

Supporting information (SI)

A Polyzwitterionic@MOF Hydrogel with Exceptionally High Water Vapor Uptake for Efficient Atmospheric Water Harvesting

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S1. Water vapor adsorption isotherms on pure LiCl, PL-3 and PML hydrogels

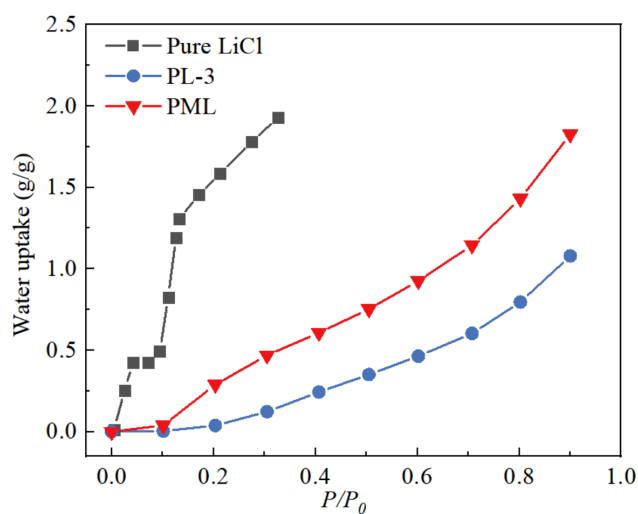


Figure S1. Water vapor adsorption isotherms on pure LiCl, PL-3 and PML hydrogels at 298 K.

S2. Optical photographs of PL hydrogels after adsorption-desorption cycle

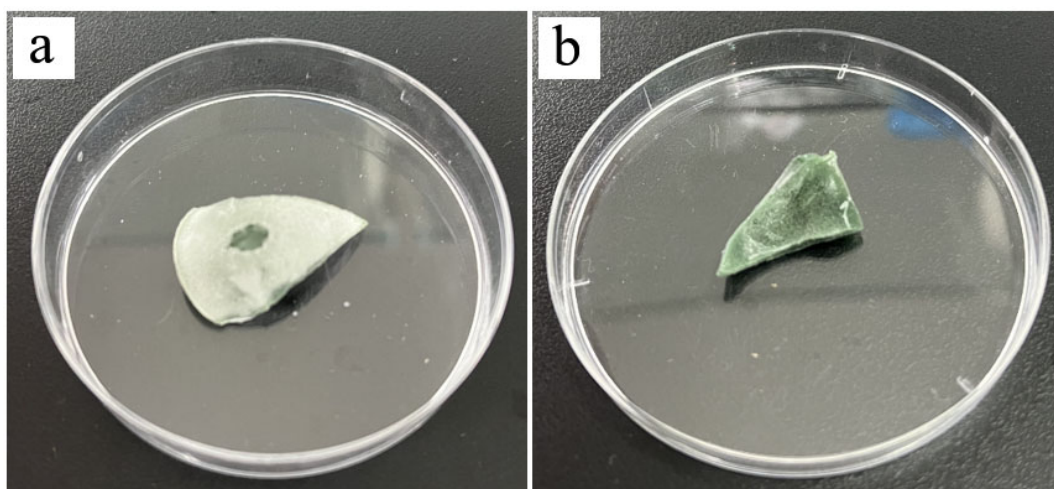


Figure S2. Optical photographs of (a) PL-2 and (b) PL-3 hydrogels after one adsorption-desorption cycle

S3. Thermogravimetric and FTIR curves of PML hydrogel before and after adsorption-desorption cycle

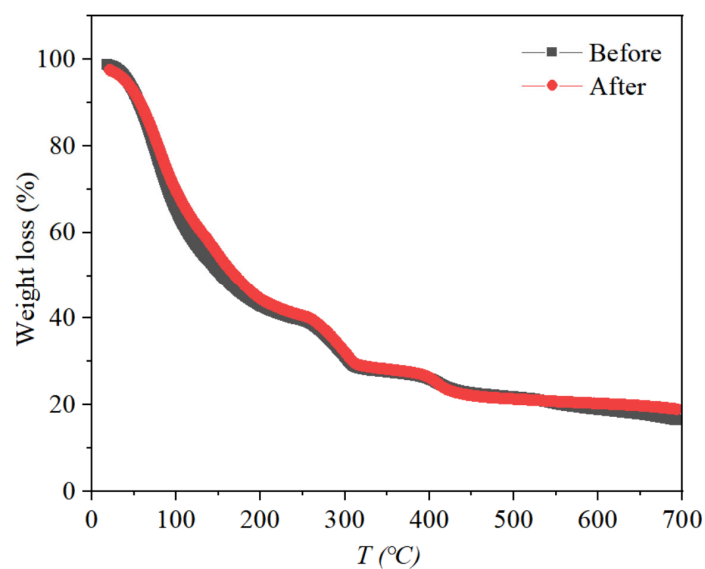


Figure S3. Thermogravimetric curves of PML hydrogel before and after an adsorption-desorption cycle

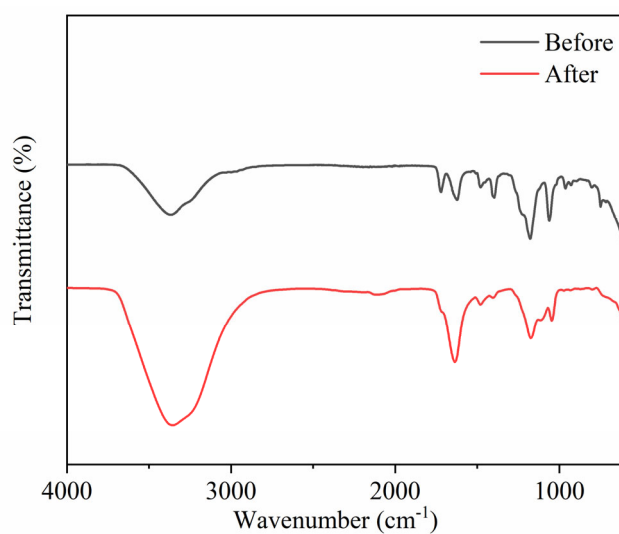


Figure S4. FTIR spectra of PML hydrogel before and after an adsorption-desorption cycle

S4. N₂ adsorption test and water adsorption isotherm of MIL-101(Cr)

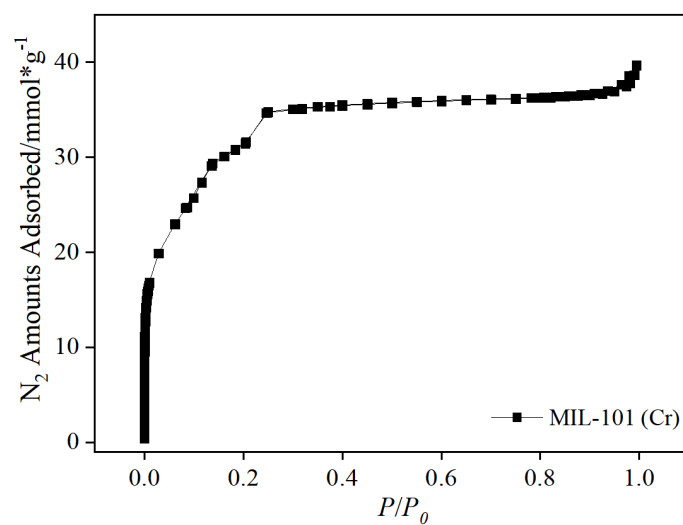


Figure S5. N₂ adsorption isotherm of MIL-101(Cr) at 298 K

Table S1. Pore structure parameters of MIL-101 (Cr)

Sample	BET (m^2/g)	Langmuir (m^2/g)	Pore Volume (cm^3/g)
MIL-101(Cr)	2602	3711	1.25

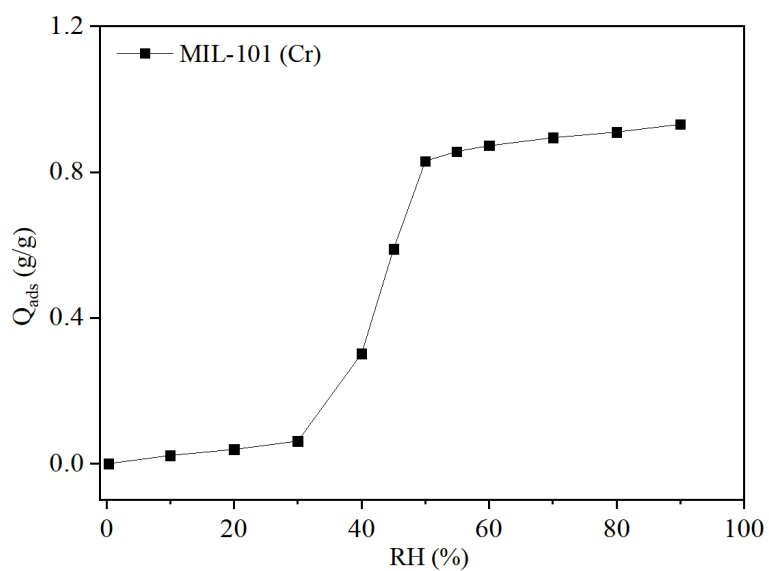


Figure S6. Water vapor adsorption isotherm of MIL-101(Cr) at 298 K

S5. Comparison of the swelling ratio of the hydrogels and other materials

Table S2. Comparison of the swelling ratio of the hydrogels for atmospheric water harvesting in the literature

Materials	Swelling ratio (g/g)	Swelling condition	Reference
PML	1.82	90% RH moisture	This study
PML	2.29	DI water	This study
PML	9.03	4 mol/L LiCl solution	This study
POG	1.54	90% RH moisture	[1]
SMAG	6.3	90% RH moisture	[2]
PAM-CNT	1.75	80% RH moisture	[3]

S6. Comparison of the water capacity of the hydrogels and other materials

Table S3. Comparison of the water capacity of sorbents in the literature.

Materials	Water vapor capacity (g/g) at RH=90%, 25°C	Water vapor capacity (g/g) at RH=30%, 25°C	Reference
PML	1.82	0.43	This study
MOF-801	0.28	0.47	[4]
MIL-101(Cr)	1.52 ^a	0.25	[5]
PNIPAM	0.21	0.03	[6]
PNIPAM@MIL-101(Cr)	3.6	0.71	[7]
PNIPAM-PPy-Cl (SMAG)	6.5	0.70	[2]
PAM-CNT-CaCl ₂	1.75 ^b	0.68 ^c	[8]
SHCP-10 (POPs)	0.76	0.21	[9]
2D ep-POP	0.40	0.1	[10]
COF-432	0.31	0.025	[11]
COFs-480-hydrazide	0.47	0.33	[12]

^a: $T_{\text{ads/des}}=30^{\circ}\text{C}$;

^b: RH=80%;

^c: RH=35%

1. Ni, F.; Qiu, N.; Xiao, P.; Zhang, C.; Jian, Y.; Liang, Y.; Xie, W.; Yan, L.; Chen, T. Tillandsia-Inspired hygroscopic photothermal organogels for efficient atmospheric water harvesting. *Angew. Chem. Int. Ed.* **2020**, *59*, 19237–19246.
2. Zhao, F.; Zhou, X.; Liu, Y.; Shi, Y.; Dai, Y.; Yu, G. Super Moisture-absorbent gels for all-weather atmospheric water harvesting. *Adv. Mater.* **2019**, *31*, 1806446.
3. Li, R.; Shi, Y.; Alsaedi, M.; Wu, M.; Shi, L.; Wang, P. Hybrid hydrogel with high water vapor harvesting capacity for deployable solar-driven atmospheric water generator. *Environ. Sci. Technol.* **2018**, *52*, 11367–11377.
4. Furukawa, H.; Gándara, F.; Zhang, Y.; Jiang, J.; Queen, W.; Hudson, M.; Yaghi, O.M. Water adsorption in porous metal-organic frameworks and related materials. *J. Am. Chem. Soc.* **2014**, *136*, 4369–4381.
5. Seo, Y.; Yoon, J.; Férey, Gérard. Porous Materials: Energy-Efficient Dehumidification over Hierarchically Porous Metal-Organic Frameworks as Advanced Water Adsorbents. *Adv. Mater.* **2012**, *24*, 806–810.
6. Karmakar, A.; Mileo, P.G.M.; Bok, I.; Peh, S.B.; Zhang, J.; Yuan, H.; Maurin, G.; Zhao, D. thermo-responsive MOF/polymer composites for temperature-mediated water capture and release. *Angew. Chem. Int. Ed.* **2020**, *59*, 11003–11009.
7. Yilmaz, G.; Meng, F.L.; Lu, W.; Abed, J.; Peh, C.K.N.; Gao, M.; Sargent, E.H.; Ho, G.W. Autonomous atmospheric water seeping MOF matrix. *Sci. Adv.* **2020**, *6*, 8605.
8. Li, R.; Shi, Y.; Alsaedi, M.; Wu, M.; Shi, L.; Wang, P. Hybrid hydrogel with high water vapor harvesting capacity for deployable solar-driven atmospheric water generator. *Environ. Sci. Technol.* **2018**, *52*, 11367–11377.
9. Schweng, P.; Mayer, F.; Galehdari, D.; Weiland, K.; Woodward, R. A Robust and Low-Cost Sulfonated Hypercrosslinked Polymer for Atmospheric Water Harvesting. *Small.* **2023**, *19*, 2304562.
10. Byun, Y.; Coskun, A. Epoxy-Functionalized Porous Organic Polymers via the Diels-Alder Cycloaddition Reaction for Atmospheric Water Capture. *Angew. Chem. Int. Ed.* **2018**, *57*, 3173–3177.
11. Nguyen, H.; Hanikel, N.; Lyle, S.; Zhu, C.; Proserpio, D.; Yaghi, O.M. A Porous Covalent Organic Framework with Voided Square Grid Topology for Atmospheric Water Harvesting. *J. Am. Chem. Soc.* **2020**, *142*, 2218–2221.
12. Nguyen, H.; Gropp, C.; Hanikel, N.; Möckel, A.; Lund, A.; Yaghi, O.M. Hydrazine-Hydrazide-Linked Covalent Organic Frameworks for Water Harvesting. *ACS Cent. Sci.* **2022**, *8*, 926–932.