

Article

# Assessment of GHG Interactions in the Vicinity of the Municipal Waste Landfill Site—Case Study

Maciej Górka <sup>1</sup>, Yaroslav Bezyk <sup>2</sup> and Izabela Sówka <sup>2,\*</sup>

- <sup>1</sup> Faculty of Earth Science and Environmental Management, University of Wrocław, Cybulskiego 32, 50-205 Wrocław, Poland; maciej.gorka@uwr.edu.pl
- <sup>2</sup> Faculty of Environmental Engineering, Wroclaw University of Science and Technology, Plac Grunwaldzki 13, 50-377 Wroclaw, Poland; jaroslaw.bezyk@pwr.edu.pl

\* Correspondence: izabela.sowka@pwr.edu.pl

Abstract: Landfills have been identified as one of the major sources of greenhouse gas (GHG) emissions and as a contributor to climate change. Landfill facilities exhibit considerable spatial and temporal variability of both methane (CH<sub>4</sub>) and carbon dioxide (CO<sub>2</sub>) rates. The present work aimed to evaluate the spatial distribution of CH<sub>4</sub> and CO<sub>2</sub> and their  $\delta^{13}$ C isotopic composition originating from a municipal landfill site, to identify its contribution to the local GHG budget and the potential impact on the air quality of the immediate surroundings in a short-term response to environmental conditions. The objective was met by performing direct measurements of atmospheric CO<sub>2</sub> and CH<sub>4</sub> at the selected monitoring points on the surface and applying a binary mixing model for the determination of carbon isotopic ratios in the vicinity of the municipal waste landfill site. Air samples were collected and analysed for isotopic composition using flask sampling with a Picarro G2201-I Cavity Ring-Down Spectroscopy (CRDS) technique. Kriging and Inverse distance weighting (IDW) methods were used to evaluate the values at unsampled locations and to map the excess of GHGs emitted from the landfill surface. The large off-site dispersion of methane from the landfill site at a 500 m distance was identified during field measurements using isotopic data. The mean  $\delta^{13}$ C of the landfill biogas emitted to the surrounded atmosphere was  $-53.9 \pm 2.2\%$ , which corresponded well to the microbial degradation processes during acetate fermentation in the waste deposits. The calculated isotopic compositions of CO<sub>2</sub> ( $\delta^{13}$ C =  $-18.64 \pm 1.75\%$ ) indicate the domination of biogenic carbon reduction by vegetation surrounding the landfill. Finally, amounts of methane escaping into the air can be limited by the appropriate landfill management practices (faster covers active quarter through separation layer), and  $CH_4$  reduction can be achieved by sealing the cover on the leachate tank.

Keywords: GHG emissions; stable isotopes; waste management; energy recovery

# 1. Introduction

Carbon dioxide (CO<sub>2</sub>) and methane (CH<sub>4</sub>) derive from a majority of natural and anthropogenic sources and play a critical role in regulating the Earth's climate by trapping heat and contributing to overall global warming [1–3]. There are limitations of the accuracy of emissions data derived from the official inventories, especially fugitive emissions from landfills, and estimates obtained from direct atmospheric measurements. Methane and carbon dioxide are major components of landfill gas (biogas), accounting for ~45–60% and ~40–60% of the total gas emissions [4], respectively; their variations depend on the composition and age of the waste [5], as well as landfill operating procedures [6]. Biogas also includes other gases, such as volatile organic compounds (VOCs), non-methane organic compounds (NMOCs), hydrogen sulfide (H<sub>2</sub>S), and ammonia (NH<sub>3</sub>), as well as trace amounts of harmful compounds [7].

Landfill gas is produced by the degradation of organic matter in the waste mix in a series of biological and physicochemical mechanisms under anaerobic conditions [5,8,9]. The production of landfill gas consists of the following five phases: (1) hydrolysis (in the



Citation: Górka, M.; Bezyk, Y.; Sówka, I. Assessment of GHG Interactions in the Vicinity of the Municipal Waste Landfill Site—Case Study. *Energies* **2021**, *14*, 8259. https://doi.org/10.3390/en14248259

Academic Editor: Rajender Gupta

Received: 7 November 2021 Accepted: 7 December 2021 Published: 8 December 2021

**Publisher's Note:** MDPI stays neutral with regard to jurisdictional claims in published maps and institutional affiliations.



**Copyright:** © 2021 by the authors. Licensee MDPI, Basel, Switzerland. This article is an open access article distributed under the terms and conditions of the Creative Commons Attribution (CC BY) license (https:// creativecommons.org/licenses/by/ 4.0/). presence of oxygen complex organic polymers, become converted into simple compounds, sugar, fatty acids, and amino acids, further  $O_2$  becomes depleted while  $CO_2$  is generated), (2) acidogenesis (anaerobic acidic phase after the oxygen in the landfill is exhausted and anaerobic bacteria convert compounds produced by hydrolysis into acetic, lactic and formic acids and alcohols;  $CO_2$  and  $H_2$  are then generated as end products), (3) acetogenesis (the methanogenesis process starts and acids from acidogenesis are decomposed into acetate, decrease in pH; the production of  $CH_4$  increases up to a level of around 70%, and the production of  $CO_2$  decreases from 70 to 40%), (4) methanogenesis (the equilibrium is reached and acetophilic methanogens directly produce  $CH_4$  and  $CO_2$  from acetate or hydrogenophilic methanogenic bacteria using hydrogen to yield  $CH_4$  and  $CO_2$ ), and (5) maturation (most organic waste is already degraded and most of the landfill gas emissions occur, then the volume of gas generation decreases with respect to the landfill stabilization) [10–12].

Emissions of landfill gases (LFG) and the presence of contaminants in wastewater (leachate) are the two major contributors to environmental impacts in the immediate surroundings [13–15]. Moreover, inappropriate waste maintenance and landfilling constitute a serious threat to public health and the welfare of the ecosystem [16,17]. The negative environmental impact can be limited by extracting the captured biogas, which can be flared or collected and purified, and used as renewable fuel whenever environmentally and economically convenient (e.g., electricity generation or vehicle fueling). Landfill gas collection begins in the extraction wells (network composed of slotted plastic pipes) installed and operated vertically and/or horizontally inside the waste mass, depending on site-specific factors [5]. It is necessary to reduce the abrasive and corrosive nature of raw landfill gas streams by treating and further removing moisture with a moisture separator and mist eliminator, and removing particulates and other impurities through the use of filtration [5,18]. Collected LFG condensate is commonly combined with the formation of landfill leachate that requires optimal system operations to conduct and store the leachate for its further treatment or disposal [5,18].

It is still difficult to assess the scale of CO<sub>2</sub> and CH<sub>4</sub> emissions from landfill facilities due to the large temporal and spatial variations in the landfill source strength [7,19,20]. The combination of high-resolution atmospheric and precise stable isotope measurements acts as an effective tool for monitoring the strength of major CO<sub>2</sub> and CH<sub>4</sub> sources and understanding GHGs biogeochemistry [21–25]. The ranges of isotopic signatures ( $\delta^{13}$ C) of carbon dioxide and methane emitted from the major source categories are very large. With respect to carbon isotopic fractionation during methanogenesis, three types of methane sources have distinct  $\delta^{13}$ C signatures, whereby biogenic CH<sub>4</sub> is heavily depleted in <sup>13</sup>C (from –75 to –55‰) [26,27] whereas thermogenic methane typically has  $\delta^{13}$ C values (from –60 to –20‰) [28,29] and pyrogenic methane is assigned more enriched values (–50 to –40‰) [21,30]. The isotopic range of  $\delta^{13}$ C (CO<sub>2</sub>) for terrestrial vegetation varies from –29 to –12‰ [31,32] whereas the  $\delta^{13}$ C composition of fossil fuel combustion ranges from –44‰ to –22‰ [22,33].

Methane released by biogas production processes, leachate characterization, and the maturity stage of the landfill site, can also be detected by measuring the stable isotopic composition of methane  $\delta^{13}C(CH_4)$  [8,34–37]. Carbon sources at landfill sites access a narrow range of isotopic methane signatures, however, the oxidation of CH<sub>4</sub> processes alongside alteration and isotopic exchange can influence the raw CH<sub>4</sub> biogenic genesis signal [37]. According to Fischer et al. [38], isotopic signatures of methane emission from gas-collection systems in landfills located in Germany and the Netherlands were in the range of -60.3 to -57.4%, with an average of  $-57.4 \pm 2.2\%$ ; samples derived from the atmospheric upwind and downwind measurements values of  $\delta^{13}C(CH_4)$  varied within the range typical for terrestrial biogenic CH<sub>4</sub> sources (from -58.0 to -54.2%, with an average of  $-55.4 \pm 1.4\%$ ). In the study conducted in Kuwait [39], reported methane isotopic signatures for samples obtained at the vicinity of a landfill site were in a range from -59.4 to -51.9%, with an average  $\delta^{13}C(CH_4)$  value of  $-56.6 \pm 3.1\%$ . A study on the isotopic

composition of methane emissions from active landfill sites in the United Kingdom [40] reveal the isotopic signature of  $\delta^{13}$ C(CH<sub>4</sub>) to be in the range -60 to -58%. The  $\delta^{13}$ C(CH<sub>4</sub>) values measured at two Danish landfills were in the range of -59.94 and -48.45% for downwind samples and ranged between -49.43 and -46.68% for upwind gas samples from the source area [41].

An isotopic analysis is important in studies on the metabolic pathways of methane production. For biogenic methane formation from the microbial decomposition of organic matter (OM), two main methanogenic pathways can be isotopically differentiated, namely, acetate fermentation (AF) and hydrogen-utilizing CO<sub>2</sub>-reduction (CR). The third major methanogenic pathway in OM degradation is through the utilization of methyl (amines and mercaptans) and methanol, yet the generated CH<sub>4</sub> is less isotopically distinct and can resemble CH<sub>4</sub> acetate fermentation. Methanogenesis leads to CH<sub>4</sub> depletion in <sup>13</sup>C compared to the organic substrate. Typical  $\delta^{13}C(CH_4)$  values, in the range –65 to –50‰,  $\Delta^{13}C(CO_2-CH_4)$  between 30 and 55‰, and  $\delta^2H < -300\%$  are indicative of an AF pathway. The CR pathway generates more <sup>13</sup>C-depleted CH<sub>4</sub>, typically in the range –60 to –80‰,  $\Delta^{13}C(CO_2-CH_4)$  between 55 and 100‰, although less <sup>2</sup>H-depleted in the range –250 to –150‰. The reason for the large  $\Delta^{13}C(CO_2-CH_4)$  is that the fractionation during CO<sub>2</sub>-reduction (multiple enzymatic steps) is higher than in cases of acetate disproportionation to CO<sub>2</sub> and CH<sub>4</sub>, and the residual CO<sub>2</sub> becomes progressively <sup>13</sup>C-enriched, depending on how open/closed the system is [8,37,38,41–44].

There have been few studies that attempted to estimate mechanisms and the relative contribution of emissions of CH<sub>4</sub> from landfill sources [45–48]. An investigation of spatial and temporal variations of CH<sub>4</sub> fluxes from the landfill facilities with the use of interpolation methods was conducted by Haro et al. [48], Gonzalez-Valencia et al. [49] and Zhang et al. [50]. In general, landfills are known to be a source of significant fugitive methane emissions, however, there is also a large spatial variability in CH<sub>4</sub> rates from similar sources in various locations across countries. Moreover, most of the investigations reported in the literature focused on: (1) the biogeochemical transformations between  $CH_4/CO_2$  from municipal waste landfill (MWL) and dissolved inorganic carbon (DIC) in leachate, so the process is quite well known; (2) investigations of geochemical and isotopic maturation in  $CH_4/CO_2$  in MWL, although mostly in reclaimed MWL and well-drilled/exploited biogas systems. Therefore, the objective of this study was to assess the spatial distribution of GHGs (CH<sub>4</sub> and CO<sub>2</sub> and their  $\delta^{13}$ C isotopic composition) originating from the municipal waste landfill (MWL) site, and to elucidate the interactions and effects of biogenic CO<sub>2</sub> components (assimilation by neighboring vegetation) to examine the MWL effect on the local atmosphere around the dump area and short-term responses to environmental factors. The analyzed MWL is quite unique because the half area is a reclaimed quarter (with biogas drilling well system and co-generator use), whereas the second half is still an active location, to which a huge amount of municipal waste (segregated and not) is supplied, which can act as a completely uncontrolled bio-reactor and a source of a significant amount of GHGs. The research tasks included field measurements of concentrations and isotopic analyses of atmospheric  $CO_2$  and  $CH_4$  in combination with landfill-site investigations and a review of existing reported emission reports, in order to provide a comprehensive approach for distinguishing the signal and confirming the off-site migration of landfill gases in the surrounding atmosphere. The closest location to the MWL is a forest as well as an arable land area, both of which can work as a possible buffer from the MWL genesis of GHGs.

#### 2. Materials and Methods

# 2.1. Site Description

The research site (51°38′03.3″ N 16°41′56.8″ E) is located in the Rudna Wielka village in the north-central part of the Lover Silesia Voivodeship (Figure 1), Góra County, Wąsosz commune. Rudna Wielka is located 80 km north-west from Wrocław (4th biggest polish city and main supplier of waste to landfill). The climate of Lower Silesian Voivodship is characterized by variability due to the topography. The northern area is on the border of the temperate zone of the oceanic and continental climate. The climatic conditions throughout the entire area of the Góra County are practically uniform and is characterized as a temperate climate—dry and warm. Temperature fluctuations here are smaller than the average fluctuations in Poland, the springs and summers are early and warm, and winters are mild. The average annual temperature is 8 °C, which is classified as high, the average temperature in July is 18 °C, and in January it is -2 °C. The average monthly atmospheric pressure ranges from 1014.4 hPa in April to 1018.2 hPa in January. The average annual sums of atmospheric precipitation over many years (1951–1980) range from 500 mm in the west to 600 mm in the east of the county. The winds prevail from the west and their average speed at a height of 10 m above the ground is 3.0–3.5 m·s<sup>-1</sup> [51].



**Figure 1.** Location of the municipal waste landfill site (51°38′03.3″ N 16°41′56.8″ E) and 32 atmospheric air sampling points on 22 August 2017. Base map source [52].

The municipal waste landfill (MWL) in Rudna Wielka was opened in 2004. The area where the waste is stored is secured by an artificial geological sealing barrier (clays) and a synthetic geomembrane so that the leachate does not penetrate into the ground and, in the next stage, into groundwater. The leachate that arises in the landfill is discharged through a drainage system to the leachate tank. Biogas in the Rudna Wielka landfill (which is produced by the decomposition of biochemical processes) is absorbed by degassing wells and then converted into energy and heat. This process allows for the reduction of biogas emissions (including methane) to the atmosphere.

The landfill in Rudna Wielka meets all the conditions stipulated by Polish law and European Union directives. The owner of the MWL aims to minimize the negative impact on the environment. The company oversees waste collection, segregation, recovery and the management of waste that is no longer suitable for use. According to the obtained data, the most often segregated waste types (typical composition) in the Rudna Wielka municipal waste landfill site received from the Voivodeship city of Wroclaw in 2017 [53] are shown in Table 1. Based on the amount of waste collected in MWL, the main component was non-combustible mixed waste (73.95% mass), which contained unsorted (mixed) municipal waste and bulky waste. The waste composition also has a high organic/biodegradable content, which constitutes 20.53% of the mass of the waste material deposited on the landfill. The waste contained a 3.46% mass of packaging fractions, (mainly glass) and residual minerals (1.55% mass), which mainly includes other non-biodegradable waste, soil and stones, a small percentage (0.51% mass) of construction and demolition debris.

Waste Component	Classification Codes for Waste in EU	Assumed (Mg·Year <sup>-1</sup> )	% Mass in Wet Basis	
Organic, biodegradable	20 02 01	23,797.33	20.53	
Non-combustible mixed waste	20 03 01, 20 03 07	85,701.55	73.95	
Minerals	20 02 02; 20 02 03	1792.27	1.55	
Textile	20 01 11	2.63	0.00	
Mixed construction and demolition debris	17 01 07; 17 09 04	587.70	0.51	
Packaging fractions:				
- glass	15 01 07	3814.43	3.29	
- multilayer packages	15 01 06	202.60	0.17	
* Total		115,898.51	100	

**Table 1.** Amount and composition of waste in the municipal waste landfill site collected in 2017, according to the survey in [53].

\* Information on waste composition received from Wrocław urban area.

The landfill for non-hazardous and inert waste consists of three quarters with a total area of 8.07 ha. The capacity of quarters 1 and 2 is approximately 960,000 m<sup>3</sup>, while the capacity of quarter 3 is 700,000 m<sup>3</sup>. Currently, headquarters No. 3 is in operation, and plots 1 and 2 have been reclaimed (Figure 1). There is an installation for mechanical and biological waste treatment in the landfill. The mechanical part consists of a modular waste segregation station (140,000 Mg·year<sup>-1</sup>) and an installation for the mechanical treatment of waste (151,200 Mg·year<sup>-1</sup>) [54].

In the biological part, the installation allows for the replacement or parallel performance of biological drying processes (100,000 Mg·year<sup>-1</sup> for the biodegradable fraction at least 0–80 mm or 21,765 Mg·year<sup>-1</sup> for biological drying of mixed waste), biological waste processing under aerobic conditions (55,000 Mg·year<sup>-1</sup>) and waste composting in the recovery process (13,000 Mg·year<sup>-1</sup>). In addition, there was an installation for composing alternative fuels (110,000 Mg·year<sup>-1</sup> and a maximum of 20 Mg·h<sup>-1</sup>) in the landfill [54,55]. At the landfill (in the reclaimed quarters 1 and 2), there is an installation for the disposal/recovery of landfill gas (accounting for 46.85% of CH<sub>4</sub>) by burning in a flare (about 447,620 m<sup>3</sup> in 2017) or for energy use in a cogeneration system [54]. The leachate water is collected with a drainage system located at the bottom of the quarters and collected in the leachate tank, and then transported to the sewage treatment plant (approximately 10,174 m<sup>3</sup> in 2017) [54] or recirculated to the top of the landfill.

# 2.2. Sampling Methodology

The ambient air samples (Figure 1) were collected in 32 locations across the study site on 22 August 2017, in 1-L PTFE bags (SKC, Bag, Tedlar<sup>®</sup>, 1 L, Single PP Fitting) by using a vacuum pump with a miniature in-line magnesium perchlorate trap. All sites were sampled windward between 12.00–3.00 PM local time c.a. 1.5–2 m above ground level. Filled Tedlar bags were subsequently analyzed within 24 h at the laboratory for  $\delta^{13}C(CO_2)$  and  $\delta^{13}C(CH_4)$ , CO<sub>2</sub> and CH<sub>4</sub> mole fraction by CRDS laser spectroscopy (G2201-i, Picarro Inc., Santa Clara, CA, USA). The sample air during measurements had a relatively constant water vapour mole fraction, typically of around 0.5–0.8%. As the background of atmospheric methane and carbon dioxide mole fraction and as  $\delta^{13}C(CH_4)$  and  $\delta^{13}C(CO_2)$  values are not affected by local biogenic and anthropogenic processes, the data for Mace Head (MHD) Ireland NOAA station (August 2017) were used [56]. The main air masses arrive in Poland from the North Atlantic area, hence MHD station (53.3260° N, 9.899° W) was deemed to be the most appropriate as a clean background sample.

# 2.3. CO<sub>2</sub> and CH<sub>4</sub> Mole Fraction and Carbon Isotope Analysis

The CO<sub>2</sub> and CH<sub>4</sub> mole fractions and  ${}^{13}C/{}^{12}C$  isotope ratios in ambient air (transferred into Tedlar bags) were measured by cavity ring-down spectroscopy (CRDS—Picarro G2201-i isotopic analyzer). Measurements were verified using fixed working reference gas in synthetic air, with a CO<sub>2</sub> concentration of  $408 \pm 2$  ppm, a  $\delta^{13}C(CO_2)$  value around  $-30.2 \pm 0.3\%$  and  $1.88 \pm 0.04$  ppm for CH<sub>4</sub> with a  $\delta^{13}C(CH_4)$  value of  $-42.6 \pm 0.4\%$ . In order to confirm the quality of methane measurements and the long-term stability of the analyzer, three standard gases with high and low concentrations of CH<sub>4</sub> were applied. The cylinders of the certified (VPDB) CH<sub>4</sub> mole fraction and the isotopic composition ranged between (standard 1) 9.5 ppm,  $\delta^{13}C(CH_4)$ : -69.8%; (standard 2) 1.88 ppm,  $\delta^{13}C(CH_4)$ : -42.6%, (standard 3) 3.3 ppm,  $\delta^{13}C(CH_4)$ : -55.1%.

#### 2.4. Meteorological Data

Weather parameters (barometric pressure, relative humidity, temperature, wind speed and direction) were recorded during the field campaign using an on-site weather station (Kestrel 5500 Weather Meter, Kestrel Instruments, Boothwyn, PA, USA). The meteorological station was placed at the level of 1.5 m a.g.l. on a portable tripod at each ambient air sampling point.

# 2.5. Data Analysis

The statistical analysis (using Statistica 13.0 Software) was carried out on samples collected over a sampling campaign to test  $CO_2$  and  $CH_4$  mole fractions, and  $\delta^{13}C$  analysed both GHG gases as well as measured meteorological parameters. Normality tests were performed using Shapiro–Wilk estimates. Due to a lack of normality, Spearman's rank correlation coefficient was used to test possible relationships between the analysed parameters.

The Keeling plot method [57] was used to determine the isotopic composition of the atmospheric carbon dioxide and methane mixing ratio, respectively. A two-end-component mixing model, for estimating the intercept and standard error of the intercept of the Keeling plot, is represented by the carbon-isotope ( $\delta^{13}$ C) signatures as a function of the inverse (1/CO<sub>2</sub>) of atmospheric gas mixing ratios, derived from a geometric mean regression as follows (Equation (1)) [58]:

$$\delta_{\rm mix} = \frac{C_{\rm atm} \times (\delta_{\rm atm} - \delta_{\rm src})}{C_{\rm mix}} + \delta_{\rm src}, \tag{1}$$

where:

 $\delta_{mix}$ —background C isotope mixing ratio;  $\delta_{src}$ —mean source C isotope mixing ratio;  $C_{mix}$ —background atmospheric C concentration;  $C_{atm}$ —C concentrations of the mean source.

The graphical relations between measured parameters were prepared using Grapher<sup>®</sup> (from Golden Software LLC, Golden, CO, USA, www.goldensoftware.com (accessed on 27 November 2021)), whereas a spatial map distribution of analysed parameters was produced using Surfer<sup>®</sup> (from Golden Software, LLC, Golden, CO, USA, www.goldensoftware. com (accessed on 27 November 2021)). The maps of CO<sub>2</sub> mole fraction, as well as  $\delta^{13}C(CO_2)$  and  $\delta^{13}C(CH_4)$ , were extrapolated using the Kriging method, whereas the CH<sub>4</sub> mole fraction map, due to the large data differences, was extrapolated using an inverse-distance to a power method [59].

# 3. Results

#### 3.1. Weather Conditions

During the investigation on 22 August 2017, the weather conditions were quite stable with occasionally small wind velocity and direction fluctuations (Table 2 and Figure 2).

No Sample	Wind Direction [○]	Wind Velocity [m/s]	Temperature [°C]	RH [%]	Pressure [hPa]	CO <sub>2</sub> [ppm]	1/CO <sub>2</sub> [ppm <sup>-1</sup> ]	δ <sup>13</sup> C(CO <sub>2</sub> ) [‰]	CH <sub>4</sub> [ppm]	1/CH <sub>4</sub> [ppm <sup>-1</sup> ]	δ <sup>13</sup> C(CH <sub>4</sub> ) [‰]
1*	249	1.6	18.6	52.8	1005.2	394.3	2,536	-7.8	2.21	0.452	-50.9
2 *	253	3.8	18.3	55.1	1005.4	388.4	2.575	-7.4	2.02	0.495	-51.6
3 *	264	3.3	17.9	55.4	1005.7	386.6	2.587	-7.3	2.01	0.497	-52.2
4 *	220	0.0	20.2	54.2	1005.5	390.5	2.561	-7.5	1.92	0.521	-51.4
5 *	272	3.7	17.5	51.3	1005.9	385.9	2.591	-7.3	1.90	0.528	-51.5
6 *	270	5.5	18.1	53.6	1006.2	386.4	2.588	-7.3	1.90	0.528	-52.1
7*	283	5.2	17.7	53.1	1006.4	388.8	2.572	-7.5	1.90	0.527	-52.0
8 *	330	3.0	17.7	55.6	1006.2	385.1	2.596	-7.2	1.90	0.525	-51.8
9*	255	3.2	18.8	54.5	1006.4	385.5	2.594	-7.4	2.16	0.463	-53.0
10 *	252	3.9	18.2	54.6	1006.5	387.2	2.583	-7.3	2.07	0.484	-52.9
11*	345	4.4	18.4	54.0	1006.2	386.7	2.586	-7.4	1.92	0.520	-53.1
12 *	297	0.8	19.5	49.6	1006.2	385.9	2.591	-7.3	2.43	0.411	-54.1
13 *	300	6.9	17.1	50.5	1006.5	390.8	2.559	-7.4	2.12	0.4/1	-55.2
14 *	267	5.1	17.0	53.5	1006.5	380.2	2.590	-7.2	2.17	0.461	-33.8
15	240	24	10.2	54.1	1006.2	202.4	2.540	-7.5	129.40	0.008	-57.9
17 *	204	2.4	20.2	52.6	1006.2	390.1	2.540	-7.5	4 49	0.033	-56.3
18 *	305	2.2	19.1	52.8	1006.0	391.0	2.558	-74	4 71	0.223	-57.7
19 *	240	6.9	18.9	48.8	1006.2	386.7	2.586	-71	2 37	0.421	-53.8
20 *	274	41	19.3	49.6	1005.9	392.8	2.546	-7.5	1.92	0.522	-52.2
21 *	276	1.7	19.2	49.2	1005.9	390.8	2.559	-7.4	2.41	0.415	-54.4
22 *	286	2.6	19.8	50.3	1005.5	390.5	2.561	-7.3	3.03	0.331	-55.9
23 *	273	3.3	18.9	50.6	1005.5	384.9	2.598	-7.4	1.89	0.528	-52.1
24 *	273	4.0	18.5	47.9	1005.0	384.9	2.598	-7.2	2.01	0.499	-52.5
25 *	300	3.3	19.9	48.8	1005.2	389.9	2.565	-7.4	3.04	0.329	-55.7
26 *	259	5.0	18.4	47.4	1005.5	388.2	2.576	-7.3	1.90	0.525	-51.6
27 *	203	0.5	18.4	55.6	1005.2	396.4	2.523	-7.6	2.50	0.400	-54.0
28 *	248	3.5	18.7	56.2	1005.4	406.4	2.461	-8.4	11.48	0.087	-56.2
29 *	27	0.6	19.3	56.5	1005.0	417.1	2.398	-12.4	23.72	0.042	-56.2
30 *	111	1.9	17.8	52.2	1004.7	395.7	2.527	-7.5	5.63	0.178	-59.1
31 *	261	4.6	18.9	55.9	1005.0	419.0	2.387	-7.8	21.12	0.047	-57.0
32 *	277	4.7	17.7	53.7	1006.5	385.0	2.598	-7.1	2.17	0.462	-52.9
Min.	27	0.0	17.1	47.4	1004.7	384.9	2.387	-12.4	1.89	0.008	-59.1
Max.	345	6.9	20.2	56.5	1006.5	419.0	2.598	-7.1	129.48	0.528	-50.9
AVG	258	3.4	18.6	52.6	1005.8	391.4	2.556	-7.6	8.48	0.380	-53.9
Median	269	3.4	18.6	53.0	1005.9	389.3	2.569	-7.4	2.17	0.461	-53.2
SD	57	1.7	0.8	2.6	0.5	8.2	0.051	0.9	22.46	0.171	2.2

**Table 2.** Meteorological parameters and geochemical data of ambient air samples gathered in the vicinity of the municipal waste landfill site during the campaign on 22 August 2017.

\* data from [60].



Figure 2. Wind rose measured on 22 August 2017 in the vicinity of the municipal waste landfill.

Weather parameters were as follows: the air temperature varied between 17.1 and 20.2 °C (average of  $18.6 \pm 0.8$  °C), wind speed varied between 0.0 and 6.9 m·s<sup>-1</sup> (average of  $3.4 \pm 1.7$  m·s<sup>-1</sup>), wind direction fluctuated between 27 and 345° (average of  $258 \pm 57^{\circ}$  which indicated prevailing westerly winds Figure 2), relative humidity varied between 47.4 and 56.5% (average of  $52.6 \pm 2.6\%$ ) and the atmospheric pressure increased from 1004.6 to 1006.5 hPa (average of  $1005.8 \pm 0.5$  hPa). The notably weak statistical relations (Table 3) between the different meteorological parameters due to very low parameter fluctuations are accidental rather than representative of environmental dependence.

**Table 3.** Spearman's rank correlation coefficient between analyzed parameters (n = 32). Statistically significant coefficients for p < 0.05 are bolded.

	Wind Direction [°]	Wind Velocity [m/s]	Temperature [°C]	RH [%]	Pressure [hPa]	CO <sub>2</sub> [ppm]	1/CO <sub>2</sub> [ppm <sup>-1</sup> ]	δ <sup>13</sup> C(CO <sub>2</sub> ) [‰]	CH <sub>4</sub> [ppm]	1/CH <sub>4</sub> [ppm <sup>-1</sup> ]
Wind velocity [m/s]	0.27									
Temperature [°C]	0.00	-0.45								
RH [%]	-0.35	-0.19	-0.17							
Pressure [hPa]	0.38	0.42	-0.36	-0.01						
CO <sub>2</sub> [ppm]	-0.40	-0.35	0.30	0.24	-0.45					
1/CO <sub>2</sub> [ppm <sup>-1</sup> ]	0.40	0.35	-0.30	-0.24	0.45	-1.00				
$\delta^{13}C(CO_2)$ [‰]	0.29	0.39	-0.37	-0.31	0.48	-0.80	0.80			
CH <sub>4</sub> [ppm]	-0.27	-0.42	0.39	0.17	-0.25	0.67	-0.67	-0.41		
$1/CH_4$ [ppm <sup>-1</sup> ]	0.27	0.42	-0.39	-0.17	0.25	-0.67	0.67	0.41	-1.00	
δ <sup>13</sup> C(CH <sub>4</sub> ) [‰]	0.03	0.24	-0.33	-0.03	0.10	-0.52	0.52	0.28	-0.88	0.88

3.2. Mole Fraction and Stable Carbon Isotopic Composition of  $CH_4$  in the Vicinity of the Municipal Waste Landfill

The CH<sub>4</sub> mole fraction measured during the 22 August 2017 sampling investigation varied from 1.89 to 129.48 ppm with an average value of  $8.48 \pm 22.46$  ppm and a median of 2.17 ppm (Table 2). At some sampling points (e.g., No. 15), the mole fraction of ambient atmospheric CH<sub>4</sub> in the vicinity of the municipal waste landfill showed significant enrichment (up to 65 times) compared to the global methane Mace Head Ireland NOAA (1.92 ppm, August 2017) background [56]. The significant differences between the average and median values indicate that some CH<sub>4</sub> hot spots were observed rather than an enrichment of the entire analyzed area (Figure 3A).

The  $\delta^{13}C(CH_4)$  in the ambient atmosphere follows the variability with <sup>13</sup>C depletion in samples in the direction of the wind from landfill relation to samples with other wind directions. The  $\delta^{13}C(CH_4)$  values vary between -59.1 and -50.9% with an average value of  $-53.9 \pm 2.2\%$  and median of -53.2%; <sup>13</sup>C was clearly depleted when compared to the Mace Head Ireland NOAA (-47.6%, August 2017) background [56]. The carbon isotope composition clearly indicates hot spots (Figure 3B), however, similar values of the average and median suggest more mixed local air than the CH<sub>4</sub> mole fraction data. The Spearman's rank correlation coefficient between CH<sub>4</sub> mole fraction data and  $\delta^{13}C(CH_4)$  values reached -0.88 (Table 3) and indicate a significant negative relationship between both parameters.



**Figure 3.** Spatial distribution of CH<sub>4</sub> mole fraction (**A**) and the  $\delta^{13}$ C(CH<sub>4</sub>) values (**B**) in the vicinity of the municipal waste landfill sampled on 22 August 2017 campaign. Brightness rectangle covered sampling points, whereas the area outside has been extrapolated.

# 3.3. Mole Fraction and Stable Carbon Isotopic Composition of $CO_2$ in the Vicinity of the Municipal Waste Landfill

The mole fraction of ambient atmospheric CO<sub>2</sub> measured in the vicinity of the municipal waste landfill showed similar values to the global carbon dioxide Mace Head Ireland NOAA (402.5 ppm, August 2017) background [56], slightly changed due to local assimilation during the vegetation season (Table 2). The CO<sub>2</sub> mole fraction during the 22 August 2017 sampling campaign varied from 384.9 to 419.0 ppm, with an average value of 391.4  $\pm$  8.2 ppm and median value of 389.3 ppm. The lack of significant differences between the average and median as well as the similarity to the background NOAA CO<sub>2</sub> mole fraction indicate that rather moderate CO<sub>2</sub> hot spots existed in the analyzed area (Figure 4A).



**Figure 4.** Spatial distribution of CO<sub>2</sub> mole fraction (**A**) and the  $\delta^{13}$ C(CO<sub>2</sub>) values (**B**) in the vicinity of the municipal waste landfill sampled on 22 August 2017 campaign. Brightness rectangle covered sampling points, whereas the area outside has been extrapolated.

The  $\delta^{13}C(CO_2)$  in the ambient atmosphere follows the variability with <sup>13</sup>C depletion in samples in the direction of the wind in relation to the landfill, observed for samples with other wind directions. However, the range of <sup>13</sup>C depletion compared to the background values indicates a negligible but notable influence on local air. The  $\delta^{13}C(CO_2)$  values vary between -12.4 and -7.1% with an average value of  $-7.6 \pm 0.9\%$  and median of -7.4%. The carbon isotope composition clearly indicates hot spots (Figure 4B), however, similar values of average and median suggest a mixed local-air similarly to the CO<sub>2</sub> mole fraction data. Moreover, most samples indicate <sup>13</sup>C enrichment compared to with the NOAA background (-8.2%, August 2017), especially on arable areas [56]. The Spearman's rank correlation coefficient between CO<sub>2</sub> mole fraction data and  $\delta^{13}C(CO_2)$  values reaches -0.80(Table 3) and indicated significant negative relationships between both parameters.

#### 4. Discussion

# 4.1. General Divagation about GHGs on Investigated Landfill

The present work reveals the adverse impact of an active municipal waste landfill on GHGs levels in the neighboring atmosphere. Some studies also provided evidence that biogeochemical processes concerning bio-waste degradation in landfills ultimately lead to increased  $CO_2$  and  $CH_4$  emissions [4,50]. The rates of landfill-gas generation depend on the composition (organic content), age (or time since emplacement), moisture content,

particle size and compaction and methods of landfilling, climate variables, etc. [7]. In general, standard gas production rates vary from 0.0007 to 0.0080 m<sup>3</sup> per kg·year<sup>-1</sup> [61].

The average mole fraction of  $CH_4$  and  $CO_2$  reported by the automatic measurement system in the biogas installation of our investigated landfill in 2017 was 46.85% and 37.73%, respectively and the amount of biogas (methane) collected and neutralized during burning in the flare and cogeneration processes (for electricity and heat production in generator) was 447,620 m<sup>3</sup>·year<sup>-1</sup> [54]. Our investigations focused on two main gases (CH<sub>4</sub> and CO<sub>2</sub>), due to the fact that official emissions of this municipal waste landfill reported in the KOBIZE (Poland's National database on greenhouse gases and other substances emissions) database from the 2017 year are: 1073.920 Mg·year<sup>-1</sup> CH4 from landfill quarters and 9.963 Mg·year<sup>-1</sup> CO<sub>2</sub> biogas used in electricity and heat generator in landfills [62]. The present results, compared to other reported data, are quite unique, due to analyzing an MWL for which a section has been reclaimed with a well-working biogas collection system, whereas the other section is still active with a completely uncontrolled emission of GHGs to the surrounding atmosphere.

The amount of biogas generated by fugitive emissions noted in [54] was 191,837 m<sup>3</sup>·year<sup>-1</sup>. In spite of effective and environmentally friendly procedures and installations such as reclaimed quarters I and II, with a biogas collection/burning system and alternative fuel production line, etc., the negative impact of GHGs on the local atmosphere should exist. The Keeling plots revealed the contribution from landfill methane and leachate treatment methane, with an average end-member  $\delta^{13}C_{source}$  of  $-58.36 \pm 0.44\%$ (Figures 5 and 6). The intercept of CH<sub>4</sub> contributed to by landfilling was also confirmed by a highly significant linear regression model ( $\mathbb{R}^2 = 0.80$ , p < 0.05). In contrast, the obtained  $\delta^{13}C_{\text{source}}$  of carbon dioxide end-member in the mixing model (Figures 5 and 6) has an average value of  $-18.64 \pm 1.75\%$ , which is in the range of photosynthetic CO<sub>2</sub> assimilation in C3 plants. The significance of biogenic CO<sub>2</sub> sources during the vegetative season was previously reported for the Wroclaw urban area [63]. In general, the ranges of  $\delta^{13}$ C values of CO<sub>2</sub> and CH<sub>4</sub> observed in our study are comparable to those from other European landfills, with a mean  $\delta^{13}C(CO_2) = -16.8 \pm 0.4\%$ ,  $\delta^{13}C(CH_4) = -52.9 \pm 5.4\%$ . The obtained values of  $\delta^{13}$ C are consistent with other isotopic studies and indicate the main pathway of CH<sub>4</sub> production in the MWL, resulting from acetate fermentation [37,38,44].



**Figure 5.** Relations between mole fraction and carbon isotopes composition of methane (**A**) and carbon dioxide (**B**) from ambient air gathered in the vicinity of municipal waste landfill sampled during the 22 August 2017 campaign. Dashed lines show the linear regression fit for the data sampled. The green dots represent average background values according Mace Head Research Station, Ireland, NOAA. Black solid/dashed lines represent expected source contribution range.



Reciprocal of CO<sub>2</sub> mole fraction [ppm<sup>-1</sup>·1000]

**Figure 6.** Relations between reciprocal of mole fraction and carbon isotopes composition of methane (**A**) and carbon dioxide (**B**) from ambient air gathered in the vicinity of municipal waste landfill sampled on 22 August 2017 campaign. Dashed lines show the linear regression fit of the data sampled. The green dots represent average background values according to Mace Head Research Station, Ireland, NOAA.

The largest sources of methane emissions from the analyzed landfill site were: (I) active landfill quarter III; (II) landfill leachate tank; (III) biogas collecting well. According to carbon dioxide, the most probable hotspots (Figure 1) were (I) the active landfill quarter III; (II) the biogas heat and power generator. The highest mole fractions of  $CH_4$  and  $CO_2$  were expected in the nearest eastward points from the landfill (Figure 1) due to the

dominant wind existing westward (Figure 2) during the sampling campaign. Finally, the field sampling campaign revealed that substantial amounts of methane escaped from the leachate tank and quarter No. 3 (in operation) in the MWL. The plume of the gases released from the mentioned sources towards the east direction had a potential negative impact on the air quality of the immediate surroundings. The enhanced methane–mole fraction and strongly diluted  $\delta^{13}C(CH_4)$  in ambient air were measured at a few points on the east side of the landfill (e.g., point No. 15—129.48 ppm; point No. 16—18.85 ppm; point No. 28—11.48 ppm) (Table 2, Figure 3).

### 4.2. CH<sub>4</sub> Balance in the Vicinity of Municipal Waste Landfill

During the sampling campaign, we observed notable CH<sub>4</sub> mole fractions not only in the nearest landfill points (Figures 1 and 3A) but also at a distance of 300–500 m eastward/southeastward. The spatial distribution (Figure 3A) found a high methane–mole fraction in the air, even 25 times above the background (e.g., No. 15 allocated on the leeward side), compared to other analyzed points in the vicinity of landfill. Moreover, samples with higher CH<sub>4</sub> mole fraction reported notable depletion of <sup>13</sup>C, confirmed by the negative –0.88 coefficient of Spearman's rank (Table 3). Graphical representation (Figure 5A) allowed us to hypothesize that: (i) the carbon isotope composition of many samples indicate biogenic landfill methane according to Levin et al. [30] data; (ii) two main CH<sub>4</sub> hotspot sources probably exist, connected with landfill methane derived from active quarter III as well as generated in the open leachate tank (Figures 1 and 3A,B). Landfill methane [30,45,64] has a similar range of  $\delta^{13}$ C(CH<sub>4</sub>) values (between –60 and –50‰) for methane generated from leachates [65] and from sewage [24].

The Keeling plot (Figure 6A) clearly indicates a biogenic origin (landfill + leachates) of methane with an intercept  $\delta^{13}C(CH_4)$  value  $-58.36 \pm 0.44\%$  as a dominant CH<sub>4</sub> source in the analyzed area. Biogenic CO<sub>2</sub> derived from decomposition processes in landfills/or soil/root respiration, as well as from anthropogenic CO<sub>2</sub> from biogas/fossil fuels burning was only connected with sample No. 29. In spite of the lack of  $\delta D(CH_4)$  values, according to the C/H isotopes plot in Coleman et al. [66], the range of our  $\delta^{13}C(CH_4)$  values indicate that the acetate-fermentation pathway dominates the CO<sub>2</sub>-reduction pathway. The  $\delta^{13}C(CO_2)$  values did not show a genetic relation with the CO<sub>2</sub>-reduction pathway, both at its preliminary stage (when <sup>13</sup>C depleted CO<sub>2</sub> exists), as well as on its final stage, where it appears to be mostly reduced and enriched in <sup>13</sup>C(CO<sub>2</sub>).

We did not discount the fact that intensive  $CO_2/CH_4$  exchange can exist in covered and reclaimed landfill part in quarters I and II, however, almost the entire biogas from these quarters was found to be captured in wells and burned in a generator. Hence, methane observed in the vicinity is probably generated mostly in situ from "fresh" municipal waste and from landfill leachates. The lack or the very negligible increase of the  $CH_4$  mole fraction, as well as the fact that it was generally not isotopically different from the background in north and northeast sampling points (near reclaimed fragment of landfill) confirmed our hypothesis. The rose-type diagram for the  $CH_4$  mole fraction (Figure 7A) and  $\delta^{13}C(CH_4)$ values (Figure 7B) revealed the dominant wind direction and velocity to be factors that controlled methane spread in the local atmosphere, in spite of pure or a lack of official statistical relations (Table 3).



**Figure 7.** The circular distribution of directional data of CH<sub>4</sub> mole fraction (**A**) and  $\delta^{13}$ C(CH<sub>4</sub>) values (**B**) from ambient air gathered in the vicinity of municipal waste landfill sampled during 22 August 2017 campaign.

# 4.3. CO<sub>2</sub> Balance in the Vicinity of Municipal Waste Landfill

Surprisingly, for a given study period, the above-atmospheric levels of CO<sub>2</sub> (>390 ppm) were possible to observe only in the nearest landfill points (No. 27–31 Figures 1 and 4A). The other sampling points reached similar or slightly lower (up to 5 ppm) CO<sub>2</sub> mole fractions compared to the background level [56]. Similarly to the CO<sub>2</sub> mole fraction, the  $\delta^{13}C(CO_2)$  values of most air samples collected further from landfills showed a slight <sup>13</sup>C enrichment compared to the background, and only the nearest (No. 28 and 29) points indicate a notable depletion in <sup>13</sup>C (up to 4.4‰). The relationship between the CO<sub>2</sub> mole fraction and  $\delta^{13}C(CO_2)$  values (Figure 5B) clearly indicated a biogenic input of the surrounding landfill agriculture fields and grassland. An assimilation process of CO<sub>2</sub> by vegetation or by soil microorganisms [67–69] caused a lower CO<sub>2</sub> mole fraction of residue carbon dioxide in ambient air, as well as the enrichment of <sup>13</sup>C due to preferential <sup>12</sup>C consumption.

The slightly marked CO<sub>2</sub> hotspot near generator burned biogas (methane) (Figure 1) and leachate tank influence the local landfill atmosphere (Table 2 and Figure 4A,B). As a result, the very intensive aerobic bacterial reactions on active quarter III are not spatially observed in  $CO_2/\delta^{13}C(CO_2)$  values plum. A total of 447,600 m<sup>3</sup> of biogas was burned in the flare and generator (0.294 MW of electric power) in 2017. However, a  $CO_2$  signal was not detected, probably due to the hot exhaust gases being conventionally raised and mixed in the upper layers of the atmosphere, although not affecting the local atmosphere.

Many authors noted extremely enriched in  $\delta^{13}C(CO_2)$  values, reaching up to +20‰ for landfill CO<sub>2</sub> and CO<sub>2</sub>/DIC in landfill leachates [8,45,64]. Therefore, in the case of the existing bacterial CO<sub>2</sub> reduction process, extremely <sup>13</sup>C enriched residual ambient CO<sub>2</sub> in landfill and neighboring sampling points have been expected. Instead of this, the slightly depleted in <sup>13</sup>C(CO<sub>2</sub>), observed in the vicinity of the landfill, was probably connected with exhaust gasses from the exploitation of heavy machinery (garbage trucks, landfill compactors) and from the biogas generator (marked on Figure 5B as landfill CO<sub>2</sub>). Conversely, the oxidation of biogas methane in landfills results in a significant depletion in <sup>13</sup>C(CO<sub>2</sub>) in the landfill atmosphere [66]. However, in this situation, both the CO<sub>2</sub> mole fraction, as well as the carbon isotopic signal of carbon dioxide have not been observed in the analysed landfill.

We hypothesize that intensive CO<sub>2</sub> reduction on this landfill does not occur, especially in active open quarter III, and only local grassland/agriculture fields were able to buffer

and assimilate CO<sub>2</sub> that potentially derived from the landfill. This thesis is confirmed by the positive ( $\rho = 0.80$ ) coefficients of Spearman's rank between CO<sub>2</sub> and  $\delta^{13}C(CO_2)$ and negative ( $\rho = -0.41$ ) between the CH<sub>4</sub> mole fraction and  $\delta^{13}C(CO_2)$  (Table 3). The Keeling plot (Figure 6B) for n = 31 points' mixing ratio source ( $\delta^{13}C_{source} = -18.64 \pm 1.75\%$ ) indicates assimilation by vegetation surrounding the landfill, rather than (i) biogenic CO<sub>2</sub> derived from decomposition processes on landfills/or soil/root respiration, as well as (ii) anthropogenic CO<sub>2</sub> from biogas/fossil fuels burning (only connected with sample No. 29).

#### 5. Conclusions

This study presented the spatial characteristic of atmospheric CO<sub>2</sub> and CH<sub>4</sub> levels in short-term responses to environmental conditions observed in the vicinity of the municipal waste landfill for our August 2017 sampling campaign. Our investigation confirmed the negative influence of the municipal waste landfill site, especially at the hot-spot zones, on the local GHG balance in the surrounding atmosphere, but less than assumed. We observe a negligible CO<sub>2</sub> mole fraction difference in relation to that of the background, as well as its carbon isotopes signal in samples at a distance of 100 m from the landfill. A very significant CO<sub>2</sub> assimilation process in the surrounded agriculture/grassland area completely compensated for the possible negative CO<sub>2</sub> input. Most of the ambient CO<sub>2</sub> samples collected around the landfill had prevailing background CO<sub>2</sub> levels (varied from 384.9 to 419.0 ppm), with an average value of  $391.4 \pm 8.2$  ppm, which were calculated by the binary mixing model isotopic signature and had a mean source value of  $-18.64 \pm 1.75\%$ .

The obtained results suggest that the isotopic content of methane in the immediate surroundings (average value of  $\delta^{13}C = -53.9 \pm 2.2\%$ ) is strongly influenced by microbial origins (acetate -fermentation pathway) in the studied landfill site. Methane contribution was likely connected with two of the landfill CH<sub>4</sub> hotspots, including (1) active quarter III and (2) the leachate system/tank. The highest excess of CH<sub>4</sub> and notable methane depletion in <sup>13</sup>C was observed c.a. 300 m eastward/northeastward from the landfill. The CH<sub>4</sub> mole fraction in the emitted gases close to the hotspots reached up to 129.48 ppm, with an average  $\delta^{13}C$  value of  $-57.3 \pm 1.0\%$ . The reclaimed quarters I and II showed negligible methane emission to the local atmosphere, due to the existing well-working biogas collection and processing system (with heat and power co-generator).

Biogas emissions showed a good correlation with air temperature and wind velocity, however, no significant relationships between relative humidity, atmospheric pressure and methane levels were observed. The effect of moderate to strong westward wind (average of  $3.4 \pm 1.7 \text{ m} \cdot \text{s}^{-1}$ ) on the dispersion of landfill origin GHG gases (CO<sub>2</sub> and CH<sub>4</sub>) in the local atmosphere dominated up to 500 m from the landfill. The dispersion of GHG emissions from landfill areas should be further investigated using in situ surveys, as their contribution to the local carbon budget vary significantly in different spatial and temporal settings. Finally, to maximize the potential of CH<sub>4</sub> mitigation, emissions from landfill sites should be reduced by utilizing waste composition as a source. Therefore, further research should focus on assessing the level of biogas production from active quarter III and the leachate tank, using chamber methods and experimental well drilling, followed by dispersion modeling, which should help to minimize the negative impact of GHGs derived from MWLs on the environment.

**Author Contributions:** Supervision, writing original draft, investigation, methodology, visualization: M.G.; writing original draft, methodology, investigation: Y.B.; writing, review and editing, funding acquisition: I.S. All authors have read and agreed to the published version of the manuscript.

Funding: This research received no external funding.

Institutional Review Board Statement: Not applicable.

Informed Consent Statement: Not applicable.

Data Availability Statement: Not applicable.

**Acknowledgments:** The authors thank Nikola Nowak for assistance with field sample collection. We would like to thank Dariusz Strąpoć for his helpful advice on principal methane production processes during the preparation of this manuscript. We are also grateful to the National Center for Emissions Management, Poland (KOBiZE) for access to emission data for particular source categories in the selected years.

**Conflicts of Interest:** The authors declare no conflict of interest. The funders had no role in the design of the study; in the collection, analyses, or interpretation of data; in the writing of the manuscript, or in the decision to publish the results.

# References

- 1. Friedlingstein, P.; O'Sullivan, M.; Jones, M.W.; Andrew, R.M.; Hauck, J.; Olsen, A.; Peters, G.P.; Peters, W.; Pongratz, J.; Sitch, S.; et al. Global Carbon Budget 2020. *Earth Syst. Sci. Data* 2020, *12*, 3269–3340. [CrossRef]
- Smith, K.R.; Desai, M.A.; Rogers, J.V.; Houghton, R.A. Joint CO<sub>2</sub> and CH<sub>4</sub> accountability for global warming. *Proc. Natl. Acad. Sci.* USA 2013, 110, E2865–E2874. [CrossRef] [PubMed]
- 3. Saunois, M.; Stavert, A.R.; Poulter, B.; Bousquet, P.; Canadell, J.G.; Jackson, R.B.; Raymond, P.A.; Dlugokencky, E.J.; Houweling, S.; Patra, P.K.; et al. The Global Methane Budget 2000–2017. *Earth Syst. Sci. Data* 2020, *12*, 1561–1623. [CrossRef]
- Pehme, K.-M.; Orupõld, K.; Kuusemets, V.; Tamm, O.; Jani, Y.; Tamm, T.; Kriipsalu, M. Field Study on the Efficiency of a Methane Degradation Layer Composed of Fine Fraction Soil from Landfill Mining. Sustainability 2020, 12, 6209. [CrossRef]
- 5. Speight, J. Natural Gas; Elsevier: Amsterdam, The Netherlands, 2019; ISBN 9780128095706.
- 6. Yang, L.; Chen, Z.; Zhang, X.; Liu, Y.; Xie, Y. Comparison study of landfill gas emissions from subtropical landfill with various phases: A case study in Wuhan, China. *J. Air Waste Manag. Assoc.* **2015**, *65*, 980–986. [CrossRef]
- 7. Robertson, T.; Dunbar, J. Guidance for evaluating landfill gas emissions from closed or abandoned facilities. In *EPA Report*; United States Environmental Protection Agency: Washington, DC, USA, 2005.
- 8. de Medeiros Engelmann, P.; dos Santos, V.H.J.M.; Barbieri, C.B.; Augustin, A.H.; Ketzer, J.M.M.; Rodrigues, L.F. Environmental monitoring of a landfill area through the application of carbon stable isotopes, chemical parameters and multivariate analysis. *Waste Manag.* **2018**, *76*, 591–605. [CrossRef]
- 9. Kjeldsen, P.; Barlaz, M.A.; Rooker, A.P.; Baun, A.; Ledin, A.; Christensen, T.H. Present and Long-Term Composition of MSW Landfill Leachate: A Review. *Crit. Rev. Environ. Sci. Technol.* 2002, 32, 297–336. [CrossRef]
- 10. Białowiec, A. Some Aspects of Environmental Impact of Waste Dumps. In *Contemporary Problems of Management and Environmental Protection*; University of Warmia and Mazury in Olsztyn: Olsztyn, Poland, 2011; ISBN 978-83-929462-7-4.
- Khalil, D.M.J.; Gupta, R.; Sharma, K. Microbiological Degradation of Municipal Solid Waste in Landfills for LFG Generation. In National Conference on Synergetic Trends in engineering and Technology (STET-2014) International Journal of Engineering and Technical Research; ISSN 2321-0869. Special Issue; Available online: https://www.erpublication.org/published\_paper/IJETR\_APRIL\_2014 \_STET\_03.pdf (accessed on 2 November 2021).
- 12. Njoku, P.O.; Odiyo, J.O.; Durowoju, O.S.; Edokpayi, J.N. A Review of Landfill Gas Generation and Utilisation in Africa. *Open Environ. Sci.* **2018**, *10*, 1–15. [CrossRef]
- 13. Vaverková, M.D. Landfill Impacts on the Environment—Review. Geosciences 2019, 9, 431. [CrossRef]
- 14. Brennan, R.B.; Healy, M.G.; Morrison, L.; Hynes, S.; Norton, D.; Clifford, E. Management of landfill leachate: The legacy of European Union Directives. *Waste Manag.* **2016**, *55*, 355–363. [CrossRef] [PubMed]
- 15. Ozbay, G.; Jones, M.; Gadde, M.; Isah, S.; Attarwala, T. Design and Operation of Effective Landfills with Minimal Effects on the Environment and Human Health. *J. Environ. Public Health* **2021**, 2021, 1–13. [CrossRef] [PubMed]
- 16. Toro, R.; Morales, L. Landfill fire and airborne aerosols in a large city: Lessons learned and future needs. *Air Qual. Atmos. Health* **2018**, *11*, 111–121. [CrossRef]
- 17. Rao, M.N.; Sultana, R.; Kota, S. Solid and Hazardous Waste Management; Elsevier: Amsterdam, The Netherlands, 2016; ISBN 9780128098769.
- 18. U.S. Environmental Protection Agency. *International Best Practices Guide for Landfill Gas Energy Projects*; United States Environmental Protection Agency: Washington, DC, USA, 2012.
- 19. Xu, L.; Lin, X.; Amen, J.; Welding, K.; McDermitt, D. Impact of changes in barometric pressure on landfill methane emission. *Glob. Biogeochem. Cycles* **2014**, *28*, 679–695. [CrossRef]
- 20. Goldsmith, C.D.; Chanton, J.; Abichou, T.; Swan, N.; Green, R.; Hater, G. Methane emissions from 20 landfills across the United States using vertical radial plume mapping. *J. Air Waste Manag. Assoc.* **2012**, *62*, 183–197. [CrossRef]
- 21. Menoud, M.; van der Veen, C.; Necki, J.; Bartyzel, J.; Szénási, B.; Stanisavljević, M.; Pison, I.; Bousquet, P.; Röckmann, T. Methane (CH<sub>4</sub>) sources in Krakow, Poland: Insights from isotope analysis. *Atmos. Chem. Phys.* **2021**, *21*, 13167–13185. [CrossRef]
- Zimnoch, M.; Necki, J.; Chmura, L.; Jasek, A.; Jelen, D.; Galkowski, M.; Kuc, T.; Gorczyca, Z.; Bartyzel, J.; Rozanski, K. Quantification of carbon dioxide and methane emissions in urban areas: Source apportionment based on atmospheric observations. *Mitig. Adapt. Strateg. Glob. Chang.* 2019, 24, 1051–1071. [CrossRef]
- 23. Pang, J.; Wen, X.; Sun, X. Mixing ratio and carbon isotopic composition investigation of atmospheric CO<sub>2</sub> in Beijing, China. *Sci. Total Environ.* **2016**, *539*, 322–330. [CrossRef] [PubMed]

- 24. Górka, M.; Lewicka-Szczebak, D.; Fuß, R.; Jakubiak, M.; Jędrysek, M.O. Dynamics and origin of atmospheric CH4 in a Polish metropolitan area characterized by wetlands. *Appl. Geochem.* **2014**, *45*, 72–81. [CrossRef]
- Stolper, D.A.; Martini, A.M.; Clog, M.; Douglas, P.M.; Shusta, S.S.; Valentine, D.L.; Sessions, A.L.; Eiler, J.M. Distinguishing and understanding thermogenic and biogenic sources of methane using multiply substituted isotopologues. *Geochim. Et Cosmochim. Acta* 2015, 161, 219–247. [CrossRef]
- Douglas, P.M.J.; Stolper, D.A.; Eiler, J.M.; Sessions, A.L.; Lawson, M.; Shuai, Y.; Bishop, A.; Podlaha, O.G.; Ferreira, A.A.; Santos Neto, E.V.; et al. Methane clumped isotopes: Progress and potential for a new isotopic tracer. Org. Geochem. 2017, 113, 262–282. [CrossRef]
- Schaefer, H.; Fletcher, S.E.M.; Veidt, C.; Lassey, K.R.; Brailsford, G.W.; Bromley, T.M.; Dlugokencky, E.J.; Michel, S.E.; Miller, J.B.; Levin, I.; et al. A 21st-century shift from fossil-fuel to biogenic methane emissions indicated by 13CH4. *Science* 2016, 352, 80–84. [CrossRef]
- Zazzeri, G.; Lowry, D.; Fisher, R.E.; France, J.L.; Lanoisellé, M.; Kelly, B.F.J.; Necki, J.M.; Iverach, C.P.; Ginty, E.; Zimnoch, M.; et al. Carbon isotopic signature of coal-derived methane emissions to the atmosphere: From coalification to alteration. *Atmos. Chem. Phys.* 2016, *16*, 13669–13680. [CrossRef]
- Tait, D.R.; Maher, D.T.; Santos, I.R. Seasonal and Diurnal Dynamics of Atmospheric Radon, Carbon Dioxide, Methane, δ13C-CO<sub>2</sub> and δ13C-CH<sub>4</sub> in a Proposed Australian Coal Seam Gas Field. *Water Air Soil Pollut.* 2015, 226, 350. [CrossRef]
- Levin, I.; Bergamaschi, P.; Dörr, H.; Trapp, D. Stable isotopic signature of methane from major sources in Germany. *Chemosphere* 1993, 26, 161–177. [CrossRef]
- Hare, V.J.; Loftus, E.; Jeffrey, A.; Ramsey, C.B. Atmospheric CO<sub>2</sub> effect on stable carbon isotope composition of terrestrial fossil archives. *Nat. Commun.* 2018, 9, 252. [CrossRef] [PubMed]
- 32. Jasek, A.; Zimnoch, M.; Gorczyca, Z.; Smula, E.; Rozanski, K. Seasonal variability of soil CO<sub>2</sub> flux and its carbon isotope composition in Krakow urban area, Southern Poland. *Isot. Environ. Health Stud.* **2014**, *50*, 143–155. [CrossRef] [PubMed]
- Graven, H.; Keeling, R.F.; Rogelj, J. Changes to Carbon Isotopes in Atmospheric CO<sub>2</sub> Over the Industrial Era and Into the Future. *Glob. Biogeochem. Cycles* 2020, 34, e2019GB006170. [CrossRef]
- 34. North, J.C.; Frew, R.D.; Van Hale, R. Can stable isotopes be used to monitor landfill leachate impact on surface waters? *J. Geochem. Explor.* **2006**, *88*, 49–53. [CrossRef]
- 35. Raco, B.; Dotsika, E.; Battaglini, R.; Bulleri, E.; Doveri, M.; Papakostantinou, K. A Quick and Reliable Method to Detect and Quantify Contamination from MSW Landfills: A Case Study. *Water Air Soil Pollut.* **2013**, 224, 1380. [CrossRef]
- 36. Wimmer, B.; Hrad, M.; Huber-Humer, M.; Watzinger, A.; Wyhlidal, S.; Reichenauer, T.G. Stable isotope signatures for characterising the biological stability of landfilled municipal solid waste. *Waste Manag.* **2013**, *33*, 2083–2090. [CrossRef] [PubMed]
- Widory, D.; Proust, E.; Bellenfant, G.; Bour, O. Assessing methane oxidation under landfill covers and its contribution to the above atmospheric CO<sub>2</sub> levels: The added value of the isotope (δ13C and δ18O CO<sub>2</sub>; δ13C and δD CH<sub>4</sub>) approach. *Waste Manag.* 2012, 32, 1685–1692. [CrossRef] [PubMed]
- Bergamaschi, P.; Lubina, C.; Königstedt, R.; Fischer, H.; Veltkamp, A.C.; Zwaagstra, O. Stable isotopic signatures (•13C, gD) of methane from European landfill sites. J. Geophys. Res. 1998, 103, 8251–8265. [CrossRef]
- 39. Al-Shalan, A.; Lowry, D.; Fisher, R.E.; Nisbet, E.G.; Zazzeri, G.; Al-Sarawi, M.; France, J.L. Methane emissions in Kuwait: Plume identification, isotopic characterisation and inventory verification. *Atmos. Environ.* **2021**, *268*, 118763. [CrossRef]
- 40. Bakkaloglu, S.; Lowry, D.; Fisher, R.; France, J.; Lanoiselle, M.; Fernandez, J. Characterization and Quantification of Methane Emissions from Waste in the UK. In Proceedings of the EGU General Assembly 2020, Online, 4–8 May 2020. [CrossRef]
- Aghdam, E.F.; Fredenslund, A.M.; Chanton, J.; Kjeldsen, P.; Scheutz, C. Determination of gas recovery efficiency at two Danish landfills by performing downwind methane measurements and stable carbon isotopic analysis. *Waste Manag.* 2018, 73, 220–229. [CrossRef] [PubMed]
- 42. Sanci, R.; Panarello, H.O. Carbon stable isotopes as indicators of the origin and evolution of CO<sub>2</sub> and CH<sub>4</sub> in urban solid waste disposal sites and nearby areas. *Environ. Earth Sci.* **2016**, *75*, 294. [CrossRef]
- 43. Whiticar, M.J. Stable isotope geochemistry of coals, humic kerogens and related natural gases. *Int. J. Coal Geol.* **1996**, *32*, 191–215. [CrossRef]
- 44. Whiticar, M.; Faber, E.; Schoell, M. Biogenic methane formation in marine and freshwater environments: CO<sub>2</sub> reduction vs. acetate fermentation—Isotope evidence. *Geochim. Et Cosmochim. Acta* **1986**, *50*, 693–709. [CrossRef]
- 45. Baldassare, F.J.; Laughrey, C.D. Identifying the sources of stray methane by using geochemical and isotopic fingerprinting. *Environ. Geosci.* **1997**, *4*, 85–94.
- 46. Tecle, D.; Lee, J.; Hasan, S. Quantitative analysis of physical and geotechnical factors affecting methane emission in municipal solid waste landfill. *Environ. Geol.* **2009**, *56*, 1135–1143. [CrossRef]
- 47. Zhang, C.; Xu, T.; Feng, H.; Chen, S. Greenhouse Gas Emissions from Landfills: A Review and Bibliometric Analysis. *Sustainability* **2019**, *11*, 2282. [CrossRef]
- 48. Haro, K.; Ouarma, I.; Nana, B.; Bere, A.; Tubreoumya, G.C.; Kam, S.Z.; Laville, P.; Loubet, B.; Koulidiati, J. Assessment of CH<sub>4</sub> and CO<sub>2</sub> surface emissions from Polesgo's landfill (Ouagadougou, Burkina Faso) based on static chamber method. *Adv. Clim. Chang. Res.* 2019, 10, 181–191. [CrossRef]

- Gonzalez-Valencia, R.; Magana-Rodriguez, F.; Martinez-Cruz, K.; Fochesatto, G.J.; Thalasso, F. Spatial and temporal distribution of methane emissions from a covered landfill equipped with a gas recollection system. *Waste Manag.* 2021, 121, 373–382. [CrossRef] [PubMed]
- 50. Zhang, C.; Guo, Y.; Wang, X.; Chen, S. Temporal and spatial variation of greenhouse gas emissions from a limited-controlled landfill site. *Environ. Int.* 2019, 127, 387–394. [CrossRef]
- 51. Climate-Data.org. Available online: https://en.climate-data.org/europe/poland/lower-silesian-voivodeship-456/ (accessed on 9 October 2021).
- 52. OpenStreetMap. Available online: OpenStreetMap.org (accessed on 11 November 2020).
- 53. Information on Individual Types of Municipal Waste Collected from the Wroclaw Urban Area. Available online: https://www. google.com/url?sa=t&rct=j&q=&esrc=s&source=web&cd=&ved=2ahUKEwjXt-vzwpr0AhUEt4sKHdw9CDcQFnoECAkQAQ& url=https%3A%2F%2Fbip.um.wroc.pl%2Fattachments%2Fdownload%2F57300&usg=AOvVaw3O4-ZvhtKb5GEwKITRZ1yG (accessed on 2 November 2021).
- 54. GIOŚ. Chief Inspectorate of Environmental Protection (GIOS, Poland). Inspection Report no. WIOŚ-Wroc 301/2018; GIOŚ: Wrocław, Poland, 2018.
- 55. Chybiński, S.; Gaworecka, M.; Olearnik, M.; Krzyśków, A. Analysis of the State of Municipal Precipitation Management in the Kobierzyce Commune; Gmina Kobierzyce: Wrocław, Poland, 2015.
- NOAA. Carbon Cycle Gases Mace Head, County Galway, Ireland. Available online: https://gml.noaa.gov/dv/iadv/graph.php? code=MHD&program=ccgg&type=lg (accessed on 28 April 2021).
- 57. Keeling, C.D. The concentration and isotopic abundances of atmospheric carbon dioxide in rural areas. *Geochim. Et Cosmochim. Acta* **1958**, *13*, 322–334. [CrossRef]
- 58. Pataki, D.E. Seasonal cycle of carbon dioxide and its isotopic composition in an urban atmosphere: Anthropogenic and biogenic effects. *J. Geophys. Res.* 2003, 108, 4735. [CrossRef]
- 59. Bezyk, Y.; Sówka, I.; Górka, M.; Blachowski, J. GIS-Based Approach to Spatio-Temporal Interpolation of Atmospheric CO<sub>2</sub> Concentrations in Limited Monitoring Dataset. *Atmosphere* **2021**, *12*, 384. [CrossRef]
- 60. Nowak, N. Assessment of Greenhouse Gas Emissions from the Municipal Waste Landfill in Rudna Wielka on the Basis of Concentrations and Carbon Isotopic Composition of Atmospheric CO<sub>2</sub> and CH<sub>4</sub>. Master's thesis, University of Wroclaw, Wroclaw, Poland, 2018.
- 61. Ezekwe, I.C.; Arokoyu, S.B. Landfill Emissions and their Urban Planning and Environmental Health Implications in Port Harcourt, South-South Nigeria. *Desenvolv. E Meio Ambiente* **2017**, *42*, 42–224. [CrossRef]
- 62. KOBiZE. Poland's National Database on Greenhouse Gases and other Substances Emissions. Data on the Waste Management Plant. Installations and Sources, Emission Levels of Selected Substances for 2016–2020; IOŚ-PIB: Warsaw, Poland, 2021.
- Bezyk, Y.; Sówka, I.; Górka, M. Assessment of urban CO<sub>2</sub> budget: Anthropogenic and biogenic inputs. Urban Clim. 2021, 39, 100949. [CrossRef]
- Hackley, K.C.; Liu, C.L.; Coleman, D.D. Environmental Isotope Characteristics of Landfill Leachates and Gases. *Ground Water* 1996, 34, 827–836. [CrossRef]
- 65. Bréas, O.; Guillou, C.; Reniero, F.; Wada, E. The Global Methane Cycle: Isotopes and Mixing Ratios, Sources and Sinks. *Isot. Environ. Health Stud.* **2001**, *37*, 257–379. [CrossRef] [PubMed]
- Coleman, D.D.; Liu, C.L.; Hackley, K.C.; Pelphrey, S.R. Isotopic Identification of Landfill Methane. AAPG Div. Environ. Geosci. J. 1995, 2, 95–103.
- 67. Schubert, B.A.; Jahren, A.H. The effect of atmospheric CO<sub>2</sub> concentration on carbon isotope fractionation in C3 land plants. *Geochim. Et Cosmochim. Acta* 2012, *96*, 29–43. [CrossRef]
- Bezyk, Y.; Dorodnikov, M.; Sówka, I. Effects of climate factors and vegetation on the CO<sub>2</sub> fluxes and δ<sup>13</sup>C from re-established grassland. *E3S Web Conf.* 2017, 22, 00017. [CrossRef]
- 69. Werth, M.; Kuzyakov, Y. 13C fractionation at the root–microorganisms–soil interface: A review and outlook for partitioning studies. *Soil Biol. Biochem.* 2010, 42, 1372–1384. [CrossRef]