

Recoverable palladium-catalyzed carbon-carbon bond forming reactions under thermomorphic mode: Stille and Suzuki-Miyaura reactions

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1. General procedures

HP 6890 GC containing a 30 m 0.250 mm HP-1 capillary column with a 0.25 mm stationary phase film thickness was used to monitor the reaction. The same GC instrument with a 5973 series mass selective detector was used to Acquire GC/MS data. The flow rate was 1 mL/ min and splitless. NMR spectra were recorded on Bruker AM 500 and Joel AM 200 using 5 mm sample tubes. CDCl₃ was the reference for ¹H and ¹³C NMR spectra. Freon® 11 (CFCl₃) was the reference for ¹⁹F NMR spectra.

2. Thermomorphing curve: plot of solubility versus temperature

Pd complexes **2a-d** and ligand **1e** were dissolved in same amount of deuterated DMF inside the NMR tubes[1]. The HCF₂(CF₂)₃CH₂OH, which was totally soluble in DMF in the temperature range of -40 to 80 °C, was used as an internal standard. The variable temperature NMR (¹H & ¹⁹F) measurements were operated between -40 to 80 °C. The 80 °C was the highest temperature limit for the NMR 500 spectrometer. The general procedures were at the set temperature intervals, after the 256 scans the CF₂ peaks of fluorinated compounds, which usually appear around -120 ppm, in ¹⁹F NMR spectrum was integrated, and its value was compared to that of the integration of the CF₂ group from standard, HCF₂(CF₂)₃CH₂OH. Accordingly, all the ¹H NMR spectra results which were also analyzed with similar procedures above were in good agreements with those of ¹⁹F NMR spectra. The set temperature point intervals are -40, -20, -10, 0, 20, 30, 40, 50 60, 70, and 80 °C.

3. Coupling products identification

The coupling products are identified by GC/MS data and NMR methods. Their GC/MS data are shown below. These products are all known compounds whose NMR data are the same as the literature data.

Stille products:

4a (R= H) $\text{C}_6\text{H}_4\text{CH}=\text{CH}_2$ (m/z; EI): 104 (M^+), 77 ($\text{M}^+ - \text{C}_2\text{H}_3$).

5a (R= CN) $(\text{CN})\text{C}_6\text{H}_4\text{CH}=\text{CH}_2$ (m/z; EI): 129 (M^+), 102 ($\text{M}^+ - \text{C}_2\text{H}_3$), 103 ($\text{M}^+ - \text{CN}$), 77 ($\text{M}^+ - \text{CN} - \text{C}_2\text{H}_3 = 77$).

6a (R= NO_2) $(\text{NO}_2)\text{C}_6\text{H}_4\text{CH}=\text{CH}_2$ (m/z; EI): 149 (M^+), 120 ($\text{M}^+ - \text{C}_2\text{H}_3$), 103 ($\text{M}^+ - \text{NO}_2$), 77 ($\text{M}^+ - \text{NO}_2 - \text{C}_2\text{H}_3$).

7a (R= CH_3) $(\text{CH}_3)\text{C}_6\text{H}_4\text{CH}=\text{CH}_2$ (m/z; EI): 118 (M^+), 93 ($\text{M}^+ - \text{C}_2\text{H}_3$), 103 ($\text{M}^+ - \text{CH}_3$), 77 ($\text{M}^+ - \text{NO}_2 - \text{C}_2\text{H}_3$).

Suzuki-Miyaura products:

12a (X= Ph, Y= CN) $(\text{CN})\text{C}_6\text{H}_4\text{C}_6\text{H}_4\text{C}_6\text{H}_5$ (m/z; EI): 255 (M^+), 229 ($\text{M}^+ - \text{CN}$), 178 ($\text{M}^+ - \text{C}_6\text{H}_5$), 153 ($\text{M}^+ - \text{C}_7\text{H}_4\text{N}$).

13a (X= H, Y= CN) $(\text{CN})\text{C}_6\text{H}_4\text{C}_6\text{H}_5$ (m/z; EI): 179 (M^+), 151 ($\text{M}^+ - \text{CN}$), 102 ($\text{M}^+ - \text{C}_6\text{H}_5$).

13b (X= H, Y= H) $\text{C}_6\text{H}_5\text{C}_6\text{H}_5$ (m/z; EI): 154 (M^+), 77 ($\text{M}^+ - \text{C}_6\text{H}_5$).

13c (X= H, Y= OCH_3) $(\text{CH}_3\text{O})\text{C}_6\text{H}_4\text{C}_6\text{H}_5$ (m/z; EI): 184 (M^+), 153 ($\text{M}^+ - \text{OCH}_3$), 107 ($\text{M}^+ - \text{C}_6\text{H}_5$).

4. Monitoring the catalyst **2b** by the ^{19}F NMR spectrum.

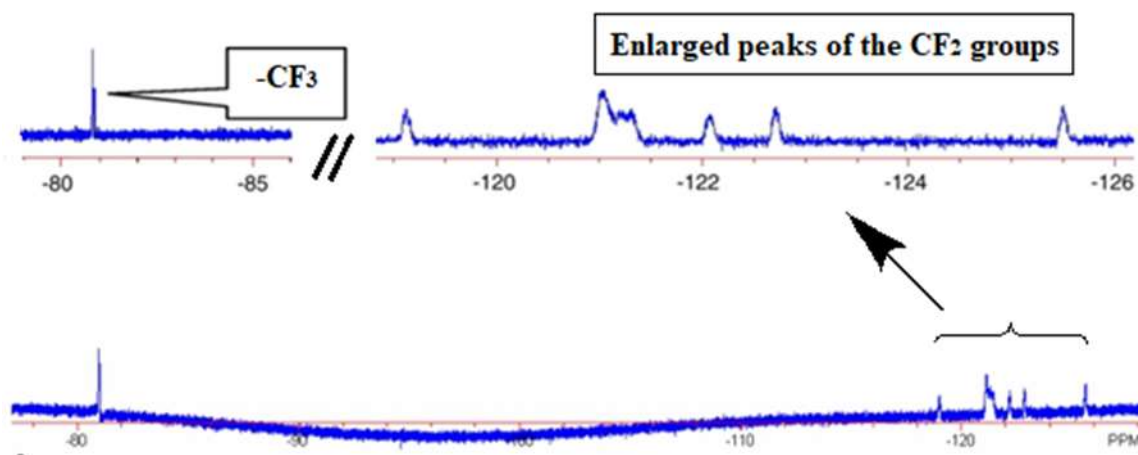


Figure S1. The comparison of ^{19}F NMR spectra of catalyst **2b** from Pd-catalyzed Suzuki-Miyaura reaction (**top image**) before the reaction; (**bottom image**) after the 8 cycles.

^{19}F NMR data of **2b** (NMR data collected in d-DMF at 90 C because of poor solubility.)[1, 2]:

^{19}F NMR (470.5 MHz, DMF- d_6 , d ppm, J Hz) 80.9 (t, 6F, $^3J_{\text{FF}} = 7.52$, $-\text{CF}_3$), CF_2 groups region: 121.0 (4F), 121.3 (20F), 122.1 (4F), 122.7 (4F), 125.5 (4F).

Reference:

1. Lu, N.; Chen, S.-C.; Chen, T.-C.; Liu, L.-K. Palladium-catalyzed Heck reaction under thermomorphic mode. *Tetrahedron Letters* **2008**, *49*, 371-375.
2. Lu, N.; Chen, Y.-C.; Chen, W.-S.; Chen, T.-L.; Wu, S.-J. Efficient, Recoverable, Copper-Free Sonogashira Reaction under FBS and Thermomorphic Mode. *J. Organomet. Chem.* **2009**, *694*, 278-284.