

Supplemental online material

Table S1 – Base thermal characteristics obtained from the DSC measurements of amorphous NIF powders.

powder	q ⁺	T _g	T _g ^p	T _{ons}	T _{p1}	T _{p2}	ΔH _c	T _m ^{ons}	T _m ^p	ΔH _m
50 - 125	0.5	30.142	33	51	53.4	55.5	61.43	164.75	167.43	99.49
	1	33.01	35.13	55.46	58.13	64.5	53.81	163.41	167.81	98.74
	2	36.09	37.66	61.82	64.91	70.19	64.91	164.99	167.99	99.76
	5	36.21	39.55	67.53	70.72	77.45	69.82	165.95	168.4	100.4
	10	37.7	41.29	72.6	76.65	86.93	66.83	165.88	169.29	97.09
	20	41.45	43.34	78.8	83.4	90.87	70.97	166.52	170.34	98.23
powder	q ⁺	T _g	T _g ^p	T _{ons}	T _{p1}	T _{p2}	ΔH _c	T _m ^{ons}	T _m ^p	ΔH _m
125-180	0.5	30.13	34.2	54.98	57.57	60.15	64.65	166.54	169.1	103.6
	1	33.71	36.39	59.98	62.44	64.82	55.31	166.43	169.41	107.2
	2	35.92	38.45	65.24	67.68	72.24	62.53	166.63	169.55	101.7
	5	37.39	40.66	71.15	74	79.03	68.08	171.22	172.42	112.9
	10	39.18	42.71	75.57	79.72	86.29	68.09	167.84	170.77	105.7
	20	40.97	43.82	81.62	85.89	92.45	76.7	168.48	171.87	110.8
powder	q ⁺	T _g	T _g ^p	T _{ons}	T _{p1}	T _{p2}	ΔH _c	T _m ^{ons}	T _m ^p	ΔH _m
180-250	0.5	32.76	35.45	59.66	62.14	67.66	51.7	163.5	167.81	99.46
	1	34.5	37.5	65.91	68.66	75.87	65.33	164.23	168.32	100.1
	2	37.28	38.76	70.68	73.6	82.66	65.91	164.85	168.59	98.97
	5	37.9	41.13	77.87	81.12	91.82	69.68	165.48	169.26	103.2
	10	39.53	43.18	83.67	87.61	99.4	71.71	166.68	170.17	105.4
	20	43.55	45.24	87.56	93.77	105.4	72.06	166.74	171.12	93.63
powder	q ⁺	T _g	T _g ^p	T _{ons}	T _{p1}	T _{p2}	ΔH _c	T _m ^{ons}	T _m ^p	ΔH _m
250-300	0.5	34.5	36.24	60.36	62.14	64.82	50.96	167.64	169.88	106.9
	1	36.08	38.29	65.43	67.39	73.03	56.24	167.92	170.11	108.9
	2	38.72	40.03	70.13	72.38	77.61	67.86	168.01	170.28	111.1
	5	39.25	42.08	75.05	78.7	85.19	67.89	168.21	170.67	108.4
	10	40.54	44.13	81.36	85.81	94.19	67.58	168.79	171.64	102.4
	20	44.64	46.5	87.57	92.9	101.45	75.65	169.43	172.72	110.7
powder	q ⁺	T _g	T _g ^p	T _{ons}	T _{p1}	T _{p2}	ΔH _c	T _m ^{ons}	T _m ^p	ΔH _m
300-500	0.5	34.82	37.82	61.3	63.66	66.56	49.48	167.4	169.95	105.9
	1	36.71	39.24	66.44	68.79	74.77	53.44	168.06	170.3	103.9
	2	39.75	41.13	71.11	73.96	79.66	64.33	168.5	170.43	108.7
	5	40.24	43.18	75.31	81.23	88.51	66.92	168.66	171.11	105.5
	10	41.61	44.92	83.16	88.3	95.61	73.58	169.19	171.79	111.4
	20	44.61	47.76	89.19	95.86	104.61	73.69	170.81	173.46	108.4

powder	q^+	T_g	T_g^p	T_{ons}	T_{p1}	T_{p2}	ΔH_c	T_m^{ons}	T_m^p	ΔH_m
500-1000	0.5	37.53	40.44	68.36	71.73	73.16	61.79	170.3	171.32	115.9
	1	38.92	41.07	72.55	74.63	77.71	66.86	170.02	171.04	118
	2	40.19	43.47	77.31	80.92	85.66	70.26	170.67	171.75	112
	5	42.21	46.76	85.75	90.39	93.12	74.24	170.88	173.26	110.4
	10	43.1	47.64	90.39	94.76	99.43	74.45	172.17	173.72	112.8
	20	45.72	51.4	94.63	101.1	106.98	77.66	172.37	175.26	115.3

Figure S1 – Graphical representation of the Moynihan’s equal areas method.

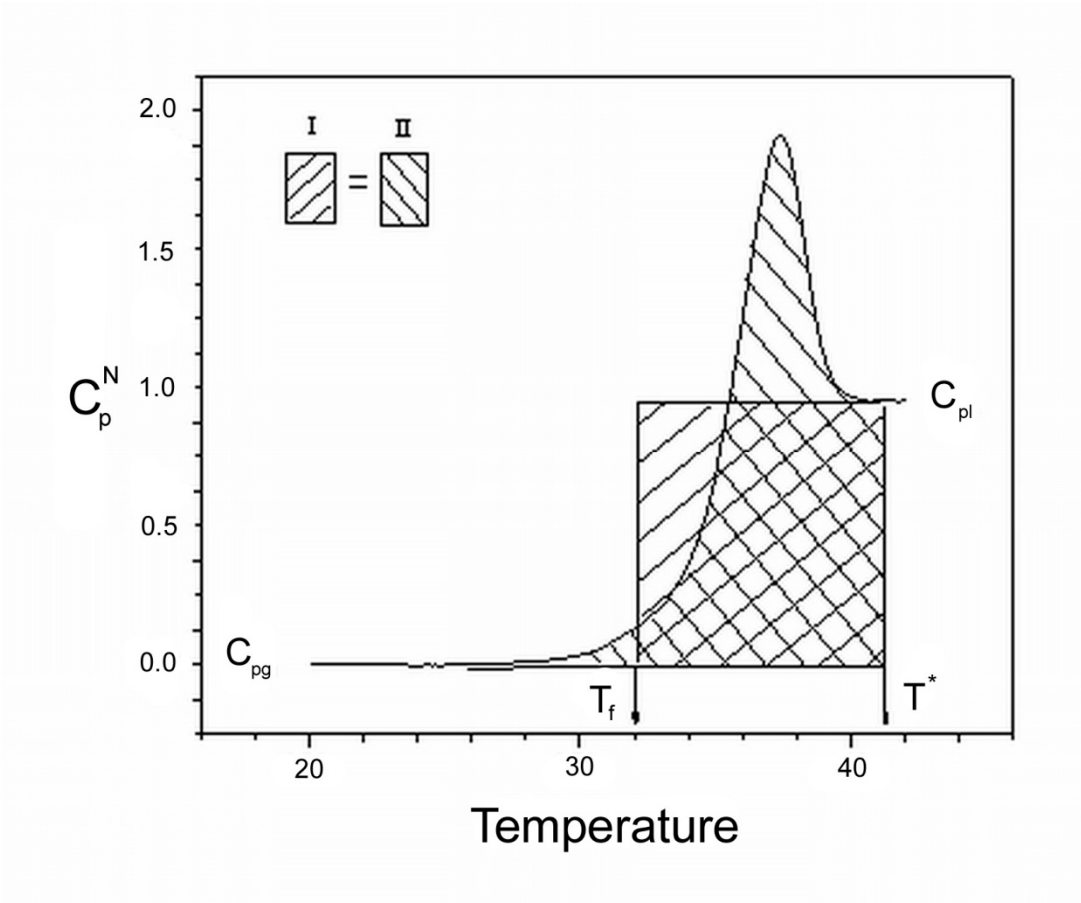


Table S2 – *The results of the sc-MKA method applied to the DSC crystallization data of NIF powders.*

50 – 125 μm						
q^+	0.5	1	2	5	10	20
$\log(A_1)$	16.61108	16.21726	16.07724	16.14603	16.13793	16.1286
E_1	114.3	114.3	114.3	114.3	114.3	114.3
N_1	0.71578	0.71685	0.6064	0.68519	0.68983	0.79353
M_1	1.08685	0.99073	0.95979	0.93186	0.93569	0.94401
$\log(A_2)$	14.43758	13.88043	13.93524	14.07167	13.91808	13.80624
E_2	104.9	104.9	104.9	104.9	104.9	104.9
N_2	0.78186	0.7479	1.13295	1.26169	0.92056	0.89957
M_2	0.83517	0.69899	0.73803	0.79265	0.7523	0.52812
$\log(A_3)$	15.84363	15.55615	15.62165	15.95344	15.71288	15.63753
E_3	114.3	114.3	114.3	114.3	114.3	114.3
N_3	0.22245	0.27432	0.40329	0.80225	0.63848	0.68753
M_3	0.54031	0.35569	0.5151	0.68013	0.60195	0.59076
$\Delta H_1/\Delta H$	0.32998	0.47018	0.49405	0.49332	0.31383	0.29648
$\Delta H_2/\Delta H$	0.48785	0.37548	0.35505	0.34149	0.30222	0.3928
ΔH	60.02713	46.53704	60.25898	63.58853	62.3003	68.49157
r	0.999283	0.999585	0.999242	0.999595	0.999925	0.999919

125 – 180 μm						
q^+	0.5	1	2	5	10	20
$\log(A_1)$	17.85998	17.73347	17.75048	17.72093	17.60956	17.63822
E_1	124	124	124	124	124	124
N_1	0.68043	0.72883	0.56739	0.5676	0.61335	0.77514
M_1	1.07068	0.98209	1.01537	0.98471	0.94121	0.97735
$\log(A_2)$	15.15411	15.02412	15.03406	15.15007	14.86001	14.53043
E_2	108.3	108.3	108.3	108.3	108.3	108.3
N_2	0.96341	0.96219	1.29088	1.45268	1.08735	0.93687
M_2	1.05626	0.9467	1.0064	1.03809	0.9236	0.6879
$\log(A_3)$	17.3237	17.53094	17.35824	17.19903	17.13645	17.10639
E_3	124	124	124	124	124	124
N_3	0.33101	0.47651	0.46852	0.45046	0.56235	0.66068
M_3	0.76354	0.8585	0.79139	0.62079	0.56778	0.57734
$\Delta H_1/\Delta H$	0.31812	0.33289	0.30173	0.33977	0.27667	0.22922
$\Delta H_2/\Delta H$	0.32352	0.34358	0.30394	0.28159	0.29465	0.36073
ΔH	61.5838	45.90207	56.707	63.54977	62.95117	70.21816
r	0.999396	0.999819	0.999248	0.999321	0.999614	0.999941

180 – 250 μm						
q^+	0.5	1	2	5	10	20
$\log(A_1)$	15.9333	15.91214	16.04718	16.04271	15.96967	15.84261
E_1	114.9	114.9	114.9	114.9	114.9	114.9
N_1	0.64929	0.57415	0.55472	0.64845	0.6739	0.72856
M_1	0.91187	0.97602	1.02611	1.03428	1.00153	0.94422
$\log(A_2)$	13.08192	13.02887	12.80204	12.77982	12.8267	12.6187
E_2	98.9	98.9	98.9	98.9	98.9	98.9
N_2	1.6848	1.92951	0.95247	0.99978	1.67649	0.91314
M_2	0.94458	0.98068	0.83802	0.83637	0.86688	0.67518
$\log(A_3)$	15.19651	15.53805	15.61423	15.47713	15.42557	15.43836
E_3	114.9	114.9	114.9	114.9	114.9	114.9
N_3	9.61E-02	0.53267	0.46286	0.54003	0.59236	0.77517
M_3	0.35289	0.73919	0.77259	0.64713	0.60521	0.62791
$\Delta H_1/\Delta H$	0.53311	0.40634	0.2221	0.22451	0.24107	0.19933
$\Delta H_2/\Delta H$	0.37859	0.34664	0.32895	0.31056	0.26705	0.28582
ΔH	41.5032	55.28769	58.35707	66.58177	64.14673	69.20402
r	0.995812	0.988324	0.994159	0.999481	0.999742	0.999872

250 – 300 μm						
q^+	0.5	1	2	5	10	20
$\log(A_1)$	16.55688	16.62673	16.70525	16.57555	16.4093	16.29358
E_1	118.1	118.1	118.1	118.1	118.1	118.1
N_1	0.66318	0.63224	0.76393	0.65537	0.96231	0.84582
M_1	0.96465	1.0244	1.0636	0.99175	0.95104	0.91974
$\log(A_2)$	14.29303	13.73655	14.2598	13.86518	13.48142	13.45624
E_2	102.5	102.5	102.5	102.5	102.5	102.5
N_2	1.35173	1.08031	2.61096	1.00295	0.83764	0.9829
M_2	1.13652	0.97191	1.15918	0.97879	0.82156	0.80632
$\log(A_3)$	16.21793	16.22497	16.28653	16.15718	16.08023	15.78885
E_3	118.1	118.1	118.1	118.1	118.1	118.1
N_3	0.56272	0.60701	0.77217	0.71259	1.05349	0.71633
M_3	0.80268	0.83076	0.83098	0.71569	0.61357	0.48534
$\Delta H_1/\Delta H$	0.35607	0.2899	0.31282	0.29095	0.3729	0.29065
$\Delta H_2/\Delta H$	0.17943	0.27634	0.31902	0.27345	0.25972	0.2761
ΔH	43.96593	52.27706	66.69253	63.1152	63.56093	68.62673
r	0.999965	0.999548	0.984491	0.999654	0.999959	0.999965

300 – 500 μm						
q^+	0.5	1	2	5	10	20
$\log(A_1)$	15.61577	15.64946	15.54124	15.42779	15.52643	15.40945
E_1	113.3	113.3	113.3	113.3	113.3	113.3
N_1	0.80745	0.7291	0.72166	0.62472	0.74299	0.8515
M_1	0.93496	0.93803	0.89804	0.83312	0.92286	0.90035
$\log(A_2)$	14.08228	13.39364	13.61231	13.34792	13.10659	13.17014
E_2	100.8	100.8	100.8	100.8	100.8	100.8
N_2	0.70373	1.07471	1.56803	0.92867	0.87385	1.1558
M_2	1.19906	0.96875	1.03623	0.91045	0.7779	0.86057
$\log(A_3)$	15.01724	15.4031	15.46926	15.29052	15.2079	15.01982
E_3	113.3	113.3	113.3	113.3	113.3	113.3
N_3	0.61782	1.64111	0.44423	0.86264	0.84911	0.82381
M_3	0.61814	0.74667	0.69827	0.68579	0.57208	0.54354
$\Delta H_1/\Delta H$	0.5233	0.57361	0.63701	0.31718	0.35316	0.26427
$\Delta H_2/\Delta H$	4.94E-02	0.24731	0.3217	0.2635	0.28645	0.25082
ΔH	47.38628	48.64121	61.95808	63.34755	68.94937	68.20955
r	0.999617	0.998836	0.989339	0.999866	0.999914	0.999977

Figure S2– graphical representation of the CR and CHR cyclic temperature programs (temperature program in the left graph and the corresponding derivative relaxation response in the right graph).

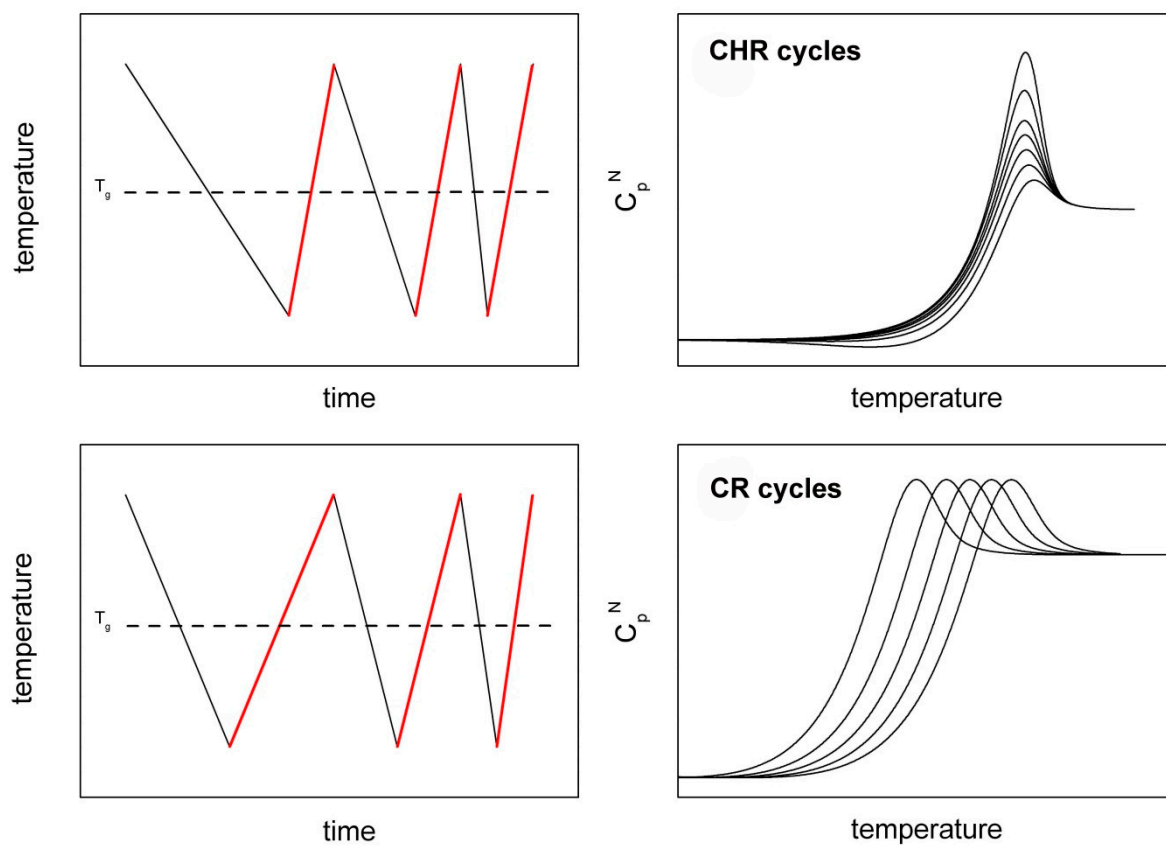
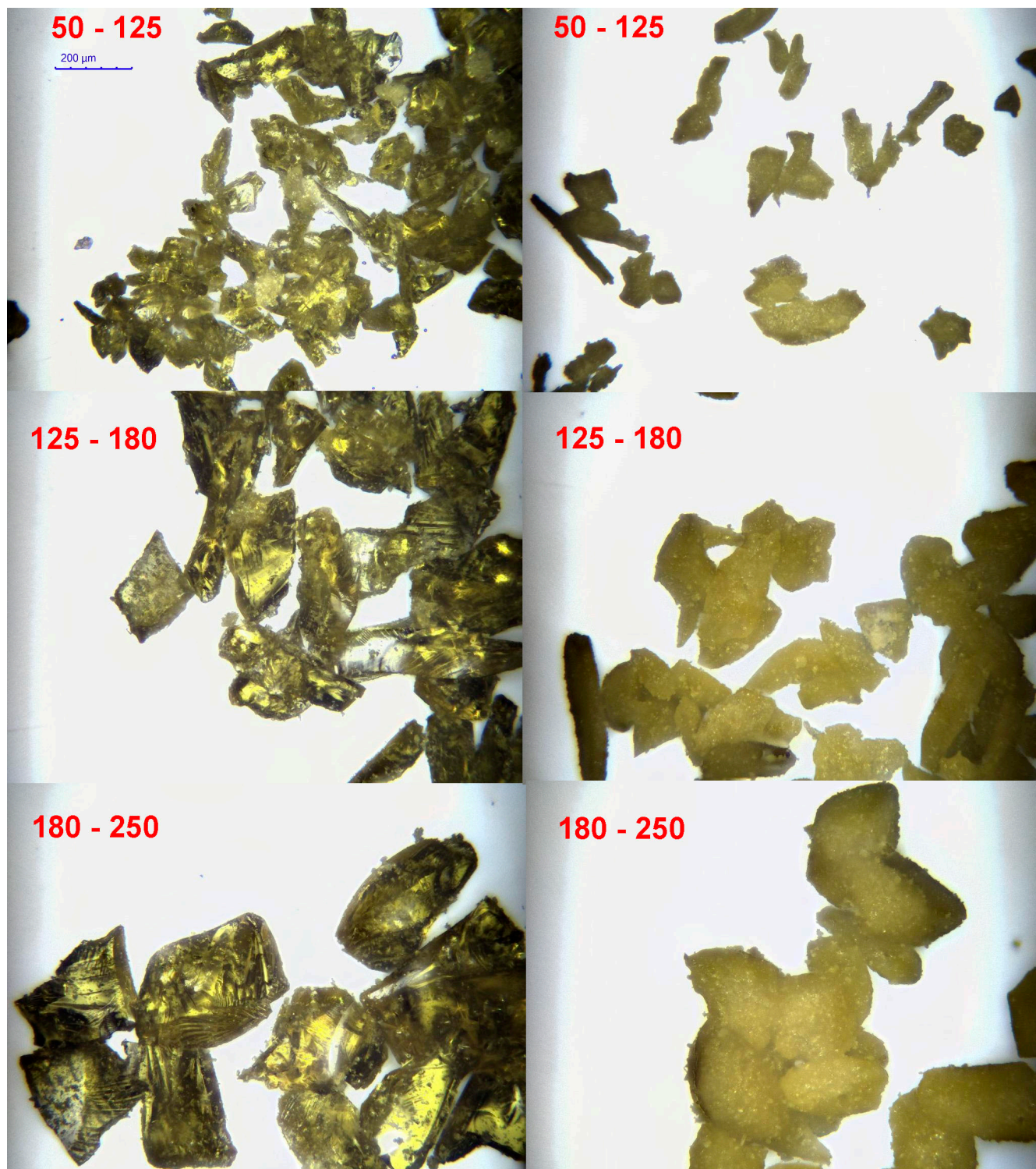
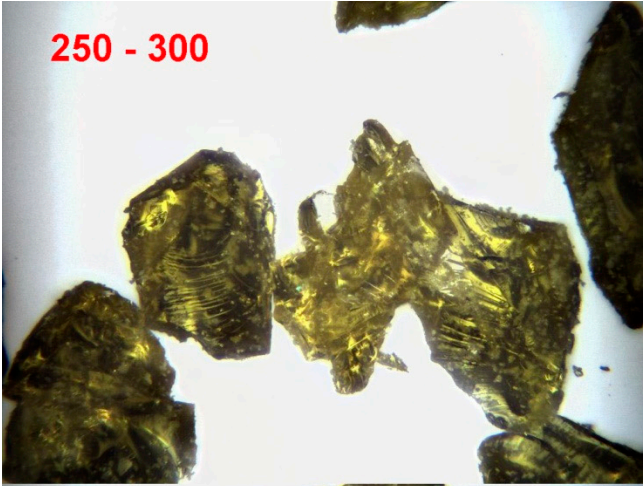


Figure S3– optical micrographs corresponding to the as-prepared (left columns) and DSC-crystallized (right columns) NIF powders. Identical scale bar as shown for the as-prepared 125 – 180 μm powder is valid for all other micrographs. The legends inserted into micrographs correspond to the powder size in μm .



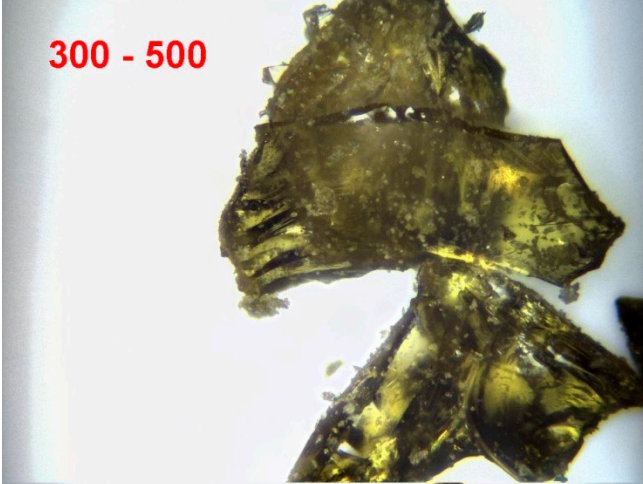
250 - 300



250 - 300



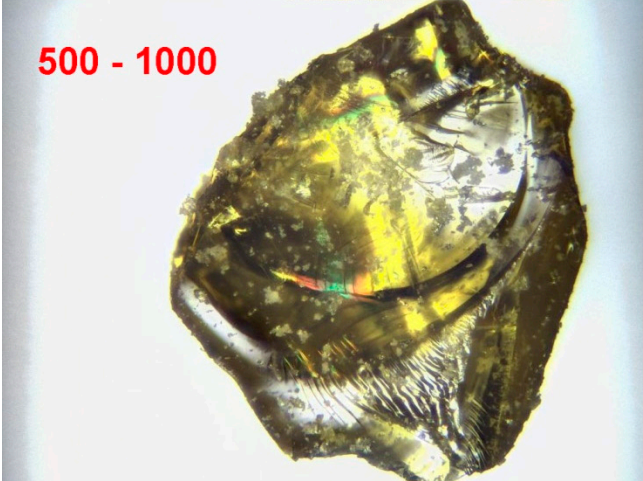
300 - 500



300 - 500



500 - 1000



500 - 1000



Figure S4— *alternative depictions of crystallization and melting enthalpies in $\text{J}\cdot\text{g}^{-1}$ (variants of Figs. 3E, 3F, 7C).*

