

Figure S1.
distribution diagram of protonation forms of tartaric acid $C_{\text{Tar}} = 1 \times 10^{-3} \text{ mol dm}^{-3}$.

Distribution

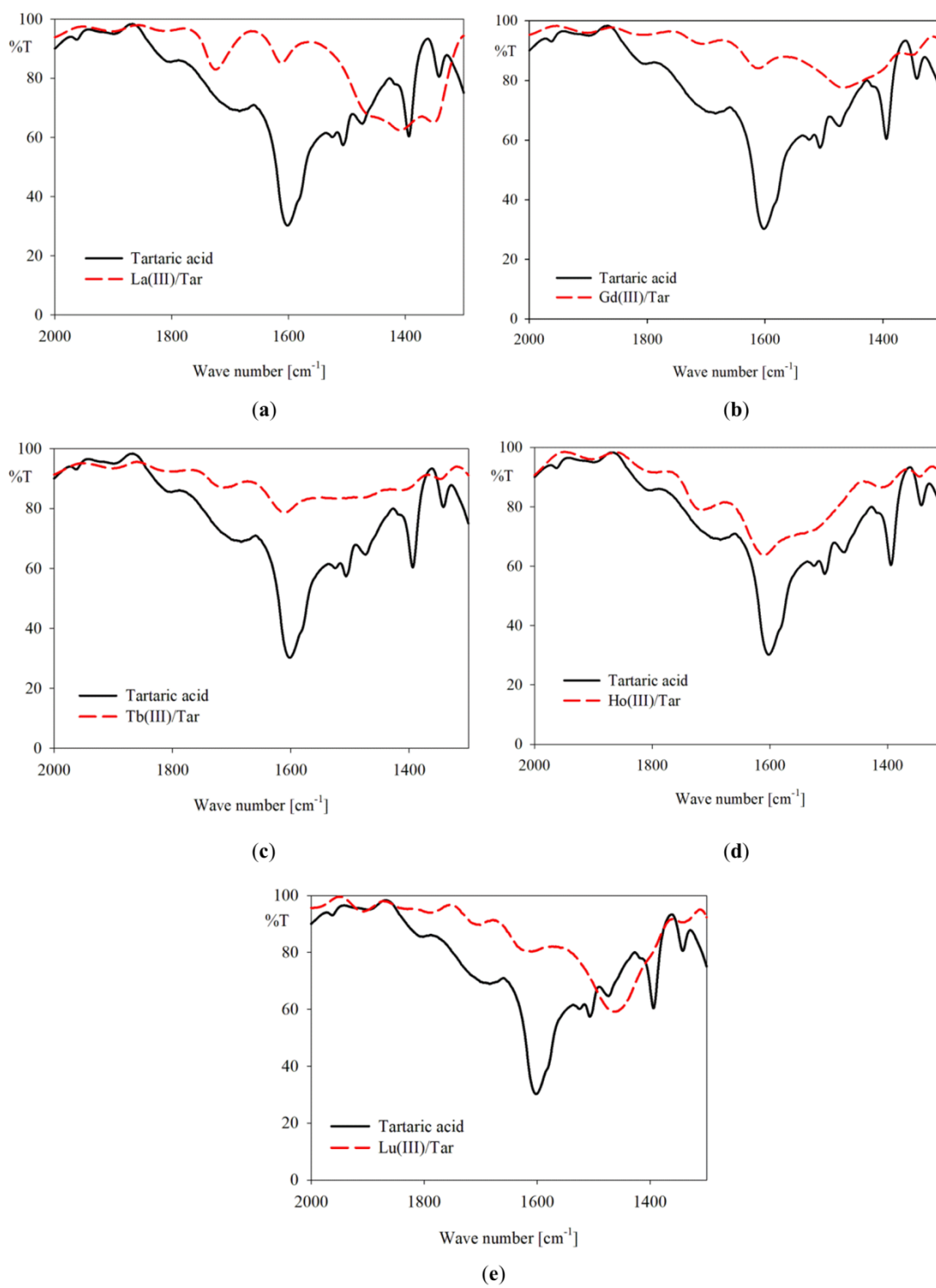


Figure S2. IR spectra of studied systems: (a) La(III)/Tar; (b) Gd(III)/Tar; (c) Tb(III)/Tar; (d) Ho(III)/Tar; (e) Lu(III)/Tar.

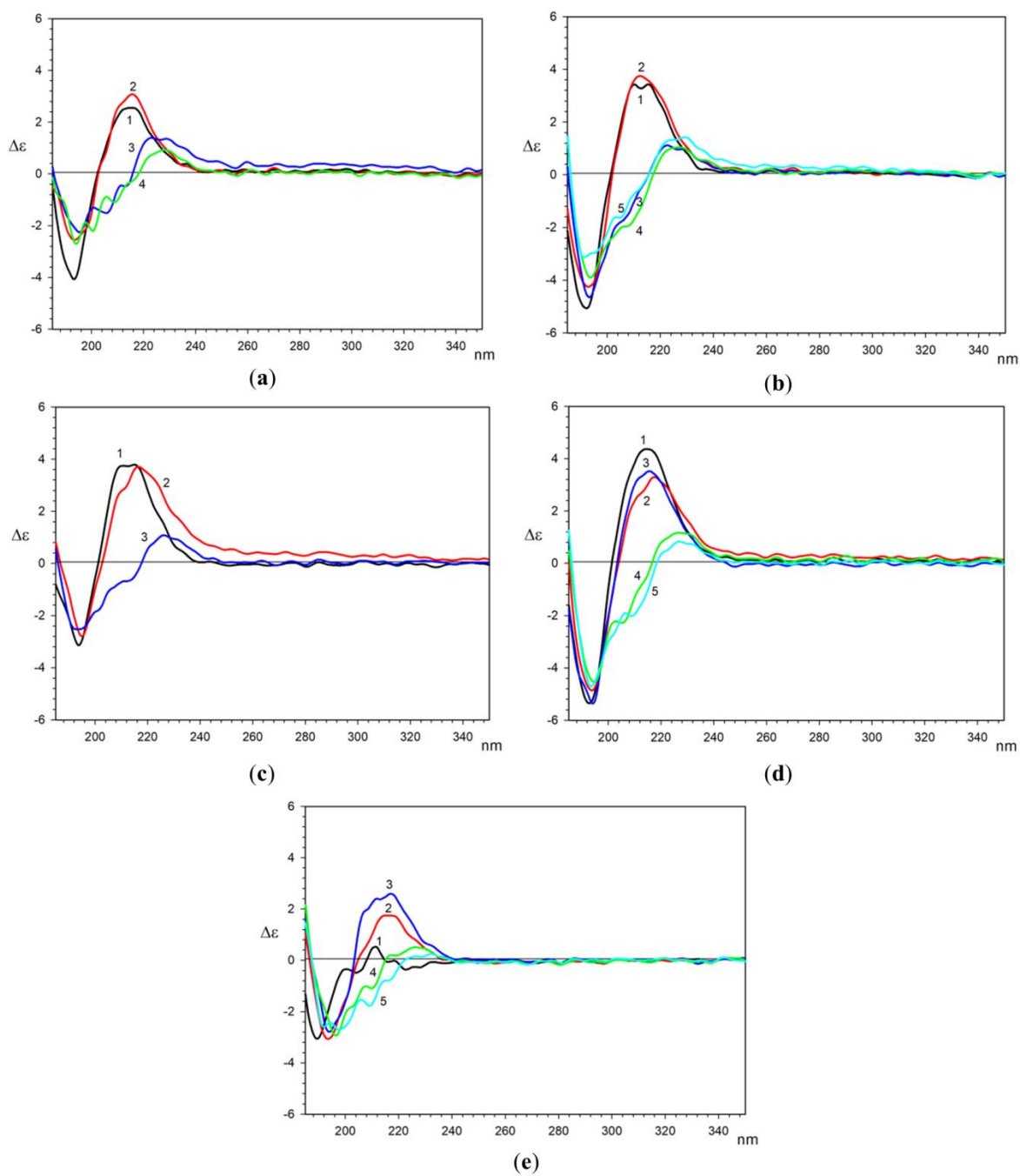


Figure S3. The CD spectra of the systems: (a) Nd/Tar: 1-pH=4.20, 2-pH=5.55, 3-pH=7.00, 4-pH=11.00; (b) Eu/Tar: 1-pH=3.80, 2-pH=5.10, 3-pH=6.80, 4-pH=10.00, 5-pH=11.00; (c) Gd/Tar: 1-pH=5.00, 2-pH=6.80, 3-pH=11.00; (d) Tb/Tar: 1-pH=3.80, 2-pH=5.00, 3-pH=6.60, 4-pH=9.90, 5-pH=11.00; (e) Lu/Tar: 1-pH=3.80, 2-pH=4.50, 3-pH=5.80, 4-pH=9.70, 5-pH=11.00.