




Article

Entropy Related to K -Banhatti Indices via Valency Based on the Presence of C_6H_6 in Various Molecules

Muhammad Usman Ghani ^{1,*}, Francis Joseph H. Campena ^{2,*}, Muhammad Kashif Maqbool ³, Jia-Bao Liu ^{4,*}, Sanaullah Dehraj ⁵, Murat Cancan ⁶ and Fahad M. Alharbi ⁷

¹ Institute of Mathematics, Khawaja Fareed University of Engineering & Information Technology, Abu Dhabi Road, Rahim Yar Khan 64200, Pakistan

² Department of Mathematics and Statistics, College of Science, De La Salle University, 2401 Taft Avenue, Manila 1004, Philippines

³ The Government Sadiq Egerton College Bahawalpur, University Chowk, Bahawalpur Cantt, Bahawalpur 63100, Pakistan

⁴ School of Mathematics and Physics, Anhui Jianzhu University, Hefei 230601, China

⁵ Department of Mathematics and Statistics, Quaid-e-Awam University of Engineering, Science and Technology, Sakrand Road, Nawabshah 67480, Pakistan

⁶ Faculty of Education, Yuzuncu Yil University, 65090 Van, Turkey

⁷ Department of Mathematics, Al-Qunfudah University College, Umm Al-Qura University, Mecca 24382, Saudi Arabia

* Correspondence: usmanghani85a@gmail.com (M.U.G.); francis.campena@dlsu.edu.ph (F.J.H.C.); liujiabao@ahjzu.edu.cn (J.-B.L.)

Abstract: Entropy is a measure of a system's molecular disorder or unpredictability since work is produced by organized molecular motion. Shannon's entropy metric is applied to represent a random graph's variability. Entropy is a thermodynamic function in physics that, based on the variety of possible configurations for molecules to take, describes the randomness and disorder of molecules in a given system or process. Numerous issues in the fields of mathematics, biology, chemical graph theory, organic and inorganic chemistry, and other disciplines are resolved using distance-based entropy. These applications cover quantifying molecules' chemical and electrical structures, signal processing, structural investigations on crystals, and molecular ensembles. In this paper, we look at K -Banhatti entropies using K -Banhatti indices for C_6H_6 embedded in different chemical networks. Our goal is to investigate the valency-based molecular invariants and K -Banhatti entropies for three chemical networks: the circumnaphthalene (CNB_n), the honeycomb (HB_n), and the pyrene (PY_n). In order to reach conclusions, we apply the method of atom-bond partitioning based on valences, which is an application of spectral graph theory. We obtain the precise values of the first K -Banhatti entropy, the second K -Banhatti entropy, the first hyper K -Banhatti entropy, and the second hyper K -Banhatti entropy for the three chemical networks in the main results and conclusion.

Keywords: C_6H_6 embedded in pyrene network; C_6H_6 embedded in circumnaphthalene network; C_6H_6 embedded in honeycomb network; K -Banhatti entropies



Citation: Ghani, M.U.; Campena, F.J.H.; Maqbool, M.K.; Liu, J.-B.; Dehraj, S.; Cancan, M.; Alharbi, F.M. Entropy Related to K -Banhatti Indices via Valency Based on the Presence of C_6H_6 in Various Molecules. *Molecules* **2023**, *28*, 452. <https://doi.org/10.3390/molecules28010452>

Academic Editors: Niels Engholm Henriksen and Francisco Torrents

Received: 7 November 2022

Revised: 18 December 2022

Accepted: 19 December 2022

Published: 3 January 2023



Copyright: © 2023 by the authors. Licensee MDPI, Basel, Switzerland. This article is an open access article distributed under the terms and conditions of the Creative Commons Attribution (CC BY) license (<https://creativecommons.org/licenses/by/4.0/>).

1. Introduction

In the late 1990s, researchers began investigating the information content of complex networks, [1] and graphs based on Shannon's entropy work [2]. Numerous quantitative measures for analyzing complex networks have been proposed [3,4] spanning a wide range of issues in discrete mathematics, computer science, information theory, statistics, chemistry, biology, and other fields [5,6]. Graph entropy measures, for example, have been widely used to characterize the structure of graph-based systems [7,8] in mathematical chemistry, biology, and computer science-related areas. The concept of graph entropy [9], developed by Rashevsky [10] and Trucco [11] has been used to quantify the structural complexity of graphs [12,13].

Chemical indices are important tools for studying different physico-chemical properties of molecules without having to conduct several tests. In the investigation of medicines, quantitative structure-activity relationships (QSAR) use mathematical computations to decipher the chemical properties [14,15]. Some researchers have analyzed the topological and K-Banhatti indices in [16,17]. Mowshowitz [18] introduced the entropy of a graph as an information-theoretic quantity. The complexity is evident here. The well-known Shannon's entropy [2] is used to calculate the entropy of a graph. Importantly, Mowshowitz interpreted his graph entropy measure as a graph's structural information content and demonstrated that this quantity satisfies important properties when used with product graphs [18]. Inspired by Dehmer and Kraus [19], it was discovered that determining the minimal values of graph entropies is difficult due to a lack of analytical methods to address this specific problem.

The first-order valence-based K-Banhatti indices [17,20,21] are, respectively, as follows:

$$B_1(G, x) = \sum_{a_i \sim a_j} x^{(V_{a_i} + V_{a_j})} \quad \text{and} \quad B_{1(G)} = \sum_{a_i \sim a_j} (V_{a_i} + V_{a_j}). \quad (1)$$

where a_i and a_j denote the atoms of a molecule, V_{a_i} and V_{a_j} represent the valency of each atom, and, if a_i and a_j are connected or have atom bonds, then we denote this by $a_i \sim a_j$. Accordingly, the second valence-based K-Banhatti index [22] and polynomial are as follows:

$$B_2(G, x) = \sum_{a_i \sim a_j} x^{(V_{a_i} \times V_{a_j})} \quad \text{and} \quad B_{2(G)} = \sum_{a_i \sim a_j} (V_{a_i} \times V_{a_j}) \quad (2)$$

The hyper K-Banhatti index and first and second polynomial types [21] are as follows:

$$HB_1(G, x) = \sum_{a_i \sim a_j} x^{(V_{a_i} + V_{a_j})^2} \quad \text{and} \quad HB_{1(G)} = \sum_{a_i \sim a_j} (V_{a_i} + V_{a_j})^2 \quad (3)$$

$$HB_2(G, x) = \sum_{a_i \sim a_j} x^{(V_{a_i} \times V_{a_j})^2} \quad \text{and} \quad HB_{2(G)} = \sum_{a_i \sim a_j} (V_{a_i} \times V_{a_j})^2 \quad (4)$$

The Banhatti indices were proposed by the Indian mathematician Kulli as a result of Milan Randić's 1972 Zagreb indices. With various techniques, such as modified and hyper-Banhatti indices, Kulli offered a number of studies on Banhatti indices. These indices have excellent associations with several chemical and nonchemical graph properties. The amount of thermal energy per unit temperature in a system that cannot be used for useful work is known as entropy [23,24]. In this article, we investigate the graphs of different molecules, namely the pyrene network, the circumnaphthalene series of benzenoid, and the honeycomb benzenoid network, to determine the K-Banhatti entropies' use of K-Banhatti indices [21,25].

2. Definitions of Entropies via K-Banhatti Indices

Manzoor et al. in [26] and Ghani et al. in [27] recently offered another strategy that is a little bit novel in the literature: applying the idea of Shannon's entropy [28] in terms of topological indices. The following formula represents the valency-based entropy:

$$ENT_{\mu(G)} = - \sum_{a_i \sim a_j} \frac{\mu(V_{a_i} V_{a_j})}{\sum_{a_i \sim a_j} \mu(V_{a_i} V_{a_j})} \log \left\{ \frac{\mu(V_{a_i} V_{a_j})}{\sum_{a_i \sim a_j} \mu(V_{a_i} V_{a_j})} \right\}. \quad (5)$$

where a_1, a_2 represents the atoms, E_G represents the edge set, and $\mu(V_{a_i} V_{a_j})$ represents the edge weight of edge $(V_{a_i} V_{a_j})$.

- **The first K-Banhatti entropy**

Let $\mu(V_{a_i}V_{a_j}) = V_{a_i} + V_{a_j}$. The first-order K-Banhatti index (1) is thus provided as

$$B_{1(G)} = \sum_{a_i \sim a_j} \{V_{a_i} + V_{a_j}\} = \sum_{a_i \sim a_j} \mu(V_{a_i}V_{a_j}).$$

The use of (5), the first K-Banhatti entropy, is shown below:

$$ENT_{B_{1(G)}} = \log(B_{1(G)}) - \frac{1}{B_{1(G)}} \log \left\{ \prod_{a_i \sim a_j} [V_{a_i} + V_{a_j}]^{[V_{a_i} + V_{a_j}]} \right\}. \quad (6)$$

- **The Second K-Banhatti entropy**

Let $\mu(V_{a_i}V_{a_j}) = V_{a_i} \times V_{a_j}$. Then, the second K-Banhatti index (2) is given by

$$B_{2(G)} = \sum_{a_i \sim a_j} \{(V_{a_i} \times V_{a_j})\} = \sum_{a_i \sim a_j} \mu(V_{a_i}V_{a_j}).$$

The use of (5), the second K-Banhatti entropy, is shown below:

$$ENT_{B_{2(G)}} = \log(B_{2(G)}) - \frac{1}{B_{2(G)}} \log \left\{ \prod_{a_i \sim a_j} [V_{a_i} \times V_{a_j}]^{[V_{a_i} \times V_{a_j}]} \right\}. \quad (7)$$

- **Entropy related to the first K-hyper Banhatti index**

Let $\mu(V_{a_i}V_{a_j}) = (V_{a_i} + V_{a_j})^2$. Then, the first K-hyper Banhatti index (3) is given by

$$HB_{1(G)} = \sum_{a_i \sim a_j} \{(V_{a_i} + V_{a_j})^2\} = \sum_{a_i \sim a_j} \mu(V_{a_i}V_{a_j}).$$

The use of (5), the first K-hyper Banhatti entropy, is shown below:

$$ENT_{HB_{1(G)}} = \log(HB_{1(G)}) - \frac{1}{HB_{1(G)}} \log \left\{ \prod_{a_i \sim a_j} [V_{a_i} + V_{a_j}]^{2[V_{a_i} + V_{a_j}]} \right\}. \quad (8)$$

- **Entropy related to the second K-hyper Banhatti index**

Let $\mu(V_{a_i}V_{a_j}) = (V_{a_i} \times V_{a_j})^2$. Then, the second K-hyper Banhatti index (4) is given by

$$HB_{2(G)} = \sum_{a_i \sim a_j} \{(V_{a_i} \times V_{a_j})^2\} = \sum_{a_i \sim a_j} \mu(V_{a_i}V_{a_j}).$$

The use of (5), the second K-hyper Banhatti entropy, is shown below:

$$ENT_{HB_{2(G)}} = \log(HB_{2(G)}) - \frac{1}{HB_{2(G)}} \log \left\{ \prod_{a_i \sim a_j} [V_{a_i} \times V_{a_j}]^{2[V_{a_i} \times V_{a_j}]} \right\}. \quad (9)$$

3. The Pyrene Network

The precise arrangement of rings in the benzenoid system offers a transformation within a sequence of benzenoid structures of the benzenoid graph, which changes the structure. The Pyrene network PY_n is a collection of hexagons, and it is a simple, connected 2D planar graph, where n represents the number of hexagons in any row of PY_n (see Figure 1). Accordingly, the Pyrene network is a series of benzenoid rings, and the total number of benzenoid rings is n^2 in $PY(n)$. We sum up the Zagreb polynomial and topological indices of $PY(n)$ in this section.

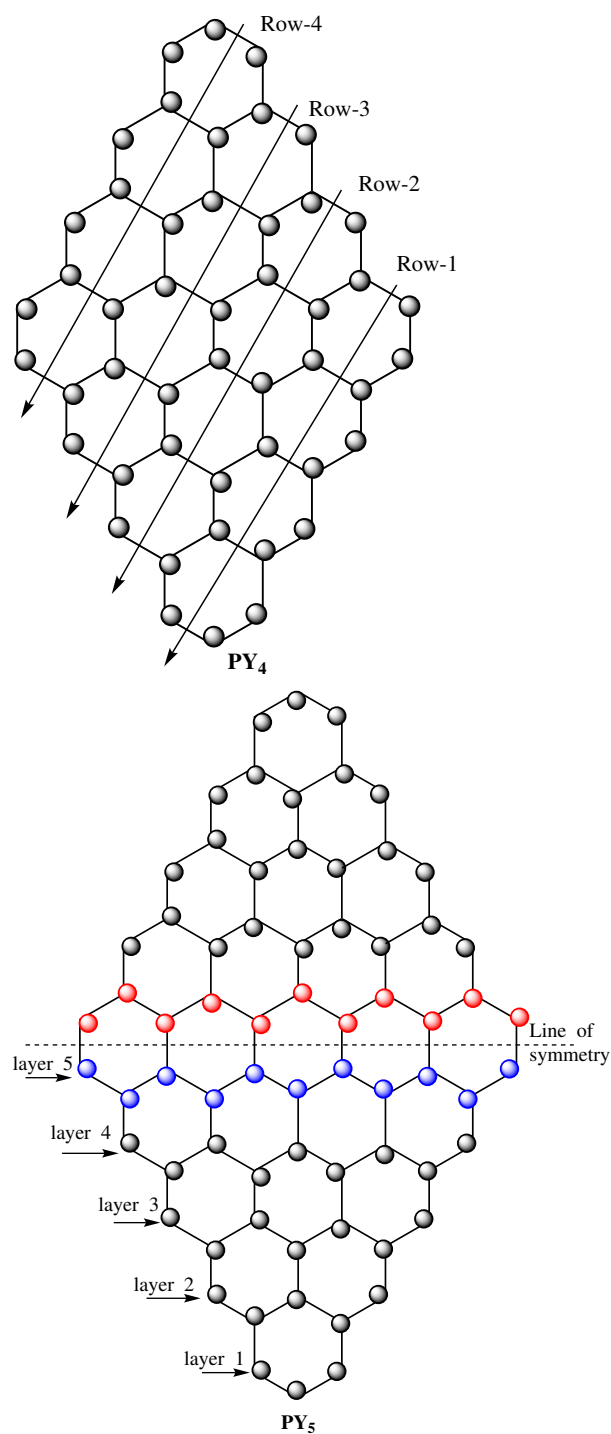


Figure 1. Pyrene network PY_n .

Results and discussion

The number of atoms and the total number of atomic bonds for PY_n are now determined. Let us consider the line of symmetry that divides PY_n into two symmetric parts, as shown in Figure 1. Let us denote the number of atoms in one symmetric portion of PY_n by x and the number of layers by l . In one symmetric part of PY_n , there are l layers of carbon

atoms for $1 \leq l \leq n$, as indicated in Figure 1. Then, an l th layer contains $2l + 1$ carbon atoms. Accordingly, we have

$$\begin{aligned} x &= \sum_{l=1}^n (2l + 1) = 3 + 5 + 7 + \dots + (2n + 1) \\ &= \frac{n}{2} \{2(3) + (n - 1)2\} \\ &= n^2 + 2n \quad (\text{the sum of an arithmetic series}) \end{aligned}$$

The number of atoms in PY_n is $2x = 2n^2 + 4n$ because of the two symmetric parts in PY_n . Furthermore, a PY_n corner atom and an atom other than a corner atom have valencies two and three, respectively. Thus, out of $2n^2 + 4n$ atoms, $4n + 2$ atoms have valency two, and $2(n^2 - 1)$ atoms have valency three. So, by using Formula (1), the number of atomic bonds in PY_n is $3n^2 + 4n - 1$. According to the valencies (two and three) of the atoms, there are three types of atomic bonds, which are (2,2), (2,3), and (3,3) in PY_n . On the basis of valency, Table 1 provides the partition of the set of atomic bonds.

The edge partition of PY_n is:

$$\begin{aligned} V_{2\sim 2} &= \{e = V_i \sim V_j, \text{ for all } V_i, V_j \text{ contained in } E(PY_n), \text{ whenever } d_{V_i} = 2, d_{V_j} = 2\}, \\ V_{2\sim 3} &= \{e = V_i \sim V_j, \text{ for all } V_i, V_j \text{ contained in } E(PY_n), \text{ whenever } d_{V_i} = 2, d_{V_j} = 3\}, \\ V_{3\sim 3} &= \{e = V_i \sim V_j, \text{ for all } V_i, V_j \text{ contained in } E(PY_n), \text{ whenever } d_{V_i} = 3, d_{V_j} = 3\}. \end{aligned}$$

This partition provides:

Table 1. Atomic bond partition of PY_n .

Atomic Bond Type	$V_{2\sim 2}$	$V_{2\sim 3}$	$V_{3\sim 3}$
Number of atom bonds	6	$8(n - 1)$	$3n^2 - 4n + 1$

• **Entropy related to the first K-Banhatti index of PY_n**

Let PY_n be the Pyrene network of C_6H_6 . The first K-Banhatti polynomial is calculated using Equation (1) and Table 1.

$$\begin{aligned} B_1(PY_n, x) &= \sum_{V_{(2\sim 2)}} x^{2+2} + \sum_{V_{(2\sim 3)}} x^{2+3} + \sum_{V_{(3\sim 3)}} x^{3+3} \\ &= 6x^4 + 8(n - 1)x^5 + (3n^2 - 4n + 1)x^6. \end{aligned} \tag{10}$$

Following the simplification of Equation (10), we obtain the first K-Banhatti index, which is given at $x = 1$ via differentiation.

$$B_1(PY_n) = 2(9n^2 + 8n - 5) \tag{11}$$

Here, we calculate the first K-Banhatti entropy of PY_n using Table 1 and Equation (11) inside Equation (6) in the following manner:

$$\begin{aligned} ENT_{B_1}(PY_n) &= \log(B_1) - \frac{1}{B_1} \log \left\{ \prod_{V_{(2,2)}} (V_{a_i} + V_{a_j})^{(V_{a_i} + V_{a_j})} \times \prod_{V_{(3,3)}} (V_{a_i} + V_{a_j})^{(V_{a_i} + V_{a_j})} \right. \\ &\times \left. \prod_{V_{(3,3)}} (V_{a_i} + V_{a_j})^{(V_{a_i} + V_{a_j})} \right\} \\ &= \log 2(9n^2 + 8n - 5) - \frac{1}{29n^2 + 8n - 5} \log \left\{ 16(4)^4 \times 8(n - 1)(5)^5 \times (3n^2 - 4n + 1)(6)^6 \right\}. \end{aligned}$$

- **The second K -Banhatti entropy of PY_n**

Let PY_n be the Pyrene network of C_6H_6 . Then, using Equation (2) and Table 1, the second K -Banhatti polynomial is

$$\begin{aligned} B_2(PY_n) &= \sum_{V_{(2\sim 2)}} x^{2 \times 2} + \sum_{V_{(2\sim 3)}} x^{2 \times 3} + \sum_{V_{(3\sim 3)}} x^{3 \times 3} \\ &= 6x^4 + 8(n-1)x^6 + (3n^2 - 4n + 1)x^9. \end{aligned} \quad (12)$$

To differentiate (34) at $x = 1$, we obtain the second K -Banhatti index:

$$B_2(PY_n) = 3(9n^2 + 4n - 5). \quad (13)$$

Here, we calculate the second K -Banhatti entropy of PY_n using Table 1 and Equation (13) in Equation (7) as described below:

$$\begin{aligned} ENT_{B_2}(PY_n) &= \log(B_2) - \frac{1}{B_2} \log \left\{ \prod_{V_{(2,2)}} (V_{a_i} \times V_{a_j})^{(V_{a_i} \times V_{a_j})} \times \prod_{V_{(2,3)}} (V_{a_i} \times V_{a_j})^{(V_{a_i} \times V_{a_j})} \right. \\ &\quad \times \left. \prod_{V_{(3,3)}} (V_{a_i} \times V_{a_j})^{(V_{a_i} \times V_{a_j})} \right\} \\ &= \log 3(9n^2 + 4n - 5) - \frac{1}{3(9n^2 + 4n - 5)} \log \left\{ 16(6^6) \right. \\ &\quad \times \left. 8(n-1)9^9 \times (3n^2 - 4n + 1)12^{12} \times 2(2st - s - t)16^{16} \right\}. \end{aligned} \quad (14)$$

- **Entropy related to the first K -hyper Bhanhatti index of PY_n**

Let PY_n be the Pyrene network of C_6H_6 . Then, using Equation (3) and Table 1, the first K -hyper Bhanhatti polynomial is

$$\begin{aligned} HB_1(PY_n) &= \sum_{V_{(2\sim 2)}} x^{(2+2)^2} + \sum_{V_{(2\sim 3)}} x^{(2+3)^2} + \sum_{V_{(3\sim 3)}} x^{(3+3)^2} \\ &= 6x^{16} + 8(n-1)x^{25} + (3n^2 - 4n + 1)x^{36} \end{aligned} \quad (15)$$

To differentiate (15) at $x = 1$, we obtain the first K -hyper Bhanhatti index

$$HB_1(PY_n) = 2(54n^2 + 28n - 34). \quad (16)$$

Here, we calculate the first K -hyper Bhanhatti entropy of PY_n using Table 1 and Equation (16) in Equation (9) as described below:

$$\begin{aligned} ENT_{HB_1}(PY_n) &= \log(HB_1) - \frac{1}{HB_1} \log \left\{ \prod_{V_{(2,2)}} (V_{a_i} + V_{a_j})^{2(V_{a_i} + V_{a_j})^2} \times \prod_{V_{(2,3)}} (V_{a_i} + V_{a_j})^{2(V_{a_i} + V_{a_j})^2} \right. \\ &\quad \times \left. \prod_{V_{(3,3)}} (V_{a_i} + V_{a_j})^{2(V_{a_i} + V_{a_j})^2} \right\} \\ &= \log 2(54n^2 + 28n - 34) - \frac{1}{2(54n^2 + 28n - 34)} \log \left\{ 16(5^{50}) \right. \\ &\quad \times \left. 8(n-1)(6^{72}) \times (3n^2 - 4n + 1)(7^{98}) \times 2(2st - s - t)(8^{128}) \right\}. \end{aligned} \quad (17)$$

- **Entropy related to the second K -hyper Bhanhatti index PY_n**

Let PY_n be the Pyrene network of C_6H_6 . Then, using Equation (4) and Table 1, the second K -hyper Bannhatti polynomial is

$$\begin{aligned}
 HB_2(PY_n) &= \sum_{V_{(2\sim 2)}} x^{(2 \times 2)^2} + \sum_{V_{(2\sim 3)}} x^{(2 \times 3)^2} + \sum_{V_{(3\sim 3)}} x^{(3 \times 3)^2} \\
 &= 6x^{16} + 8(n-1)x^{36} + (3n^2 - 4n + 1)x^{81}.
 \end{aligned}
 \tag{18}$$

To differentiate (18) at $x = 1$, we obtain the second K -hyper Bannhatti index

$$HB_2(PY_n) = 3(81n^2 - 12n - 37).
 \tag{19}$$

Here, we calculate the second K -hyper Bannhatti entropy of PY_n using Table 1 and Equation (19) in Equation (9) as described below:

$$\begin{aligned}
 ENT_{HB_1}(PY_n) &= \log(HB_1) - \frac{1}{HB_1} \log \left\{ \prod_{V_{(2,2)}} (V_{a_i} \times V_{a_j})^{2(V_{a_i} \times V_{a_j})^2} \times \prod_{V_{(2,3)}} (V_{a_i} \times V_{a_j})^{2(V_{a_i} \times V_{a_j})^2} \right. \\
 &\quad \times \left. \prod_{V_{(3,3)}} (V_{a_i} \times V_{a_j})^{2(V_{a_i} \times V_{a_j})^2} \right\} \\
 &= \log 3(81n^2 - 12n - 37) - \frac{1}{3(81n^2 - 12n - 37)} \log \left\{ 16(6)^{72} \right. \\
 &\quad \times \left. 8(n-1)9^{81} \times (3n^2 - 4n + 1)12^{288} \times 2(2st - s - t)16^{512} \right\}
 \end{aligned}
 \tag{20}$$

Characteristics of K-Bannhatti Indices of PY_n

Here, we contrast the K -Bannhatti indices, namely B_1 , B_2 , HB_1 , and HB_2 for PY_n quantitatively and visually in Table 2 and Figure 2, respectively.

Table 2. Numerical values of K -Bannhatti indices of PY_n .

Values of n	2	3	4	5	6	7	8	9	10	11	12
B_1	94	200	342	520	734	984	1270	1592	1950	2344	2774
B_2	117	264	465	720	1029	1392	1809	2280	2805	3384	4017
HB_1	476	1072	1884	2912	4156	5616	7292	9184	11,292	13,616	16,156
HB_2	789	1968	3633	5784	8421	11,544	15,153	19,248	23,829	28,896	34,449

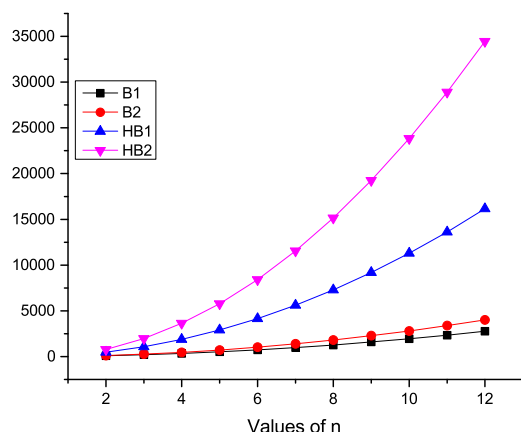


Figure 2. Cont.

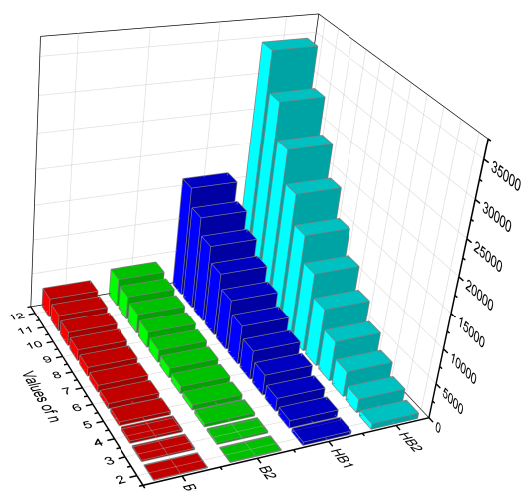


Figure 2. Graphical representation of K -Banhatti indices of PY_n .

4. Circumnaphthalene Series of Benzenoid

Circumnaphthalene is similar to the benzenoid polycyclic aromatic hydrocarbons with the formula $C_{32}H_{14}$ and the ten peri-fused six-member rings in figure CNB_2 . Ovalene is a chemical that is reddish-orange in color. It is only slightly soluble in solvents, such as benzenoid, toluene, and dichloromethane. The circumnaphthalene series of benzenoids is designated by CNB_n , where “ n ” is the number of benzenoid rings in the corner, as seen in Figure 3.

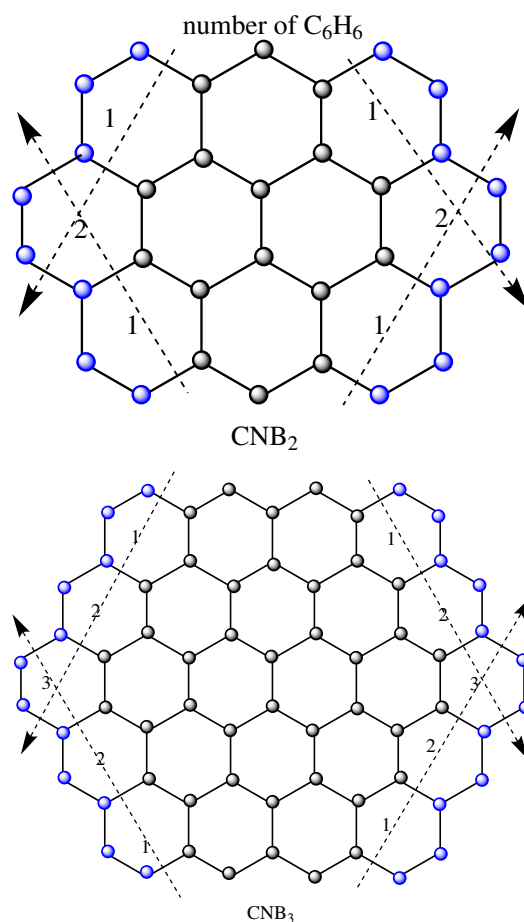


Figure 3. Circumnaphthalene series of benzenoid CNB_n .

Results and Discussion

In Figure 3, we have the following three partitions of the carbon atoms in CNB_n :

$$\begin{aligned}
 V_{2\sim 2} &= \left\{ V_i \sim V_j = e, \forall V_i, V_j \in E(CNB_n) \mid d_{CNB_n}(u) = 2, d_{CNB_n}(v) = 2 \right\}, \\
 V_{2\sim 3} &= \left\{ V_i \sim V_j = e, \forall V_i, V_j \in E(CNB_n) \mid d_{CNB_n}(u) = 2, d_{CNB_n}(v) = 3 \right\}, \\
 V_{3\sim 3} &= \left\{ V_i \sim V_j = e, \forall V_i, V_j \in E(CNB_n) \mid d_{CNB_n}(u) = 3, d_{CNB_n}(v) = 3 \right\}.
 \end{aligned}$$

These partitions provide us with the atomic bond partition of the CNB_n network (see Table 3).

Table 3. Atomic bond partition of CNB_n network.

Types of Atomic Bond	$V_{2\sim 2}$	$V_{2\sim 3}$	$V_{3\sim 3}$
$m_{V_i \sim V_j}$	6	$4(3n - 5)$	$9n^2 - 27n + 19$

• **Entropy related to the 1st K-Banhatti index of CNB_n**

Let CNB_n be the circumnaphthalene series of benzenoid of C_6H_6 . Then, using Equation (1) and Table 3, the first K-Banhatti polynomial is

$$\begin{aligned}
 B_1(CNB_n, x) &= \sum_{V_{(2\sim 2)}} x^{2+2} + \sum_{V_{(2\sim 3)}} x^{2+3} + \sum_{V_{(3\sim 3)}} x^{3+3} \\
 &= 6x^4 + 4(3n - 5)x^5 + (9n^2 - 27n + 19)x^6. \tag{21}
 \end{aligned}$$

Following the simplification of Equation (21), we obtain the first K-Banhatti index, which is given at $x = 1$ via differentiation.

$$B_1(CNB_n) = 2(27n^2 - 51n + 19). \tag{22}$$

Here, we calculate the first K-Banhatti entropy of CNB_n using Table 1 and Equation (24) in Equation (6) in the following manner:

$$\begin{aligned}
 ENT_{B_1}(CNB_n) &= \log(B_1) - \frac{1}{B_1} \log \left\{ \prod_{V_{(2,2)}} (V_{a_i} + V_{a_j})^{(V_{a_i} + V_{a_j})} \times \prod_{V_{(2,3)}} (V_{a_i} + V_{a_j})^{(V_{a_i} + V_{a_j})} \right. \\
 &\quad \times \left. \prod_{V_{(3,3)}} (V_{a_i} + V_{a_j})^{(V_{a_i} + V_{a_j})} \right\} \\
 &= \log 2(27n^2 - 51n + 19) - \frac{1}{2(27n^2 - 51n + 19)} \log \left\{ 16(4)^4 \right. \\
 &\quad \times \left. 4(3n - 5)(5)^5 \times (9n^2 - 27n + 19)(6)^6 \right\}.
 \end{aligned}$$

• **The second K-Banhatti entropy of CNB_n**

Let CNB_n be the circumnaphthalene series of benzenoid of C_6H_6 . Then, using Equation (2) and Table 1, the second K-Banhatti polynomial is

$$\begin{aligned}
 B_2(CNB_n) &= \sum_{V_{(2\sim 2)}} x^{2 \times 2} + \sum_{V_{(2\sim 3)}} x^{2 \times 3} + \sum_{V_{(3\sim 3)}} x^{3 \times 3} \\
 &= 6x^4 + 4(3n - 5)x^6 + (9n^2 - 27n + 19)x^9. \tag{23}
 \end{aligned}$$

To differentiate (23) at $x = 1$, we obtain the second K-Banhatti index

$$B_2(CNB_n) = 3(27n^2 - 57 + 25). \tag{24}$$

Here, we calculate the second K -Banhatti entropy of CNB_n using Table 3 and Equation (24) in Equation (7) as described below:

$$\begin{aligned} ENT_{B_2}(CNB_n) &= \log(B_2) - \frac{1}{B_2} \log \left\{ \prod_{V_{(2,2)}} (V_{a_i} \times V_{a_j})^{(V_{a_i} \times V_{a_j})} \times \prod_{V_{(2,3)}} (V_{a_i} \times V_{a_j})^{(V_{a_i} \times V_{a_j})} \right. \\ &\quad \times \left. \prod_{V_{(3,3)}} (V_{a_i} \times V_{a_j})^{(V_{a_i} \times V_{a_j})} \right\} \\ &= \log 3(27n^2 - 57 + 25) - \frac{1}{3(27n^2 - 57 + 25)} \log \left\{ 6(4^4) \right. \\ &\quad \times \left. 4(3n - 5)6^6 \times (9n^2 - 27n + 19)9^9 \right\}. \end{aligned} \quad (25)$$

- **Entropy related to the first K -hyper Banhatti index of CNB_n**

Let CNB_n be the circumnaphthalene series of benzenoid of C_6H_6 . Then, using Equation (3) and Table 3, the first K -hyper Banhatti polynomial is

$$\begin{aligned} HB_1(CNB_n) &= \sum_{V_{(2\sim 2)}} x^{(2+2)^2} + \sum_{V_{(2\sim 3)}} x^{(2+3)^2} + \sum_{V_{(3\sim 3)}} x^{(3+3)^2} \\ &= 6x^{16} + 4(3n - 5)x^{25} + (9n^2 - 27n + 19)x^{36} \end{aligned} \quad (26)$$

To differentiate (26) at $x = 1$, we obtain the first K -hyper Banhatti index

$$HB_1(CNB_n) = 4(81n^2 - 168n + 70). \quad (27)$$

Here, we calculate the first K -hyper Banhatti entropy of CNB_n using Table 1 and Equation (27) in Equation (9) as described below:

$$\begin{aligned} ENT_{HB_1}(CNB_n) &= \log(HB_1) - \frac{1}{HB_1} \log \left\{ \prod_{V_{(2,2)}} (V_{a_i} + V_{a_j})^{2(V_{a_i} + V_{a_j})^2} \times \prod_{V_{(2,3)}} (V_{a_i} + V_{a_j})^{2(V_{a_i} + V_{a_j})^2} \right. \\ &\quad \times \left. \prod_{V_{(3,3)}} (V_{a_i} + V_{a_j})^{2(V_{a_i} + V_{a_j})^2} \right\} \\ &= \log 4(81n^2 - 168n + 70) - \frac{1}{4(81n^2 - 168n + 70)} \log \left\{ 6(4^{32}) \right. \\ &\quad \times \left. 4(3n - 5)5^{50} \times (9n^2 - 27n + 19)(6^{72}) \right\} \end{aligned} \quad (28)$$

- **Entropy related to the second K -hyper Banhatti index CNB_n**

Let CNB_n be the circumnaphthalene series of benzenoid of C_6H_6 . Then, using Equation (4) and Table 3, the second K -hyper Banhatti polynomial is

$$\begin{aligned} HB_2(CNB_n) &= \sum_{V_{(2\sim 2)}} x^{(2 \times 2)^2} + \sum_{V_{(2\sim 3)}} x^{(2 \times 3)^2} + \sum_{V_{(3\sim 3)}} x^{(3 \times 3)^2} \\ &= 6x^{16} + 4(3n - 5)x^{36} + (9n^2 - 27n + 19)x^{81}. \end{aligned} \quad (29)$$

To differentiate (29) at $x = 1$, we obtain the second K -hyper Banhatti index

$$HB_2(CNB_n) = 3(243n^2 - 585n + 305). \quad (30)$$

Here, we calculate the second K -hyper Banhatti entropy of CNB_n using Table 3 and Equation (30) in Equation (9) as described below:

$$\begin{aligned}
 ENT_{HB_2}(CNB_n) &= \log(HB_2) - \frac{1}{HB_2} \log \left\{ \prod_{V(2,2)} (V_{a_i} \times V_{a_j})^{2(V_{a_i} \times V_{a_j})^2} \times \prod_{V(2,3)} (V_{a_i} \times V_{a_j})^{2(V_{a_i} \times V_{a_j})^2} \right. \\
 &\quad \times \left. \prod_{V(3,3)} (V_{a_i} \times V_{a_j})^{2(V_{a_i} \times V_{a_j})^2} \right\} \\
 &= \log 3(243n^2 - 585n + 305) - \frac{1}{3(243n^2 - 585n + 305)} \log \left\{ 6(4)^{32} \right. \\
 &\quad \times \left. 4(3n - 5)6^{72} \times (9n^2 - 27n + 19)9^{162} \right\} \tag{31}
 \end{aligned}$$

Characteristics of K-Banhatti Indices of CNB_n

Here, we contrast the K-Banhatti indices, namely B_1 , B_2 , HB_1 , and HB_2 for CNB_n quantitatively and visually in Table 4 and Figure 4, respectively.

Table 4. Numerical comparison of topological indices of CNB_n .

Values of n	2	3	4	5	6	7	8	9	10	11	12
B_1	50	218	494	8787	1370	1970	2678	3494	4418	5450	6590
B_2	75	57	291	678	1245	1965	2847	3891	5097	6465	7995
HB_1	280	232	1180	2776	5020	7812	11,452	15,640	20,476	25,960	32,092
HB_2	915	321	2211	5559	10,365	16,629	24,351	33,531	44,169	56,265	69,819

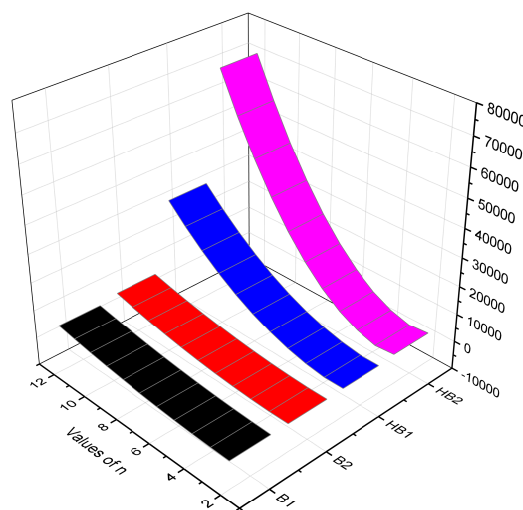
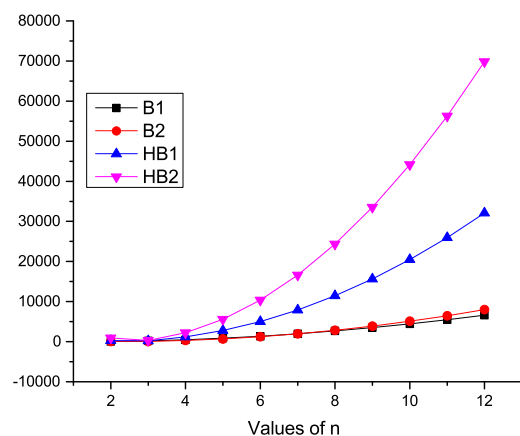


Figure 4. Graphical representation of K-Banhatti indices of CNB_n .

5. The Honeycomb Benzenoid Network

In this section, we introduce a chemical compound that has received more and more attention in recent years, partly due to its applications in chemistry. Honeycomb networks are formed when hexagonal tiling is used recursively in a specific pattern. HB_n denotes an n -dimensional honeycomb network, where n is the number of Benzene rings from center to top, center to bottom, or center to each corner of HB_n , as shown in Figure 5.

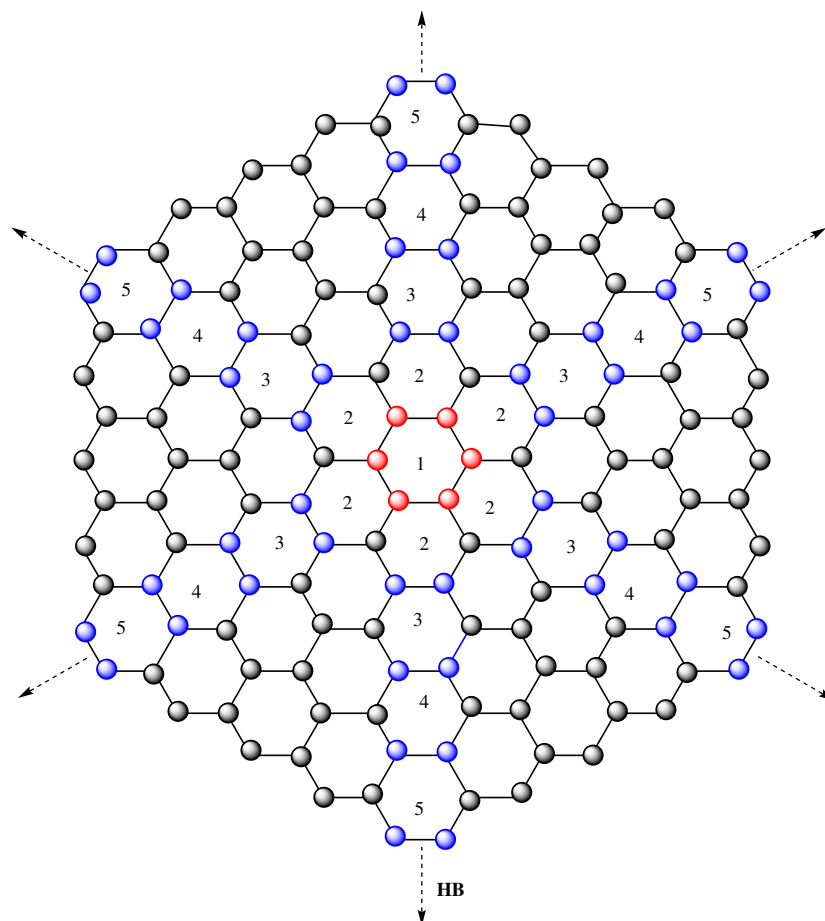


Figure 5. The honeycomb benzenoid network.

Results and Discussion

The honeycomb network HB_n is created by adding a layer of hexagons around the boundary of $HB_{(n-1)}$. In the honeycomb benzenoid network, a $6n$ amount of atoms has valency two, and $6n^2 - 6n$ atoms have valency three. According to the valency of each atom in HB_n , the atomic bonds are classified into three types: $2 \sim 2$, $2 \sim 3$, and $3 \sim 3$ (see Figure 5).

$$\begin{aligned}
 E_{G_{2 \sim 2}} &= \left\{ e = u \sim v, \forall u, v \in E(HB_n) \mid d_u = 2, d_v = 2 \right\}, \\
 E_{G_{2 \sim 3}} &= \left\{ e = u \sim v, \forall u, v \in E(HB_n) \mid d_u = 2, d_v = 3 \right\}, \\
 E_{G_{3 \sim 3}} &= \left\{ e = u \sim v, \forall u, v \in E(HB_n) \mid d_u = 3, d_v = 3 \right\}.
 \end{aligned}$$

Thus, according to the above partition of the atomic bonds, there is $3n(3n - 1)$ total number of atomic bonds used in the honeycomb benzenoid network. The atomic bond partition of HB_n is shown in Table 5.

Table 5. Atomic bond partition of HB_n .

Types of Atomic Bonds	$E_{G_{2\sim 2}}$	$E_{G_{2\sim 3}}$	$E_{G_{3\sim 3}}$
Cardinality of atomic bonds	6	$12(n - 1)$	$(9n^2 - 15n + 6)$

• **Entropy related to the first K -Banhatti index of HB_n**

Let HB_n be the honeycomb benzenoid network of C_6H_6 . Then, using Equation (1) and Table 5, the first K -Banhatti polynomial is

$$\begin{aligned}
 B_1(HB_n, x) &= \sum_{V_{(2\sim 2)}} x^{2+2} + \sum_{V_{(2\sim 3)}} x^{2+3} + \sum_{V_{(3\sim 3)}} x^{3+3} \\
 &= 6x^4 + 12(n - 1)x^5 + (9n^2 - 15n + 6)x^6.
 \end{aligned}
 \tag{32}$$

Following the simplification of Equation (32), we obtain the first K -Banhatti index given at $x = 1$ via differentiation.

$$B_1(HB_n) = 2(27n^2 - 15n - 26).
 \tag{33}$$

Here, we calculate the first K -Banhatti entropy of HB_n using Table 5 and Equation (33) in Equation (6) in the following manner:

$$\begin{aligned}
 ENT_{B_1}(HB_n) &= \log(B_1) - \frac{1}{B_1} \log \left\{ \prod_{V_{(2,2)}} (V_{a_i} + V_{a_j})^{(V_{a_i} + V_{a_j})} \times \prod_{V_{(2,3)}} (V_{a_i} + V_{a_j})^{(V_{a_i} + V_{a_j})} \right. \\
 &\quad \times \left. \prod_{V_{(3,3)}} (V_{a_i} + V_{a_j})^{(V_{a_i} + V_{a_j})} \right\} \\
 &= \log 2(27n^2 - 15n - 26) - \frac{1}{2(27n^2 - 15n - 26)} \log \left\{ 6(4)^4 \right. \\
 &\quad \times \left. 12(n - 1)(5)^5 \times (9n^2 - 15n + 6)(6)^6 \right\}.
 \end{aligned}$$

• **The second K -Banhatti entropy of HB_n**

Let HB_n be the honeycomb benzenoid network of C_6H_6 . Then, using Equation (2) and Table 5, the second K -Banhatti polynomial is

$$\begin{aligned}
 B_2(HB_n) &= \sum_{V_{(2\sim 2)}} x^{2 \times 2} + \sum_{V_{(2\sim 3)}} x^{2 \times 3} + \sum_{V_{(3\sim 3)}} x^{3 \times 3} \\
 &= 6x^4 + 12(n - 1)x^6 + (9n^2 - 15n + 6)x^9.
 \end{aligned}
 \tag{34}$$

To differentiate (34) at $x = 1$, we obtain the second K -Banhatti index

$$B_2(HB_n) = 81n^2 - 87n + 30.
 \tag{35}$$

Here, we calculate the second K -Banhatti entropy of HB_n using Table 5 and Equation (35) in Equation (7) as described below

$$\begin{aligned}
 ENT_{B_2}HB_n &= \log(B_2) - \frac{1}{B_2} \log \left\{ \prod_{V_{(2,2)}} (V_{a_i} \times V_{a_j})^{(V_{a_i} \times V_{a_j})} \times \prod_{V_{(2,3)}} (V_{a_i} \times V_{a_j})^{(V_{a_i} \times V_{a_j})} \right. \\
 &\quad \times \left. \prod_{V_{(3,3)}} (V_{a_i} \times V_{a_j})^{(V_{a_i} \times V_{a_j})} \right\} \\
 &= \log(81n^2 - 87n + 30) - \frac{1}{81n^2 - 87n + 30} \log \left\{ 6(4^4) \right. \\
 &\quad \times \left. 12(n - 1)6^6 \times (9n^2 - 15n + 6)9^9 \right\}.
 \end{aligned}
 \tag{36}$$

- **Entropy related to the first K -hyper Bannhatti index of HB_n**

Let HB_n be the honeycomb benzenoid network of C_6H_6 . Then, using Equation (3) and Table 5, the first K -hyper Bannhatti polynomial is

$$\begin{aligned}
 HB_1(HB_n) &= \sum_{V_{(2\sim 2)}} x^{(2+2)^2} + \sum_{V_{(2\sim 3)}} x^{(2+3)^2} + \sum_{V_{(3\sim 3)}} x^{(3+3)^2} \\
 &= 6x^{16} + 12(n-1)x^{25} + (9n^2 - 15n + 6)x^{36}
 \end{aligned}
 \tag{37}$$

To differentiate (37) at $x = 1$, we obtain the first K -hyper Bannhatti index

$$HB_1(HB_n) = 12(27n^2 - 20n + 1).
 \tag{38}$$

Here, we calculate the first K -hyper Bannhatti entropy of HB_n using Table 5 and Equation (38) in Equation (9) as described below:

$$\begin{aligned}
 ENT_{HB_1}HB_n &= \log(HB_1) - \frac{1}{HB_1} \log \left\{ \prod_{V_{(2,2)}} (V_{a_i} + V_{a_j})^{2(V_{a_i} + V_{a_j})^2} \times \prod_{V_{(2,3)}} (V_{a_i} + V_{a_j})^{2(V_{a_i} + V_{a_j})^2} \right. \\
 &\times \left. \prod_{V_{(3,3)}} (V_{a_i} + V_{a_j})^{2(V_{a_i} + V_{a_j})^2} \right\} \\
 &= \log 12(27n^2 - 20n + 1) - \frac{1}{12(27n^2 - 20n + 1)} \log \left\{ 6(4^{32}) \right. \\
 &\times \left. 12(n-1)(5^{50}) \times (9n^2 - 15n + 6)(6^{72}) \right\}.
 \end{aligned}
 \tag{39}$$

- **Entropy related to the second K -hyper Bannhatti index HB_n**

Let HB_n be the honeycomb benzenoid network of C_6H_6 . Then, using Equation (4) and Table 5, the second K -hyper Bannhatti polynomial is

$$\begin{aligned}
 HB_2(HB_n) &= \sum_{V_{(2\sim 2)}} x^{(2 \times 2)^2} + \sum_{V_{(2\sim 3)}} x^{(2 \times 3)^2} + \sum_{V_{(3\sim 3)}} x^{(3 \times 3)^2} \\
 &= 6x^4 + 12(n-1)x^{36} + (9n^2 - 15n + 6)x^{81}.
 \end{aligned}
 \tag{40}$$

To differentiate (40) at $x = 1$, we obtain the second K -hyper Bannhatti index

$$HB_2(HB_n) = 3(243n^2 - 261n + 26).
 \tag{41}$$

Here, we calculate the second K -hyper Bannhatti entropy of HB_n using Table 5 and Equation (41) in Equation (9), as described below:

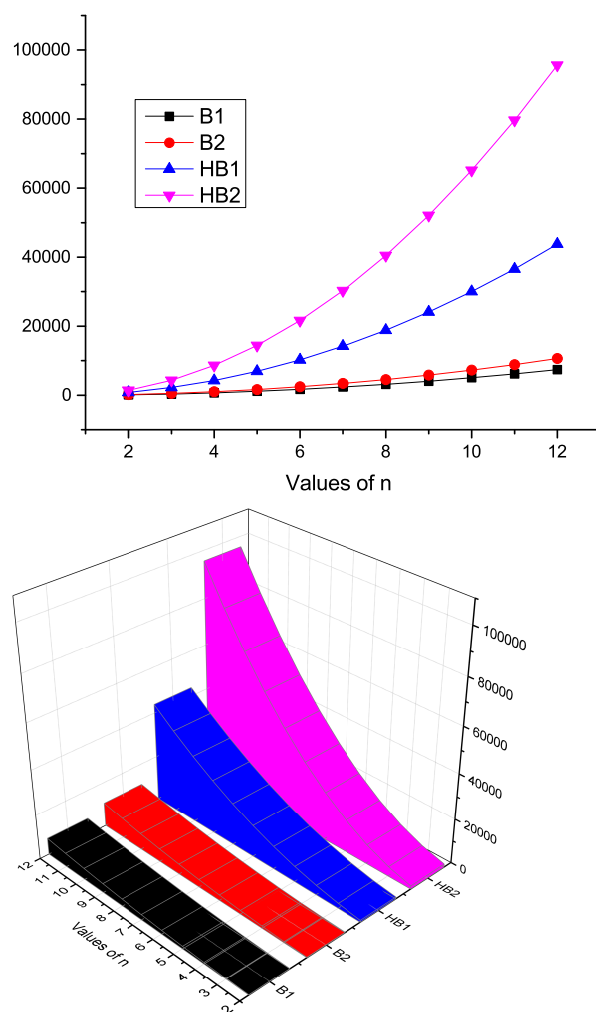
$$\begin{aligned}
 ENT_{HB_2}HB_n &= \log(HB_2) - \frac{1}{HB_2} \log \left\{ \prod_{V_{(2,2)}} (V_{a_i} \times V_{a_j})^{2(V_{a_i} \times V_{a_j})^2} \times \prod_{V_{(2,3)}} (V_{a_i} \times V_{a_j})^{2(V_{a_i} \times V_{a_j})^2} \right. \\
 &\times \left. \prod_{V_{(3,3)}} (V_{a_i} \times V_{a_j})^{2(V_{a_i} \times V_{a_j})^2} \right\} \\
 &= \log 3(243n^2 - 261n + 26) - \frac{1}{3(243n^2 - 261n + 26)} \log \left\{ 6(4)^{32} \right. \\
 &\times \left. 12(n-1)6^{72} \times (9n^2 - 15n + 6)9^{162} \right\}.
 \end{aligned}
 \tag{42}$$

Characteristics of K -Bannhatti Indices of HB_n

Here, we contrast the K -Bannhatti indices, namely B_1 , B_2 , HB_1 , and HB_2 for HB_n quantitatively and visually in Table 6 and Figure 6, respectively.

Table 6. Numerical comparison of topological indices of HB_n .

Values of n	2	3	4	5	6	7	8	9	10	11	12
B_1	104	344	692	1148	1712	2384	3164	4052	5048	6152	7364
B_2	180	498	978	1620	2424	3390	4518	5808	7260	8874	10650
HB_1	828	2208	4236	6912	10,236	14,208	18,828	24,096	30,012	36,576	43,788
HB_2	1428	4290	8610	14,388	21,624	30,318	40,470	52,080	65,148	79,674	95,658

**Figure 6.** Graphical representation of K -Banhatti indices of HB_n .

6. Conclusions

By using Shannon's entropy and the entropy definitions of Chen et al., we looked into the graph entropies connected to a novel information function and assessed the link between degree-based topological indices and degree-based entropies in this work. Industrial chemistry has a strong foundation in the concept of distance-based entropy. The Pyrene network, PY_n ; the circumnaphthalene series of benzenoid, CNB_n ; and the honeycomb benzenoid network, HB_n were studied, and their valency-based K -Banhatti indices were estimated using four K -Banhatti polynomials with a set partition and an atom bonds approach. The acquired results are valuable for anticipating numerous molecular features of chemical substances, such as boiling point, π electron energy, pharmaceutical configuration, and many more concepts. Our results can be applied to determine the electronic structure, signal processing, physicochemical reactions, and complexity of molecules and molecular ensembles for PY_n , CNB_n , and HB_n . Together with chemical structure, thermodynamic entropy, energy, and computer sciences, the K -Banhatti entropy can be crucial to linking

different fields and serving as the basis for future interdisciplinary research. We intend to extend this idea to different chemical structures in the future, opening up new directions for study in this area.

Author Contributions: Conceptualization, M.U.G. and F.J.H.C.; methodology, M.K.M.; software, J.-B.L.; validation, F.M.A.; formal analysis, F.M.A.; investigation, F.M.A., S.D.; resources, F.J.H.C.; data curation, M.C.; writing—original draft preparation, M.U.G.; writing—review and editing, F.J.H.C.; visualization, M.U.G.; supervision, M.K.M.; project administration, M.U.G.; funding acquisition, F.J.H.C. All authors have read and agreed to the published version of the manuscript.

Funding: This research received no external funding.

Institutional Review Board Statement: Not applicable for studies not involving humans or animals.

Informed Consent Statement: Not applicable.

Data Availability Statement: No data available to support this study.

Conflicts of Interest: The authors declare no conflict of interest.

Sample Availability: No data available.

References

1. Tag El Din, E.S.M.; Sultan, F.; Ghani, M.U.; Liu, J.B.; Dehraj, S.; Cancan, M.; Alharbi, F.M.; Alhushaybari, A. Some Novel Results Involving Prototypical Computation of Zagreb Polynomials and Indices for SiO₄ Embedded in a Chain of Silicates. *Molecules* **2022**, *28*, 201. [\[CrossRef\]](#)
2. Shannon, C.; Weaver, W. *The Mathematical Theory of Communication*; University of Illinois Press: Urbana, IL, USA, 1949.
3. Ali, A.; Furtula, B.; Redžepović, I.; Gutman, I. Atom-bond sum-connectivity index. *J. Math. Chem.* **2022**, *60*, 2081–2093. [\[CrossRef\]](#)
4. Ghani, M.U.; Kashif Maqbool, M.; George, R.; Ofem, A.E.; Cancan, M. Entropies Via Various Molecular Descriptors of Layer Structure of H₃BO₃. *Mathematics* **2022**, *10*, 4831. [\[CrossRef\]](#)
5. Liu, J.B.; Arockiaraj, M.; Arulperumjothi, M.; Prabhu, S. Distance based and bond additive topological indices of certain repurposed antiviral drug compounds tested for treating COVID-19. *Int. J. Quantum Chem.* **2021**, *121*, e26617. [\[CrossRef\]](#) [\[PubMed\]](#)
6. Liu, J.B.; Wang, S.; Wang, C.; Hayat, S. Further results on computation of topological indices of certain networks. *IET Control Theory Appl.* **2017**, *11*, 2065–2071. [\[CrossRef\]](#)
7. Liu, J.B.; Gao, W.; Siddiqui, M.K.; Farahani, M.R. Computing three topological indices for Titania nanotubes TiO₂ [m, n]. *AKCE Int. J. Graphs Comb.* **2016**, *13*, 255–260. [\[CrossRef\]](#)
8. Liu, J.B.; Shafiq, M.K.; Ali, H.; Naseem, A.; Maryam, N.; Asghar, S.S. Topological Indices of mth Chain Silicate Graphs. *Mathematics* **2019**, *7*, 42. [\[CrossRef\]](#)
9. Alam, A.; Ghani, M.U.; Kamran, M.; Shazib Hameed, M.; Hussain Khan, R.; Baig, A.Q. Degree-Based Entropy for a Non-Kekulean Benzenoid Graph. *J. Math.* **2022**, *2022*, 2288207.
10. Rashevsky, N. Life, information theory, and topology. *Bull. Math. Biophys.* **1955**, *17*, 229–235. [\[CrossRef\]](#)
11. Trucco, E. A note on the information content of graphs. *Bull. Math. Biophys.* **1956**, *18*, 129–135. [\[CrossRef\]](#)
12. Chu, Y.M.; Khan, A.R.; Ghani, M.U.; Ghaffar, A.; Inc, M. Computation of Zagreb Polynomials and Zagreb Indices for Benzenoid Triangular & Hourglass System. *Polycycl. Aromat. Compd.* **2022**, *2022*, 1–10. [\[CrossRef\]](#)
13. Liu, J.B.; Younas, M.; Habib, M.; Yousaf, M.; Nazeer, W. M-Polynomials and Degree-Based Topological Indices of VC 5 C 7 [p, q] and HC 5 C 7 [p, q] Nanotubes. *IEEE Access* **2019**, *7*, 41125–41132. [\[CrossRef\]](#)
14. Ghani, M.U.; Sultan, F.; El Sayed, M.; Cancan, M.; Ali, S. SiO₄ characterization in a chain and C₆ H₆ embedded in a Non-kekulean structure for Kulli Temperature indices. *Res. Square* **2022**. [\[CrossRef\]](#)
15. Zhang, Y.F.; Ghani, M.U.; Sultan, F.; Inc, M.; Cancan, M. Connecting SiO₄ in Silicate and Silicate Chain Networks to Compute Kulli Temperature Indices. *Molecules* **2022**, *27*, 7533. [\[CrossRef\]](#) [\[PubMed\]](#)
16. Hamid, K.; Muhammad, H.A.B.; Iqbal, M.W.; Hamza, M.A.; Bhatti, S.U.; Aqeel, M. Topological analysis empowered bridge network variants by dharwad indices. *Online Open Access* **2022**, *41*. [\[CrossRef\]](#)
17. Wang, W.; Naeem, M.; Rauf, A.; Riasat, A.; Aslam, A.; Anoh Yannick, K. On Analysis of Banhatti Indices for Hyaluronic Acid Curcumin and Hydroxychloroquine. *J. Chem.* **2021**, *2021*, 7468857. [\[CrossRef\]](#)
18. Mowshowitz, A. Entropy and the complexity of graphs: I. An index of the relative complexity of a graph. *Bull. Math. Biophys.* **1968**, *30*, 175–204. [\[CrossRef\]](#)
19. Dehmer, M.; Varmuza, K.; Borgert, S.; Emmert-Streib, F. On entropy-based molecular descriptors: Statistical analysis of real and synthetic chemical structures. *J. Chem. Inf. Model.* **2009**, *49*, 1655–1663. [\[CrossRef\]](#)
20. Hamid, K.; Iqbal, M.W.; Arif, E.; Mahmood, Y.; Khan, A.S.; Kama, N.; Azmi, A.; Ikram, A. K-Banhatti Invariants Empowered Topological Investigation of Bridge Networks. *CMC-Comput. Mater. Contin.* **2022**, *73*, 5423–5440. [\[CrossRef\]](#)

21. Kulli, V. On K Banhatti indices of graphs. *J. Comput. Math. Sci.* **2016**, *7*, 213–218.
22. Anjum, M.S.; Safdar, M.U. K Banhatti and K hyper-Banhatti indices of nanotubes. *Eng. Appl. Sci. Lett.* **2019**, *2*, 19–37. [[CrossRef](#)]
23. Amigó, J.M.; Balogh, S.G.; Hernández, S. A brief review of generalized entropies. *Entropy* **2018**, *20*, 813. [[CrossRef](#)] [[PubMed](#)]
24. Tsai, M.H. Physical properties of high entropy alloys. *Entropy* **2013**, *15*, 5338–5345. [[CrossRef](#)]
25. Gao, W.; Muzaffar, B.; Nazeer, W. K-Banhatti and K-hyper Banhatti indices of dominating David derived network. *Open J. Math. Anal.* **2017**, *1*, 13–24. [[CrossRef](#)]
26. Manzoor, S.; Chu, Y.M.; Siddiqui, M.K.; Ahmad, S. On topological aspects of degree based entropy for two carbon nanosheets. *Main Group Met. Chem.* **2020**, *43*, 205–218. [[CrossRef](#)]
27. Ghani, M.U.; Sultan, F.; Tag El Din, E.S.M.; Khan, A.R.; Liu, J.B.; Cancan, M. A Paradigmatic Approach to Find the Valency-Based K-Banhatti and Redefined Zagreb Entropy for Niobium Oxide and a Metal–Organic Framework. *Molecules* **2022**, *27*, 6975. [[CrossRef](#)]
28. Yang, W.; Xu, K.; Lian, J.; Ma, C.; Bin, L. Integrated flood vulnerability assessment approach based on TOPSIS and Shannon entropy methods. *Ecol. Indic.* **2018**, *89*, 269–280. [[CrossRef](#)]

Disclaimer/Publisher’s Note: The statements, opinions and data contained in all publications are solely those of the individual author(s) and contributor(s) and not of MDPI and/or the editor(s). MDPI and/or the editor(s) disclaim responsibility for any injury to people or property resulting from any ideas, methods, instructions or products referred to in the content.