

Supplementary Materials

Molecular Dynamics Approaches Dissect Molecular Mechanisms Underlying Methylene Blue–Glycosaminoglycan Interactions

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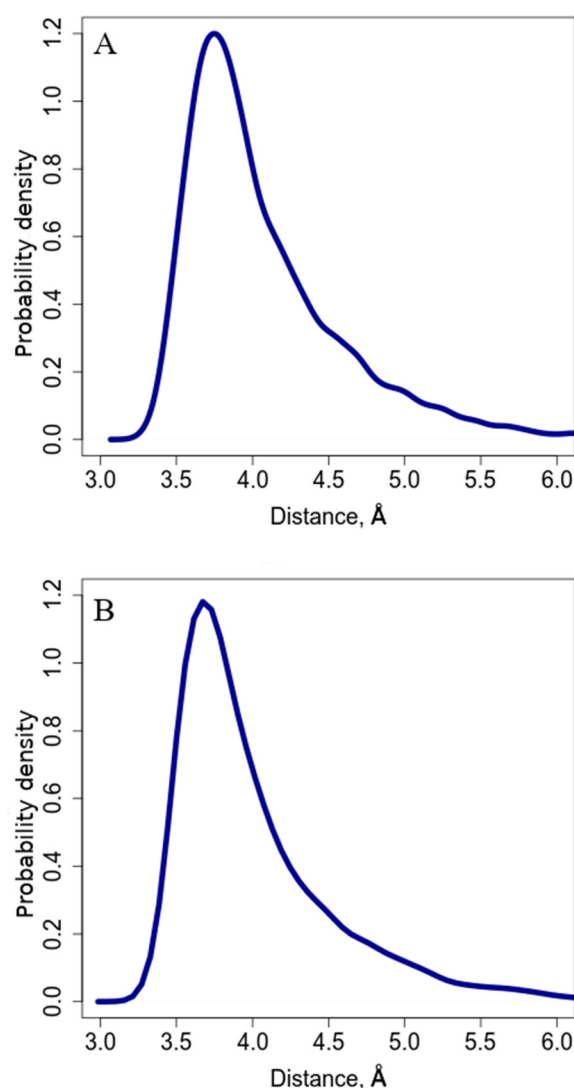


Figure S1. Density of probability for the distance between the planes defined by the aromatic carbons of two MB molecules in a stack for the: A) two unbound MB and B) MB-HP dp10.

Table S1. Comparison of the binding free energies in MB-GAG dp10 complexes analyzed in the present study to the previously obtained binding ratios by Jia et al. [35].

GAG dp10	$\Delta G/\text{drugs number}^*$	Binding ratio [35]**
deHP	+	-
HP	+++	+
HS1	+	+++
HS2	+++	++
HS3	+	+++
CS4	++	-
CS6	+++	-

*+: $\Delta G < 0$ kcal/mol; ++: $-5 \text{ kcal/mol} < \Delta G < -3 \text{ kcal/mol}$; +++: $\Delta G < -5 \text{ kcal/mol}$; **+ binding ratio > 0 ; ++ $3 < \text{binding ratio} < 4$; +++ binding ratio > 4 (higher binding ratio corresponds to stronger binding).