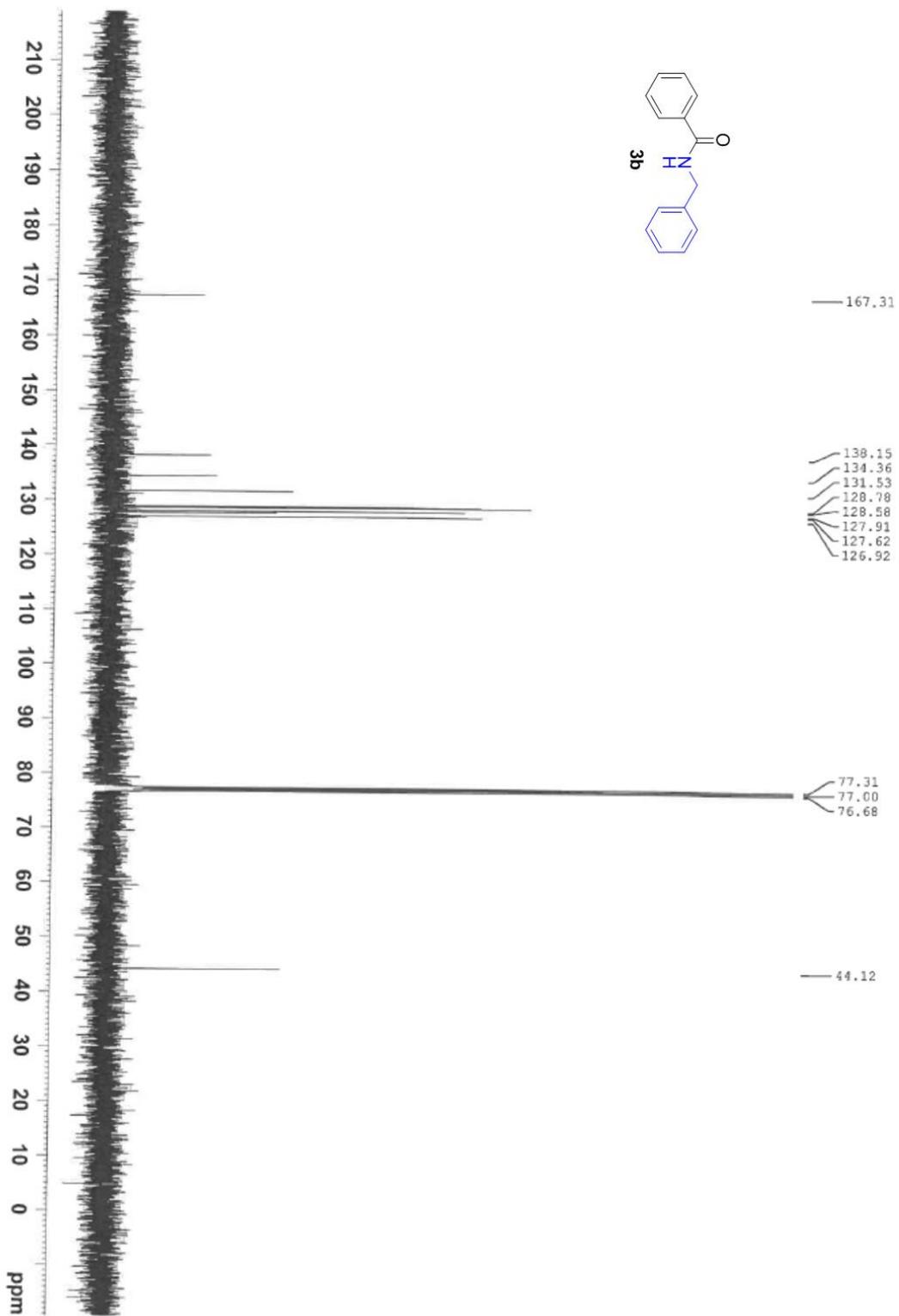
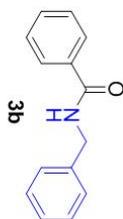
#### 4. NMR Spectra of Products

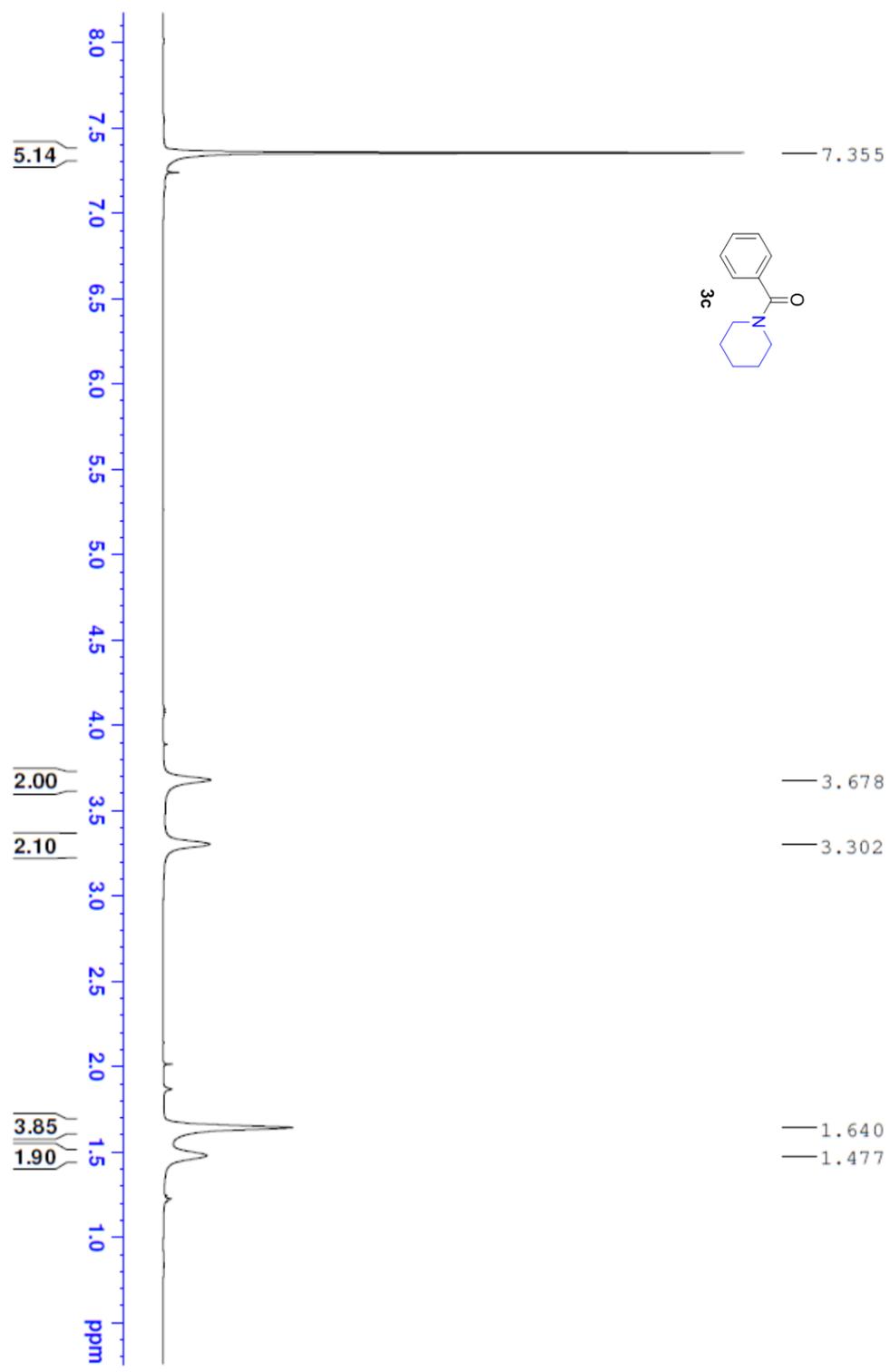




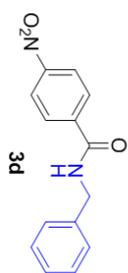


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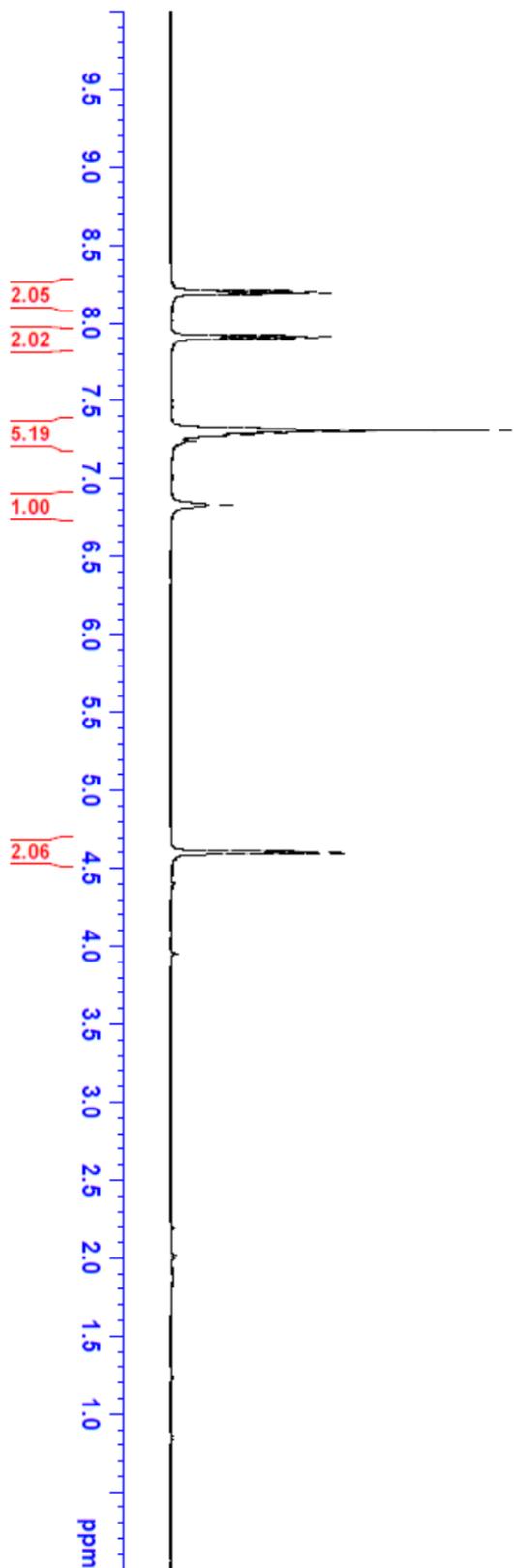


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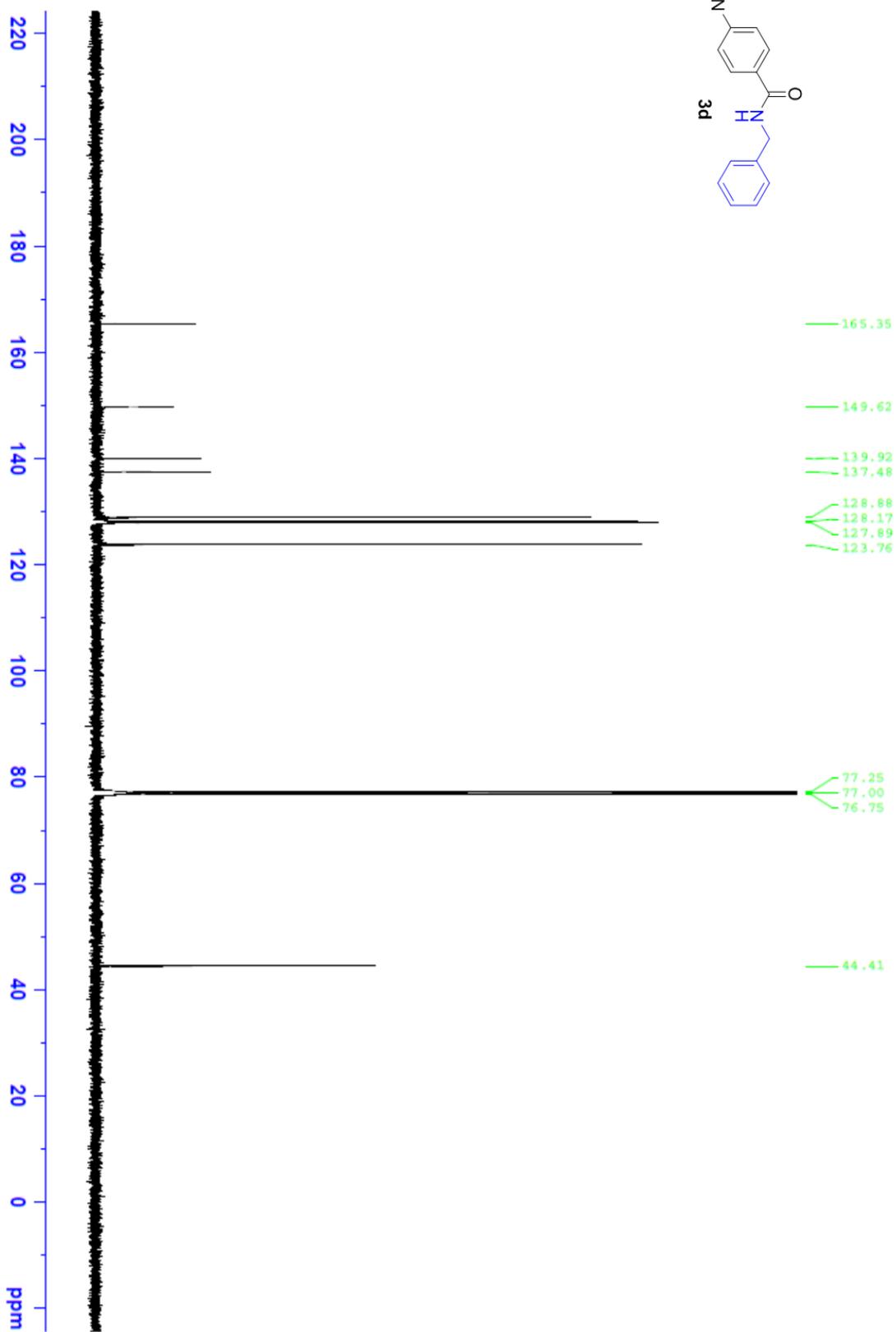
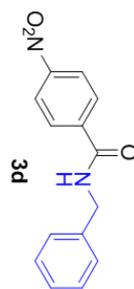


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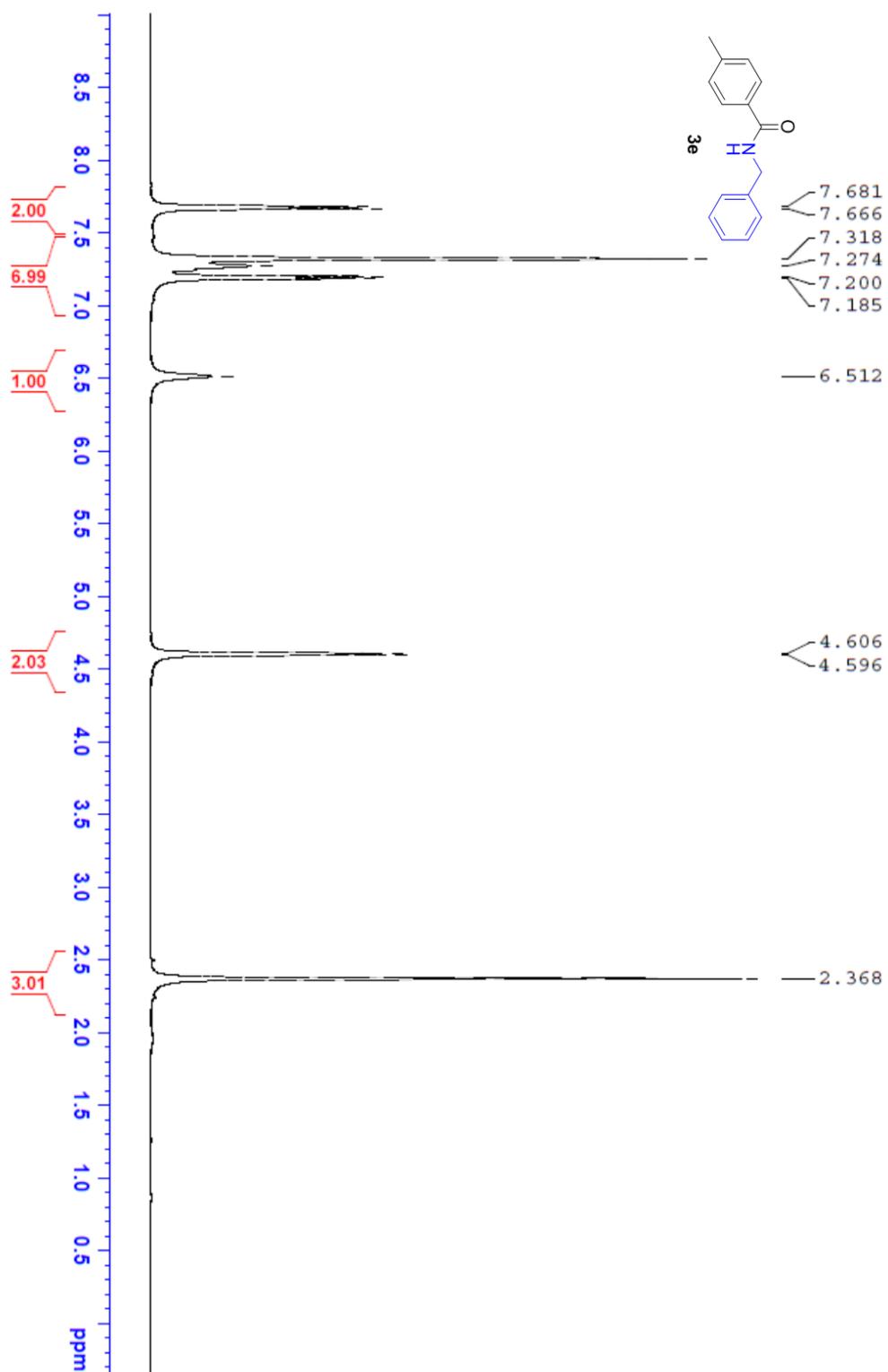
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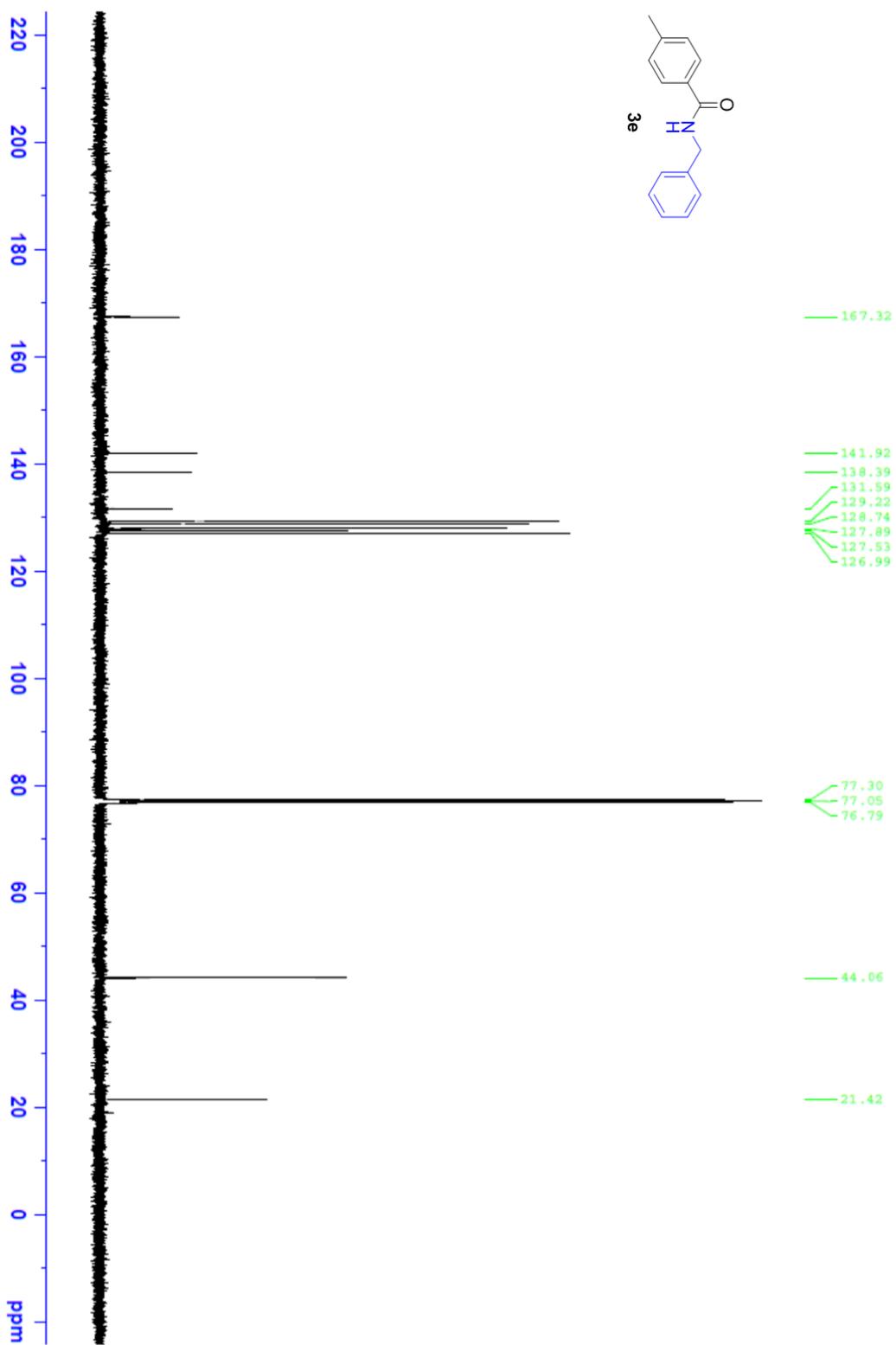
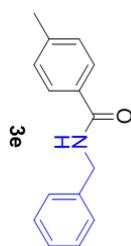
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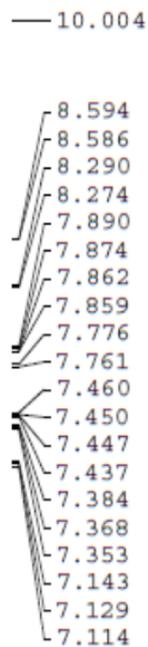
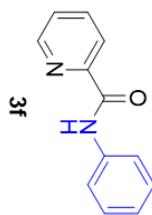
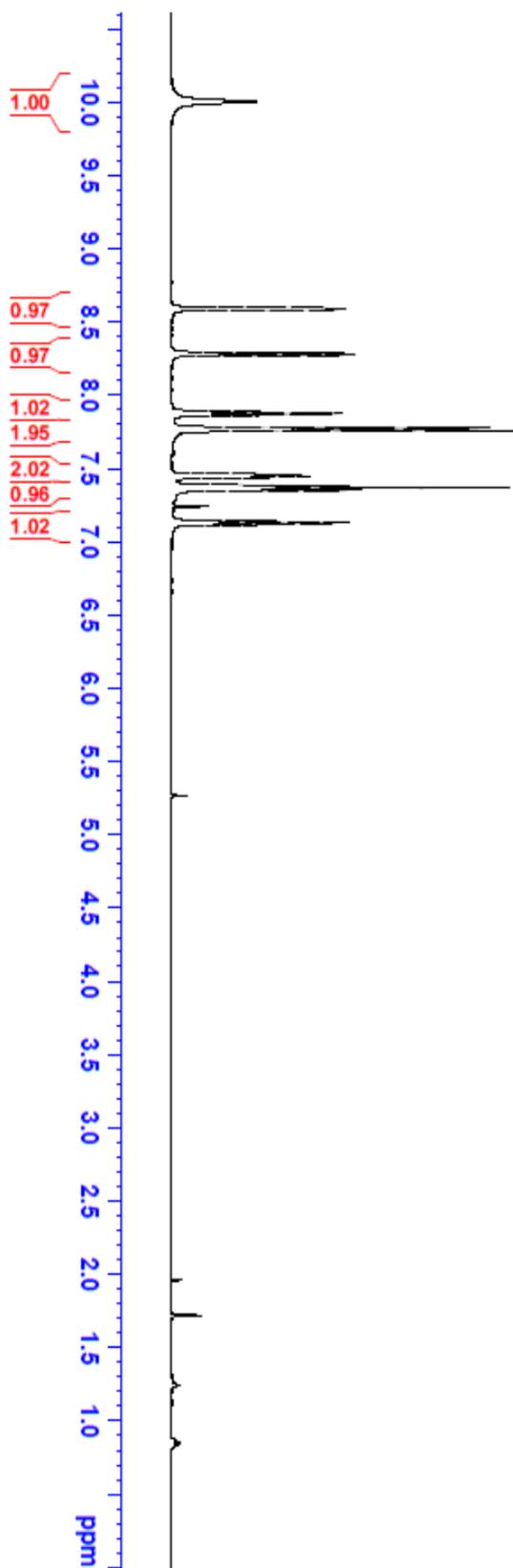


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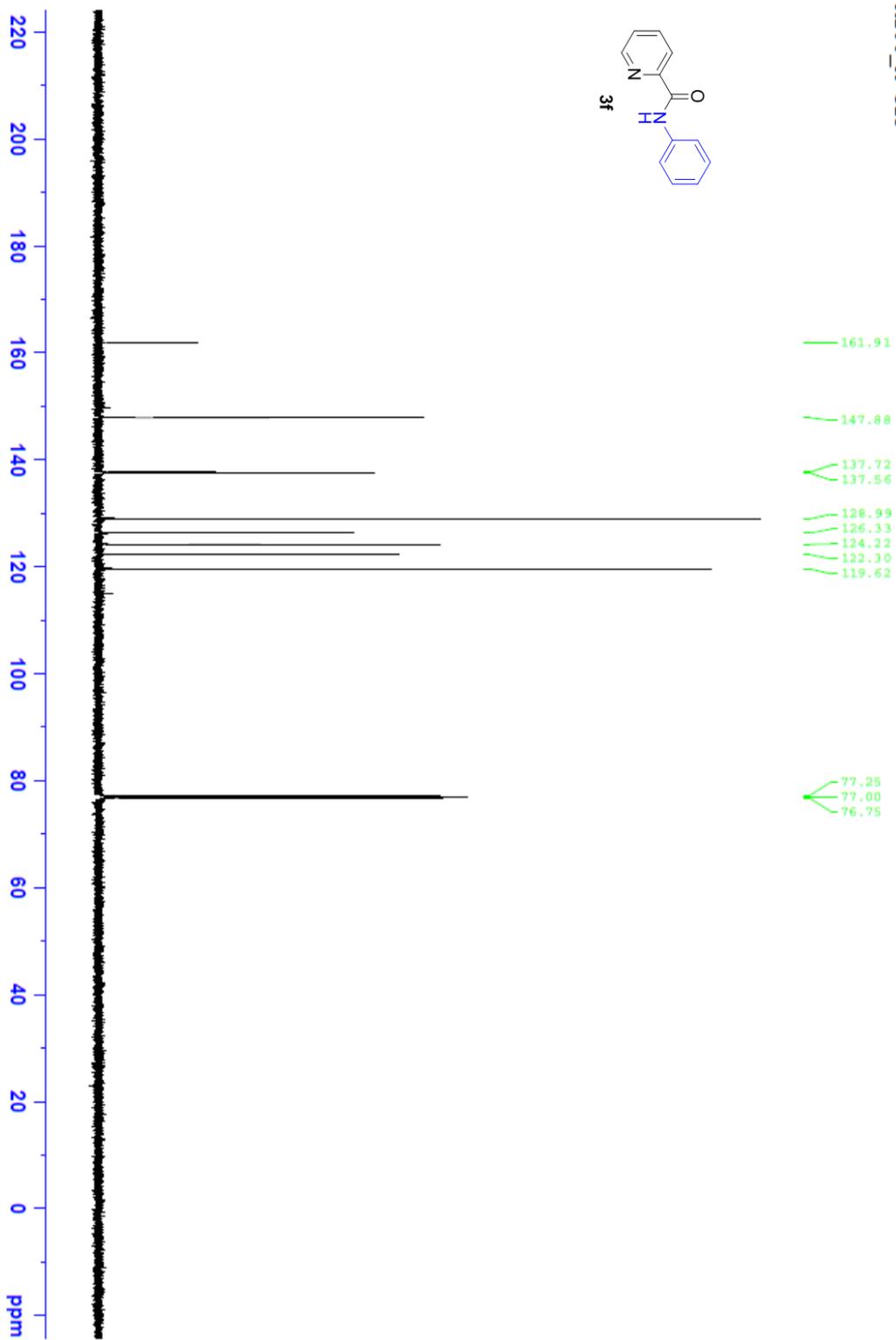
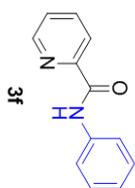


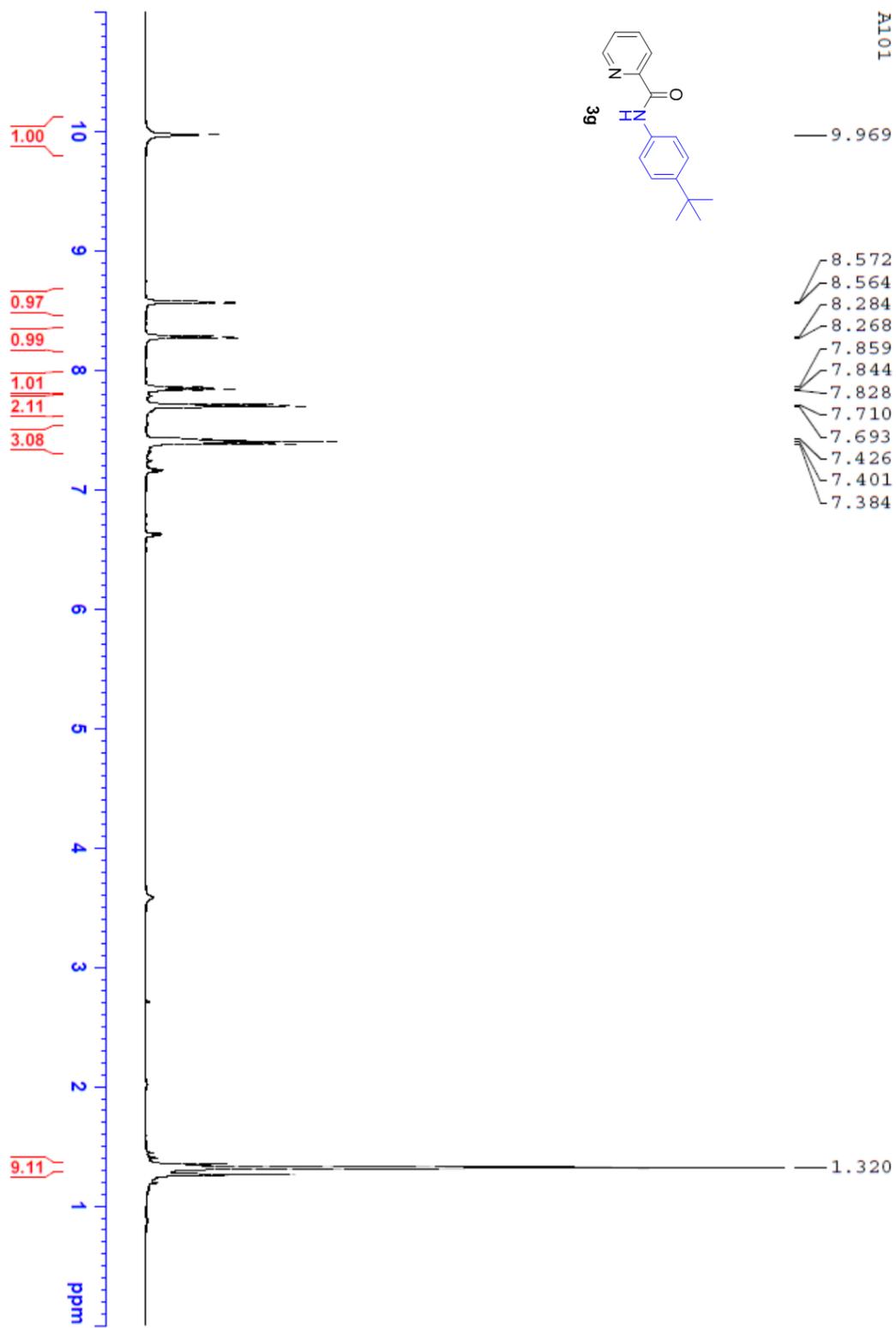
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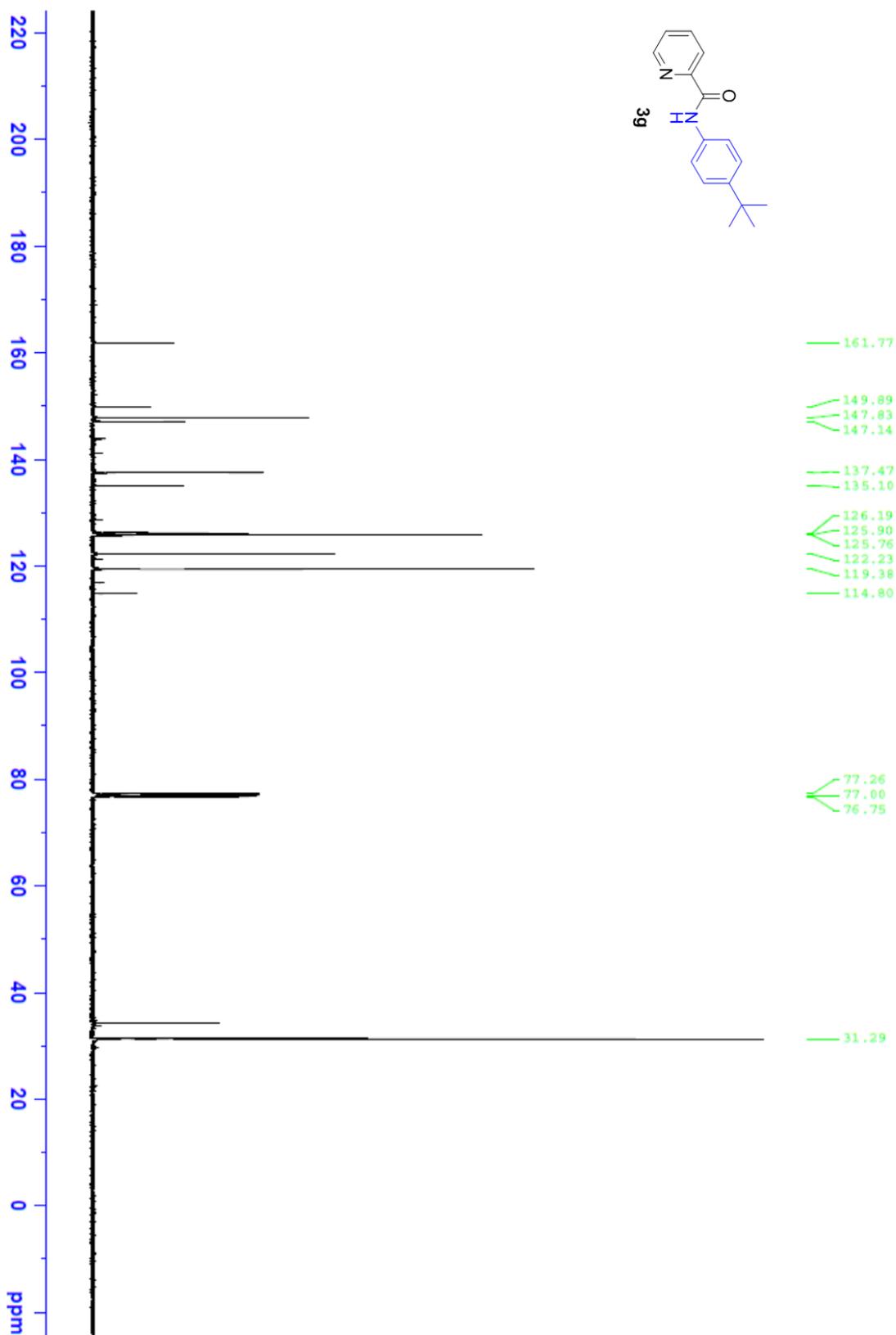
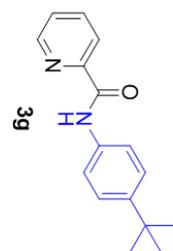


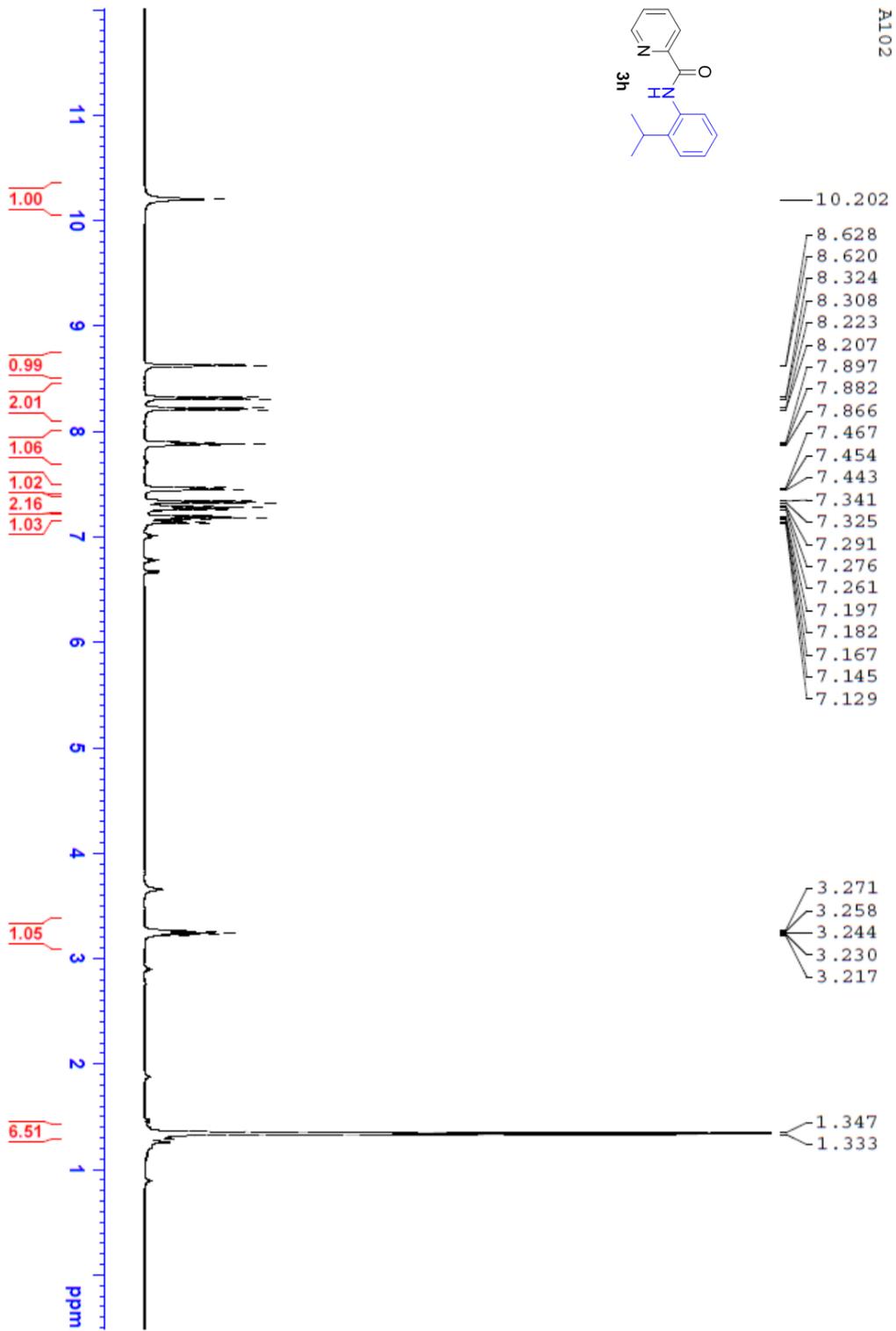
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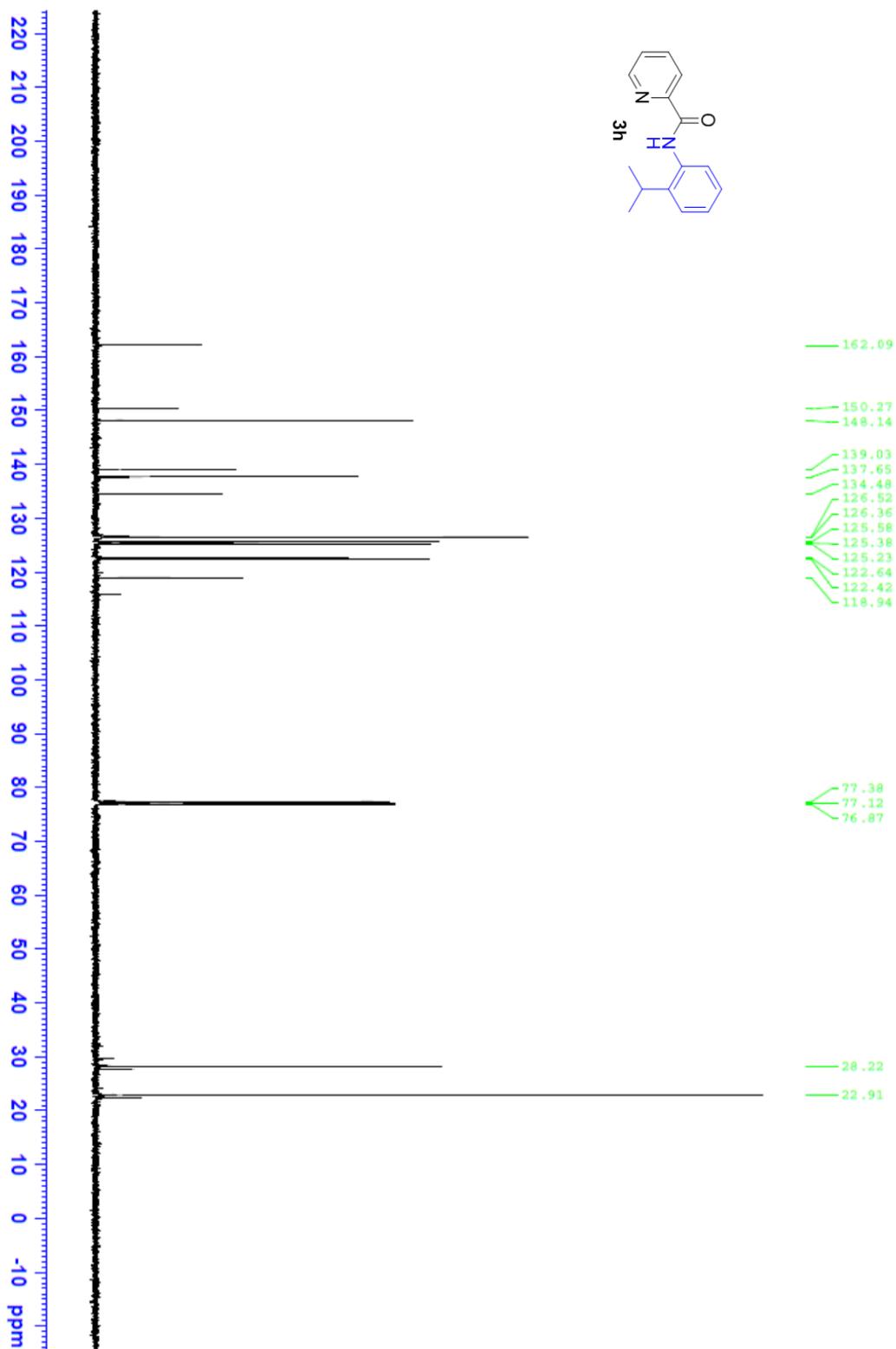
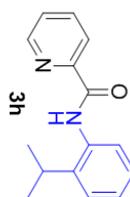


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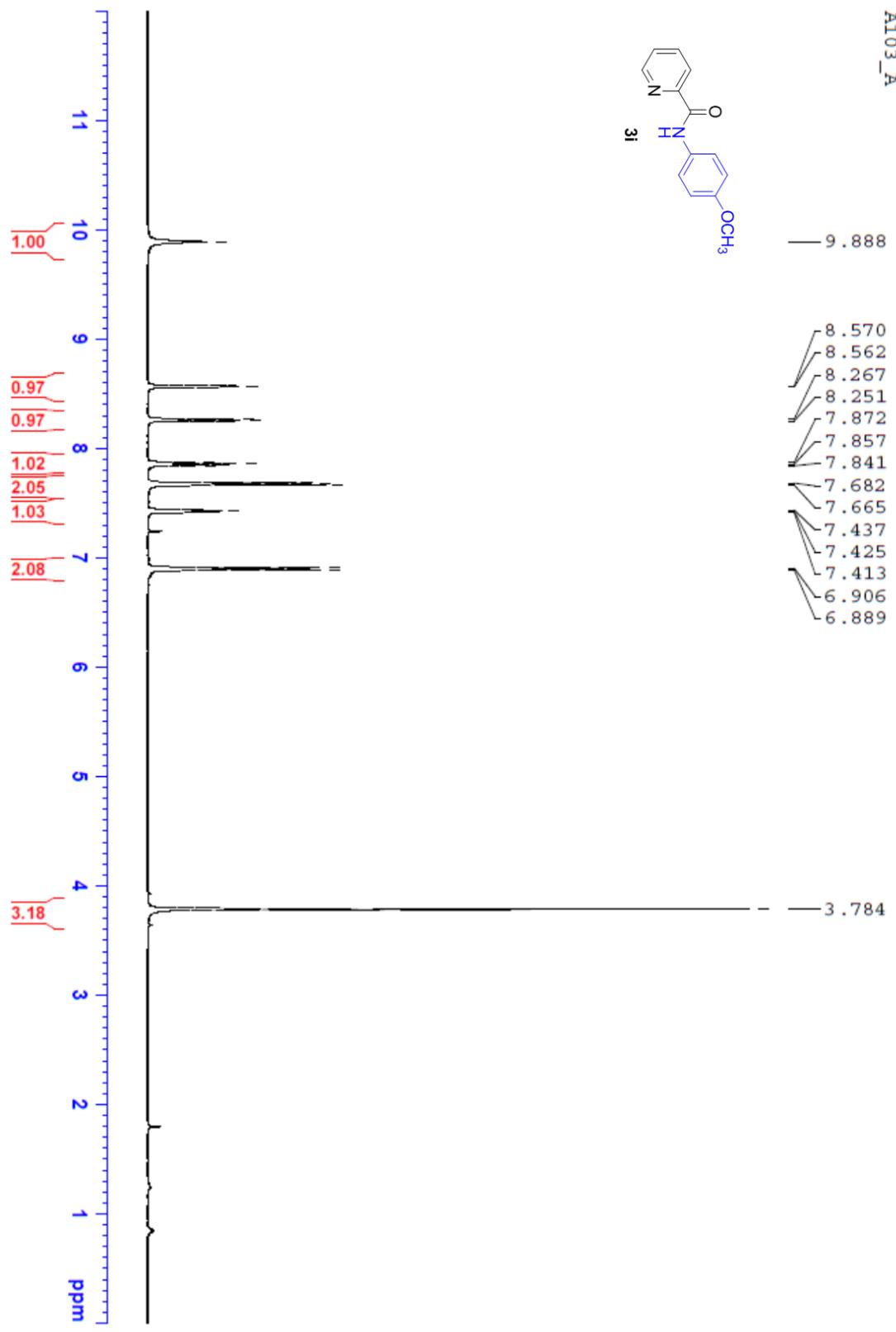
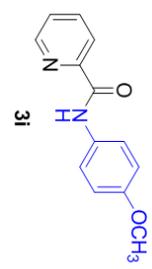




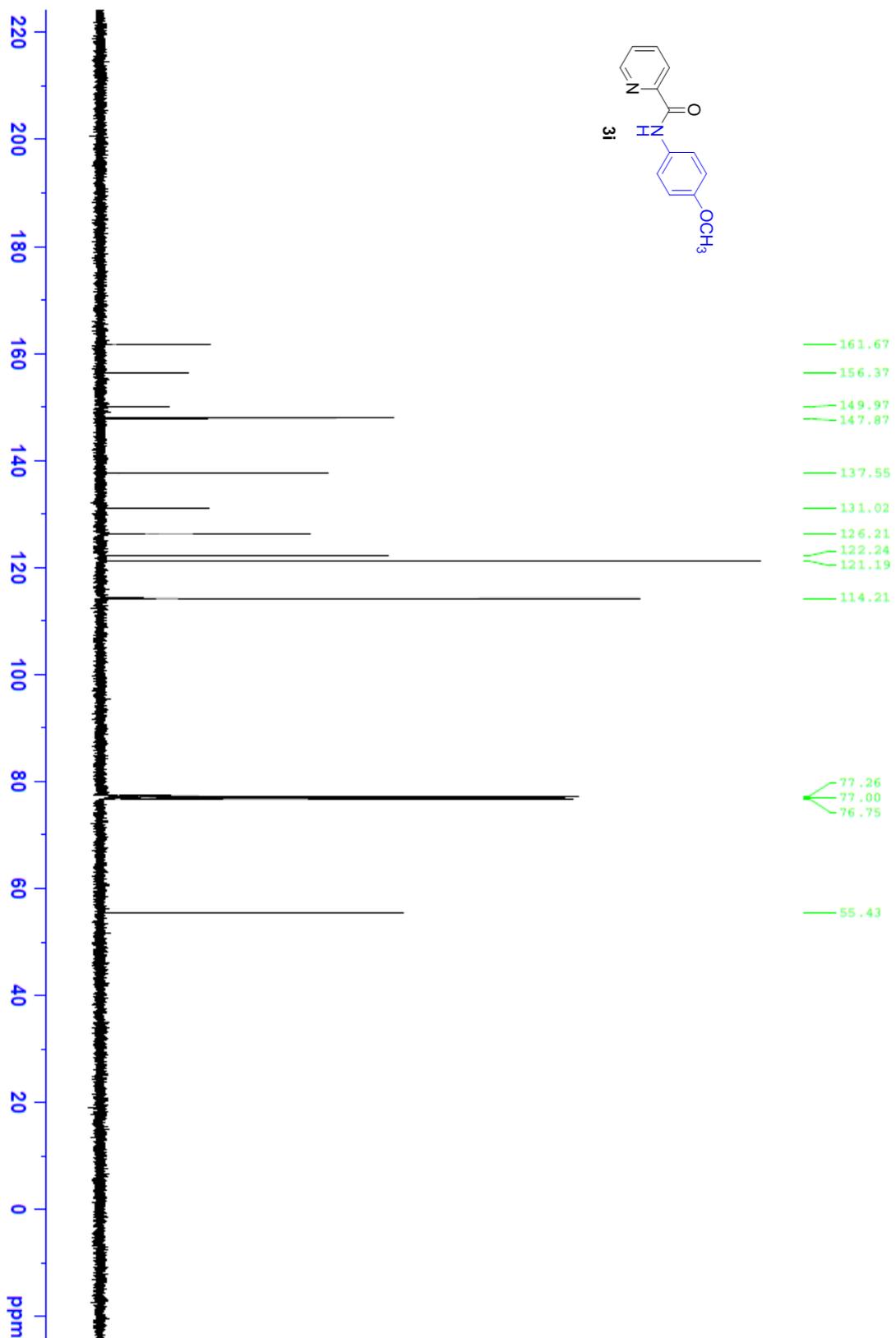
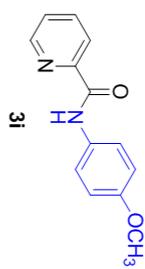
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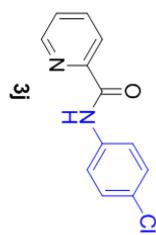
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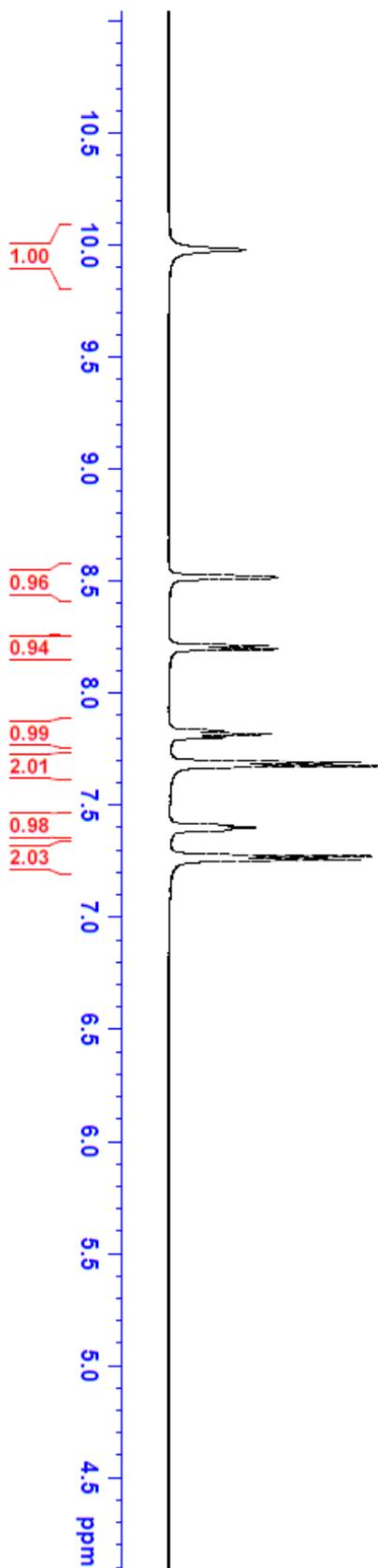


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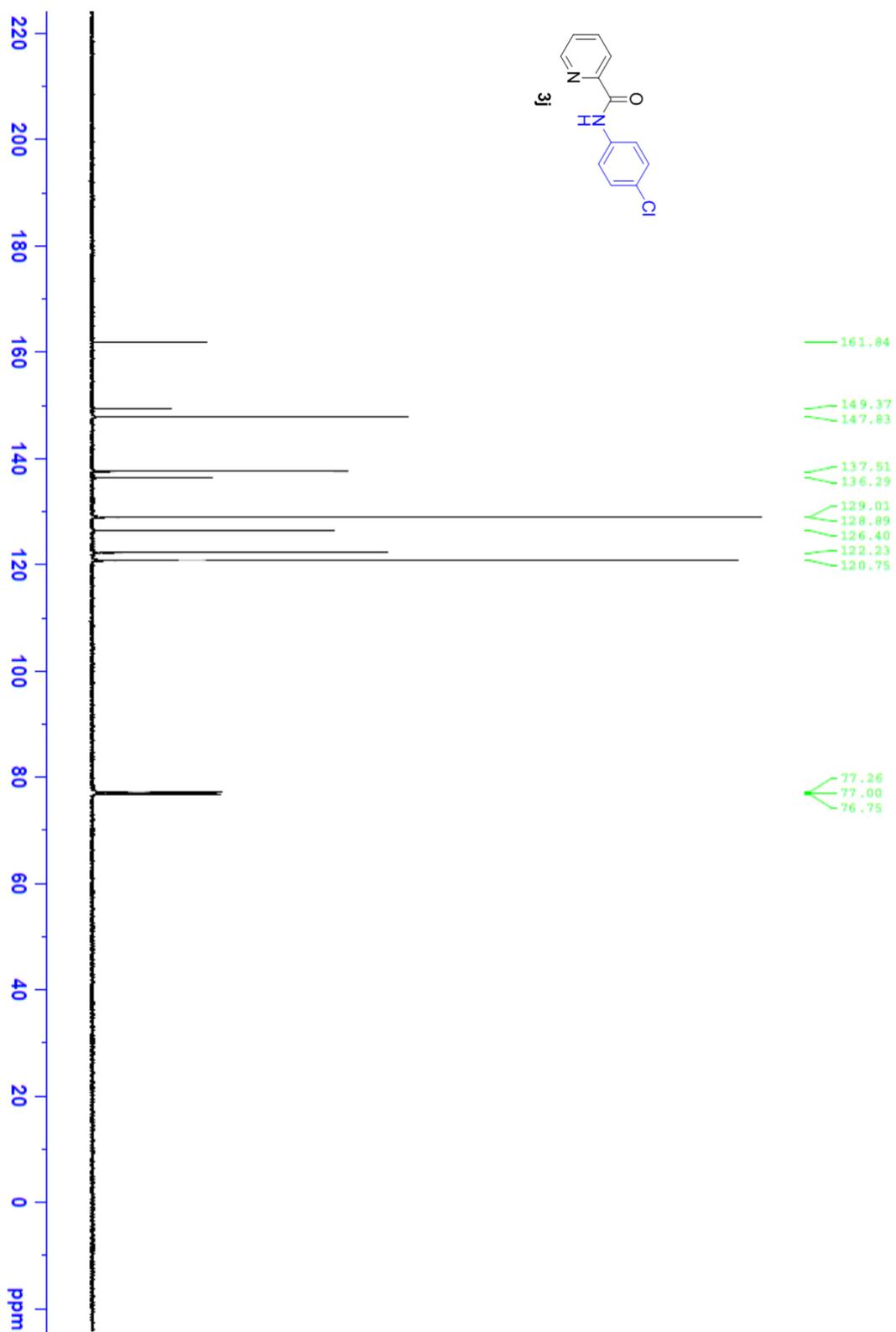
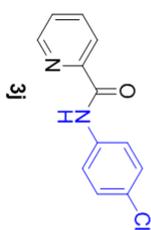


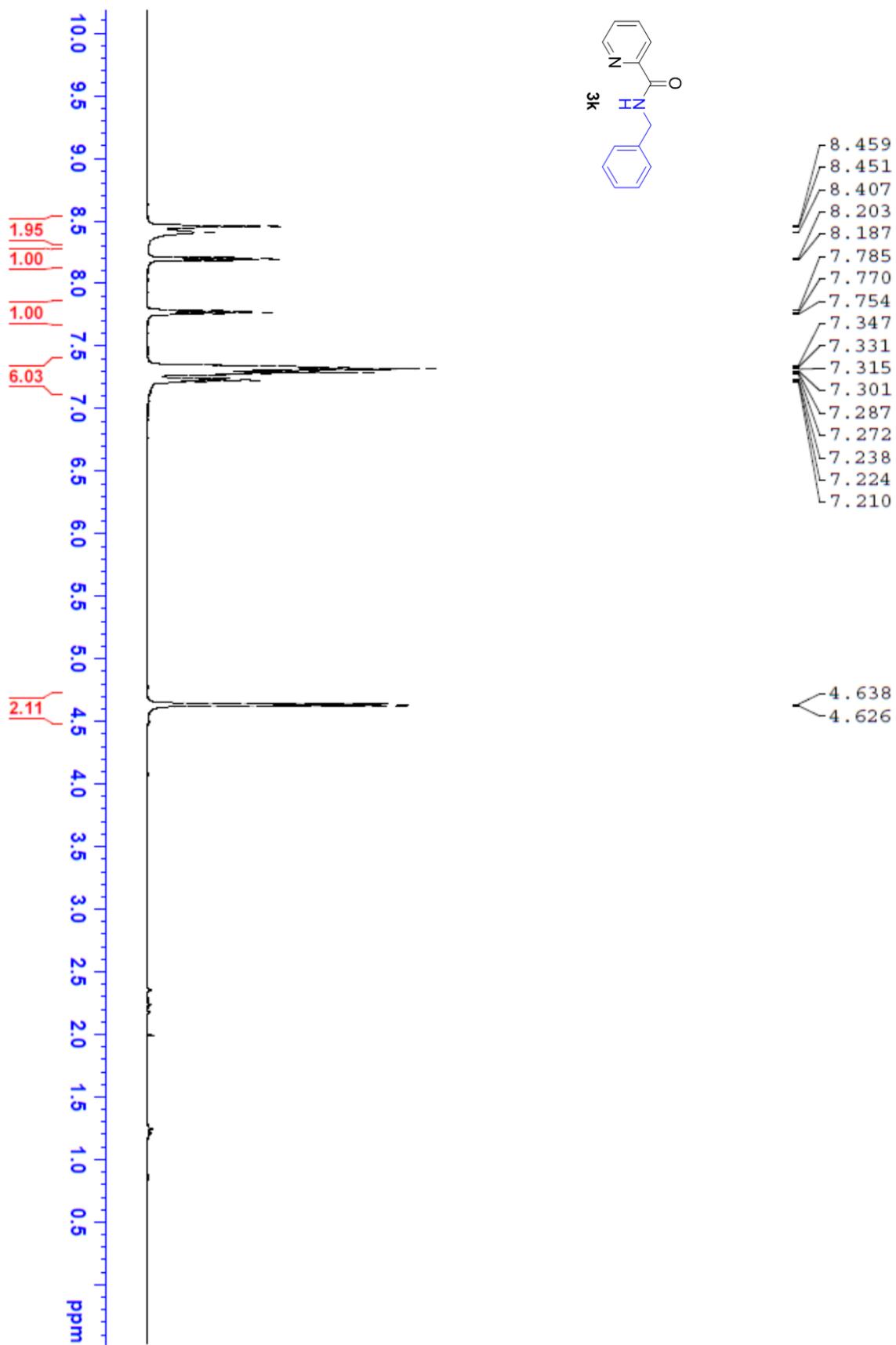
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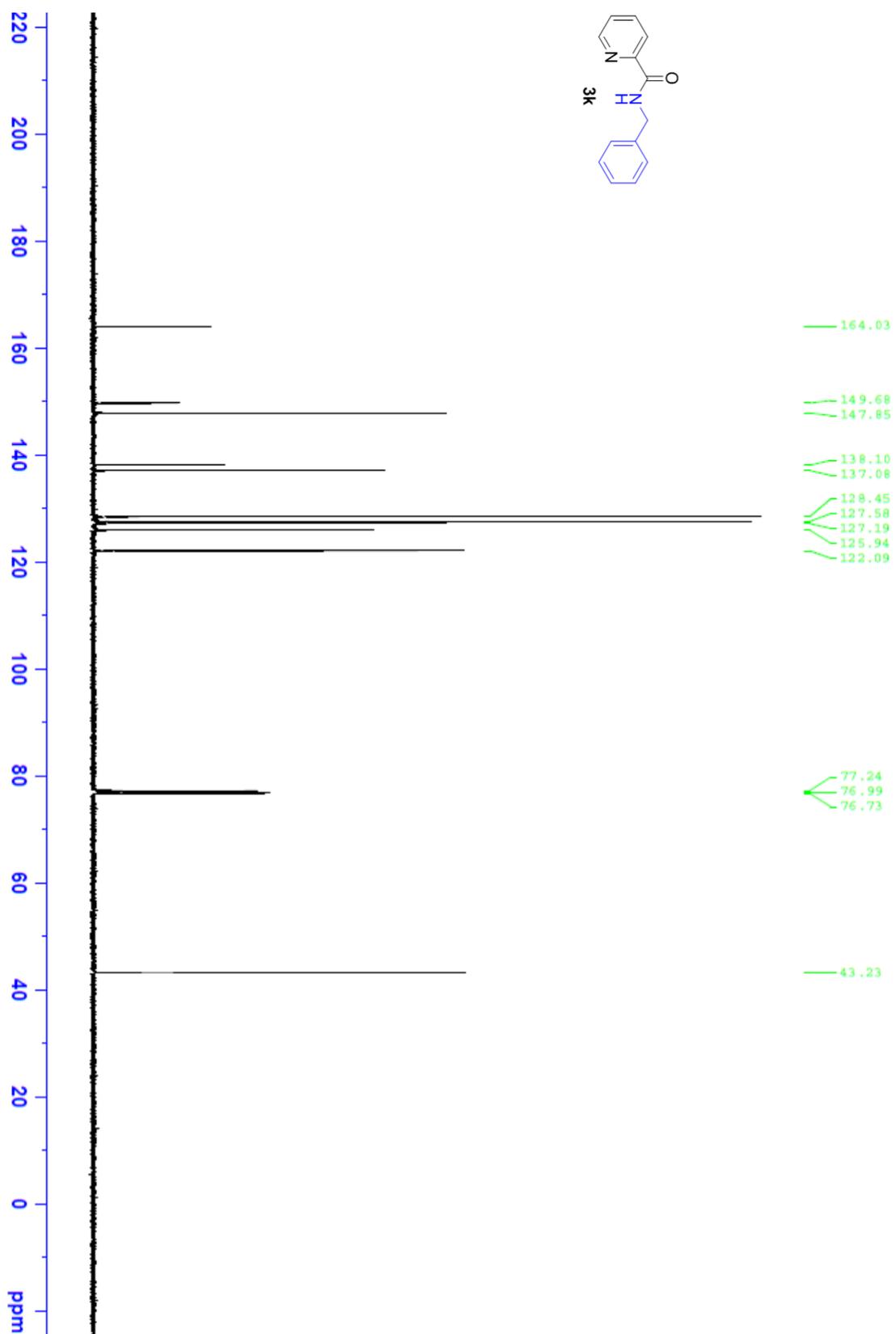
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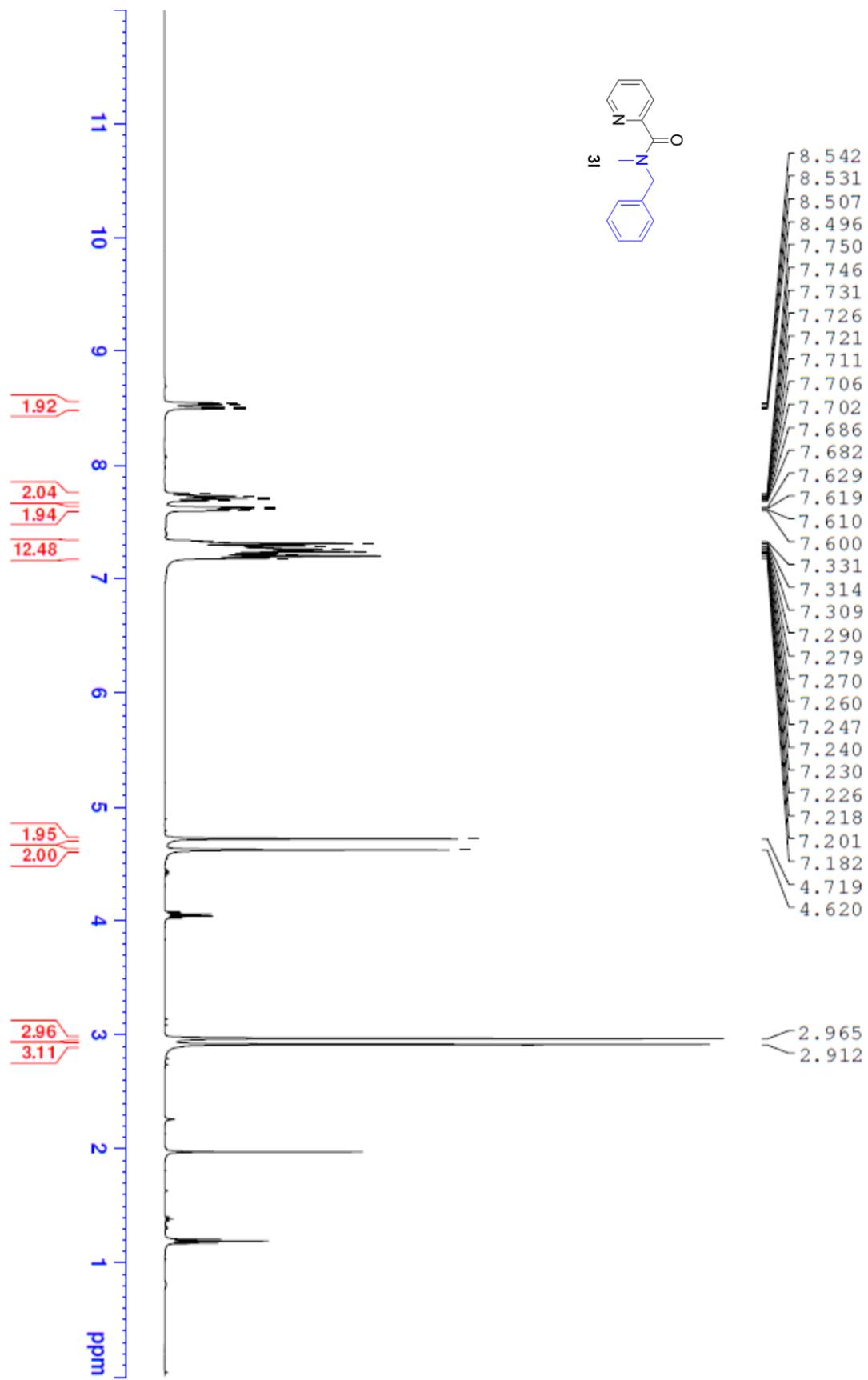


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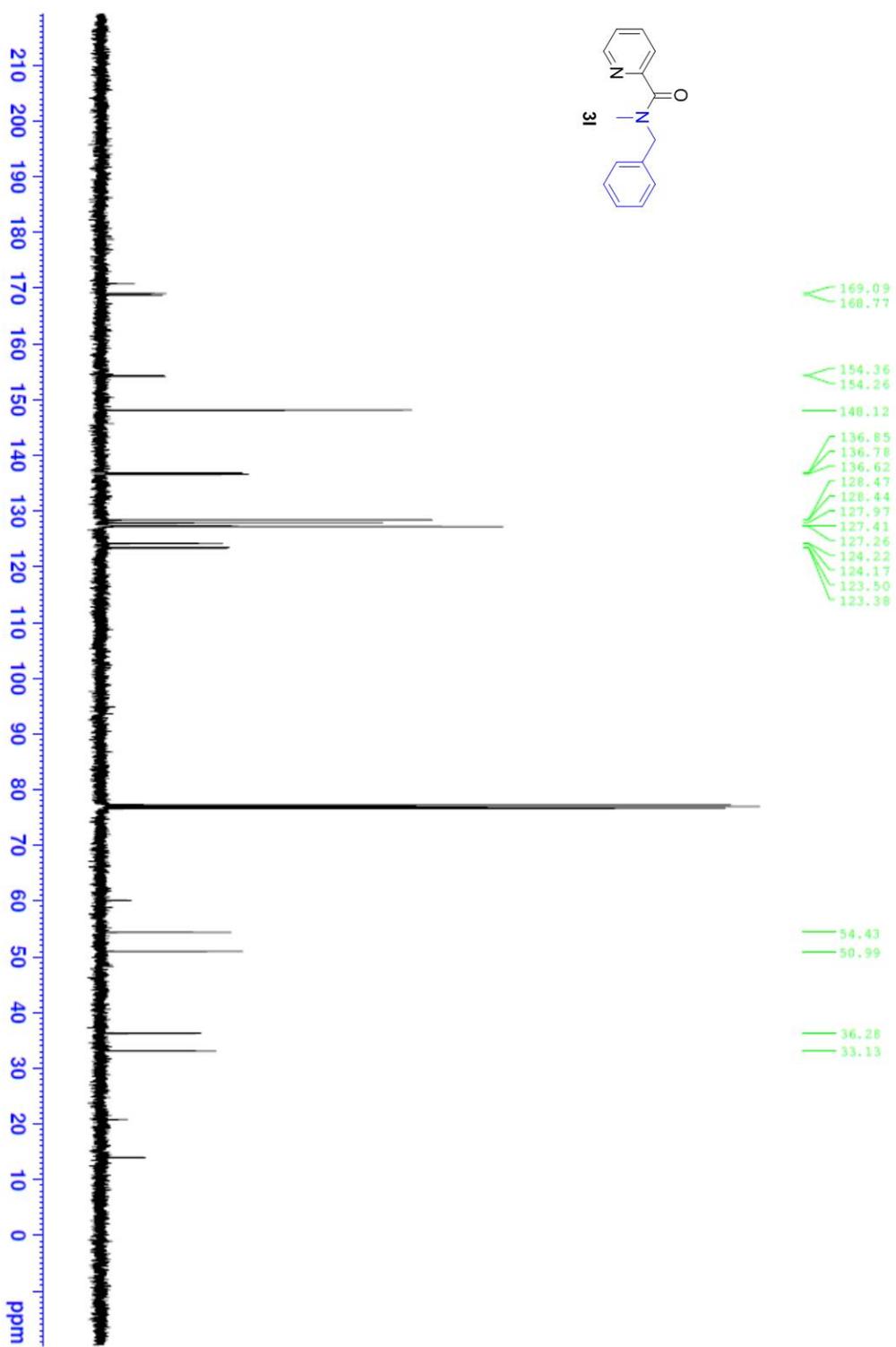
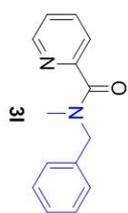


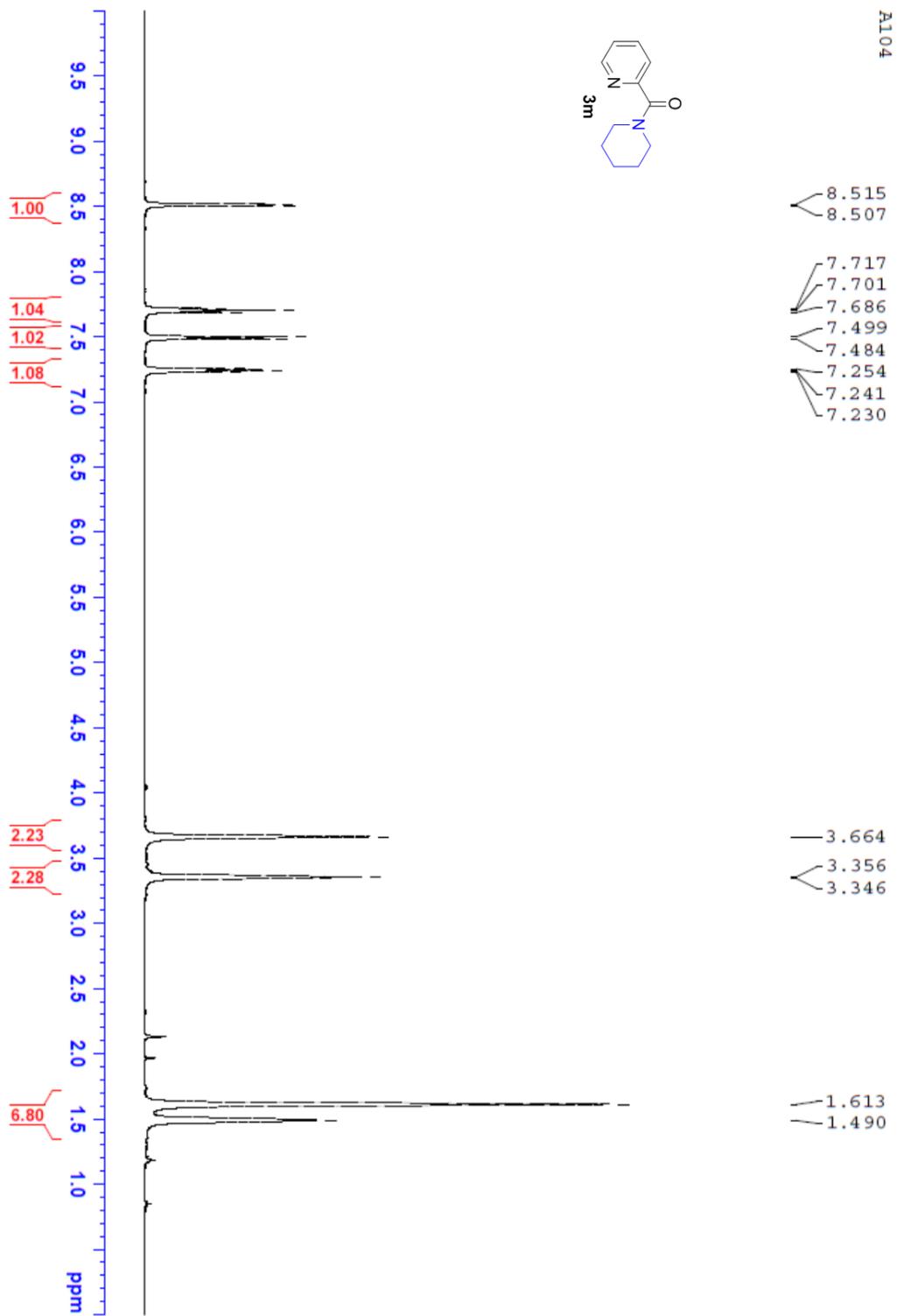




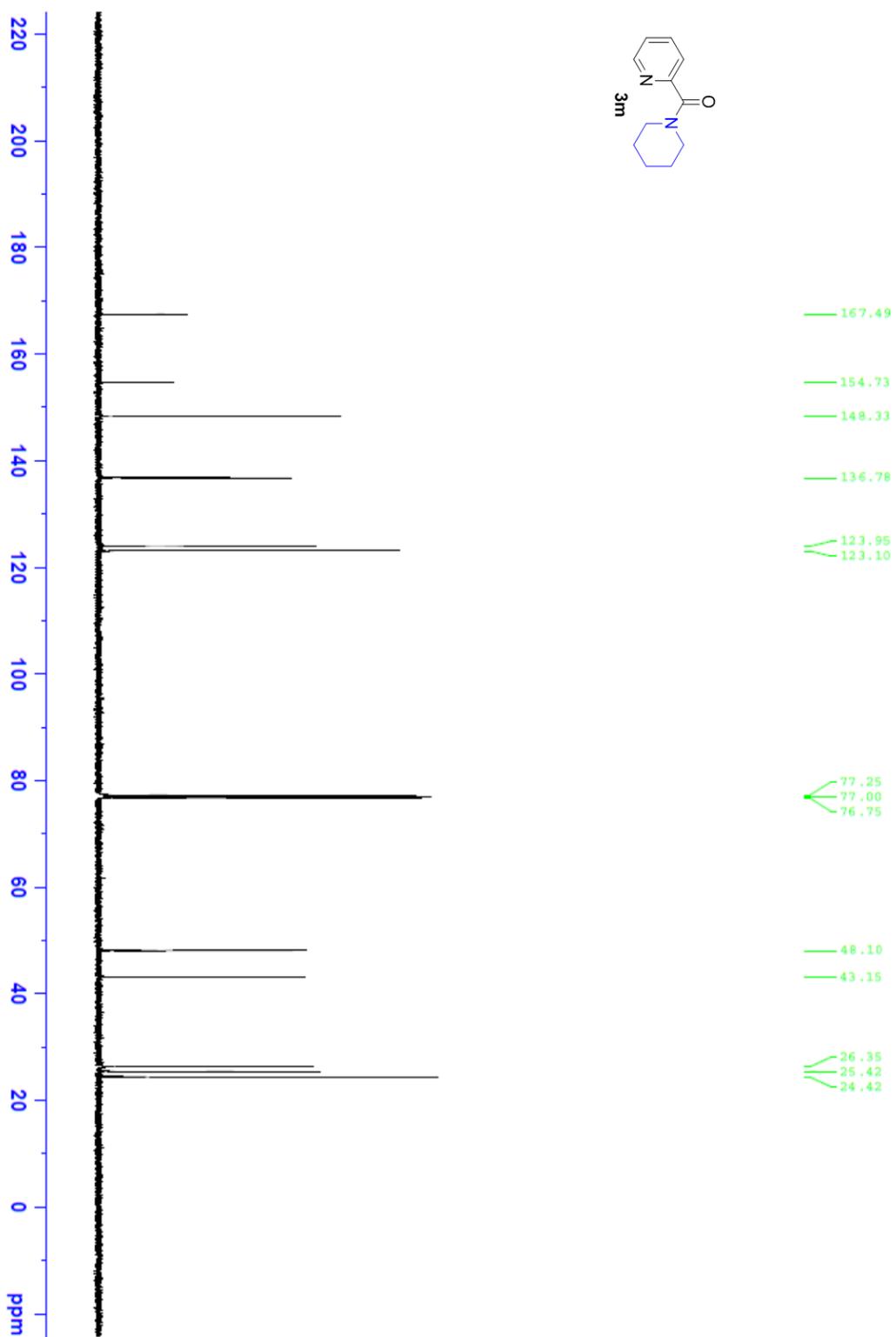
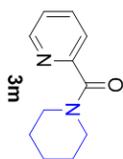


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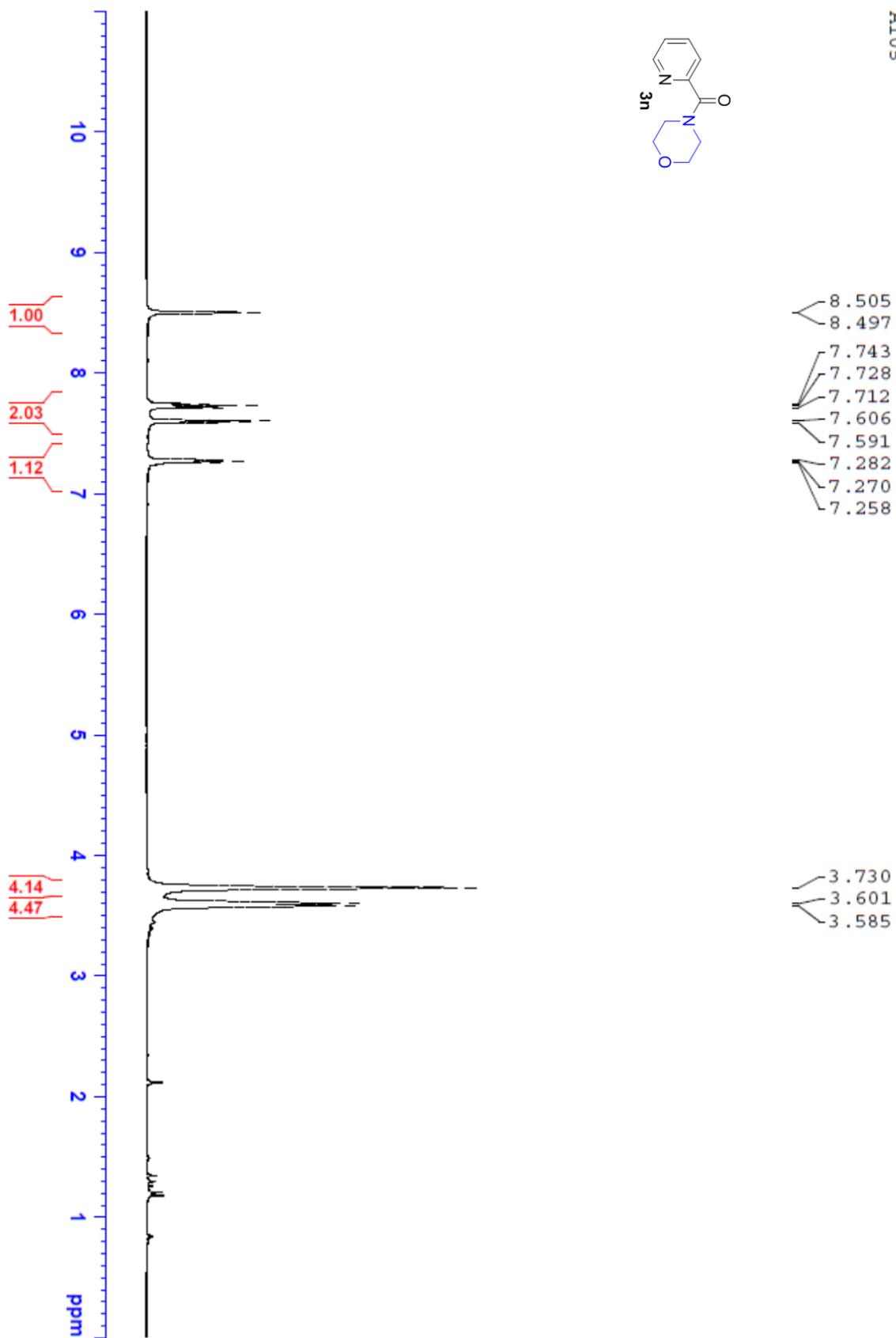
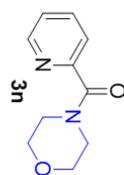




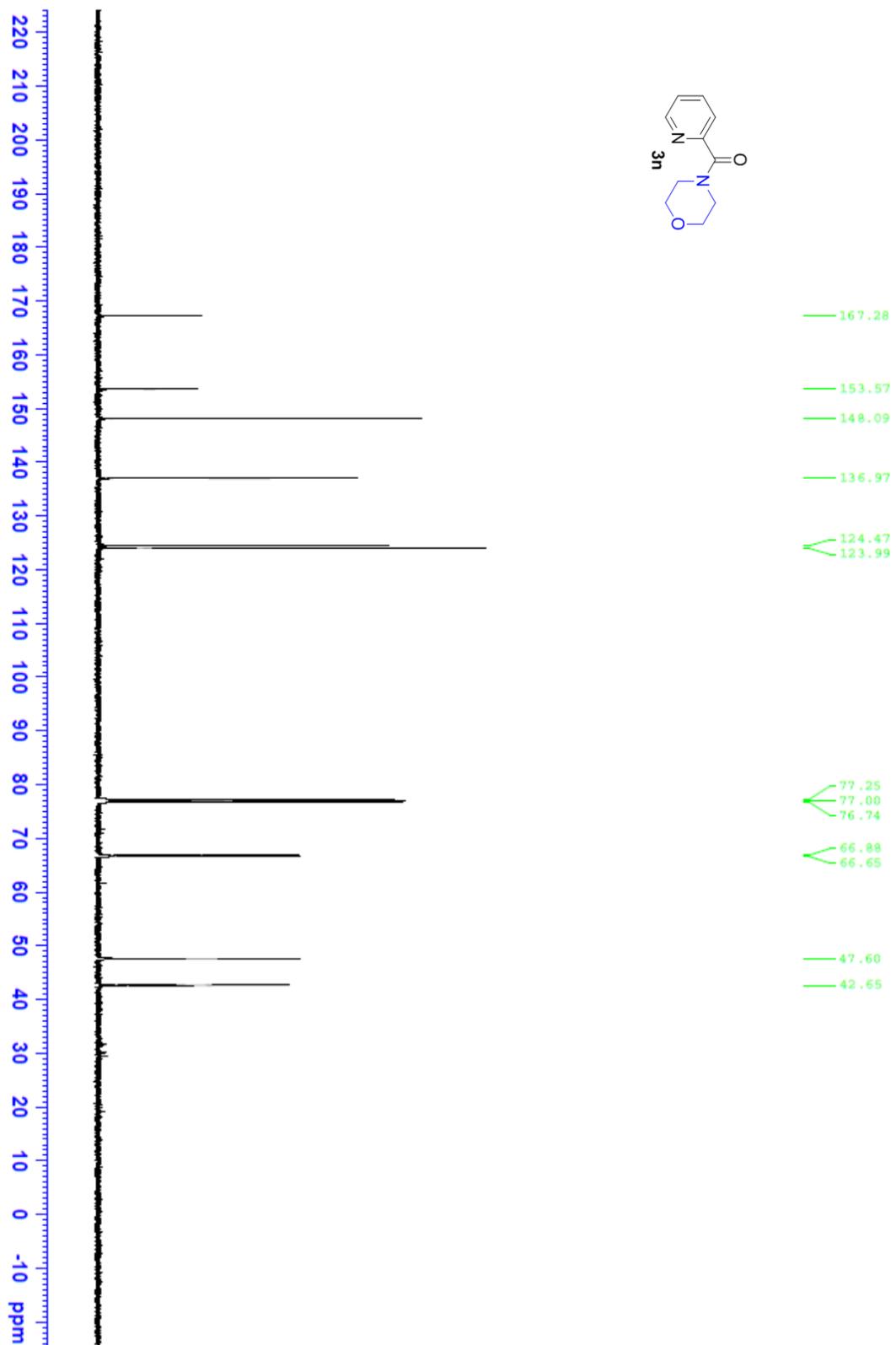
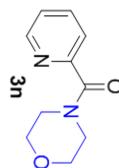
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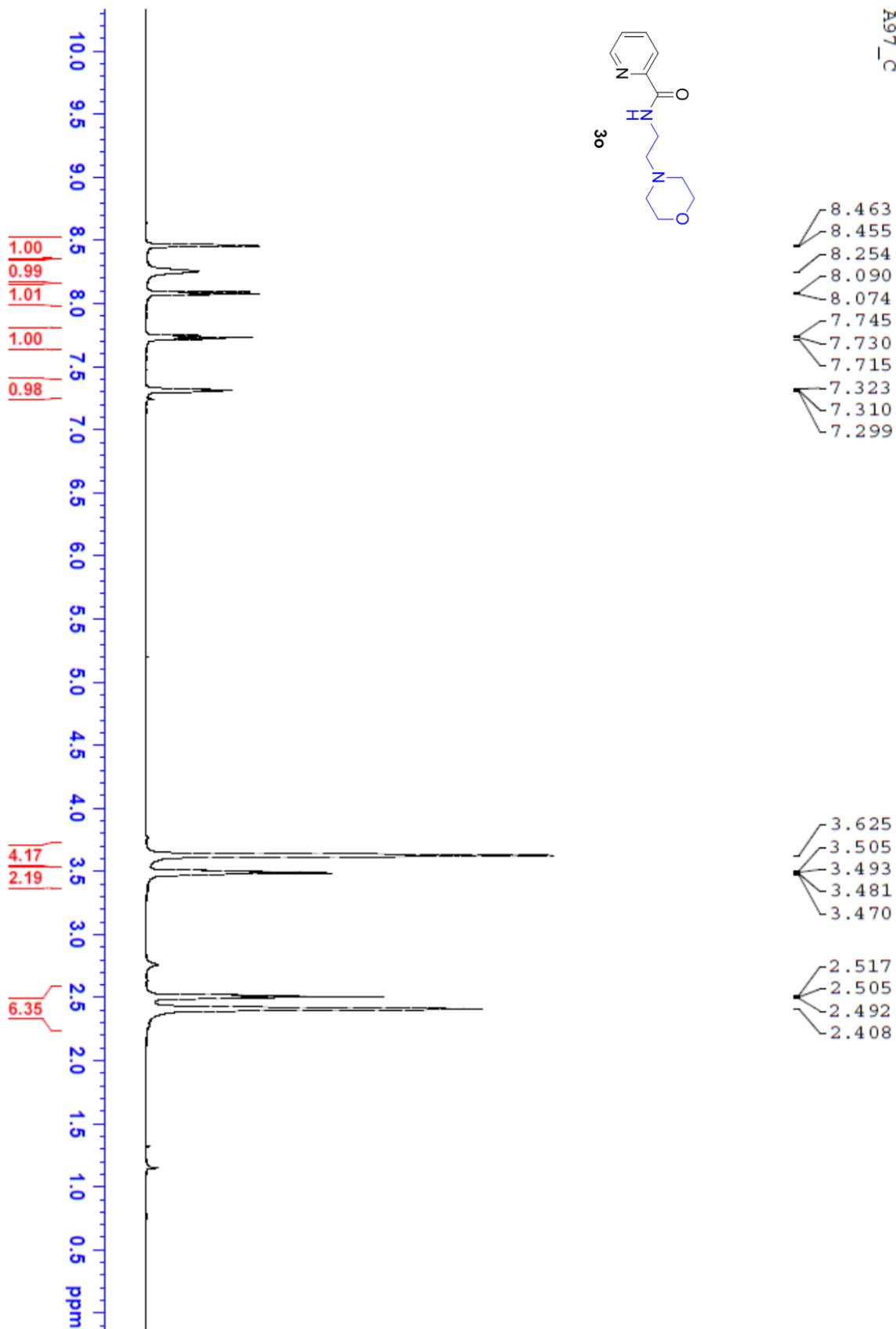
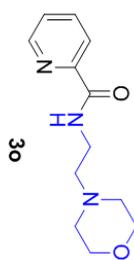
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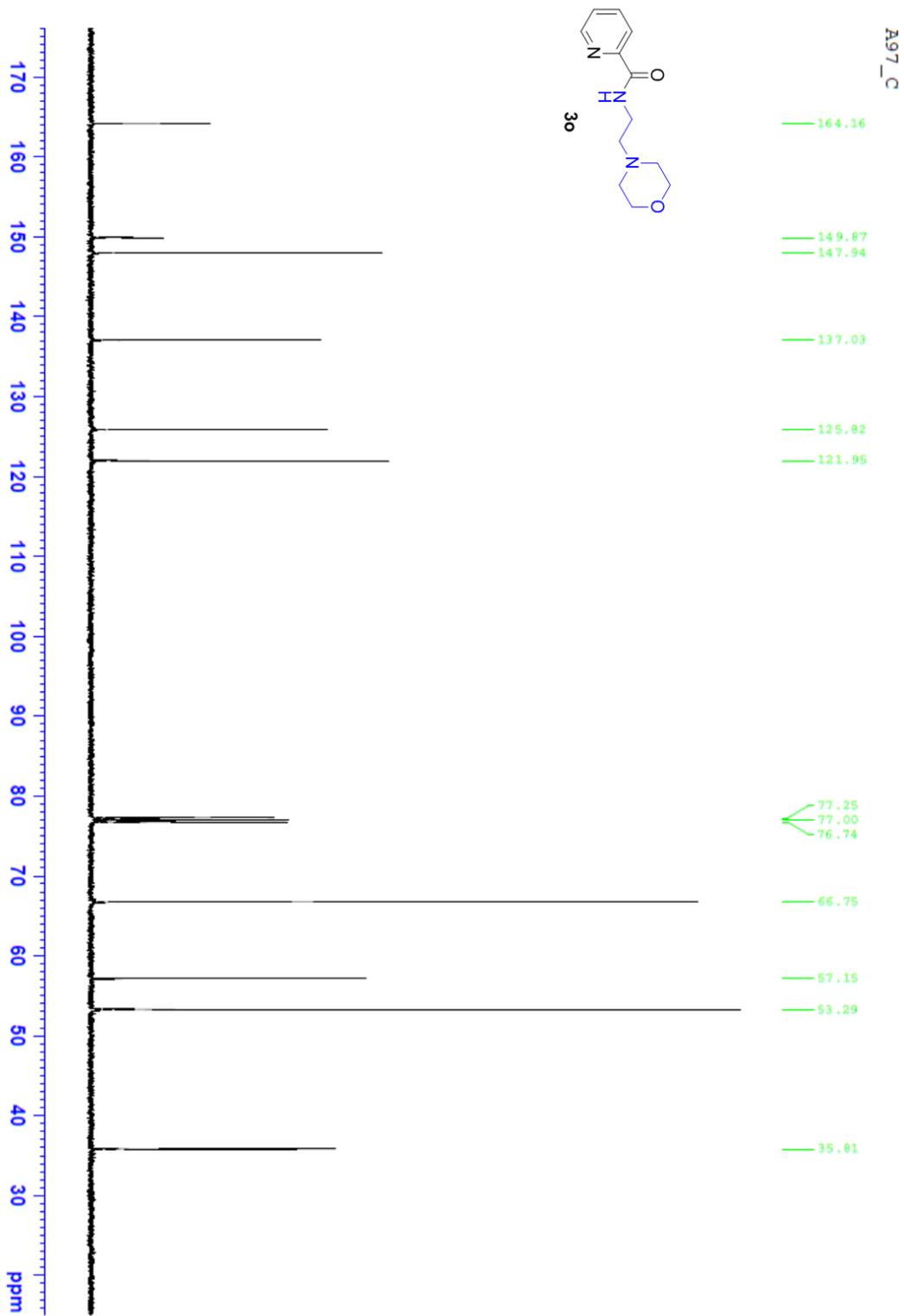


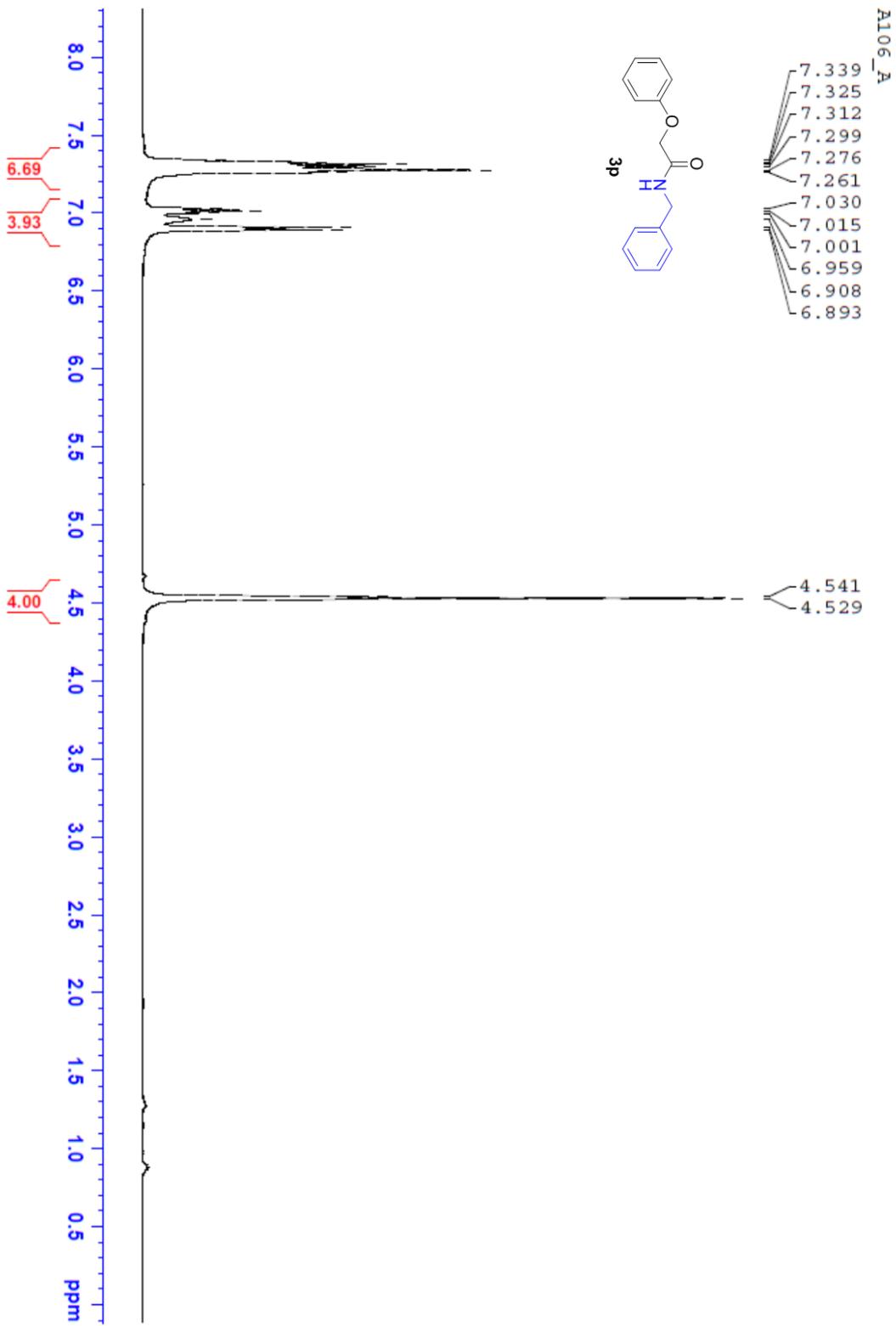
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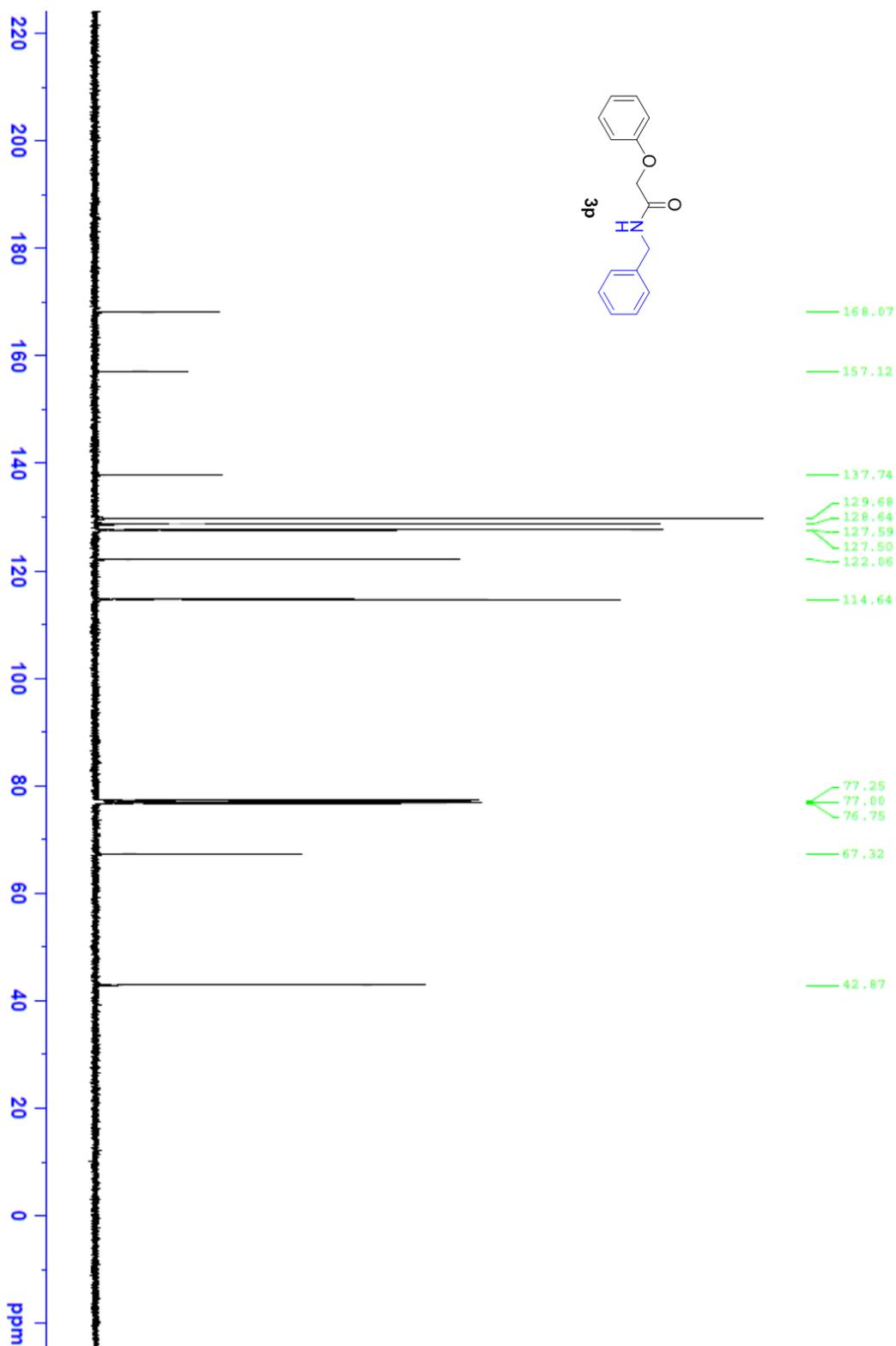
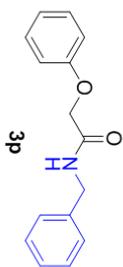
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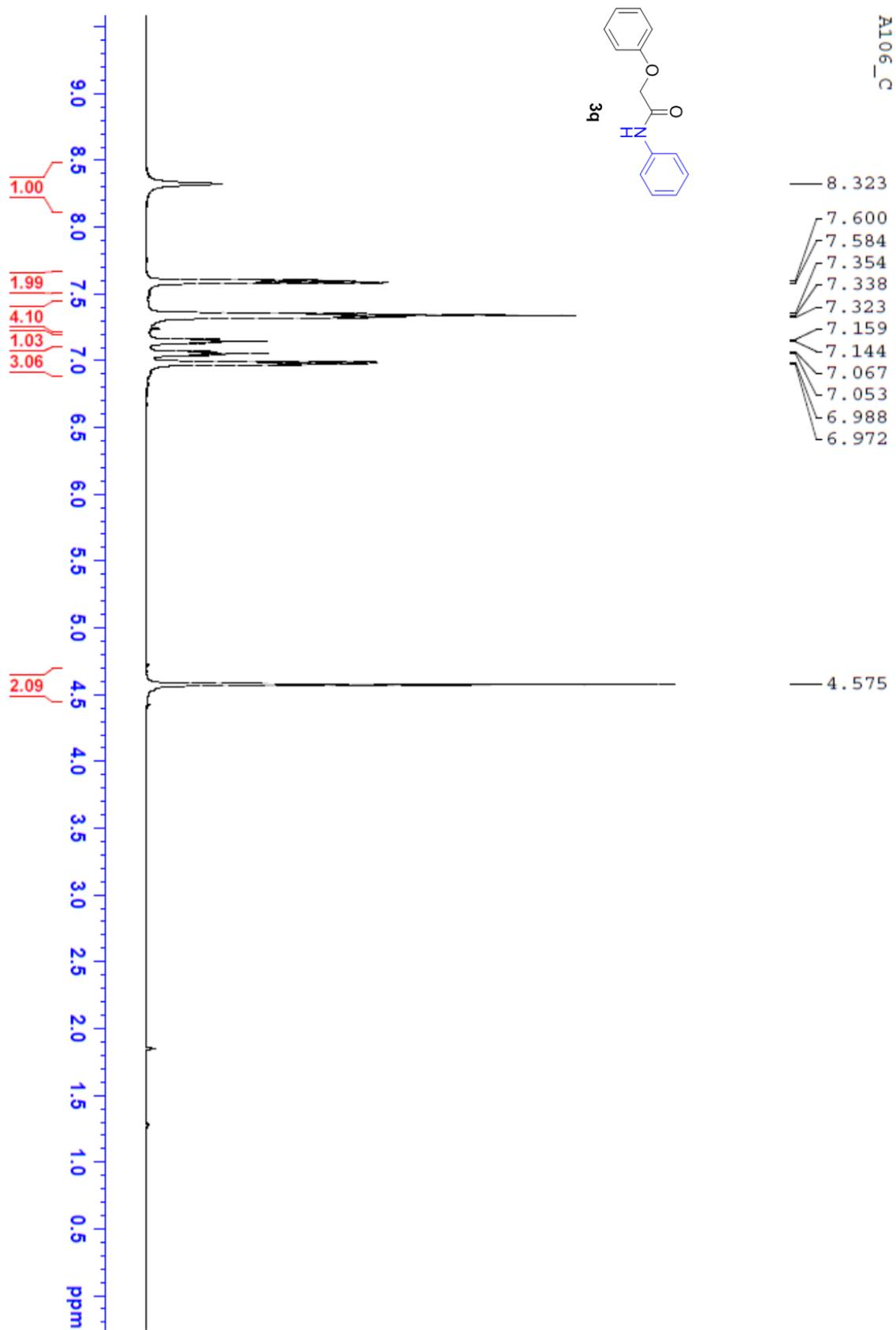




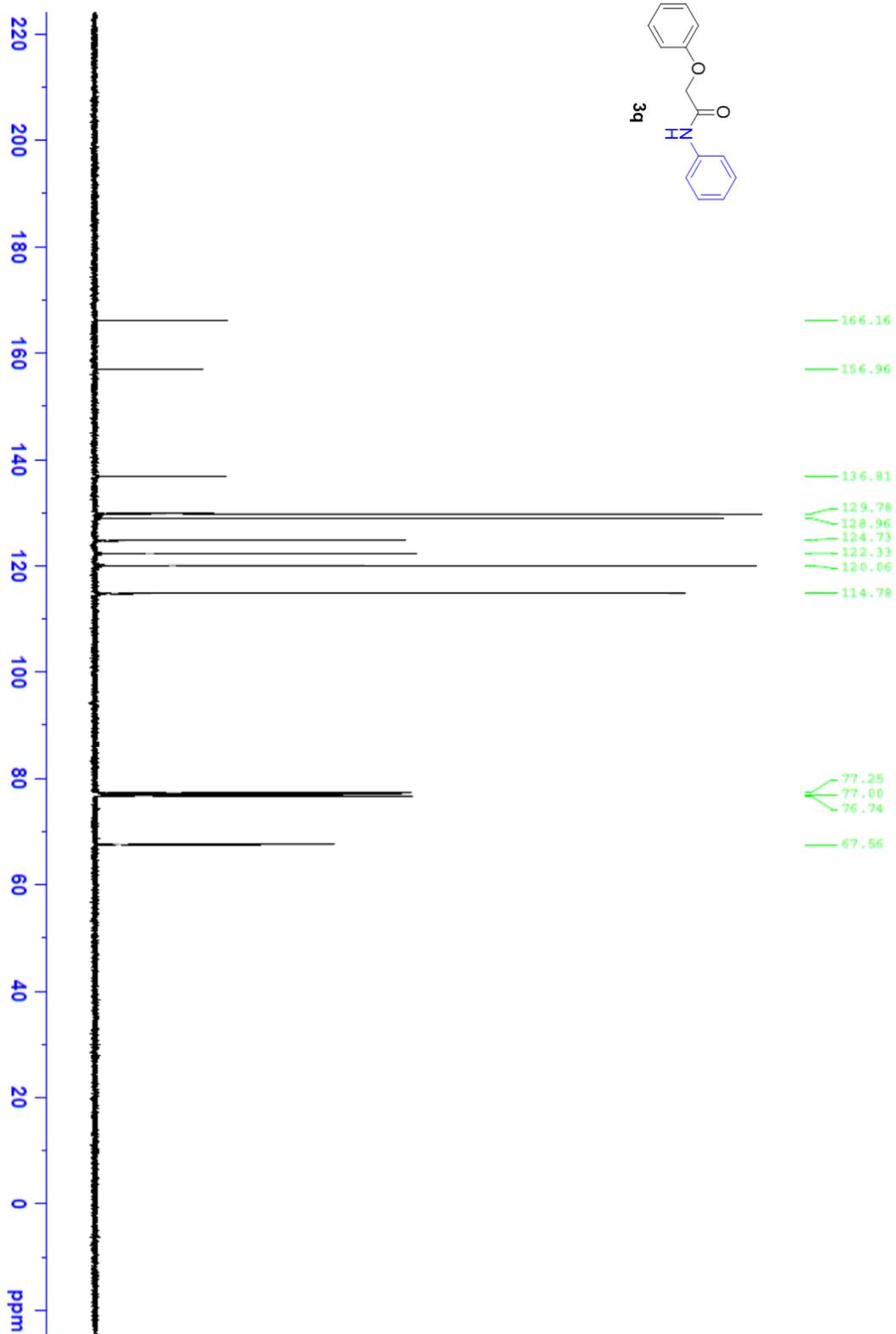
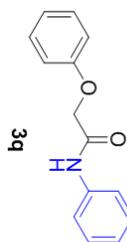


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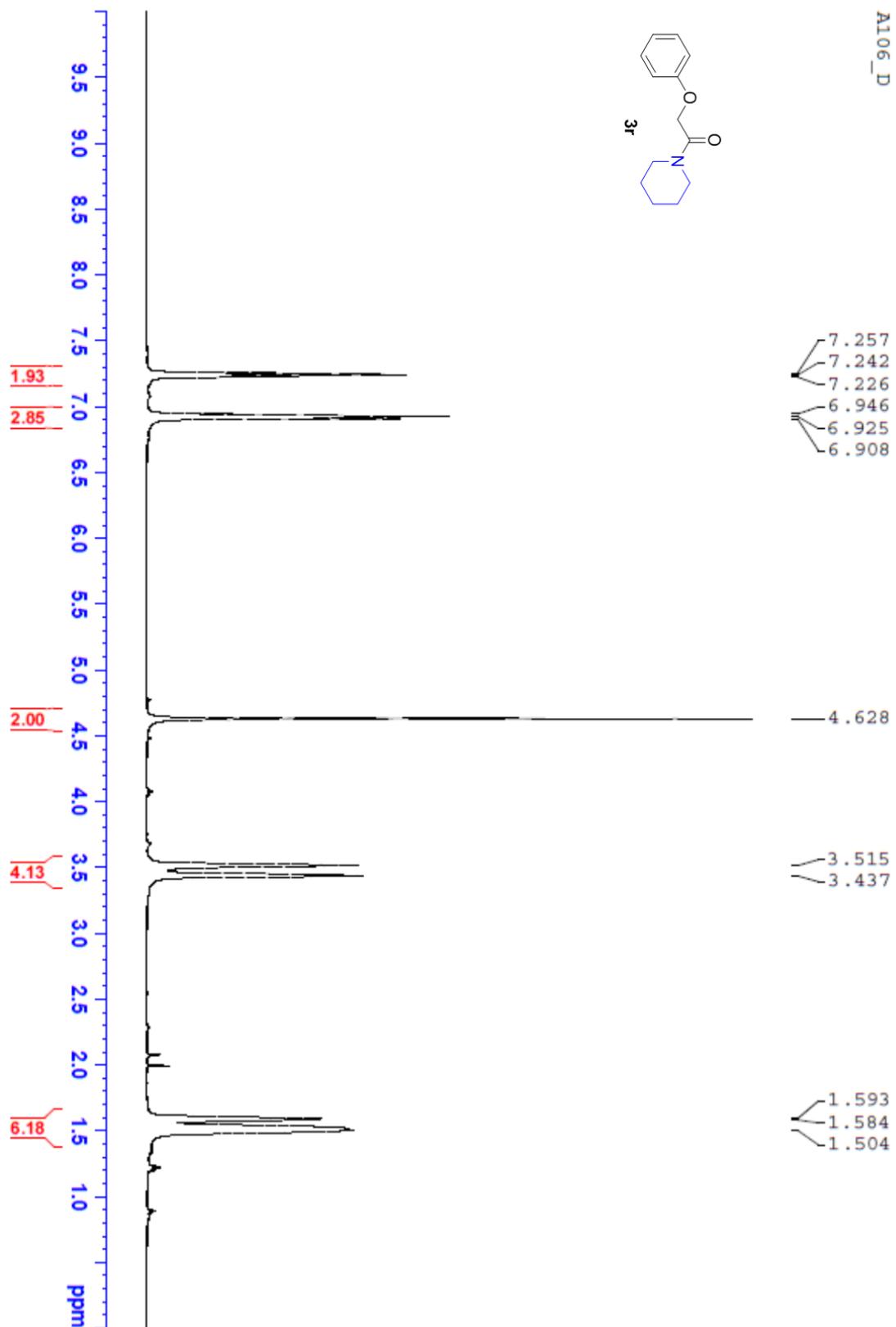
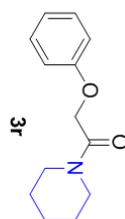




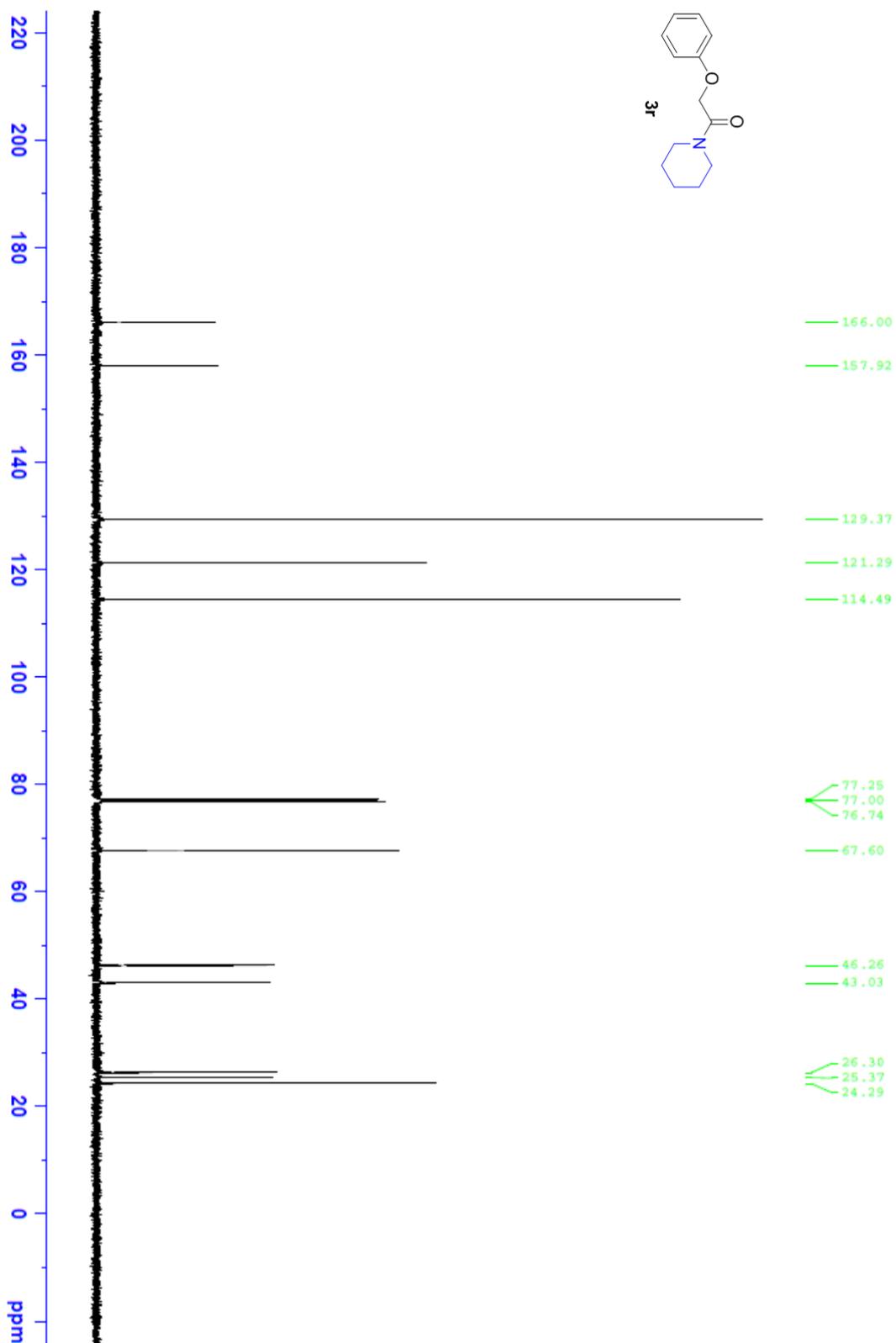
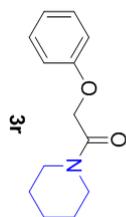
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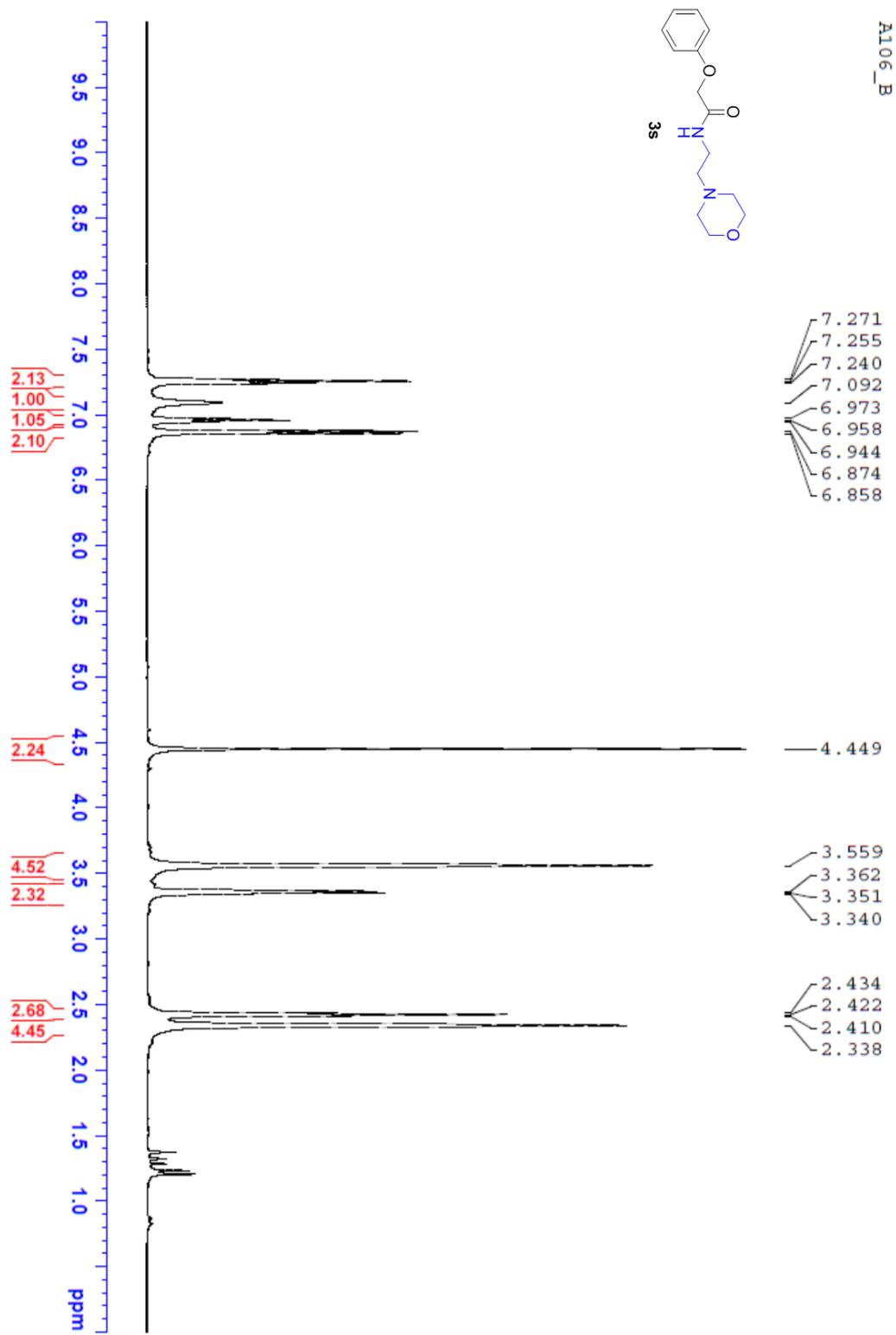


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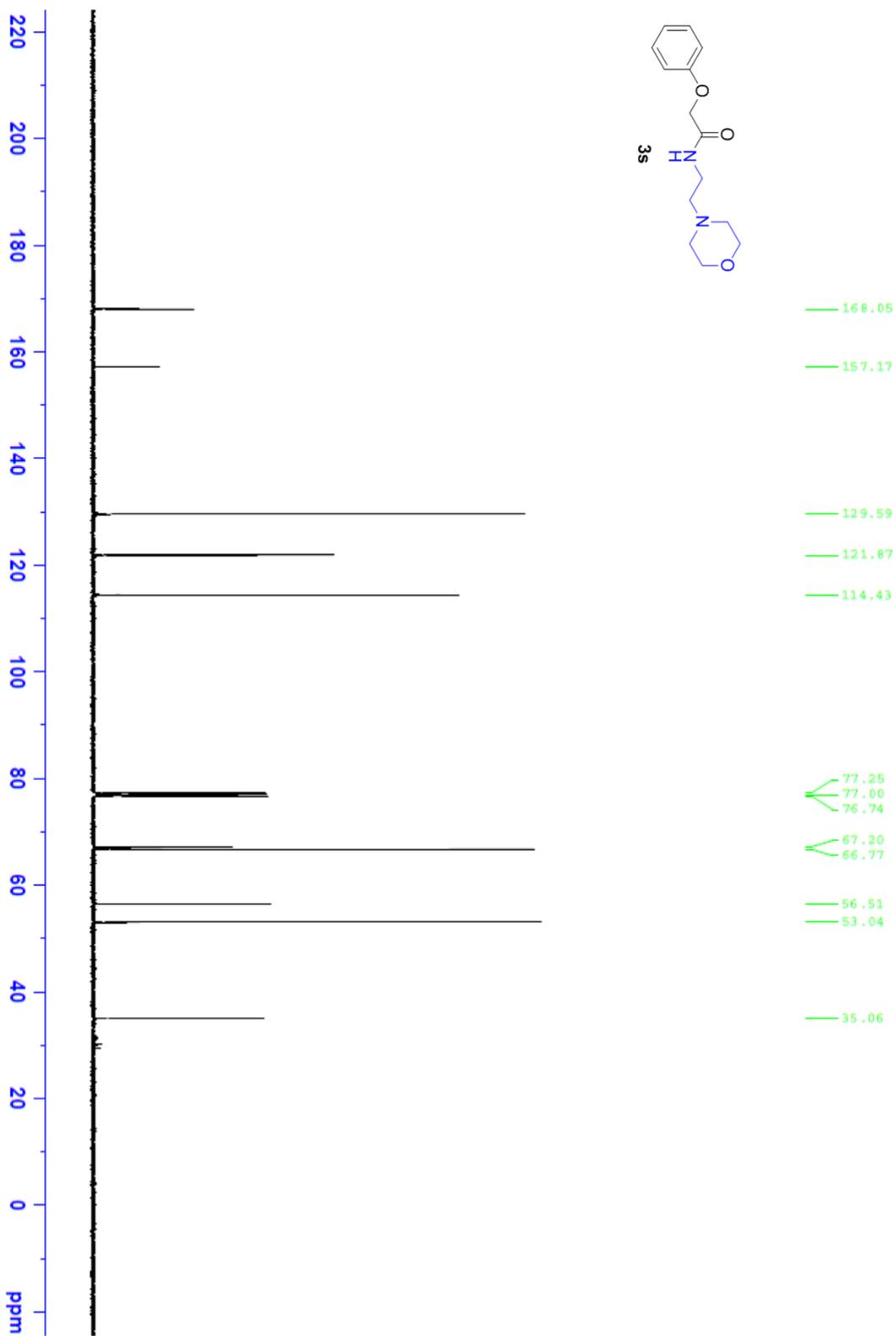
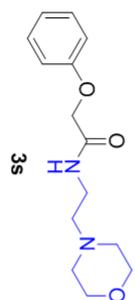


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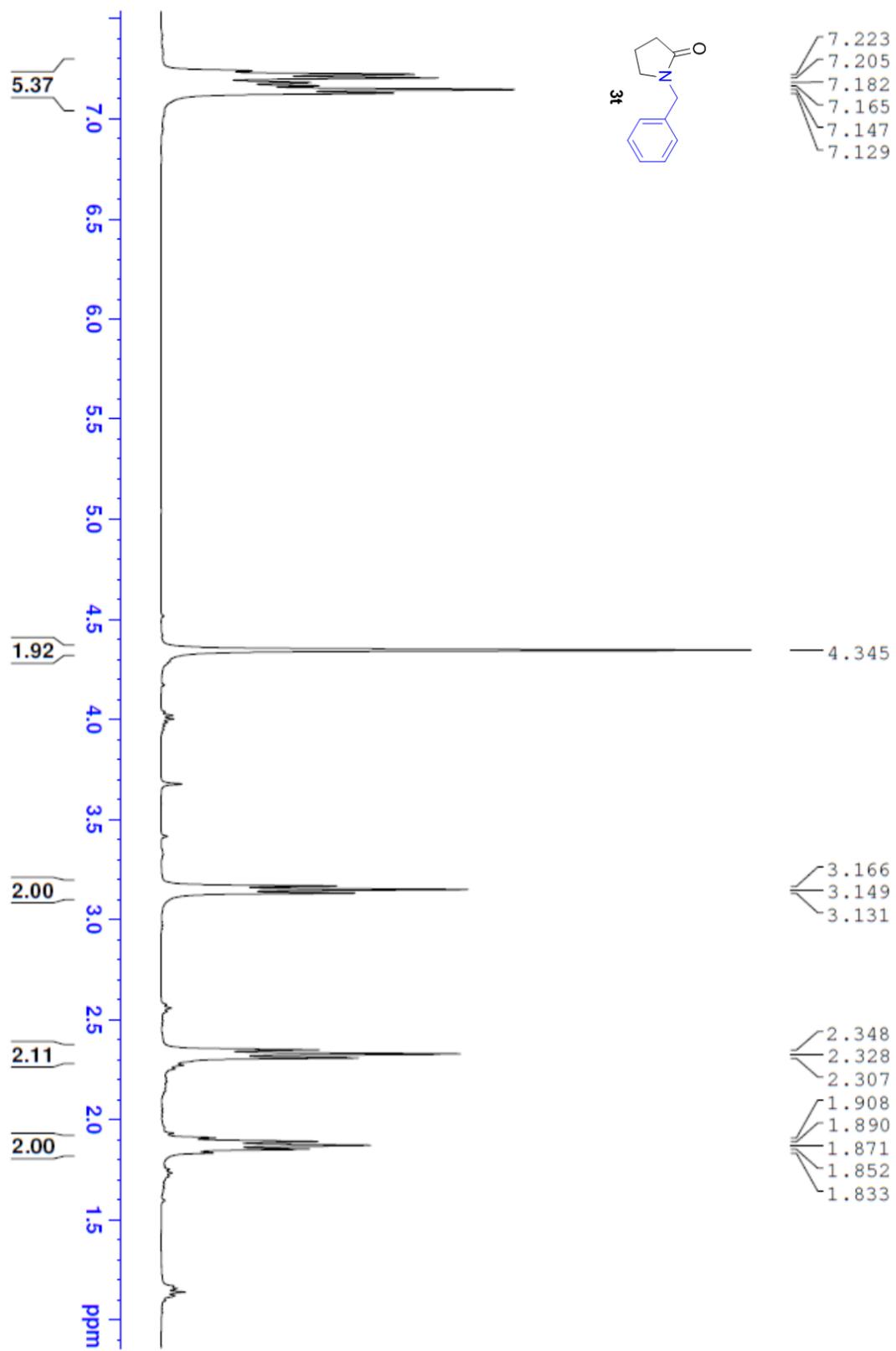




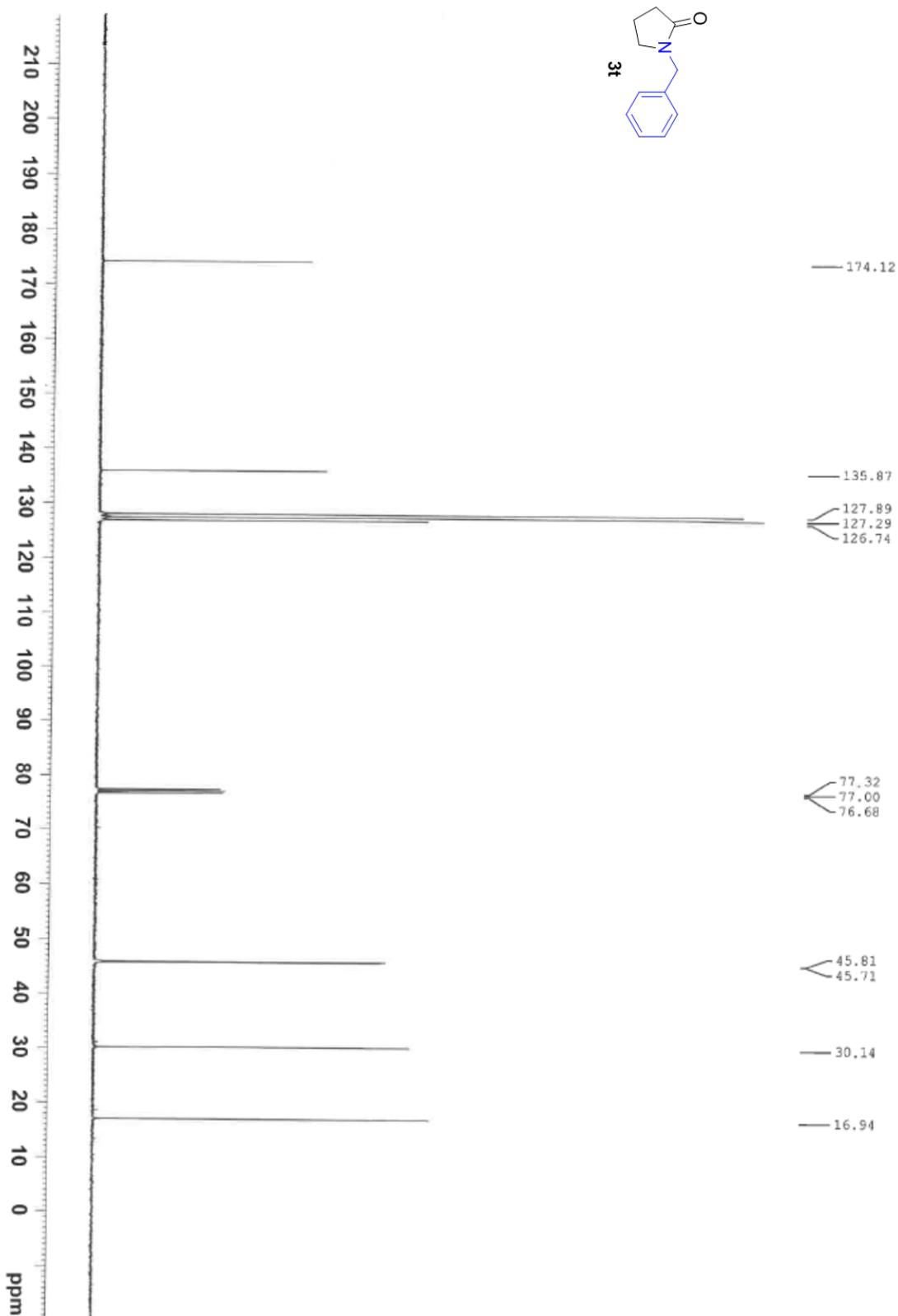
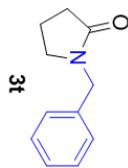
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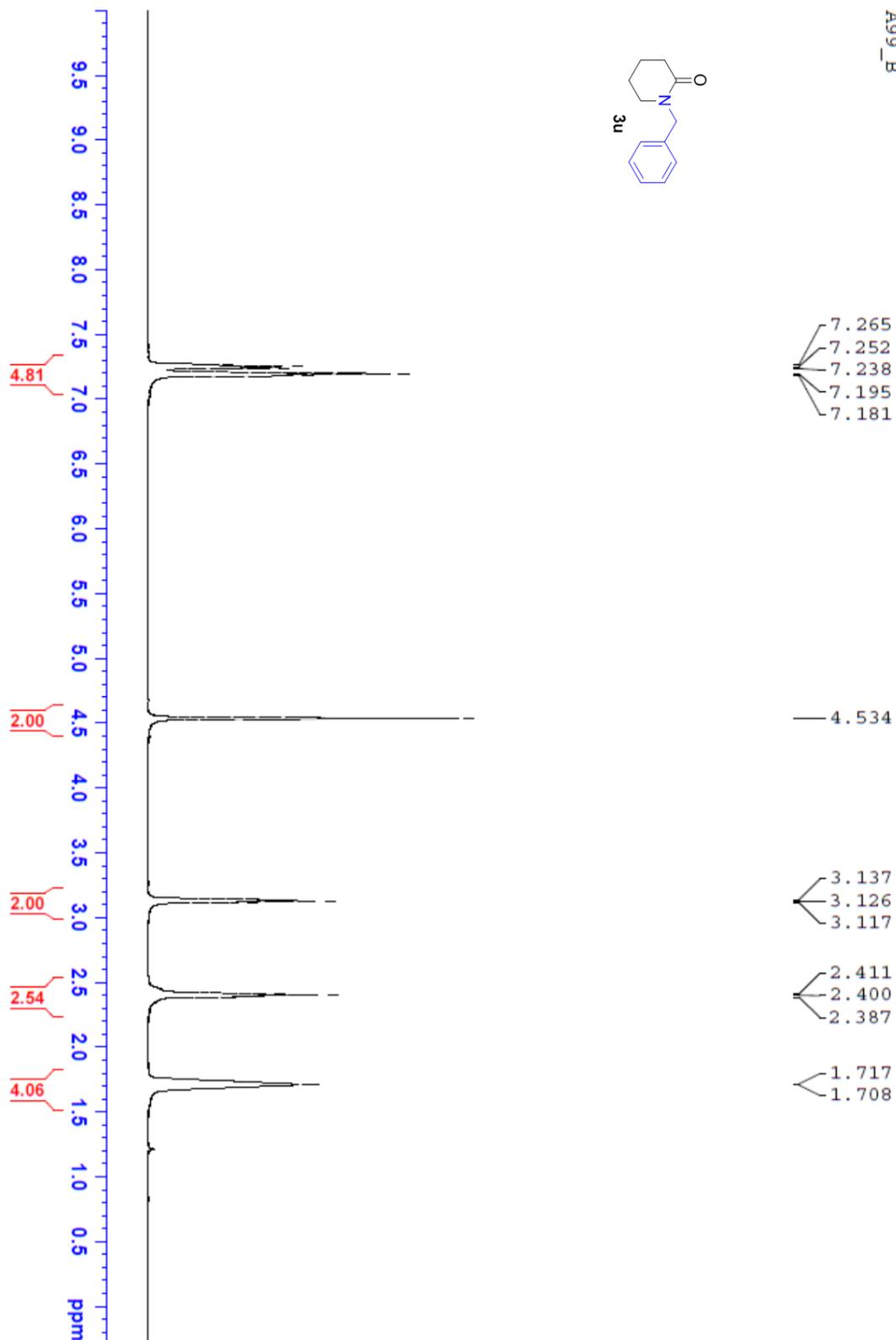
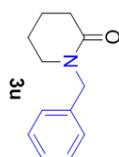
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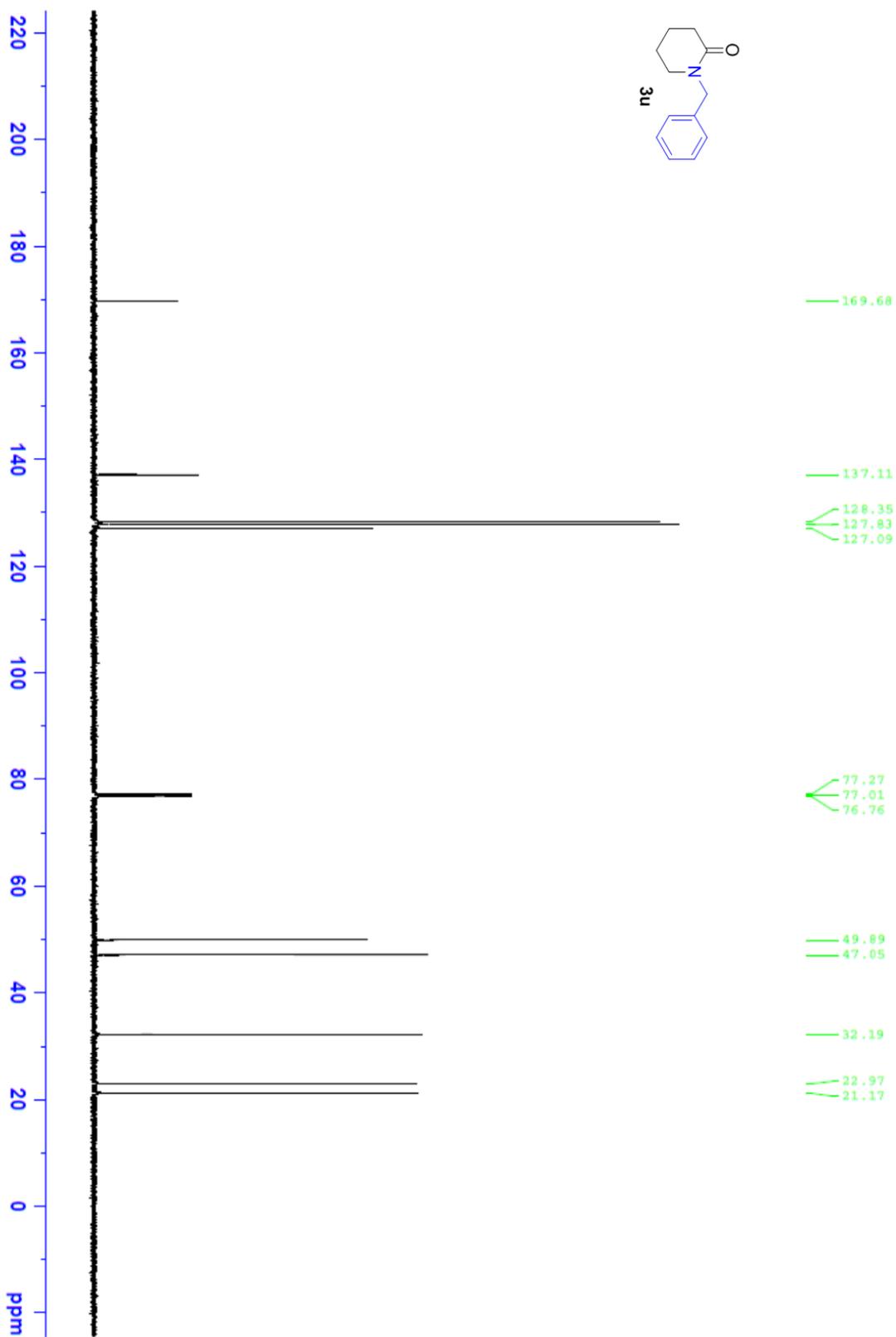
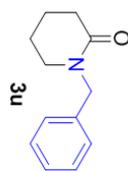
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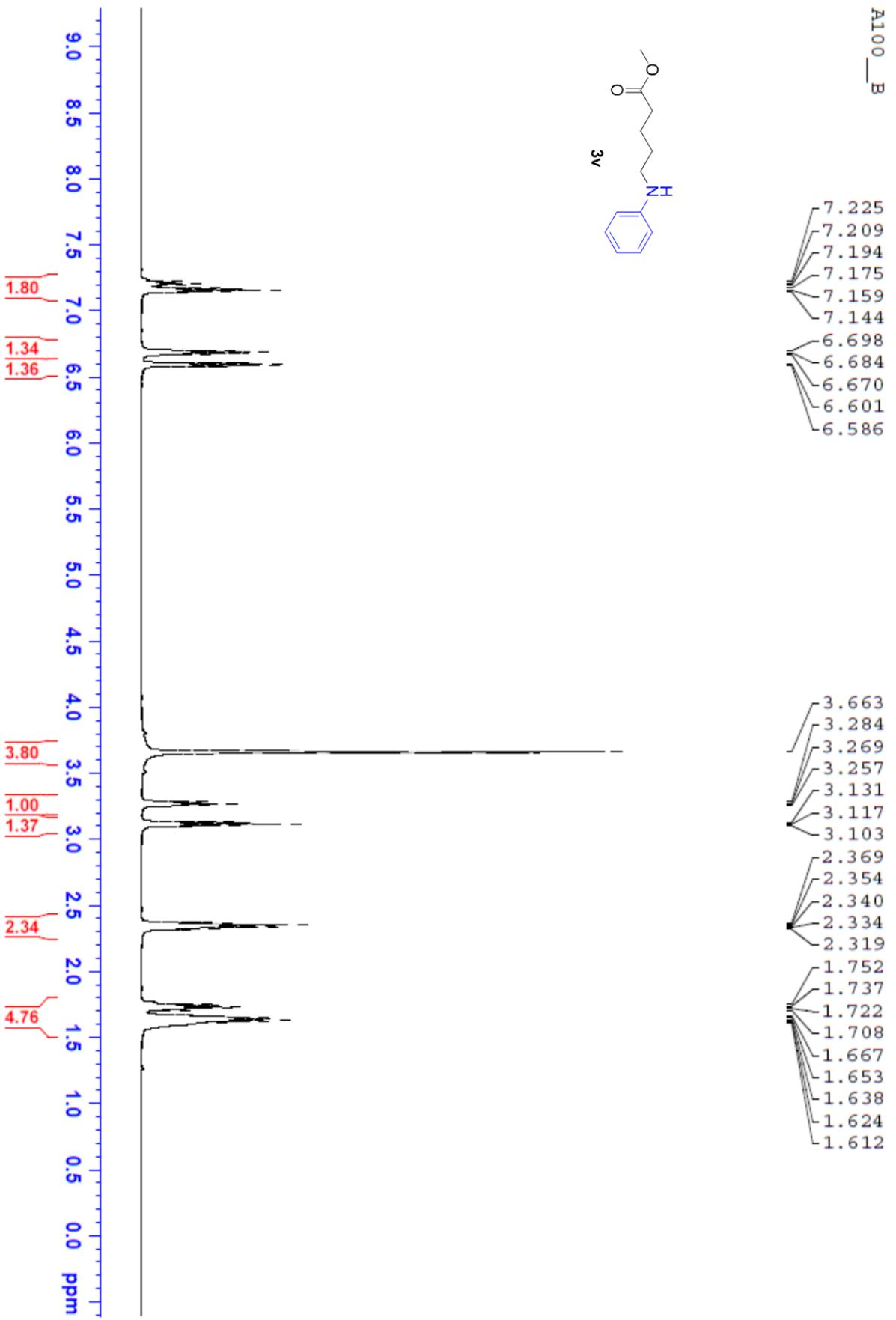


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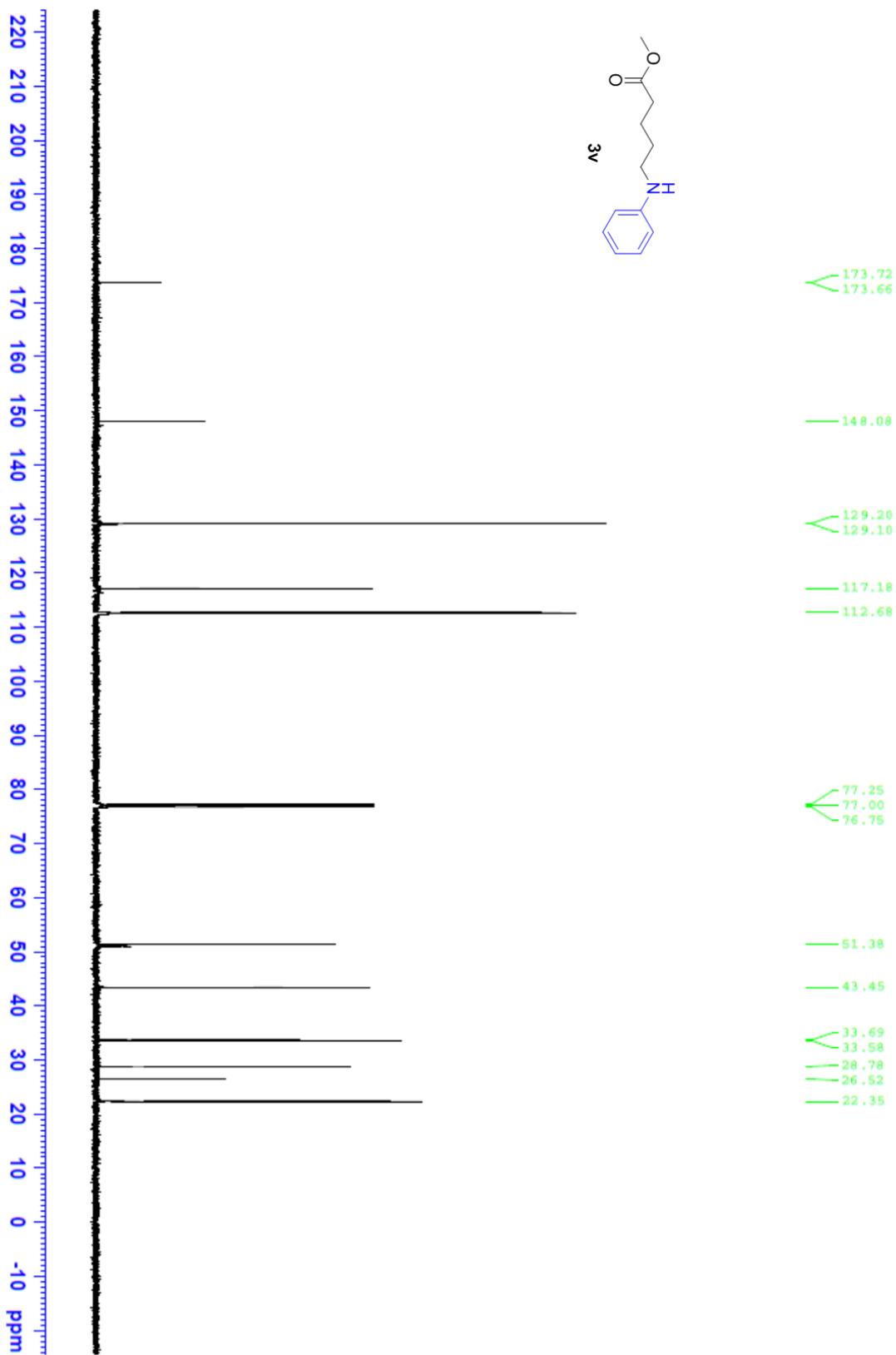
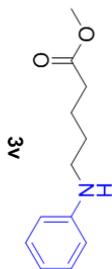


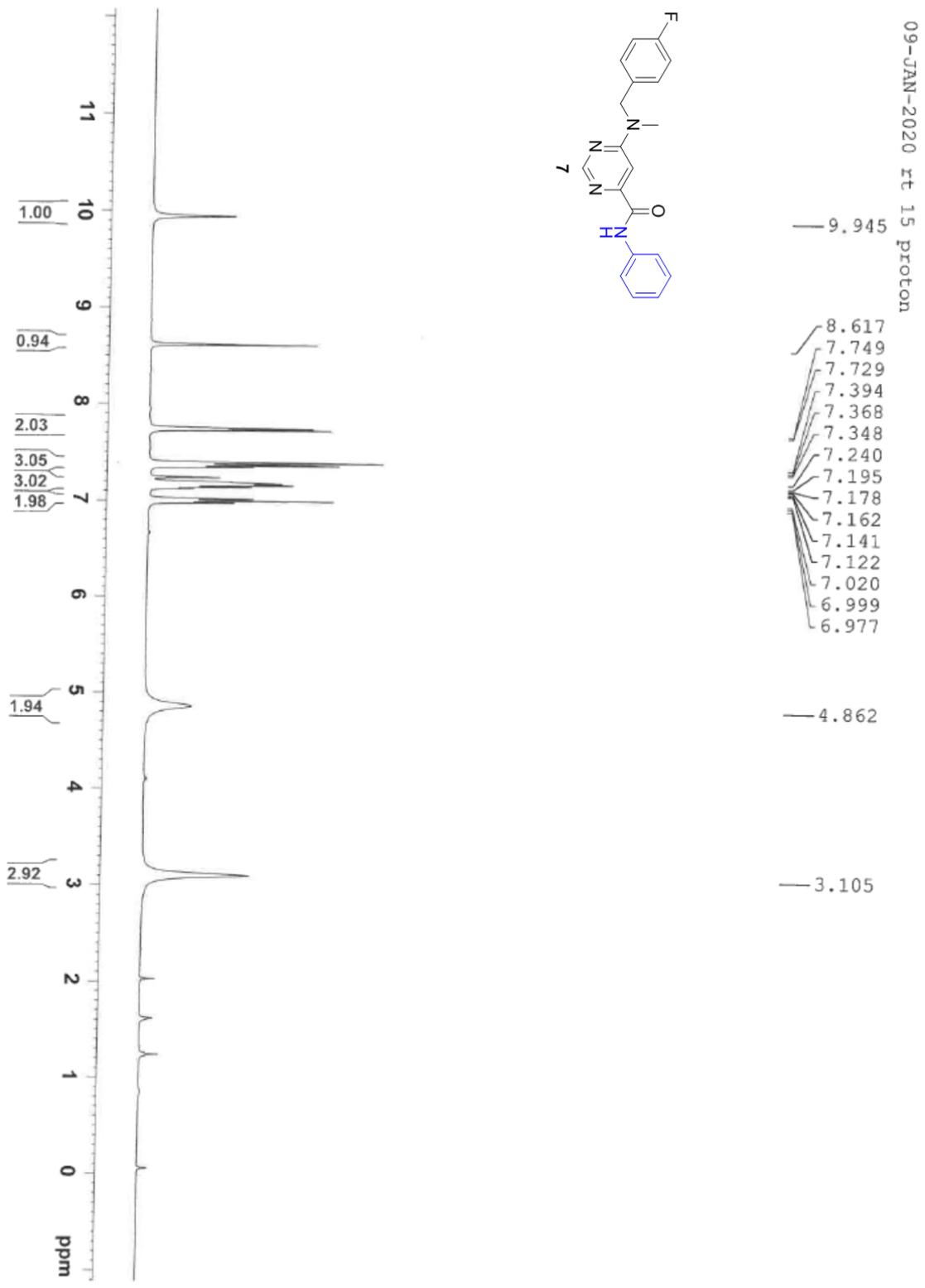
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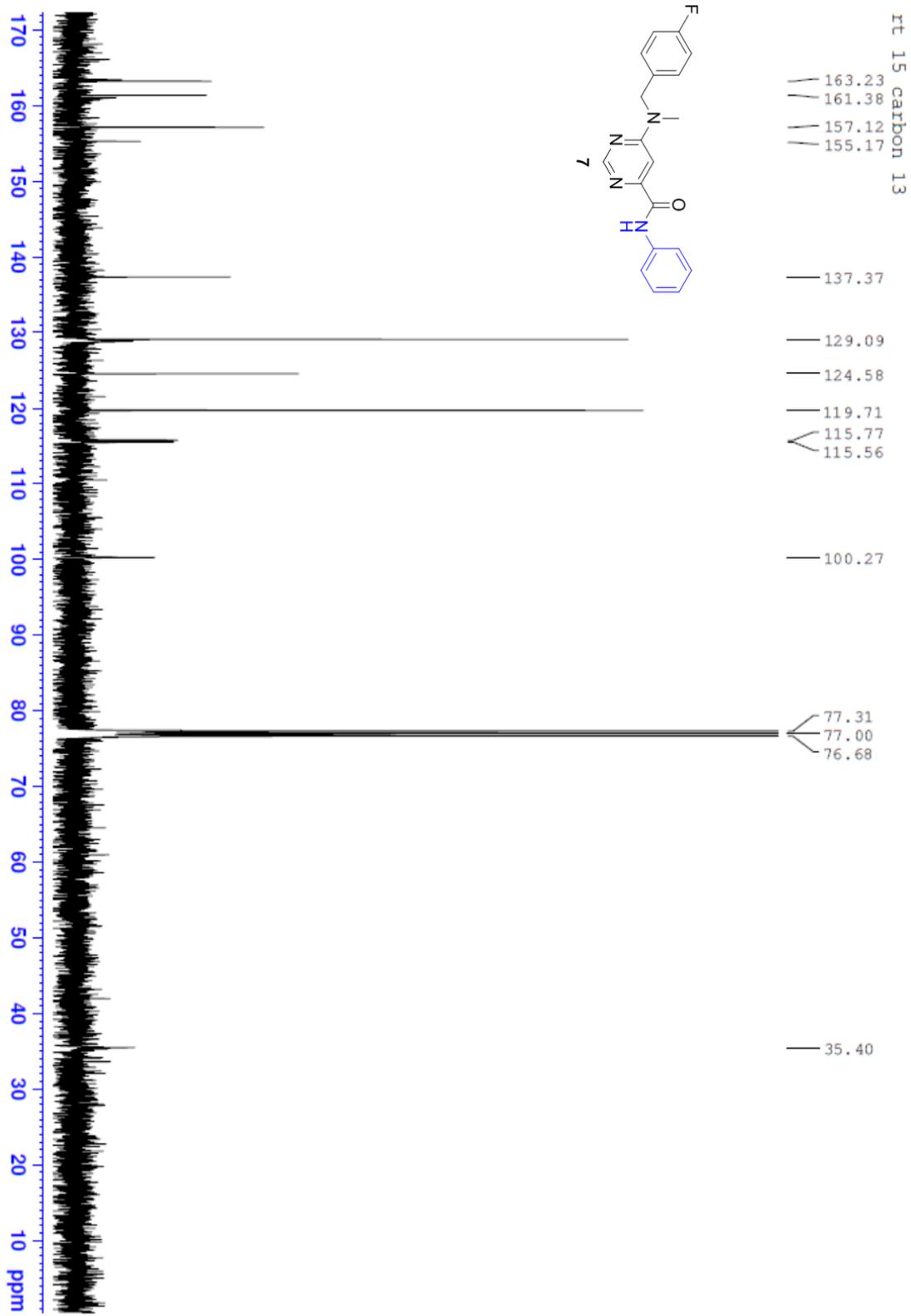


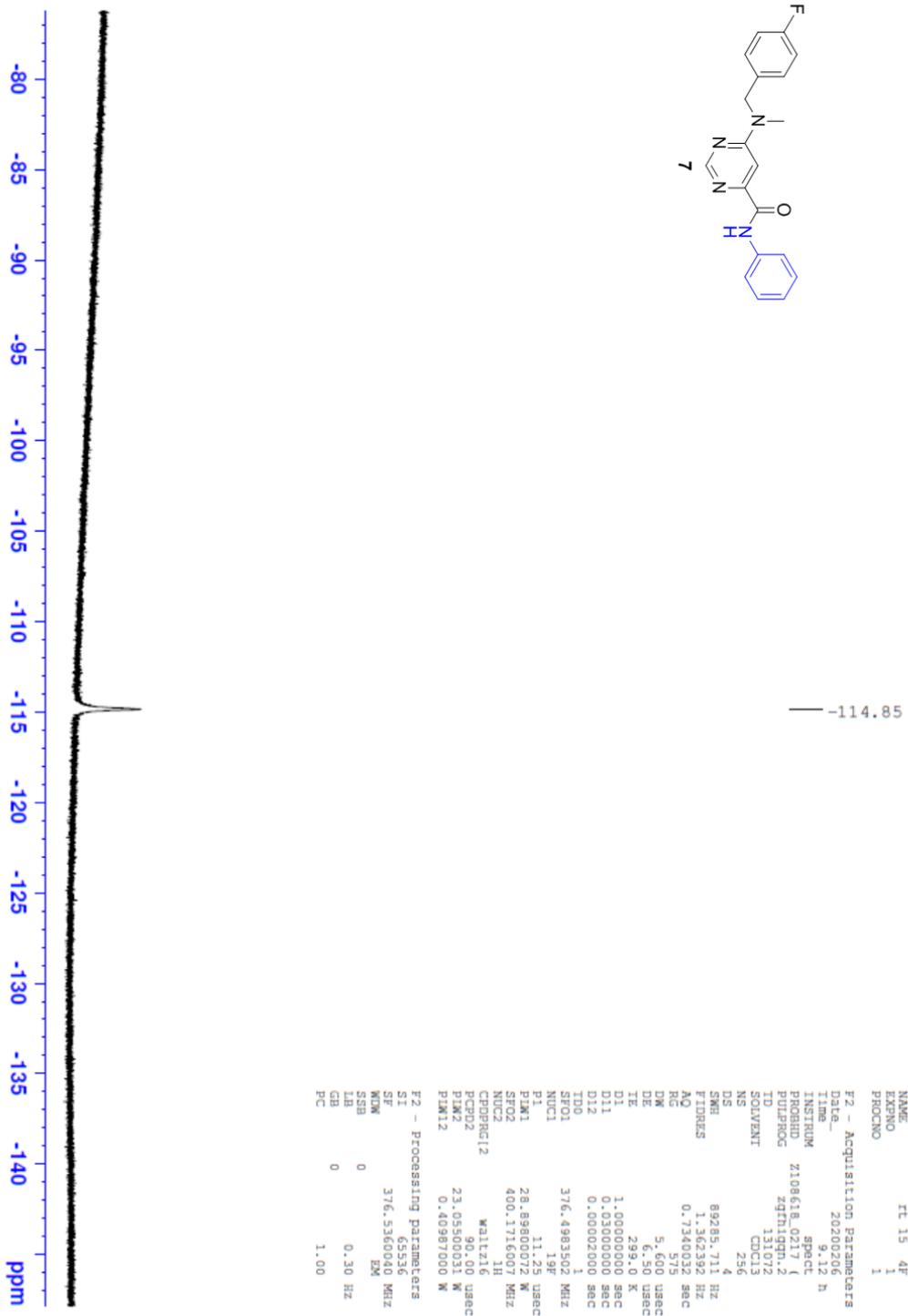
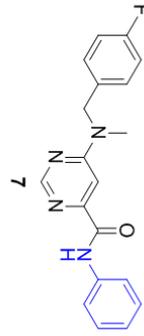


A100\_B









Current Data Parameters  
 NAME FT 15 4f  
 EXPNO 1  
 PROCNO 1

F2 - Acquisition Parameters  
 Date\_ 20200206  
 Time 9.12 h  
 INSTRUM spect  
 PROBHD Z108618.0217 (zqf19qn.2  
 PULPROG zgpg30  
 TD 131072  
 ID C012  
 SOLVENT CDCl3  
 NS 256  
 DS 4  
 SFO1 89285.711 Hz  
 FIDRES 1.362392 Hz  
 AQ 0.7340032 sec  
 RG 575

DE 5.600 usec  
 TE 299.0 K  
 D1 1.00000000 sec  
 D11 0.03000000 sec  
 D12 0.00020000 sec  
 TD 1  
 SFO1 376.4983502 MHz  
 NUCL1 13C  
 P1 1.198 usec  
 F1 28.8980122 MHz  
 SFO2 400.1716007 MHz  
 NUCL2 1H

CPDPRG12 waltz16  
 PCPD2 90.00 usec  
 P1M2 23.05500031 W  
 P1M12 0.40987000 W

F2 - Processing parameters  
 SI 65536  
 SF 376.5360040 MHz  
 WDM 0  
 SSB 0  
 LB 0.30 Hz  
 GB 0  
 PC 1.00