

Supplementary material for

Indirect photodegradation of sulfamethoxazole and trimethoprim by hydroxyl radical in aquatic environment: mechanisms, transformation products and ecotoxicity evaluation

Jiaoxue Yang¹, Guochun Lv¹, Chenxi Zhang², Zehua Wang¹, Xiaomin Sun^{1,*}

¹ Environment Research Institute, Shandong University, Qingdao 266237, China

² College of Biological and Environmental Engineering, Binzhou University,

Binzhou 256600, China

*Corresponding author: Xiaomin Sun, sxmwch@sdu.edu.cn

Figures and Table Captions

Figure. S1. The pathways of $\bullet\text{OH}$ -initiated reaction of SMX in aquatic environment with the Gibbs free energy barriers ΔE_b and reaction energies ΔE_r (unit: kcal mol⁻¹)

Figure. S2. The pathways of $\bullet\text{OH}$ -initiated reaction of TMP in aquatic environment with the Gibbs free energy barriers ΔE_b and reaction energies ΔE_r (unit: kcal mol⁻¹)

Table S1 Calculated rate constants (M⁻¹ s⁻¹) between 273 and 328 K in the reaction of SMX and $\bullet\text{OH}$

Table S2 Calculated rate constants (M⁻¹ s⁻¹) between 273 and 328 K in the reaction of TMP and $\bullet\text{OH}$

Table S3 The acute and chronic toxic criterion (unit: mg L⁻¹)

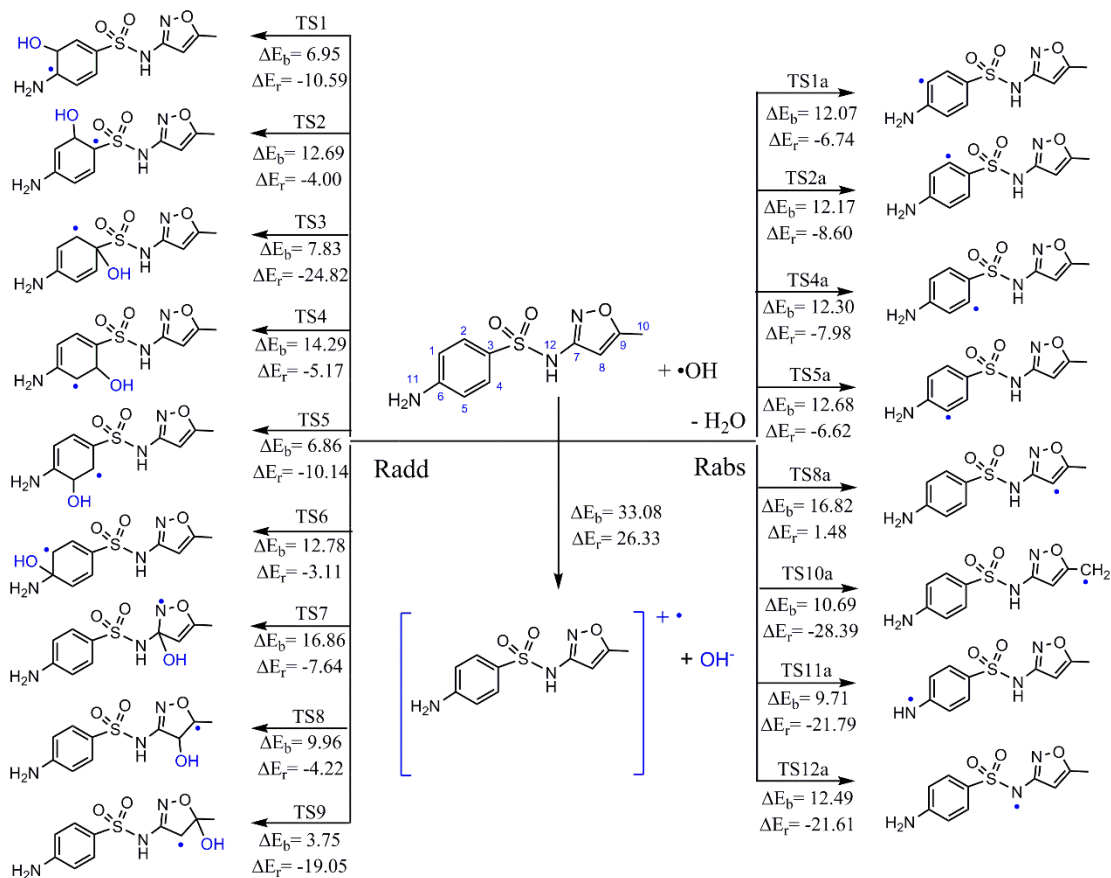


Figure. S1. The pathways of $\bullet\text{OH}$ -initiated reaction of SMX in aquatic environment with the Gibbs free energy barriers ΔE_b and reaction energies ΔE_r (unit: kcal mol⁻¹)

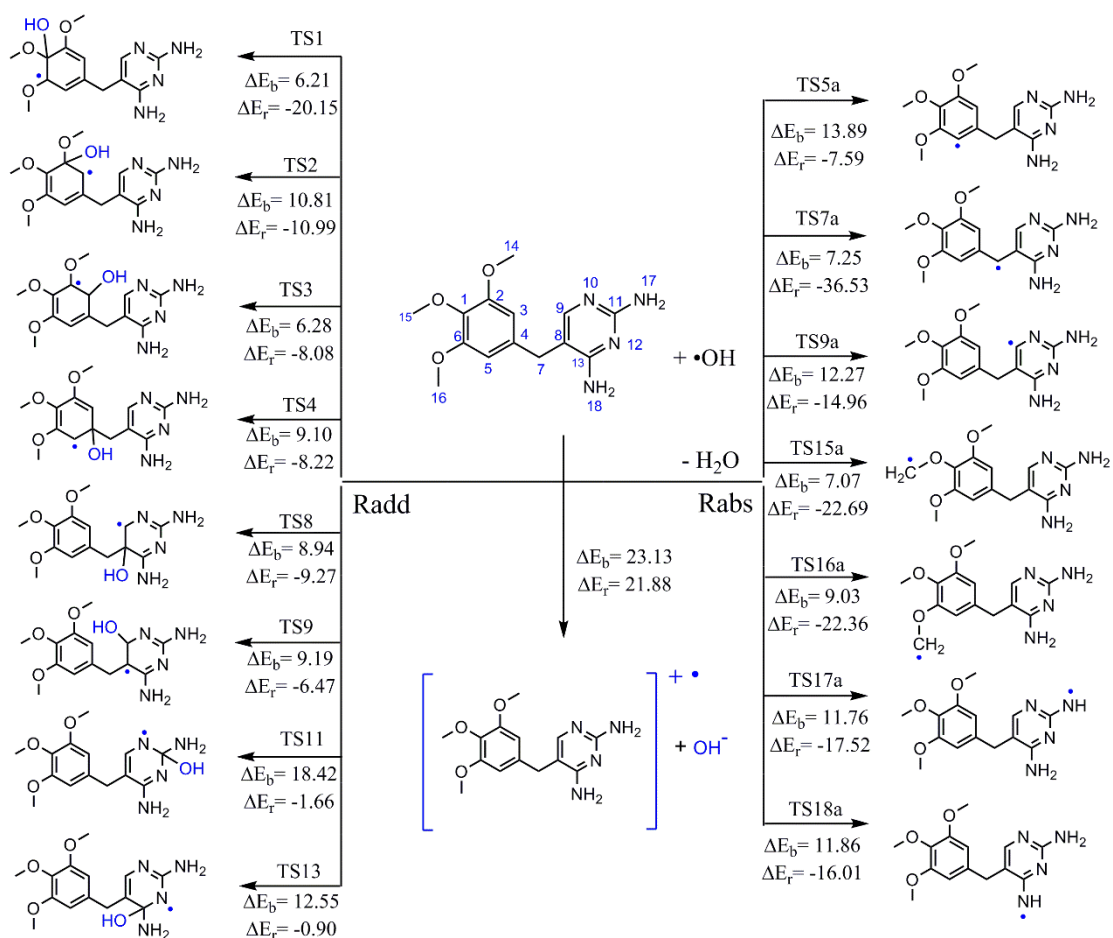


Figure. S2. The pathways of $\bullet\text{OH}$ -initiated reaction of TMP in aquatic environment with the Gibbs free energy barriers ΔE_b and reaction energies ΔE_r (unit: kcal mol^{-1})

Table S1 Calculated rate constants ($\text{M}^{-1} \text{s}^{-1}$) between 273 and 328 K in the reaction of SMX and $\bullet\text{OH}$

| T(K) | 273 | 288 | 298 | 313 | 328 |
|---------------------|--------------------|--------------------|----------------------------------|--------------------|--------------------|
| R _{add1} | 6.65×10^7 | 5.75×10^7 | 5.27×10^7 | 4.68×10^7 | 4.21×10^7 |
| R _{add2} | 2.07×10^3 | 3.08×10^3 | 3.94×10^3 | 5.54×10^3 | 7.56×10^3 |
| R _{add3} | 1.34×10^7 | 1.24×10^7 | 1.19×10^7 | 1.12×10^7 | 1.07×10^7 |
| R _{add4} | 1.32×10^2 | 2.05×10^2 | 2.68×10^2 | 3.91×10^2 | 5.51×10^2 |
| R _{add5} | 7.71×10^7 | 6.72×10^7 | 6.18×10^7 | 5.53×10^7 | 5.00×10^7 |
| R _{add6} | 1.82×10^3 | 2.59×10^3 | 3.22×10^3 | 4.35×10^3 | 5.73×10^3 |
| R _{add7} | 1.47 | 2.64 | 3.78 | 6.19 | 9.73 |
| R _{add8} | 3.04×10^5 | 3.26×10^5 | 3.40×10^5 | 3.62×10^5 | 3.84×10^5 |
| R _{add9} | 1.73×10^5 | 1.85×10^5 | 1.93×10^5 | 2.05×10^5 | 2.17×10^5 |
| R _{abs1a} | 1.30×10^4 | 1.87×10^4 | 2.34×10^4 | 3.19×10^4 | 4.24×10^4 |
| R _{abs2a} | 2.09×10^4 | 2.42×10^4 | 2.66×10^4 | 3.02×10^4 | 3.40×10^4 |
| R _{abs4a} | 1.22×10^4 | 1.57×10^4 | 1.83×10^4 | 2.28×10^4 | 2.77×10^4 |
| R _{abs5a} | 4.64×10^3 | 6.74×10^3 | 8.47×10^3 | 1.16×10^4 | 1.56×10^4 |
| R _{abs8a} | 2.29 | 5.40 | 9.11 | 18.8 | 36.4 |
| R _{abs10a} | 1.35×10^5 | 1.61×10^5 | 1.79×10^5 | 2.09×10^5 | 2.40×10^5 |
| R _{abs11a} | 7.59×10^5 | 8.43×10^5 | 8.99×10^5 | 9.85×10^5 | 1.07×10^6 |
| R _{abs12a} | 2.05×10^4 | 2.54×10^4 | 2.89×10^4 | 3.45×10^4 | 4.06×10^4 |
| k _{total} | 1.58×10^8 | 1.39×10^8 | 1.28×10^8 | 1.15×10^8 | 1.05×10^8 |
| experiment | | | (5.8 ± 0.2) $\times 10^9$ | | |

Table S2 Calculated rate constants ($\text{M}^{-1} \text{s}^{-1}$) between 273 and 328 K in the reaction of
TMP and $\bullet\text{OH}$

| T(K) | 273 | 288 | 298 | 313 | 328 |
|---------------------|--------------------|--------------------|--------------------|--------------------|--------------------|
| R _{add1} | 2.89×10^8 | 2.17×10^8 | 1.82×10^8 | 1.43×10^8 | 1.16×10^8 |
| R _{add2} | 7.16×10^4 | 7.90×10^4 | 8.41×10^4 | 9.18×10^4 | 9.98×10^4 |
| R _{add3} | 2.58×10^8 | 1.91×10^8 | 1.59×10^8 | 1.23×10^8 | 9.82×10^7 |
| R _{add4} | 1.39×10^6 | 1.43×10^6 | 1.46×10^6 | 1.51×10^6 | 1.55×10^6 |
| R _{add8} | 2.91×10^6 | 2.74×10^6 | 2.40×10^6 | 1.95×10^6 | 1.77×10^6 |
| R _{add9} | 1.32×10^6 | 1.31×10^6 | 1.30×10^6 | 1.29×10^6 | 1.29×10^6 |
| R _{add11} | 0.07 | 0.17 | 0.29 | 0.61 | 1.18 |
| R _{add13} | 3.03×10^3 | 3.92×10^3 | 4.59×10^3 | 5.72×10^3 | 6.99×10^3 |
| R _{abs5a} | 6.19×10^2 | 9.35×10^2 | 1.20×10^3 | 1.71×10^3 | 2.36×10^3 |
| R _{abs7a} | 4.89×10^7 | 4.19×10^7 | 3.81×10^7 | 3.36×10^7 | 3.00×10^7 |
| R _{abs9a} | 2.71×10^3 | 2.63×10^3 | 2.50×10^3 | 2.39×10^3 | 2.27×10^3 |
| R _{abs15a} | 1.12×10^8 | 8.99×10^7 | 7.86×10^7 | 6.53×10^7 | 5.53×10^7 |
| R _{abs16a} | 2.20×10^6 | 2.32×10^6 | 2.40×10^6 | 2.52×10^6 | 2.65×10^6 |
| R _{abs17a} | 2.99×10^4 | 3.65×10^4 | 4.12×10^4 | 4.89×10^4 | 5.70×10^4 |
| R _{abs18a} | 4.02×10^4 | 4.91×10^4 | 5.54×10^4 | 6.53×10^4 | 7.59×10^4 |
| k _{total} | 9.72×10^8 | 7.35×10^8 | 6.21×10^8 | 4.94×10^8 | 4.03×10^8 |
| experiment | | | 8.66×10^9 | | |

Table S3 The acute and chronic toxic criterion (unit: mg L⁻¹)

| Acute toxicity ¹ | Chronic toxicity ² | Toxicity grade |
|--|-------------------------------|----------------|
| LC ₅₀ >100 or EC ₅₀ >100 | ChV>10 | Not harmful |
| 10< LC ₅₀ <100 or 10< EC ₅₀ <100 | 1<ChV<10 | Harmful |
| 1< LC ₅₀ <10 or 1< EC ₅₀ <10 | 0.1<ChV<1 | Toxic |
| LC ₅₀ <1 or EC ₅₀ <1 | ChV<0.1 | Very toxic |

¹ Criteria set by the European Union (described in Annex VI of Directive 67/548/EEC)

² Criteria set by the Chinese hazard evaluation guidelines for new chemical substances (HJ/T 154–2004)