



## Supplementary Materials

# Hydrolysis and Photolysis Kinetics, and Identification of Degradation Products of the Novel Bactericide 2-(4-Fluorobenzyl)-5-(Methylsulfonyl)-1,3,4-Oxadiazole in Water

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$^1\text{H}$  NMR,  $^{13}\text{C}$  NMR,  $^{19}\text{F}$  NMR spectra of FBEZF's degradation product P<sub>1</sub> (2-(4-fluorobenzyl)-5-methoxy-1,3,4-oxadiazole) were recorded on a JEOL ECX 500 NMR spectrometer operated at room temperature and 500 MHz using DMSO-d<sub>6</sub> as a solvent and TMS as an internal standard. MS spectra were recorded on LC/MSD Trap VL.

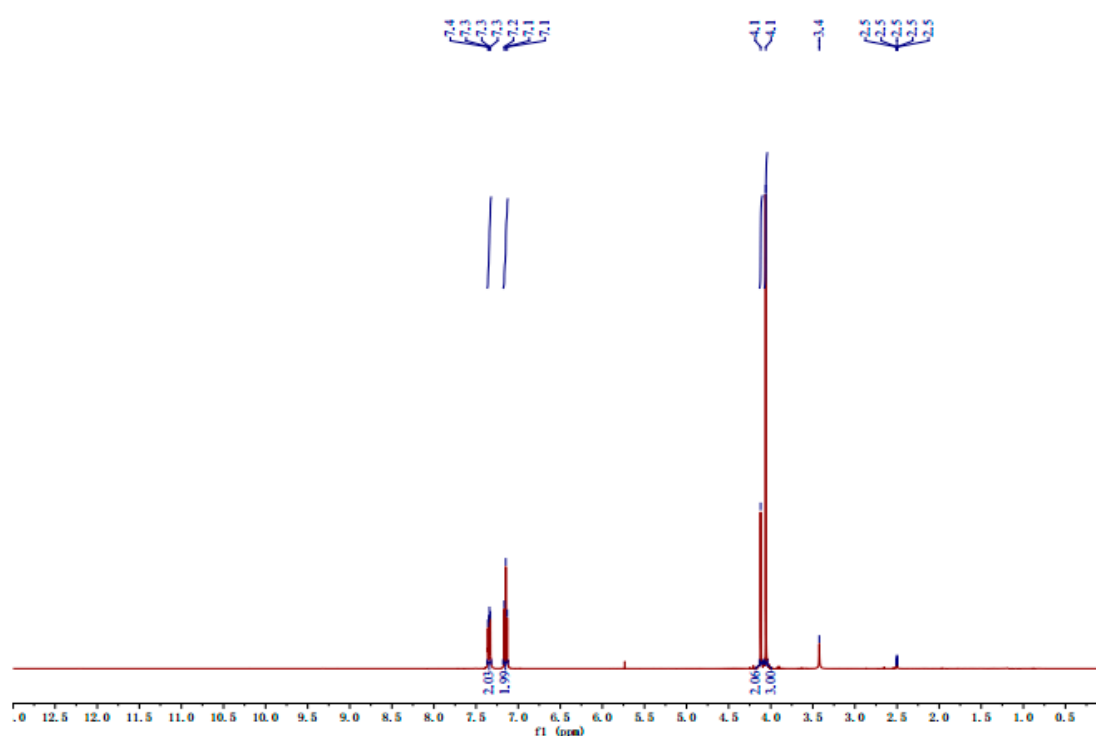


Figure S1.  $^1\text{H}$  NMR spectrum of degradation product P<sub>1</sub>.

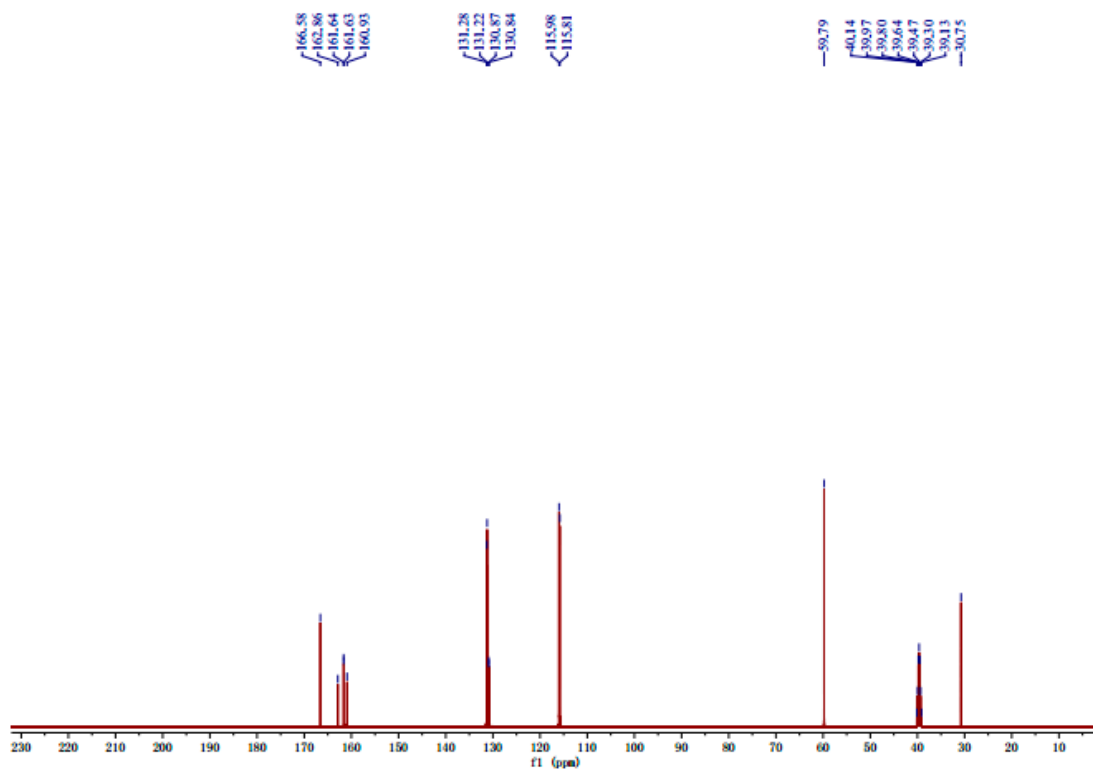


Figure S2. <sup>13</sup>C NMR spectrum of degradation product P1.

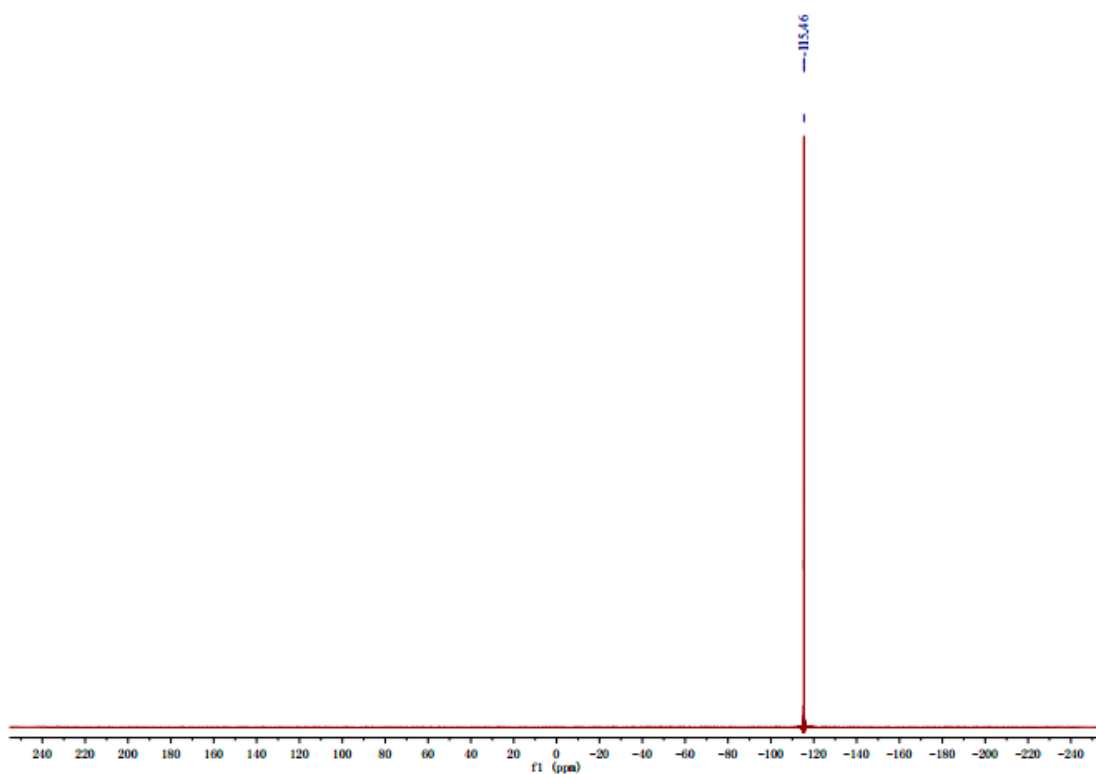
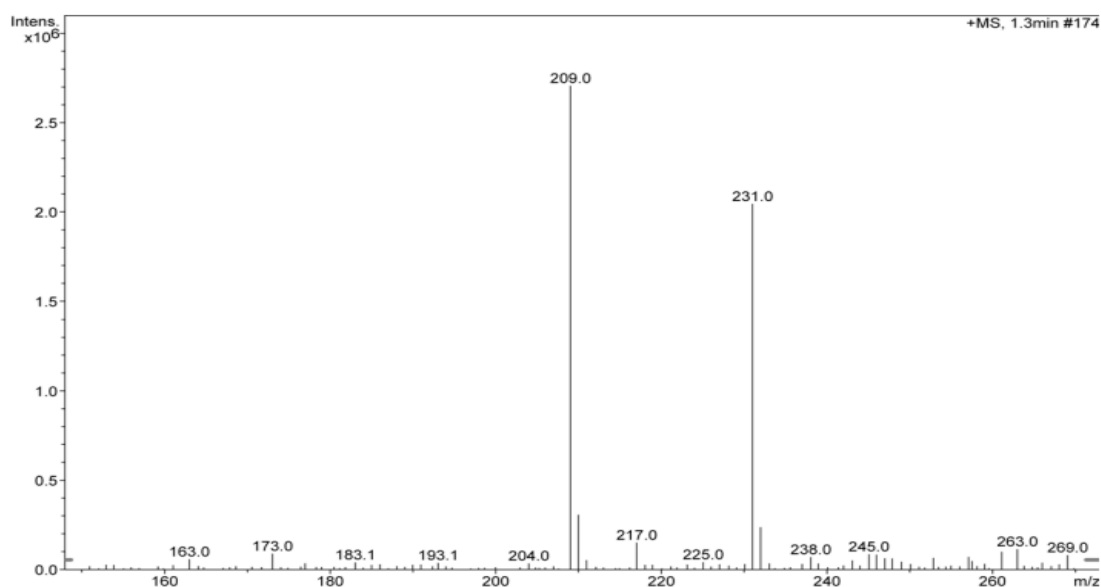


Figure S3. <sup>19</sup>F NMR spectrum of degradation product P1.



**Figure S4.** Mass Spectrum of degradation product P<sub>i</sub>.

**Table S1.** Hydrolysis parameters of FBEZF in water with different pH values.

Initial concentration (mg L <sup>-1</sup> )	Temperature (°C)	pH	Kinetic equation	K (d <sup>-1</sup> )	Half-life (d)	R <sup>2</sup>
5.0	25	5	$C_t = 5.0386e^{-0.048t}$	0.048	14.44	0.9832
5.0	25	7	$C_t = 4.9574e^{-0.432t}$	0.432	1.60	0.9972
5.0	25	9	/	/	/	/

**Table S2.** Hydrolysis parameters of FBEZF with different initial concentrations in water.

Initial concentration (mg L <sup>-1</sup> )	Temperature (°C)	pH	Kinetic equation	K (h <sup>-1</sup> )	Half-life (h)	R <sup>2</sup>
1.0	25	7	$C_t = 0.7419e^{-0.019t}$	0.019	36.48	0.9869
5.0	25	7	$C_t = 4.9754e^{-0.018t}$	0.018	38.51	0.9972
10.0	25	7	$C_t = 11.765e^{-0.022t}$	0.022	31.51	0.9979

**Table S3.** Hydrolysis parameters of FBEZF in water under different temperature.

Initial concentration (mg L <sup>-1</sup> )	Temperature (°C)	pH	Kinetic equation	K (h <sup>-1</sup> )	Half-life (h)	R <sup>2</sup>
5.0	15	7	$C_t = 4.9709e^{-0.009t}$	0.009	77.02	0.9949
5.0	25	7	$C_t = 4.9754e^{-0.018t}$	0.018	38.51	0.9972
5.0	35	7	$C_t = 4.8602e^{-0.035t}$	0.035	19.80	0.9967
5.0	45	7	$C_t = 5.5072e^{-0.231t}$	0.231	3.00	0.9948

**Table S4.** Photolysis parameters of FBEZF with different initial concentrations in water.

Initial concentration (mg L <sup>-1</sup> )	Temperature (°C)	pH	Kinetic equation	K (h <sup>-1</sup> )	Half-life (h)	R <sup>2</sup>
1.0	25	7	$C_t = 0.9655e^{-0.079t}$	0.079	8.77	0.9892
5.0	25	7	$C_t = 4.2420e^{-0.083t}$	0.061	8.35	0.9909
10.0	25	7	$C_t = 9.7266e^{-0.080t}$	0.080	8.66	0.9955