

Supporting information

A Sensor Array Realized by a Single Flexible TiO₂/POMs Film to Contactless Detection of Triacetone Triperoxide

Xiaorong Lü ¹, Puqi Hao ¹, Guanshun Xie ¹, Li Gao ^{1,*}, Junyuan Duan ^{2,*} and Bingxin Liu ^{1,*}

¹ Qinghai Provincial Key Laboratory of New Light Alloys, Qinghai Provincial Engineering Research Center of High Performance Light Metal Alloys and Forming, Qinghai University, Xining 810016, China; jwplxr@sina.com (X.L.); pheghost@outlook.com (P.H.); guanshunxie@126.com (G.X.)

² State Key Laboratory of Material Processing and Die & Mould Technology, School of Material Sciences and Engineering, Huazhong University of Science and Technology (HUST), Wuhan 430074, China

* Correspondence: love_lier@163.com (L.G.); junyuanduan@sina.com (J.D.); liubx408@nenu.edu.cn (B.L.); Tel.: +86-1869-723-5043 (B.L.)

Table S1. The available techniques for detection of TATP.

Detection techniques	Sensing materials	Limits of detection	Response time	Ref
Semiconductor based vapor sensor	organic-semiconductor sensors	<100 ppb	30 s	[1]
	SnO ₂ and WO ₃	12 ppb	-	[2]
	metal oxide catalyst	8 ppm	2 min	[3]
	In ₂ O ₃ nanoparticles	2.9 ppb	120 s	[4]
Ion mobility spectrometry	-	1.2 ng	-	[5]
Colorimetric method	colorimetric sensor	< 2 ppb	-	[6]
	silver nanoparticles	20 nM	-	[7]
	N,N-dimethyl-p-phenylene diamine	0.1 mg·L ⁻¹	-	[8]
Electrochemical method	Molecularly-imprinted polymer	26.9 µg·L ⁻¹	-	[9]
	Fe ^{II/III} ethylenediaminetetraacetate	0.89	-	[10]

Table S2. The atomic percentage results of TiO₂/PW₁₁ from XPS.

Name	Peak BE	FWHM eV	Area(P) CPS.eV	Atomic %
W4f	35.39	1.19	61905.31	3.94
Ti2p	458.56	1.15	123995.77	17.6
O1s	530.04	1.47	154800.67	52.12
P2p	133.1	0.38	2507.5	1.38

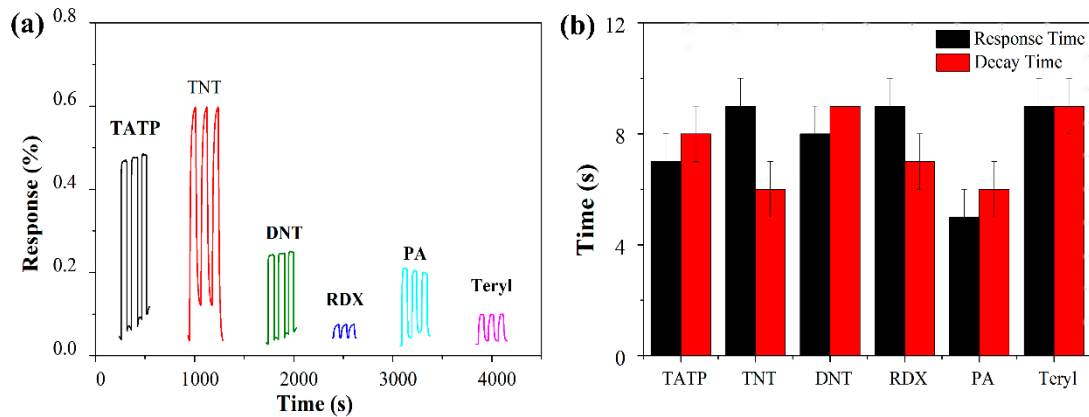


Figure S1. Response and response time and decay time of TiO₂ under 365 nm illumination.

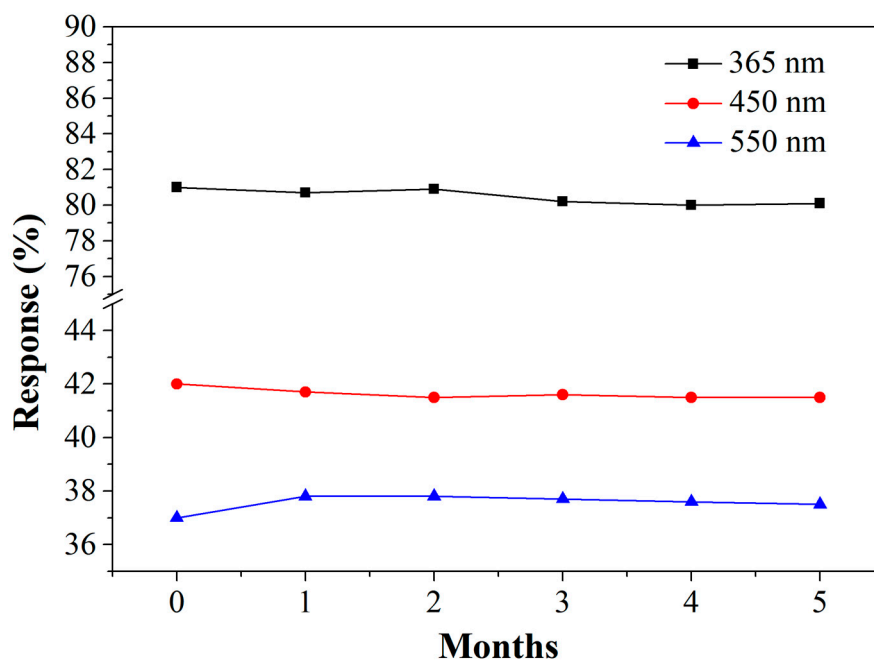


Figure S2. The long-term stability of sensor film under the illumination of 365 nm, 450nm and 550nm.

References

1. Capua, E.; Cao, R.; Sukenik, C.N.; Naaman, R. Detection of triacetone triperoxide (TATP) with an array of sensors based on non-specific interactions. *Sens. Actuators B Chem.* **2009**, *140*, 122–127.
2. Warmer, J.; Wagner, P.; Schöning, M.J.; Kaul, P. Detection of triacetone triperoxide using temperature cycled metal-oxide semiconductor gas sensors. *Phys. Status Solidi* **2015**, *212*, 1289–1298.
3. Amani, M.; Chu, Y.; Waterman, K.L.; Hurley, C.M.; Platek, M.J.; Gregory, O.J. Detection of triacetone triperoxide (TATP) using a thermodynamic based gas sensor. *Sens. Actuators B Chem.* **2012**, *162*, 7–13.
4. Chen, W.-H.Z. and W.-D.Z. and L.-Y. Highly sensitive detection of explosive triacetone triperoxide by an In₂O₃ sensor. *Nanotechnology* **2010**, *21*, 315502.
5. Wen, M.; Jiang, L.; Liu, W.; Cheng, S.; Wang, W.; Chen, C.; Liang, X.; Zhou, Q.; Peng, L.; Li, J.; et al. Sensitive Detection of Triacetone Triperoxide (TATP) by Acetone-Assisted Photoionization Ion Mobility Spectrometry. *J. Chinese Mass Spectrom. Soc.* **2014**, *356*, 481–487.

6. Lin, H.; Suslick, K.S. A Colorimetric Sensor Array for Detection of Triacetone Triperoxide Vapor. *2010*, *132*, 15519–15521.
7. Üzer, A.; Durmazel, S.; Erçağ, E.; Apak, R. Determination of hydrogen peroxide and triacetone triperoxide (TATP) with a silver nanoparticles—based turn-on colorimetric sensor. *Sens. Actuators B Chem.* **2017**, *247*, 98–107.
8. Can, Z.; Üzer, A.; Türkekul, K.; Erçağ, E.; Apak, R. Determination of Triacetone Triperoxide with a N,N-Dimethyl-p-phenylenediamine Sensor on Nafion Using Fe₃O₄ Magnetic Nanoparticles. *Anal. Chem.* **2015**, *87*, 9589–9594.
9. Mamo, S.K.; Gonzalez-rodriguez, J. Development of a Molecularly Imprinted Polymer-Based Sensor for the Electrochemical Determination of Triacetone Triperoxide (TATP). *Sensors* **2014**, *14*, 23269–23282.
10. Laine, D.F.; Roske, C.W.; Cheng, I.F. Electrochemical detection of triacetone triperoxide employing the electrocatalytic reaction of iron(II/III)-ethylenediaminetetraacetate and hydrogen peroxide. *Anal. Chim. Acta* **2008**, *608*, 56–60.