

Supplementary Information

Table S1. Parameter for HPLC-MSⁿ analysis of champacyclin (**1a**) and synthetic reference peptides (**1b**, **2**).

Table S2. Parameter for NMR analytics of the natural product champacyclin (**1a**).

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Table S5. Parameters for GC/MS analysis of amino acids and dipeptides.

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Figure S1. LC-MS analysis. (A) Total ion chromatogram (TIC). (B) Extracted ion chromatogram (EIC) of (**1a**) (*m/z* 912-914) (C) Mass spectrum for the EIC as shown in (B).

Figure S2. In-source-fragmentation-MSⁿ-experiments for (**1a**). (A) SID for (**1a**) with 70 eV. MS² for the 185 Da isobaric ion fragment *y*₂/*b*₇*y*₃ with the ESI-(+)-Ion-Trap (CE 15 eV). (B) MS³ on 86 Da ion with ESI-(+)-Ion-Trap (CE 20 eV) for the isobaric ions *y*₂/*b*₇*y*₃. Formation of 69 Da Ion is indicative for an Ile-residue. (C) Possible mechanism for the subsequent MSⁿ-fragmentation series suggesting at least one Ile at position 6 and/or 8.

Figure S3. HR-ESI-(+)-SID-Orbitrap-MS of (A) natural product (**1a**) with SID at 65eV (B) Head-to-tail cyclization N^α_(L)Lys¹-CO-(*AlloD*)Ile⁸ in (**1b**) (C) Head-to-side-chain cyclization N^ε_(L)Lys-CO-(*AlloD*)Ile⁸ in (**2**). Note: the relative abundance for the *b*₁ Ion *m/z* 129 [M + H]⁺ is dramatically decreased in (C). Thus is indicative that the natural product in A (**1a**) is head-to-tail cyclized.

Figure S4. ¹H-NMR of (**1a**) in *d*₆-DMSO at 600 MHz.

Figure S5. ¹H-(PRESAT)-NMR of (**1a**) in *d*₆-DMSO at 600 MHz. PRESAT was used for water suppression.

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Figure S7. HMQC-TOCSY (spin lock 80 ms) of (**1a**) in *d*₆-DMSO at 600 MHz.

Figure S8. ¹H-¹³C-HMBC of (**1a**) in *d*₆-DMSO at 600 MHz.

Figure S9. ¹H-¹H-COSY of (**1a**) in *d*₆-DMSO at 600 MHz.

Figure S10. ¹H-¹H-NOESY (mixing time 100 ms) of (**1a**) in *d*₆-DMSO at 600 MHz.

Figure S11. ¹H-¹H-TOCSY (spin lock 60 ms) data of (**1a**) in *d*₆-DMSO at 600 MHz.

Figure S12. HPLC-ESI-(+)-MS analysis of partial hydrolysate of (**1a**) after 7 h at 110 °C with 6 M HCl. (A) EICs for dipeptides, (B) EICs for tripeptides (C) EIC for tetrapeptides. Note: Absolute stereochemistry and constitution of dipeptides was assigned by GC/MS analysis. This analysis shows the status of the hydrolysis reaction, required for subsequent derivatization and analytics. Formation of lysine-dipeptides could not be observed.

Figure S13. Chiral GC-MS analysis of total hydrolysate of (**1a**) on a Chirasil[®]-(*L*)-Val column. Amino acids were analyzed as their *N*-pentafluoropropionic 2-propyl esters and identified by their [M + H]⁺-Ions and their fragmentation products using positive chemical ionization (PCI) with methane as reagent gas.

Figure S14. Chiral GC-PCI-MS analysis on a Chirasil[®]-(*L*)-Val column of Xle amino acid standards as *N*-pentafluoropropionic 2-propyl esters and identified by their [M + H]⁺-Ions using positive chemical ionization (PCI) with methane as reagent gas.

Figure S15. PCI-MS spectra of *N*-trifluoroacetyl-(*L*)Ile-(*D*)Ala-methyl ester found in partial hydrolysate of (1a) at $R_t = 19.97$ min on a Chirasil[®]-(*L*)-Val column.

Figure S16. EI-MS spectra of *N*-trifluoroacetyl-(*L*)Ile-(*D*)Ala-methyl ester found in partial hydrolysate of (1a) at $R_t = 19.97$ min on a Chirasil[®]-(*L*)-Val column.

Figure S17. PCI-MS spectra of *N*-trifluoroacetyl-(*L*)Ile-(*L*)Ile-methyl ester found in partial hydrolysate of (1a) at $R_t = 22.5$ min on a Chirasil[®]-(*L*)-Val column.

Figure S18. EI-MS spectra of *N*-trifluoroacetyl-(*L*)Ile-(*L*)Ile-methyl ester found in partial hydrolysate of (1a) at $R_t = 22.5$ min on a Chirasil[®]-(*L*)-Val column.

Figure S19. PCI-MS spectra of *N*-trifluoroacetyl-(*D*)Leu-(*L*)Ile-methyl ester found in partial hydrolysate of (1a) at $R_t = 24.6$ min on a Chirasil[®]-(*L*)-Val column.

Figure S20. EI-MS spectra of *N*-trifluoroacetyl-(*D*)Leu-(*L*)Ile-methyl ester found in partial hydrolysate of (1a) at $R_t = 24.6$ min on a Chirasil[®]-(*L*)-Val column.

Figure S21. PCI-MS spectra of *N*-trifluoroacetyl-(*L*)Ile-(*D*)Phe-methyl ester found in partial hydrolysate of (1a) at $R_t = 46.68$ min on a Chirasil[®]-(*L*)-Val column.

Figure S22. EI-MS spectra of *N*-trifluoroacetyl-(*L*)Ile-(*D*)Phe-methyl ester found in partial hydrolysate of (1a) at $R_t = 46.68$ min on a Chirasil[®]-(*L*)-Val column.

Figure S23. PCI-MS spectra of synthetic *N*-trifluoroacetyl-Leu-Leu-methyl ester.

Figure S24. EI-MS spectra of synthetic *N*-trifluoroacetyl-Leu-Leu-methyl ester.

Figure S25. PCI-MS spectra of synthetic *N*-trifluoroacetyl-Ile-Ile-methyl ester.

Figure S26. EI-MS spectra of synthetic *N*-trifluoroacetyl-Ile-Ile-methyl ester.

Figure S27. PCI-MS spectra of synthetic *N*-trifluoroacetyl-Leu-Ile-methyl ester.

Figure S28. EI-MS spectra of synthetic *N*-trifluoroacetyl-Leu-Ile-methyl ester.

Figure S29. PCI-MS spectra of synthetic *N*-trifluoroacetyl-Ile-Leu-methyl ester.

Figure S30. EI-MS spectra of synthetic *N*-trifluoroacetyl-Ile-Leu-methyl ester.

Figure S31. PCI-MS spectra of synthetic *N*-trifluoroacetyl-Ala-Leu-methyl ester.

Figure S32. EI-MS spectra of synthetic *N*-trifluoroacetyl-Ala-Leu-methyl ester.

Figure S33. PCI-MS spectra of synthetic *N*-trifluoroacetyl-Leu-Ala-methyl ester.

Figure S34. EI-MS spectra of synthetic *N*-trifluoroacetyl-Leu-Ala-methyl ester.

Figure S35. PCI-MS spectra of synthetic *N*-trifluoroacetyl-Ala-Ile-methyl ester.

Figure S36. EI-MS spectra of synthetic *N*-trifluoroacetyl-Ala-Ile-methyl ester.

Figure S37. PCI-MS spectra of synthetic *N*-trifluoroacetyl-Ile-Ala-methyl ester.

Figure S38. EI-MS spectra of synthetic *N*-trifluoroacetyl-Ile-Ala-methyl ester.

Figure S39. PCI-MS spectra of synthetic *N*-trifluoroacetyl-Leu-Phe-methyl ester.

Figure S40. EI-MS spectra of synthetic *N*-trifluoroacetyl-Leu-Phe-methyl ester.

Figure S41. PCI-MS spectra of synthetic *N*-trifluoroacetyl-Phe-Leu-methyl ester.

Figure S42. EI-MS spectra of synthetic *N*-trifluoroacetyl-Phe-Leu-methyl ester.

Figure S43. PCI-MS spectra of synthetic *N*-trifluoroacetyl-Ile-Phe-methyl ester.

Figure S44. EI-MS spectra of synthetic *N*-trifluoroacetyl-Ile-Phe-methyl ester.

Figure S45. PCI-MS spectra of synthetic *N*-trifluoroacetyl-Phe-Ile-methyl ester.

Figure S46. EI-MS spectra of synthetic *N*-trifluoroacetyl-Phe-Ile-methyl ester.

Table S1. Parameter for HPLC-MSⁿ analysis of champacyclin (**1a**) and synthetic reference peptides (**1b**, **2**).

| Parameter for HPLC-MS ⁿ Analysis | | | |
|---|--|-----------|--|
| Mass Spectrometer | LTQ Orbitrap XL Mass Spectrometer (Thermo Fisher Scientific, Waltham, MA, USA) | | |
| HPLC | 1260 HPLC-system (Agilent, Santa Clara, CA, USA) | | |
| Column | Thermo, Waltham, MA, USA, Hypersil-Gold, 5 μ m, 50 \times 2.1 mm | | |
| Solvent System | A: H ₂ O + 0.1% HCOOH B: ACN + 0.1% HCOOH | | |
| Flow Rate | 0.25 mL/min | | |
| Gradient | B% | Time | |
| | 5% B | 0 min | |
| | 5% B | 1 min | |
| | 100% B | 6 min | |
| | 100% B | 10 min | |
| | 5% B | 10.10 min | |
| | 5% B | 13 min | |

Table S2. Parameter for NMR analytics of the natural product champacyclin (**1a**).

| Parameter for the NMR Analytics | |
|---------------------------------|--|
| NMR-Spectrometer | DRX 600 MHz (¹ H) (Bruker, Karlsruhe, Germany) |
| Solvent | <i>d</i> ₆ -DMSO, 500 μ L |
| Probe Head | TXI 5 mm, with Z-Gradient |

Table S3. NMR parameters for ¹H- and 2D-NMR experiments for (**1a**) measurements on a DRX 600 (Bruker, Karlsruhe, Germany) NMR spectrometer at the FMP Berlin.

| Experiment | Pulse Program | Scans | TD (F1/F2) | Additional Information |
|--------------------------------------|---------------|-------|------------|------------------------|
| ¹ H-NMR | zg30 | 64 | 65536 | / |
| ¹ H-NMR-PRESAT | MF1hpresat | 32 | 8192 | / |
| ¹ H- ¹ H-COSY | MFdqfcosypre | 8 | 4096/1024 | / |
| ¹ H- ¹ H-NOESY | MFnoesypre | 32 | 4096/1024 | 100 ms |
| ¹ H- ¹ H-TOCSY | Psdipsi2pre | 8 | 4096/1024 | 60 ms |
| ¹ H- ¹³ C-HMQC | MFbihmqceaf2 | 128 | 1024/512 | / |
| ¹ H- ¹³ C-HMBC | MFhmbceaf2 | 80 | 4096/1024 | / |
| HSQC-TOCSY | MFbihmqctocf2 | 8 | 1024/512 | 80 ms |

Table S4. Parameter for HPLC-MS analysis of Partial hydrolysis of the natural product champacyclin (**1a**).

| Parameter for HPLC-MS Analysis | | | | | | | | | | | | | | | | | | | |
|--------------------------------|--|----|------|------|-------|------|-------|-------|--------|-------|-------|--------|---------|--------|-------|------|---------|------|--------|
| Mass Spectrometer | QQQ-MS-6460 (Agilent Technologies, Waldbronn Germany) | | | | | | | | | | | | | | | | | | |
| HPLC | Agilent 1290 UHPLC-system (Agilent Technologies, Waldbronn, Germany) | | | | | | | | | | | | | | | | | | |
| Column | Agilent, Waldbronn Germany, Eclipse Plus C18 RRHD column, 1.8 μm , 2.1 \times 50 mm | | | | | | | | | | | | | | | | | | |
| Solvent System | A: H ₂ O + 0.1% HCOOH B: ACN + 0.1% HCOOH | | | | | | | | | | | | | | | | | | |
| Flow Rate | 0.3 mL/min | | | | | | | | | | | | | | | | | | |
| Gradient | <table border="1"> <thead> <tr> <th>B%</th> <th>Time</th> </tr> </thead> <tbody> <tr> <td>5% B</td> <td>0 min</td> </tr> <tr> <td>5% B</td> <td>1 min</td> </tr> <tr> <td>20% B</td> <td>20 min</td> </tr> <tr> <td>70% B</td> <td>6 min</td> </tr> <tr> <td>100% B</td> <td>6.5 min</td> </tr> <tr> <td>100% B</td> <td>8 min</td> </tr> <tr> <td>5% B</td> <td>8.5 min</td> </tr> <tr> <td>5% B</td> <td>10 min</td> </tr> </tbody> </table> | B% | Time | 5% B | 0 min | 5% B | 1 min | 20% B | 20 min | 70% B | 6 min | 100% B | 6.5 min | 100% B | 8 min | 5% B | 8.5 min | 5% B | 10 min |
| B% | Time | | | | | | | | | | | | | | | | | | |
| 5% B | 0 min | | | | | | | | | | | | | | | | | | |
| 5% B | 1 min | | | | | | | | | | | | | | | | | | |
| 20% B | 20 min | | | | | | | | | | | | | | | | | | |
| 70% B | 6 min | | | | | | | | | | | | | | | | | | |
| 100% B | 6.5 min | | | | | | | | | | | | | | | | | | |
| 100% B | 8 min | | | | | | | | | | | | | | | | | | |
| 5% B | 8.5 min | | | | | | | | | | | | | | | | | | |
| 5% B | 10 min | | | | | | | | | | | | | | | | | | |

Table S5. Parameters for GC/MS analysis of amino acids and dipeptides.

| Parameters for GC/MS Analysis | |
|--|---|
| GC Column | Chirasil [®] -(L)-Val (Agilent CP7495, Waldbronn Germany), 200 $^{\circ}\text{C}$, 25 m, 250 μm \times 0.12 μm |
| Temperature Program <i>N</i> -trifluoroacetyl methyl ester:dipeptides | 140 $^{\circ}\text{C}$ (10 min isothermal), 190 $^{\circ}\text{C}$, 5 $^{\circ}\text{C}/\text{min}$ (33min isothermal), 195 $^{\circ}\text{C}$, 5 $^{\circ}\text{C}/\text{min}$ (20 min isothermal) |
| Temperature Program <i>N</i> -pentafluoropropionic 2-propyl ester:amino acids | 70 $^{\circ}\text{C}$ (5 min isothermal), 100 $^{\circ}\text{C}$, 2 $^{\circ}\text{C}/\text{min}$, 190 $^{\circ}\text{C}$, 3.5 $^{\circ}\text{C}/\text{min}$ (10 min isothermal) |
| Gas Chromatograph | GC/MS 5975C (Agilent Technologies, Waldbronn, Germany) |
| Mass Spectrometer (GC-MS) | |
| Scan | Full-Scan, 50–800 m/z |
| Heater/MSD-Transfer Line | 300 $^{\circ}\text{C}/280$ $^{\circ}\text{C}$ |
| Flow | 1.2 mL/min, 40.3 cm/s with He as carrier gas |
| PCI-Mode | Energy: 105.2 eV, Emission: 182 μA , MS-Source: 300 $^{\circ}\text{C}$, MS Quad: 150 $^{\circ}\text{C}$ Collision gas: Methane with flow rate 19% |
| EI-Mode using CI-Source | Energy: 105.2 eV, Emission: 250 μA , MS-Source: 300 $^{\circ}\text{C}$, MS Quad: 150 $^{\circ}\text{C}$ Collision gas: Methane with flow rate 0% |
| Single Ion Monitoring (SIM) for Higher Sensitivity | SIM was used with ions represented in all <i>N</i> -trifluoroacetyl methyl ester dipeptide derivatives #1: m/z 389 [M + H] ⁺ (Phe-Xle/Xle-Phe) #2: m/z 313 [M + H] ⁺ (Ala-Xle/Xle-Ala) #3: m/z 355 [M + H] ⁺ (Xle-Xle) |

Table S6. Peak areas and relative Quantification of the *N*-pentafluoropropionic 2-propyl ester derivatives of the total hydrolysate of (**1a**) as shown in Figure S13.

| Amino Acids (<i>N</i> -Pentafluoropropionic 2-Propyl Ester Derivatives) | Peak Areas (Absolute) RTE Integrator | Ratio/Racemization Factor (%) |
|--|---|-------------------------------|
| (<i>D</i>)Ala:(<i>L</i>)Ala | 64,586,729: <i>n.d.</i> | <i>n.d.</i> |
| (Allo <i>D</i>)Ile:(<i>L</i>)Ile | 40,548,258:114,743,611 | 1:2.8 |
| (<i>D</i>)Leu:(<i>L</i>)Leu | 92,808,033:5,641,417 | 93.9%:6.1% |
| (<i>D</i>)Phe:(<i>L</i>)Phe | 97,483,637:3,178,567 | 96.8%:3.2% |
| (<i>D</i>)Lys:(<i>L</i>)Lys | 3,493,779:69,688,561 | 95%:5.0% |

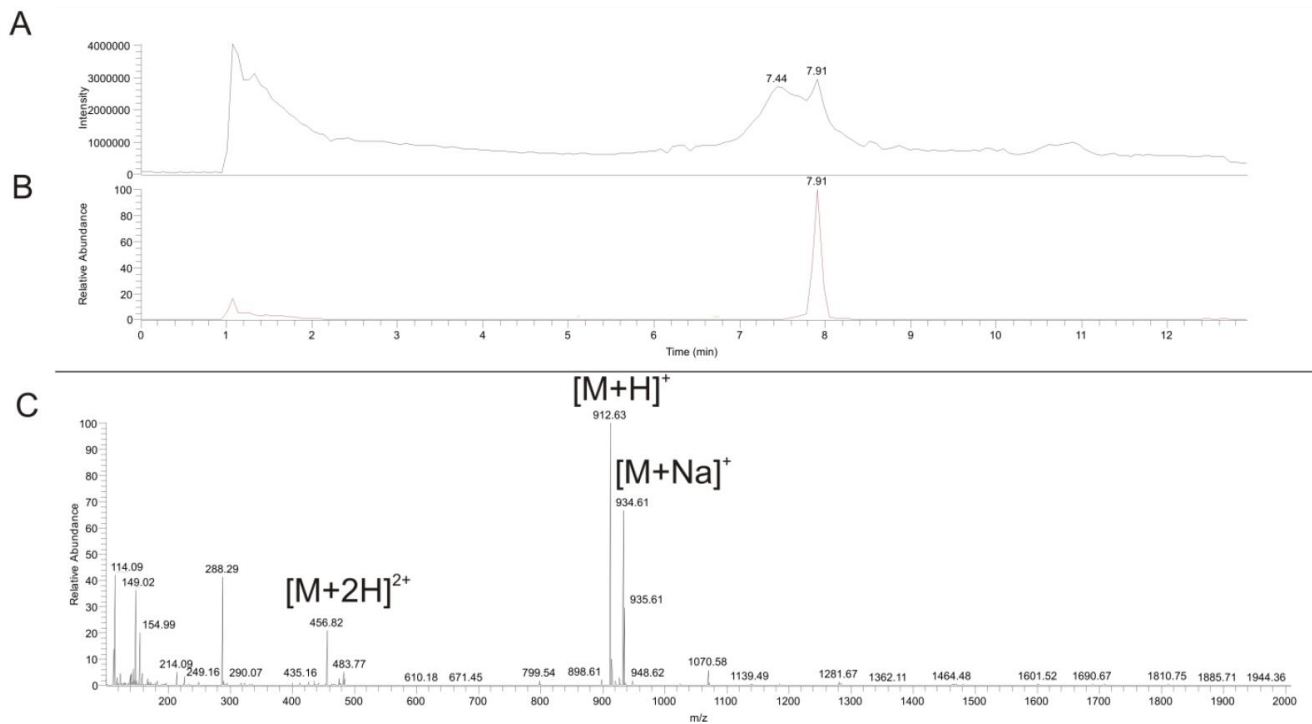
n.d. = not determined.**Figure S1.** LC-MS analysis. (A) Total ion chromatogram (TIC). (B) Extracted ion chromatogram (EIC) of (**1a**) (*m/z* 912-914) (C) Mass spectrum for the EIC as shown in (B).

Figure S2. In-source-fragmentation- MS^n -experiments for (1a). (A) SID for (1a) with 70 eV. MS^2 for the 185 Da isobaric ion fragment y_2/b_7y_3 with the ESI-(+)-Ion-Trap (CE 15 eV). (B) MS^3 on 86 Da ion with ESI-(+)-Ion-Trap (CE 20 eV) for the isobaric ions y_2/b_7y_3 . Formation of 69 Da Ion is indicative for an Ile-residue. (C) Possible mechanism for the subsequent MS^n -fragmentation series suggesting at least one Ile at position 6 and/or 8.

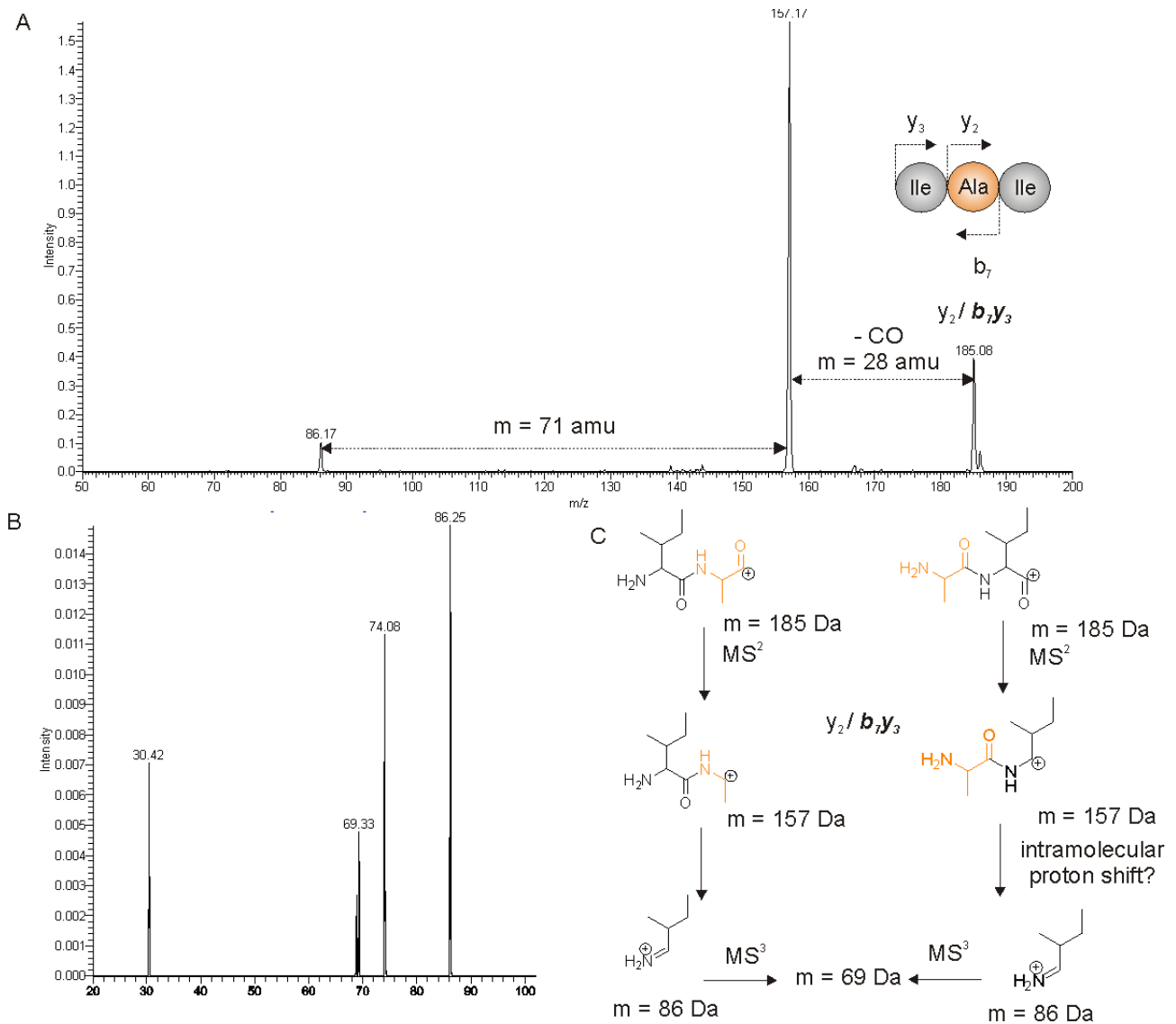


Figure S3. HR-ESI(+)-SID-Orbitrap-MS of (A) natural product (**1a**) with SID at 65eV (B) Head-to-tail cyclization $N^{\alpha}_{(L)}\text{Lys}^1\text{-CO}_{-(AlloD)}\text{Ile}^8$ in (**1b**) (C) Head-to-side-chain cyclization $N^{\zeta}_{-(L)}\text{Lys-CO}_{-(AlloD)}\text{Ile}^8$ in (**2**). Note: the relative abundance for the b_1 Ion m/z 129 $[\text{M} + \text{H}]^+$ is dramatically decreased in (C). Thus is indicative that the natural product in A (**1a**) is head-to-tail cyclized.

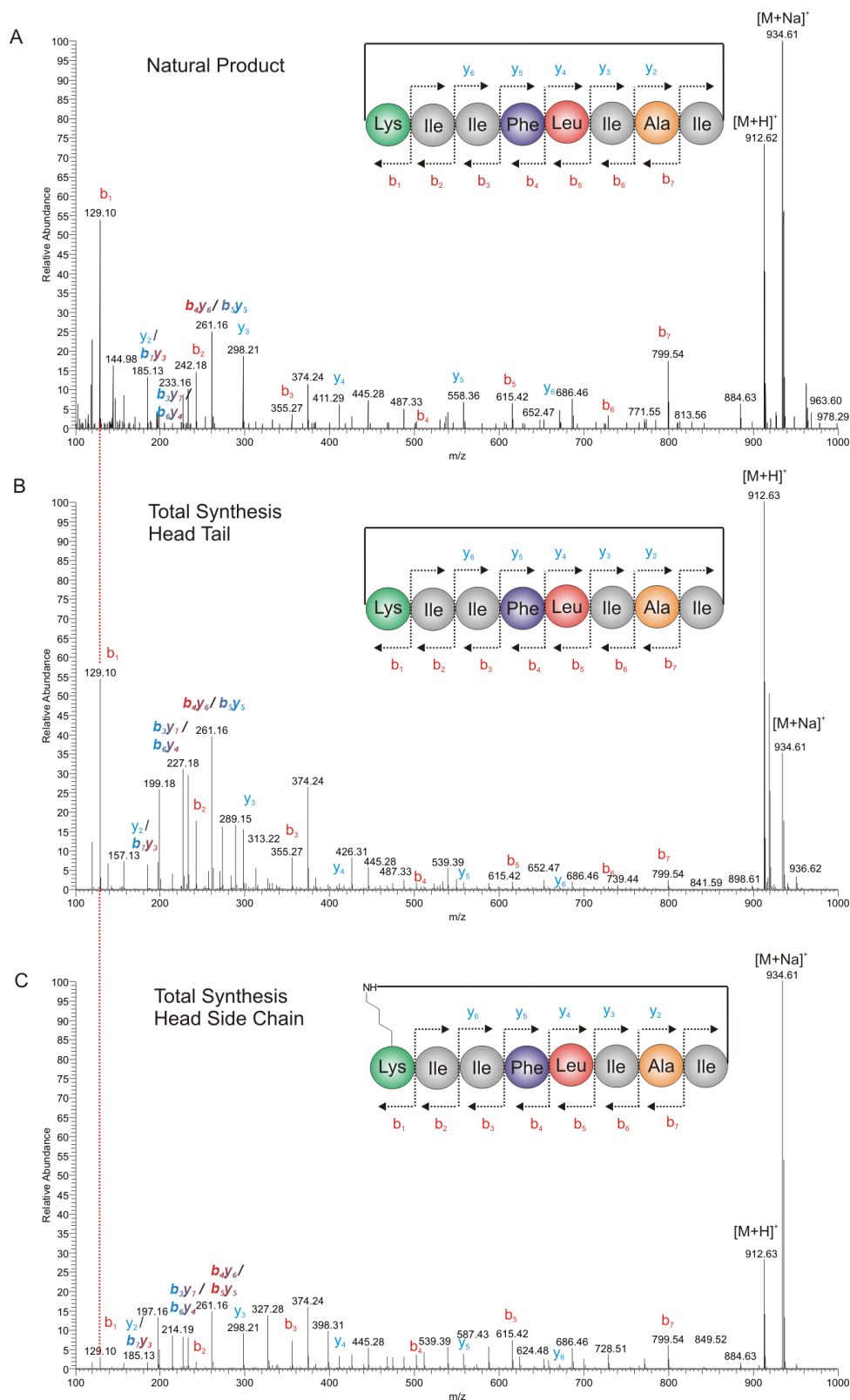


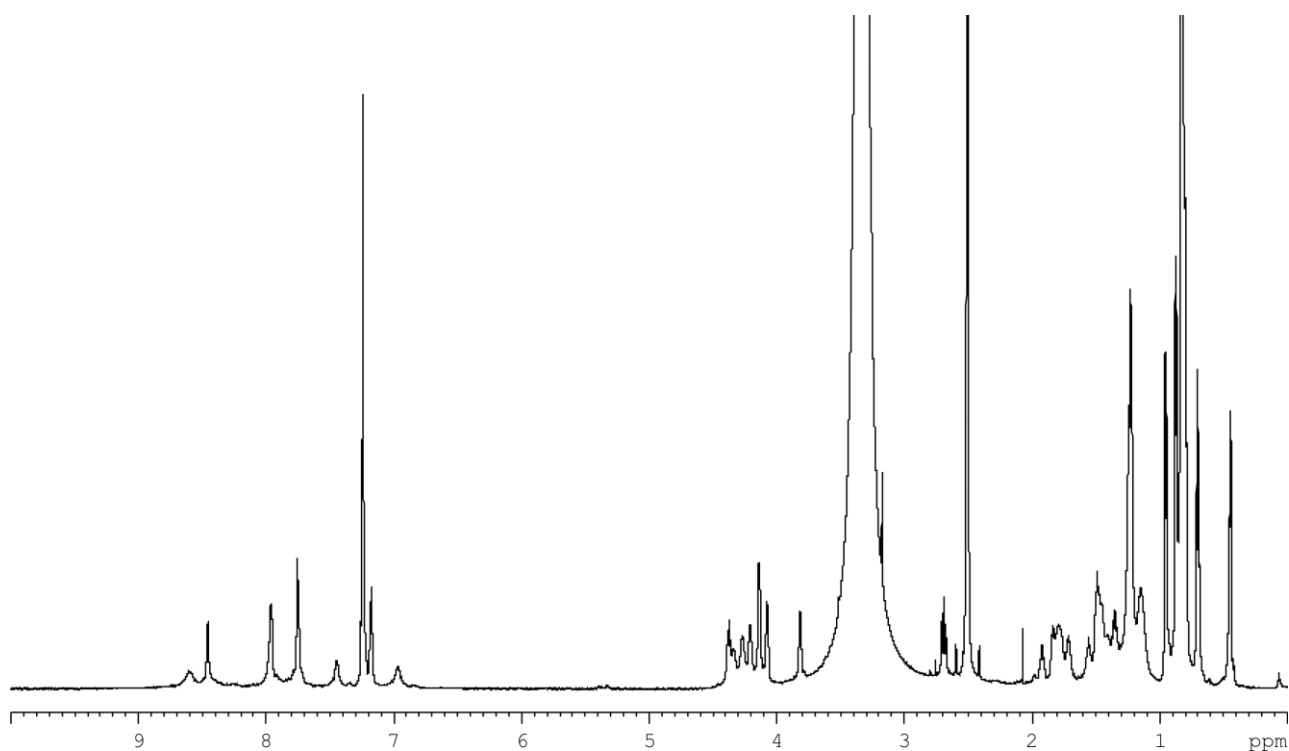
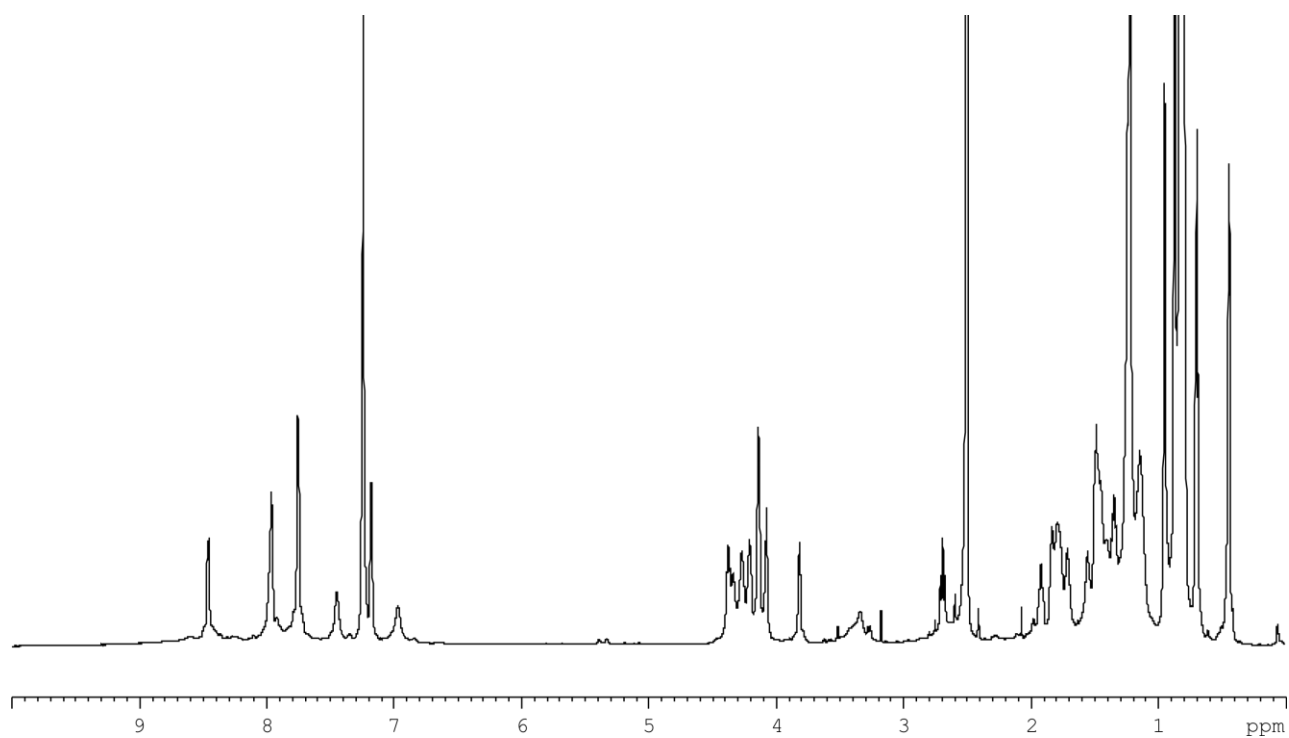
Figure S4. ^1H -NMR of (**1a**) in d_6 -DMSO at 600 MHz.**Figure S5.** ^1H -(PRESAT)-NMR of (**1a**) in d_6 -DMSO at 600 MHz. PRESAT was used for water suppression.

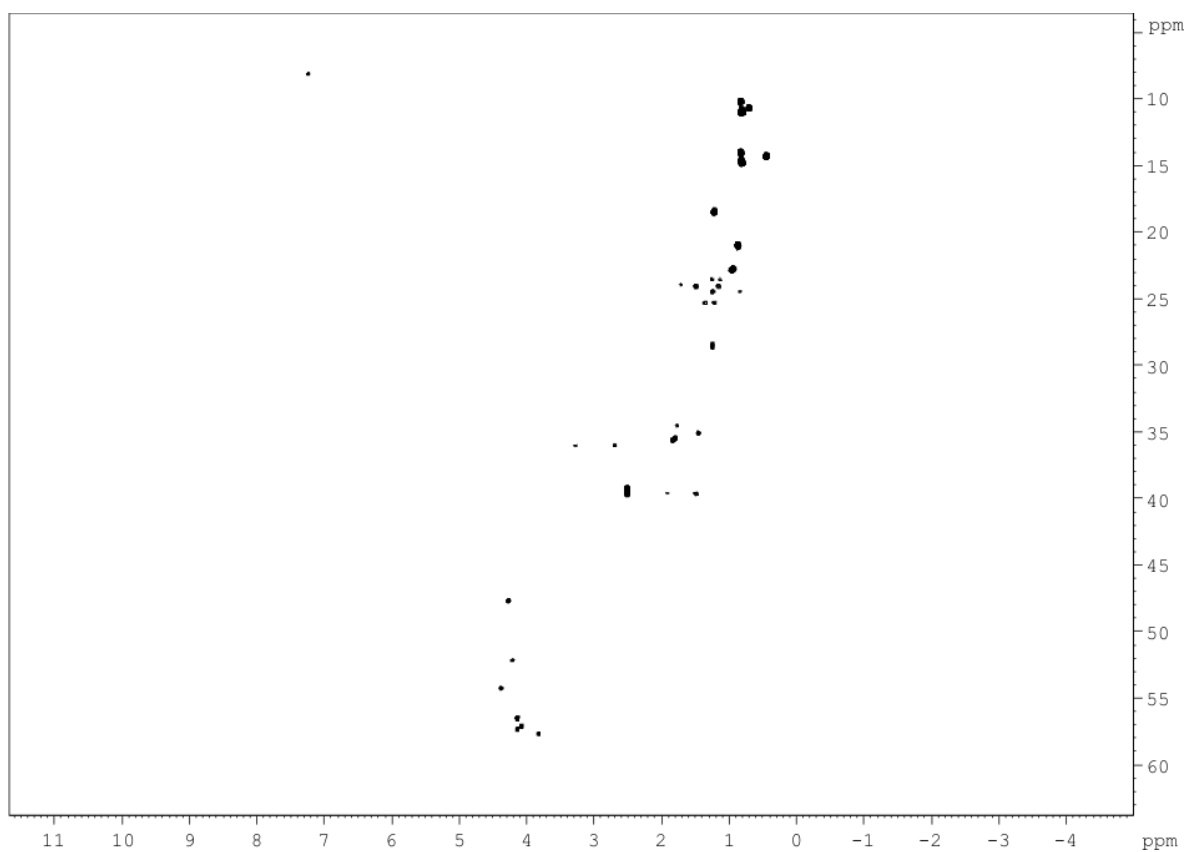
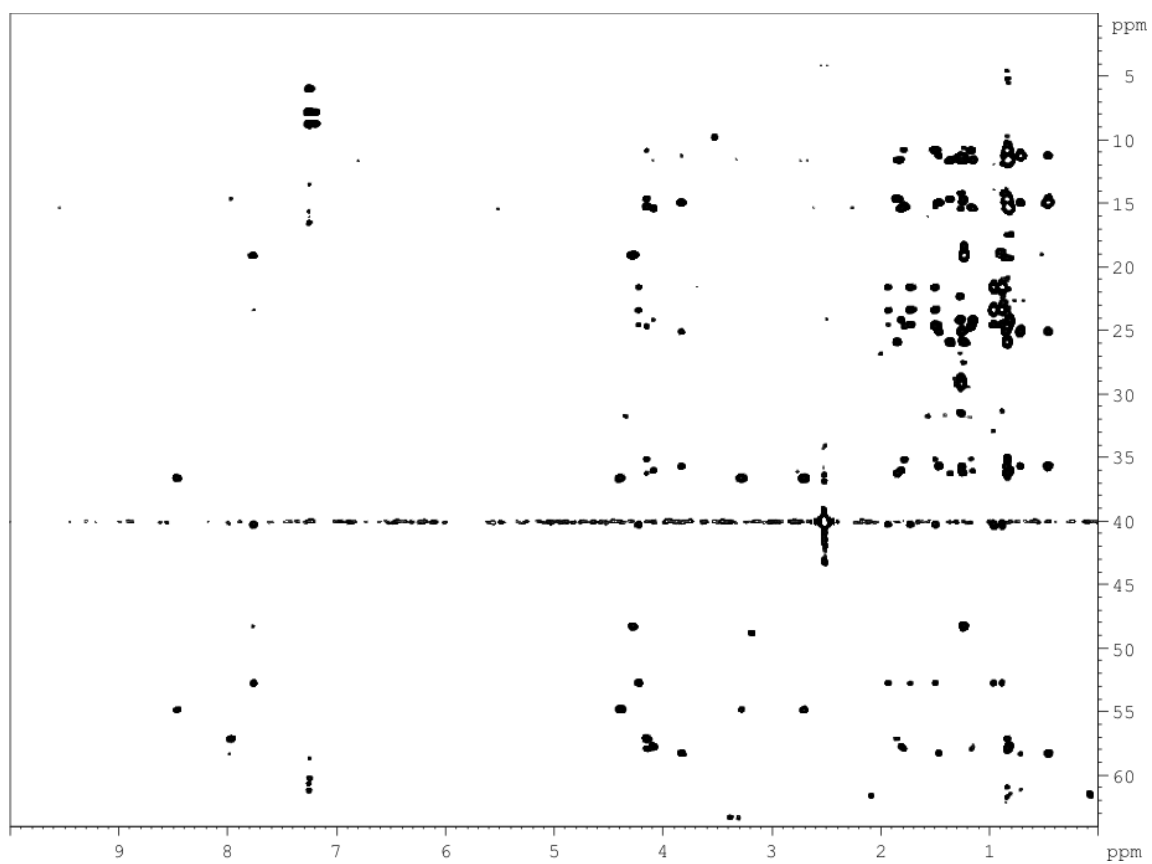
Figure S6. ^1H - ^{13}C -HMQC of (1a) in d_6 -DMSO at 600 MHz.**Figure S7.** HMQC-TOCSY (spin lock 80 ms) of (1a) in d_6 -DMSO at 600 MHz.

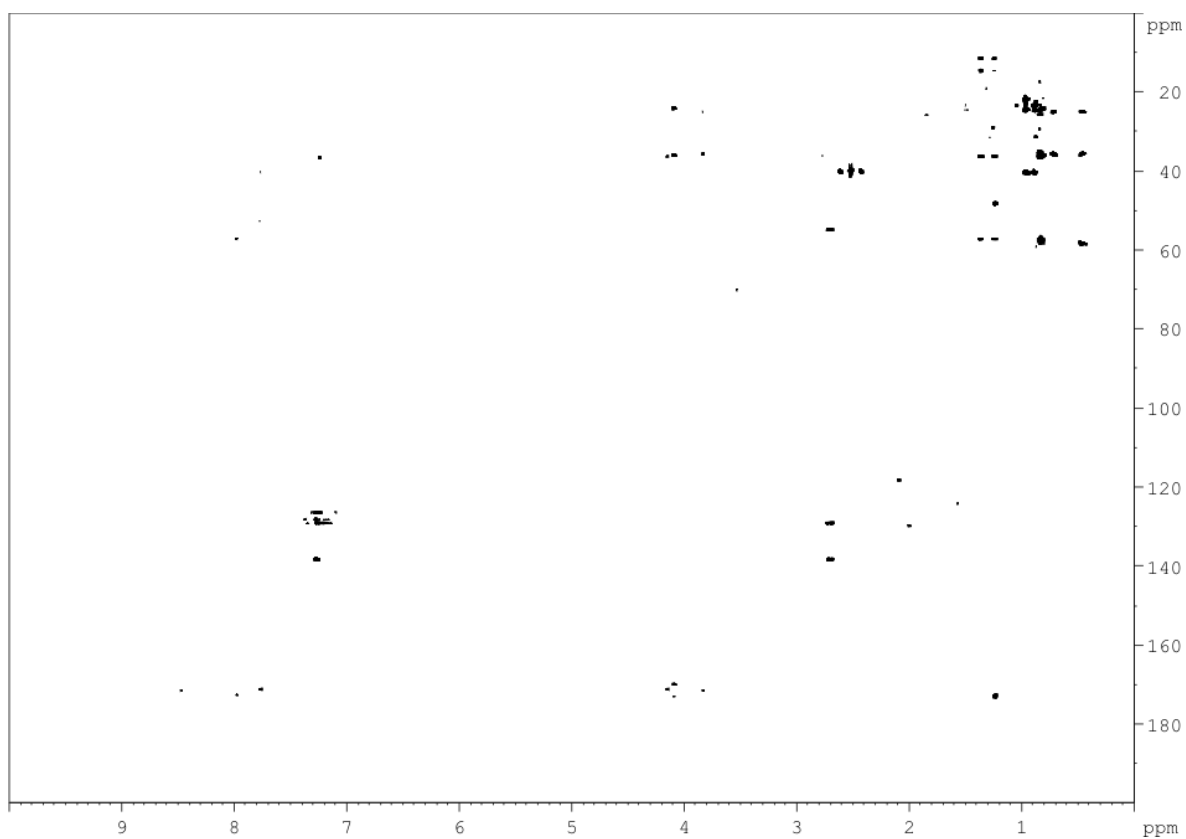
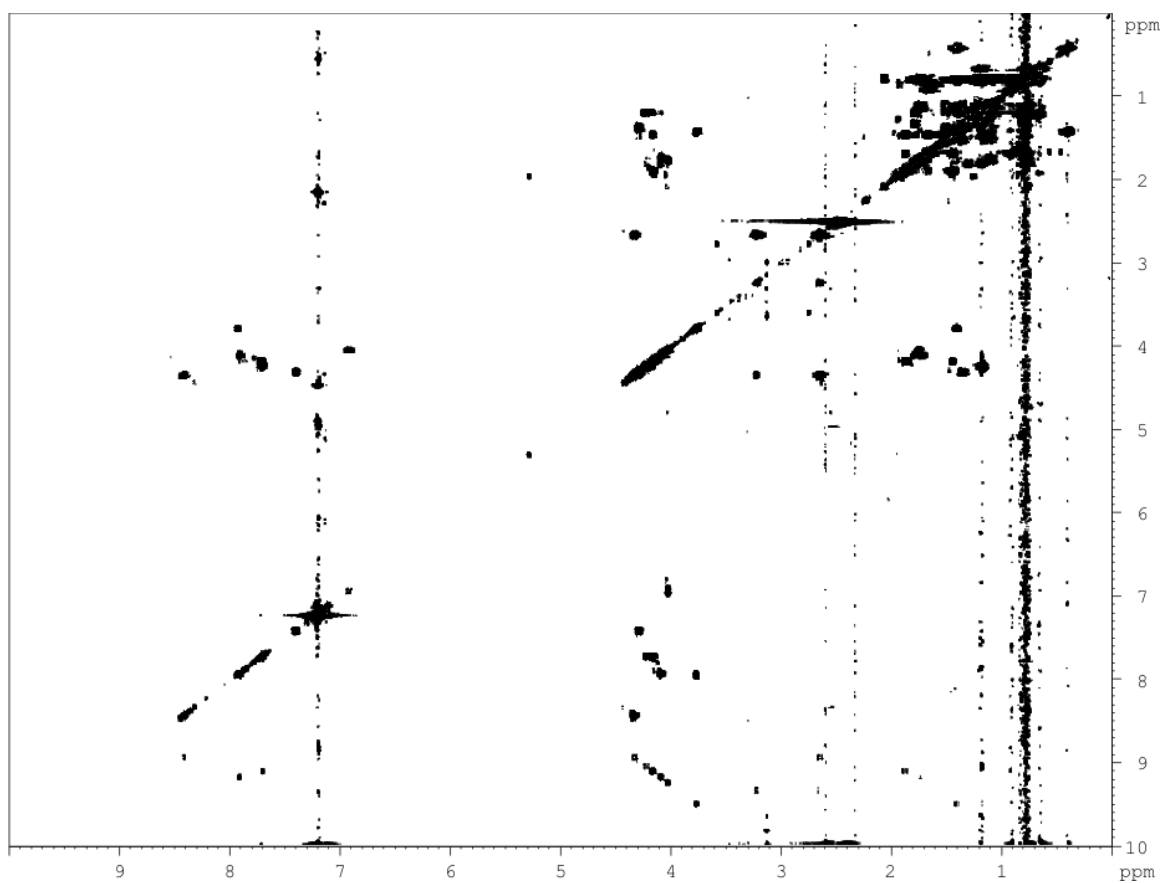
Figure S8. ^1H - ^{13}C -HMBC of (1a) in d_6 -DMSO at 600 MHz.**Figure S9.** ^1H - ^1H -COSY of (1a) in d_6 -DMSO at 600 MHz.

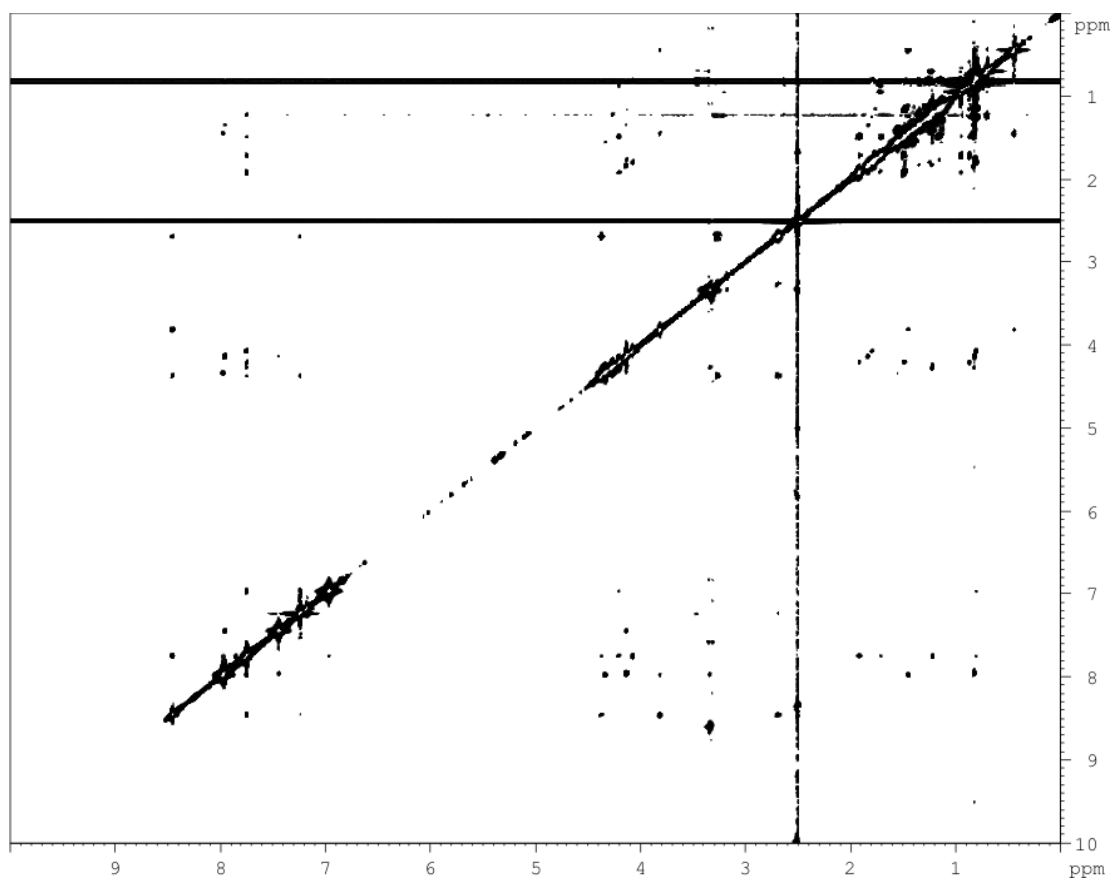
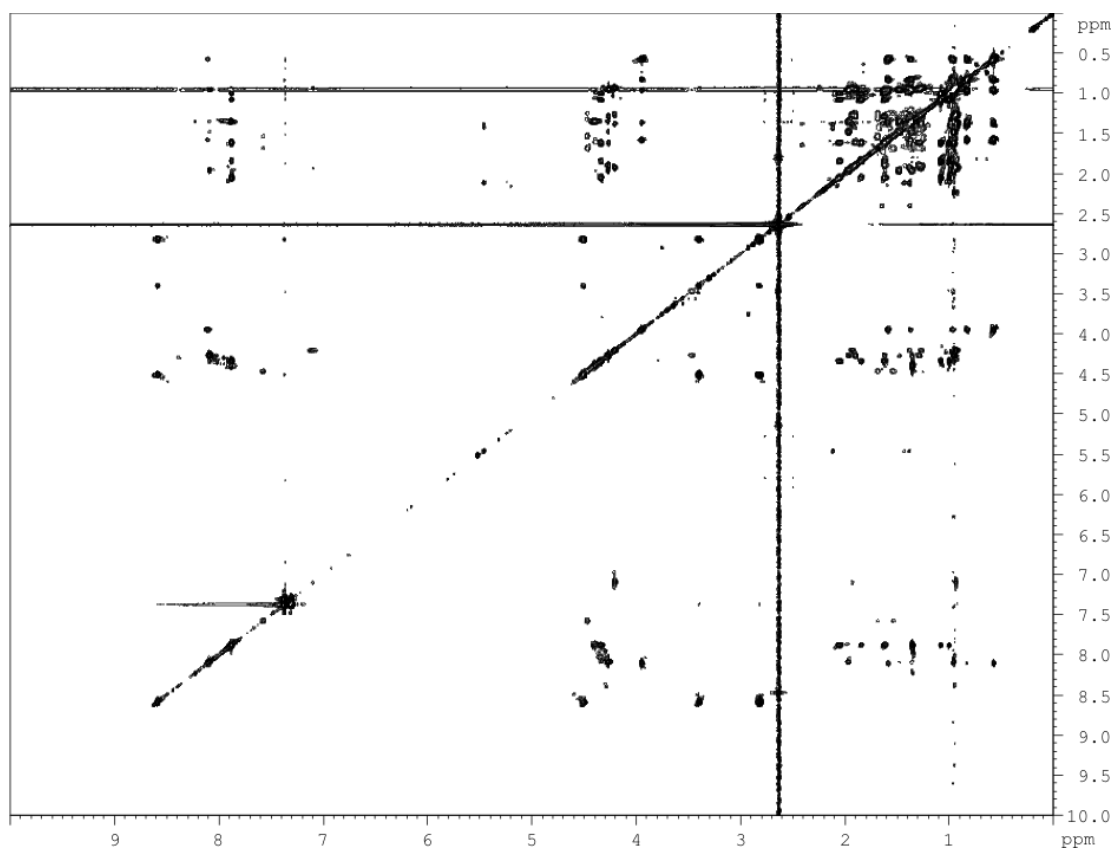
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Figure S12. HPLC-ESI-(+)-MS analysis of partial hydrolysate of (**1a**) after 7 h at 110 °C with 6 M HCl. (A) EICs for dipeptides, (B) EICs for tripeptides (C) EIC for tetrapeptides. Note: Absolute stereochemistry and constitution of dipeptides was assigned by GC/MS analysis. This analysis shows the status of the hydrolysis reaction, required for subsequent derivatization and analytics. Formation of lysine-dipeptides could not be observed.

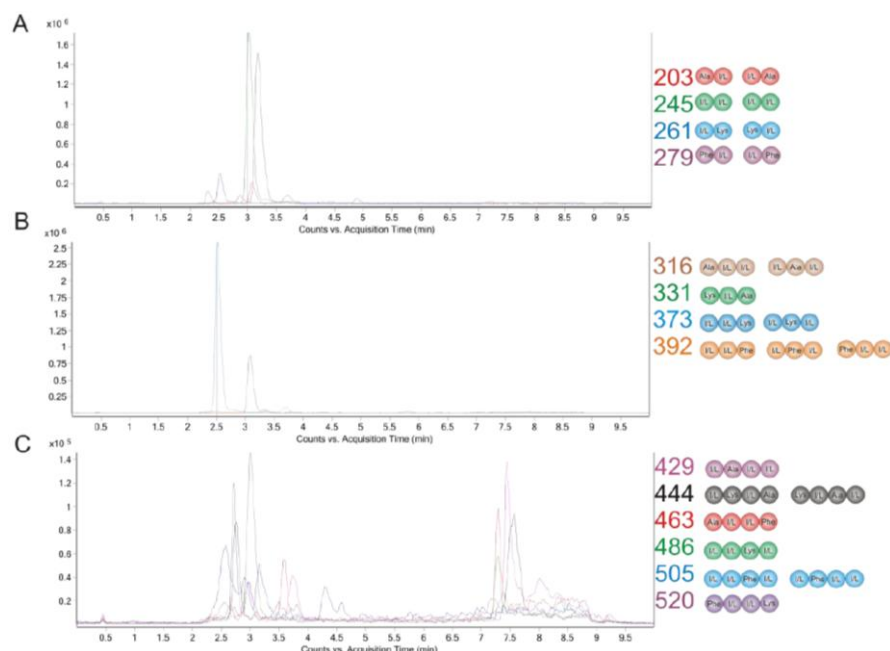


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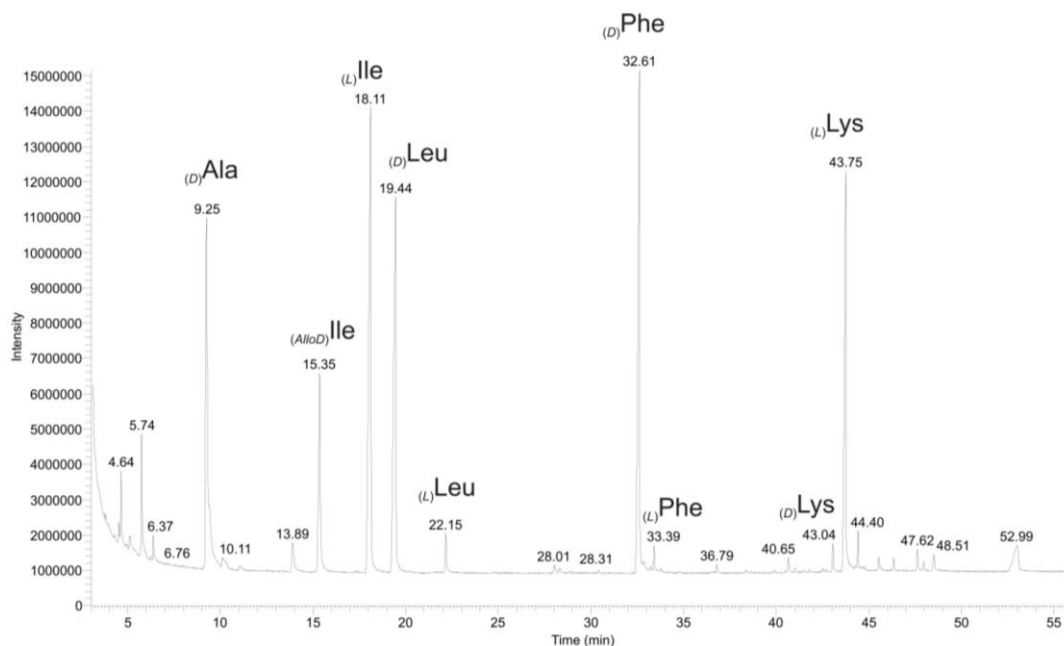


Figure S14. Chiral GC-PCI-MS analysis on a Chirasil[®]-(L)-Val column of Xle amino acid standards as *N*-pentafluoropropionic 2-propyl esters and identified by their $[M + H]^+$ -Ions using positive chemical ionization (PCI) with methane as reagent gas.

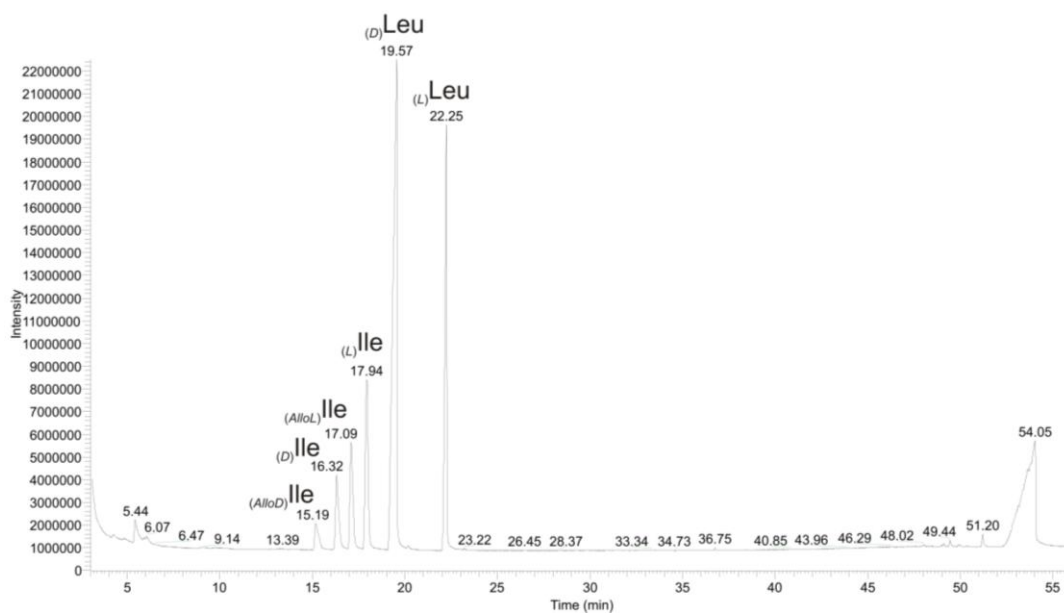


Figure S15. PCI-MS spectra of *N*-trifluoroacetyl-(L)Ile-(D)Ala-methyl ester found in partial hydrolysate of (1a) at $R_t = 19.97$ min on a Chirasil[®]-(L)-Val column.

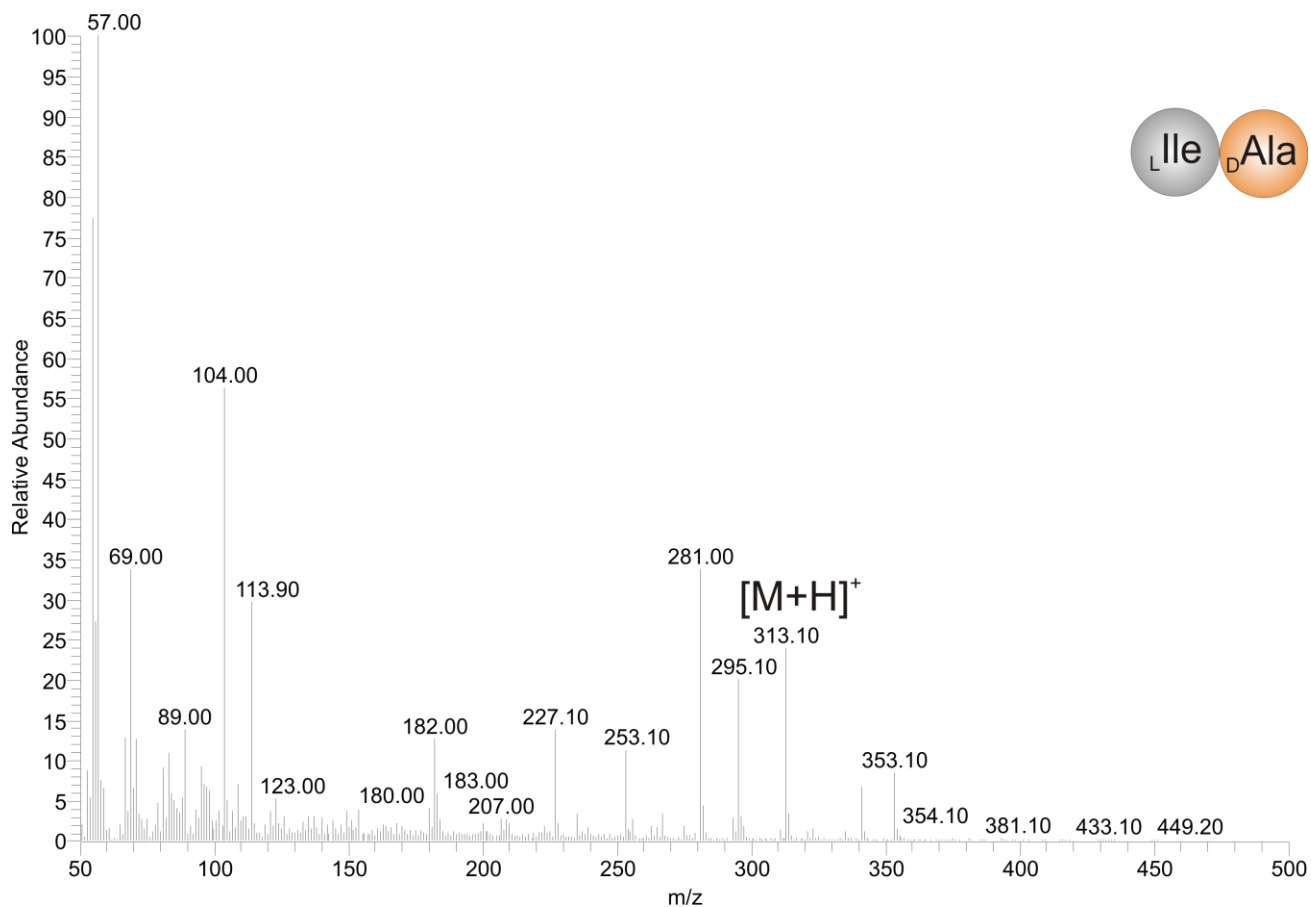


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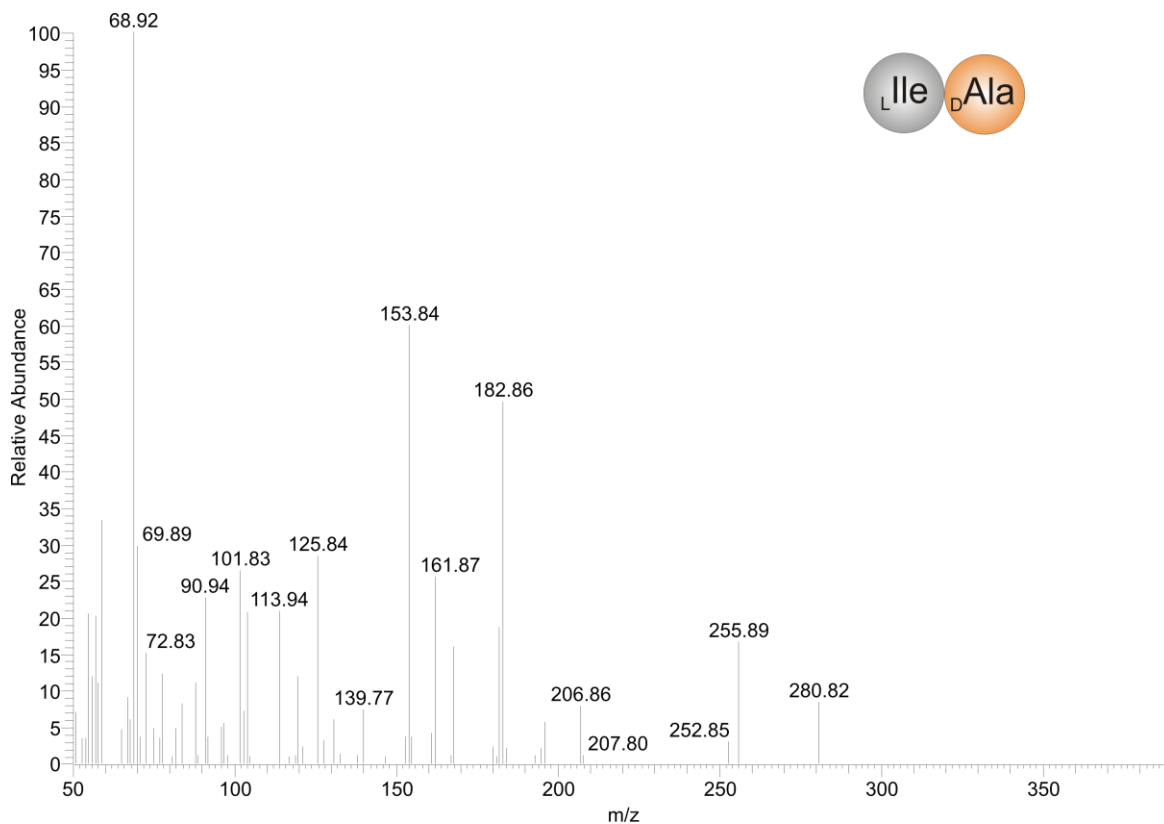


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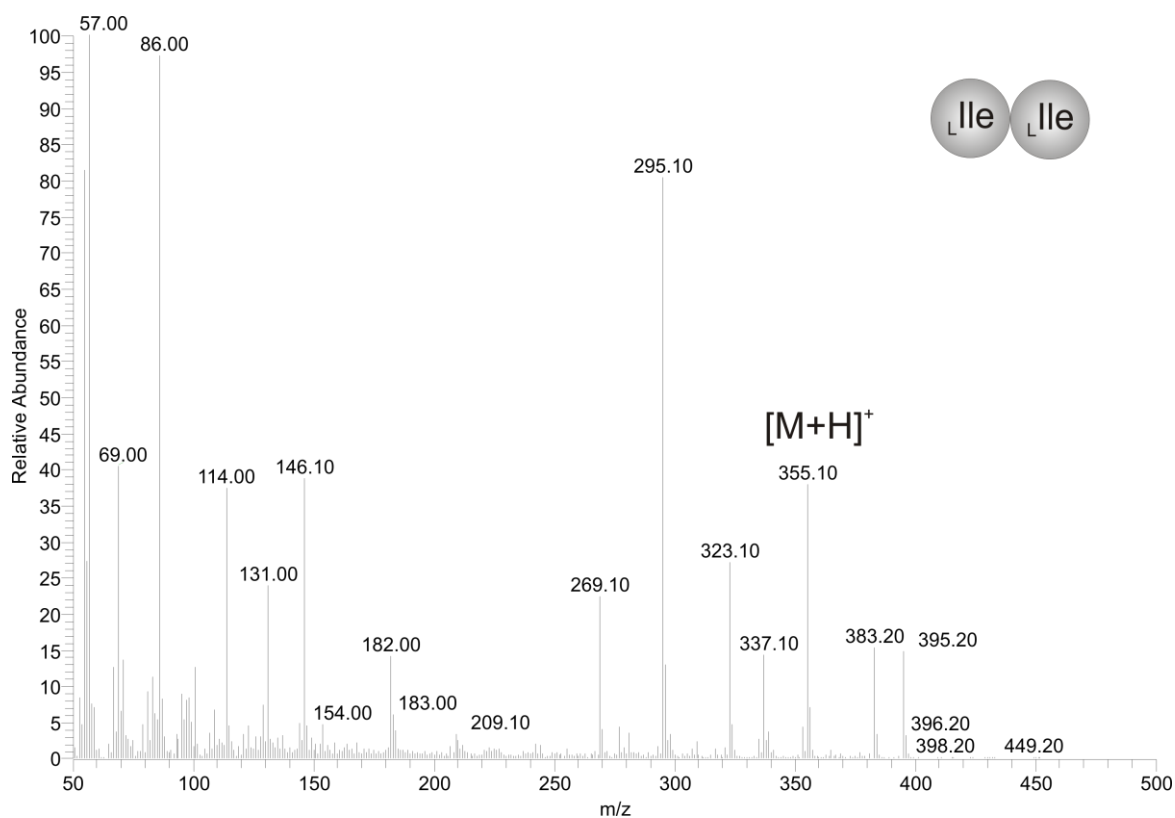


Figure S18. EI-MS spectra of *N*-trifluoroacetyl-(*L*)Ile-(*L*)Ile-methyl ester found in partial hydrolysate of (**1a**) at $R_t = 22.5$ min on a Chirasil[®]-(*L*)-Val column.

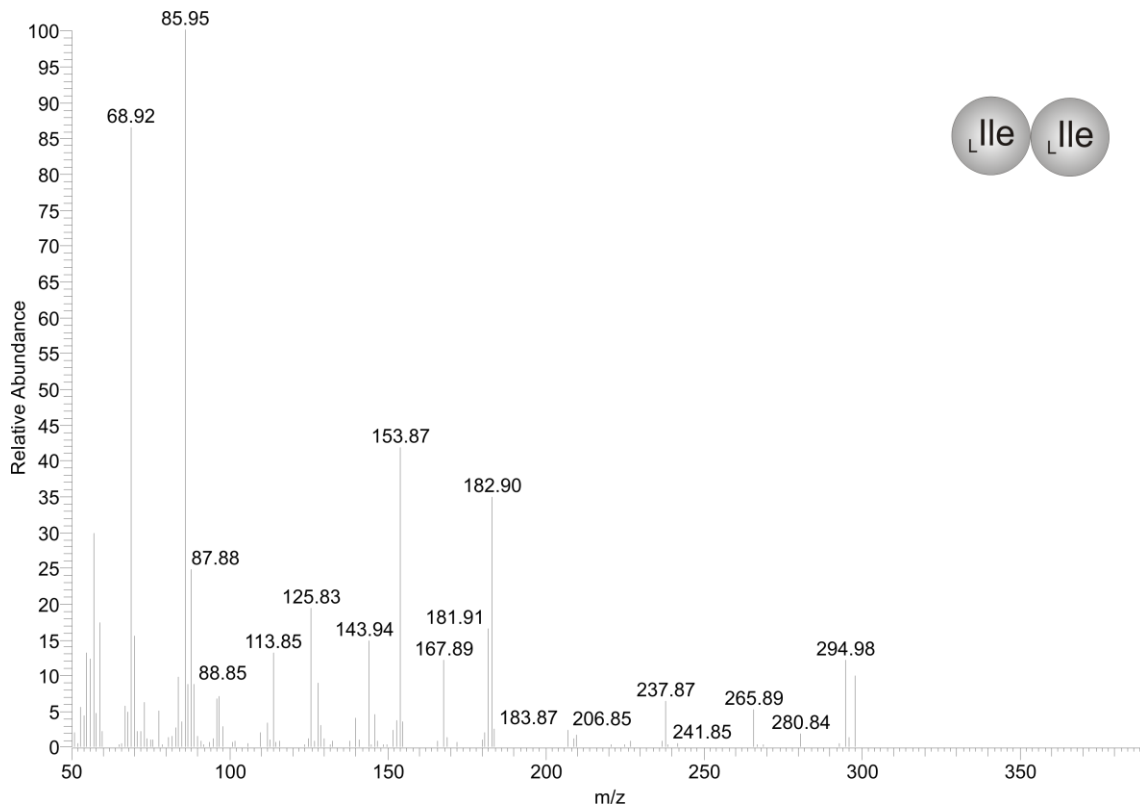


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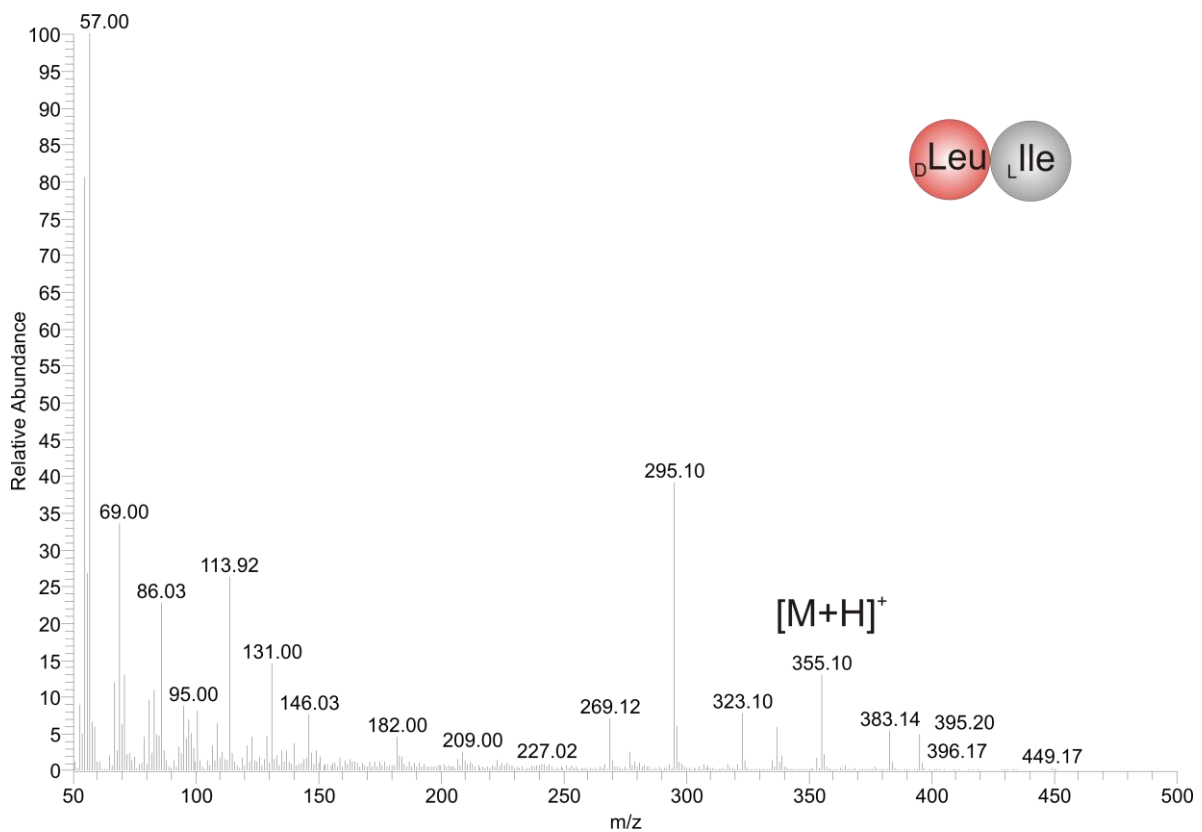


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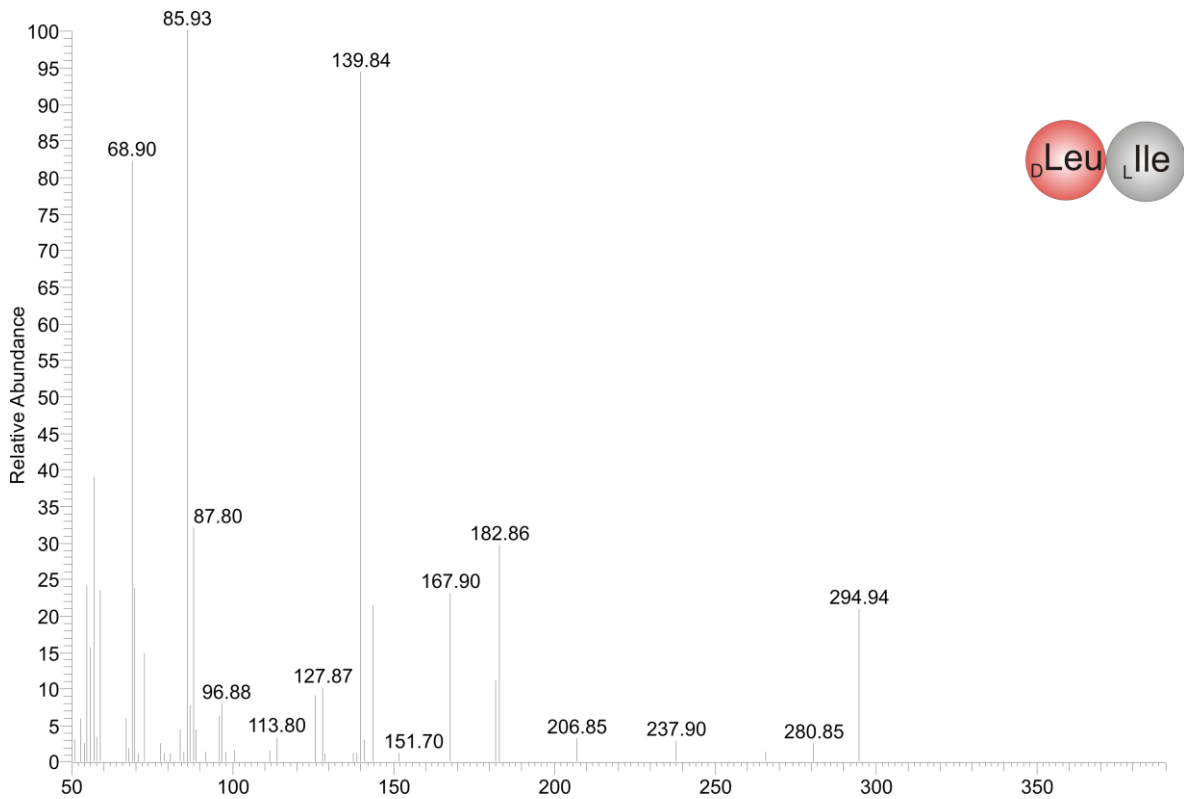


Figure S21. PCI-MS spectra of *N*-trifluoroacetyl-^(L)Ile-^(D)Phe-methyl ester found in partial hydrolysate of (1a) at $R_t = 46.68$ min on a Chirasil[®]-(L)-Val column.

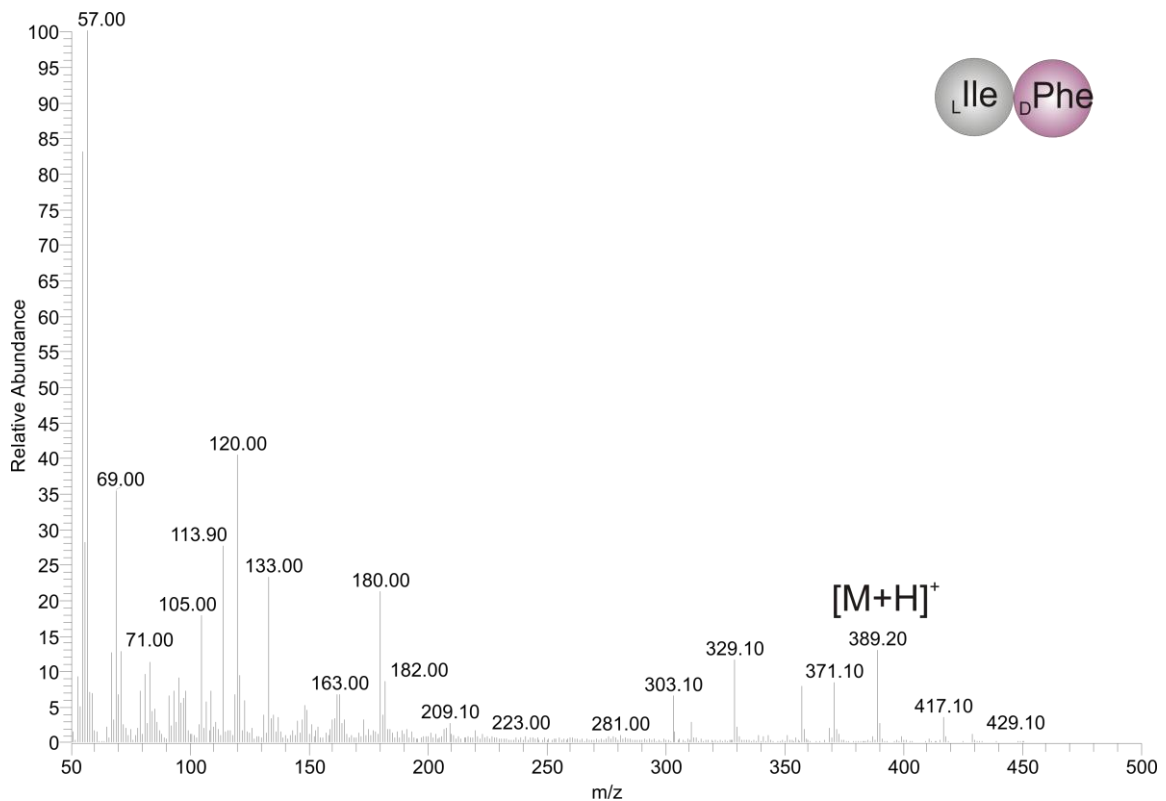


Figure S22. EI-MS spectra of *N*-trifluoroacetyl-(*L*)Ile-(*D*)Phe-methyl ester found in partial hydrolysate of (**1a**) at $R_t = 46.68$ min on a Chirasil[®]-(*L*)-Val column.

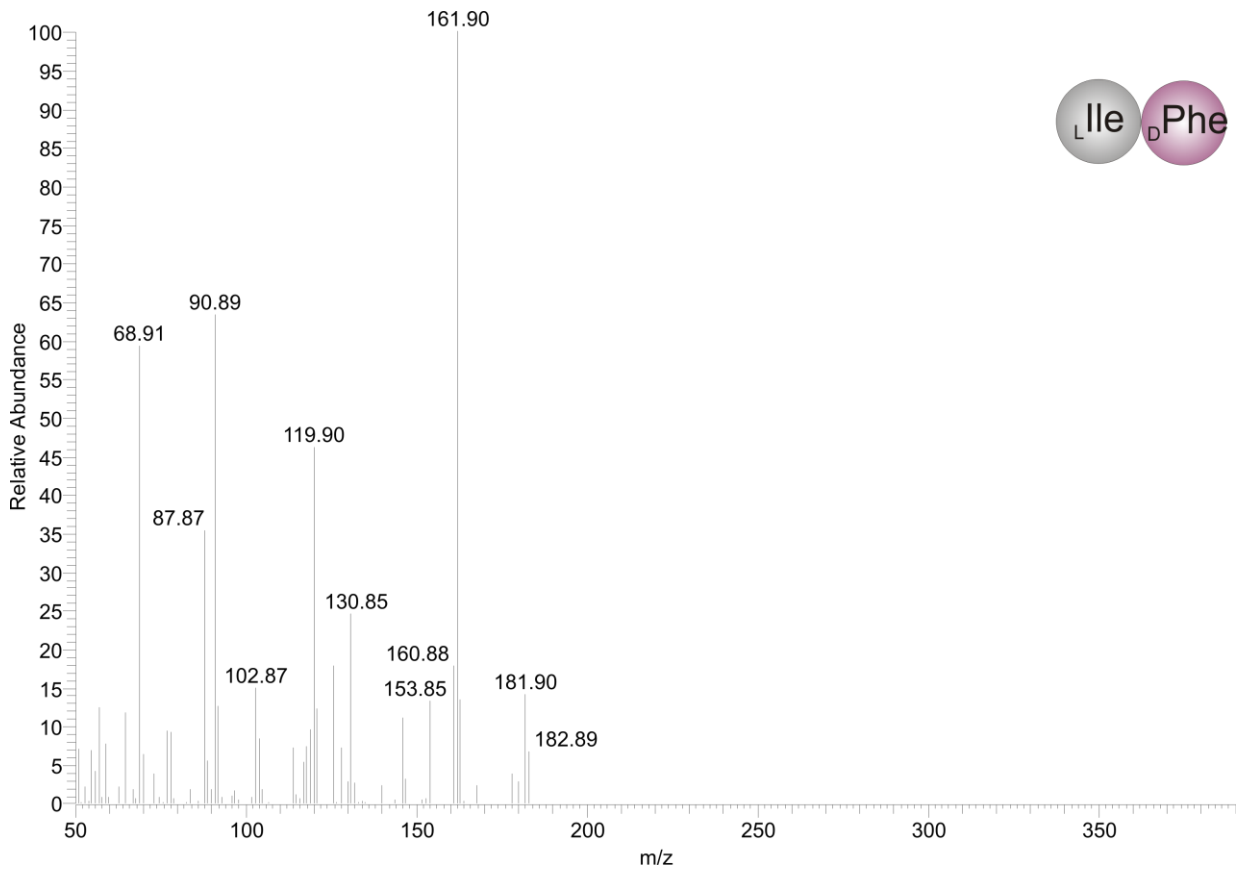


Figure S23. PCI-MS spectra of synthetic *N*-trifluoroacetyl-Leu-Leu-methyl ester.

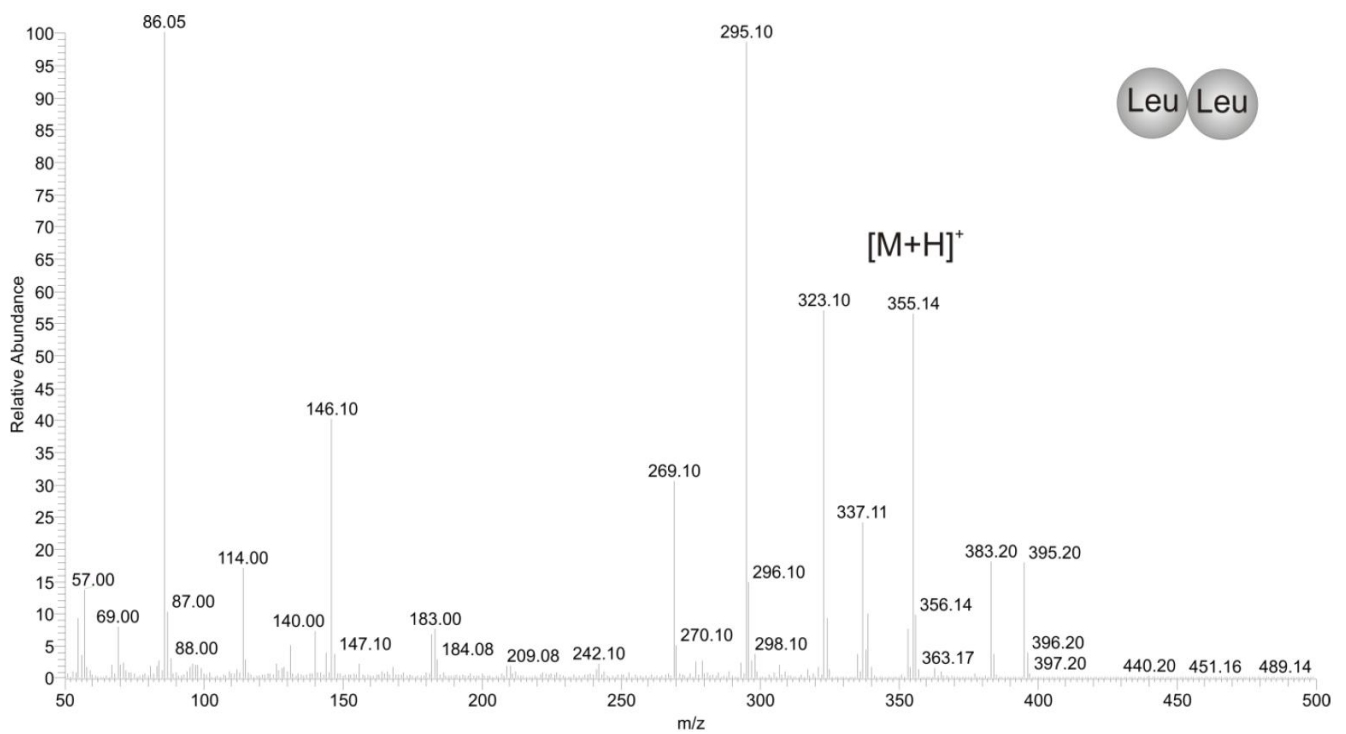


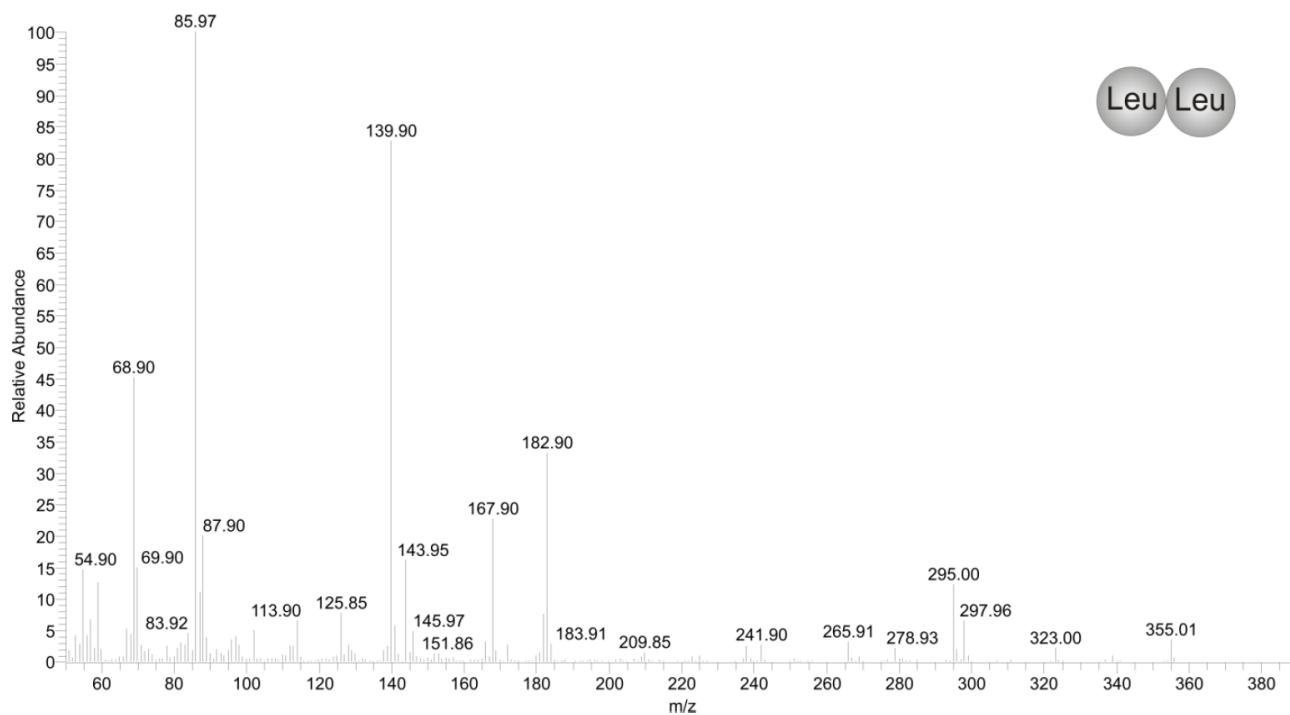
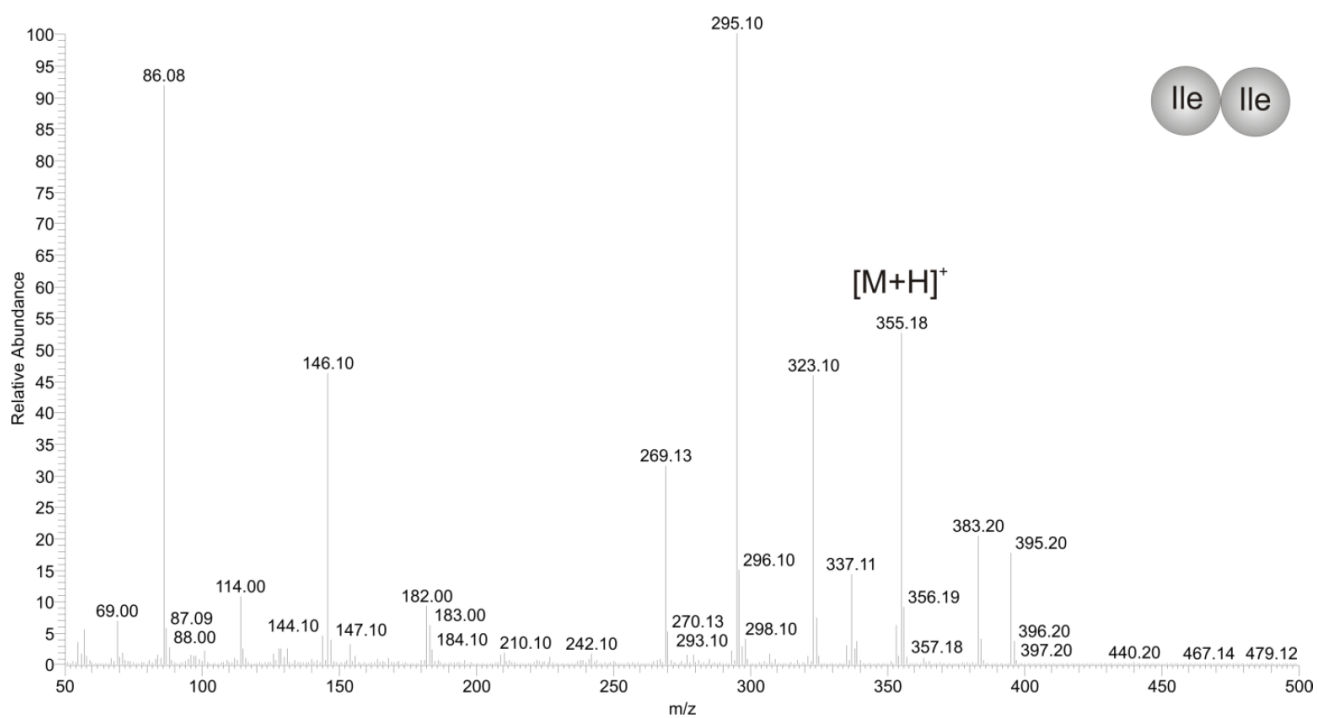
Figure S24. EI-MS spectra of synthetic *N*-trifluoroacetyl-Leu-Leu-methyl ester.**Figure S25.** PCI-MS spectra of synthetic *N*-trifluoroacetyl-Ile-Ile-methyl ester.

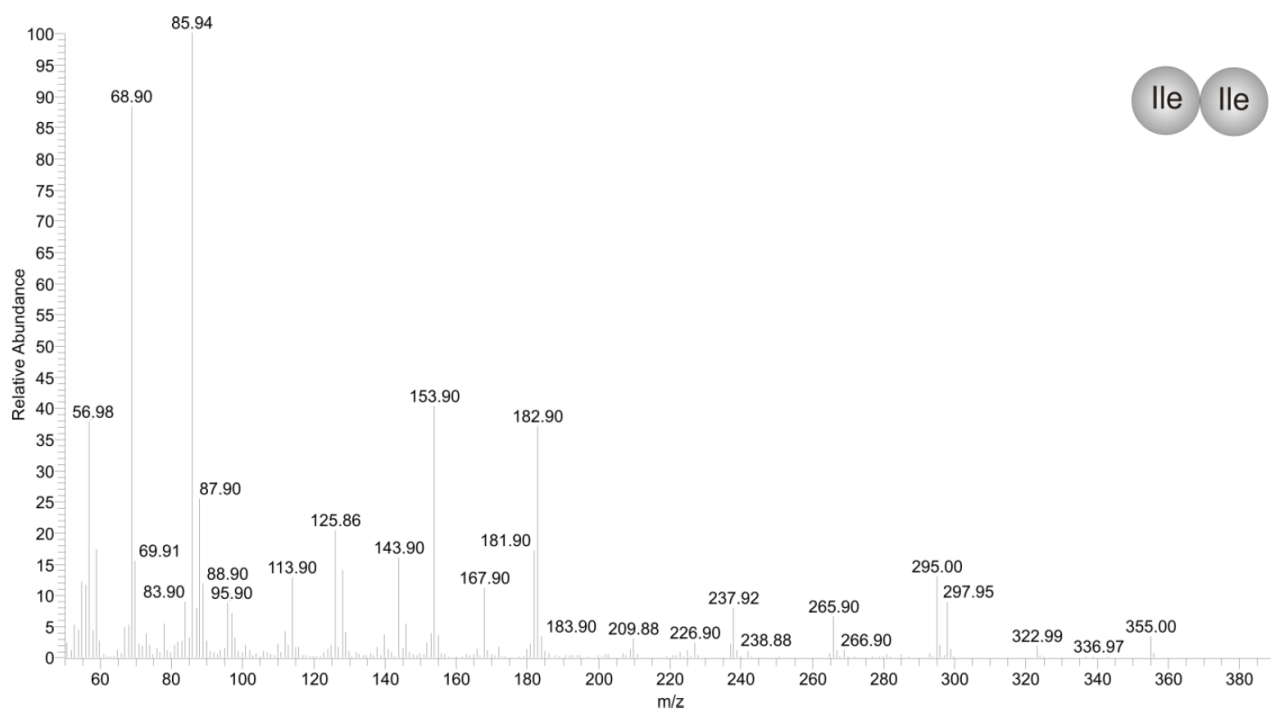
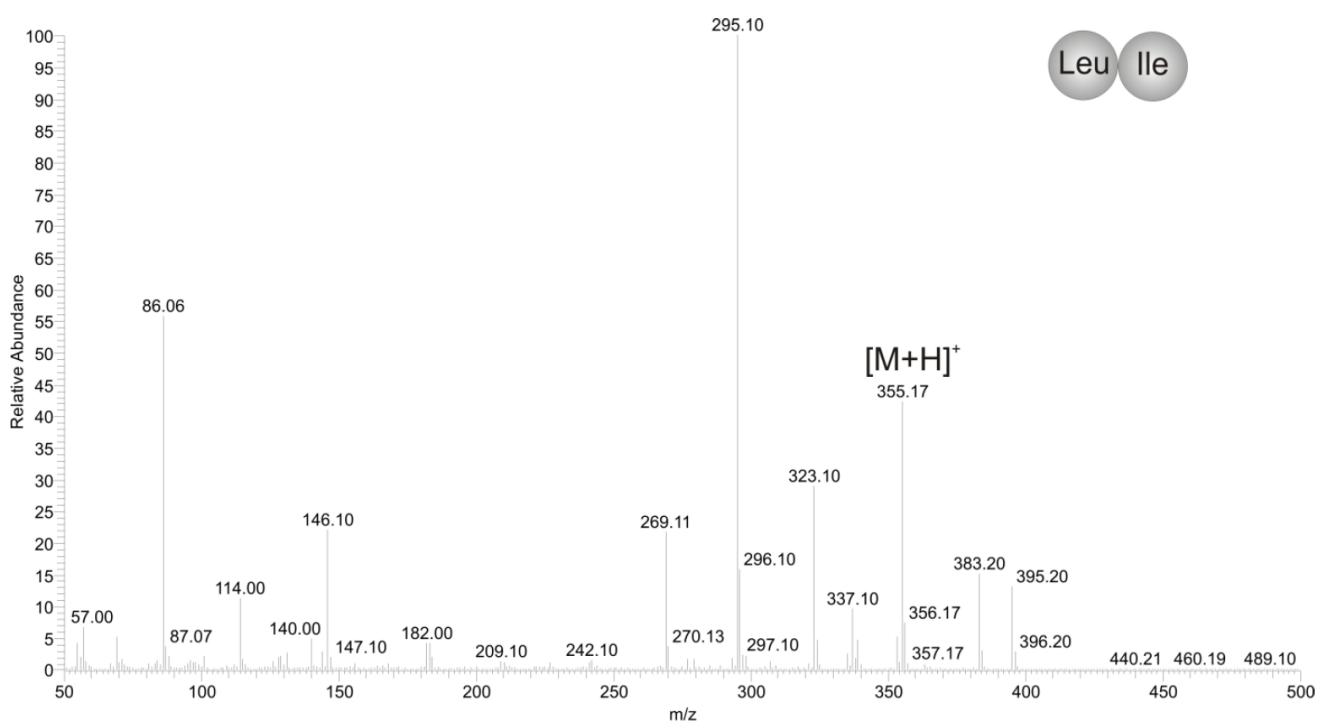
Figure S26. EI-MS spectra of synthetic *N*-trifluoroacetyl-Ile-Ile-methyl ester.**Figure S27.** PCI-MS spectra of synthetic *N*-trifluoroacetyl-Leu-Ile-methyl ester.

Figure S28. EI-MS spectra of synthetic *N*-trifluoroacetyl-Leu-Ile-methyl ester.

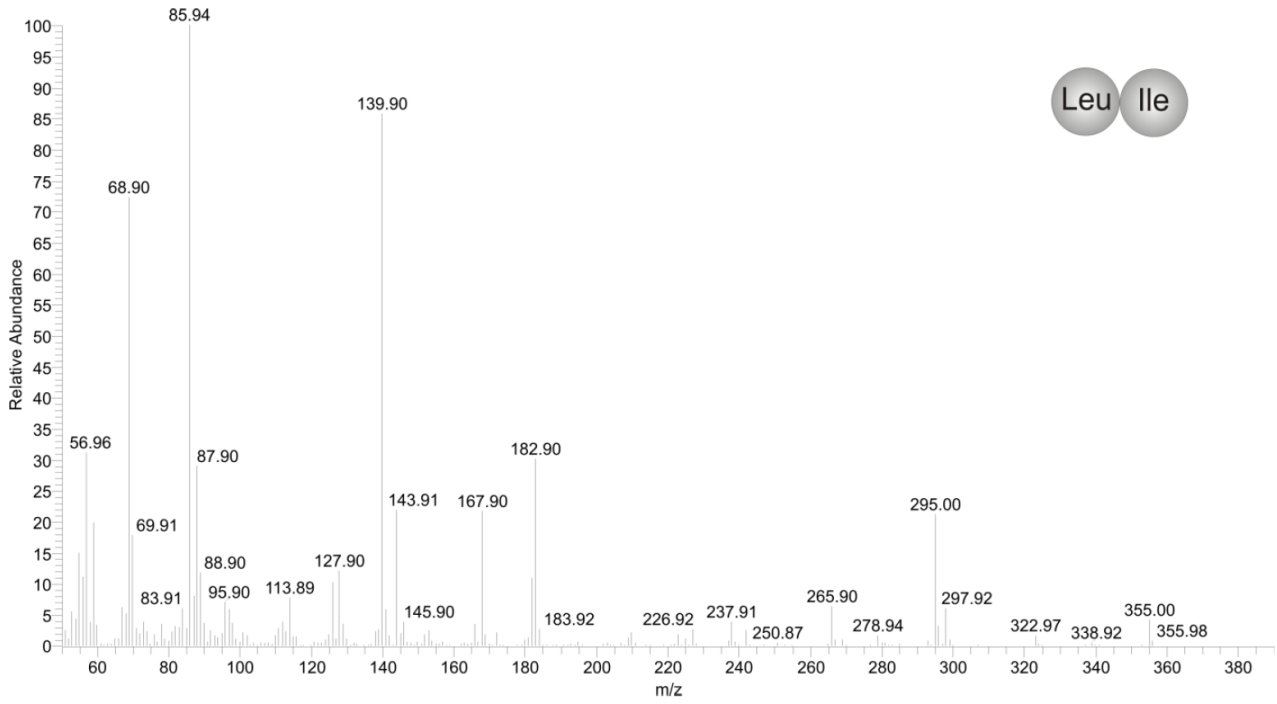


Figure S29. PCI-MS spectra of synthetic *N*-trifluoroacetyl-Ile-Leu-methyl ester.

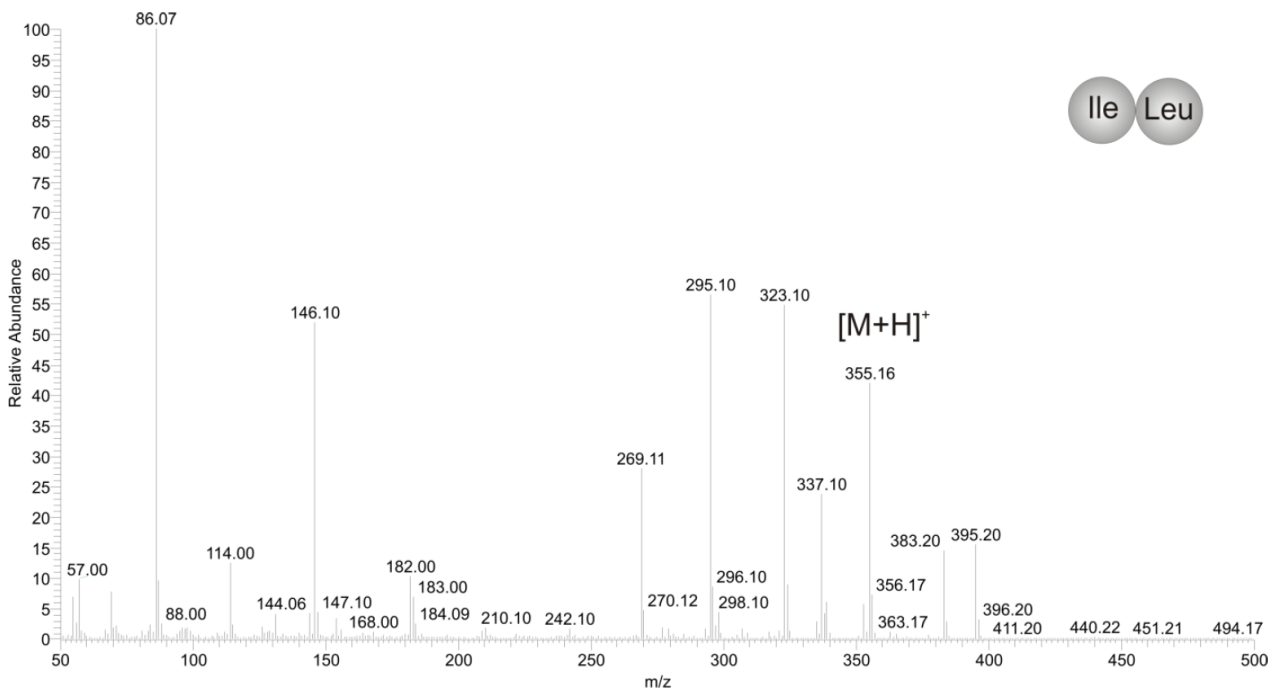


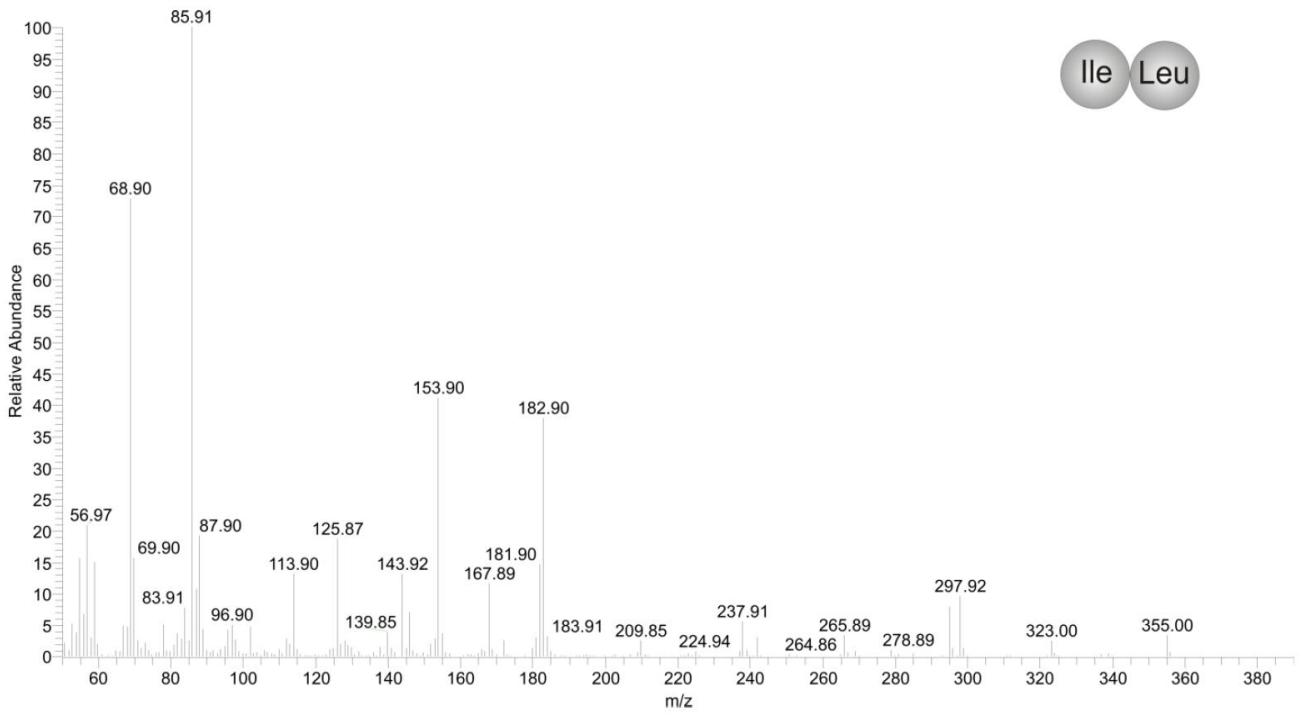
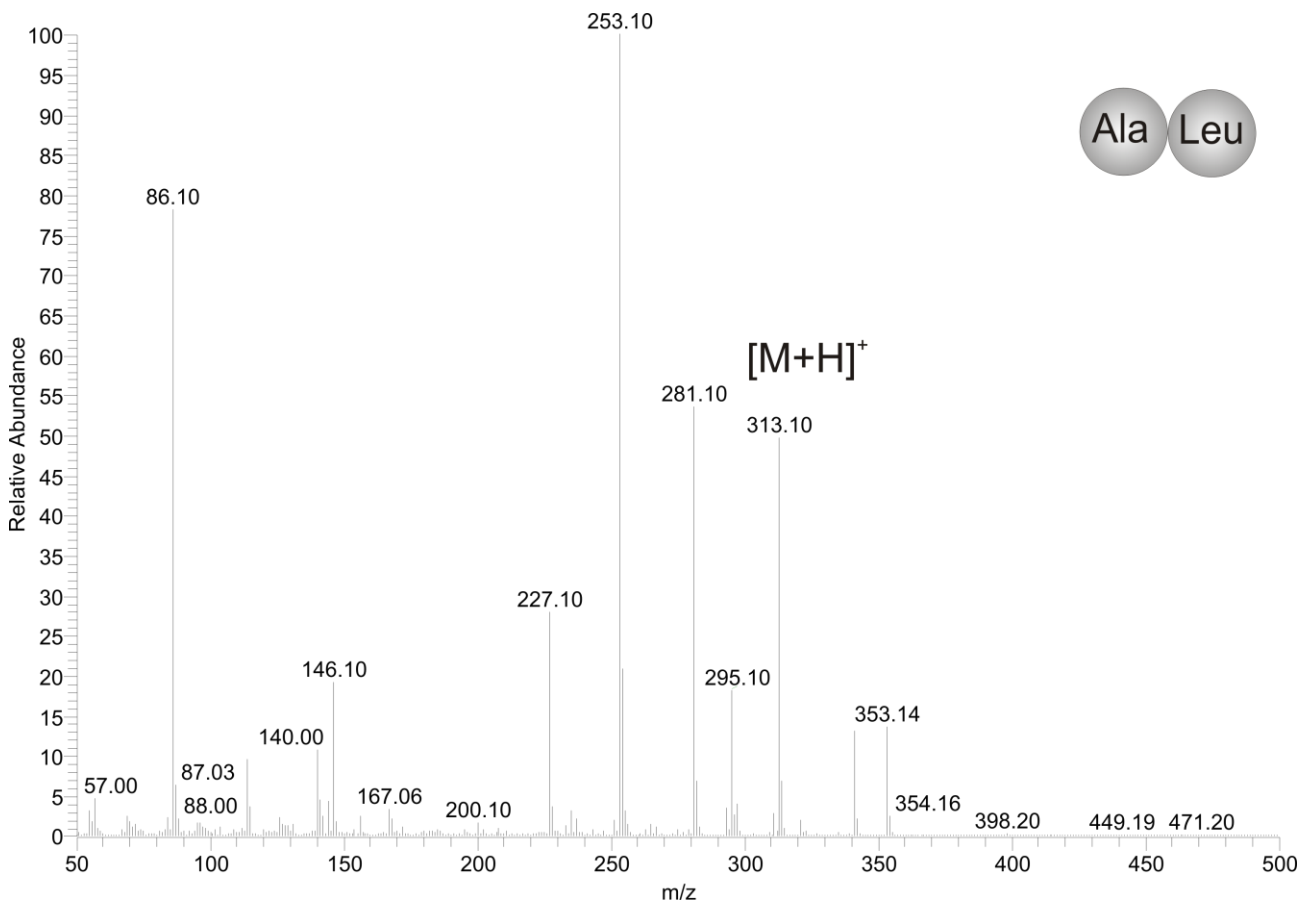
Figure S30. EI-MS spectra of synthetic *N*-trifluoroacetyl-Ile-Leu-methyl ester.**Figure S31.** PCI-MS spectra of synthetic *N*-trifluoroacetyl-Ala-Leu-methyl ester.

Figure S32. EI-MS spectra of synthetic *N*-trifluoroacetyl-Ala-Leu-methyl ester.

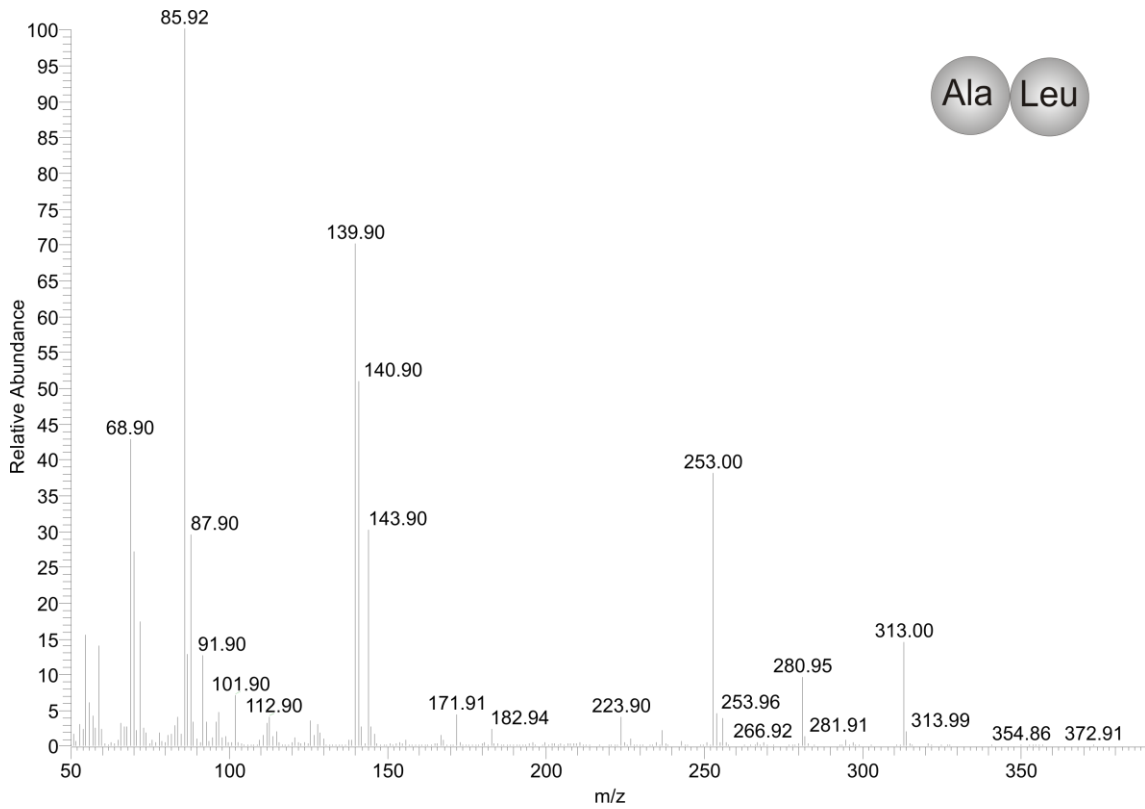


Figure S33. PCI-MS spectra of synthetic *N*-trifluoroacetyl-Leu-Ala-methyl ester.

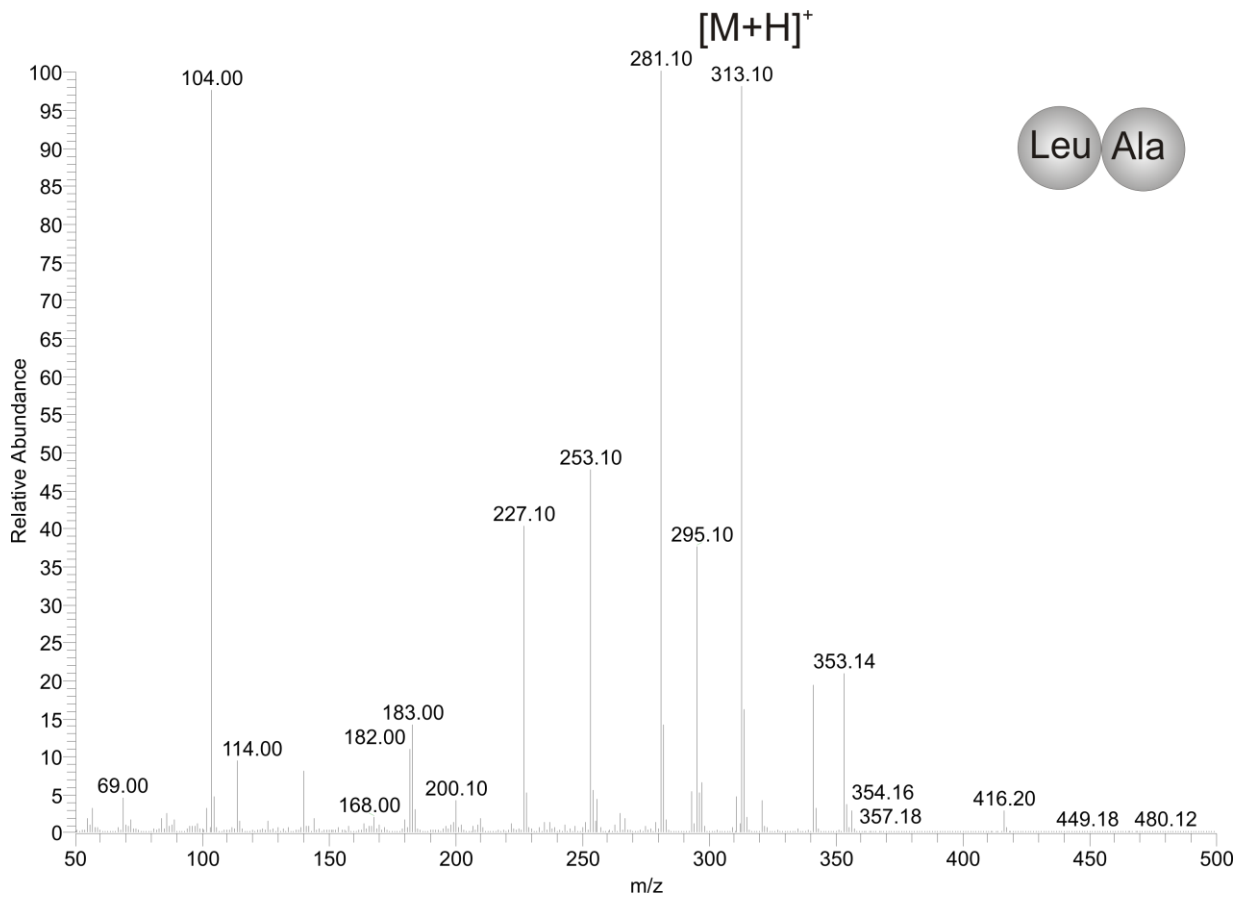


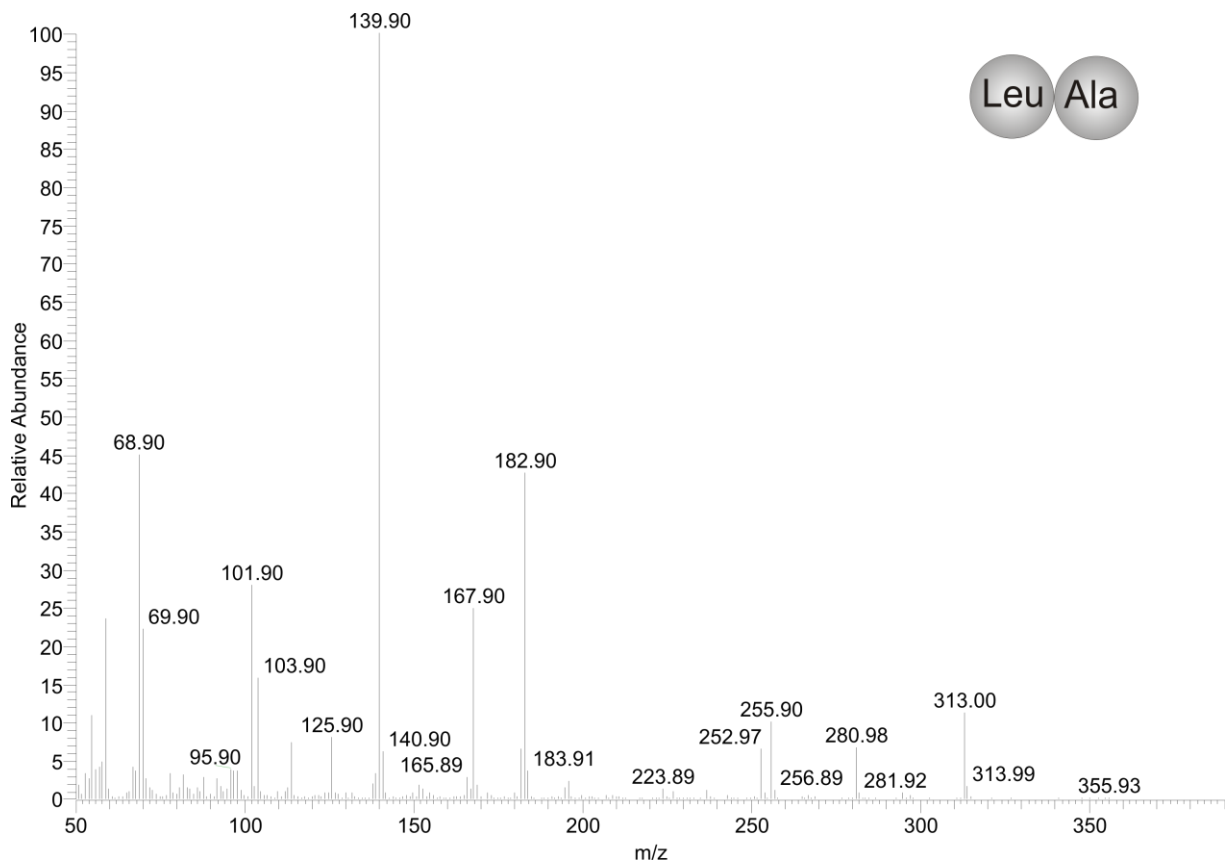
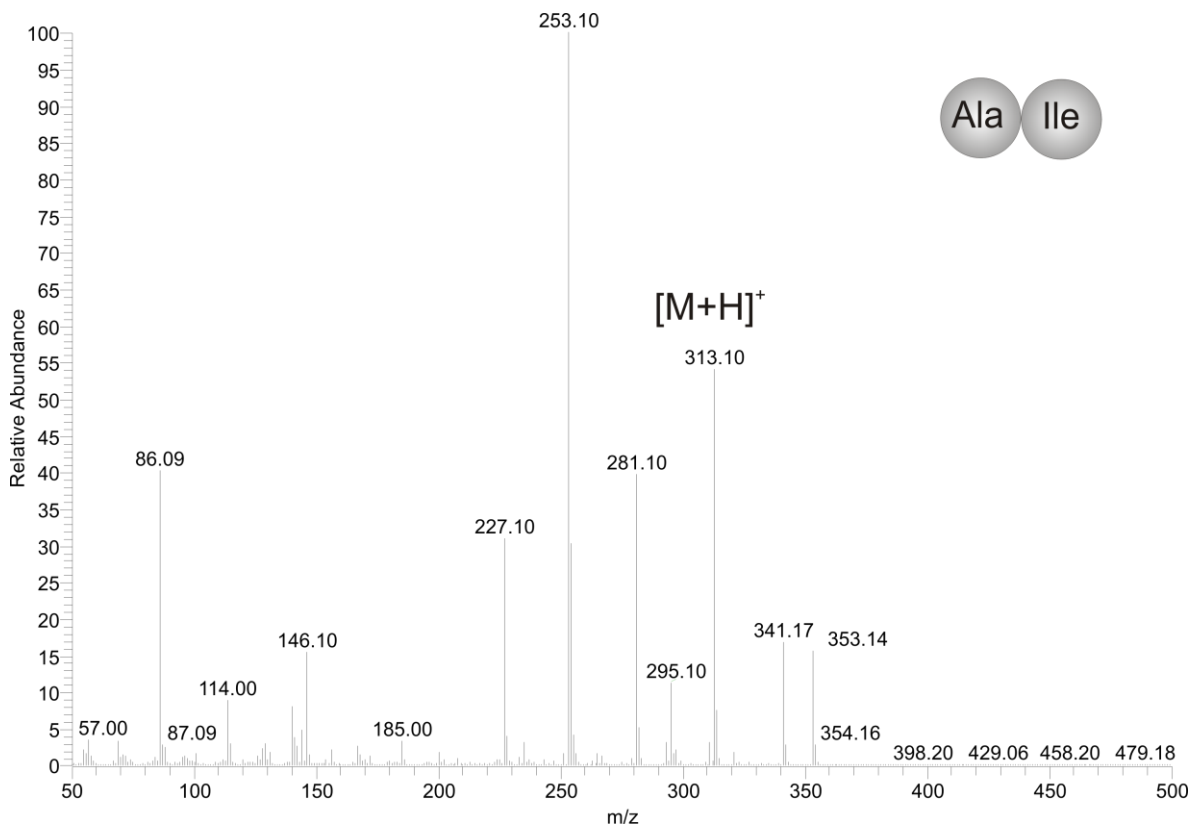
Figure S34. EI-MS spectra of synthetic *N*-trifluoroacetyl-Leu-Ala-methyl ester.**Figure S35.** PCI-MS spectra of synthetic *N*-trifluoroacetyl-Ala-Ile-methyl ester.

Figure S36. EI-MS spectra of synthetic *N*-trifluoroacetyl-Ala-Ile-methyl ester.

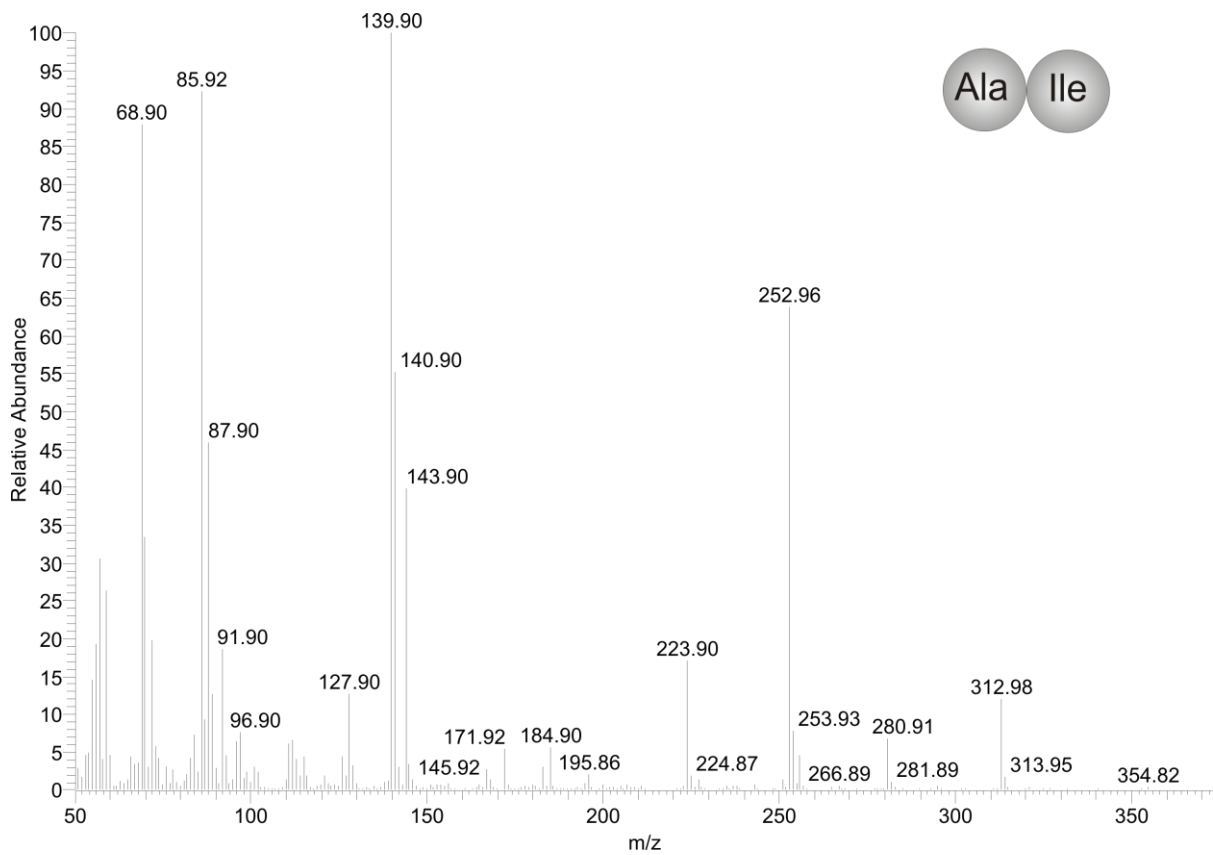


Figure S37. PCI-MS spectra of synthetic *N*-trifluoroacetyl-Ile-Ala-methyl ester.

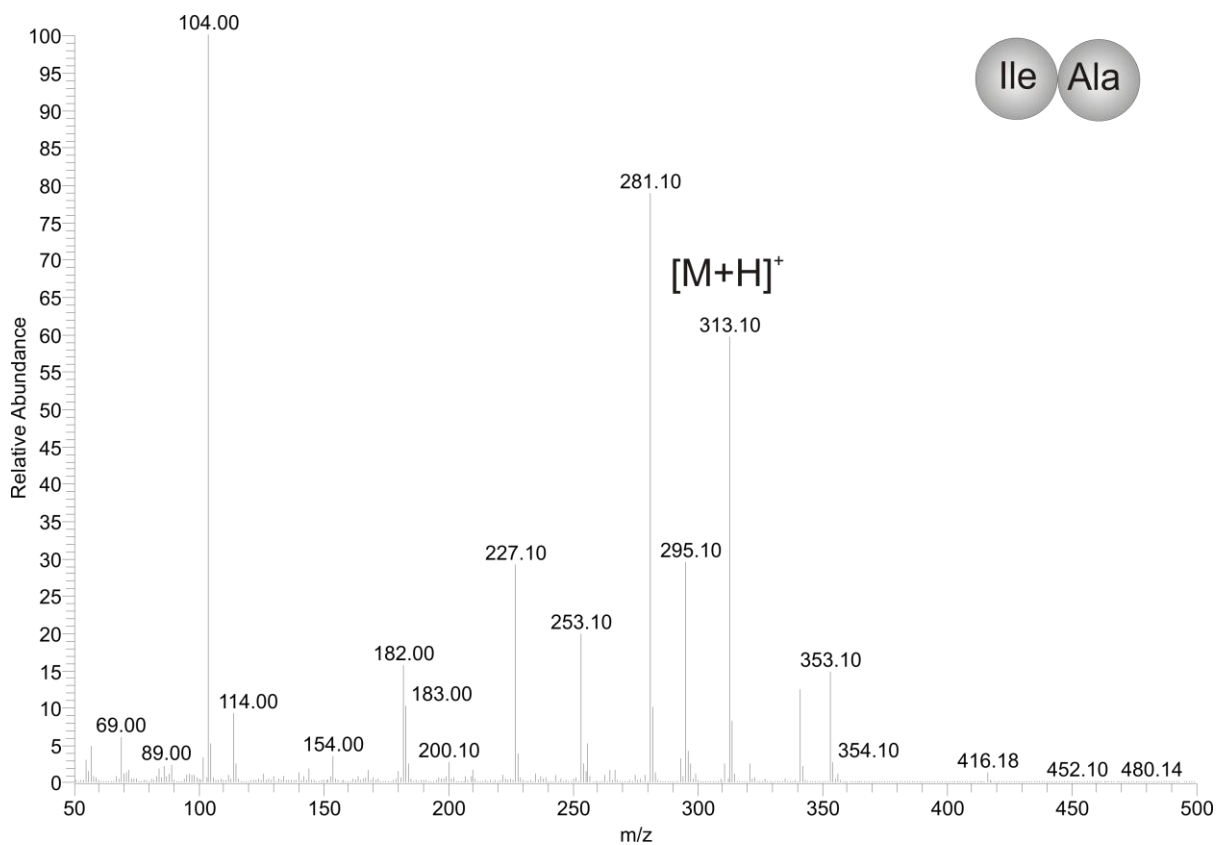


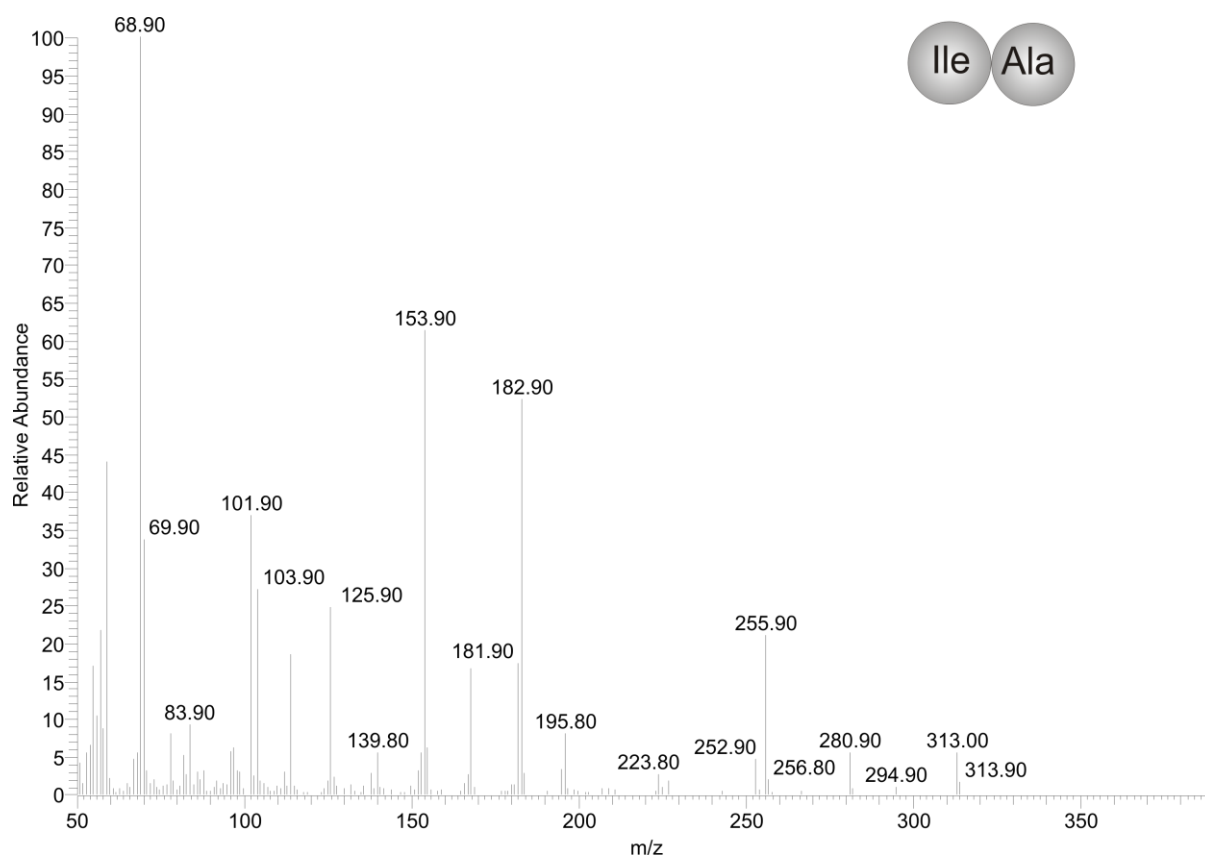
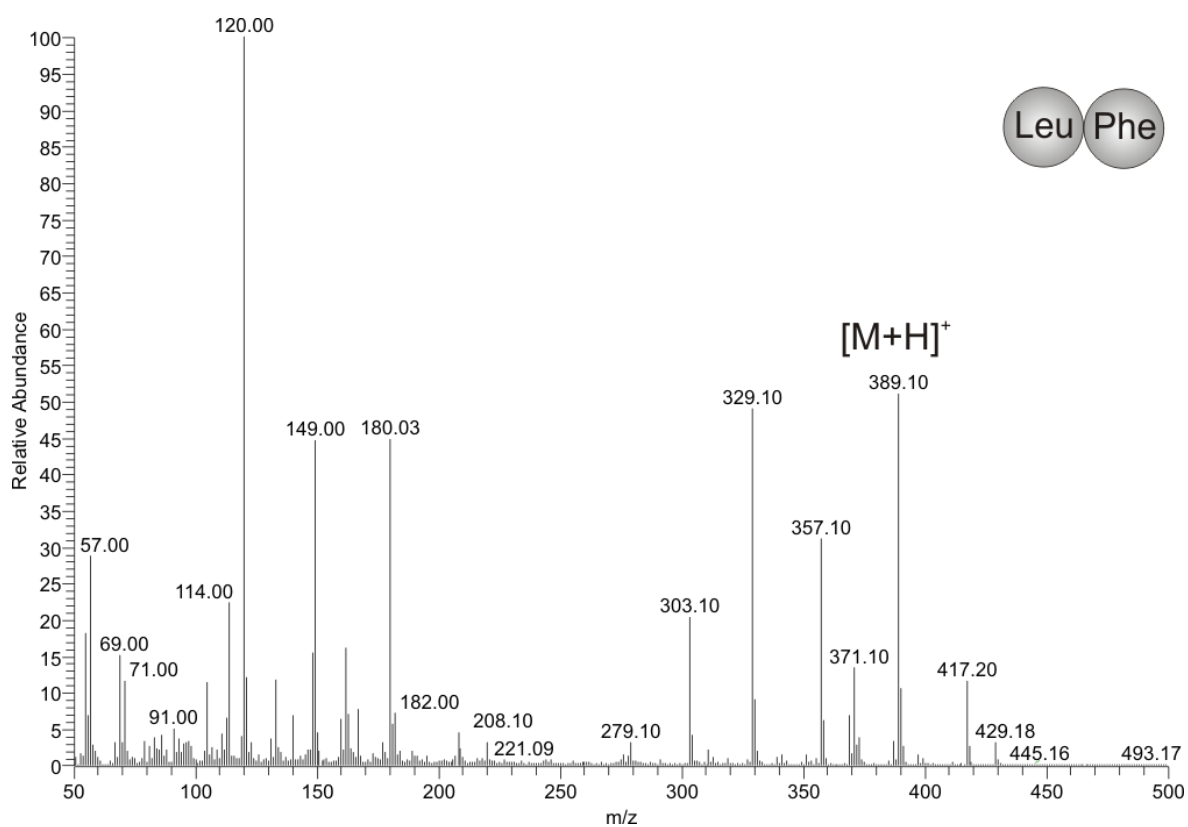
Figure S38. EI-MS spectra of synthetic *N*-trifluoroacetyl-Ile-Ala-methyl ester.**Figure S39.** PCI-MS spectra of synthetic *N*-trifluoroacetyl-Leu-Phe-methyl ester.

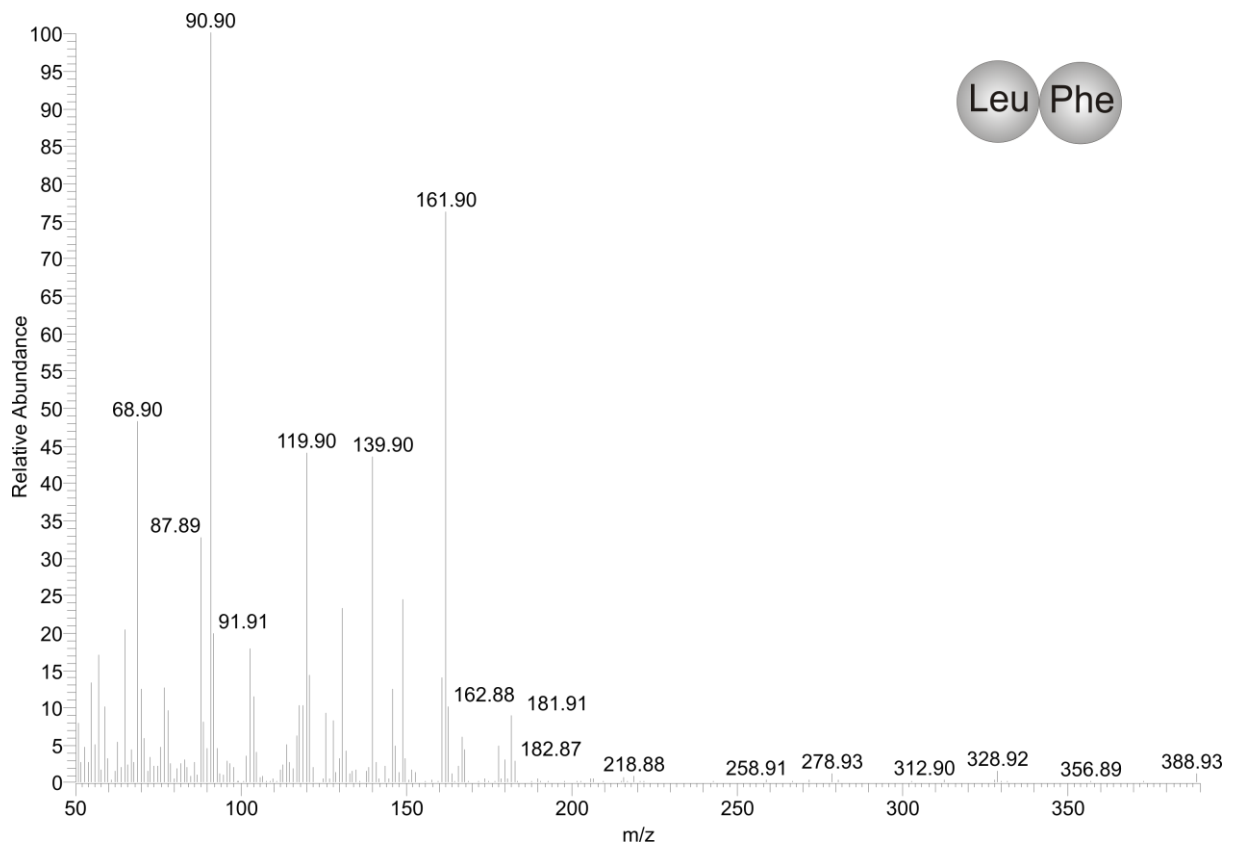
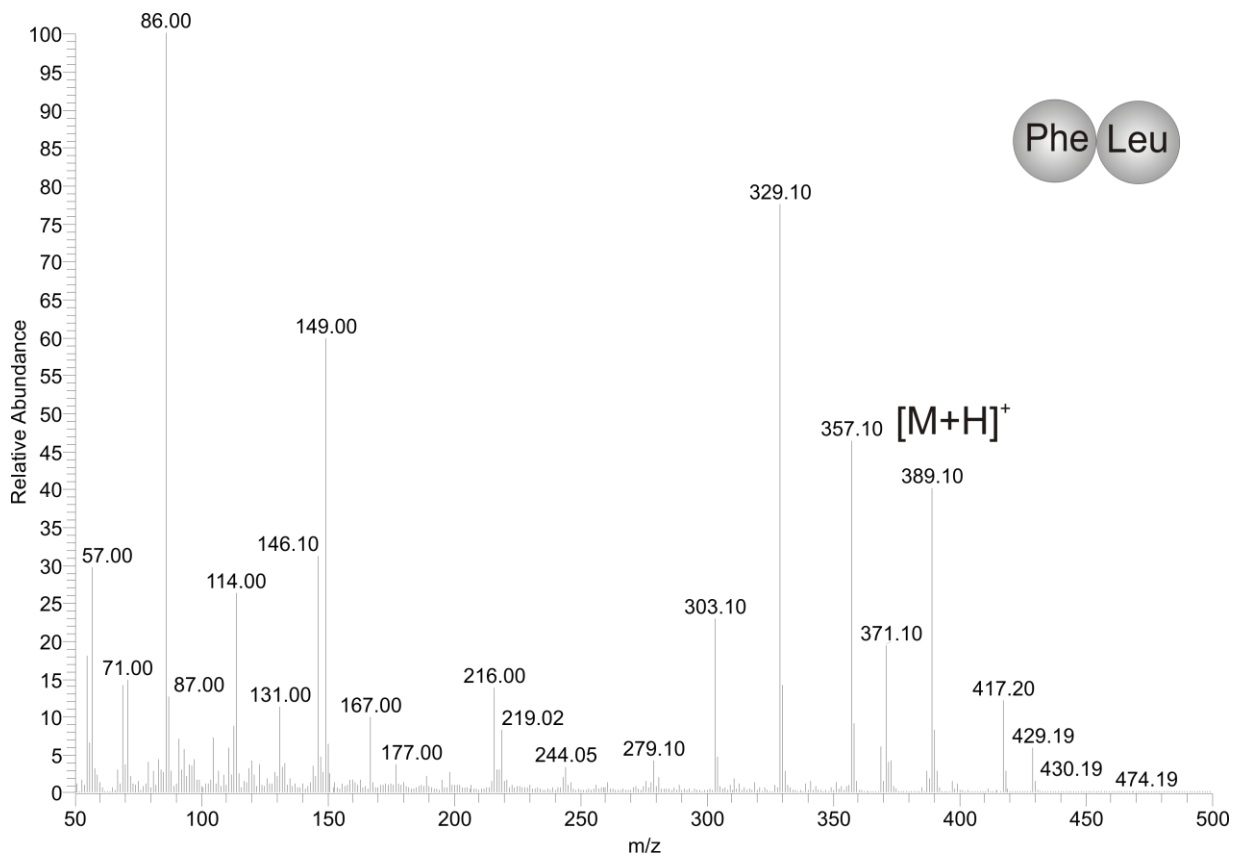
Figure S40. EI-MS spectra of synthetic *N*-trifluoroacetyl-Leu-Phe-methyl ester.**Figure S41.** PCI-MS spectra of synthetic *N*-trifluoroacetyl-Phe-Leu-methyl ester.

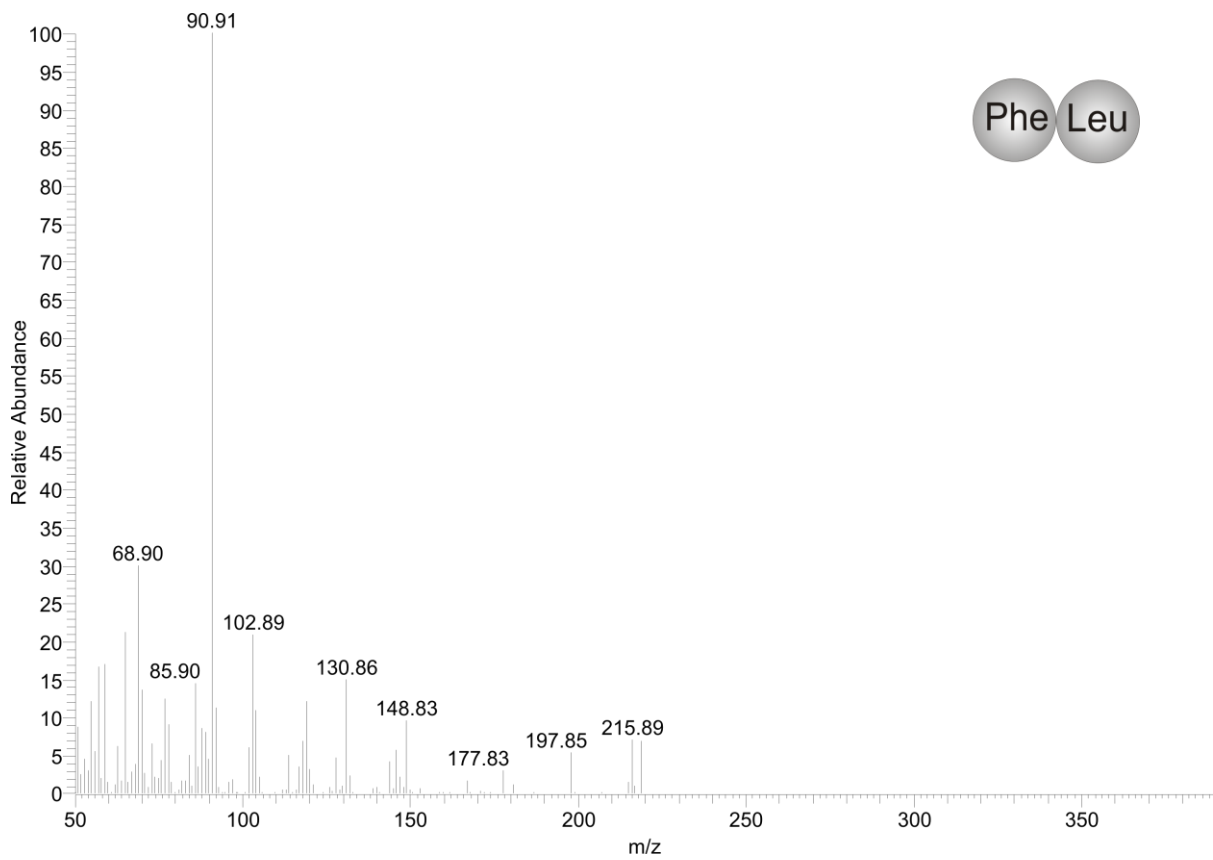
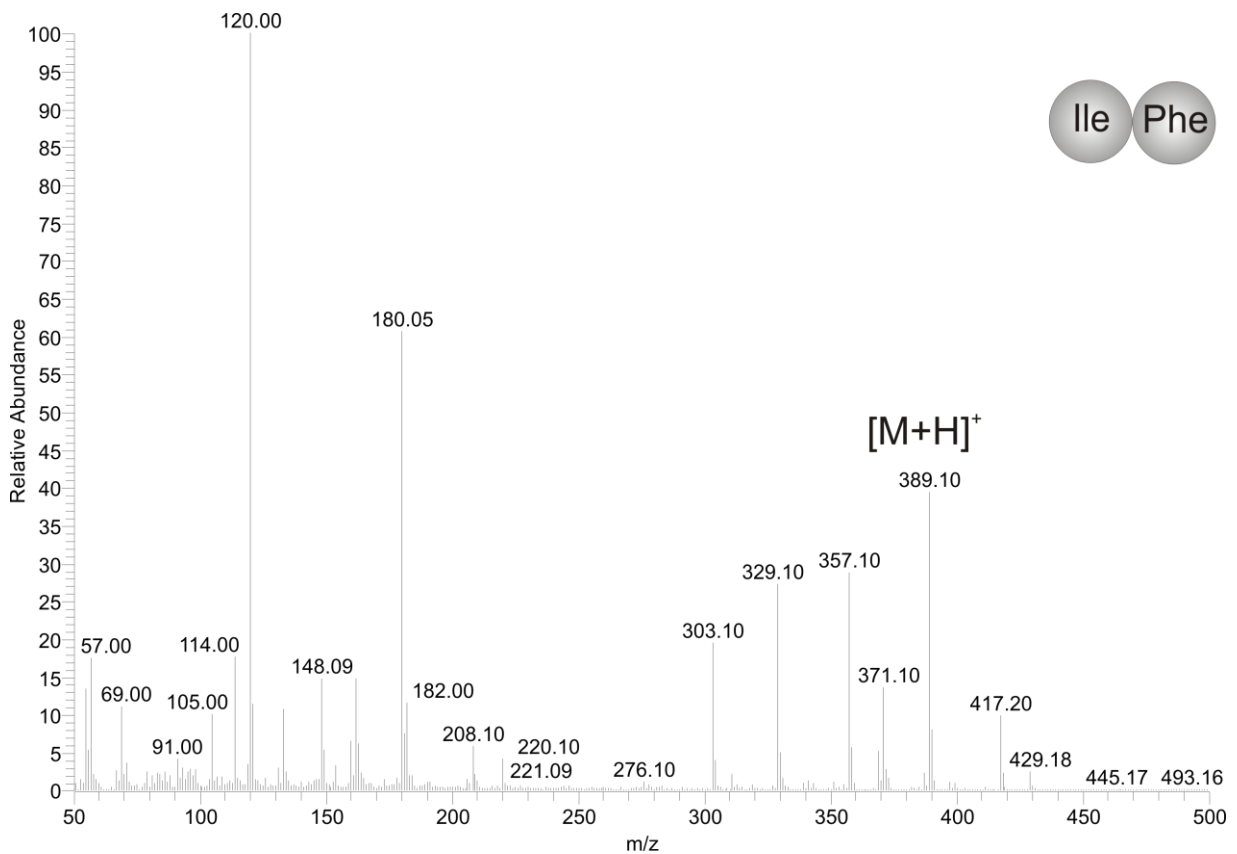
Figure S42. EI-MS spectra of synthetic *N*-trifluoroacetyl-Phe-Leu-methyl ester.**Figure S43.** PCI-MS spectra of synthetic *N*-trifluoroacetyl-Ile-Phe-methyl ester.

Figure S44. EI-MS spectra of synthetic *N*-trifluoroacetyl-Ile-Phe-methyl ester.

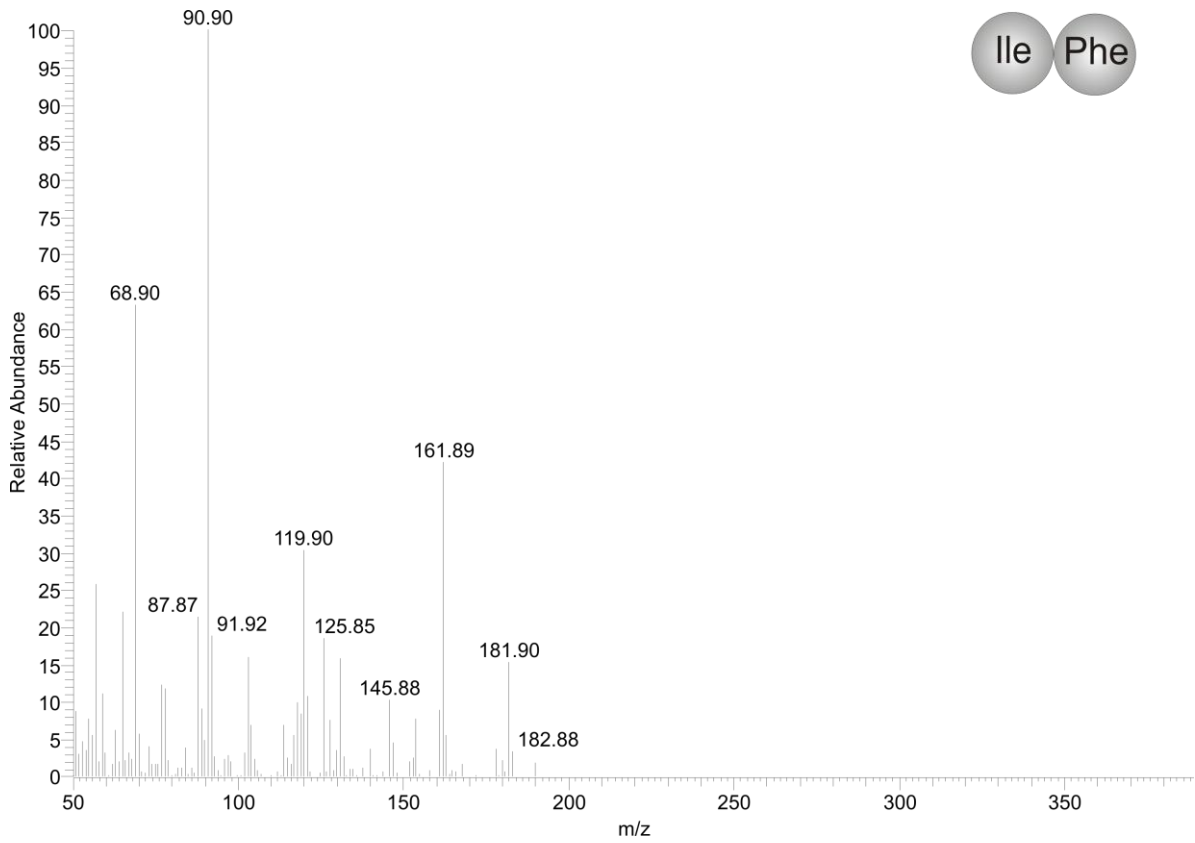


Figure S45. PCI-MS spectra of synthetic *N*-trifluoroacetyl-Phe-Ile-methyl ester.

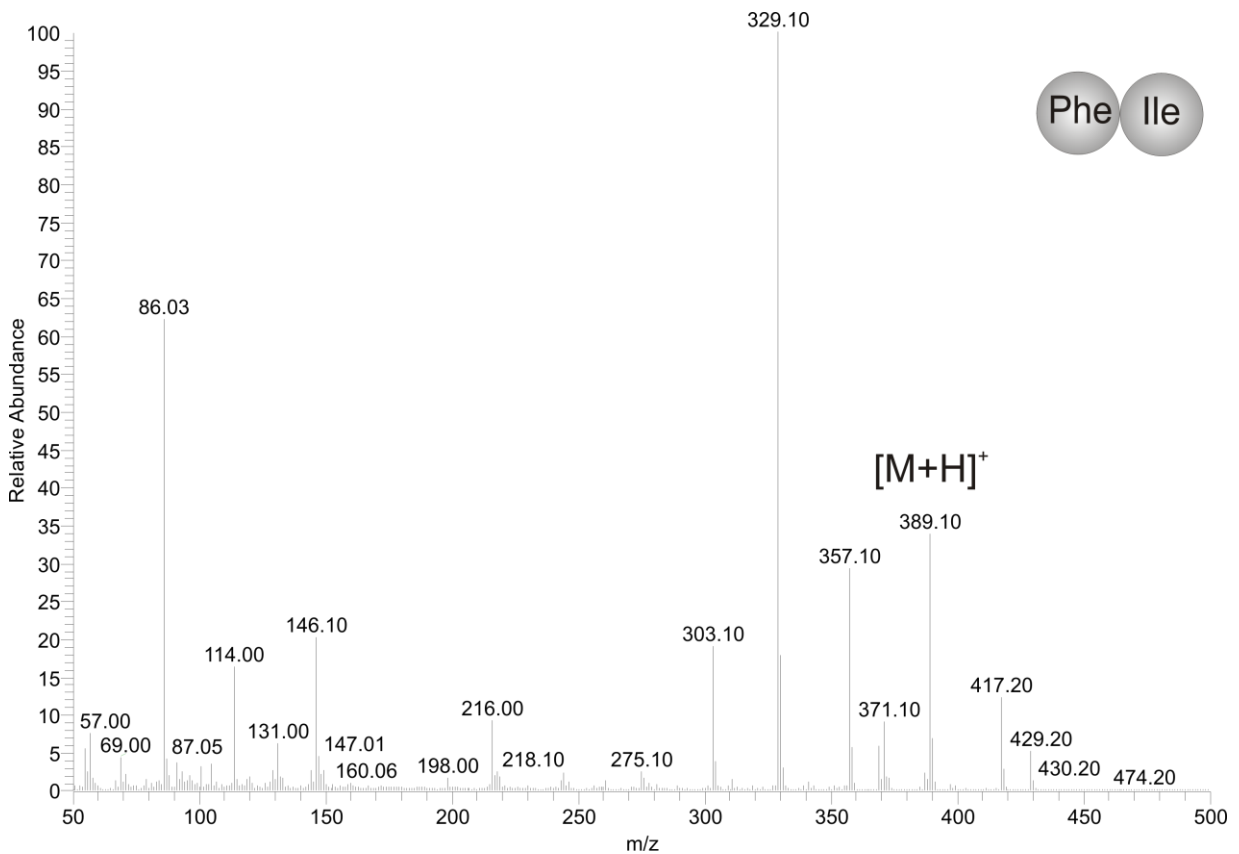


Figure S46. EI-MS spectra of synthetic *N*-trifluoroacetyl-Phe-Ile-methyl ester.