

Linker-Free Magnetite-Decorated Gold Nanoparticles (Fe₃O₄-Au): Synthesis, Characterization, and Application for Electrochemical Detection of Arsenic (III)

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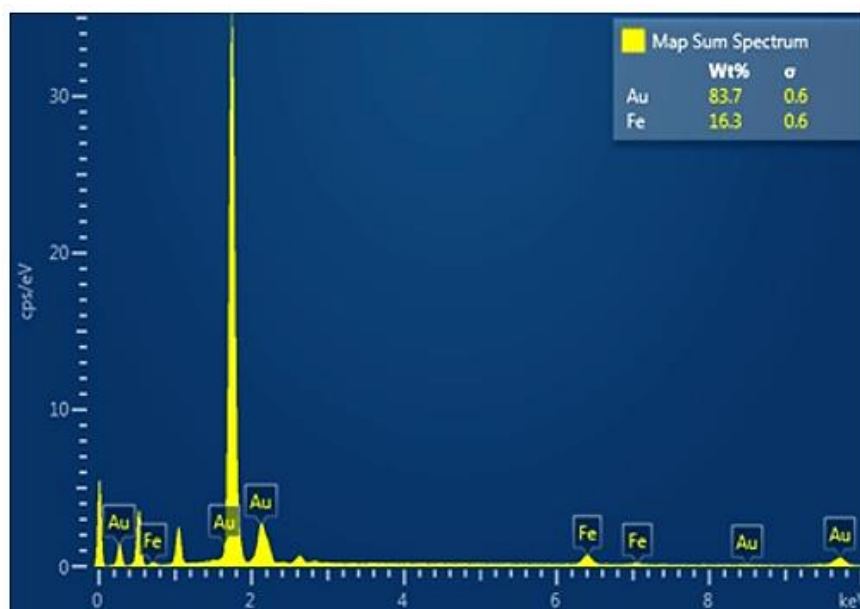


Figure S1. EDX spectrum of the Fe₃O₄-Au nanocomposite on a piece of Pd/Pt-sputtered silicon wafer with the weight percentage.

Table S1. The recipe of synthetic raw wastewater.

Compounds	Amount (mg) added to 500 mL
CH ₃ COONa	128.20
CH ₃ COONH ₄	120.44
KH ₂ PO ₄	21.97
NaHCO ₃	62.50
FeCl ₂	0.188
MnSO ₄	0.02
ZnSO ₄	0.18
NaCl	292.50
Humic acid sodium	25.00
MgSO ₄	12.50

Table S2. Comparison of different electrodes for the detection of As(III).

Electrodes	Technique	Linear range (µg/L)	LOD (µg/L)	Reference
AuNP/BDD electrode	SWASV	100-1500	20	[1]
Au-Pd/GCE	ASV	1-25	0.5	[2]
nano-Au/PANI/GCE	SWV	610-3050	0.4	[3]
AuNPs-PCWEs	CS	2-50	2.2	[4]
rGO/Fe ₃ O ₄ /SPCE	SWASV	2-20	0.3	[5]
AuNPs/GCE	SWASV	0-15	0.25	[6]
NanoPt-Fe(III)/MWCNT/GCE	ASV	1.35-14.5	0.75	[7]
Fe ₃ O ₄ -Au-IL/GCE	SWASV	0-100	0.22	This work

AuNP/BDD: gold nanoparticles on a boron-doped diamond; nano-Au/PANI/GCE: nanogold-particle/polyaniline-modified Glassy Carbon Electrode; SWV: Square Wave Voltammetric; AuNPs-PCWEs: gold-modified paper-based carbon working electrodes; CS: Chronoamperometric Stripping; rGO: reduced graphene oxide; SPCE: screen-printed carbon electrode; platinum-iron(III) nanoparticles modified multi-walled carbon nanotube on glassy carbon electrode: nanoPt-Fe(III)/MWCNT/GCE; GCE: glassy carbon electrode

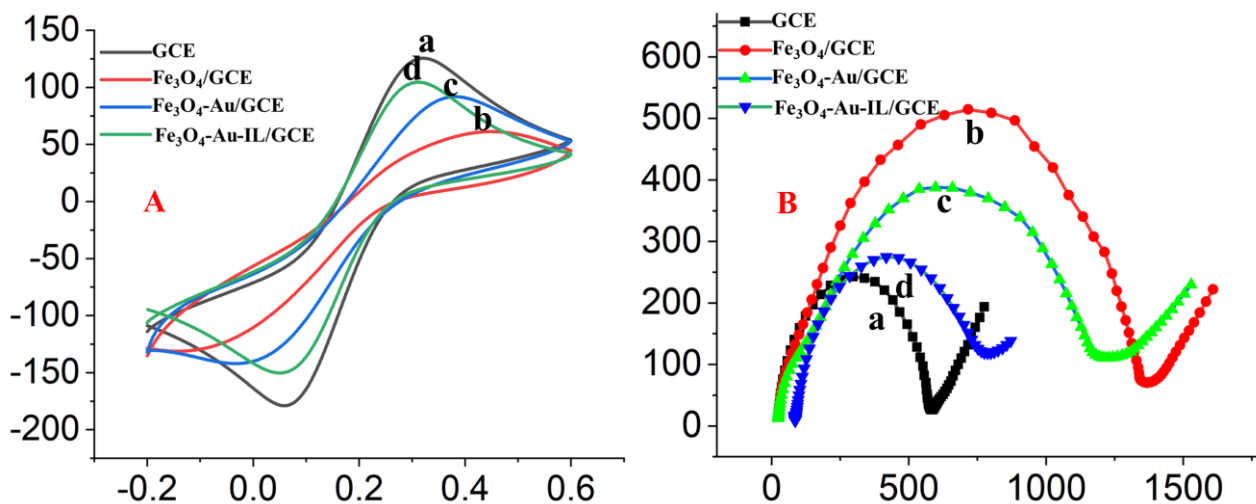


Figure S2. (A) Cyclic voltammograms and (B) Nyquist plot of electrochemical impedance spectra of 5mM $[\text{Fe}(\text{CN})_6]^{3-/4-}$ in 0.1 M KCl: (a) bare GCE, (b) $\text{Fe}_3\text{O}_4/\text{GCE}$, (c) $\text{Fe}_3\text{O}_4\text{-Au}/\text{GCE}$ and (d) $\text{Fe}_3\text{O}_4\text{-Au-IL}/\text{GCE}$.

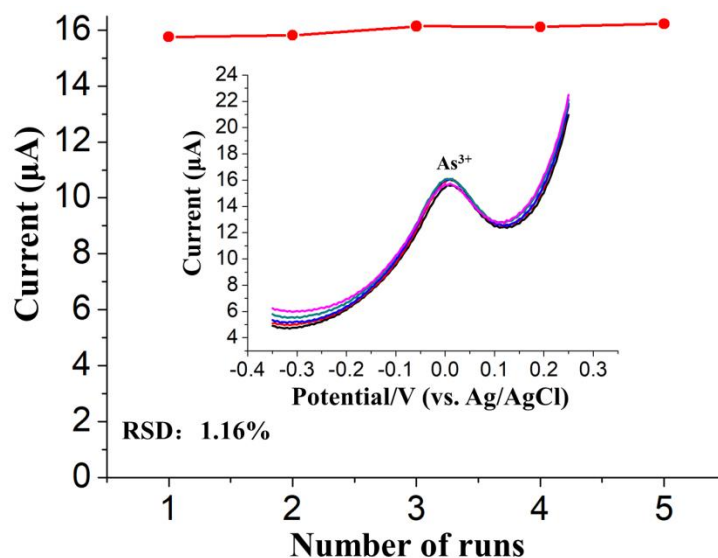


Figure S3. Five repetitive stripping current measurements of 60 µg/L As(III) using the Fe₃O₄-Au-IL/GCE in 0.2 M acetate buffer (pH 5.0). The insets are the data collected from every SWASV response for five runs.

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