

Article

Structural, Spectroscopic, and Dynamic Properties of $Li_2^+ (X^2 \Sigma_g^+)$ in Interaction with Krypton Atom

Samah Saidi ^{1,2,*} , Nesrine Mabrouk ¹, Jamila Dhiflaoui ¹  and Hamid Berriche ^{1,3} 

¹ Laboratory of Interfaces and Advanced Materials, Physics Department, Faculty of Sciences of Monastir, Avenue de l'Environnement, Monastir 5019, Tunisia; dhiflaouijamila@gmail.com (J.D.); hamid.berriche@aurak.ac.ae (H.B.)

² Department of Physics, College of Science and Humanities in Al-Kharj, Prince Sattam Bin Abdulaziz University, Al-Kharj 11942, Saudi Arabia

³ Mathematics and Natural Sciences Department, School of Arts and Sciences, American University of Ras Al Khaimah, Ras Al-Khaimah 10021, United Arab Emirates

* Correspondence: s.saidi@psau.edu.sa

Abstract: We report a computational study of the potential energy surface (PES) and vibrational bound states for the ground electronic state of $Li_2^+ Kr$. The PES was calculated in Jacobi coordinates at the Restricted Coupled Cluster method RCCSD(T) level of calculation and using aug-cc-pVnZ ($n = 4$ and 5) basis sets. Afterward, this PES is extrapolated to the complete basis set (CBS) limit for correction. The obtained interaction energies were, then, interpolated numerically using the reproducing kernel Hilbert space polynomial (RKHS) approach to produce analytic expressions for the 2D-PES. The analytical PES is used to solve the nuclear Schrodinger equation to determine the bound states' eigenvalues of $Li_2^+ Kr$ for a $J = 0$ total angular momentum configuration and to understand the effects of orientational anisotropy of the forces and the interplay between the repulsive and attractive interaction within the potential surface. In addition, the radial and angular distributions of some selected bound state levels, which lie below, around, and above the T-shaped 90° barrier well, are calculated and discussed. We note that the radial distributions clearly acquire a more complicated nodal structure and correspond to bending and stretching vibrational motions "mode" of the Kr atom along the radial coordinate, and the situation becomes very different at the highest bound states levels with energies higher than the T-shaped 90° barrier well. The shape of the distributions becomes even more complicated, with extended angular distributions and prominent differences between even and odd states.

Keywords: RCCSD(T) method; RKHS interpolation; vibrational quantum bound states; radial and angular distributions



Citation: Saidi, S.; Mabrouk, N.; Dhiflaoui, J.; Berriche, H. Structural, Spectroscopic, and Dynamic Properties of $Li_2^+ (X^2 \Sigma_g^+)$ in Interaction with Krypton Atom.

Molecules **2023**, *28*, 5512. <https://doi.org/10.3390/molecules28145512>

Academic Editor: Igor Reva

Received: 23 May 2023

Revised: 10 July 2023

Accepted: 17 July 2023

Published: 19 July 2023



Copyright: © 2023 by the authors. Licensee MDPI, Basel, Switzerland. This article is an open access article distributed under the terms and conditions of the Creative Commons Attribution (CC BY) license (<https://creativecommons.org/licenses/by/4.0/>).

1. Introduction

Over the past few decades, rare gas media and matrices have attracted great attention as fascinating environments for investigating the spectroscopy, structure, and dynamics embedded inside or deposited on their surfaces. Rare gas atoms are renowned as non-reactive solvents [1–3] and provide a model system for both theoretical and experimental investigations involving a large number of atoms, representing a high degree of freedom. Furthermore, as their condensates exhibit transparency across a broad range of radiation energies, inert gas matrices offer the advantage of selectively exciting chromophores embedded within them. Considering the case of alkali atoms, these matrices have proven useful in studying the spectroscopy, photo-association, and optical absorption spectra of alkali atoms [2,4,5]. Matrix isolation spectroscopy (MIS) is a widely utilized technique in various research fields, particularly for investigating alkali atoms in solid matrices of

noble gases at cryogenic temperatures [6], as well as studying the high-spin states of alkali molecules [7–10].

The exceptional characteristics of rare gas atoms, combined with the growing interest in studying alkali atoms following the observation of the photo-association of cold alkali atoms and their Bose–Einstein condensation [4], have made the interaction between alkali atoms and rare gas atoms a prominent research theme in numerous studies. Notably, both experimental and theoretical works [4,11–29] have been dedicated to exploring the interaction between alkali neutral/ionic dimers and rare gas atoms. The primary focus of these studies is calculating the potential energy surfaces (PES) of dimers with a single rare gas atom. Subsequently, these surfaces are fitted with suitable analytical forms, which are then utilized for dynamic studies such as bound state calculations, as well as for structural investigations, such as the geometric minimization of large-sized clusters to examine the micro-solvation of the embedded dimers.

From an experimental perspective, recent work by Kristensen et al. [21] introduced the use of Coulomb explosion induced by an intense fs laser pulse as a means of studying alkali dimers (Rb_2 , Li_2 , Na_2 and K_2) on He droplets. They demonstrated that Coulomb explosion enables the determination, within a single measurement, of whether alkali dimers, Ak_2 , on the surface of He nanodroplets, are formed in either the $X^1 \Sigma_g^+$ ground state or in the lowest-lying triplet state $a^3 \Sigma_u^+$. Fuchs et al. [19] experimentally investigated the collisional cross-sections of lithium dimer Li_2 in selected vibrational states with He and Kr atoms. An der Lan et al. [13] presented high-resolution mass spectra of $Na^+ He_n$, $K^+ He_n$, $Na_2^+ He_n$ and $K_2^+ He_n$, formed via the electron ionization of doped helium droplets. They observed two distinct anomalies in $Na_2^+ He_n$ clusters at $n = 2$ and $n = 6$.

On the theoretical side, Guillon et al. [11] recently performed diffusion and path integral quantum Monte Carlo studies of the rubidium dimer in its ground triplet state in a helium environment [11]. The focus of their work was the influence of helium atoms on the rotational motion of the Rb_2 dimer. Alharzali et al. [12,22], Bodo et al. [17,23,29], and Marinetti et al. [26] focused on the structure and dynamic properties of cationic dimers (Li_2^+ , Na_2^+ and K_2^+) interacting with helium atoms using RCCSDT and Post Hartree–Fock approaches. They found that for all three species, the helium atom preferred linear attachment to the cationic dimer rather than a T-shaped configuration. In the case of larger clusters, they observed the solvation of cationic dimers within the helium clusters. Douady et al. [20] investigated the solvation of ionic sodium dimers in argon clusters. They confirmed that the cationic dimer enters the clusters rather than resides at the surface. This behavior is attributed to the relatively strong interaction between the closed-shell neon atoms and the cationic dimer. Zanuttini et al. [15] employed a pseudopotential technique and molecular dynamics with a surface hopping approach to investigate the structure and optical absorption of Li_2^+ , Na_2^+ , and K_2^+ alkali dimers in neon clusters. They concluded that the modification of the PES by surrounding neon atoms indicated the strong confinement of the lowest energy states of the three cationic dimers. Saidi et al. [18] explored the structure and stability of the lithium dimer with xenon atoms, while Ghanmi et al. [25] performed a structural, energetic, and spectroscopic study of the potassium cationic dimer interacting with rare gas atoms, such as Ar , Kr , and Xe .

In this paper, we present a computational investigation of the structure and dynamics of the lithium cationic Li_2^+ dimer interacting with a Kr atom. Section 3 provides an overview of the computational method details employed in this study, including the basis sets and extrapolation schemes used to calculate the interaction energies of the $Li_2^+ Kr$ complex. This potential energy surface is, then, employed in the calculation of bound state levels after being reproduced via the RKHS method [30]. The results are presented and discussed in Section 2. The used methodologies are detailed in Section 3.

2. Results and Discussion

The resulting interaction energies were employed to calculate the spectroscopic constants Re and De for each configuration corresponding to angles ranging from 0° to 90° .

The found equilibrium distances (Re) and the well depths (De) for the Li_2^+Kr using the RCCSD(T) method and different basis are summarized in Table 1.

Table 1. Equilibrium distance (Re) and the well depth (De) for the Li_2^+Kr using the RCCSD(T) method and different basis sets. Re are in Å and De in cm^{-1} .

Basis Set	AVQZ		AV5Z		CBS[Q5]	
	Re	De	Re	De	Re	De
θ ($^\circ$)						
0	4.22	1730	4.17	1737	4.17	1745
10	4.14	1715	4.13	1722	4.13	1730
20	4.01	1698	4.01	1706	4.01	1715
30	3.88	1564	3.87	1572	3.87	1580
40	3.69	1391	3.68	1400	3.68	1408
50	3.52	1122	3.51	1130	3.51	1139
60	3.45	775	3.45	783	3.44	791
70	3.61	479	3.60	485	3.59	492
80	3.85	340	3.83	346	3.82	353
90	3.93	304	3.92	310	3.90	317

From Table 1, it can be observed that the well depth of the Li_2^+Kr complex at all geometries increases when moving from AVQZ to CBS[Q5]. Taken the example of their collinear arrangement ($\theta = 0^\circ$) De varies from 1730 cm^{-1} to 1737 cm^{-1} for the RCCSD(T)/AVQZ and RCCSD(T)/AV5Z calculations, respectively. However, upon computing the RCCSD(T)/CBS[Q5] using the RCCSD(T)/AVQZ/AV5Z results, the energies increase to 1745 cm^{-1} . This indicates that the RCCSD(T)/CBS[Q5] energies are lower than those obtained from the RCCSD(T)/AVQZ and RCCSD(T)/AV5Z estimations.

A comparison of the RCCSD(T)/AVQZ, RCCSD(T)/AV5Z, and RCCSD(T)/CBS[Q5] extrapolations is depicted in Figure 1, considering all angular orientations. It can be observed that despite a slight shift between the three curves, the energies obtained from the RCCSD(T)/AVQZ/AV5Z calculations are lower than those from the RCCSD(T)/CBS[Q5] extrapolation. Therefore, the RCCSD(T)/CBS[Q5] energies will be considered as reference data in the subsequent sections.

It is worth noting that as the angle increases from $\theta = 0^\circ$ to $\theta = 90^\circ$, the well depth decreases from $De = 1745\text{ cm}^{-1}$ to $De = 317\text{ cm}^{-1}$. This implies that the linear configuration is the most stable. This arrangement can be explained by considering that the Li_2^+ dimer in its ground state can be approximated as two closed-shell Li^+ cores with an electron cloud located between them. Consequently, the short-range repulsion between the electron and the krypton atom, combined with the attraction between the two Li^+ cores and the closed-shell Kr atom, favors the positioning of the krypton atom at the ends of the dimer.

It is important to mention that the interaction of the lithium dimer with noble gas atoms has been extensively studied recently due to its simplicity and significance. Studies conducted by Zanuttini et al. [15] and Alharzali et al. [22] have contributed to this research area. In Table 2, we have grouped the findings from these studies with the current results obtained for Li_2^+Kr . It is evident that the repulsive interactions decrease in their impact compared to the attractive long-range interaction forces when moving from Li_2^+He to Li_2^+Kr . In fact, the well depths in their linear configuration increase from 380 cm^{-1} for Li_2^+He to 700 cm^{-1} for Li_2^+Ne and further to 1745 cm^{-1} for Li_2^+Kr . Hence, the interactions of the three rare gas atoms (He , Ne , Kr) with the lithium dimer become increasingly more attractive, exhibiting larger attractive cores as one moves from helium to krypton.

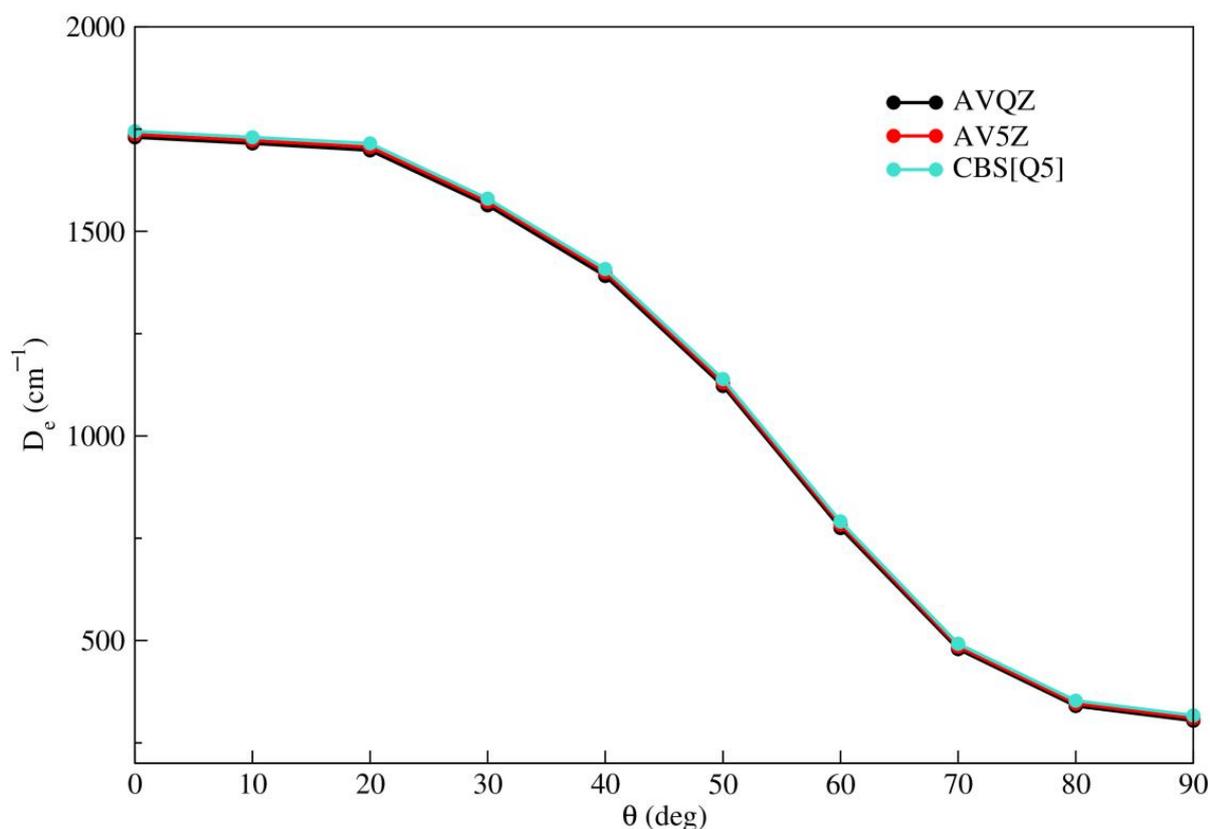


Figure 1. Minimum energy values obtained from RCCSD(T) level of theory using the AVQZ and AV5Z basis sets and CBS extrapolation as a function of θ (deg).

Table 2. Trends of equilibrium distances (R_e) and the well depths (D_e) for $Li_2^+(X^2\Sigma_g^+)$ alkali dimer in interaction with He, Ne and Kr rare gas atoms. R_e are in Å and D_e in cm^{-1} .

Complex	R_e	D_e	Method/References
$Li_2^+(X^2\Sigma_g^+)He$	3.53	342	RCCSD(T)/CBS[Q5] [24]
$Li_2^+(X^2\Sigma_g^+)Ne$	-	700	Pseudopotential [15]
$Li_2^+(X^2\Sigma_g^+)Kr$	4.01	1745	RCCSD(T)/CBS[Q5]

The RKHS potential curves together with the ab-initio RCCSD(T)/CBS[Q5] interaction energies along the R coordinate for each θ angle from 0° to 90° are illustrated in Figure 2. A good agreement is observed between the RKHS potential and the RCCSD(T)/CBS[Q5] calculations for all orientations.

To further verify the quality of the fit we compute the relative error $\Delta E(\%)$ between the original ab initio RCCSD(T)/CBS[Q5] and the values of the RKHS potential for all orientations. The results are presented in Table 3. We note that the relative error does not exceed 0.175%.

The angular minimum energy path for all configurations of the Li_2^+Kr complex, obtained from the RKHS and the RCCSD(T)/CBS[Q5] potentials, are depicted in Figure 3. Several observations could be concluded from the behavior of the angular minimum energy path. Firstly, the RKHS fitting method demonstrates good performance in describing the ab-initio RCCSD(T)/CBS[Q5] PES for all orientations. Secondly, as the configuration changes from linear to T-shaped, the attractive effect decreases compared to the increasing potential values. Furthermore, there is a non-equally spaced energy between successive orientations, and the differences become more pronounced at angles between 30° and 60° . This phenomenon can be attributed to the emergence of repulsive effects, which become

more important than the attractive ones as the *Kr* atom approaches the center of mass of the ionic dimer Li_2^+ .

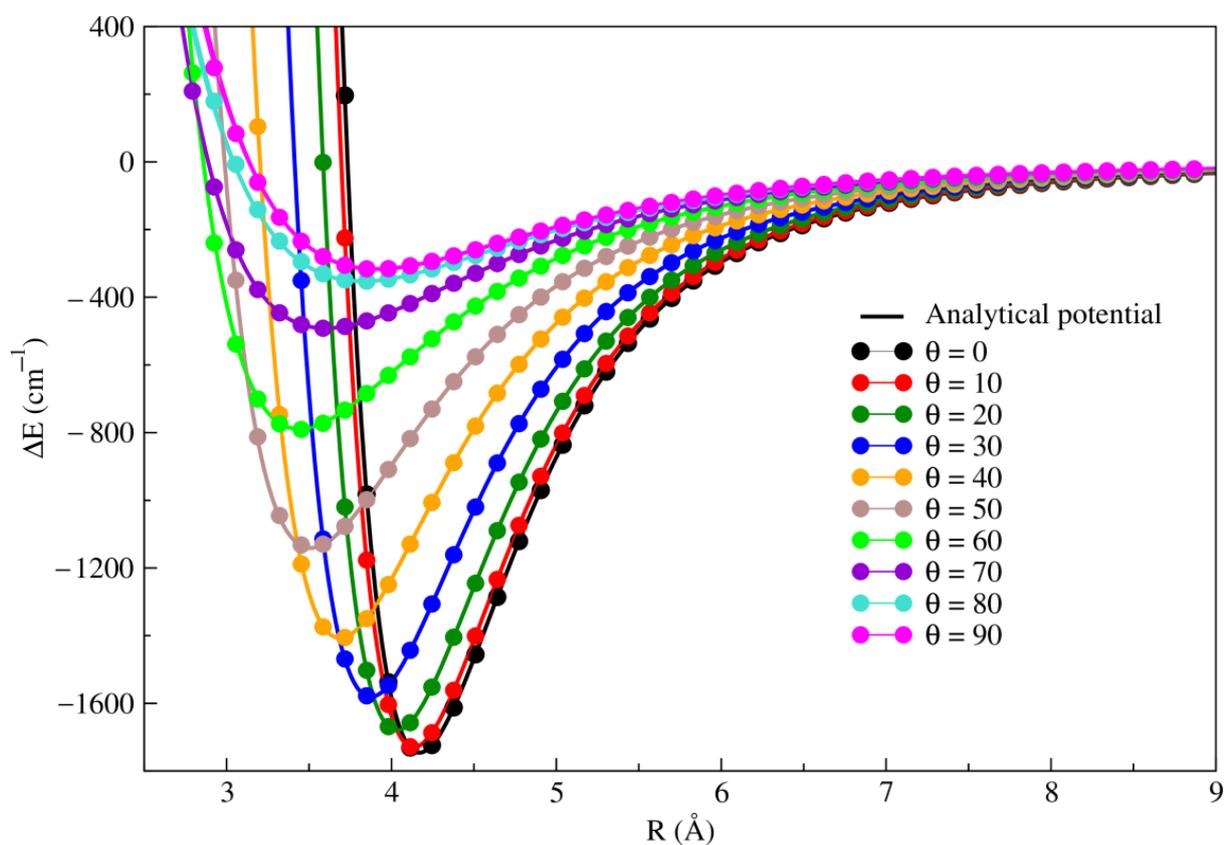


Figure 2. Comparison between the 2D-PES RKHS analytical fitting and the RCCSD(T)/CBS(Q5) ab-initio surfaces of Li_2^+Kr .

Table 3. The equilibrium distance (Re) and the well depth (De) of Li_2^+Kr complex obtained with the RCCSD(T)/CBS[Q5] and the RKHS method for θ ranging from 0° to 90° .

θ ($^\circ$)	RCCSD(T)/CBS[Q5]		RKHS		ΔE (%)
	Re (\AA)	De (cm^{-1})	Re (\AA)	De (cm^{-1})	
0°	4.17	1745	4.17	1746	0.057
10°	4.13	1730	4.13	1730	0.000
20°	4.01	1715	4.03	1715	0.000
30°	3.87	1580	3.87	1581	0.063
40°	3.68	1408	3.68	1407	0.071
50°	3.51	1139	3.50	1141	0.175
60°	3.44	791	3.44	790	0.126
70°	3.59	492	3.59	492	0.000
80°	3.82	353	4.82	353	0.000
90°	3.90	317	4.90	317	0.000

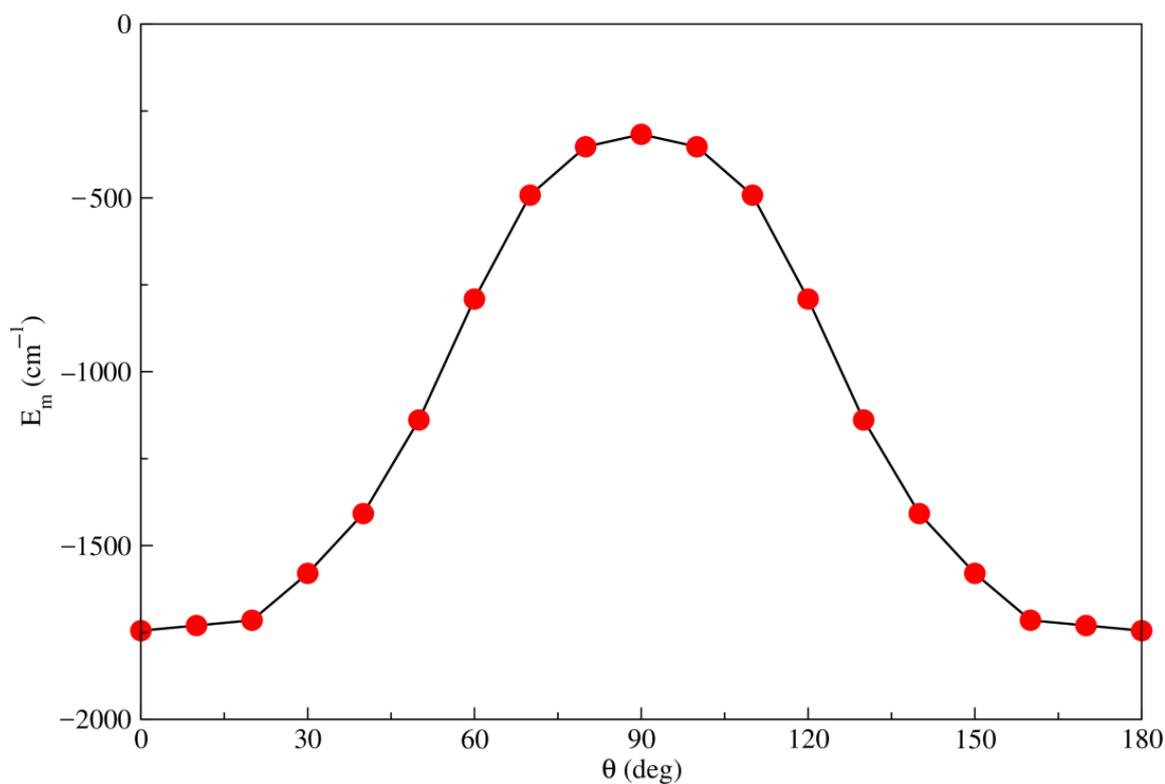


Figure 3. Minimum energy values obtained from the RKHS analytical interpolation with the corresponding RCCSD(T)/CBS[Q5] interaction energies as a function of angular orientations θ .

Figure 4 displays a two-dimensional contour plot in the (θ, R) plane, representing the fitted RCCSD(T)/CBS[Q5] potential of $\text{Li}_2^+ \text{Kr}$. In this plot, we observe the presence of two symmetrical minima, which correspond to the linear orientations. These minima have a well depth of 1746 cm^{-1} located at an internuclear distance $R = 4.17 \text{ \AA}$.

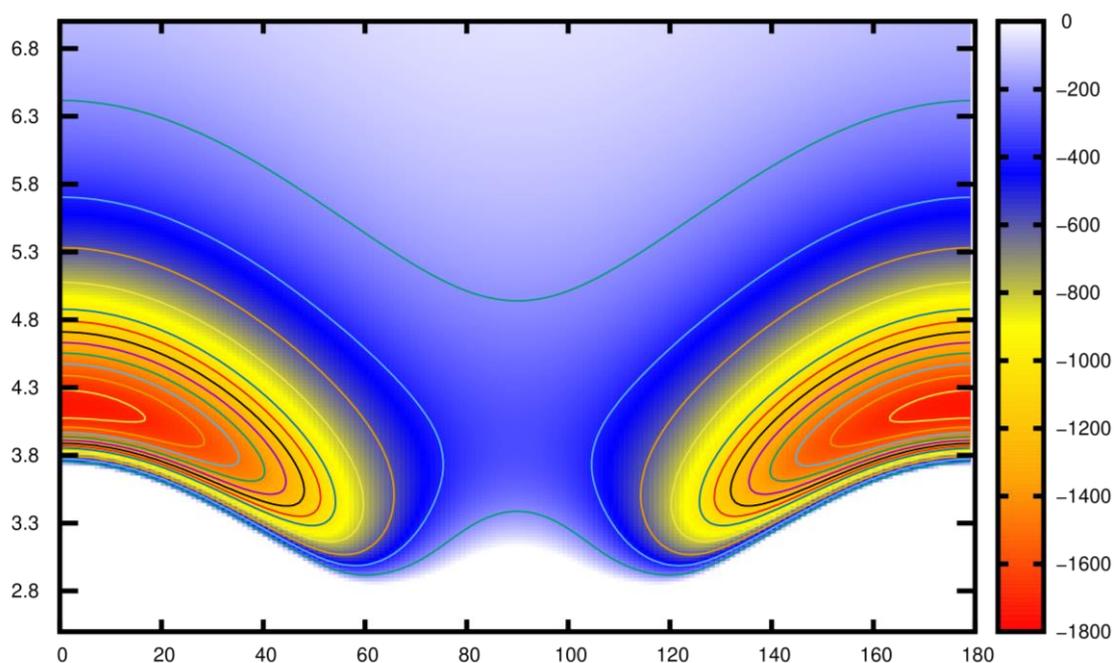


Figure 4. Two-dimensional contour plot of the analytical potential using the RKHS analytical interpolation in the (θ, R) plane.

Based on the parity of the quantum number j and the total wave function, with respect to the exchange of the two identical Li^+ atoms, the results will be divided into two types (even/odd). It is expected to obtain a significant number of bound states when we consider the energy of the linear arrangement, 1746 cm^{-1} . In our calculations, we obtained 272/260 states for even and odd symmetry, respectively. In Table 4, we list only the energies of the lowest bound states and some selected states that lie below, around, and above the T-shaped 90° barrier well for both even and odd symmetry. The other values are reported in the Supplementary Materials.

Table 4. Vibrational energies (in cm^{-1}) of the lowest and some selected bound states ($J = 0$) that lie below, around, and above the T-shaped 90° barrier well of the $Li_2^+ Kr$ complex.

n	$j = \text{even/odd}$
0	−1633.440613/−1633.440613
1	−1562.620803/−1562.620803
2	−1492.719188/−1492.719188
3	−1481.617393/−1481.617393
4	−1422.606448/−1422.606448
5	−1405.444466/−1405.444466
6	−1352.065571/−1352.065571
7	−1339.178576/−1339.178576
8	−1331.488421/−1331.488421
9	−1281.201931/−1281.201931
10	−1263.686846/−1263.686846
114	−335.655266/−335.655351
115	−335.190171/−335.190064
116	−326.764948/−326.764891
117	−322.031270/−322.031221
118	−318.591568/−318.591555
119	−316.216834/−316.216848
240	−25.004181/−18.144948
250	−14.747134/−8.049236
260	−6.805258/−0.477206
272	−0.298997/-

From the full table, we observe that up to $n = 78$, the even and odd vibrational states are fully degenerated. However, the higher-lying even and odd parity states exhibit different energies, and the energy difference becomes more pronounced for levels above the potential barrier in the T-shaped configuration. Furthermore, Table 4 shows that the energy of the lowest vibrational level is -1666 cm^{-1} , with a zero-point energy (ZPE) of about 88 cm^{-1} , which corresponds to only 5% of the well depth.

Figures 5–7 illustrate the radial and angular distributions for selected states lying below, around, and above the T-shaped 90° barrier for both even and odd symmetries, respectively.

For the lowest bound states wave functions, such as $n = 0-5$, there is a minimal angular population in the region of the 90° barrier well, indicating a strong localization of the Kr atom in the collinear geometric configuration. The ground state wave function exhibits nodal structures, with angular distributions symmetrically positioned at $\theta = 0^\circ$ and 180° , while the radial distribution peaks at $R = 4.3\text{ \AA}$, which is close to the equilibrium intermolecular distance of the linear geometry. The low-lying excited states can be attributed to the stretching and bending motions of the Kr atom. The bending motions correspond to the $n = 1$ and $n = 2$ levels, as evidenced by the presence of nodes in their angular distributions. The $n = 3$ state clearly displays nodes in its radial distribution, indicating stretching motions.

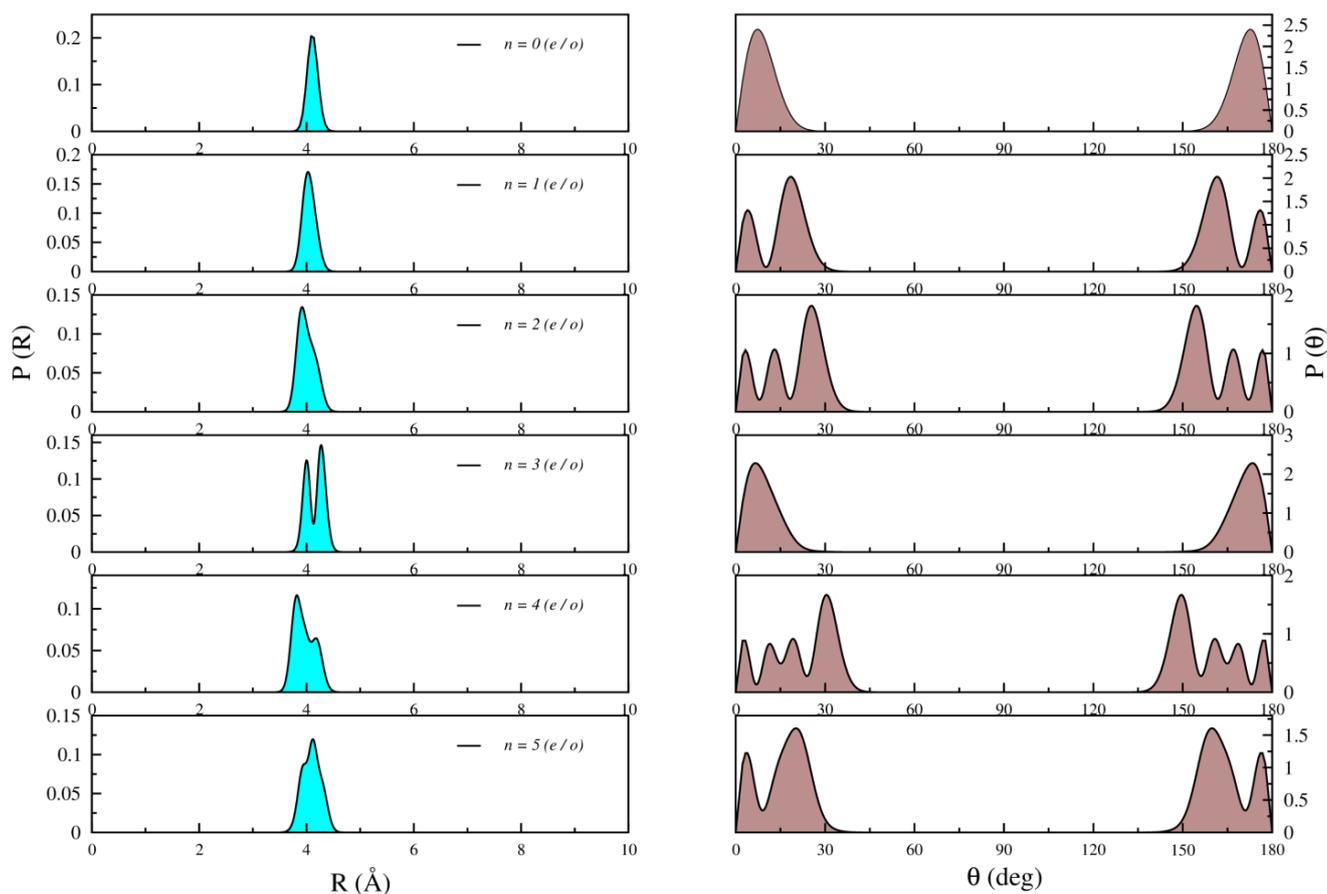


Figure 5. Radial (left) and angular (right) distributions of both even (e) and odd (o) states of Li_2^+Kr complex for $n = 0-5$.

For the bound states that lie around the 90° barrier well (just below and above it), such as $n = 114-119$, Figure 6 illustrates more complex nodal patterns in both the radial and angular distributions. These distributions extend to larger R distances and configurations, and some population appears in the region of the T-shaped barrier. This behavior can be associated with the mixed stretch-bending vibrational “mode” of the Kr atom along the radial and angular coordinates. Additionally, noticeable differences between even and odd states become apparent in both radial and angular probability distributions for levels $n = 114$ and above.

For the highest bound states, as depicted in Figure 7, the shape of the distributions becomes even more complicated, with extended angular distributions and prominent differences between even and odd parity states.

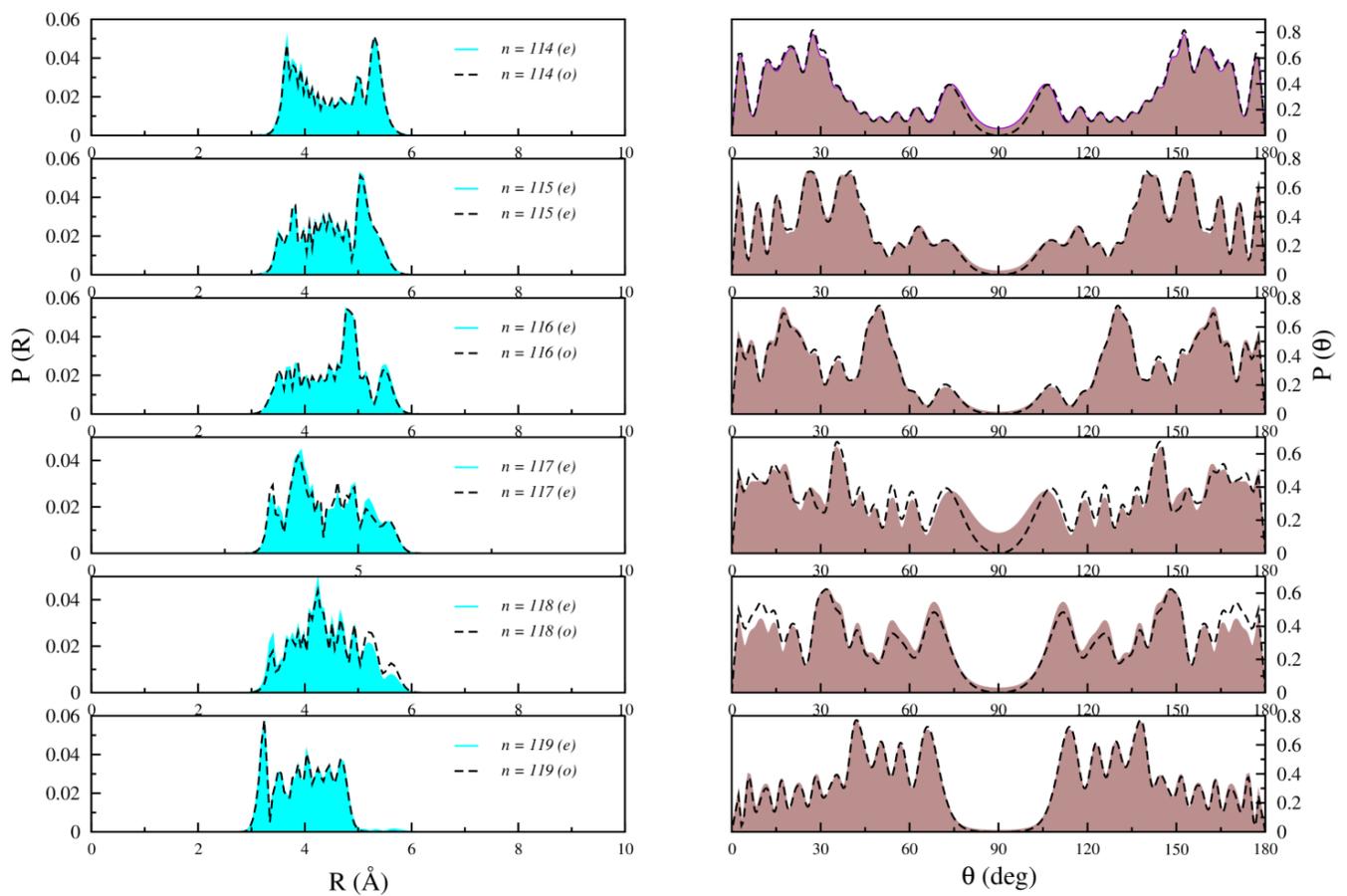


Figure 6. Computed radial (left) and angular (right) distributions for $n = 114$ to $n = 119$ bound states of $Li_2^+ Kr$ for both even (right line) and odd (dashed line) states.

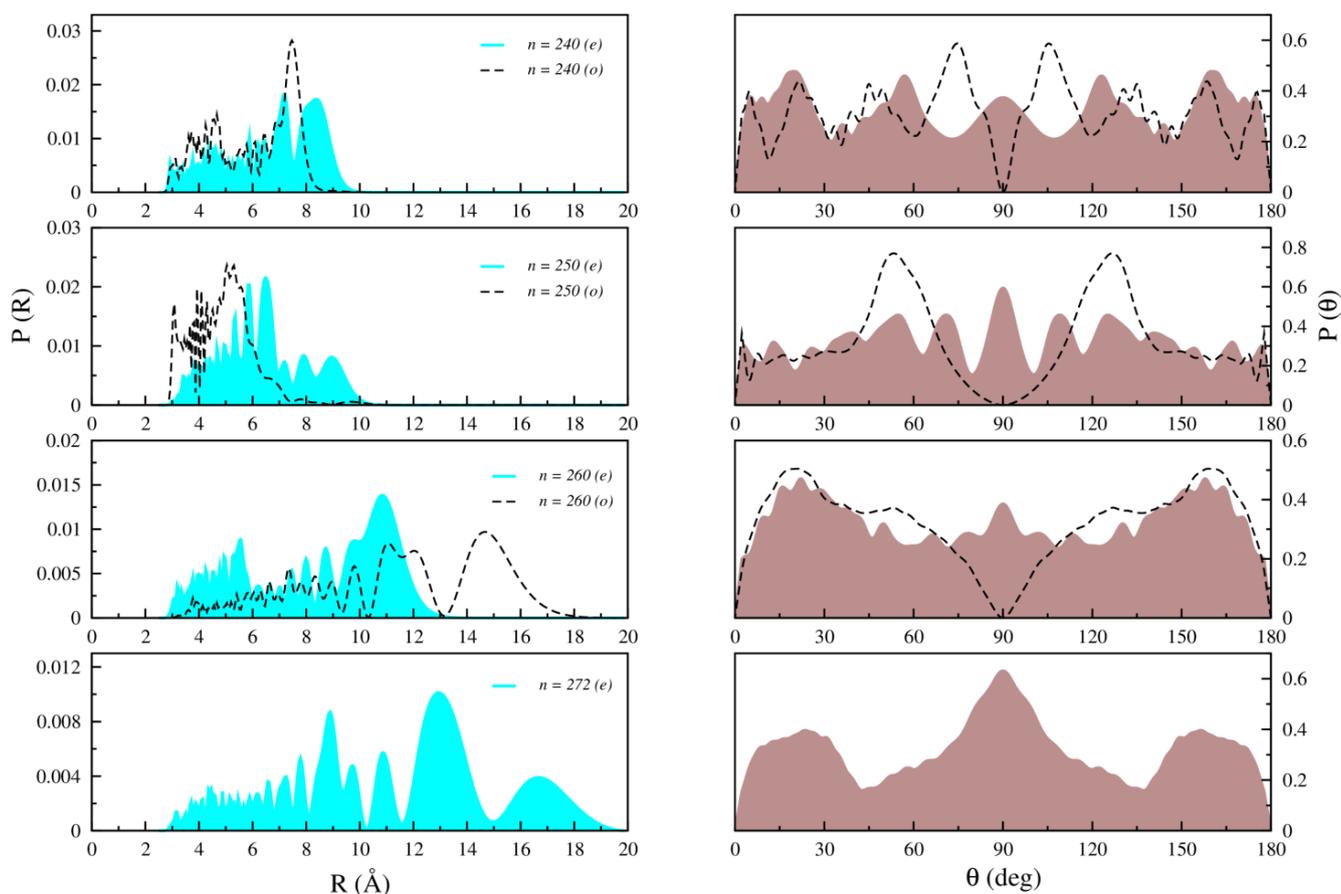


Figure 7. Computed radial (left) and angular (right) distributions for $n = 240, 250, 260,$ and 272 bound states of $Li_2^+ Kr$ for both even (right line) and odd (dashed line) states.

3. Computational Methods

3.1. Potential Energy Surface

3.1.1. Ab Initio Calculations

To describe the intermolecular interactions between the krypton atom and the diatomic molecule Li_2^+ , Jacobi coordinates (r, R, θ) were employed, as illustrated in Figure 8. Here, r represents the equilibrium distance of Li_2^+ , R denotes the distance between the center of mass of Li_2^+ and the krypton atom, and θ represents the Jacobi angle between the vectors r and R .

The calculations were carried out by keeping the diatomic Li_2^+ frozen at its experimental equilibrium distance, $r_e = 3.11$ Å, obtained via pulsed optical–optical double resonance spectroscopy [16]. The internuclear distance R was varied from 2 Å to 20 Å with an irregular step, while the Jacobi angle θ was varied from 0° to 90° with a step of 10°.

In this present work, the Molpro 2010 Package [31] was used to perform all ab initio calculations. The potential energy surface of $Li_2^+ Kr$ was calculated using the restricted Hartree–Fock calculation followed by a single-reference restricted open-shell coupled cluster method with single, double, and no-iterative triple excitations RCCSD(T) method [32,33]. This method is known for its high level of accuracy in describing electronic correlation effects.

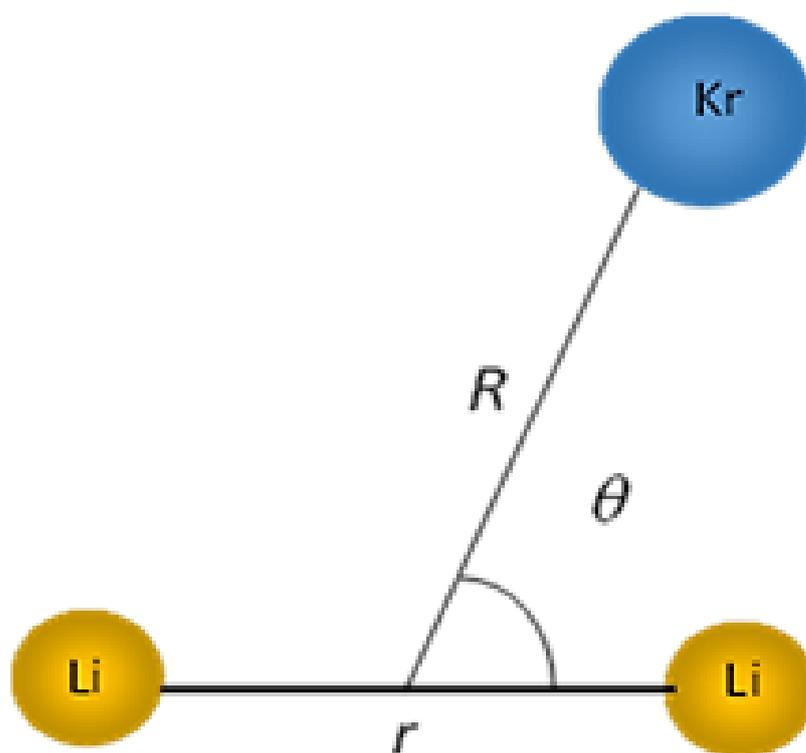


Figure 8. Jacobi coordinates (r , R , θ) of $Li_2^+ Kr$ complex.

To assess the validity of the RCCSD(T) method employed, correlation factors D_1 and T_1 were calculated for all investigated geometries. The T_1 and D_1 diagnostics, as defined in reference [34], were used for this purpose. The T_1 diagnostic is given by $T_1 = \sqrt{\frac{\sum_i^{occ} \sum_a^{vir} (t_i^a)^2}{n}}$, where n is the number of electrons, and (t_i^a) represents the single excitation amplitudes. The D_1 diagnostic is defined as $D_1(CCS) = \|T\|_2$, where $\|T\|_2$ is the Euclidean norm of the matrix T calculated from the CCSD wave function. The results of the calculations are summarized in Table 5.

Table 5. T_1 and D_1 diagnostics for $Li_2^+ Kr$ complex around the equilibrium positions.

θ ($^\circ$)	T_1	D_1
0 $^\circ$	0.00337867	0.00727516
10 $^\circ$	0.00346117	0.00761505
20 $^\circ$	0.00973038	0.0087485
30 $^\circ$	0.00407142	0.00994863
40 $^\circ$	0.00455682	0.01166247
50 $^\circ$	0.00510274	0.01350919
60 $^\circ$	0.00546915	0.01473819
70 $^\circ$	0.00508529	0.01356188
80 $^\circ$	0.00430078	0.01101894
90 $^\circ$	0.00399547	0.00998581

The obtained results indicate that both, T_1 and D_1 factors, are below the threshold values of 0.02 and 0.025, respectively, as suggested by Lee and Taylor [34]. This confirms the suitability of the mono-configurational approach RCCSD(T) chosen for this study.

For both Li and Kr atoms, the aug-cc-pVn ($n = 4, 5$) Z basis sets [35] were utilized. The energies obtained were then extrapolated to the complete basis set (CBS) limit using a two-parameter expression [36] applied to the correlation energies as follows: $E_n = E_{CBS} + \frac{A}{n^3}$. Here, $n = 4$ and 5 correspond to AVQZ and AV5Z basis sets, respectively, E_n is the computed total energy corresponding to that basis set, E_{CBS} is the CBS extrapolated energy, and A

is a fitting parameter. The energies from the AVQZ and AV5Z basis set calculations were employed for the extrapolation.

In the calculations of the internuclear interaction energies $V(r_e, R, \theta)$ between Kr and Li_2^+ , the standard counterpoise method of Boys and Bernardi [37] was used to correct the basis set superposition error (BSSE) at all configurations. This correction is applied according to the equation: $V(r_e, R, \theta) = E_{Li_2^+Kr}(r_e, R, \theta) - [E_{Li_2^+}(r_e, R, \theta) + E_{Kr}(r_e, R, \theta)]$, where $E_{Li_2^+Kr}(r_e, R, \theta)$ represents the total energy of the complex, and $E_{Li_2^+}(r_e, R, \theta)$ and $E_{Kr}(r_e, R, \theta)$ are the energies of the Li_2^+ and Kr monomers, respectively, calculated in the full basis set of the system.

3.1.2. Analytical Representation of the Ab Initio Surface

In order to conduct dynamic investigations, an analytical representation of the ab-initio potential energy surface (PES) is necessary. In this study, the 2D-PES of Li_2^+Kr was interpolated using the reproducing kernel Hilbert space (RKHS) procedure developed by Ho and Rabitz [30]. The two-dimensional potential function is given by the following:

$$V(r_e, R, \theta) = \sum_{i=1}^{N_R} \sum_{j=1}^{N_\theta} v_{ij} q_1^{2,3}(R_i, R) q_2(y_j, y) \quad (1)$$

$$V^{RKHS}(r_e, R, \theta) = \sum_{i=1}^{N_R} \sum_{j=1}^{N_\theta} V_{ij} q_1^{n,m}(R_i, R) q_2(y_j, y) \quad (2)$$

In the above equations, $y = \cos\theta$, and N_R and N_θ represent the number of calculated ab initio energies in the R and θ coordinates, respectively. The v_{ij} coefficients are determined by solving Equation (1), where $V(r_e, R_i, \theta_j)$ represents the ab initio RCCSD(T)/CBS[Q5] energy at each (R_i, θ_j, r_e) grid point.

The one-dimensional distance-like $q_1^{n,m}$ and angle-like q_2 reproducing kernels are expressed as follows:

$$q_1^{n,m}(x, x') = n^2 x_{>}^{-(m+1)} B(m+1, n) {}_2F_1\left(-n+1, m+1; n+m+1; \frac{x_{<}}{x_{>}}\right) \quad (3)$$

$$q_2(y, y') = \sum_l \frac{(2l+1)}{2} P_l(y) P_l(y') \quad (4)$$

where $x_{>}$ and $x_{<}$ refer to the maximum and minimum values of x and x' , respectively. B and ${}_2F_1$ represent the Beta and Gauss hypergeometric functions [38], respectively. P_l denotes the Legendre polynomials with $l = 0, 2, 4, 6, 8, 10, 12$.

Based on the previous work by Alharzali et al. [12,24] for similar interactions (Li_2^+He , Na_2^+He and K_2^+He), it is assumed that the 2D-PES $V(R, \theta)$ of Li_2^+Kr is a smooth function with derivatives up to the second order, $n = 2$, in both R and θ . Additionally, to account for the dominant dispersion interaction between the Kr atom and the ionic dimer Li_2^+ , a weighting factor $w(x) = x^{-m}$ with $m = 3$ is introduced.

3.2. Bound States Calculation

The calculations of bound states energies and wave functions are crucial steps for studying dynamics, such as vibrational predissociation and photoionization. In this study, variational quantum bound states calculations were performed using the fitted RKHS potential energies. The bound states energies were obtained by diagonalizing the Hamiltonian expressed in Jacobi coordinates, as follows:

$$\hat{H} = -\frac{\hbar^2}{2\mu_1} \frac{\partial^2}{\partial^2 R} + \frac{\hat{j}^2}{2\mu_2 r_e^2} + \frac{\hat{l}^2}{2\mu_1 R^2} + V(r_e, R, \theta) \quad (5)$$

In the above equation, $\frac{1}{\mu_1} = \frac{1}{m_{Kr}} + \frac{1}{2m_{Li_2^+}}$ and $\frac{1}{\mu_2} = \frac{1}{m_{Li}} + \frac{1}{m_{Li}}$ are the reduced masses of the Li_2^+Kr complex and the Li_2^+ dimer, respectively, where m_{Li} and m_{Kr} are the atomic

masses of *Li* and *Kr* atoms. \hat{j} and \hat{l} represent the angular momenta associated with the vectors r (rotational momentum of the dimer) and R (orbital angular momentum), respectively. The sum of these angular momenta gives the total angular momentum \hat{J} , which was taken as zero in these calculations. $V(r_e, R, \theta)$ represents the fitted 2D-RKHS potential, where r_e is the fixed equilibrium bond length of the diatomic Li_2^+ .

In the framework of zero total angular momentum $J = 0$, a product of radial and angular basis functions is used to represent the Hamiltonian. For the angular coordinate, the employed basis function is an orthonormalized Legendre polynomial $P_j \cos(\theta)$, where j ranges up to 40 for even and odd symmetry, respectively. For the radial coordinate, a discrete variable representation (DVR) basis set, based on particle in a box eigenfunctions [39], was employed. A 300 points DVR was used over a range from $R = 2.5$ to 20 \AA , and a convergence criterion of 10^{-6} was established.

4. Conclusions

In this study, we conducted structural and dynamic investigations of the lithium cationic dimer, Li_2^+ , interacting with the *Kr* atom. The two-dimensional potential energy surface was computed using the RCCSD(T) method and the aug-cc-pVnZ ($n = 4, 5$) basis sets, and then extrapolated to its CBS[Q5] limit. Subsequently, the RKHS method was employed for numerical interpolation to generate the RCCSD(T)/CBS[Q5] analytical potentials. Remarkably, this interpolation method accurately matched the numerical curves for all orientations used to determine the two main spectroscopic constants, Re and De , for each configuration. The analysis of the results revealed that the linear configuration, where the krypton atom is linearly attached to the lithium dimer Li_2^+ is found to be more stable than the T-shaped configuration.

The fitted 2D-RKHS potential was employed to calculate the vibrational bound state energies, resulting in a significant number of bound states as expected. It is worth noting the limited contribution of the zero-point energy (ZPE), which accounts for only 5% of the well depth. Additionally, the system exhibits very large amplitude motions, both as stretching and bending modes in its lower states, but rapidly transformed into non-regular features of their wave function in the excited states.

The obtained bound states, along with the potential energy surfaces, have various applications. They can be utilized to investigate vibrational predissociation or photoionization processes of Li_2Kr complex. Furthermore, an analytical fitting of the obtained 2D-PES can be employed to explore the micro-solvation process of the lithium cationic dimer in a krypton matrix. These valuable results could assist in experimental settings and help in the interpretation of observations.

Supplementary Materials: The following supporting information can be downloaded at: <https://www.mdpi.com/article/10.3390/molecules28145512/s1>, Table S1: Vibrational energies (in cm^{-1}) of all bound states ($J = 0$) of the Li_2^+Kr complex.

Author Contributions: Conceptualization, S.S. and J.D.; methodology, S.S., J.D. and N.M.; investigation, H.B., S.S. and J.D.; data curation, N.M.; writing—original draft preparation, S.S. and J.D.; writing—review and editing, S.S., J.D. and H.B.; project administration, S.S. All authors have read and agreed to the published version of the manuscript.

Funding: This work was supported by the Deputyship for Research and Innovation, Ministry of Education in Saudi Arabia (Project No. IF-PSAU-2021/01/18968).

Institutional Review Board Statement: Not applicable.

Informed Consent Statement: Not applicable.

Data Availability Statement: The data that support the findings of this study are available from the corresponding author upon reasonable request.

Acknowledgments: The authors extend their appreciation to the Deputyship for Research & Innovation, Ministry of Education in Saudi Arabia for funding this research work through the project number (IF-PSAU-2021/01/18968).

Conflicts of Interest: The authors declare no conflict of interest.

Sample Availability: Not applicable.

References

1. Liu, Q.; Wang, J.K.; Zewail, A.H. Femtosecond dynamics of dissociation and recombination in solvent cages. *Nature* **1993**, *364*, 427–430. [[CrossRef](#)]
2. Apkarian, V.A.; Schwentner, N. Molecular Photodynamics in Rare Gas Solids. *Chem. Rev.* **1999**, *99*, 1509–1514. [[CrossRef](#)]
3. Greenblatt, B.J.; Zanni, M.T.; Neumark, D.M. Photodissociation of $I_2^- Ar_n$ Clusters Studied with Anion Femtosecond Photoelectron Spectroscopy. *Science* **1997**, *276*, 1675–1678. [[CrossRef](#)]
4. Kokouline, V.; Dulieu, O.; Kosloff, R.; Masnou-Seeuws, F. Mapped Fourier methods for long-range molecules: Application to perturbations in the $Rb_2(O_u^+)$ photoassociation spectrum. *J. Chem. Phys.* **1999**, *110*, 9865–9876. [[CrossRef](#)]
5. Weyhmann, W.; Pipkin, F.M. Optical Absorption Spectra of Alkali Atoms in Rare-Gas Matrices. *Phys. Rev.* **1965**, *137*, A490. [[CrossRef](#)]
6. Braggio, C.; Calabrese, R.; Carugno, G.; Fiscelli, G.; Guarise, M.; Khanbekyan, A.; Noto, A.; Passante, R.; Rizzuto, L.; Ruoso, G.; et al. Spectroscopy of Alkali Atoms in Solid Matrices of Rare Gases: Experimental Results and Theoretical Analysis. *Appl. Sci.* **2022**, *12*, 6492. [[CrossRef](#)]
7. Rodríguez-Cantano, R.; López-Durán, D.; González-Lezana, T.; Delgado-Barrio, G.; Villarreal, P.; Yurtsever, E.; Gianturco, F.A. Spin-polarized Rb_2 interacting with bosonic He atoms: Potential energy surface and quantum structures of small clusters. *J. Phys. Chem. A* **2012**, *116*, 2394–2404. [[CrossRef](#)] [[PubMed](#)]
8. Higgins, J.; Callegari, C.; Reho, J.; Stienkemeier, F.; Ernst, W.E.; Gutowski, M.; Scoles, G. Helium Cluster Isolation Spectroscopy of Alkali Dimers in the Triplet Manifold. *J. Phys. Chem. A* **1998**, *102*, 4952–4965. [[CrossRef](#)]
9. Schulz, C.P.; Claas, P.; Schumacher, D.; Stienkemeier, F. Formation and Stability of High-Spin Alkali Clusters. *Phys. Rev. Lett.* **2004**, *92*, 013401–013404. [[CrossRef](#)]
10. Vongehr, S.; Kresin, V.V. Unusual pickup statistics of high-spin alkali agglomerates on helium nanodroplets. *J. Chem. Phys.* **2003**, *119*, 11124–11129. [[CrossRef](#)]
11. Guillon, G.; Zanchet, A.; Leino, M.; Viel, A.; Zillich, R.E. Theoretical Study of Rb_2 in He_N : Potential Energy Surface and Monte Carlo Simulations. *J. Phys. Chem. A* **2011**, *115*, 6918–6926. [[CrossRef](#)] [[PubMed](#)]
12. Alharzali, N.; Rodríguez-Segundo, R.; Prosmi, R. Modelling interactions of cationic dimers in He droplets: Microsolvation trends in $He_n K_2^+$ clusters. *Phys. Chem. Chem. Phys.* **2021**, *23*, 7849–7859. [[CrossRef](#)]
13. An der Lan, L.; Bartl, P.; Leidlmair, C.; Jochum, R.; Denifl, S.; Echt, O.; Scheier, P. Solvation of Na^+ , K^+ and their dimers in helium. *Chem. Eur. J.* **2012**, *18*, 4411–4418. [[CrossRef](#)] [[PubMed](#)]
14. Prosmi, R.; Delgado-Barrio, G.; Villarreal, P.; Yurtsever, E.; Coccia, E.; Gianturco, F.A. Structuring a Quantum Solvent around a Weakly Bound Dopant: The $He - Cs_2(^3\Sigma_u)$ Complex. *J. Phys. Chem. A* **2009**, *113*, 14718–14729. [[CrossRef](#)]
15. Zanuttini, D.; Douady, J.; Jacquet, E.; Giglio, E.; Gervais, B. Structure and photoabsorption properties of cationic alkali dimers solvated in neon clusters. *J. Chem. Phys.* **2010**, *133*, 174503. [[CrossRef](#)]
16. Bernheim, R.A.; Gold, L.P.; Tipton, T. Rydberg states of 7Li_2 by pulsed optical-optical double resonance spectroscopy: Molecular constants of $^7Li_2^+$. *J. Chem. Phys.* **1983**, *78*, 3635–3646. [[CrossRef](#)]
17. Bodo, E.; Gianturco, F.A.; Yurtsever, E.; Yurtsever, M. Neutral and ionic dopants in helium clusters: Interaction forces for the $Li_2(a^3\Sigma_u^+) - He$ and $Li_2^+(X^2\Sigma_g^+) - He$ complexes. *Mol. Phys.* **2005**, *103*, 3223–3231. [[CrossRef](#)]
18. Saidi, S.; Ghanmi, C.; Hassen, F.; Berriche, H. Ab initio Study of the Potential Energy Surface and Stability of the $Li_2^+(X^2\Sigma_g^+)$ Alkali Dimer in Interaction with a Xenon Atom. *Prog. Theor. Chem. Phys.* **2012**, *26*, 321–330. [[CrossRef](#)]
19. Fuchs, M.; Toennies, J.P. Scattering of highly vibrationally excited Li_2 from He and Kr . *J. Chem. Phys.* **1986**, *85*, 7062–7076. [[CrossRef](#)]
20. Douady, J.; Jacquet, E.; Giglio, E.; Zanuttini, D.; Gervais, B. Solvation of Na_2^+ in Ar_n clusters. I. Structures and spectroscopic properties. *J. Chem. Phys.* **2008**, *129*, 184303–184313. [[CrossRef](#)]
21. Kristensen, H.H.; Kranabetter, L.; Schouder, C.A.; Stapper, C.; Arlt, J.; Mudrich, M.; Stapelfeldt, H. Quantum-State-Sensitive Detection of Alkali Dimers on Helium Nanodroplets by Laser-Induced Coulomb Explosion. *Phys. Rev. Lett.* **2022**, *128*, 93201. [[CrossRef](#)] [[PubMed](#)]
22. Alharzali, N.; Berriche, H.; Villarreal, P.; Prosmi, R. Theoretical Study of Cationic Alkali Dimers Interacting with He : $Li_2^+ - He$ and $Na_2^+ - He$ van der Waals Complexes. *J. Phys. Chem. A* **2019**, *123*, 7814–7821. [[CrossRef](#)] [[PubMed](#)]
23. Bodo, E.; Sebastianelli, F.; Gianturco, F.A.; Yurtsever, E.; Yurtsever, M. Ab initio quantum dynamics with very weak van der Waals interactions: Structure and stability of small $Li_2^+(X^1\Sigma_g^+) - (He)_n$ clusters. *J. Chem. Phys.* **2004**, *120*, 9160–9166. [[CrossRef](#)] [[PubMed](#)]
24. Grebenev, S.; Toennies, J.P.; Vilesov, A.F. Superfluidity within a small helium-4 cluster: The microscopic andronikashvili experiment. *Science* **1998**, *279*, 2083–2086. [[CrossRef](#)]
25. Ghanmi, C.; Nakbi, H.; Al-Qarni, H.J.; Alharzali, N.; Berriche, H. Structure, energetics, and spectroscopy of the $K_2^+(X^2\Sigma_g^+)$ interacting with the noble gas atoms Ar , Kr and Xe . *J. Mol. Graph. Model.* **2023**, *120*, 108413–108418. [[CrossRef](#)]

26. Marinetti, F.; Uranga-Piña, L.; Coccia, E.; López-Durán, D.; Bodo, E.; Gianturco, F.A. Microsolvation of Cationic Dimers in ^4He Droplets: Geometries of $A_2^+(He)_N$ ($A = \text{Li}, \text{Na}, \text{K}$) from Optimized Energies. *J. Phys. Chem. A* **2007**, *111*, 12289–12294. [[CrossRef](#)]
27. Stienkemeier, F.; Ernst, W.E.; Higgins, J.; Scoles, G. On the use of liquid helium cluster beams for the preparation and spectroscopy of the triplet states of alkali dimers and other weakly bound complexes. *J. Chem. Phys.* **1995**, *102*, 615–617. [[CrossRef](#)]
28. Albertini, S.; Martini, P.; Schiller, A.; Schöbel, H.; Ghavidel, E.; Ončák, M.; Echt, O.; Scheier, P. Electronic transitions in Rb_2^+ dimers solvated in helium. *Theor. Chem. Acc.* **2021**, *140*, 29–40. [[CrossRef](#)]
29. Bodo, E.; Yurtsever, E.; Yurtsever, M.; Gianturco, F.A. Ionic dimers in He droplets: Interaction potentials for $\text{Li}_2^+ - \text{He}$, $\text{Na}_2^+ - \text{He}$, and $\text{K}_2^+ - \text{He}$ and stability of the smaller clusters. *J. Chem. Phys.* **2006**, *124*, 074320–074332. [[CrossRef](#)]
30. Ho, T.S.; Rabitz, H. A general method for constructing multidimensional molecular potential energy surfaces from ab initio calculations. *J. Chem. Phys.* **1996**, *104*, 2584–2597. [[CrossRef](#)]
31. Werner, H.J.; Knowles, P.J.; Knizia, G.; Manby, F.R.; Schütz, M. Molpro: A general-purpose quantum chemistry program package. *Wiley Interdiscip. Rev. Comput. Mol. Sci.* **2012**, *2*, 242–253. [[CrossRef](#)]
32. Čížek, J. On the Correlation Problem in Atomic and Molecular Systems. Calculation of Wavefunction Components in Ursell-Type Expansion Using Quantum-Field Theoretical Methods. *J. Chem. Phys.* **1966**, *45*, 4256–4266. [[CrossRef](#)]
33. Čížek, J. On the Use of the Cluster Expansion and the Technique of Diagrams in Calculations of Correlation Effects in Atoms and Molecules. In *Advances in Chemical Physics*; LeFebvre, R., Moser, C., Eds.; Wiley: Hoboken, NJ, USA, 1969; Volume 14, pp. 35–89. [[CrossRef](#)]
34. Lee, T.J.; Taylor, P.R. A diagnostic for determining the quality of single-reference electron correlation methods. *Int. J. Quantum Chem.* **1989**, *36*, 199–207. [[CrossRef](#)]
35. Dunning, T.H., Jr. Gaussian basis sets for use in correlated molecular calculations. I. The atoms boron through neon and hydrogen. *J. Chem. Phys.* **1989**, *90*, 1007–1023. [[CrossRef](#)]
36. Schwartz, C. Importance of Angular Correlations between Atomic Electrons. *Phys. Rev.* **1962**, *126*, 1015–1019. [[CrossRef](#)]
37. Boys, S.F.; Bernardi, F. The calculation of small molecular interactions by the differences of separate total energies. Some procedures with reduced errors. *Mol. Phys.* **1970**, *19*, 553–566. [[CrossRef](#)]
38. Abramowitz, M.; Irene, S. *Handbook of Mathematical Functions with Formulas, Graphs, and Mathematical Tables*, 9th ed.; Dover: New York, NY, USA, 1964; p. 470.
39. Prosimiti, R.; Cunha, C.; Villarreal, P.; Delgado-Barrio, G. The van der Waals potential energy surfaces and structures of He-ICl and Ne-ICl clusters. *J. Chem. Phys.* **2002**, *117*, 7017–7023. [[CrossRef](#)]

Disclaimer/Publisher’s Note: The statements, opinions and data contained in all publications are solely those of the individual author(s) and contributor(s) and not of MDPI and/or the editor(s). MDPI and/or the editor(s) disclaim responsibility for any injury to people or property resulting from any ideas, methods, instructions or products referred to in the content.