

Article

Riminophenazine Derivatives as Potential Antituberculosis Agents: Synthesis, Biological, and Electrochemical Evaluations

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Abstract: A series of novel riminophenazine derivatives, having ionizable alkyl substituents at N-5 and a variety of substituents on the C-3 imino nitrogen, at C-8 and on the pendant aryl group, have been designed and synthesized. Preliminary investigations into the relationship between lipophilicity, redox potential, and antimycobacterial activity were conducted, using the in vitro activity against *Mycobacterium tuberculosis* H₃₇Rv, mammalian cytotoxicity, and the redox potential of the compounds determined by cyclic voltammetry as measures. Results revealed an activity “cliff” associated with C-8 substitution (**10l** and **10m**) that, along with defined redox activity, point to a new class of riminophenazines as potential anti-tuberculosis agents having reasonable activity (MIC₉₉ ~1 μM).

Keywords: riminophenazines; alkyl substituent; drug discovery; *Mycobacterium tuberculosis*



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1. Introduction

Tuberculosis (TB), caused by *Mycobacterium tuberculosis* in man, continues to be a serious threat to public health [1]. Despite the extensive control measures in place, the disease burden still remains high globally, mainly as a result of multidrug-resistant strains and co-infection with HIV [2,3]. Additionally, approximately one-third of the world's population have latent TB [4]. The disease occurs globally, with 22 countries bearing 80% of the total number of cases of active TB. This distribution occurs largely in the developing world, with sub-Saharan Africa being the worst affected [5,6]. An interesting class of tricyclic heterocycles, called riminophenazines, has shown promising activity against TB, leprosy, and cancer [7–12]. These compounds have a phenazine ring substituted on one of the ring nitrogens forming the central phenazine ring, as well as on one or both of the adjacent fused benzene rings. The designation “rimino” was coined to indicate the ‘R’ substituent on the imino moiety typical of this bioactive class of compounds [8]. This specific substituent is important in defining this bioactivity. A typical riminophenazine is clofazimine (Figure 1), which was first discovered by Barry et al. for the treatment of TB [13]. They proposed an intracellular redox cycling mechanism as the basis for the activity, in effect functioning as redox ‘traps’ that unproductively deplete cells of FADH and/or NAD(P)H. Clofazimine has also proved particularly active against *M. leprae*, which is the causative agent of leprosy [12,14,15]. Although the material (clofazimine) displayed high in vitro activity against TB, animal studies using guinea pig and monkey models failed to show in vivo activity [16]. Poor oral absorption of the drug has been implicated in certain studies, while species-specific activity may also be part of the problem based on considerable in vivo activity observed in hamsters and mice [17]. In vitro activity against the *M. avium complex* (MAC), an opportunistic pathogen in patients with HIV/AIDS, has

also been demonstrated [18–22]. A major hindrance to the use of riminophenazines in medicine is the observed staining of the dermis and retina on prolonged use, which was likely as the result of deposition of these lipophilic materials in fatty tissues [11].

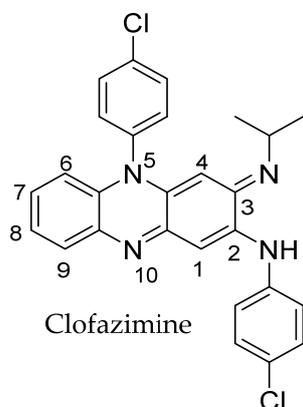


Figure 1. Structure of clofazimine.

Previous SAR investigations using riminophenazines focused on varying the alkylimino group at C-3 and the phenyl group at N-5 as handles to improve activity and pharmacokinetic behavior [23–26]. Liu et al. proved that the phenazine nucleus is key to the activity of the materials [24]. In addition, while a phenyl substituent at N-5 is typical of bioactive riminophenazines, cycloalkyl groups are also well-tolerated [27]. The current state of SAR knowledge with respect to the riminophenazine compounds is summarized in Figure 2 [28].

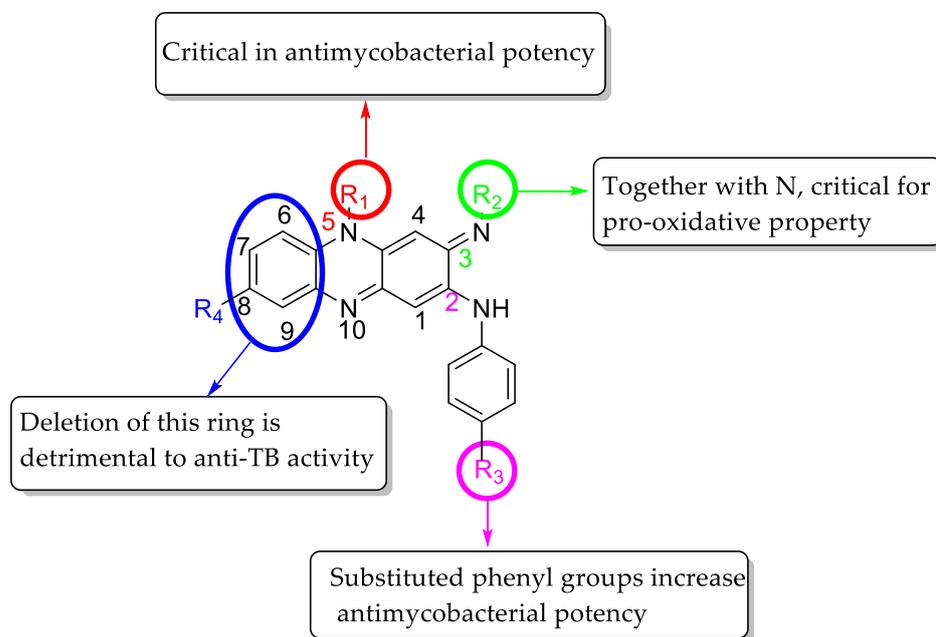


Figure 2. Structure–activity relationships of the riminophenazine system. IUPAC numbering is included for reference. Substituents targeted in this investigation are labeled R₁–R₄.

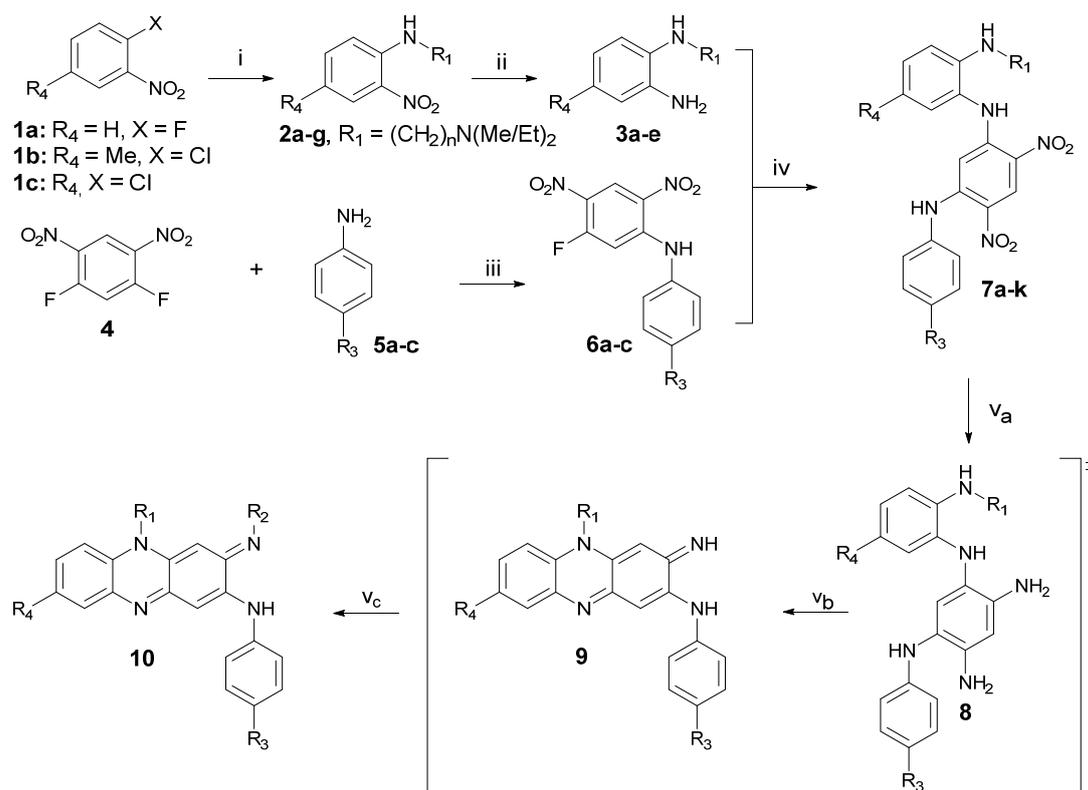
With these facts and the foregoing SAR in mind, an investigation was undertaken to determine the impact of including an ionizable group at R₁ on the riminophenazine skeleton on antimycobacterial activity. This may provide a means of solubilizing the material, in so doing diminishing accumulation in the fatty tissues of the skin. We particularly focused on using a limited number of short tertiary aminoalkyl chains for this purpose. In an attempt to optimize and/or modulate the impact of the amino moiety on function, a selection

of substituents at R_2 – R_4 , having distinct physicochemical properties, were included for examination. The activity of these new compounds was assessed against the redox potential of the materials and their inherent lipophilicity.

2. Results and Discussion

2.1. Chemistry

Several tertiary amines, in the form of *N,N*-diethyl- or *N,N*-dimethyl substituted ethylene- or propylenediamines, as well as *N,N*-dimethylhydrazine were chosen as candidates to generate the desired ionizable group at R_1 . Scheme 1 shows the synthetic sequence followed.



Scheme 1. (i) $R_1 = \text{Me}_2\text{N}(\text{CH}_2)_2$: IPA or neat, 90–100 °C, 16–24 h, 74%; $R_1 = \text{Me}_2\text{N}(\text{CH}_2)_3$ - or $\text{Et}_2\text{N}(\text{CH}_2)_3\text{Et}_2\text{NC}_2\text{H}_4$:- K_2CO_3 , DMF, 90 °C, 24 h, 52–72%; (ii) 5% Pd/C, H_2NNH_2 , 80 °C, 2–3 h; or 5% Pd/C, H_2 , rt, 4 h, 43–90%; (iii) 1,4-dioxane, Et_3N , 70–80 °C, 4 h, 71–83%; (iv) DIPEA, dry EtOH, reflux, 16 h, 50–84%; (v) (a) Zn, AcOH, 50 °C, 1 h; or 5% Pd/C, H_2 , MeOH:THF, rt, 12 h; then (b) 5% Pd/C, H_2 , THF:MeOH, 12 h; then medical oxygen, MeOH, rt, 12 h; subdivide then (c) R_2NH_2 , AcOH, dioxane, 110 °C, 48 h, 6–37%.

Three starting materials containing the ortho-halonitrobenzene structure **1** were commercially available, namely the parent compound, 1-fluoro-2-nitrobenzene, 1,4-dichloro-2-nitrobenzene, and 4-chloro-3-nitrotoluene. It was envisaged that the extra methyl or chloro group's impact on lipophilicity might have a bearing on activity while also modulating the redox potential. *Ips*o-substitution of the halo groups ortho to the nitro moiety in these materials using each of the three aliphatic amines chosen proceeded smoothly to yield substituted 2-nitroanilines **2** [29]. Attempts to introduce *N,N*-dimethylhydrazine were not successful. Catalytic hydrogenation of the nitro function in **2** gave substituted 1,2-phenylenediamines **3** in reasonable yield [25], with the exception of the chloro analogues, for which facile protodehalogenation was noted under these conditions. This series was subsequently abandoned from the sequence. Concurrently, compounds **6** were synthesized by *ip*so-substitution of one of the fluorine groups of 1,5-difluoro-2,4-dinitrobenzene **4**. Three anilines **5** featuring a variety of para-substituents with varying electronic properties were

used for this purpose, affording desired anilines **6**. Key intermediates **7** were obtained in modest yields by similar substitutions of each of the remaining fluoro groups in anilines **6** by the primary amino groups of phenylenediamines **3**. Reduction of the nitro groups in **7** under either acidic or neutral conditions, followed by immediate oxidative cyclization without prior isolation of tetramines **8**, afforded the desired iminophenazines **9** [25,30]. These latter two sets of compounds were not isolated because of difficulties in purification and questionable stability. Transimination of the unsubstituted imino moiety in compound **9** with commercially available isobutylamine, isopentylamine, or cyclohexylamine, was followed by column chromatography and preparative TLC to afford the target riminophenazines **10** in low yields [25,30].

2.2. Antimycobacterial Assessment

The compounds obtained were screened in triplicate against *M. tuberculosis* H₃₇Rv using the MGIT 960 system at 1, 3, and 5 μ M concentrations, with compounds showing MIC₉₉ \approx 1 μ M further tested in triplicate in the range 0.0625–1 μ M to refine the values determined. Cytotoxicity was determined using WI-38 cells (Human Fetal Lung Fibroblast) as a model of rapidly dividing normal mammalian cells. Although 26 compounds were synthesized, selected compounds were investigated, mainly due to the low quantities obtained synthetically. The data are summarized in Table 1.

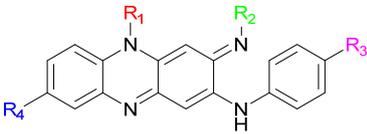
Lipophilicity (LogP) is an important property for riminophenazine derivatives, as the skin discoloration side effect of clofazimine is closely related to its high lipophilicity. As shown in Table 1, the LogP values for the most new compounds are lower than that of clofazimine (7.50), ranging from 3.43 to 5.02. Modifications of the phenazine ring at position N-5 with alkyl diamines provided compounds having lower LogP values compared to clofazimine. Generally, replacement of the phenyl group at the N-5 position by an alkyl group lowered LogP values, depending on the structure of the alkyl groups. Compounds with a dimethylaminoethyl substituent at the N-5 position displayed significantly reduced lipophilicity as compared to the corresponding dimethylaminopropyl/diethylaminopropyl substituted compounds, as exemplified by compound **10o** (LogP 3.43) versus compound **10b** (LogP 4.21), compound **10g** (LogP 3.64) versus compound **10i** (LogP 3.74), and compound **10c** (LogP 4.31) versus compound **10f** (LogP 5.02). These results are consistent with previous observation made by Yi-li et al., who also observed that improved antibacterial activity is dependent on increasing lipophilicity, a major factor for skin pigmentation [27]. In addition, the benzyl substituent at the C-2 position with hydrogen, methyl, and ethoxy at the para position showed significantly reduced LogP values, with lipophilicity increasing as R₃ was changes from H to OEt to Me. This was exemplified by compounds **10o**, **10n**, and **10g** with LogP values of (H: 3.43 < OEt: 3.64 < Me: 3.91). The same trend was observed for compounds **10a** (LogP 3.75), **10h** (LogP 3.96) and **10c** (LogP 4.31). **10o** could serve as an interesting lead for identifying new analogues with further decreased lipophilicity.

In terms of cytotoxicity, there are no general trends or variances with chain length or ring substitution. The R₂ = cHex series in each case appears to be the least toxic set of compounds (compare **10d**, **10e** with **10c**, **10f**, or **10i**, **10j**, **10k** with **10g**, **10h**). Compounds in the R₃ = OEt, R₄ = H series tend to be among the least toxic members of each contiguous series. Further substitution of the system (compare **10g** and **10l**, **10f** and **10m**) increases the toxicity.

With regard to antimycobacterial action, most of the activity appears to be as a result of toxicity rather than selective bactericidal action. The exceptions are **10l**, **10m**, and **10p** with both **10l** and **10m** having R₃, R₄ = Me as substituents, along with Me₂N(CH₂)₂- at N-5. Similarly, **10p** has a methyl substituent at R₃ which might also contribute to its activity. Compounds **10l**, **10m**, and **10p** exhibited potent activity comparable to that of the control drug, isoniazid, and also the well-known riminophenazine drug, clofazimine, all with MIC values < 1 μ M. In each case, an activity cliff (MIC₉₉ \approx 1 μ M) is encountered relative to all the other compounds, particularly conjoiners **10d** and **10e** (relative to **10l**) and **10c** and **10f** (relative to **10m**). The selectivity index of **10l**, **10m**, and **10p** (defined as cytotoxicity

IC_{50}/MIC_{99}), $SI \geq 3$ is modest at best, but it does indicate that interesting activity is still possible using the ionizable groups included in the design. Most of the rest of the materials show an $SI < 1.5$ by comparison.

Table 1. Structure–activity assessment of riminophenazines 10.



The chemical structure shows a riminophenazine core with four substituent positions: R₁ (red), R₂ (green), R₃ (pink), and R₄ (blue).

Compound	R ₁	R ₂	R ₃	R ₄	LogP ¹	IC ₅₀ (μM) ²	MIC ₉₉ (μM) ³
* Clofazimine	4-Clphenyl	iPr	Cl	H	7.50	68.62	<1
10b	Et ₂ N(CH ₂) ₃ -	iBu	H	H	4.21	4.44	3
10o	Me ₂ N(CH ₂) ₂ -	iBu	H	H	3.43	2.45	3
10d	Me ₂ N(CH ₂) ₃ -	iBu	Me	H	4.02	2.47	3
10e	Et ₂ N(CH ₂) ₃ -	iBu	Me	H	4.70	1.63	5
10l	Me ₂ N(CH ₂) ₂ -	iBu	Me	Me	4.73	2.86	<1
10n	Me ₂ N(CH ₂) ₂ -	iBu	Me	H	3.91	3.89	3
10g	Me ₂ N(CH ₂) ₂ -	iBu	OEt	H	3.64	12.30	5
10i	Me ₂ N(CH ₂) ₃ -	iBu	OEt	H	3.74	4.81	5
10j	Et ₂ N(CH ₂) ₃ -	iBu	OEt	H	4.42	5.35	3
10k	Et ₂ N(CH ₂) ₃ -	iPent	OEt	H	4.77	3.81	5
10a	Me ₂ N(CH ₂) ₂ -	cHex	H	H	3.75	9.01	3
10c	Me ₂ N(CH ₂) ₂ -	cHex	Me	H	4.31	5.99	5
10p	Me ₂ N(CH ₂) ₃ -	cHex	Me	H	4.34	5.08	<1
10f	Et ₂ N(CH ₂) ₃ -	cHex	Me	H	5.02	4.37	3
10m	Me ₂ N(CH ₂) ₂ -	cHex	Me	Me	4.41	3.64	<1
10h	Me ₂ N(CH ₂) ₂ -	cHex	OEt	H	3.96	11.10	5
INH							<1

¹ LogP calculated using ChemDraw version 12.0; ² Cytotoxicity against rapidly dividing human fibroblasts (WI38 cell line); ³ MIC₉₉: lowest concentration of the material inhibiting 99% of *M. tuberculosis* H₃₇Rv growth in culture. * Data taken from literature [26,27], INH = Isoniazid.

2.3. Cyclic Voltammetry Studies

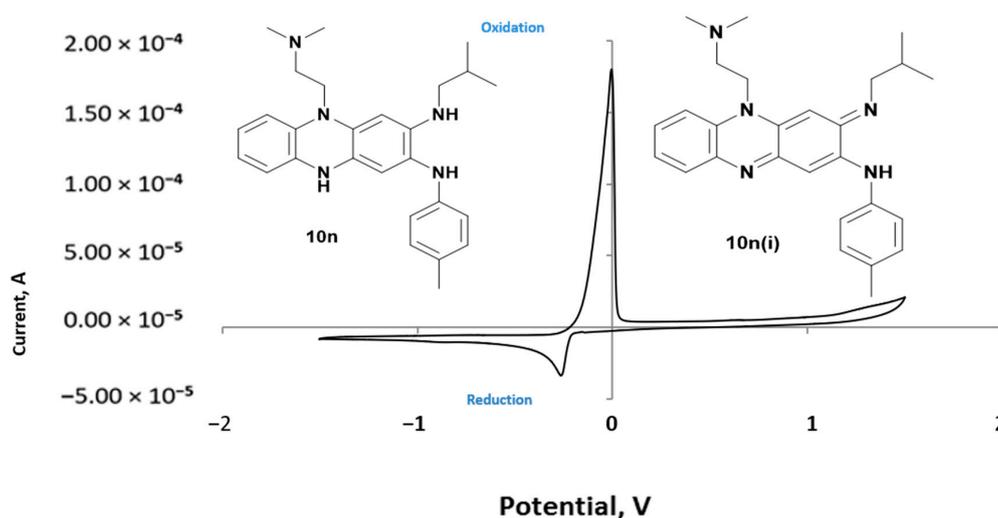
Cyclic voltammetry was used to determine the redox potential of the compounds, as a measure of the electronic impact the various substituents introduced on the riminophenazine skeleton may have. The results are summarized in Table 2. The cyclic voltammograms of the compounds each displayed a fully reversible one-electron transfer process. For example, the cyclic voltammogram of **10n** (Figure 3) showed that the redox cycle was fully reversible: increasingly negative (reducing) applied potential through a minimum anodic current produced reduced species on the left (**10n**), while an increasingly positive (oxidative) potential produces a maximum cathodic current to afford the fully oxidized species [**10n(i)**] to the right. All the compounds had negative half-wave potentials, indicating that these riminophenazines are easily oxidized—as observed in the laboratory [31].

It is interesting to note that the half-wave potential of the compounds proved to be a modest indicator of activity. All the compounds having MIC₉₉ < 3 μM showed $E_{1/2}$ values between −0.20 V and −0.13 V. Compounds (**10b**, **10c**, **10e**, **10f**, **10j**) with $E_{1/2} > -0.22$ V were generally inactive. This may indicate, in conjunction with the structural requirements for the observed activity cliff, that there is both an electronic/redox potential aspect, as well as some form of specific steric/binding determinant operative in this series of compounds that may be further exploited to improve the observed activity.

Table 2. Redox potentials (V vs. Ag/Ag⁺) of selected riminophenazines.

Compound	E_{pa}/V	E_{pc}/V	$E_{1/2}/V$	ΔE
10b	-0.0392	-0.41	-0.22	0.37
10e	-0.215	-0.971	-0.59	0.76
10i	-0.0218	-0.28	-0.15	0.25
10n	-0.0021	-0.26	-0.13	0.26
10j	-0.077	-0.38	-0.23	0.30
10c	-0.0055	-0.49	-0.24	0.35
10f	-0.0606	-0.41	-0.24	0.35
10m	-0.0305	-0.37	-0.20	0.34

Measured in CH₃CN containing 0.1 M [n-Bu₄N][ClO₄] at a scan rate of 100 mV s⁻¹ and referenced to Ag/Ag⁺. E_{pa} = anodic potential peak. E_{pc} = cathodic potential peak. $E_{1/2}$ = Half wave potential [$E_{1/2} = (E_{pa} + E_{pc})/2$]. $\Delta E = E_{pa} - E_{pc}$.

**Figure 3.** Cyclic voltammogram (CV) of **10n**.

3. Materials and Methods

The reactions were monitored for their completion by thin layer chromatography (TLC) using aluminium-backed Macherey-Nagel ALUGRAM Sil G/UV254 plates pre-coated with 0.25 mm silica gel 60Å. Column chromatography was carried out on silica gel 60 (particle size 200–300 mesh) with a silica to compound ratio of 30:1 by mass or neutral alumina as the adsorbent for conventional preparative chromatography.

NMR spectra were recorded on a Varian INOVA 400 MHz system or a Varian PremiumShield VNMRs 600 MHz system operating at 399.94 or 599.74 MHz, respectively. Chemical shifts (δ) are reported in parts per million (ppm) relative to tetramethylsilane (TMS) or against the residual protonated solvent signal as reference. Coupling constants (J) are reported in Hz.

Melting points were obtained on a Reichert Hot Stage or a Mettler FP62 melting point apparatus. High-resolution mass spectra were recorded on a Waters Acquity UPLC coupled in tandem to a Waters photodiode array (PDA) detector and a SYNAPT G1 HDMS mass spectrometer.

3.1. Synthesis Procedures

General Procedure for Final Compounds **10a–p**

The chosen dinitro compound **7** (3 mmol) and 5% palladium on carbon (3 mmol) in 2:3 (*v/v*) methanol:THF was stirred at room temperature under 275 kPa hydrogen atmosphere for 12 h in a steel pressure vessel. The orange solution turned maroon after the completion.

The catalyst was filtered off through a celite pad, and the filtrate was concentrated under reduced pressure. The gummy maroon residue was dissolved in methanol and stirred at room temperature under pure oxygen (99.5%) atmosphere (345 kPa) for 12 h. The mixture was concentrated under reduced pressure to give a dark red/black product. The crude residue was purified on silica using 5–15% methanol:chloroform as eluent and used immediately without characterization. The material obtained was divided among a choice of amines (isobutylamine, cyclohexylamine, or isopentylamine) (25 eq.) in dioxane (3–4 mL) and glacial acetic acid (0.2 eq.) and heated at 110 °C in a sealed or closed tube for 48 h. After cooling to room temperature, the material was concentration in vacuo, and the residue obtained was purified by column chromatography on silica gel [2–5% (*v/v*) methanol:chloroform as eluent]. Further purification using preparative TLC was required using the same eluent to ensure adequate purity of the isolated materials.

(*E*)-5-[2-(Dimethylamino)ethyl]-3-cyclohexylimino-2-phenylamino-3,5-dihydrophenazine (**10a**). Dark red solid (20.4 mg, 20.4%); R_f 0.46 [10% (*v/v*) methanol: chloroform]; mp. 94–96 °C (ethanol). ^1H NMR (400 MHz, CD_3OD): δ 7.33 (dd, $J = 7.93, 1.03$ Hz, 1H, H 9), 7.27 (dd, $J = 8.51, 7.78$ Hz, 2H, H 3'), 7.21–7.17 (m, H 2'), 7.15 (d, $J = 7.70$ Hz, 1H, H 6), 7.11 (d, $J = 8.48$ Hz, 1H, H 7), 6.99 (dt, $J = 13.1, 7.3$ Hz, 2H, H 8), 6.53 (s, 1H, H 1), 5.74 (s, 1H, H 4), 3.91 (br s, 2H, NCH_2CH_2), 3.48–3.36 (m, 1H, =NCH), 2.46–2.35 (m, 2H, CH_2NMe_2), 2.26 (s, 6H, NMe_2), 1.93–1.75, 1.75–1.63, 1.49–1.34, and 1.33–1.24 [$4 \times$ m, 10H, $\text{NCH}(\text{C}_5\text{H}_{10})$]; ^{13}C NMR (101 MHz, CD_3OD): δ 152.4 (C 3), 151.7 (C-10a), 145.1 (C 2), 141.4 (C 1'), 136.7 (C 9a), 134.7 (C-4a), 131.5 (C 5a), 130.6 (C 3'), 129.7 (C 4'), 128.9 (C 9), 124.7 (C 7), 124.2 (C 8), 122.4 (C 2'), 114.0 (C 6), 99.9 (C 1), 89.0 (C 4), 59.3 (=NCH), 54.8 (CH_2NMe_2), 46.1 (NMe_2), 44.8 (NCH_2), 35.0 [$\text{CH}(\text{CH}_2)_2$], 27.2 [$\text{CH}_2(\text{CH}_2)_2$], 26.3 [$\text{CH}_2(\text{CH}_2)_2$]; HRMS (ESI-TOF+): m/z calculated for $\text{C}_{28}\text{H}_{34}\text{N}_5$: 440.2814; found: 440.2810 (MH+). IR (cm^{-1}): 3209 (N-H), 2923 (Ar-C-H), 1508 (C=N), 1463 (Ar-C=C), 1241 (C-N).

(*E*)-5-[3-(Diethylamino)propyl]-3-isobutylimino-2-phenylamino-3,5-dihydrophenazine (**10b**). Dark red solid (12.5 mg, 37%); R_f 0.42 [20% (*v/v*) methanol: chloroform]; mp. 102–104 °C (ethanol). ^1H NMR (400 MHz, CD_3OD): δ 8.19 (br d, $J = 8.79$ Hz, 1H, H 6), 8.03 (dd, $J = 8.35, 1.42$ Hz, 1H, H 9), 7.89 (ddd, $J = 8.76, 7.10, 1.51$ Hz, 1H, H 7), 7.70 (ddd, $J = 8.18, 7.10, 0.93$ Hz, 1H, H 8), 7.50–7.45 (m, 2H, H 3'), 7.41–7.37 (m, 2H, H 2'), 7.29 (s, 1H, H 1), 7.23 (ddt, $J = 7.30, 7.23, 1.17$ Hz, 1H, H 4'), 6.77 (s, 1H, H 4), 4.93 (br t, $J = 7.62$ Hz, 2H, NCH_2CH_2), 3.54 (d, $J = 7.13$ Hz, 2H, NCH_2CH), 2.88 (t, $J = 7.08$ Hz, 2H, CH_2NEt_2), 2.73 [q, $J = 7.16$ Hz, 4H, $\text{N}(\text{CH}_2\text{CH}_3)_2$], 2.29 (dquin, $J = 13.49, 6.78$ Hz, 1H, CHMe_2), 2.17 (dt, $J = 15.11, 7.43$ Hz, 2H, $-\text{CH}_2-$), 1.16 (d, $J = 6.64$ Hz, 6H, CHMe_2) and 1.15 [t, $J = 7.13$ Hz, 6H, $\text{N}(\text{CH}_2\text{CH}_3)_2$]; ^{13}C NMR (101 MHz, CD_3OD): 154.4 (C 3), 147.2 (C 10a), 142.6 (C 2), 141.3 (C 1'), 140.4 (C 9a), 135.5 (C 4a), 133.7 (C 7), 131.5 (C 9), 131.0 (C 3'), 130.7 (C 5a), 128.5 (C 8), 126.2 (C 4'), 123.7 (C 2'), 117.2 (C 6), 107.7 (C 1), 90.2 (C 4), 53.9 (NCH_2CH), 50.6 (CH_2NEt_2), 48.3 [$\text{N}(\text{CH}_2\text{CH}_3)_2$], 47.6 (NCH_2CH_2), 29.2 (CHMe_2), 25.7 ($-\text{CH}_2-$), 21.2 (CHMe_2) and 11.6 [$\text{N}(\text{CH}_2\text{CH}_3)_2$]. HRMS (ESI-TOF+): m/z calculated for $\text{C}_{29}\text{H}_{38}\text{N}_5$: 456.3127; found: 456.3114 (MH+). IR (cm^{-1}): 3381 (N-H), 2980 (Ar-C-H), 1509 (C=N), 1461 (Ar-C=C), 1250 (C-N).

(*E*)-5-[2-(Dimethylamino)ethyl]-3-cyclohexylimino-2-(4-methylphenyl)amino-3,5-dihydrophenazine (**10c**). Maroon solid (38 mg, 12% yield); R_f 0.39 [10% (*v/v*) methanol: chloroform]; mp. 110–112 °C (ethanol). ^1H NMR (400 MHz, CD_3OD): δ 7.39 (dd, $J = 7.91, 1.37$ Hz, 1H, H 6), 7.19 (br d, $J = 7.70$ Hz, 1H, H 9), 7.24 (ddd, $J = 8.40, 7.03, 1.37$ Hz, 1H, H 7), 7.12 (d, $J = 8.49$ Hz, 2H, H 2'), 7.10 (d, $J = 8.89$ Hz, 2H, H 3'), 7.04 (ddd, $J = 8.01, 6.93, 1.07$ Hz, 1H, H 8), 6.53 (s, 1H, H 1), 5.82 (s, 1H, H 4), 4.00 (br t, $J = 7.30$ Hz, 2H, NCH_2CH_2), 3.50 (br td, 1H, =NCH), 2.50 (br t, $J = 7.81$ Hz, 2H, CH_2NMe_2), 2.34 (s, 6H, NMe_2), 2.29 (3H, s, C-4' Me), 1.93–1.75, 1.75–1.63, and 1.53–1.35 [$10\text{H}, 3 \times$ m, $\text{NCH}(\text{C}_5\text{H}_{10})$]; ^{13}C NMR (101 MHz, CD_3OD): δ 152.4 (C 3), 151.6 (C-10a), 145.4 (C 2), 138.6 (C 9a), 136.8 (C 1'), 134.6 (C 4'), 134.0 (C 4a), 131.3 (C 7), 131.1 (C 3'), 129.6 (C 9), 128.8 (C 5a), 124.2 (C 8), 122.6 (C 2'), 113.9 (C 6), 99.5 (C 1), 89.0 (C 4), 59.2 (=NCH), 54.8 (CH_2NMe_2), 46.1 (NMe_2), 44.8 (NCH_2), 35.0 [$\text{CH}(\text{CH}_2)_2$], 27.2 [$\text{CH}_2(\text{CH}_2)_2$], 26.3 (C-4' Me), 21.1 [$\text{CH}_2(\text{CH}_2)_2$]; HRMS (ESI-TOF+): m/z

calculated for $C_{29}H_{36}N_5$: 454.2971; found: 454.2956 (MH⁺); IR (cm⁻¹): 3292 (N-H), 2920 (Ar-C-H), 1517 (C=N) and 1465 (Ar-C=C), 1240 (C-N).

(*E*)-5-[3-(Dimethylamino)propyl]-3-isobutylimino-2-(4-methylphenyl)amino-3,5-dihydrophenazine (**10d**). Dark maroon crystals (49 mg; 13%); R_f 0.30 [20% (v/v) methanol:chloroform]; mp. > 230 °C (ethanol). ¹H NMR (600 MHz, CD₃OD): δ 8.25 (br d, J = 8.30 Hz, 1H, H 6), 8.23 (d, J = 7.90 Hz, 1H, H 9), 8.08 (t, J = 7.52 Hz, 1H, H 7), 7.90 (t, J = 7.50 Hz, 1H, H 8), 7.66 (d, J = 8.25 Hz, 2H, H 3'), 7.40 (d, J = 8.40 Hz, 2H, H 2'), 7.40 (s, 1H, H 1), 6.88 (s, 1H, H 4), 5.03 (br t, J = 7.63 Hz, 2H, NCH₂CH₂), 3.80 (d, J = 7.04 Hz, 2H, NCH₂CH), 2.93 (t, J = 6.90 Hz, 2H, CH₂NMe₂), 2.77 (s, 6H, NMe₂), 2.71 (s, C-4' 3H, Me), 2.58 (quin, J = 7.04 Hz, CHMe₂), 2.44 (dt, J = 15.26, 7.34 Hz, 2H, -CH₂-) and 1.52 (d, J = 6.75 Hz, 6H, CHMe₂); ¹³C NMR (151 MHz, CD₃OD): 154.4 (C 3), 147.0 (C 10a), 142.8 (C 2), 140.7 (C 9a), 138.4 (C 1'), 136.6 (C 4'), 135.6 (C 4a), 133.8 (C 7), 131.7 (C 9), 131.6 (C 3'), 130.6 (C 5a), 128.8 (C 8), 124.3 (C 2'), 117.4 (C 6), 107.4 (C 1), 90.3 (C 4), 57.1 (CH₂NMe₂), 53.4 (NCH₂CH), 47.5 (NMe₂), 45.6 (NCH₂CH₂), 29.1 (CHMe₂), 26.1 (-CH₂-), 21.2 (C-4' Me) and 21.1 (CHMe₂); HRMS (ESI-TOF⁺): m/z calculated for $C_{28}H_{36}N_5$: 442.2971; found: 442.2892 (MH⁺). IR (cm⁻¹): 3108 (N-H), 2954 (Ar-C-H), 1512 (C=N), 1459 (Ar-C=C), 1239 (C-N).

(*E*)-5-[3-(Diethylamino)propyl]-3-isobutylimino-2-(4-methylphenyl)amino-3,5-dihydrophenazine (**10e**). Dark red solid (22.4 mg, 5.8%); R_f 0.38 [10% (v/v) methanol:chloroform]; mp. 80–82 °C (ethanol). ¹H NMR (400 MHz, CD₃OD): δ 8.11 (br d, J = 8.89 Hz, 1H, H 6), 7.99 (br d, J = 8.30 Hz, 1H, H 9), 7.82 (br t, 1H, J = 7.80 Hz, H 7), 7.64 (br t, J = 7.80 Hz, 1H, H 8), 7.27 (q, J = 8.89 Hz, 4H, H 3', H 2'), 7.17 (s, 1H, H 1), 6.68 (s, 1H, H 4), 4.84 (br t, J = 8.30 Hz, 2H, NCH₂CH₂), 3.49 (d, J = 7.03 Hz, 2H, NCH₂CH), 2.76 (t, J = 6.93 Hz, 2H, CH₂NEt₂), 2.64 [q, J = 7.13 Hz, 4H, N(CH₂CH₃)₂], 2.38 (s, 3H, C-4' Me), 2.25 (dquin, J = 13.54, 6.75 Hz, 1H, CHMe₂), 2.15–2.06 (m, J = 8.50, 2H, -CH₂-), 1.13 [d, J = 7.18 Hz, 6H, N(CH₂CH₃)₂], and 1.10 (t, J = 6.64 Hz, 6H, CHMe₂); ¹³C NMR (101 MHz, CD₃OD): 154.2 (C 3), 147.9 (C 10a), 143.9 (C 2), 140.1 (C 9a), 138.5 (C 1'), 136.3 (C 4'), 135.3 (C 4a), 133.1 (C 7), 131.2 (C 9), 131.5 (C 3'), 130.7 (C 5a), 128.0 (C 8), 124.0 (C 2'), 116.9 (C 6), 107.0 (C 1), 90.4 (C 4), 54.0 (NCH₂CH), 51.0 (CH₂NEt₂), 48.3 [N(CH₂CH₃)₂], 47.5 (NCH₂CH₂), 29.5 (CHMe₂), 26.1 (-CH₂-), 21.4 (CHMe₂), 21.3 (C-4' Me) and 11.9 [N(CH₂CH₃)₂]; HRMS (ESI-TOF⁺): m/z calculated for $C_{30}H_{40}N_5$: 470.3284; found: 470.3291 (MH⁺); IR (cm⁻¹): 3367 (N-H), 2980 (Ar-C-H), 1517 (C=N), 1464 (Ar-C=C), 1250 (C-N).

(*E*)-5-[3-(Diethylamino)propyl]-3-cyclohexylimino-2-(4-methylphenyl)amino-3,5-dihydrophenazine (**10f**). Dark red solid (35.9 mg, 9.7%); R_f 0.36 [20% (v/v) methanol:chloroform]; mp. 108–110 °C (ethanol). ¹H NMR (400 MHz, CD₃OD): δ 8.19 (br d, J = 8.79 Hz, 1H, H 6), 8.04 (dd, J = 8.40, 1.37 Hz, 1H, H 9), 7.87 (ddd, J = 8.69, 7.13, 1.57 Hz, 1H, H 7), 7.70 (ddd, J = 8.30, 7.13, 0.88 Hz, 1H, H 8), 7.29 (d, J = 8.01 Hz, 2H, H 3'), 7.26 (d, J = 9.18 Hz, 2H, H 2'), 7.22 (s, 1H, H 1), 6.81 (s, 1H, H 4), 4.98–4.88 (m, 2H, NCH₂), 4.07 (br t, J = 7.81 Hz, 1H, =NCH), 2.81 (t, J = 7.13 Hz, 2H, CH₂NEt₂), 2.67 [q, J = 7.10 Hz, 4H, N(CH₂CH₃)₂], 2.38 (s, 3H, C-4' Me), 2.23–2.17 (m, 2H, -CH₂-), 1.82–1.18 [4 × m, 10H, NCH(C₅H₁₀)], and 1.10 [t, J = 7.23 Hz, 6H, N(CH₂CH₃)₂]; ¹³C NMR (101 MHz, CD₃OD): δ 152.8 (C 3), 147.3 (C-10a), 143.2 (C 2), 140.5 (C 9a), 138.4 (C 1'), 136.6 (C 4'), 135.5 (C 4a), 133.4 (C 7), 131.6 (C 3'), 131.5 (C 9), 130.6 (C 5a), 128.5 (C 8), 124.3 (C 2'), 117.2 (C 6), 106.8 (C 1), 90.1 (C 4), 55.3 (=NCH), 50.9 (CH₂NEt₂), 48.3 [N(CH₂CH₃)₂], 47.6 (NCH₂), 33.1 [CH(CH₂)₂], 26.7 (-CH₂-), 26.2 [CH₂(CH₂)₂], 21.1 (C-4' Me) and 11.8 [N(CH₂CH₃)₂]; HRMS (ESI-TOF⁺): m/z calculated for $C_{32}H_{42}N_5$: 496.3440; found: 496.3460 (MH⁺). IR (cm⁻¹): 3367 (N-H), 2980 (Ar-C-H), 1511 (C=N), 1462 (Ar-C=C), 1250 (C-N).

(*E*)-5-[2-(Dimethylamino)ethyl]-3-isobutylimino-2-(4-ethoxyphenyl)amino-3,5-dihydrophenazine (**10g**). Dark red solid (193 mg, 47%); R_f 0.32 [10% (v/v) methanol:chloroform]; mp. 208–210 °C (ethanol). ¹H NMR (400 MHz, CD₃OD): δ 8.09 (br d, J = 8.98 Hz, 1H, H 6), 8.04 (br d, J = 7.81 Hz, 1H, H 9), 7.83 (dd, J = 8.49, 6.93 Hz, 1H, H 7), 7.65 (t, J = 7.42 Hz, 1H, H 8), 7.29 (d, J = 8.79 Hz, H 2'), 7.09 (d, J = 8.89 Hz, 2H, H 3'), 7.03 (s, 1H, H 1), 6.87 (s, 1H, H 4), 4.90 (dd, J = 7.42, 6.54 Hz, 2H, NCH₂CH₂), 4.09 (q, J = 6.97 Hz, 2H, OCH₂CH₃), 3.52 (d, J = 7.0 Hz, 2H, NCH₂CH), 2.89 (t, J = 7.37 Hz, 2H, CH₂NMe₂), 2.46 (s, NMe₂), 2.26

(sep, $J = 6.63$ Hz, 1H, CHMe₂), 1.43 (t, $J = 6.98$ Hz, 3H, OCH₂CH₃), 1.14 (d, $J = 6.64$ Hz, 6H, CHMe₂); ¹³C NMR (101 MHz, CD₃OD): δ 158.5 (C 4'), 154.1 (C 3), 147.2 (C 10a), 143.5 (C 2), 140.4 (C 9a), 135.5 (C 4a), 133.3 (C 1'), 133.1 (C 7), 131.3 (C 9), 130.4 (C 5a), 128.5 (C 8), 126.4 (C 2'), 116.9 (C 3'), 116.8 (C 6), 105.7 (C 1), 90.3 (C 4), 64.9 (OCH₂CH₃), 56.6 (CH₂NMe₂), 53.5 (NCH₂CH), 47.9 (NCH₂CH₂), 46.1 (NMe₂), 28.9 (CHMe₂), 20.9 (CHMe₂) and 15.2 (OCH₂CH₃); HRMS (ESI-TOF+): m/z calculated for C₂₈H₃₆N₅O: 458.2920; found: 458.2891 (MH+); IR (cm⁻¹): 3128 (N-H), 2954 (Ar-C-H), 1507 (C=N), 1464 (Ar-C=C), 1230 (C-N).

(*E*)-5-[2-(Dimethylamino)ethyl]-3-cyclohexylimino-2-(4-ethoxyphenyl)amino-3,5-dihydrophenazine (10h). Dark red solid (45 mg, 14%); R_f 0.42 [10% (*v/v*) methanol:chloroform]; mp. 222–224 °C (ethanol). ¹H NMR (400 MHz, CD₃OD): δ 8.04–7.96 (m, 2H, H 6, H-9), 7.82 (dd, $J = 11.5$, 4.2 Hz, 1H, H 7), 7.65 (br t, $J = 7.45$ Hz, 1H, H 8), 7.30–7.25 (m, 2H, H 2'), 7.04 (s, 1H, H 1), 7.02 (d, $J = 8.80$ Hz, 2H, H 3'), 6.83 (s, 1H, H 4), 4.85 (observed, 2H, NCH₂), 4.08 (q, $J = 6.85$ Hz, 2H, OCH₂CH₃), 3.92 (m, 2H, =NCH), 2.85 (t, $J = 7.63$ Hz, 2H, CH₂NMe₂), 2.46 (s, 6H, NMe₂), 2.23–2.15, 1.96–1.85, 1.83–1.75 and 1.62–1.55 [4 × m, 10H, NCH(C₅H₁₀)], 1.40 (t, $J = 6.90$ Hz, 3H, OCH₂CH₃); ¹³C NMR (101 MHz, CD₃OD): δ 158.6 (C 4'), 152.7 (C 3), 147.7 (C 10a), 144.1 (C 2), 140.2 (C 9a), 135.6 (C 4a), 133.3 (C 7), 133.2 (C 1'), 131.1 (C 9), 130.6 (C 5a), 128.2 (C 8), 126.6 (C 2'), 117.0 (C 3'), 116.7 (C 6), 105.1 (C 1), 90.4 (C 4), 65.1 (OCH₂CH₃), 56.8 (=NCH), 55.8 (CH₂NMe₂), 47.7 (NCH₂), 46.3 (NMe₂), 33.3 [CH(CH₂)₂], 26.7 and 26.2 [CH₂(CH₂)₂] and 15.3 (OCH₂CH₃); HRMS (ESI-TOF+): m/z calculated for C₃₀H₃₈N₅O: 484.3076; found: 484.3048 (MH+).

(*E*)-5-[3-(Dimethylamino)propyl]-3-isobutylimino-2-(4-ethoxyphenyl)amino-3,5-dihydrophenazine (10i). Dark red solid (41.0 mg, 15%); R_f 0.40 [20% (*v/v*) methanol:chloroform]; mp. 185–187 °C (ethanol). ¹H NMR (400 MHz, CD₃OD): δ 8.21 (d, $J = 8.79$ Hz, 1H, H 6), 8.05 (d, $J = 8.40$ Hz, 1H, H 9), 7.84 (ddd, $J = 9.00$, 7.19, 1.27 Hz, 1H, H 7), 7.72 (t, $J = 7.70$ Hz, 1H, H 8), 7.30 (d, $J = 8.89$ Hz, 2H, H 2'), 7.11 (s, 1H, H 1), 7.03 (d, $J = 8.98$ Hz, 2H, H 3'), 6.87 (s, 1H, H 4), 4.98 (t, $J = 7.80$ Hz, 2H, NCH₂CH₂), 4.08 (q, $J = 6.93$ Hz, 2H, OCH₂CH₃), 3.55 (d, $J = 7.03$ Hz, 2H, NCH₂CH), 2.61 (t, $J = 6.83$ Hz, 2H, CH₂NMe₂), 2.34 (s, 6H, NMe₂), 2.27 (sep, $J = 6.77$ Hz, 1H, CHMe₂), 2.18 (br t, $J = 6.74$ Hz, 2H, -CH₂-), 1.41 (t, $J = 6.98$ Hz, 3H, OCH₂CH₃), 1.15 (d, $J = 6.64$ Hz, 6H, CHMe₂); ¹³C NMR (101 MHz, CD₃OD): δ 158.7 (C 4'), 154.1 (C 3), 147.2 (C 10a), 143.7 (C 2), 140.6 (C 9a), 135.5 (C 4a), 133.5 (C 1'), 133.3 (C 7), 131.5 (C 9), 130.5 (C 5a), 128.7 (C 8), 126.6 (C 3'), 117.3 (C 6), 117.0 (C 2'), 106.0 (C 1), 90.3 (C 4), 65.1 (OCH₂CH₃), 57.3 (CH₂NMe₂), 53.6 (NCH₂CH), 47.6 (NCH₂CH₂), 46.0 (NMe₂), 29.1 (CHMe₂), 26.4 (-CH₂-), 21.1 (CHMe₂), and 15.3 (OCH₂CH₃); HRMS (ESI-TOF+): m/z calculated for C₂₉H₃₈N₅O: 472.3076; found: 472.3048 (MH+); IR (cm⁻¹): 3085 (N-H), 2955 (Ar-C-H), 1502 (C=N), 1462 (Ar-C=C), 1232 (C-N).

(*E*)-5-[3-(Diethylamino)propyl]-3-isobutylimino-2-(4-ethoxyphenyl)amino-3,5-dihydrophenazine (10j). Dark purple solid (22.1 mg, 6.2%); R_f 0.37 [20% (*v/v*) methanol:chloroform]; mp. 205–207 °C (ethanol). ¹H NMR (400 MHz, CD₃OD): δ 7.98 (d, $J = 8.79$ Hz, 1H, H 6), 7.84 (dd, $J = 8.24$, 1.28 Hz, 1H, H 9), 7.69 (ddd, $J = 8.63$, 7.17, 1.40 Hz, 1H, H 7), 7.52 (ddd, $J = 8.20$, 6.95, 0.70 Hz, 1H, H 8), 7.16 (d, $J = 9.03$ Hz, 2H, H 2'), 6.91 (s, 1H, H 1), 6.88 (d, $J = 8.91$ Hz, 2H, H 3'), 6.55 (s, 1H, H 4), 4.71 (t, $J = 7.80$ Hz, 2H, NCH₂CH₂), 3.99 (q, $J = 7.00$ Hz, 2H, OCH₂CH₃), 3.39 (d, $J = 7.08$ Hz, 2H, NCH₂CH), 2.66 (t, $J = 6.96$ Hz, 2H, CH₂NEt₂), 2.55 [q, $J = 7.12$ Hz, 4H, N(CH₂CH₃)₂], 2.14 (sep, $J = 6.76$ Hz, 1H, CHMe₂), 2.03–1.94 (m, 2H, -CH₂-), 1.29 (t, $J = 7.02$ Hz, 3H, OCH₂CH₃), 1.01 (d, $J = 6.59$ Hz, 6H, CHMe₂), and 0.97 [t, $J = 7.14$ Hz, 6H, N(CH₂CH₃)₂]; ¹³C NMR (101 MHz, CD₃OD): δ 158.5 (C 4'), 154.0 (C 3), 148.0 (C 10a), 144.2 (C 2), 139.9 (C 9a), 135.5 (C 4a), 133.4 (C 1'), 132.7 (C 7), 130.9 (C 9), 130.6 (C 5a), 127.8 (C 8), 126.3 (C 3'), 116.9 (C 2'), 116.8 (C 6), 104.6 (C 1), 89.9 (C 4), 65.1 (OCH₂CH₃), 54.7 (CH₂NEt₂), 50.9 (NCH₂CH), 48.3 [N(CH₂CH₃)₂], 47.4 (NCH₂CH₂), 29.5 (CHMe₂), 25.8 (-CH₂-), 21.2 (CHMe₂), 15.3 (OCH₂CH₃) and 11.9 [N(CH₂CH₃)₂]; HRMS (ESI-TOF+): m/z calculated for C₃₁H₄₂N₅O: 500.3389; found: 500.3391 (MH+); IR (cm⁻¹): 3365 (N-H), 2980 (Ar-C-H), 1509 (C=N), 1462 (Ar-C=C), 1233 (C-N).

(*E*)-5-[3-(Diethylamino)propyl]-3-isopentylimino-2-(4-ethoxyphenyl)amino-3,5-dihydrophenazine (10k). Dark purple solid (19.1 mg, 6.2%) after crystallization; R_f 0.39 [20% (*v/v*) methanol:

chloroform]; mp. 198–200 °C (ethanol). ¹H NMR (400 MHz, CD₃OD): δ 8.17 (d, *J* = 8.79 Hz, 1H, H 6), 8.02 (dd, *J* = 8.30, 1.37 Hz, 1H, H 9), 7.85 (ddd, *J* = 8.74, 7.13, 1.42 Hz, 1H, H 7), 7.69 (ddd, *J* = 8.30, 6.90, 0.70 Hz, 1H, H 8), 7.29 (d, *J* = 8.89 Hz, 2H, H 2'), 7.05 (d, *J* = 8.89 Hz, 2H, H 3'), 7.02 (s, 1H, H 1), 6.74 (s, 1H, H 4), 4.91 (dd, *J* = 8.40, 7.40 Hz, 2H, NCH₂CH₂), 4.07 (q, *J* = 7.03 Hz, 2H, OCH₂CH₃), 3.34–3.29 (m, 2H, =NCH₂), 2.79 (t, *J* = 6.93 Hz, 2H, CH₂NEt₂), 2.66 [q, *J* = 7.13 Hz, 4H, N(CH₂CH₃)₂], 1.95–1.81 (m, 3H, CH₂CHMe₂), 2.15 (quin, *J* = 7.13 Hz, 2H, -CH₂-), 1.42 (t, *J* = 6.98 Hz, 3H, OCH₂CH₃), 1.11 (d, *J* = 7.2 Hz, 6H, CHMe₂) and 1.08 [d, *J* = 6.3 Hz, 6H, N(CH₂CH₃)₂]; ¹³C NMR (101 MHz, CD₃OD): δ 158.6 (C 4'), 153.8 (C 3), 147.5 (C 10a), 143.8 (C 2), 140.4 (C 9a), 135.3 (C 4a), 133.3 (C 1'), 133.1 (C 7), 131.3 (C 9), 130.5 (C 5a), 128.4 (C 8), 126.4 (C 3'), 117.1 (C 6), 116.9 (C 2'), 105.3 (C 1), 89.9 (C 4), 65.1 (OCH₂CH₃), 50.9 (CH₂NEt₂), 48.3 [N(CH₂CH₃)₂], 47.4 (NCH₂CH₂), 44.9 (=NCH₂), 38.3 (CH₂CH), 27.6 (CHMe₂), 25.9 (-CH₂-), 23.1 (CHMe₂), 15.3 (OCH₂CH₃) and 11.8 [N(CH₂CH₃)₂]; HRMS (ESI-TOF+): *m/z* calculated for C₃₂H₄₄N₅O: 514.3546; found: 514.3546 (MH+); IR (cm⁻¹): 3381 (N-H), 2980 (Ar-C-H), 1507 (C=N), 1462 (Ar-C=C), 1253 (C-N).

(*E*)-5-[2-(Dimethylamino)ethyl]-3-isobutylimino-8-methyl-2-(4-methylphenyl)amino-3,5-dihydrophenazine (**10l**). Dark red solid (23.0 mg, 11%); *R*_f 0.37 [10% (*v/v*) methanol: chloroform]; mp. 148–150 °C (ethanol). ¹H NMR (600 MHz, CDCl₃ + CD₃OD): δ 7.12 (br s, 1H, H 6), 7.08–7.04 (m, 3H, H 9, H-3'), 7.04–6.99 (m, 3H, H-7, H 2'), 6.46 (s, 1H, H 1), 5.68 (s, 1H, H 4), 3.96 (t, *J* = 7.19 Hz, 2H, NCH₂CH₂), 3.07 (d, *J* = 6.90 Hz, 2H, NCH₂CH), 2.43 (t, *J* = 8.22 Hz, 2H, CH₂NMe₂), 2.25 (s, 6H, NMe₂), 2.22 (s, 3H, C-8 Me), 2.18 (s, 3H, C-4' Me), 1.92 (dquin, *J* = 13.41, 6.74 Hz, 1H, CHMe₂), and 0.94 (d, *J* = 6.75 Hz, 6H, CHMe₂); ¹³C NMR (151 MHz, CD₃OD): δ 153.7 (C 3), 151.2 (C 10a), 144.8 (C 2), 138.7 (C 1'), 137.0 (C 4a), 134.6 (C 9a), 134.6 (C 5a), 134.0 (C 4'), 131.2 (C 8), 131.1 (C 3'), 129.2 (C 9), 128.7 (C 7), 122.5 (C 2'), 114.1 (C 6), 99.9 (C 1), 88.5 (C 4), 58.9 (NCH₂CH), 54.6 (CH₂NMe₂), 46.0 (NMe₂), 45.1 (NCH₂), 31.1 (CHMe₂), 21.6 (CHMe₂), 21.1 (C-4' Me) and 20.9 (C-8 Me); HRMS (ESI-TOF+): *m/z* calculated for C₂₈H₃₆N₅: 442.2971; found: 442.2831 (MH+); IR (cm⁻¹): 3221 (N-H), 2959 (Ar-C-H), 1518 (C=N) and 1464 (Ar-C=C), 1241 (C-N).

(*E*)-3-(Cyclohexylimino)-5-[2-(dimethylamino)ethyl]-8-methyl-*N*-*p*-tolyl-3,5-dihydrophenazin-2-amine (**10m**). Maroon solid (26.0 mg, 21%); *R*_f 0.50 [10% (*v/v*) methanol:chloroform]; mp. 155–157 °C (ethanol). ¹H NMR (400 MHz, CD₃OD): δ 7.44 (br d, *J* = 8.80 Hz, 1H, H 6), 7.38 (br s, 1H, H 9), 7.30 (br dd, *J* = 8.80, 1.60 Hz, 3H, H 7), 7.14 (d, *J* = 8.01 Hz, 2H, H 3'), 7.08 (d, *J* = 8.60 Hz, 2H, H 2'), 6.73 (s, H 1), 1H, 6.22 (s, 1H, H 4), 4.37 (br t, *J* = 7.03 Hz, 2H, NCH₂), 3.68–3.57 (m, 1H, =NCH), 2.60 (t, *J* = 7.60 Hz, 2H, CH₂NMe₂), 2.31 (s, 6H, NMe₂), 2.29 (s, 3H, C-8 Me), 2.26 (s, 3H, C-4' Me), 2.05–1.91, 1.85–1.73, 1.71–1.58 and 1.50–1.31 [4 × m, 10H, NCH(C₅H₁₀)]; ¹³C NMR (101 MHz, CD₃OD): δ 152.3 (C 3), 148.8 (C 10a), 143.7 (C 2), 138.7 (C 1'), 136.9 (C 4a), 135.5 (C 9a), 134.4 (C 5a), 133.3 (C 4'), 131.3 (C 8), 131.2 (C 3'), 129.4 (C 9), 128.8 (C 7), 123.4 (C 2'), 115.3 (C 6), 103.0 (br, C 1), 89.5 (C 4), 57.1 (=NCH), 56.0 (CH₂NMe₂), 46.6 (NCH₂), 46.1 (NMe₂), 33.9 [CH(CH₂)₂], 26.8 and 26.2 [CH₂(CH₂)₂], 21.0 (C-4' Me) and 20.98 (C-8 Me); HRMS (ESI-TOF+): *m/z* calculated for C₃₀H₃₉N₅: 469.3205; found: 469.3041 (MH+). IR (cm⁻¹): 3231 (N-H), 2922 (Ar-C-H), 1512 (C=N) and 1460 (Ar-C=C), 1242 (C-N).

(*E*)-5-[2-(Dimethylamino)ethyl]-3-(isobutylimino)-*N*-*p*-tolyl-3,5-dihydrophenazin-2-amine (**10n**). Dark red solid (98 mg, 17% yield); *R*_f 0.50 [20% (*v/v*) methanol: chloroform]; mp. 140–142 °C (ethanol). ¹H NMR (600 MHz, CD₃OD₃): δ 8.14 (d, *J* = 8.8 Hz, 1 H, H-6), 8.09 (d, *J* = 8.3 Hz, 1 H, H-7), 7.93 (t, *J* = 7.9 Hz, 1 H, H-9), 7.74 (t, *J* = 7.6 Hz, 1-H, H-8), 7.47 (d, *J* = 7.9 Hz, 2 H, H-3'), 7.38 (d, *J* = 7.7 Hz, 2 H, 2'), 7.37 (s, 1 H, H-1), 6.93 (s, 1 H, H-4), 5.05 (t, *J* = 7.3 Hz, 2 H, NCH₂), 3.58 (s, 3 H, Ph-CH₃) 3.56 (d, *J* = 7.2 Hz, 2 H, NCH₂CH), 2.98–2.92 (m, 2 H, CH₂NEt₂), 2.46 (s, 2.28, 6H, CH₂NMe₂), (t, *J* = 13.6, 6.7 Hz, 1 H, CHMe₂), 1.14 (d, *J* = 6.6 Hz, 6 H, CHMe₂). ¹³C NMR (151 MHz, CD₃OD₃): δ 153.2 (C 3), 149.9 (C-10a), 143.8 (C 2), 137.4 (C 1'), 137.2 (C-4a), 133.7 (C-9a), 133.4 (C-5a), 130.1 (C 3), 129.8 (C 7), 129.4 (C 9), 128.8 (C 8), 124.0 (C-4'), 122.6 (C 2'), 112.9 (C 6), 100.2 (C 1), 87.6 (C 4), 56.8 (CH₂NMe₂), 54.6 (NCH₂CH), 46.1 [N(CH₃)₂], 45.1 (C 1'), 29.6 (CHMe₂), 21.2 (CHMe₂), 21.1 (PhCH₃); HRMS

(ESI-TOF+): m/z calculated for $C_{27}H_{34}N_5$: 428.2814; found: 428.2823 (M+H)⁺. IR (cm^{-1}): 3301 (N-H), 2975 (Ar-C-H), 1510 (C=N) and 1465 (Ar-C=C), 1238 (C-N).

(*E*)-5-[2-(Dimethylamino)ethyl]-3-(isobutylimino)-*N*-phenyl-3,5-dihydrophenazin-2-amine (**10o**). Dark maroon tacky compound (6.40 mg, 25%); R_f 0.39 [10% (*v/v*) methanol: chloroform]; mp. 119–121 °C (ethanol). ¹H NMR (400 MHz, CDCl₃): δ 7.72 (d, $J = 7.80$ Hz, 1 H, H-6), 7.40–7.32 (m, 6 H, H-7, H-9, H-2', H-3'), 7.07 (t, $J = 7.22$ Hz, 1 H, H 8), 6.92 (s, 1 H, H-1), 6.06 (s, 1 H, H-4), 4.29 (br s, 2 H, H 1'), 3.32 (d, $J = 7.24$ Hz, 2 H, NCH₂CH), 2.72 (t, $J = 7.80$ Hz, 2-H, CH₂NMe₂), 2.43 [s, 6-H, N(CH₃)₂], 2.15 (br s, 1-H, CHMe₂), 1.07 (d, $J = 6.63$ Hz, 6-H N(CH₂CH₃)₂). ¹³C NMR (101 MHz, CDCl₃): δ 153.1 (C 3), 150.3 (C 10a), 143.5 (C 2), 140.1 (C 1'), 137.0 (C 4a), 133.4 (C 9a), 131.1 (C 5a), 130.1 (C 7), 129.5 (C 9), 129.1 (C 3'), 129.0 (C 8), 122.2 (C 2', C 4'), 112.8 (C 6), 100.2 (C 1), 87.6 (C 4), 57.6 (CH₂NEt₂), 54.5 (C 2), 46.1 [N(CH₃)₂], 45.0 (C 1'), 29.9 (CHMe₂), 20.9 (CHMe₂). HRMS (ESI-TOF+): m/z calculated for $C_{27}H_{33}N_5$: 428.2382; found: 428.2378 (M+H)⁺. IR (cm^{-1}): 3289 (N-H), 2950 (Ar-C-H), 1513 (C=N), 1464 (Ar-C=C), 1239 (C-N).

(*E*)-3-(Cyclohexylimino)-5-[(3-dimethylamino)propyl]-*N*-*p*-tolyl-3,5-dihydrophenazin-2-amine (**10p**). A red solid (97.4 mg, 26%); R_f 0.43 [20% (*v/v*) methanol:chloroform]; mp. 212–215 °C (ethanol). ¹H NMR (400 MHz, CD₃OD): δ 7.65 (d, $J = 7.50$ Hz, 2 H, H 6, H 9), 7.50 (t, $J = 8.30$ Hz, 1 H, H 7), 7.32 (t, $J = 7.60$ Hz, 1 H, H 8), 7.22 (q, $J = 8.44$ Hz, 4 H, H 2'', H 3''), 6.81 (s, 1 H, H 1), 6.24 (s, 1 H, H 4), 4.45–4.26 (m, 2 H, H 1'), 3.83–3.63 (m, 1 H, H 1*), 2.52 (t, $J = 7.04$ Hz, 2 H, H 3'), 2.35 (s, 3 H, PhCH₃), 2.31 [s, 6 H, N(CH₃)₂], 2.07–1.94 (m, 4 H, H 2', H 2*), 1.90 (br s, 2 H, H 2*), 1.77 (br d, $J = 12.7$ Hz, 2 H, H 4*), 1.61–1.45 (m, 4 H, H 3*). ¹³C NMR (151 MHz, CD₃OD): δ 152.6 (C 3), 150.3 (C 10a), 145.0 (C 2), 138.6 (C 1''), 138.0 (C 4a), 135.4 (C 9a), 134.7 (C 5a), 131.3 (C 3''), 131.2 (C 7), 130.8 (C 4''), 129.6 (C 9), 125.6 (C 8), 123.4 (C 2''), 115.3 (C 6), 101.7 (C 1), 89.3 (C 4), 65.6 (C 1*), 57.7 (C 3'), 46.0 [N(CH₃)₂], 45.7 (C 1'), 34.4 (C 2*), 27.0 (C 4*), 26.3 (C 2'), 25.2 (C 3*), 21.1 (PhCH₃). HRMS (ESI-TOF+): m/z calculated for $C_{30}H_{38}N_5$: 468.3726; found: 468.3102 (M+H)⁺. IR (cm^{-1}): 3253 (N-H), 2922 (Ar-C-H), 1514 (C=N), 1459 (Ar-C=C), 1210 (C-N).

3.2. Method for Cyclic Voltammetry

Cyclic voltammetry was carried out using a BAS-100W electrochemical analyzer using a one-compartment three-electrode system composed of an Ag/Ag⁺ reference electrode (0.01 M AgNO₃), a platinum wire as the auxiliary electrode, and a platinum disc as the working electrode. The supporting electrolyte solution was 0.1 M aqueous tetrabutylammonium perchlorate in dried acetonitrile (100 mL). All measurements were carried out at 1, 3, or 5 μM of the riminophenazines, at a scan rate of 100 mV.s⁻¹, unless otherwise stated. All analyses were run in duplicate under an atmosphere of nitrogen at room temperature. The solutions were deoxygenated by purging under a stream of nitrogen through the solution for 4 min prior each run. The platinum disc electrode was polished after every run.

3.3. Antitubercular Testing (MIC₉₉ Determination) Method

A working stock solution of 10 μM of the test compound in fresh DMSO was prepared and stored at -70 °C until used. All compounds were tested at 5, 3, and 1 μM point concentrations for the initial screens. Mycobacterium tuberculosis H₃₇Rv inocula were prepared from cultures grown on Lowenstein Jensen (LJ) slants. Mycobacterial suspensions were prepared in saline, and the turbidity was adjusted to 0.5 McFarland units. Aliquots of 100 μL were inoculated into MGIT tubes and incubated at 37 °C until the inoculum became positive (about three days). Then, 0.5 mL of positive MGIT cultures were transferred into MGIT tubes containing the test compounds. Controls consisting of a 1:100 dilution of the positive MGIT culture control in saline, 1.2% DMSO control in saline (concentration equivalent to that used when introducing the drug substance), and a 0.05 μg/mL isoniazid (INH) positive control in saline were also prepared. For mycobacterial growth evaluation, the MGIT 960 system (Becton Dickinson, Sparks, MD, USA) was used, where *M. tuberculosis* growth is observed through fluorescent changes due to oxygen consumption during

mycobacterial growth. Incubation at 37 °C was continued in the MGIT system, and the growth units (GU) were recorded hourly. For MIC₉₉ evaluations, the 1% bacterial control culture was used; the MIC₉₉ of the compound was determined relative to the growth units of the control (GU₄₀₀). When the GU of the control reached 400, the results were interpreted. If the drug-containing tube showed GU > 400, the material was defined as inducing resistance; if it showed GU ≤ 400, susceptibility of the mycobacteria to the drug substance was indicated at that concentration of drug substance. Compounds that showed GU = 0 at 1 μM were considered to have an MIC ≤ 1 μM, and further MIC determinations on these materials using concentrations ranging from 0.0625–1 μM were performed. All experiments were conducted in triplicate, with three independent experiments performed each time and the averages are reported. Statistically, the percentage range of standard deviation from mean showed the two concentrations (3 and 5 μM) having a close range as compared to the lowest concentration (see Supplementary data).

3.4. Protocol for Toxicity Evaluation (IC₅₀) Method

The WI-38 cell line—normal Human Fetal Lung Fibroblast from ECACC was routinely maintained as a monolayer cell culture at 37 °C, 5% CO₂, 95% air, and 100% relative humidity in EMEM containing 10% fetal bovine serum, 2 mM L-glutamine, and 50 μg/mL gentamicin. For screening experiment, the cells (21–50 passages) were inoculated in 96-well microtiter plates at plating densities of 10,000 cells/well and were incubated for 24 h. After 24 h, the cells were treated with the experimental drugs, which were previously dissolved in DMSO and diluted in medium to produce five concentrations. The process was performed in triplicate and the data presented are means of three repeats performed in triplicate. Cells without drug addition served as control. The blank contains complete medium without cells. Parthenolide was used as a standard. The plates were incubated for 48 h after addition of the compounds. Viable cells were fixed to the bottom of each well with cold 50% trichloroacetic acid, washed, dried, and dyed by SRB. Unbound dye was removed, and protein-bound dye was extracted with a 10 mM Tris base for optical density determination at the wavelength 540 nm using a multiwell spectrophotometer. Data analysis was performed using GraphPad Prism software. 50% of cell growth inhibition (IC₅₀) was determined by non-linear regression.

4. Conclusions

A series of variously substituted phenazines having ionizable aminoalkyl chains installed at N-5 of the riminophenazine system as part of the strategy to develop less lipophilic compounds with high anti-TB activity are reported. While it is known that lipophilicity is an important determinant of antimycobacterial activity as well as pigment deposition in fatty tissues [14], the inclusion of aminoalkyl sidechains has generated reasonably active materials with the potential to further enhance activity while still providing a “handle” for improved compounds. Remarkably, compounds with methyl groups at both R₃ and R₄ positions with Me₂N(CH₂)₂- at N-5 position (**10l** and **10m**) demonstrated both lower ClogP values and were also potent against *M. tuberculosis*. The data obtained give a preliminary indication that the more commonly used N-5 phenyl moieties may be substituted without abrogating activity. Future work will focus on refining this further, followed by pharmacokinetic testing.

Supplementary Materials: The following are available online, ¹H and ¹³C NMR spectra of the synthesized target compounds.

Author Contributions: M.V.B. synthesized all the target compounds, collected and analyzed the experimental data, drafted and finalized the manuscript. C.v.d.W. conceptualized and supervised the research, provided intellectual leadership, corrected, edited and reviewed the manuscript. E.M.M. co-designed the research. A.N. performed the biological assays. All authors have read and agreed to the published version of the manuscript.

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