

Supplementary materials

Coordination chemistry of phosphate groups in systems including copper(II) ions, phosphoethanolamine and pyrimidine nucleotides

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The set of EPR, UV-Vis, ^{13}C NMR, ^{31}P NMR spectra of the MLH_4L^- type complexes as well as corresponding to the free ligand ^{13}C NMR and ^{31}P NMR spectra

1. **Cu(enP)H₄(TMP) pH=2.5**
2. **Cu(enP)H₄(UMP) pH=3.0**
3. **Cu(enP)H₄(CMP) pH=2.5**

Cu(enP)H₄(TMP) pH=2.5

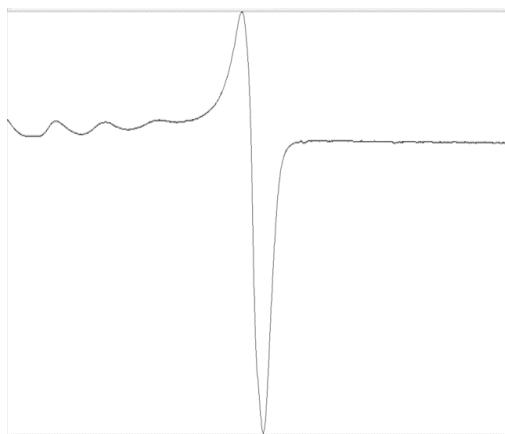


Figure S1. EPR spectrum

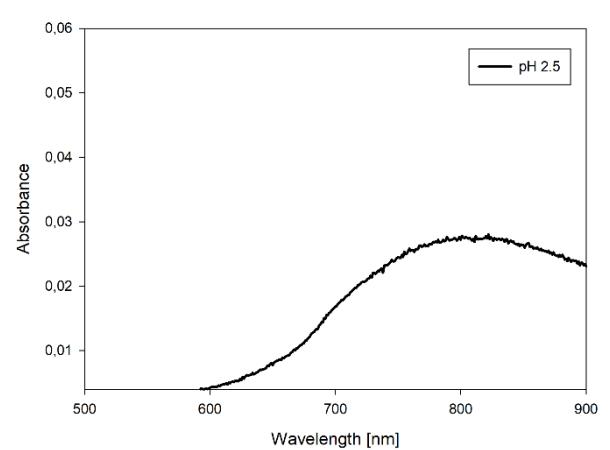


Figure S2. UV-Vis spectrum (C=0.002 M)

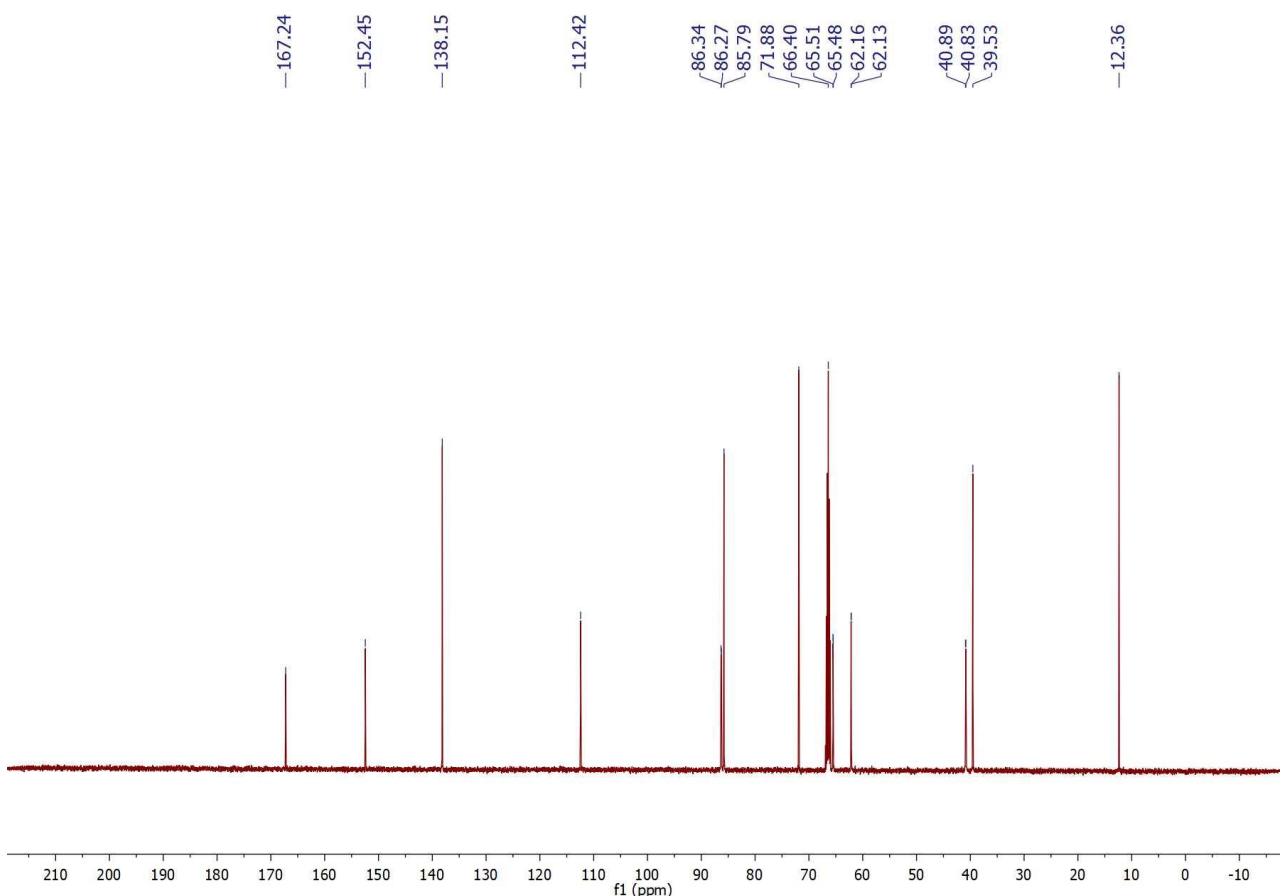


Figure S3. Cu(enP)H₄(TMP) ¹³ C NMR spectrum

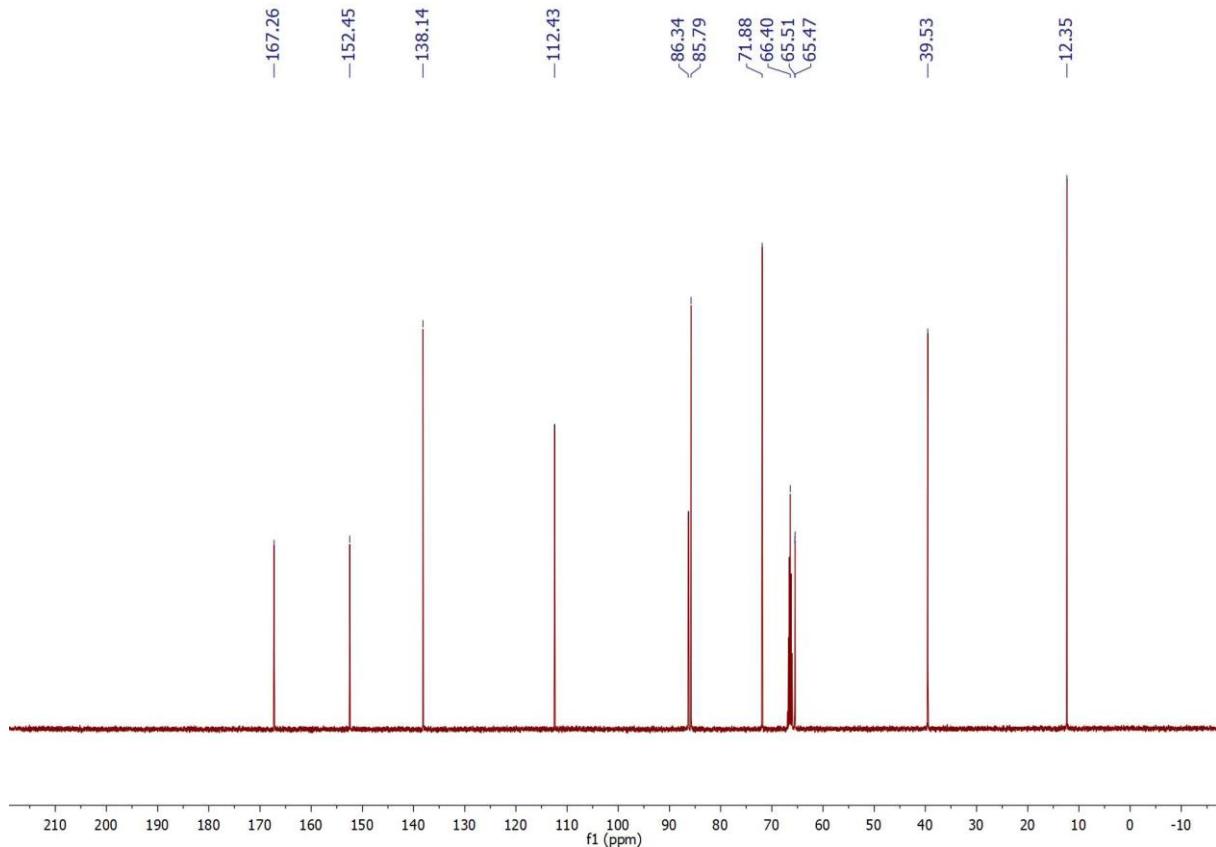


Figure S4. TMP ^{13}C NMR spectrum

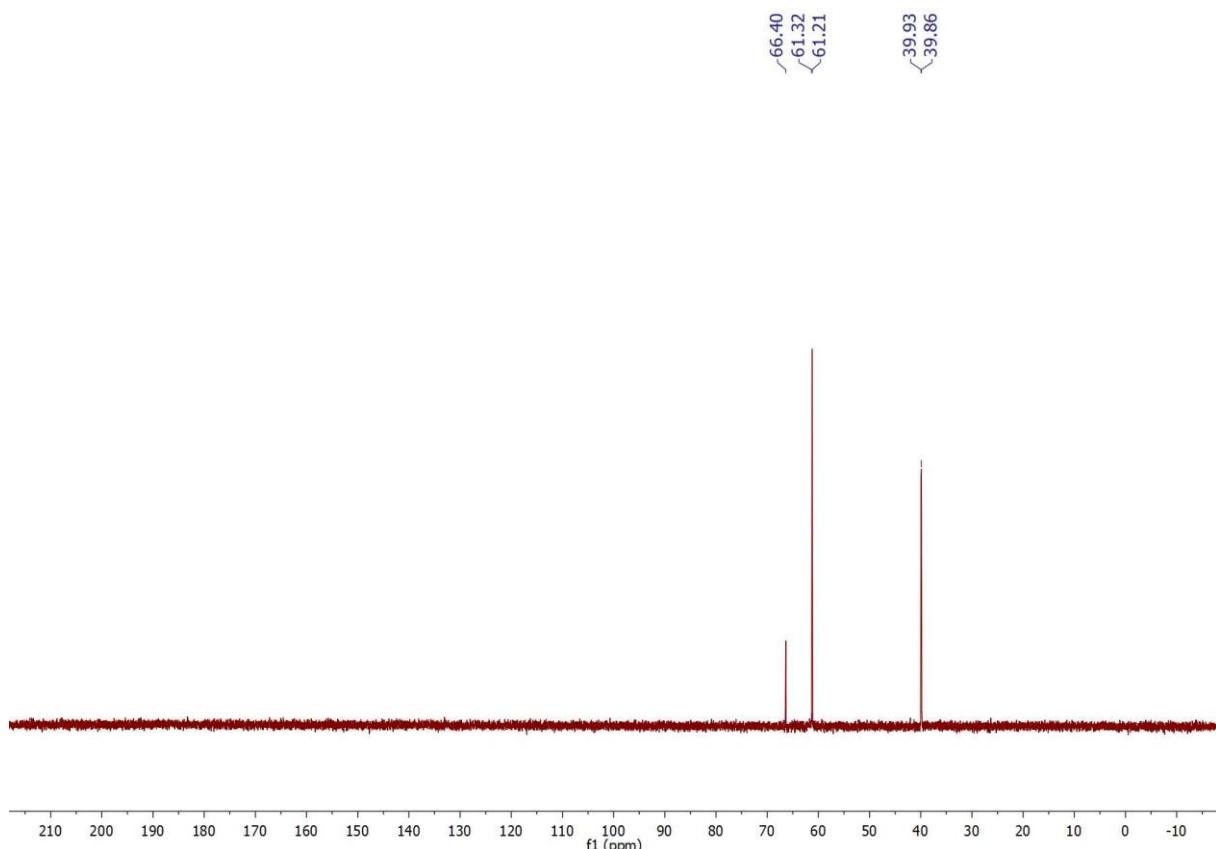


Figure S5. enP ^{13}C NMR spectrum

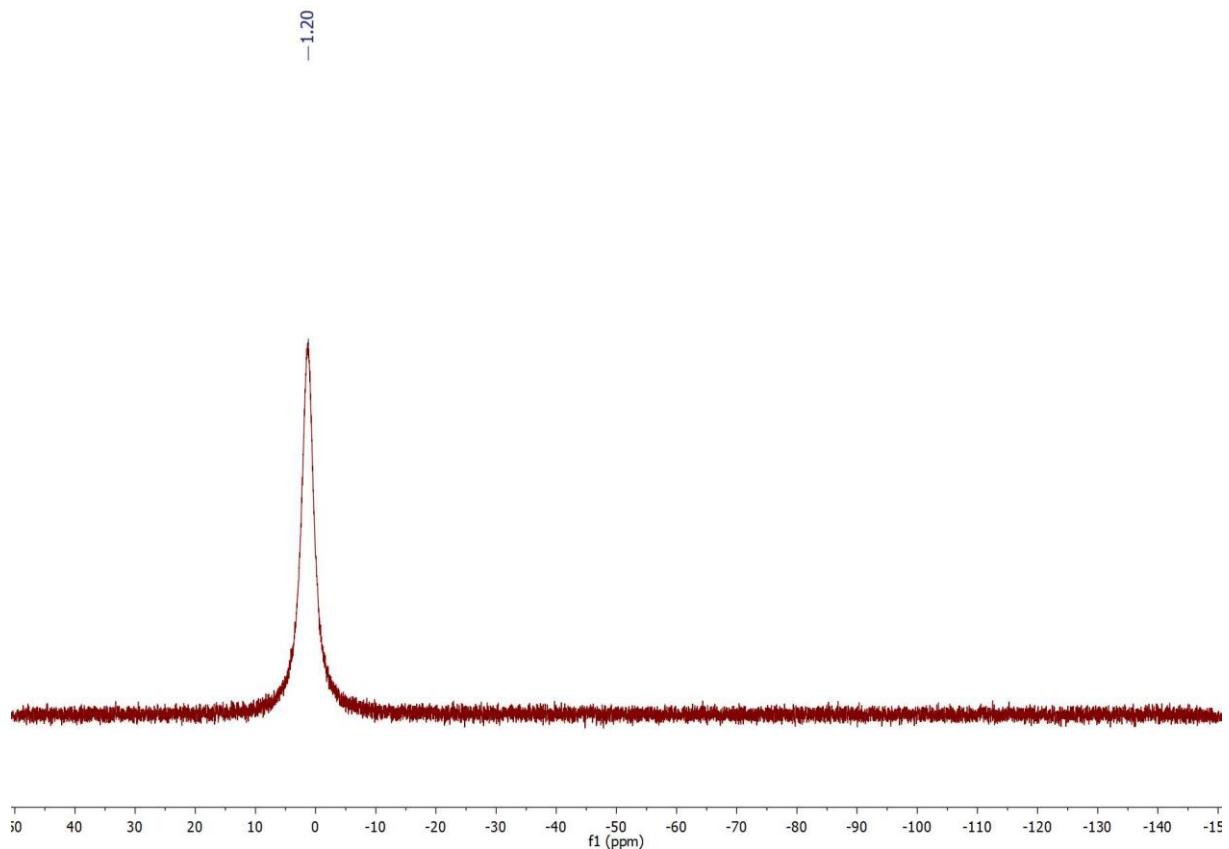


Figure S6. $\text{Cu}(\text{enP})\text{H}_4(\text{TMP})$ ^{31}P NMR spectrum

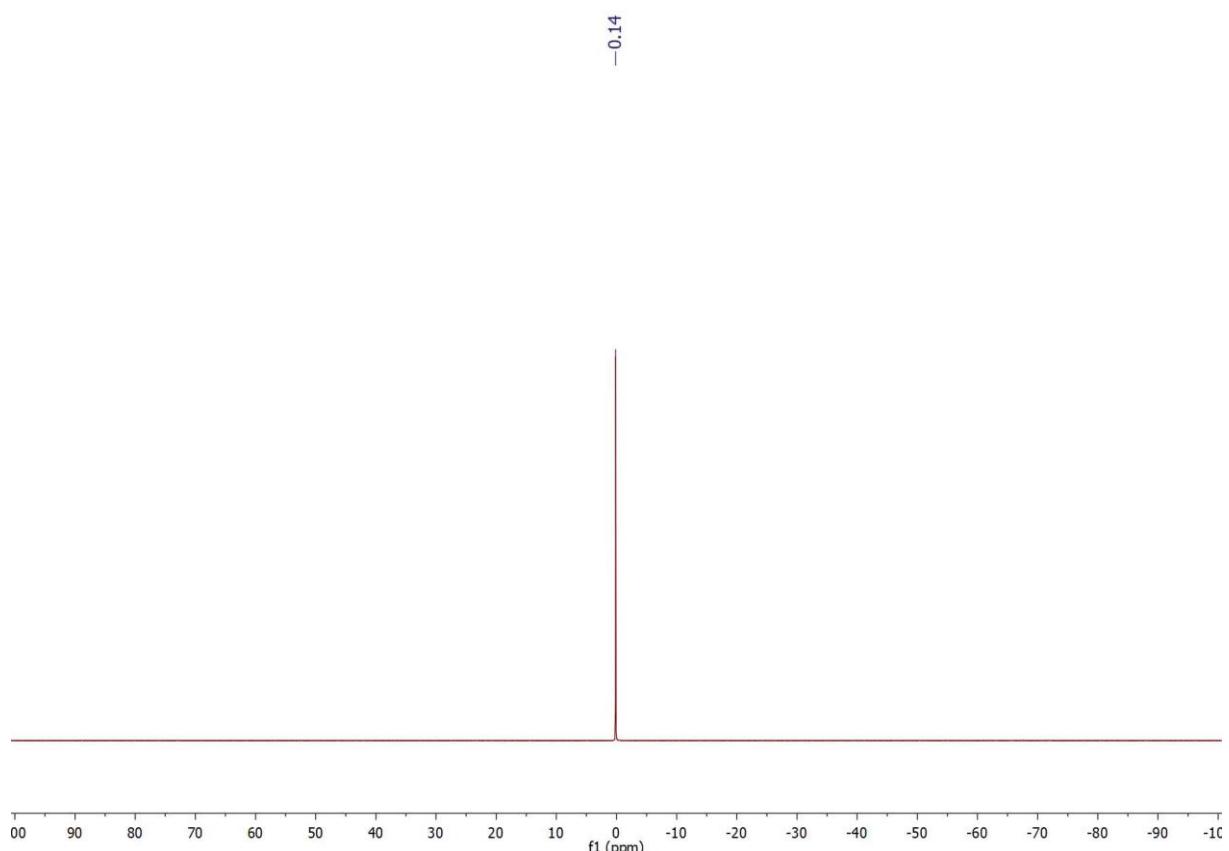


Figure S7. TMP ^{31}P NMR spectrum

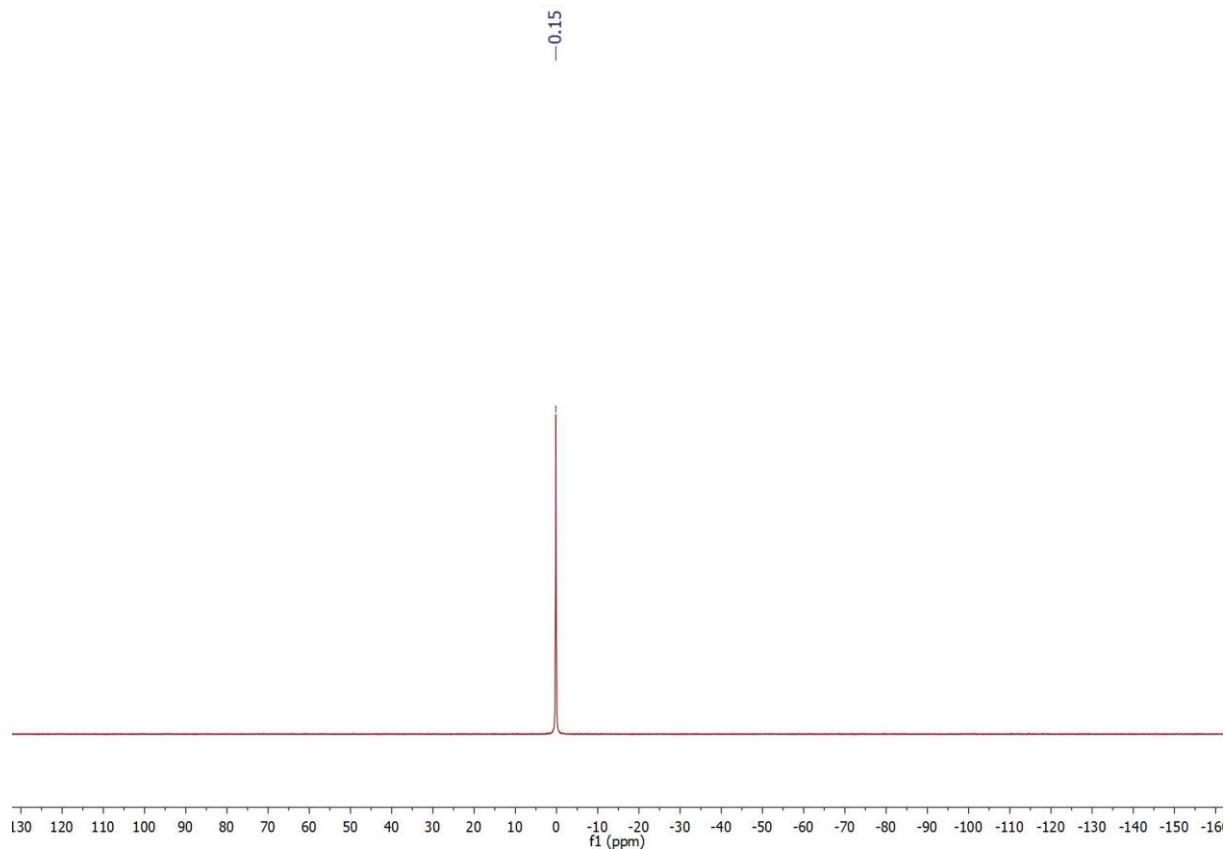


Figure S8. enP ^{31}P NMR spectrum

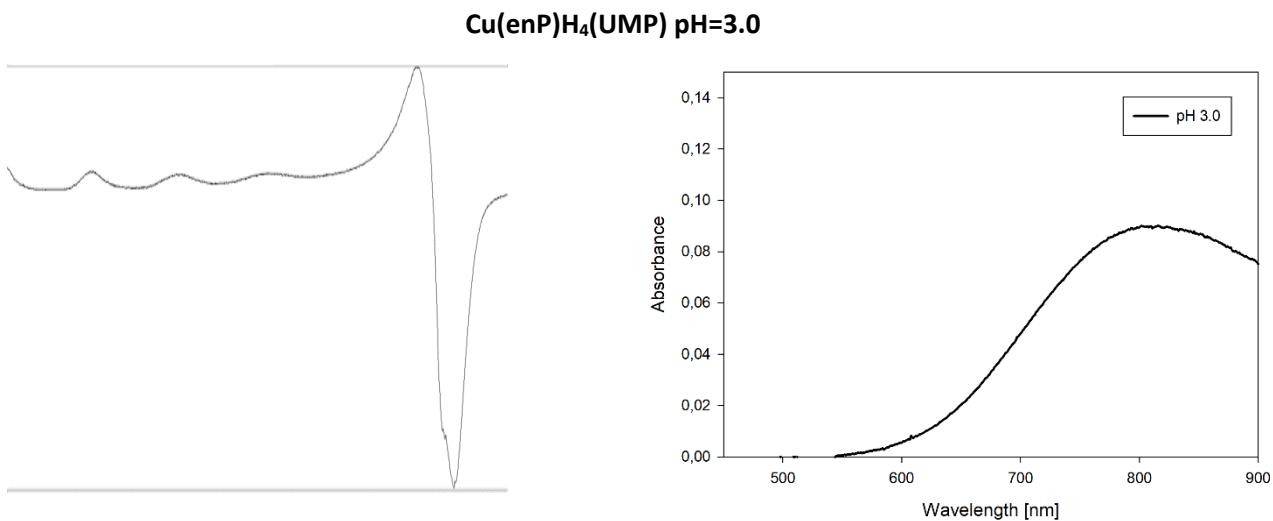


Figure S9. EPR spectrum

Figure S10. UV-Vis spectrum (C=0.01 M)

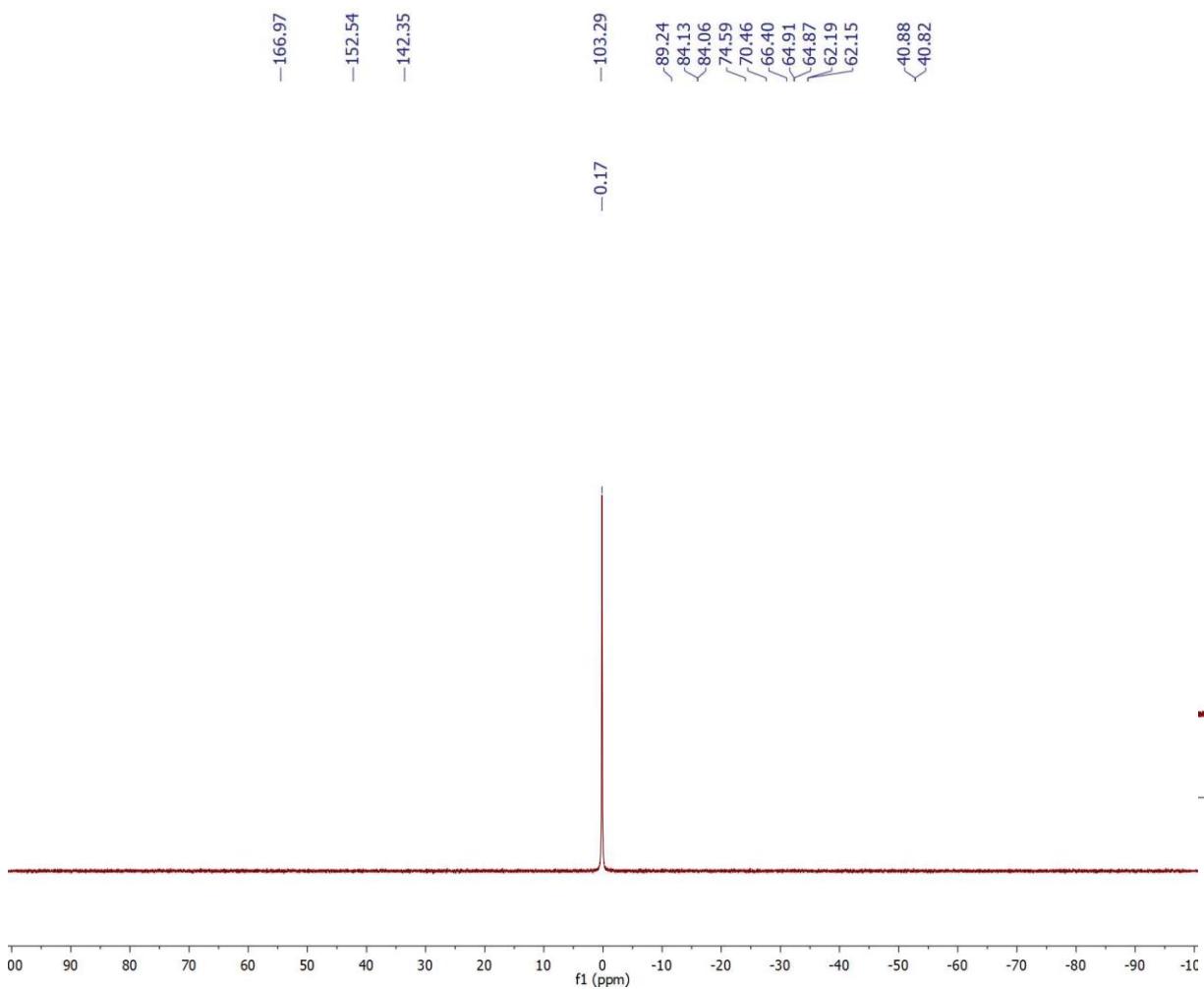


Figure S11. Cu(enP)H₄(UMP) ¹³ C NMR spectrum

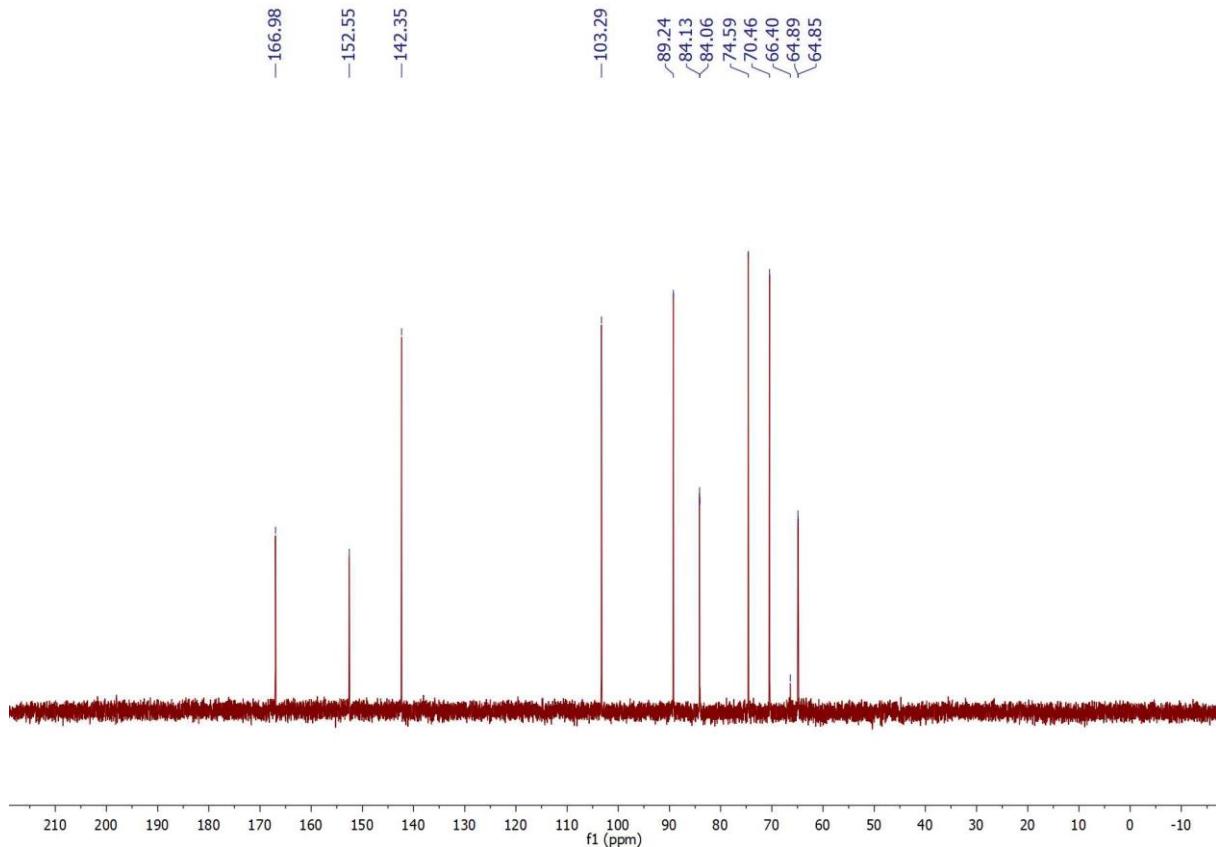


Figure S12. UMP ^{13}C NMR spectrum

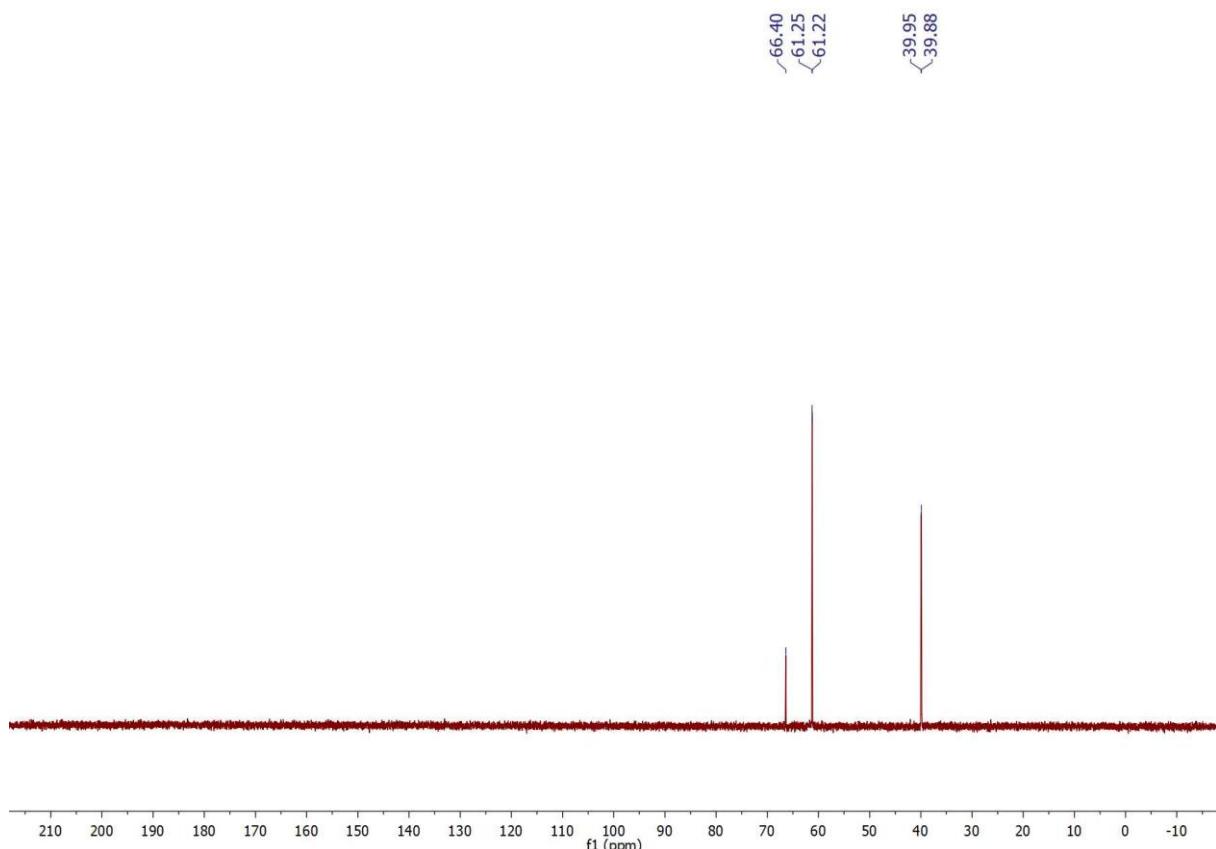


Figure S13. enP ^{13}C NMR spectrum

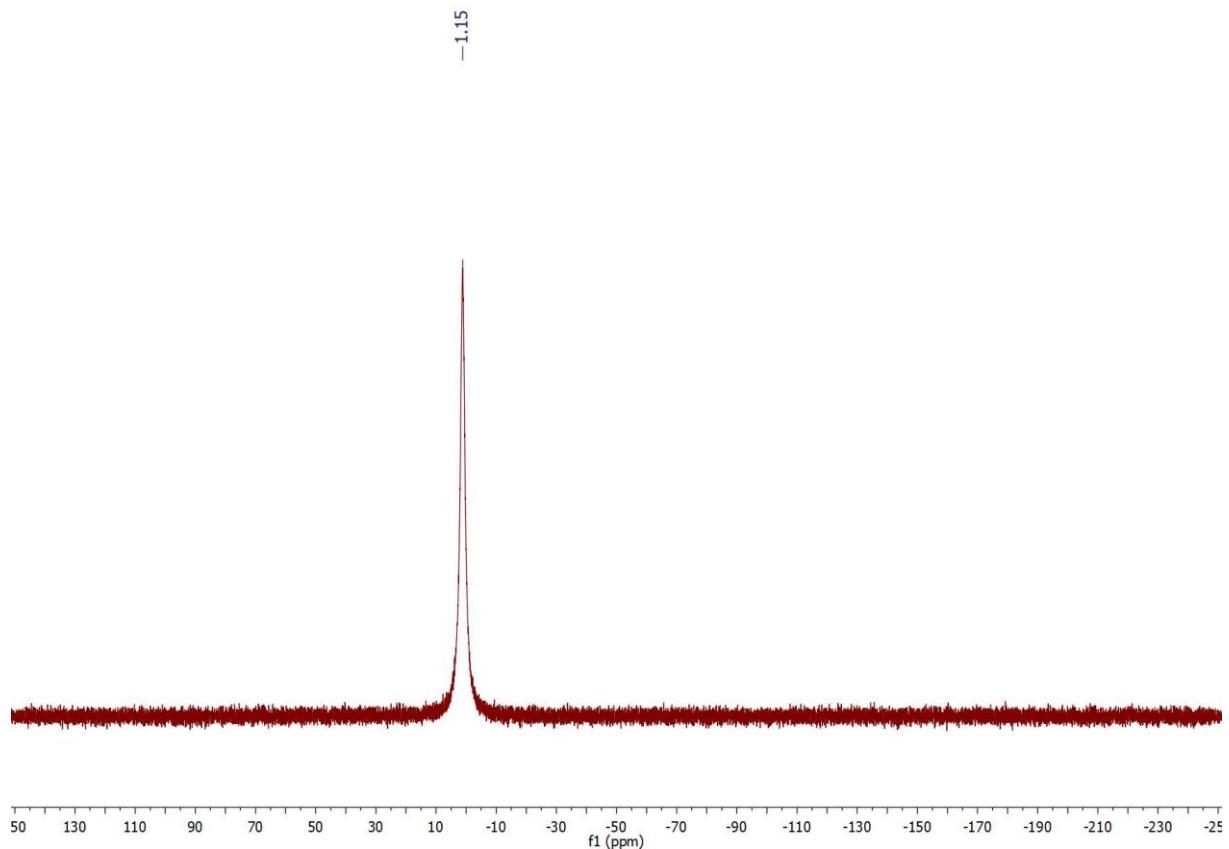


Figure S14. $\text{Cu}(\text{enP})\text{H}_4(\text{UMP})$ ^{31}P NMR spectrum

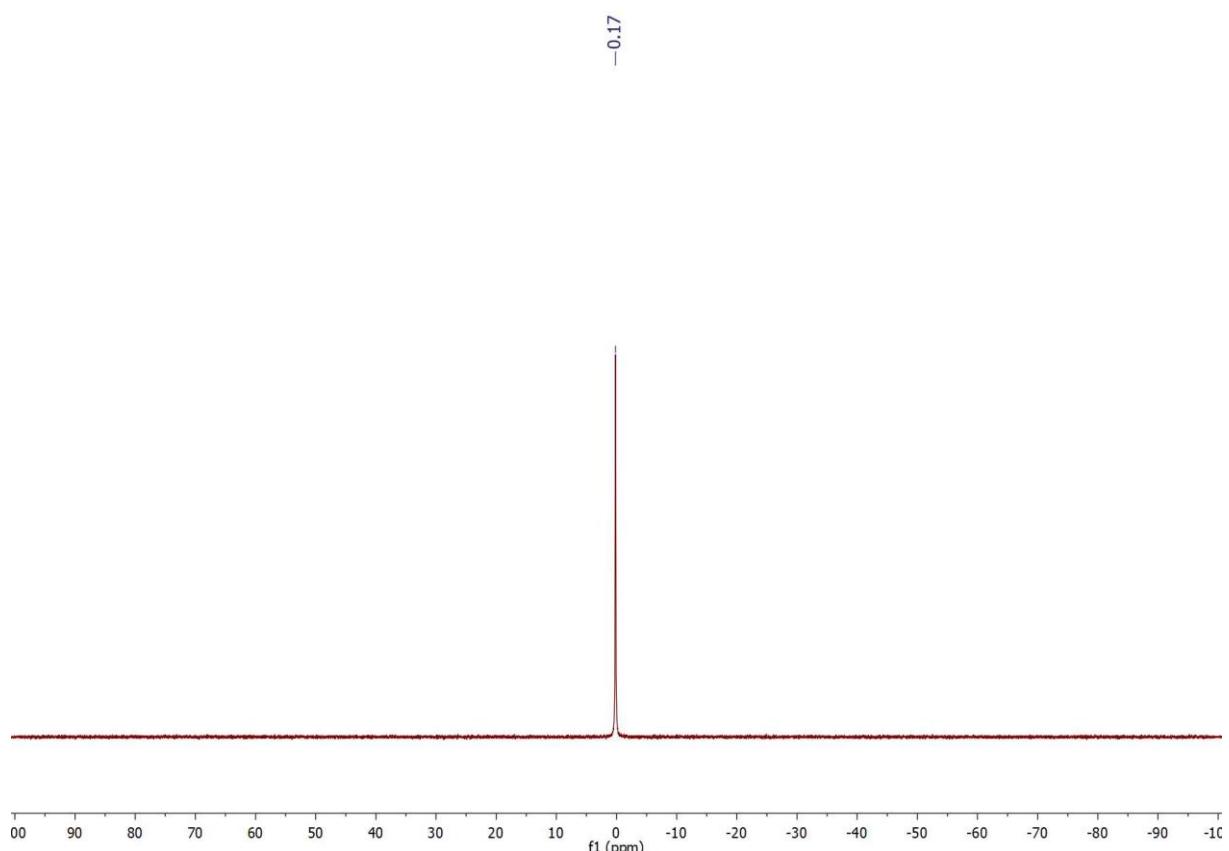


Figure S15. UMP ^{31}P NMR spectrum

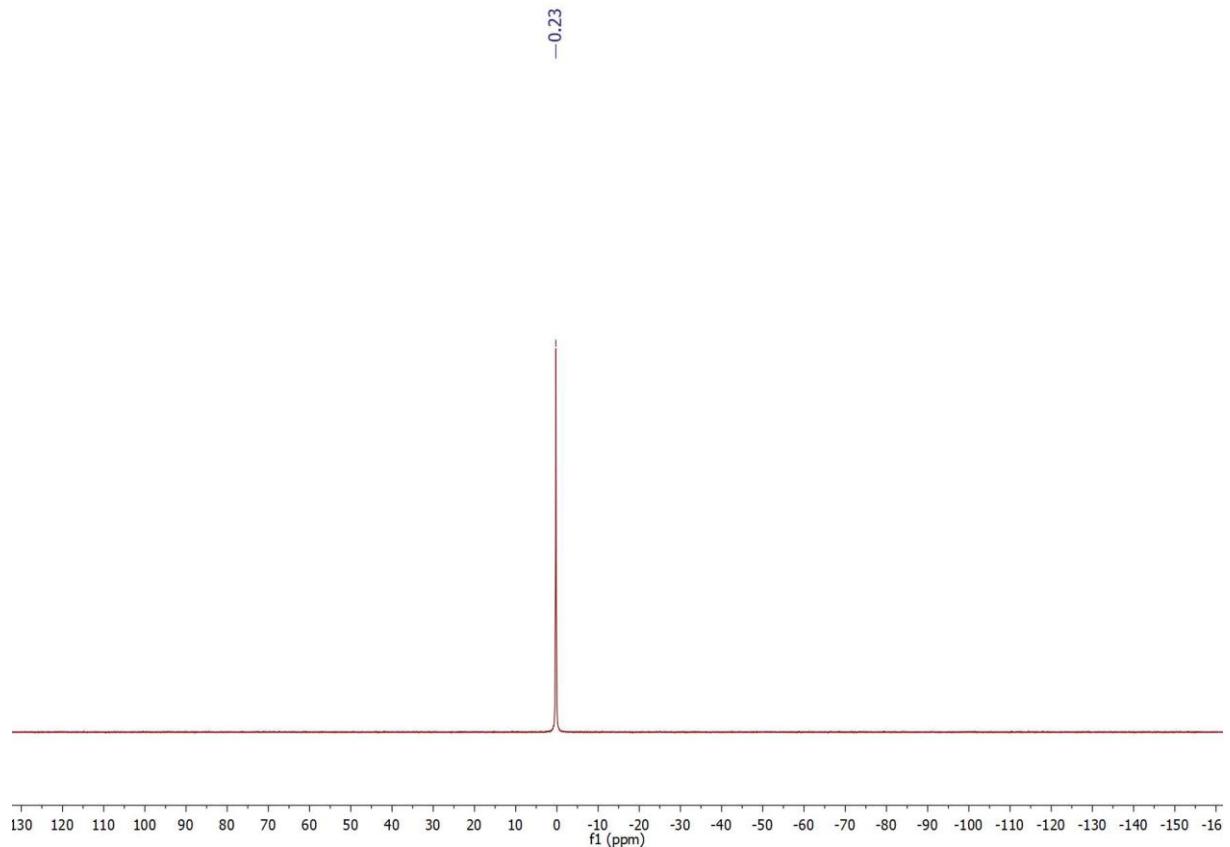


Figure S16. enP ^{31}P NMR spectrum

Cu(enP)H₄(CMP) pH = 2.5

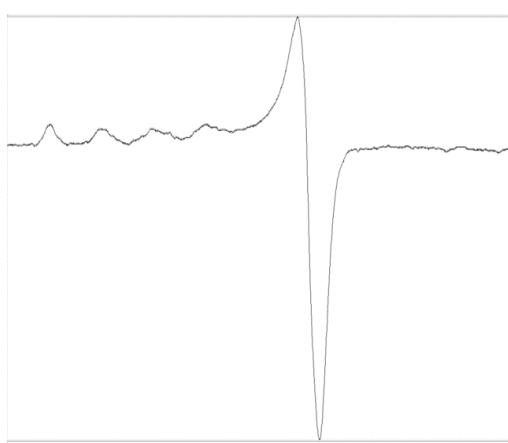


Figure S17. EPR spectrum

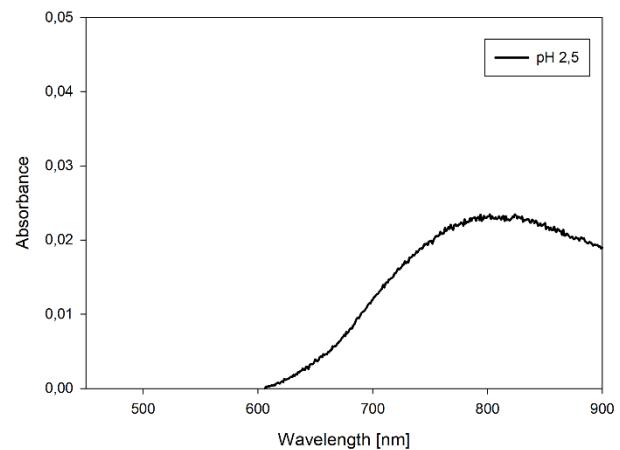


Figure S18. UV-Vis spectrum (C=0.002 M)

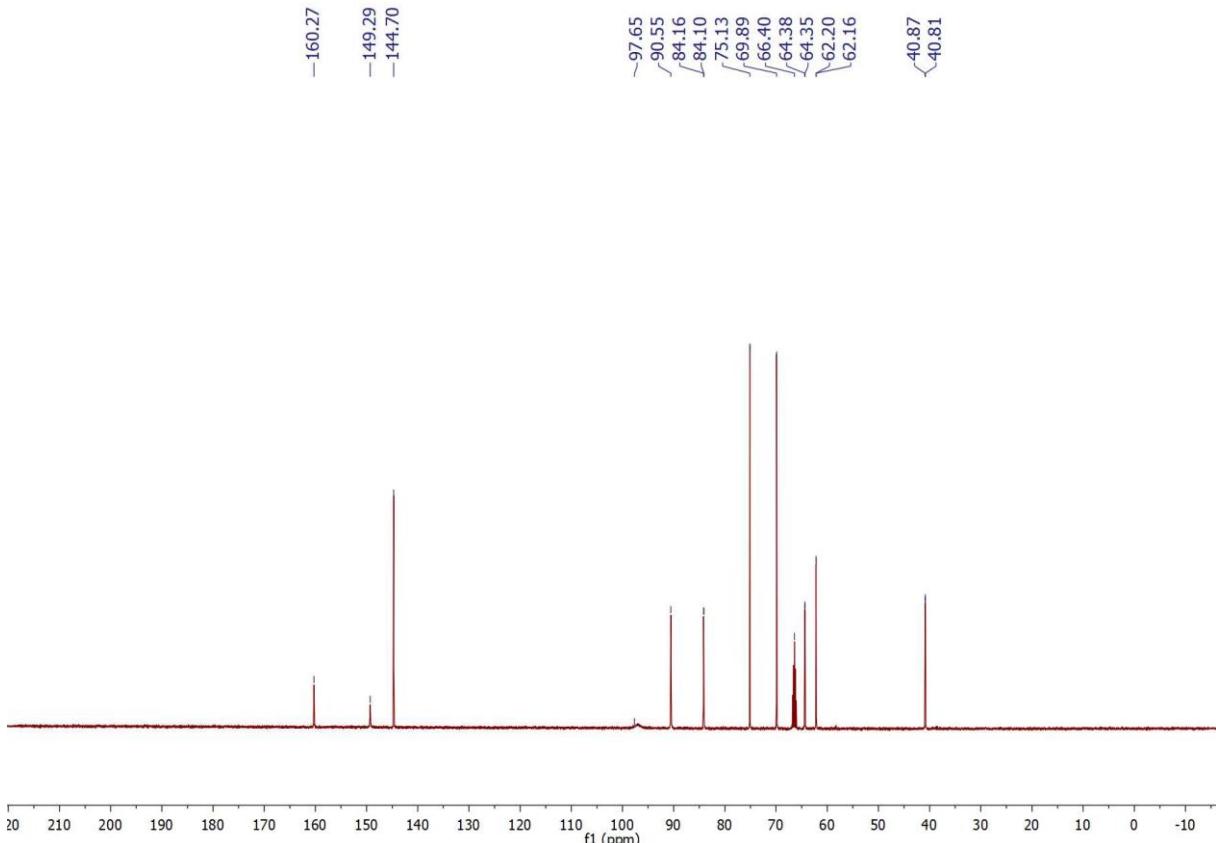


Figure S19. Cu(enP)H₄(CMP) ¹³ C NMR spectrum

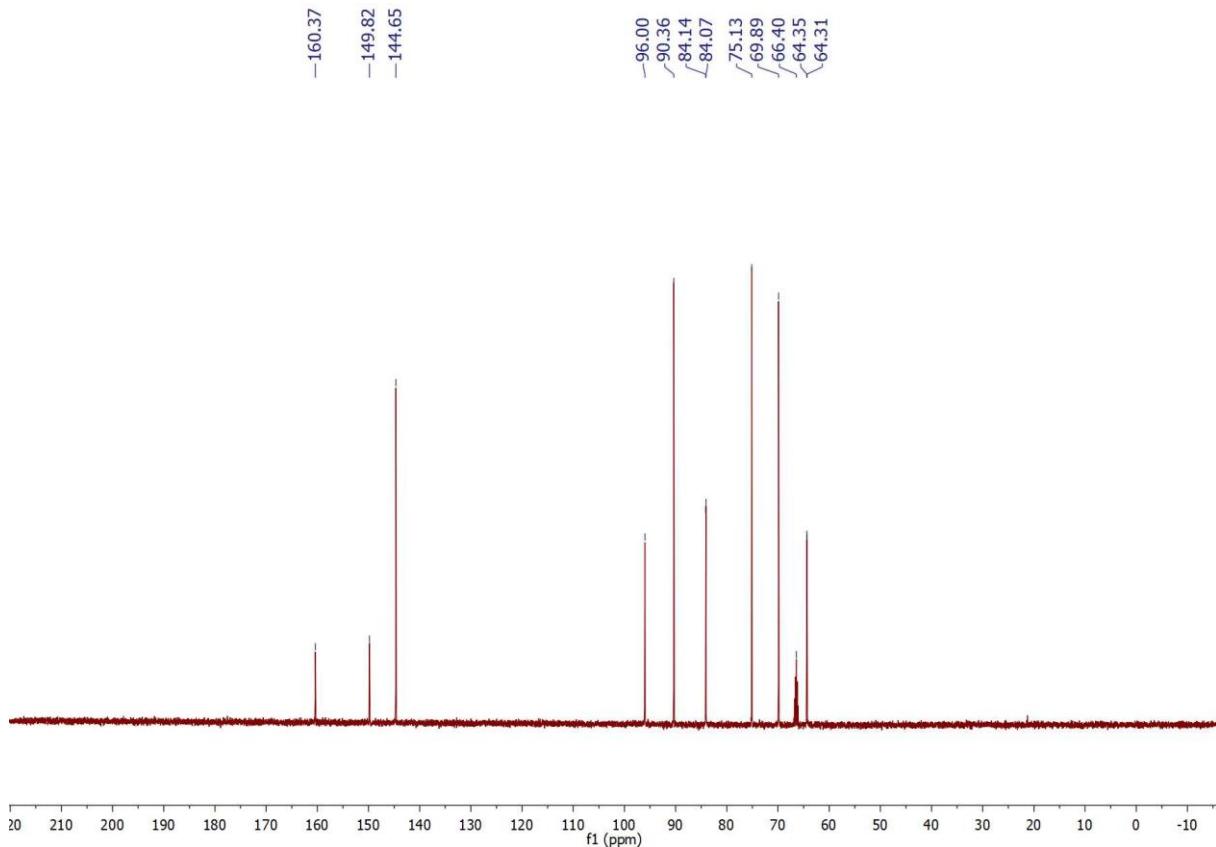


Figure S20. CMP ^{13}C NMR spectrum

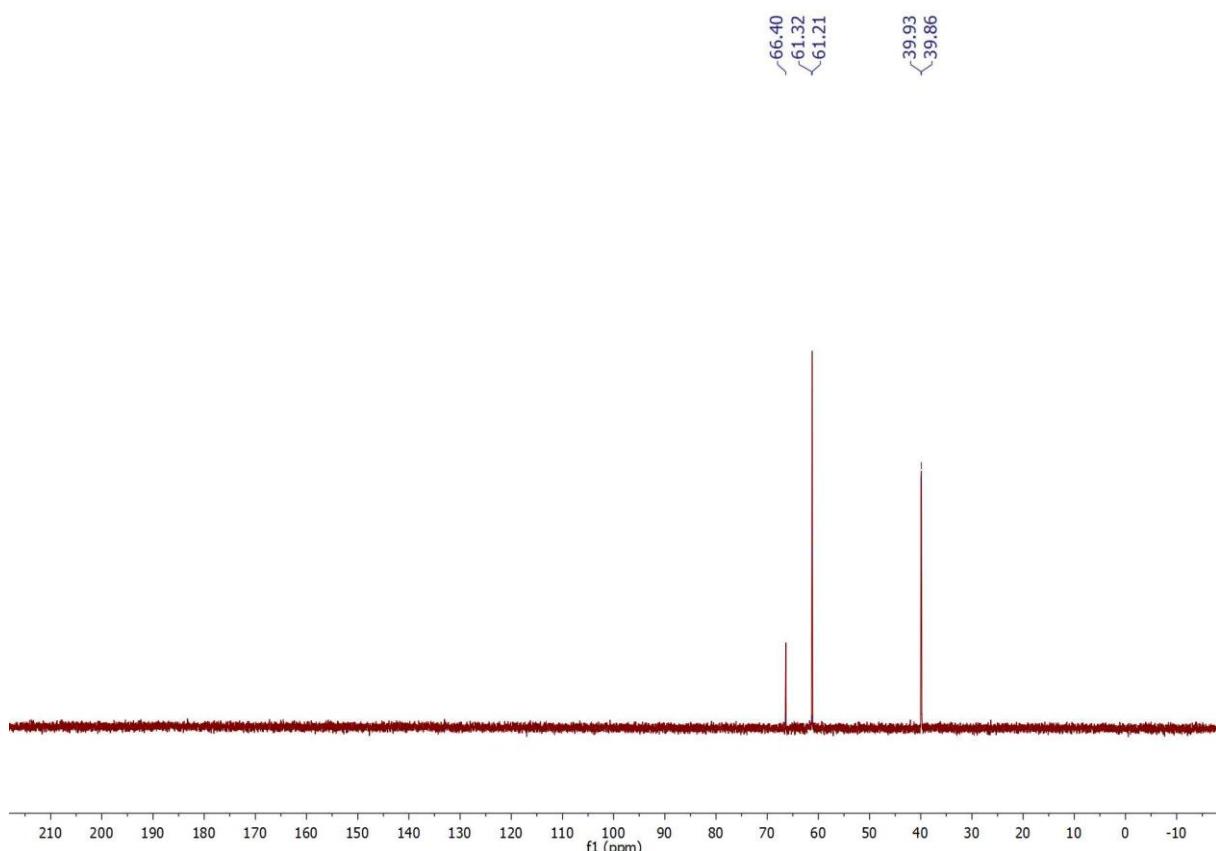


Figure S21. enP ^{13}C NMR spectrum

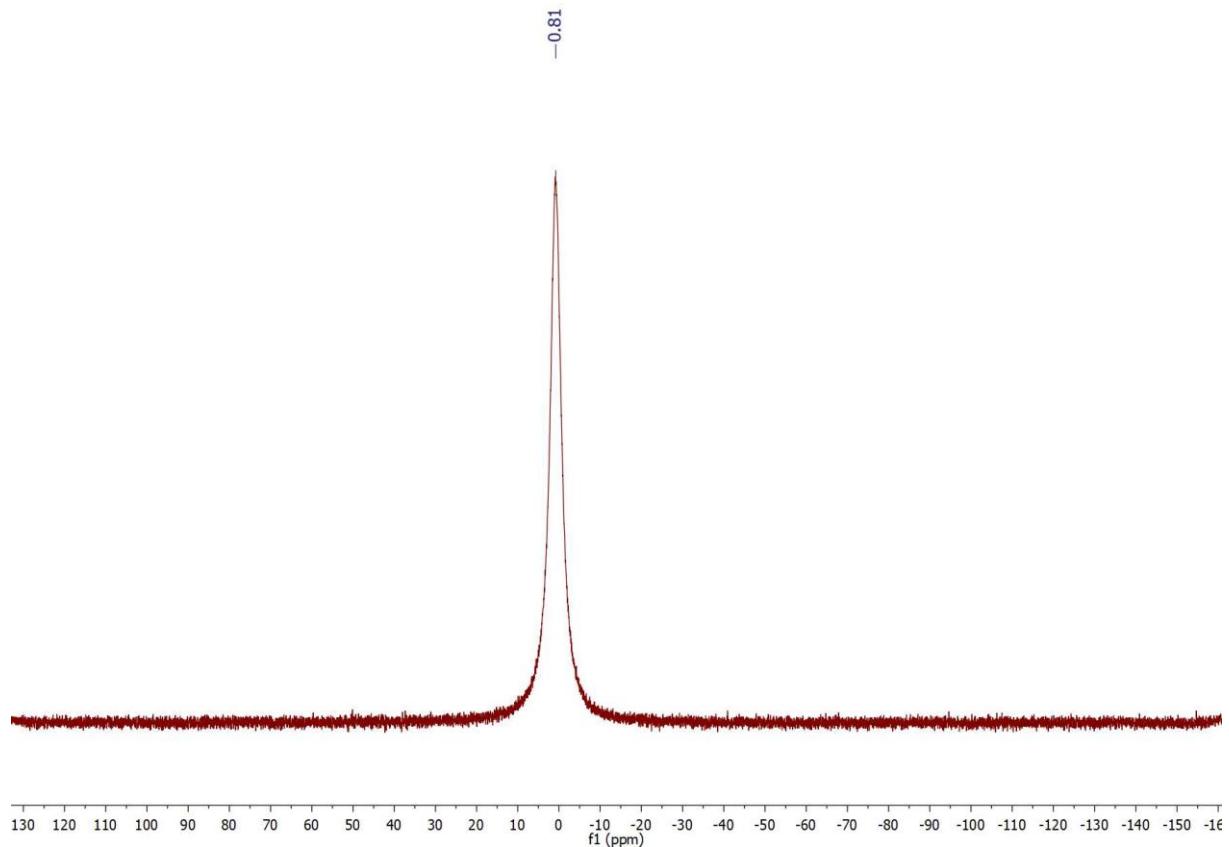


Figure S22. $\text{Cu}(\text{enP})\text{H}_4(\text{CMP})$ ^{31}P NMR spectrum

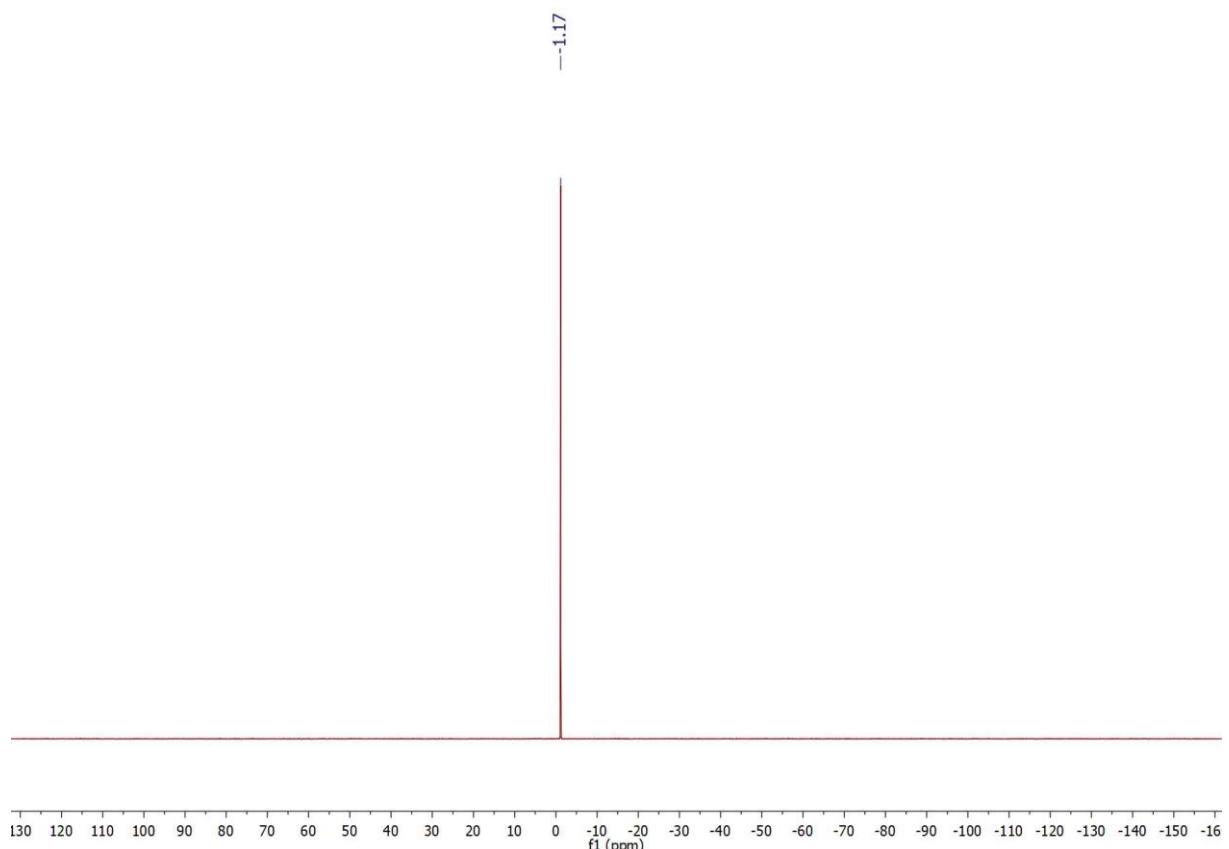


Figure S23. CMP ^{31}P NMR spectrum

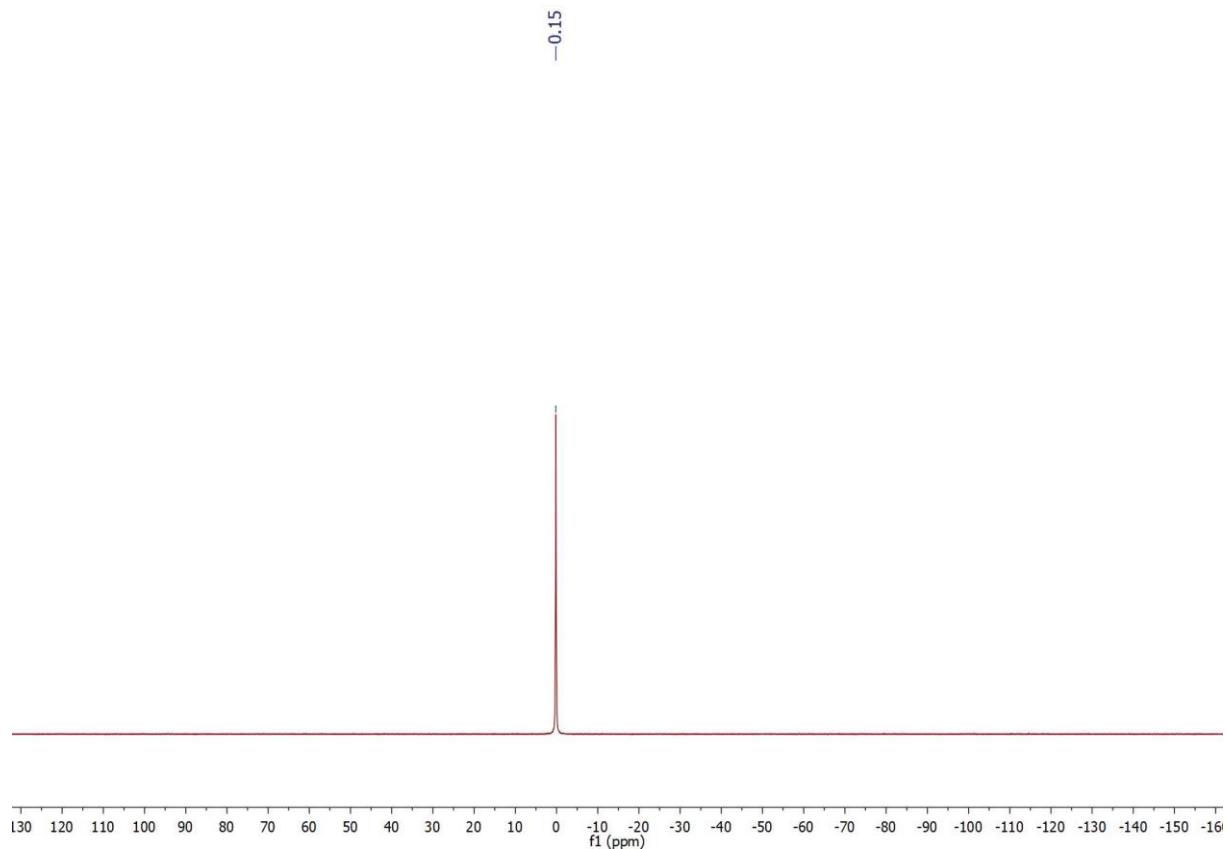


Figure S24. enP ^{31}P NMR spectrum