

Supporting information for:

Zinc-chelating mechanism of sea cucumber (*Stichopus japonicus*)-derived synthetic peptides

Xiaoyang Liu^{1,2}, Zixu Wang¹, Fawen Yin^{1,2}, YuXin Liu^{1,2}, Ningbo Qin^{1,2}, Yoshimasa Nakamura³, Fereidoon Shahidi⁴, Chenxu Yu^{2,5}, Dayong Zhou^{1,2,*}, Beiwei Zhu^{1,2}

¹ School of Food Science and Technology, Dalian Polytechnic University, Dalian, China, 116034

² National Engineering Research Center of Seafood, Dalian, PR China, 116034

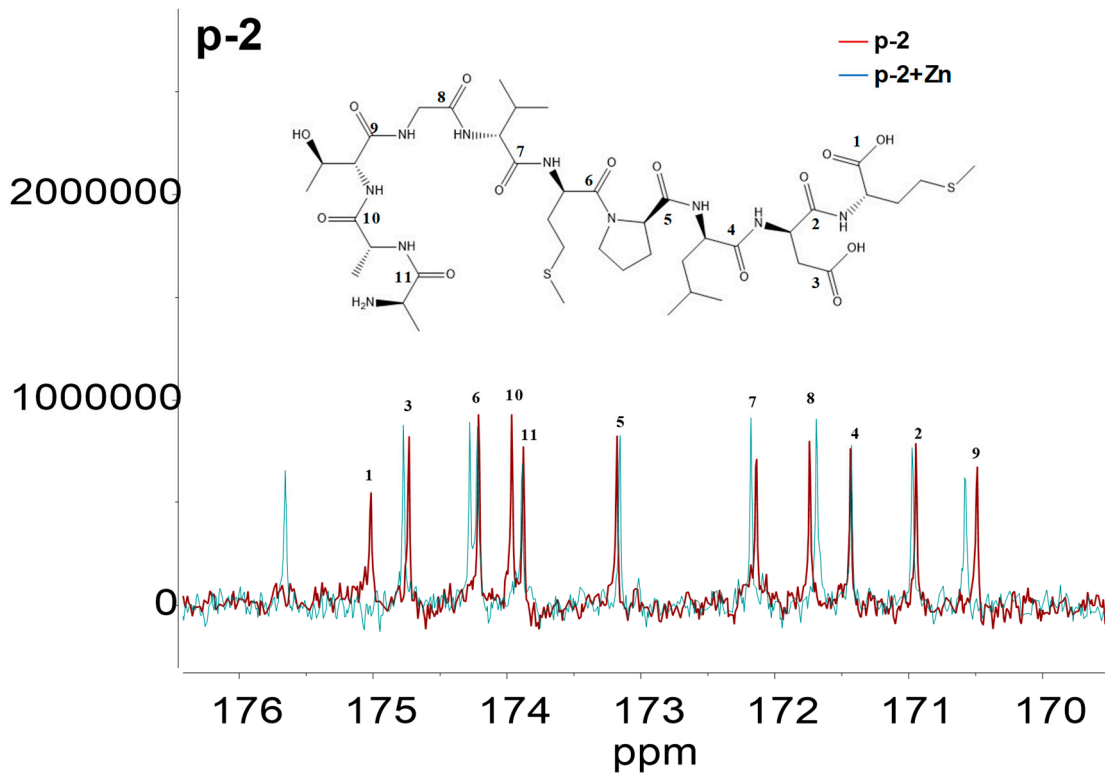
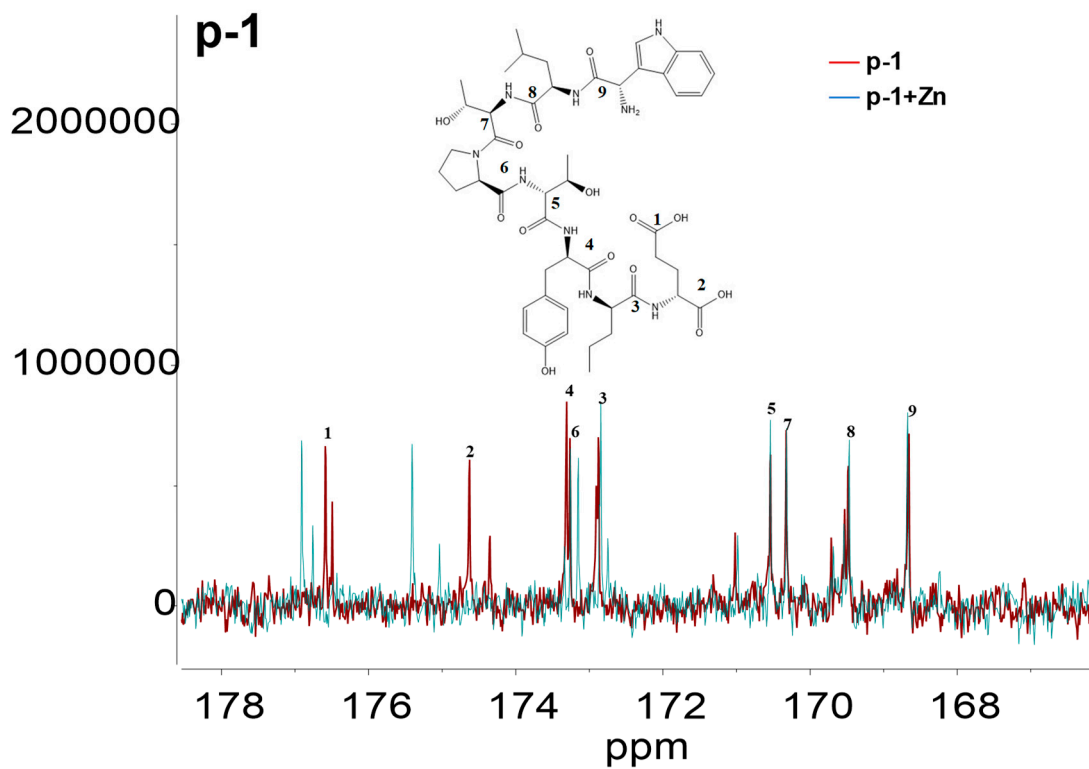
³ Graduate School of Environmental and Life Science, Okayama University, Okayama, Japan, 700-8530.

⁴ Department of Biochemistry, Memorial University of Newfoundland, St. John's, NL A1B 3X9, Canada.

⁵ Department of Agricultural and Biosystems Engineering, Iowa State University, IA 50011, USA

Corresponding Authors:

Dayong Zhou^{1,2,*}. E-mail: zdyzfl@163.com; Tel: +86-411 86323453



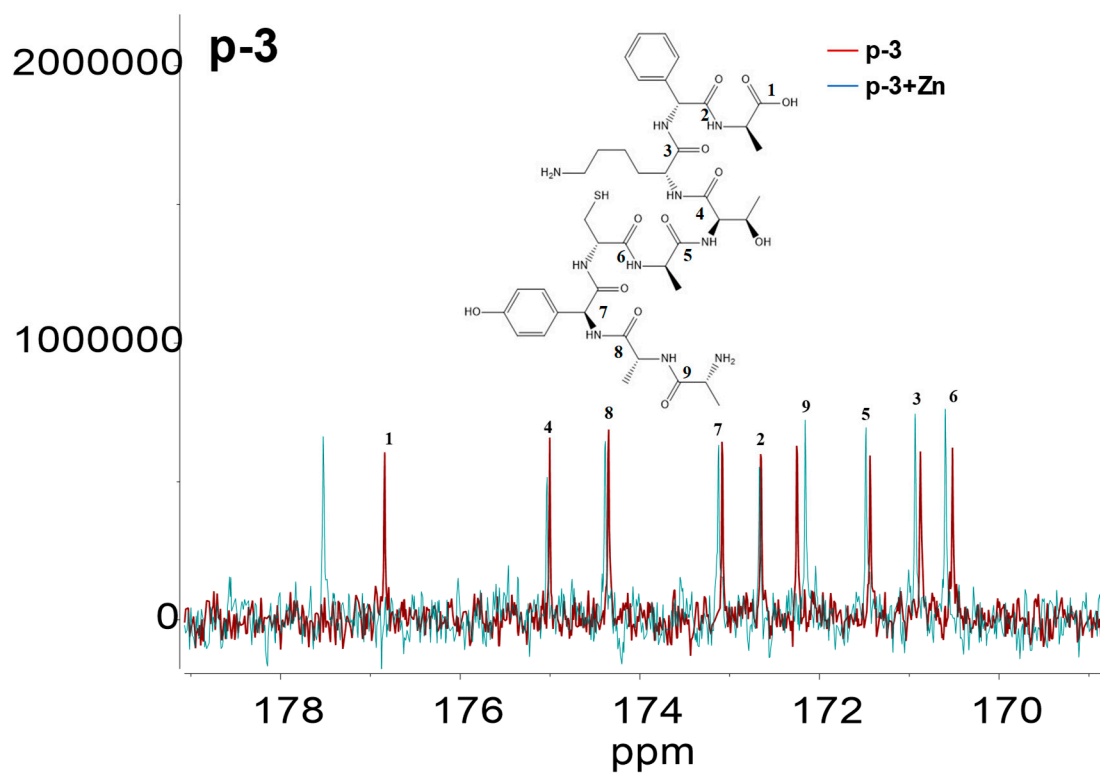
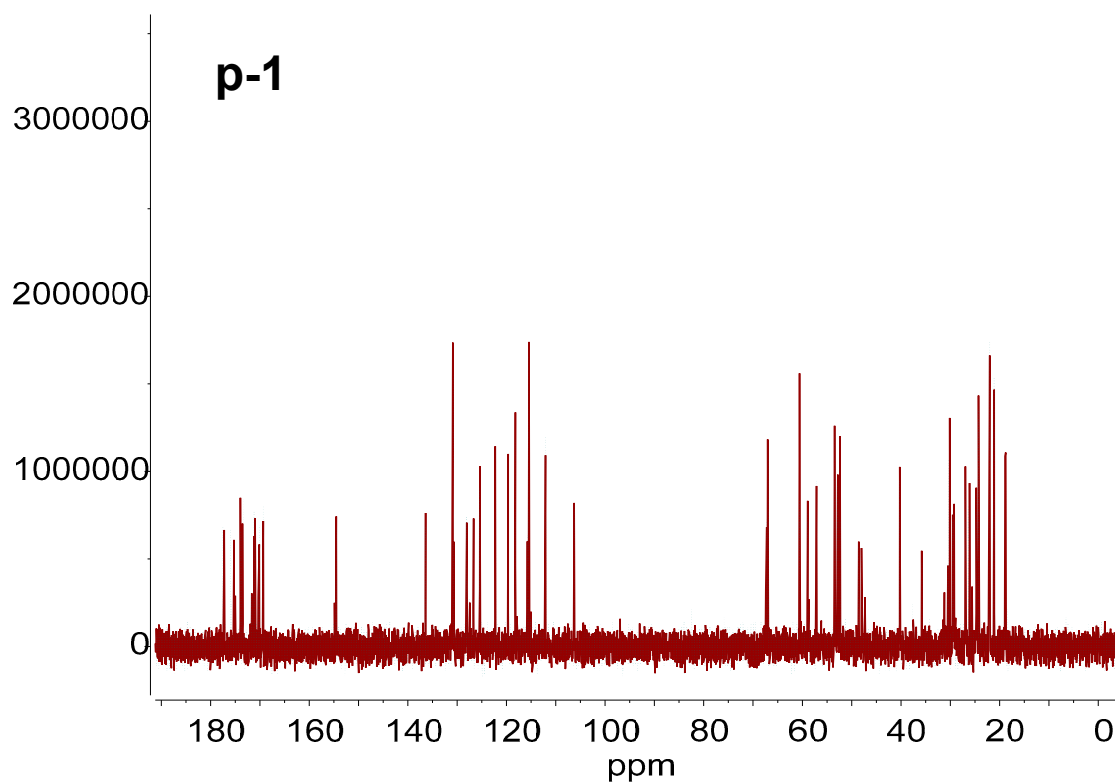


Figure. S1

^{13}C NMR spectra of ZCPs and ZCP-zinc complexes from 160 to 180 ppm. Numbers represent designated carboxyl and acylamino.



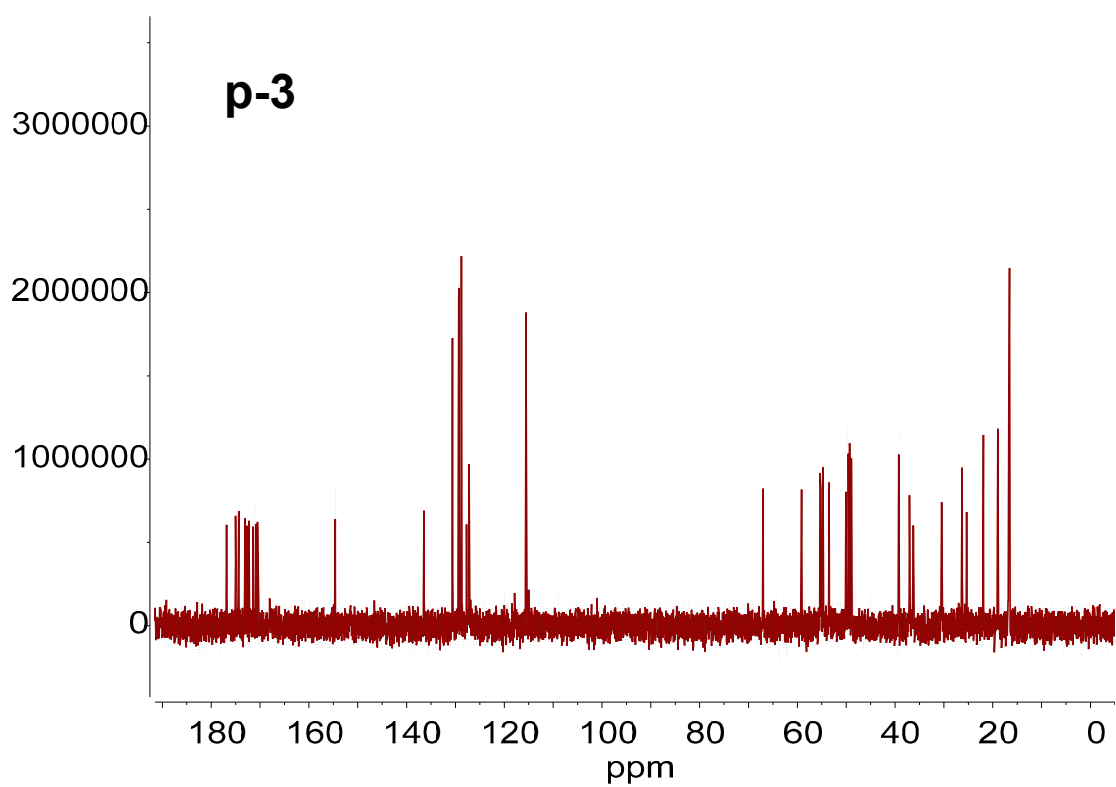
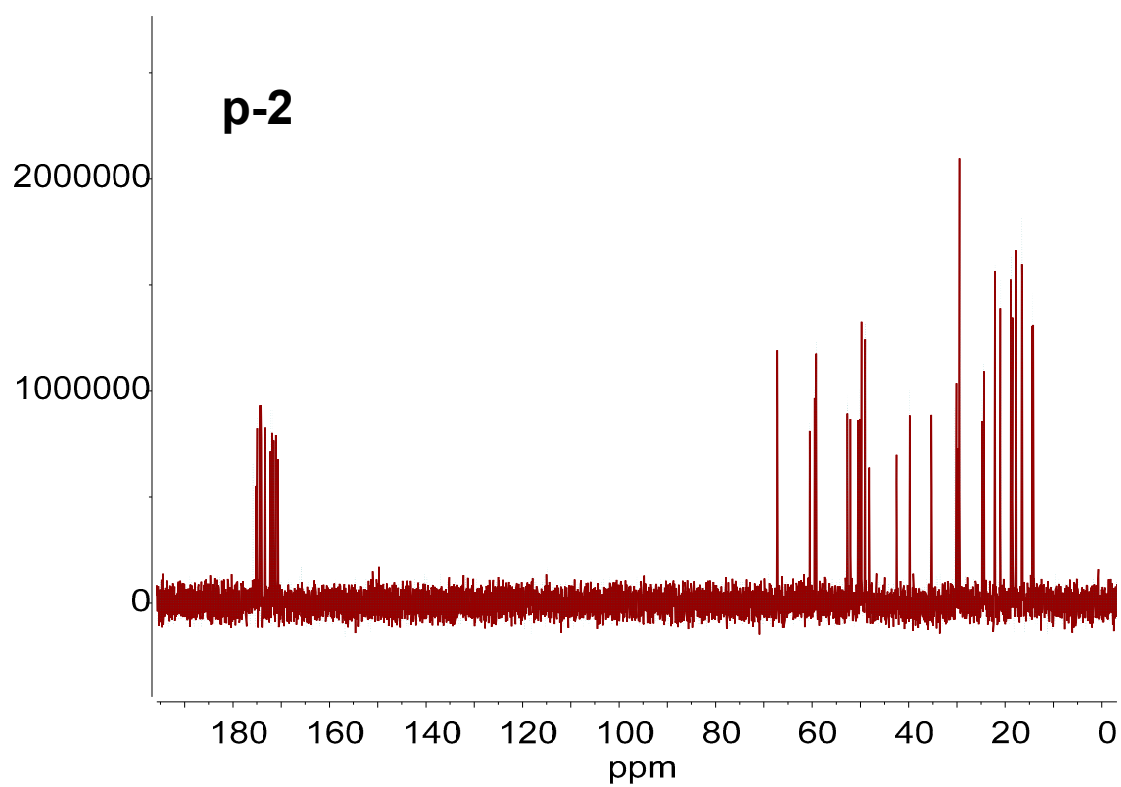


Figure.S2

Full spectra of ¹³C NMR spectra of ZCPs and ZCP-zinc complexes.

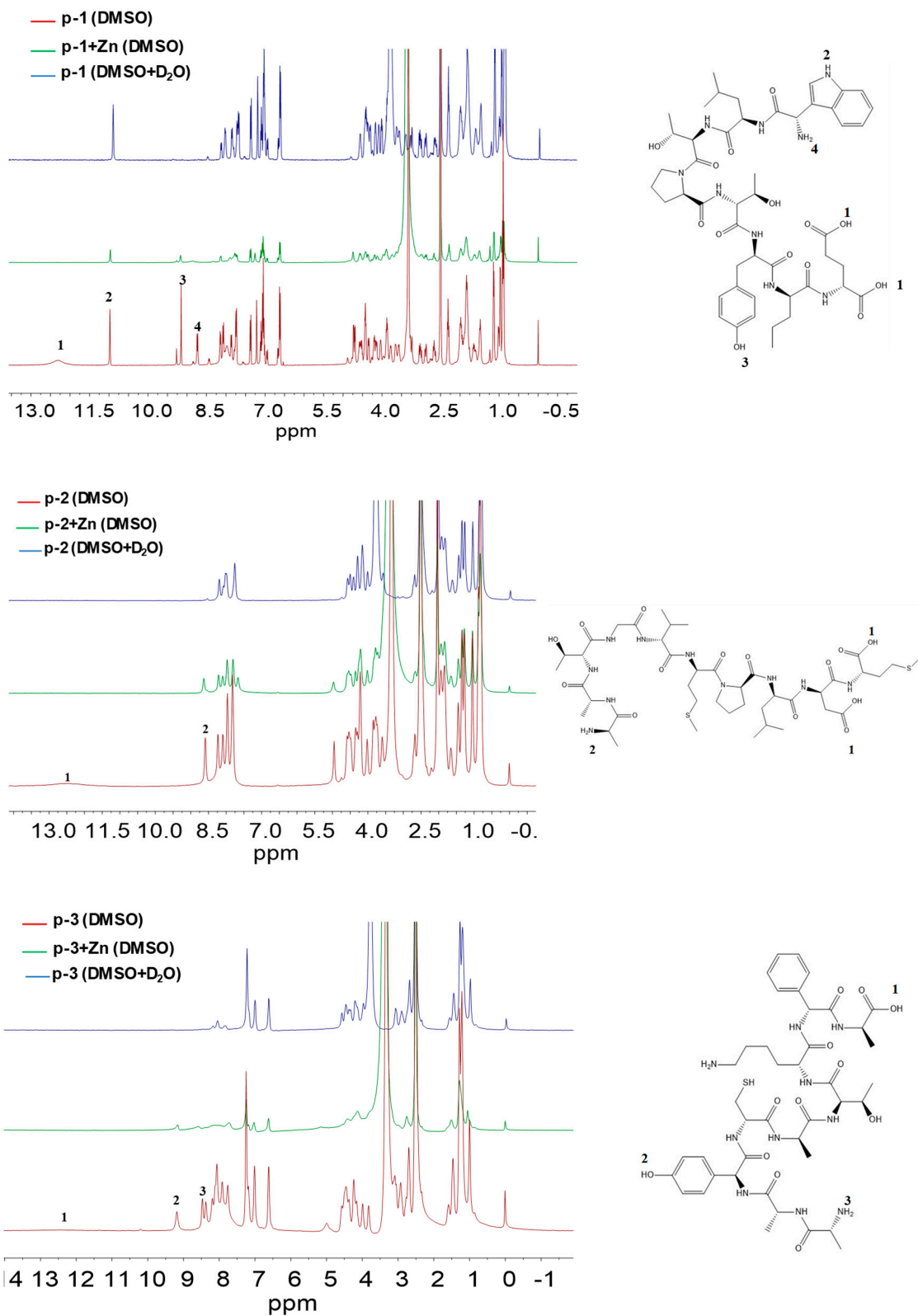


Figure.S3

¹H NMR spectra of ZCPs and ZCP-zinc complexes (dissolved in DMSO-d₆ or DMSO-d₆/10%D₂O). Numbers represent designated hydrogen.

