

# Evaluation of biological activity of new 1,2,4-triazole derivatives containing propionic acid moiety

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Table S1. A summary of the yield and melting point of compounds **3a-3g** obtained under different reaction conditions (methods A-D)

compound	method	solvent	temp.	time	yield	melting point
<b>3a</b>	A	diethyl ether	ambient	2 days	55.39%	197-199 °C
<b>3a</b>	B	toluene	boiling	4 h	69.36%	199-202 °C
<b>3a</b>	C	chloroform	boiling	1 h	29.90%	198-200 °C
<b>3b</b>	A	diethyl ether	ambient	2 days	85.76%	178-180 °C
<b>3b</b>	B	toluene	boiling	1 h	52.94%	181-183 °C
<b>3b</b>	C	chloroform	boiling	1.5 h	44.12%	183-185 °C
<b>3c</b>	A	diethyl ether	ambient	2 days	73.53%	157-160 °C
<b>3c</b>	B	toluene	boiling	1 h	31.58%	157-159 °C
<b>3c</b>	C	chloroform	boiling	1 h	35.29%	158-160 °C
<b>3d</b>	B	toluene	boiling	2.5 h	32.83%	186-189 °C
<b>3d</b>	D	alkaline cyclization*	boiling	2 h	31.88%	189-191 °C
<b>3e</b>	A	diethyl ether	ambient	2 days	80.82%	235-237 °C
<b>3e</b>	B	toluene	boiling	1 h	85.71%	237-240 °C
<b>3e</b>	C	chloroform	boiling	1.5 h	66.47%	237-239 °C
<b>3f</b>	A	diethyl ether	ambient	2 days	64.30%	208-211 °C
<b>3f</b>	B	toluene	boiling	1 h	53.79%	208-211 °C
<b>3g</b>	A	diethyl ether	ambient	2 days	76.47%	251-254 °C
<b>3g</b>	B	toluene	boiling	1.5 h	74.93%	250-252 °C

\* alkaline cyclization of intermediate obtained in diethyl ether

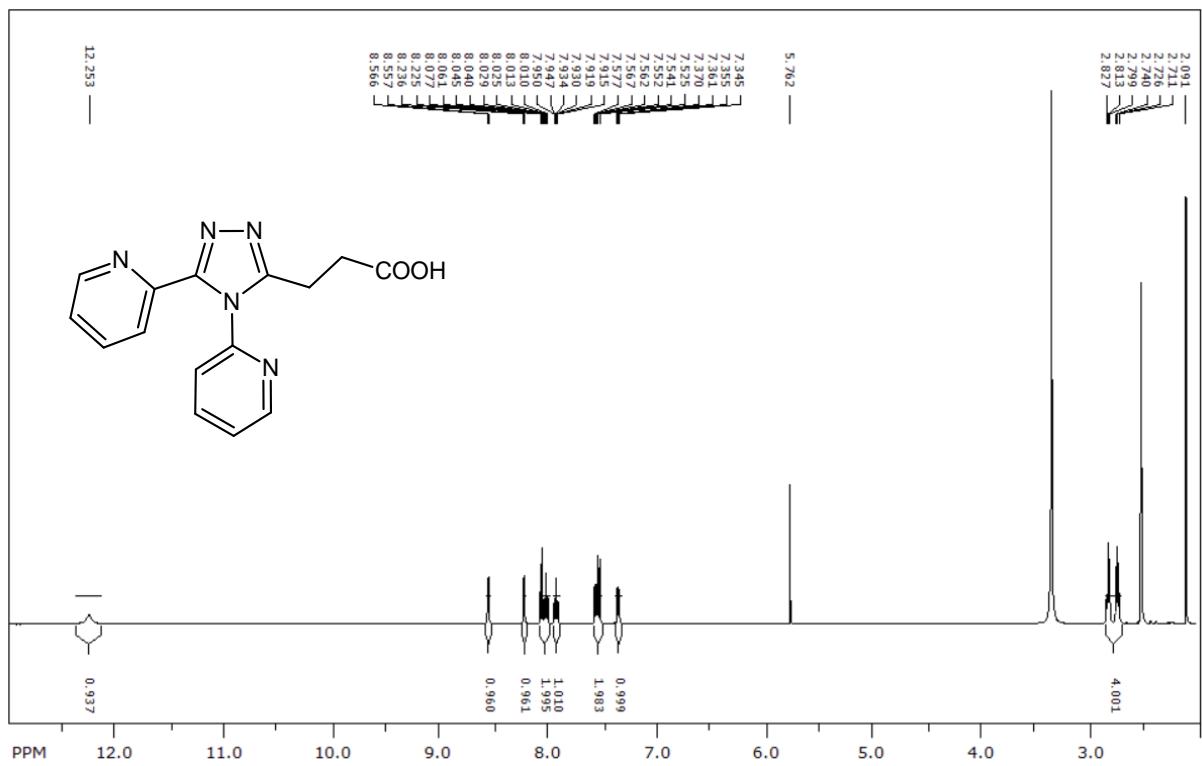


Figure S1.  $^1\text{H}$  NMR spectrum of compound **3a** (in  $\text{DMSO-d}_6$ )

(5.762 ppm – impurity of dichloromethane formed in the NMR study, 2.091 – impurity of acetone in the NMR study)

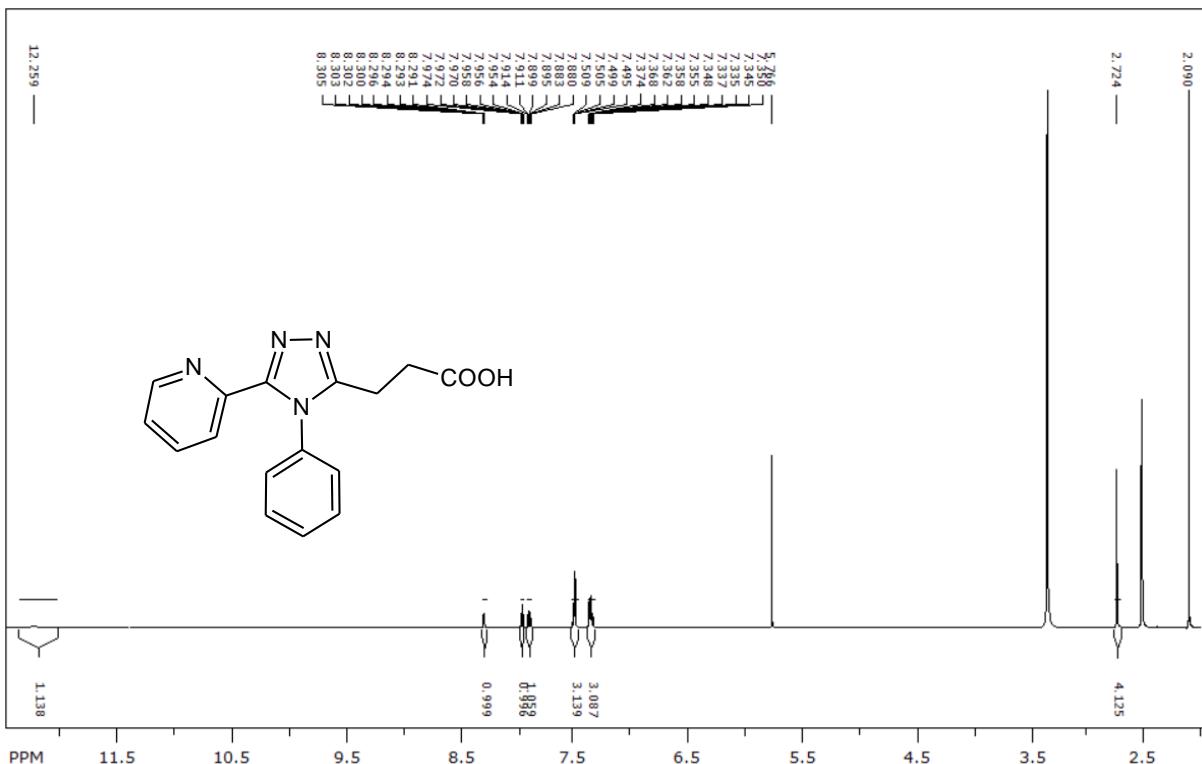


Figure S2.  $^1\text{H}$  NMR spectrum of compound **3b** (in  $\text{DMSO-d}_6$ )

(5.762 ppm – impurity of dichloromethane formed in the NMR study, 2.091 – impurity of acetone in the NMR study)

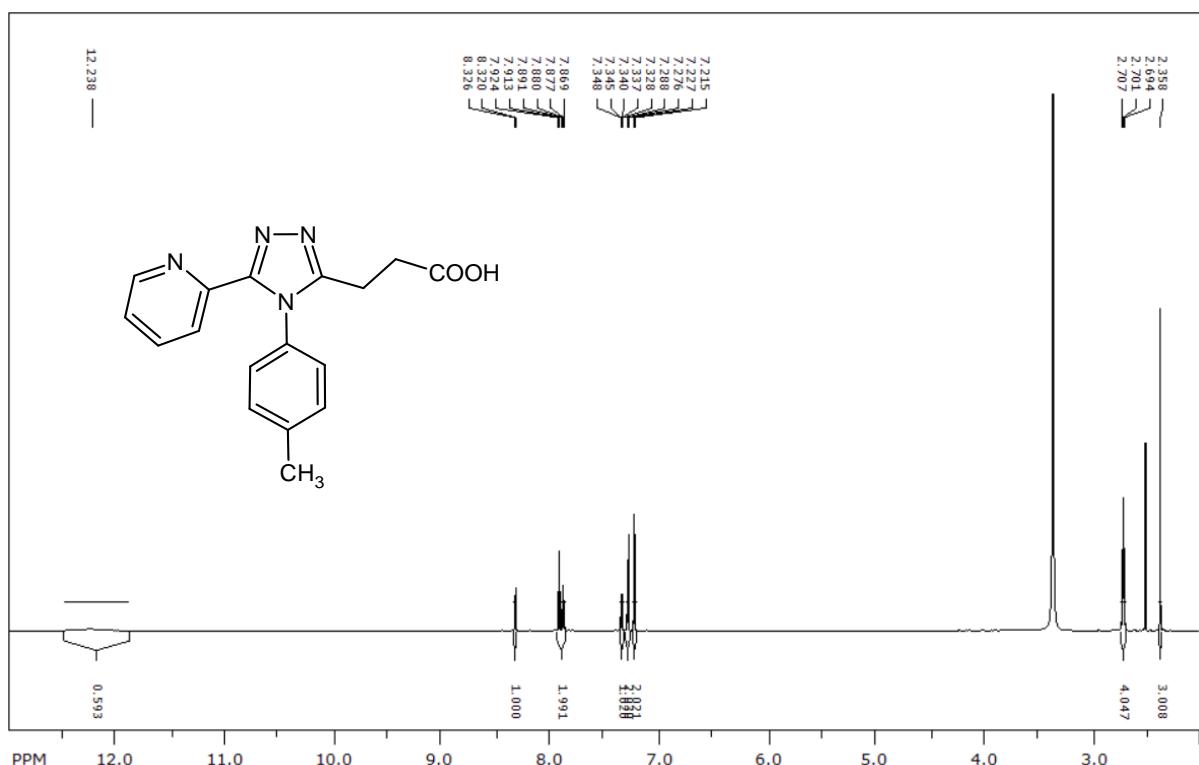


Figure S3.  $^1\text{H}$  NMR spectrum of compound **3c** (in  $\text{DMSO-d}_6$ )

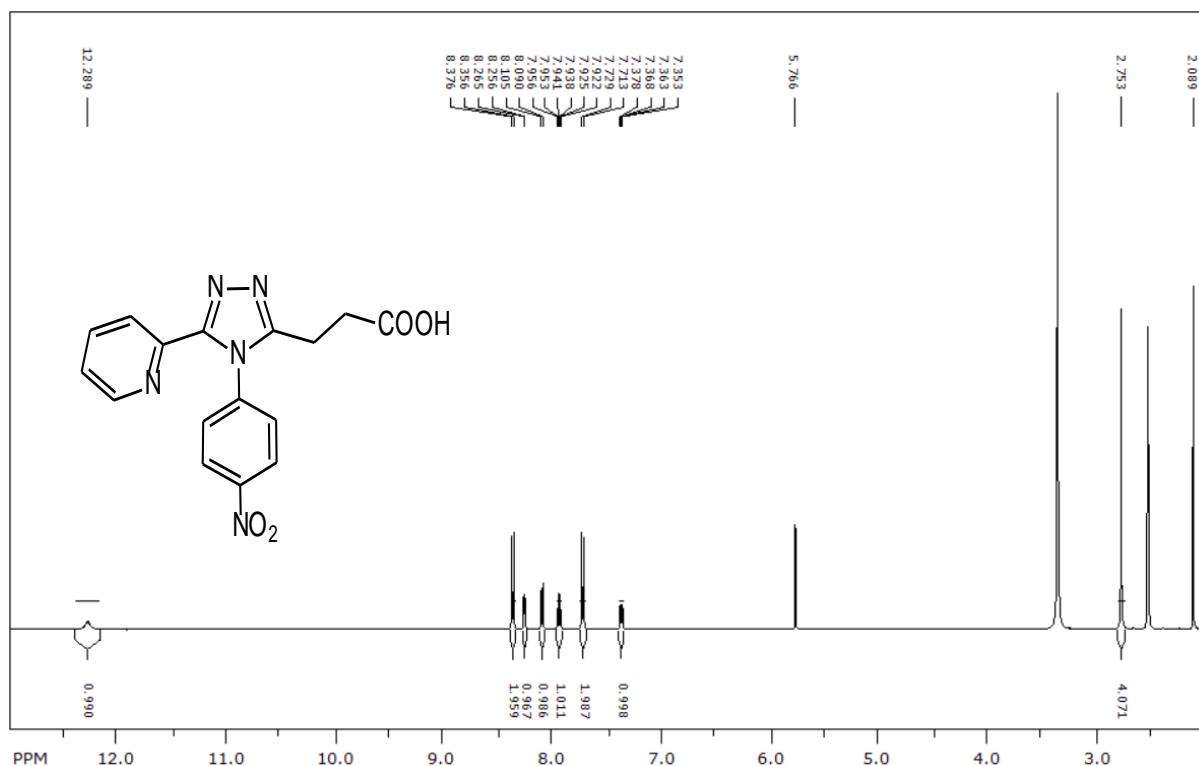


Figure S4.  $^1\text{H}$  NMR spectrum of compound **3d** (in  $\text{DMSO-d}_6$ )

(5.762 ppm – impurity of dichloromethane formed in the NMR study, 2.089 – impurity of acetone in the NMR study)

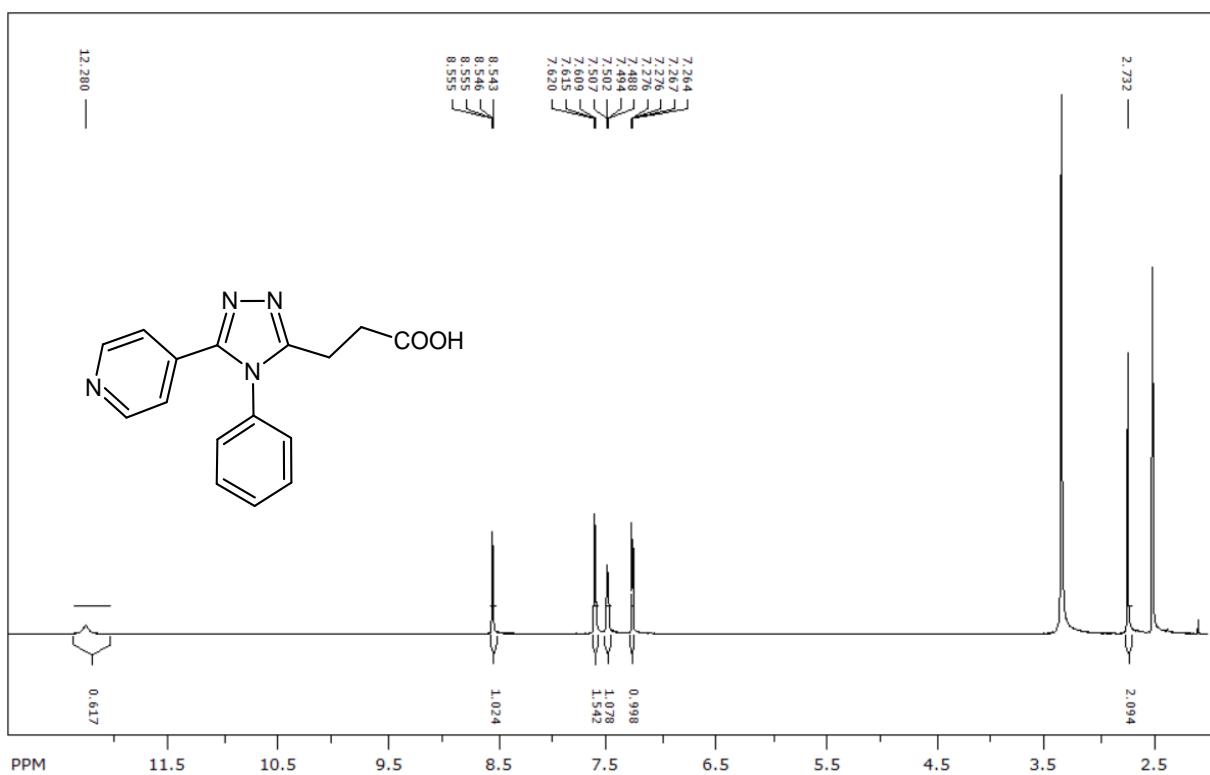


Figure S5.  $^1\text{H}$  NMR spectrum of compound 3e (in  $\text{DMSO-d}_6$ )

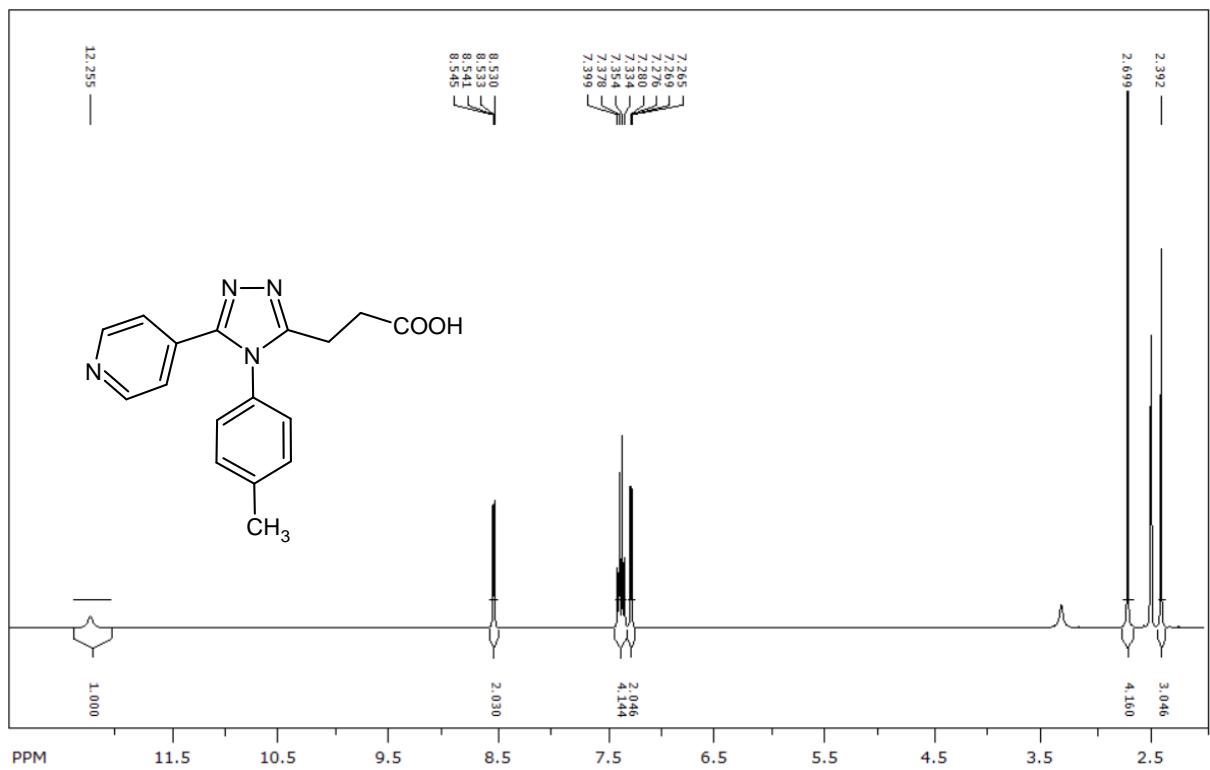


Figure S6.  $^1\text{H}$  NMR spectrum of compound 3f (in  $\text{DMSO-d}_6$ )

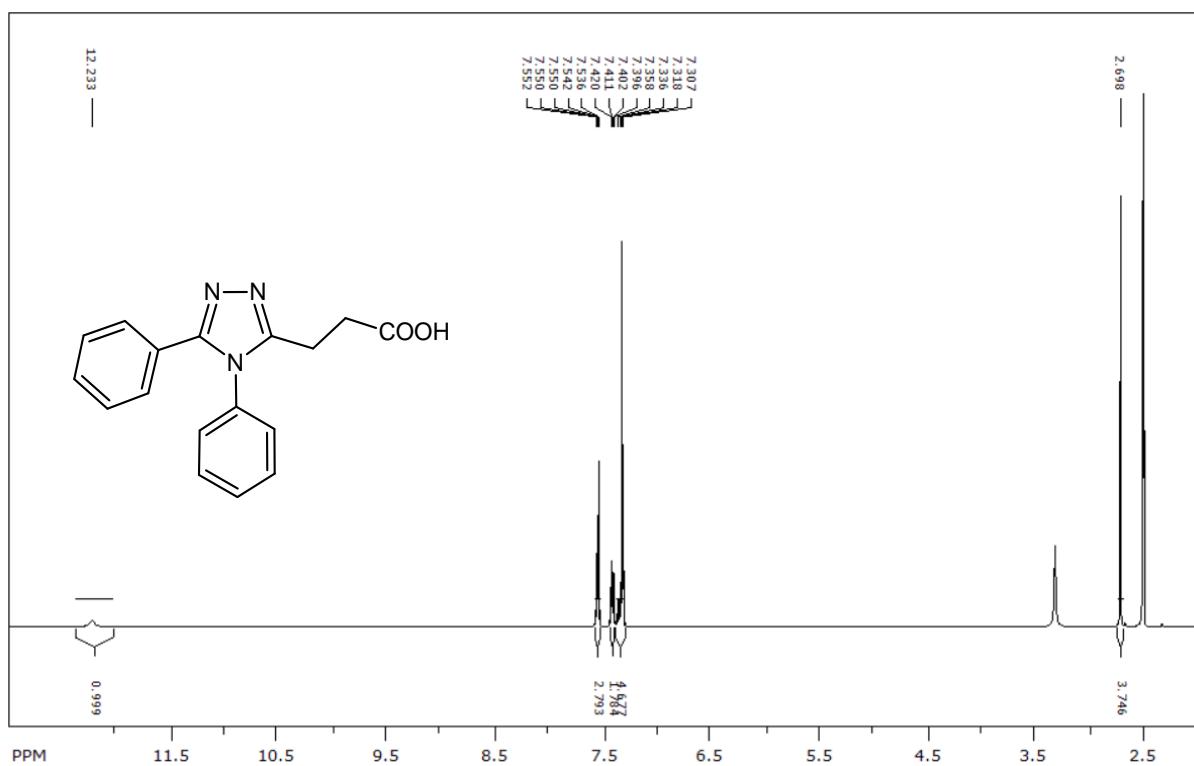


Figure S7. <sup>1</sup>H NMR spectrum of compound **3g** (in DMSO-d<sub>6</sub>)

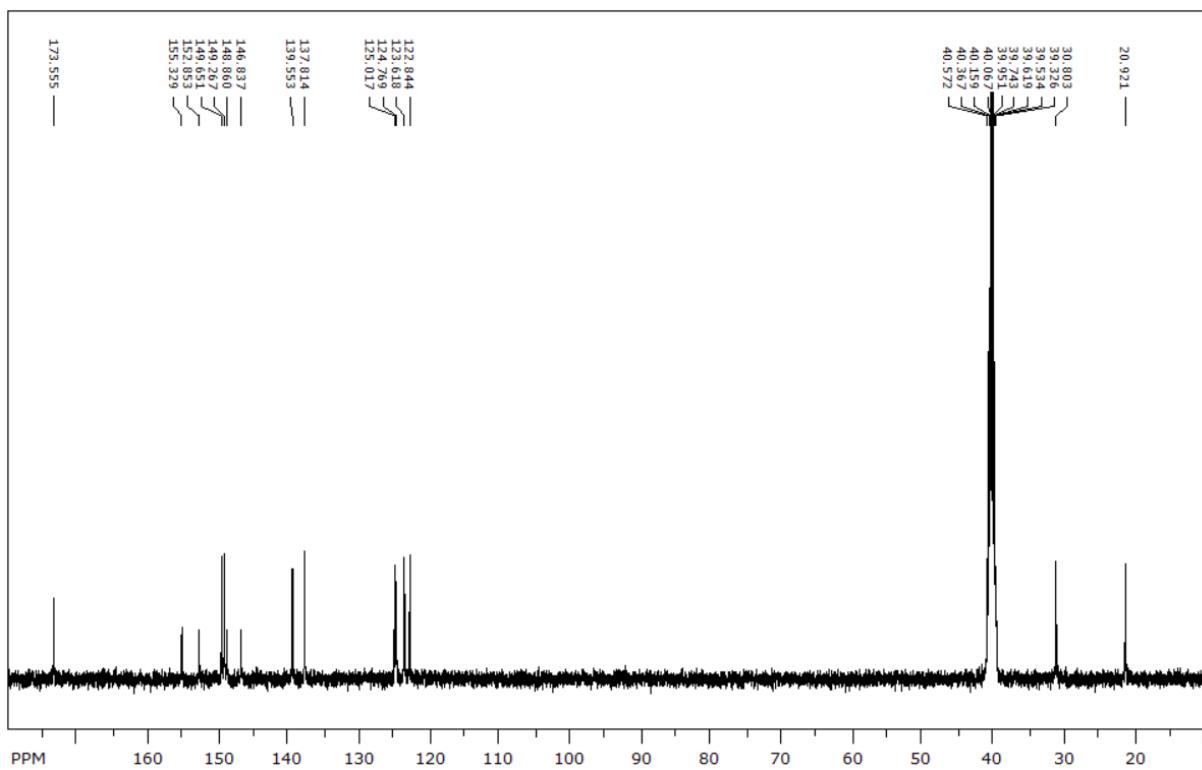


Figure S8. <sup>13</sup>C NMR spectrum of compound **3a** (in DMSO-d<sub>6</sub>)

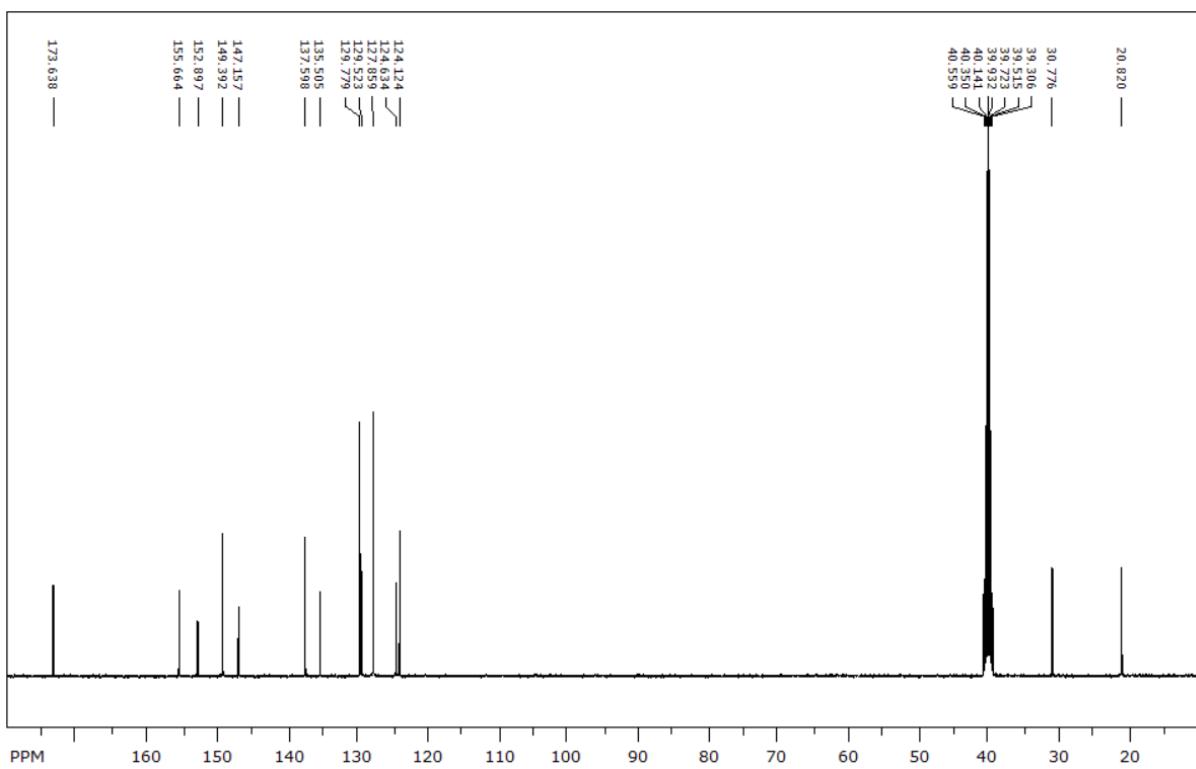


Figure S9. <sup>13</sup>C NMR spectrum of compound **3b** (in DMSO-d<sub>6</sub>)

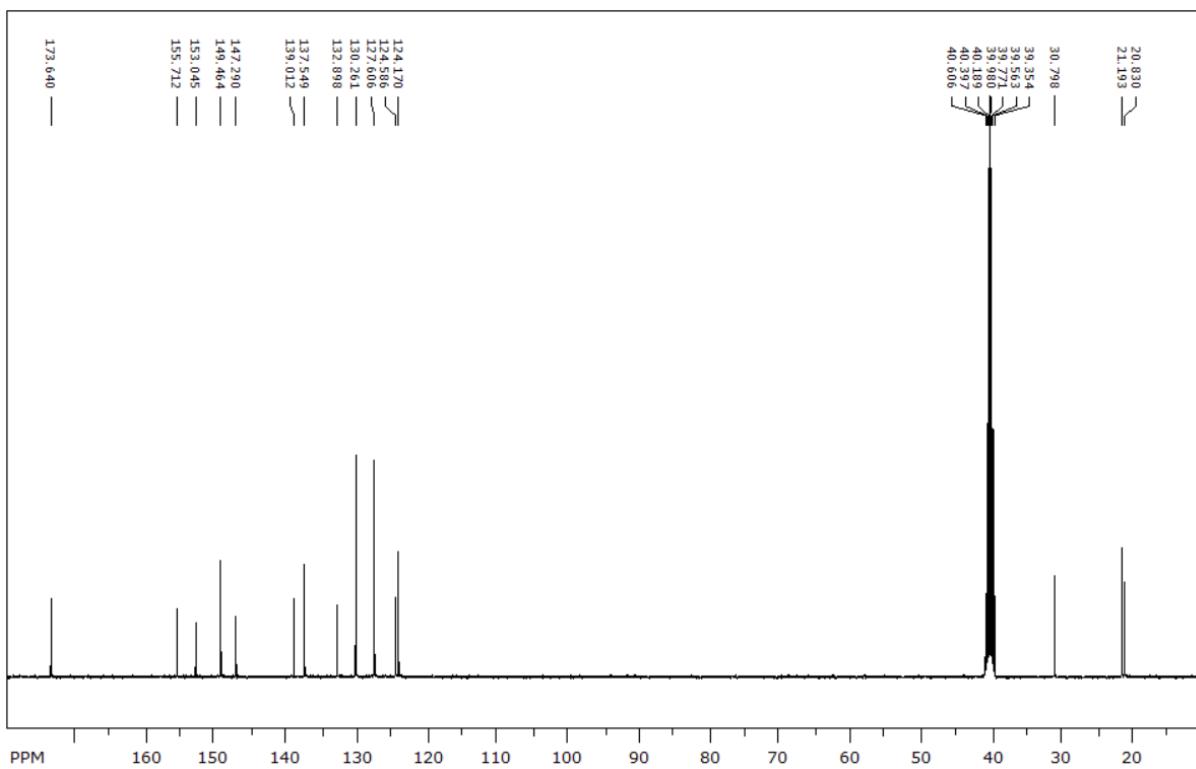


Figure S10. <sup>13</sup>C NMR spectrum of compound **3c** (in DMSO-d<sub>6</sub>)

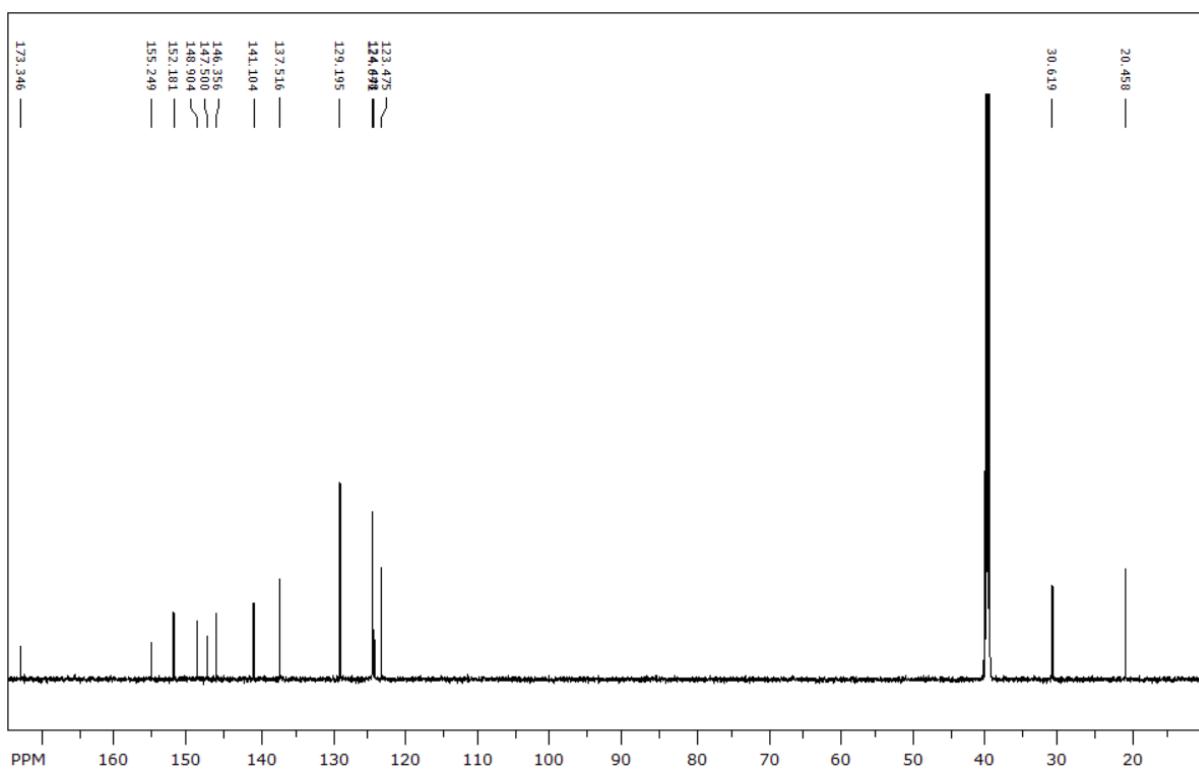


Figure S11. <sup>13</sup>C NMR spectrum of compound **3d** (in DMSO-d<sub>6</sub>)

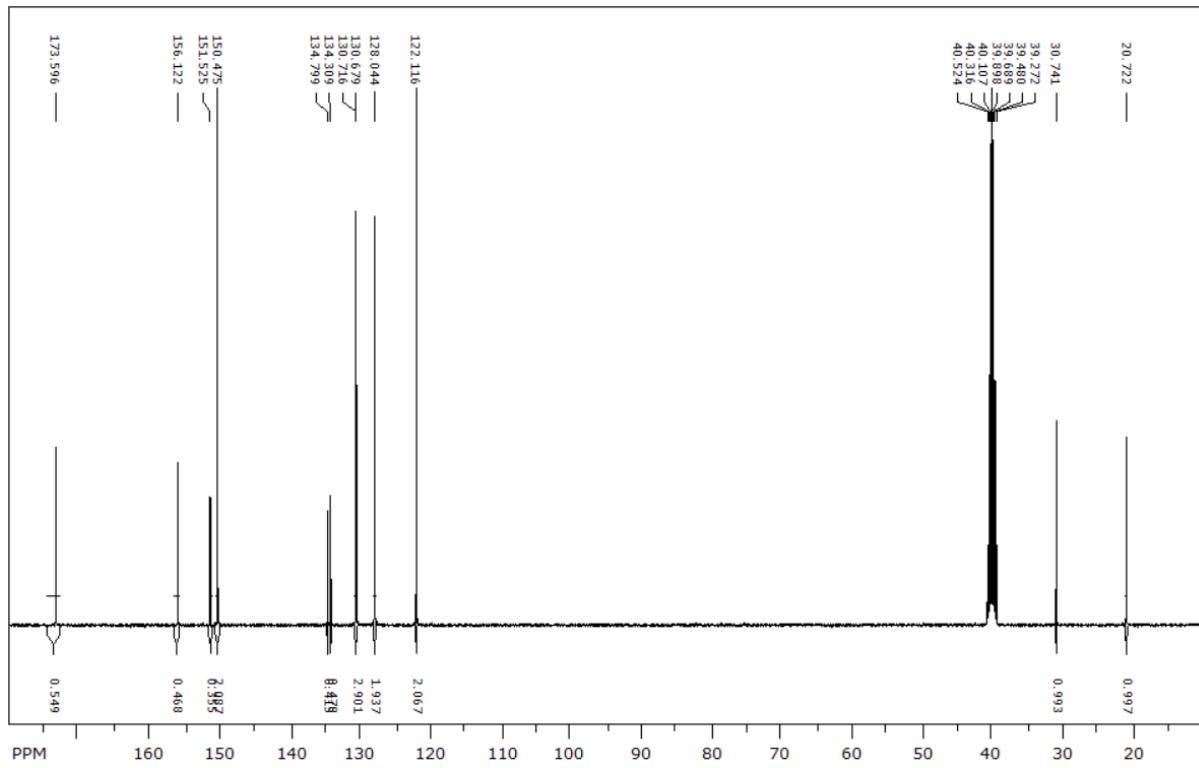


Figure S12. <sup>13</sup>C NMR spectrum of compound **3e** (in DMSO-d<sub>6</sub>)

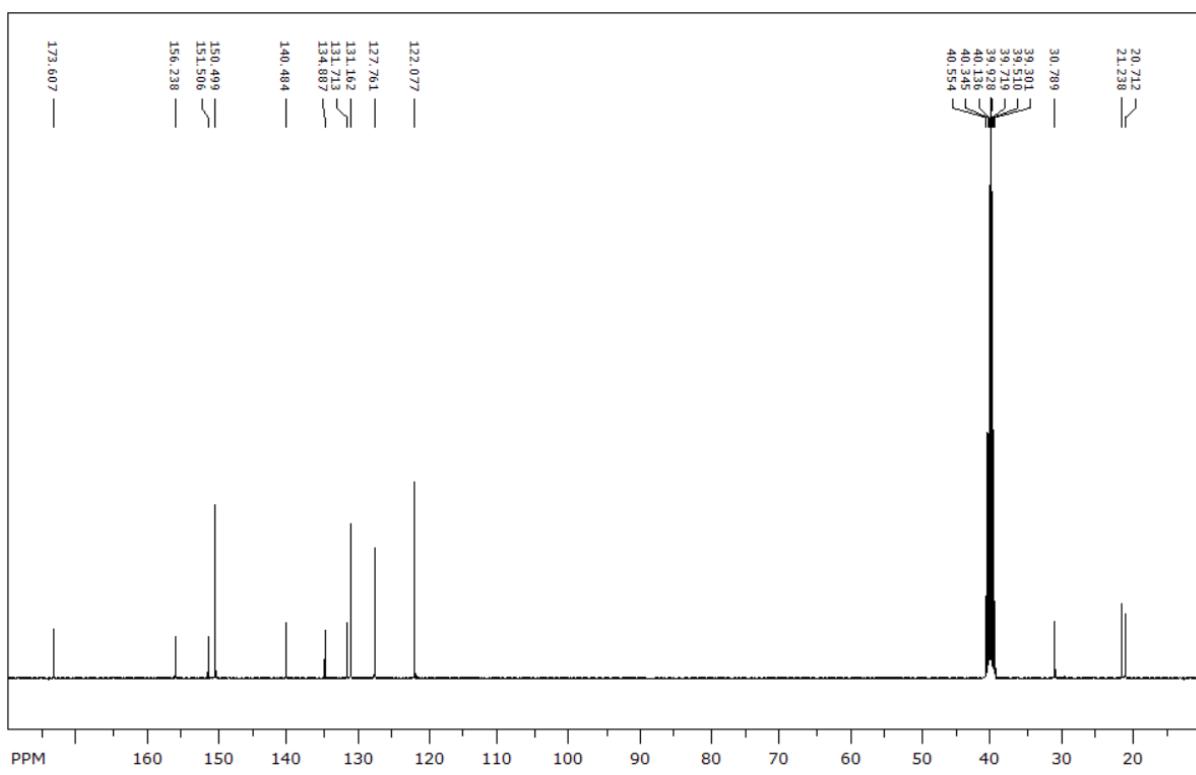


Figure S13. <sup>13</sup>C NMR spectrum of compound 3f (in DMSO-d<sub>6</sub>)

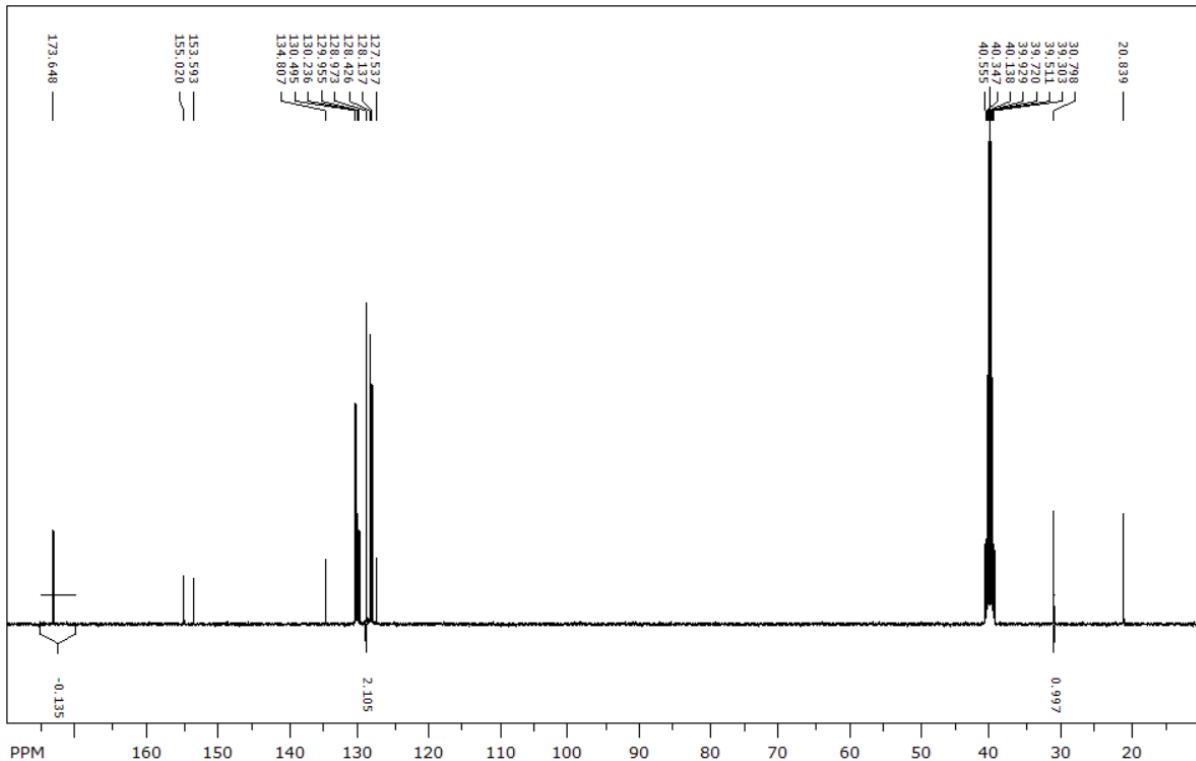


Figure S14. <sup>13</sup>C NMR spectrum of compound 3g (in DMSO-d<sub>6</sub>)

**Elemental Composition Report****Page 1****Single Mass Analysis**

Tolerance = 7.0 PPM / DBE: min = -1.5, max = 90.0

Element prediction: Off

Number of isotope peaks used for i-FIT = 9

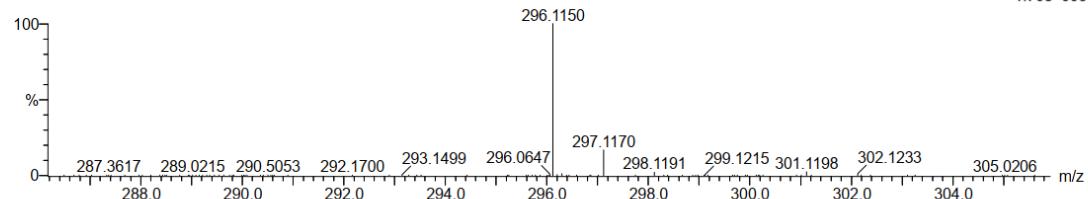
Monoisotopic Mass, Even Electron Ions

70 formula(e) evaluated with 1 results within limits (all results (up to 1000) for each mass)

Elements Used:

C: 0-20 H: 0-25 N: 1-6 O: 0-5

230207\_YA 9 (0.147) Cm (8.9-3.6)

TOF MS ES+  
1.70e+006

Mass	Calc. Mass	mDa	PPM	DBE	i-FIT	Norm	Conf (%)	Formula
296.1150	296.1147	0.3	1.0	11.5	345.7	n/a	n/a	C15 H14 N5 O2

Figure S15. HRMS spectrum of compound 3a

**Elemental Composition Report****Page 1****Single Mass Analysis**

Tolerance = 7.0 PPM / DBE: min = -1.5, max = 90.0

Element prediction: Off

Number of isotope peaks used for i-FIT = 9

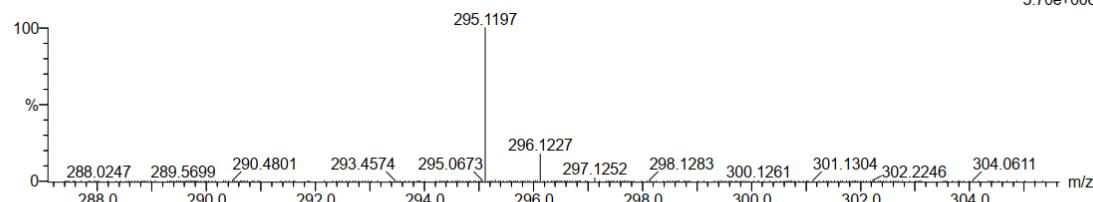
Monoisotopic Mass, Even Electron Ions

48 formula(e) evaluated with 1 results within limits (all results (up to 1000) for each mass)

Elements Used:

C: 0-20 H: 0-25 N: 1-4 O: 0-5

230207 ACA 26 (0.374) Cm (26.29-3.6)

TOF MS ES+  
5.70e+006

Mass	Calc. Mass	mDa	PPM	DBE	i-FIT	Norm	Conf (%)	Formula
295.1197	295.1195	0.2	0.7	11.5	1420.2	n/a	n/a	C16 H15 N4 O2

Figure S16. HRMS spectrum of compound 3b

**Elemental Composition Report****Page 1****Single Mass Analysis**

Tolerance = 7.0 PPM / DBE: min = -1.5, max = 90.0

Element prediction: Off

Number of isotope peaks used for i-FIT = 9

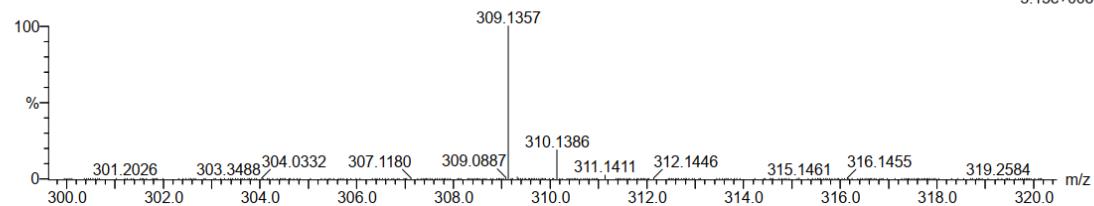
## Monoisotopic Mass, Even Electron Ions

43 formula(e) evaluated with 1 results within limits (all results (up to 1000) for each mass)

Elements Used:

C: 0-20 H: 0-25 N: 1-4 O: 0-5

230207\_XA 40 (0.548) Cm (40:41)

TOF MS ES+  
3.15e+006Minimum: -1.5  
Maximum: 5.0 7.0 90.0

Mass	Calc. Mass	mDa	PPM	DBE	i-FIT	Norm	Conf (%)	Formula
309.1357	309.1352	0.5	1.6	11.5	1340.5	n/a	n/a	C17 H17 N4 O2

Figure S17. HRMS spectrum of compound 3c

**Single Mass Analysis**

Tolerance = 5.0 PPM / DBE: min = -1.5, max = 90.0

Element prediction: Off

Number of isotope peaks used for i-FIT = 9

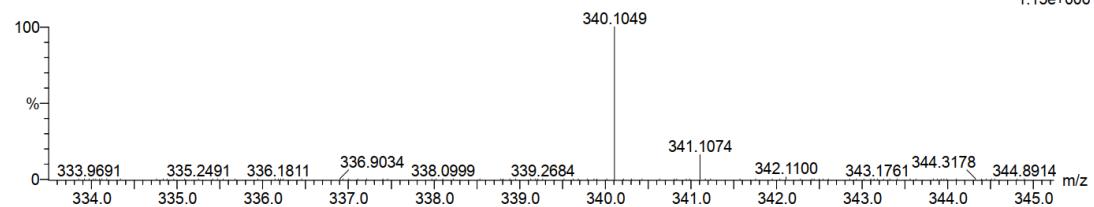
## Monoisotopic Mass, Even Electron Ions

185 formula(e) evaluated with 1 results within limits (all results (up to 1000) for each mass)

Elements Used:

C: 0-45 H: 0-45 N: 0-6 O: 0-6

230310\_probka\_AA\_pos\_A 25 (0.365) Cm (25:30-4:6)

TOF MS ES+  
1.15e+006Minimum: -1.5  
Maximum: 5.0 5.0 90.0

Mass	Calc. Mass	mDa	PPM	DBE	i-FIT	Norm	Conf (%)	Formula
340.1049	340.1046	0.3	0.9	12.5	480.0	n/a	n/a	C16 H14 N5 O4

Figure S18. HRMS spectrum of compound 3d

**Elemental Composition Report****Page 1****Single Mass Analysis**

Tolerance = 7.0 PPM / DBE: min = -1.5, max = 90.0  
Element prediction: Off  
Number of isotope peaks used for i-FIT = 9

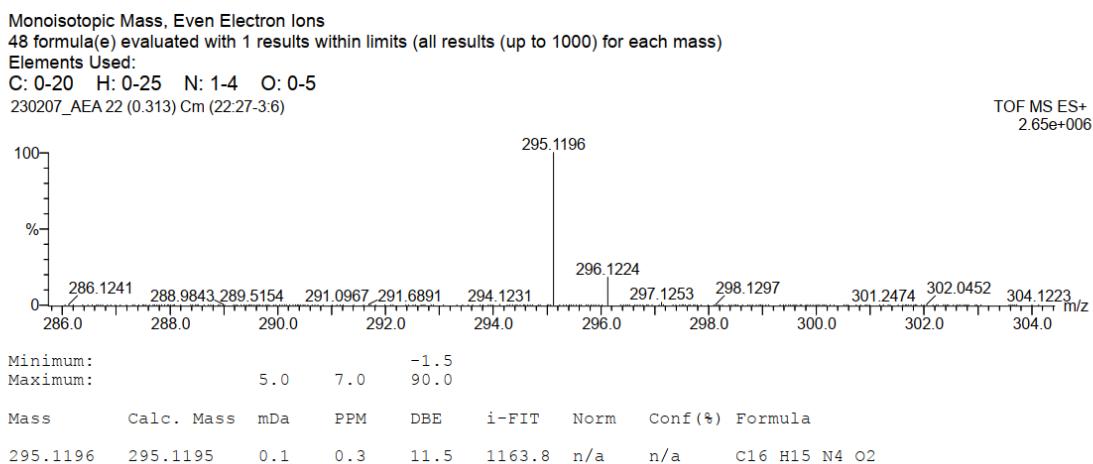


Figure S19. HRMS spectrum of compound 3e

**Elemental Composition Report****Page 1****Single Mass Analysis**

Tolerance = 7.0 PPM / DBE: min = -1.5, max = 90.0  
Element prediction: Off  
Number of isotope peaks used for i-FIT = 9

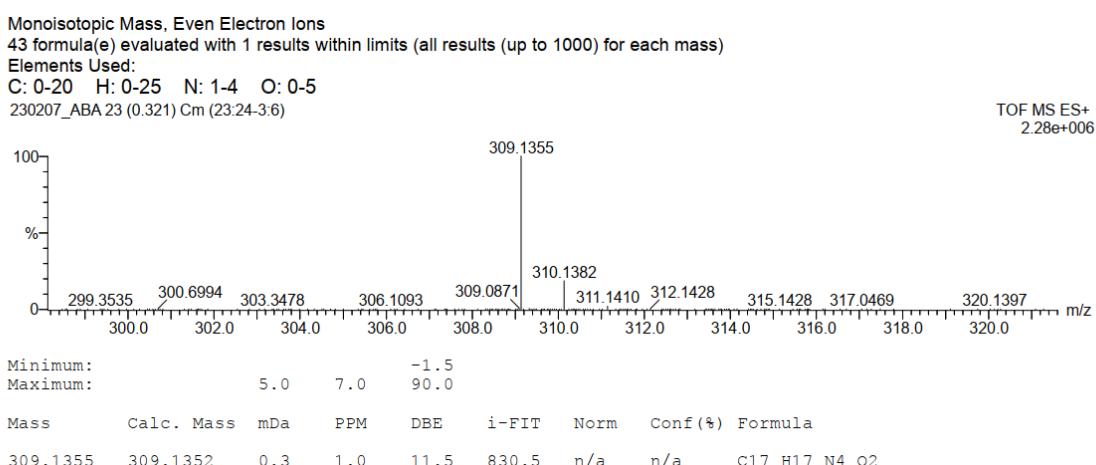


Figure S20. HRMS spectrum of compound 3f

**Single Mass Analysis**

Tolerance = 7.0 PPM / DBE: min = -1.5, max = 90.0  
 Element prediction: Off  
 Number of isotope peaks used for i-FIT = 9

## Monoisotopic Mass, Even Electron Ions

45 formula(e) evaluated with 1 results within limits (all results (up to 1000) for each mass)

## Elements Used:

C: 0-20 H: 0-25 N: 1-4 O: 0-5  
 230207\_ADA 23 (0.321) Cm (23:24-3:6)

TOF MS ES+  
 2.29e+006

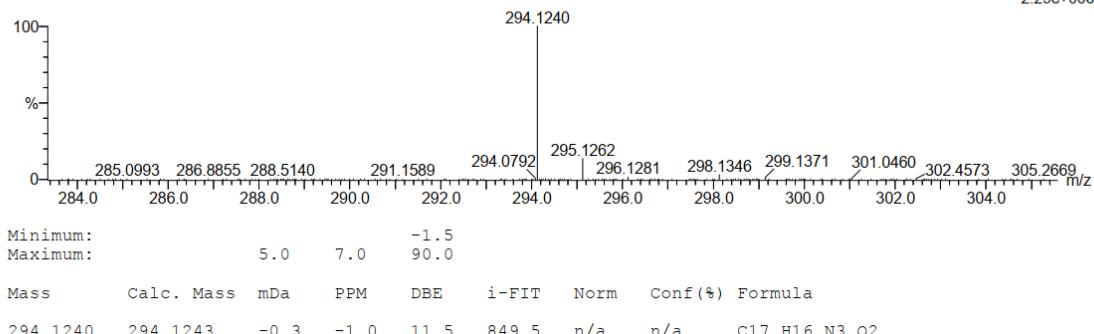


Figure S21. HRMS spectrum of compound **3g**

Table S2. The results of cell apoptosis assay of compounds **3a-3g** and ibuprofen at concentration 100 µg/mL in PBMC 24 h cultured measured by flow cytometry.

	<b>viable cells</b>	<b>early apoptotic cells</b>	<b>late apoptotic cells</b>	<b>necrotic cells</b>
control	96.45%	0.58%	0.40%	2.56%
control with 2% DMSO	94.19%	2.08%	0.74%	2.98%
<b>IBU</b>	96.01%	1.39%	0.77%	1.84%
<b>3c</b>	96.72%	1.22%	0.76%	1.31%
<b>3b</b>	94.89%	1.64%	1.51%	1.96%
<b>3c</b>	95.71%	1.68%	1.26%	1.36%
<b>3d</b>	94.71%	1.84%	1.82%	1.62%
<b>3e</b>	95.74%	0.73%	0.47%	3.05%
<b>3f</b>	95.34%	1.57%	0.73%	2.36%
<b>3g</b>	95.62%	1.92%	0.63%	1.84%

Table S3. Minimal inhibitory concentrations of compounds **3a-3g**, ciprofloxacin (CIP) and amphotericin B (AmB).

strain	MIC $\mu\text{g/mL}$								
	<b>3a</b>	<b>3b</b>	<b>3c</b>	<b>3d</b>	<b>3e</b>	<b>3f</b>	<b>3g</b>	<b>CIP</b>	<b>AmB</b>
<i>E. coli</i> ATCC 25922	>512	512	>512	512	>512	512	256	0.015	
<i>Y. enterocolitica</i> O3	512	512	512	512	512	512	512	0.125	
<i>S. aureus</i> ATCC 25923	>512	>512	>512	>512	>512	>512	256	0.25	
<i>M. smegmatis</i>	128	256	256	512	128	256	512	0.5	
<i>C. albicans</i> ATCC 90028	>512	>512	>512	512	>512	512	512	-	0.125