

Article

# The Effect of Deposition Time Optimization on the Photovoltaic Performance of $\text{Sb}_2\text{Se}_3$ Thin-Film Solar Cells

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**Abstract:** Antimony selenide ( $\text{Sb}_2\text{Se}_3$ ) photovoltaic thin-film materials have been recognized as suitable thin-film photovoltaic candidates for sustainable development due to the low toxicity of their constituent elements and abundant reserves. In this study, we employed the close space sublimation (CSS) method to fabricate solar cells with the FTO/ $\text{SnO}_2$ / $\text{Sb}_2\text{Se}_3$ /P3HT/C device architecture. By optimizing the deposition time, we achieved (hk1) orientation-preferred  $\text{Sb}_2\text{Se}_3$  films, the optimized device exhibited a peak efficiency of 5.06%. This work investigated the growth mechanism of antimony selenide using a complete characterization technique, while the experimental parameters were simulated and matched using Widget Provided Analysis of Microelectronic and Photonic Structures (wxAMPS) showing excellent potential in the deposition of optoelectronic thin films by close space sublimation.

**Keywords:**  $\text{Sb}_2\text{Se}_3$  solar cells; close-spaced sublimation; deposition time; simulation

## 1. Introduction

As a key technology for electrical energy conversion, solar cells have great potential to replace non-renewable fossil fuels [1,2]. Thin-film solar cells have offered more possibilities for the promotion and utilization of renewable energy in areas such as photovoltaic buildings and intelligent robots due to their many advantages, such as high material utilization efficiency, low energy consumption, high power generation rate, and flexible scalability [3–5]. New types of group III and V materials, like Copper Zinc Tin Sulfide (CZTS), Cadmium Telluride (CdTe), and metal halide perovskites, have been applied in thin-film solar cells [6–8]. However, the inherent toxicity of cadmium, the scarcity of elements such as tellurium, indium, and gallium, and the sensitivity of metal halide chalcogenides and CZTS materials to oxidation, sulfation, and hydrolysis have severely limited the sustainable development of these thin-film solar cell technologies [9–12].

In recent years, the emerging photovoltaic thin-film material antimony selenide ( $\text{Sb}_2\text{Se}_3$ ) has been considered a very promising candidate for future thin-film solar cells [13,14]. The toxicity of the constituent elements Sb and Se is extremely low, and they are abundant on earth, ensuring a sustainable supply of Sb and Se [15,16]. The bandgap of  $\text{Sb}_2\text{Se}_3$  films ranges between 1.1 eV and 1.3 eV, characterized by high short-wavelength light absorption ( $>10^5 \text{ cm}^{-1}$ ), high electron mobility ( $10 \text{ cm}^2\text{V}^{-1}\text{s}^{-1}$ ), and low material toxicity [17]. As an emerging solar cell, various manufacturing techniques for  $\text{Sb}_2\text{Se}_3$  films have been reported, such as close-spaced sublimation (CSS) [18], chemical molecular beam deposition (CMBD) [19], vapor transport deposition (VTD) [20], and electrodeposition [21]. Additionally, magnetron sputtering [22], chemical bath deposition (CBD) [23], and ultrasonic spray pyrolysis (USP) [24] techniques have been successfully employed in the fabrication processes of antimony selenide solar cells. The application of these technologies has not only facilitated improvements in the performance of antimony selenide solar cells but also provided new strategies for the commercialization of future solar cells.

$\text{Sb}_2\text{Se}_3$  has a unique quasi-one-dimensional crystal structure consisting mainly of  $(\text{Sb}_4\text{Se}_6)_n$  ribbons. The carrier transport efficiency along the  $(\text{Sb}_4\text{Se}_6)_n$  ribbons is much



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higher than that across the ribbons [25,26]. Therefore, to optimize  $\text{Sb}_2\text{Se}_3$  solar cells, it was essential to select appropriate manufacturing techniques to precisely control the growth orientation of the  $\text{Sb}_2\text{Se}_3$  thin films and ensure that their vertical conductivity was enhanced. Moreover, by engineering means to further optimize the crystal structure and orientation of  $\text{Sb}_2\text{Se}_3$  materials, it was anticipated to further improve their application performance in photovoltaic and photoelectric fields. Lin et al. demonstrated that  $\text{Sb}_2\text{Se}_3$  with (hk1) orientation proved more favorable for enhancing carrier collection compared to (hk0) orientation.  $\text{Sb}_2\text{Se}_3$  films oriented along (211) and (221) exhibit lower resistivity as opposed to the (120) orientation [27]. Furthermore, Zhou et al. found that the (001)-oriented  $\text{Sb}_2\text{Se}_3$  film could provide excellent carrier migration efficiency [28]. Capitalizing on the particular characteristic of  $\text{Sb}_2\text{Se}_3$  crystals, Mai's team employed the CBD method to fabricate high-quality  $\text{Sb}_2\text{Se}_3$  nanorod arrays along the [001] direction, providing profound insights into the 1D structure and applications of  $\text{Sb}_2\text{Se}_3$  crystals [29]. Therefore, meticulous control over the crystal orientation of  $\text{Sb}_2\text{Se}_3$  thin films to attain a preferred (hk1) orientation is indispensable for optimizing their conversion efficiency in solar cells.

$\text{Sb}_2\text{Se}_3$  has a relatively high saturation vapor pressure in the range of 450 to 500 °C, which provides ideal conditions for the formation of stable compounds in the gas phase [30]. Wang et al. demonstrated the growth process of  $\text{Sb}_2\text{Se}_3$  films and the formation mechanism of  $\text{Sb}_2\text{Se}_3$  nanorods appearing on  $\text{Sb}_2\text{Se}_3$  films by the VTD method [31]. Tao et al. demonstrated the potential of the CSS technique in investigating the microstructure of the  $\text{Sb}_2\text{Se}_3$  photoactive layer by independently controlling the source and substrate temperatures, resulting in high-quality  $\text{Sb}_2\text{Se}_3$  thin films with a (hk1)-preferred orientation [32]. Its sublimation properties allow it to transform directly from solid to gas at lower temperatures, which makes the CSS method a suitable technique to be utilized for controlling the deposition process of antimony selenide thin films. The CSS technique, as an advanced deposition approach equipped with real-time monitoring and control devices, enables precise optimization of deposition conditions, facilitating finer control over film quality and properties. Rijal et al. utilized a closed-space sublimation followed by a selenization approach to fabricate  $\text{Sb}_2\text{Se}_3$  solar cells, investigating the influence of selenization temperature on the structural and morphological properties of  $\text{Sb}_2\text{Se}_3$  films as well as the corresponding photovoltaic performance of the devices [33]. Fan et al. templated the growth of vertically oriented columnar  $\text{Sb}_2\text{Se}_3$  absorber layers via closed-space sublimation, achieving high-quality  $\text{Sb}_2\text{Se}_3$  absorber layers with passivation by controlling the crystal grain orientation and density of  $\text{Sb}_2\text{Se}_3$  seeds [34].

Additionally, numerical simulation, a vital theoretical tool for probing photovoltaic device performance, facilitates precise modifications to device structure and film parameters. Cao et al. utilized wxAMPS simulation software to model and simulate three different electron transport layer configurations ( $\text{CdS}$ ,  $\text{ZnO}$ , and  $\text{SnO}_2$ ) applied to  $\text{Sb}_2\text{Se}_3$  solar cells, demonstrating that  $\text{Sb}_2\text{Se}_3$  thin-film solar cells can achieve excellent optoelectronic performance with a simple device structure [35]. Memari et al. simulated colloidal CPQD solar cells with a 14.61% efficiency using minimal fitting parameters in SCAPS-1D and, through experimental and simulation analysis, comprehensively evaluated the photovoltaic performance of high-efficiency, flexible CPQD solar cells from a research standpoint [36]. Comparing simulation results with experimental data revealed the model's shortcomings and biases, thus enhancing the accuracy and reliability of the simulations. Data fitting and analysis facilitated a deeper understanding of the performance and behavior of solar cells, inspiring strategies to improve energy conversion efficiency. This included evaluating the performance of solar cells under different lighting conditions and exploring ways to optimize performance through material and structural adjustments [37]. Additionally, data fitting was crucial for optimizing manufacturing techniques, not only deepening the understanding of thin-film solar cell technology but also revealing which parameters were most critical and how to adjust them for optimal performance.

In this study, high-quality  $\text{Sb}_2\text{Se}_3$  films with (hk1)-preferred orientation were successfully deposited using the CSS technique and applied to the preparation of solar cell

devices with FTO/SnO<sub>2</sub>/Sb<sub>2</sub>Se<sub>3</sub>/P3HT/C structure. The effects of deposition time on the crystal orientation, interfacial properties and device performance of Sb<sub>2</sub>Se<sub>3</sub> solar cells were systematically investigated, and the formation mechanism of (hk1)-oriented Sb<sub>2</sub>Se<sub>3</sub> films was discussed in detail. The results indicated that appropriate deposition times facilitated the achievement of uniform grain distribution, moderate grain size, and reduced defect density in Sb<sub>2</sub>Se<sub>3</sub> films, which contributed to enhanced carrier transport efficiency and, consequently, improved photovoltaic performance of the devices. Furthermore, through simulation and emulation, the optimal device's experimental data were successfully matched, validating the rationality of the experimental design and the accuracy of the experimental operations. Additionally, simulation allowed researchers to gain a deeper understanding of how these processes affected the overall performance of the devices, thereby providing theoretical guidance for optimizing device design and material selection. This study demonstrated the significant potential of CSS technology in depositing photovoltaic thin films with specific crystal orientations and served as a reference for integrating experimental and simulation analyses in the study of thin-film solar cells.

## 2. Materials and Methods

### 2.1. Materials

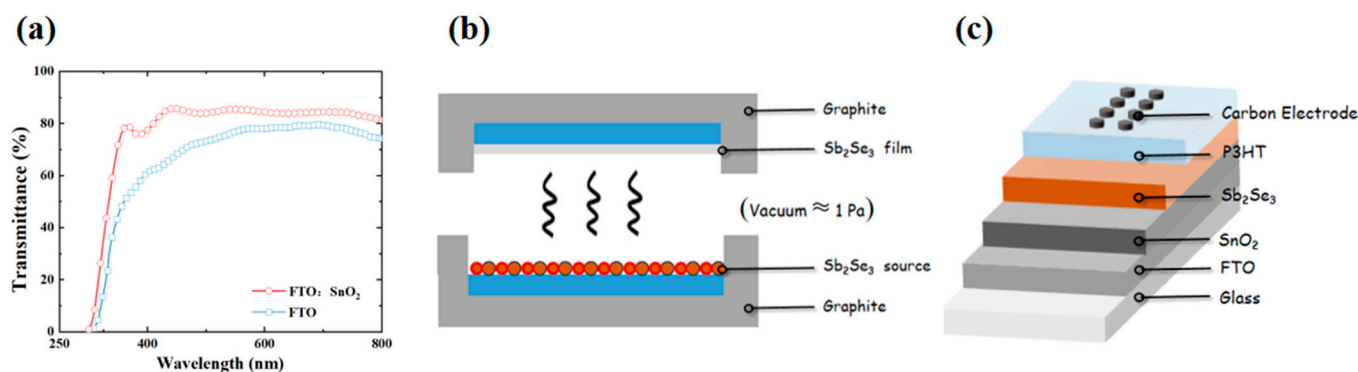
The FTO substrate (TEC15) was supplied by Youxuan New Energy Technology Co., Ltd. (Shenyang, China). The SnO<sub>2</sub> colloid solution (tin (IV) oxide, 15% in H<sub>2</sub>O colloidal dispersion) and 1,2-Dichlorobenzene (o-DCB, 99%) were purchased from Alfa Aesar. Sb<sub>2</sub>Se<sub>3</sub> powder (99.999%) was purchased from Ketai New Material Co., Ltd. (Nanchang, China). Polymer Light Technology Inc. (Xi'an, China) supplied the Regioregular poly (3-hexylthiophene) (P3HT). The low-temperature conductive carbon paste was supplied by MaterWin Co., Ltd. (Shanghai, China).

### 2.2. Device Fabrication

The composition of the Sb<sub>2</sub>Se<sub>3</sub> solar cell under investigation is depicted in Figure 1a. The FTO substrate (2.5 × 2.5 cm) underwent sequential ultrasonic cleaning with deionized water, ethanol, and isopropanol in a glass cleaning rack, with each cleaning lasting 8 min. Subsequently, any residual alcohol solution was removed from the glass surface using nitrogen and dried in a homemade UV-Ozone cleaner for 30 min. This cleaning procedure ensures the FTO substrate is thoroughly cleaned, thereby guaranteeing accurate and reproducible experimental outcomes. Following this, the SnO<sub>2</sub> electron transport layer was prepared using a SnO<sub>2</sub> nanoparticle dispersion as a precursor solution. The pristine SnO<sub>2</sub> colloidal solution was spin-coated onto the cleaned FTO substrate twice at 2000 r/min for 60 s. After the spin-coating process, the FTO substrate was transferred to a heated bench and annealed in an air atmosphere at 150 °C for 30 min to form a dense tin dioxide electron transport layer, resulting in the formation of a dense SnO<sub>2</sub> electron transport layer. Tin dioxide was chosen as the electron transport layer due to its excellent electron transport properties and suitable electron mobility, thereby enhancing the charge transfer efficiency of antimony selenide solar cells. Furthermore, the high transmittance of tin dioxide enhances the light absorption of antimony selenide solar cells, maximizing the conversion of light energy into electrical energy.

By forming a dense SnO<sub>2</sub> film on the FTO substrate, effective prevention of photogenerated carrier recombination was achieved, thereby maximizing the photoelectric conversion efficiency. In addition, we conducted transmittance tests on the FTO-SnO<sub>2</sub>, with Figure 1a displaying the transmittance for both the FTO substrate without spin-coated SnO<sub>2</sub> and that with FTO-SnO<sub>2</sub>. After the fabrication of the SnO<sub>2</sub> electron transport layer on the FTO substrate, the transmittance of the FTO-SnO<sub>2</sub> films did not decrease; the transmittance at the FTO substrate at 400 nm was approximately 61%, while it increased to around 72% for the FTO-SnO<sub>2</sub> films. This can be attributed to the lower surface roughness of FTO-SnO<sub>2</sub>, which resulted in reduced light scattering and increased direct transmittance. Furthermore, the transmittance of FTO-SnO<sub>2</sub> remained stable at around 80% for wavelengths of approx-

imately 500 nm and above, and its high transmittance in the visible range (400~550 nm) enables light to reach the light-absorbing layer more effectively.



**Figure 1.** (a) Transmission spectra of FTO films and FTO:SnO<sub>2</sub> films; (b) Structure of the studied Sb<sub>2</sub>Se<sub>3</sub> solar cell; (c) Schematic diagram of the CSS technique to deposit Sb<sub>2</sub>Se<sub>3</sub> film.

Subsequently, Sb<sub>2</sub>Se<sub>3</sub> powder was loaded into the chamber of a customized CSS device as the evaporation source, facilitating the vacuum evaporation of the Sb<sub>2</sub>Se<sub>3</sub> film onto the SnO<sub>2</sub> electron transport layer. A schematic representation of the preparation process is provided in Figure 1b. The deposition rate was controlled by resistive wire current, and Sb<sub>2</sub>Se<sub>3</sub> films were deposited in a vacuum of 1 Pa with an evaporation distance of 10 mm. Conducting the sublimation process under vacuum prevents interference from gases or impurities, ensuring a stable environment for antimony selenide powder sublimation and preventing oxidation and other undesired reactions. The graphite substrate of the Sb<sub>2</sub>Se<sub>3</sub> source was connected to the instrument panel with a sensor in our CSS device, and the temperature difference between the source and the target was strictly controlled and monitored. The temperature difference between the source temperature and the target temperature was tested to be within  $\pm 5$  °C. Before deposition, the substrate was preheated for a period of time. Once the evaporation source temperature reaches 250 °C, the SnO<sub>2</sub>-coated FTO glass substrate is transferred from the loading chamber to the CSS unit. Subsequently, the temperature is held at 350 °C for 20 min to ensure uniform temperature distribution in the chamber for the antimony selenide powder. Deposition of the Sb<sub>2</sub>Se<sub>3</sub> film then begins at 520 °C, with the shutter between the deposition source and the substrate closed to initiate the growth of Sb<sub>2</sub>Se<sub>3</sub> grains on the substrate. Deposition times are set at 40, 50, 60, and 70 s, respectively. After completion of the process, the shutter is opened, the heater is turned off, and the film is returned to the loading chamber to cool to room temperature before removal, achieved by purging the chamber with nitrogen gas to release the vacuum. During the cooling phase, special care was taken to ensure a gradual reduction in temperature to avoid thermal shock, which could potentially induce cracks or dislocations within the crystalline structure of the films. The cooling rate was meticulously controlled, with a steady decrease in temperature over a predefined time period, to allow for the proper annealing of the films. This annealing step is vital for relieving any internal stresses developed during the deposition process and for promoting the crystallinity of the Sb<sub>2</sub>Se<sub>3</sub> films.

Moreover, a 20 mg/mL solution of P3HT was prepared by accurately weighing 20 mg of P3HT and dissolving it in 1 mL of 1,2-Dichlorobenzene. The solution was then stirred at 50 °C for 5 h using a magnetic stirrer to ensure complete dissolution. Subsequently, 80  $\mu$ L drops of the P3HT solution were dispensed onto the sample film using a pipette gun and spin-coated at 2000 rpm for 60 s. Then, anneal in an air atmosphere at a temperature of 120 °C for 10 min to uniformly disperse the P3HT hole transport layer on the Sb<sub>2</sub>Se<sub>3</sub> thin film. As an organic semiconductor material with good hole transport properties, P3HT can effectively transport photogenerated holes from the Sb<sub>2</sub>Se<sub>3</sub> film to the electrode, which helps to improve the charge collection efficiency of the photovoltaic cell. In the

processing of P3HT hole transport layers, the annealing temperature had a significant impact on promoting the rearrangement and interaction of molecular chains within the material, thereby affecting device performance. Higher annealing temperatures were found to improve the orderliness of P3HT chains, potentially increasing the photoelectric conversion efficiency of the films. However, excessively high annealing temperatures could also lead to structural damage in P3HT, impacting the overall performance of the films. Moreover, the P3HT hole transport layer can regulate the energy level difference between  $\text{Sb}_2\text{Se}_3$  and the electrode, which helps to improve the photogenerated carrier transport efficiency and thus enhance the photoelectric conversion efficiency.

Finally, carbon electrodes were fabricated using carbon paste. This process involved placing a homemade mask plate on the surface of the film and then uniformly applying the carbon paste using a scraping method. Subsequently, the carbon electrodes were annealed in an air atmosphere at 150 °C for 30 min to promote curing. Carbon electrodes were chosen because they do not involve the use of rare materials or harmful chemicals in their preparation and are environmentally friendly in line with sustainable development. At the same time, carbon electrodes have high chemical stability at room temperature, which can withstand chemical reactions during battery operation and extend the service life of the battery. The structure of the final prepared  $\text{Sb}_2\text{Se}_3$  thin-film solar cell device is shown in Figure 1c. The electrodes possessed a circular cross-section with an area of 0.07 cm<sup>2</sup>.

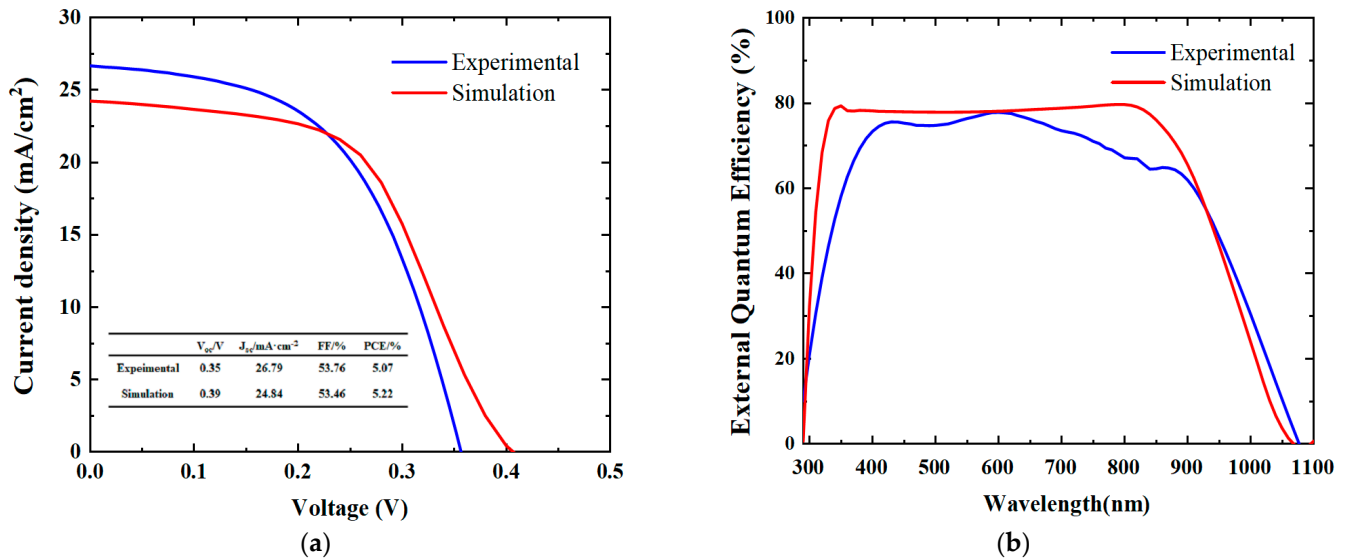
### 2.3. Characterization

The microstructure of  $\text{Sb}_2\text{Se}_3$  thin films was characterized by scanning electron microscopy (SEM) (Sigma, Zeiss, Jena, Germany), and surface topography and cross-sectioning of material samples were carried out to realize the analysis of sample crystallization and densification. The X-ray diffraction (XRD) patterns of the  $\text{Sb}_2\text{Se}_3$  films were analyzed using a diffractometer (XRD-7000, Shimadzu, Tokyo, Japan). CuK $\alpha$  radiation was employed as the radiation source, with an incident wavelength of 1.5406 Å and a voltage of 40 kV. The scanning range was set between 10° and 60°, and the samples were characterized at a scanning speed of 10° min<sup>-1</sup>. The solar simulator (SS100A, Photo Emission Tech Inc., Moorpark, CA, USA) was employed for the measurement of current density–voltage (J–V) characteristics. The solar simulator, class AAA, generated incident solar radiation with an intensity of 100 mW/cm<sup>2</sup> using the AM 1.5 spectrum at a test temperature of 25 °C. Electrical performance parameters such as open-circuit voltage ( $V_{oc}$ ), short-circuit current density ( $J_{sc}$ ), fill factor (FF), photoelectric conversion efficiency (PCE) were obtained for solar cell installations under light. The external quantum efficiency (EQE) spectra were obtained using the quantum efficiency system (SCS100, Zolix Co., Ltd., Beijing, China) for the analysis of photovoltaic response efficiency and the loss of photocurrent in solar cells, indicating the ratio of the number of charge carriers to the number of incident photons introduced externally. The scanning range was set from 300 to 1100 nm, with a scanning speed of 5 nm/s.

## 3. Results

The EQE and J–V curves of the experimental results were simulated and analyzed using the wxAMPS software, as shown in Figure 2. During the simulation, we considered two key electrode materials as a function of FTO (tin fluoride oxide) and C (carbon), which are 4.4 and 5.0 eV, respectively. Accurately setting these figures of merit was crucial for simulating the optoelectronic device's behavior. In the optical part of the simulation, we considered the incidence of sunlight from the FTO side and set the light reflectivity of the top and bottom electrodes to 0% and 100%, respectively. Surface recombination velocities for the front and back contacts are both  $1 \times 10^7$  cm/s. This design fully takes into account the light absorption and reflection processes and provides the basis for more accurate simulation results. In terms of environmental parameters, we set the external working environment to standard conditions: a temperature of 300 K, an AM 1.5 G solar spectrum, and an incident light intensity of 100 mW/cm<sup>2</sup>. These settings allowed us to

simulate environments similar to the actual operating conditions during the simulation of the experimental results and thus predict the performance of the photovoltaic devices more accurately.



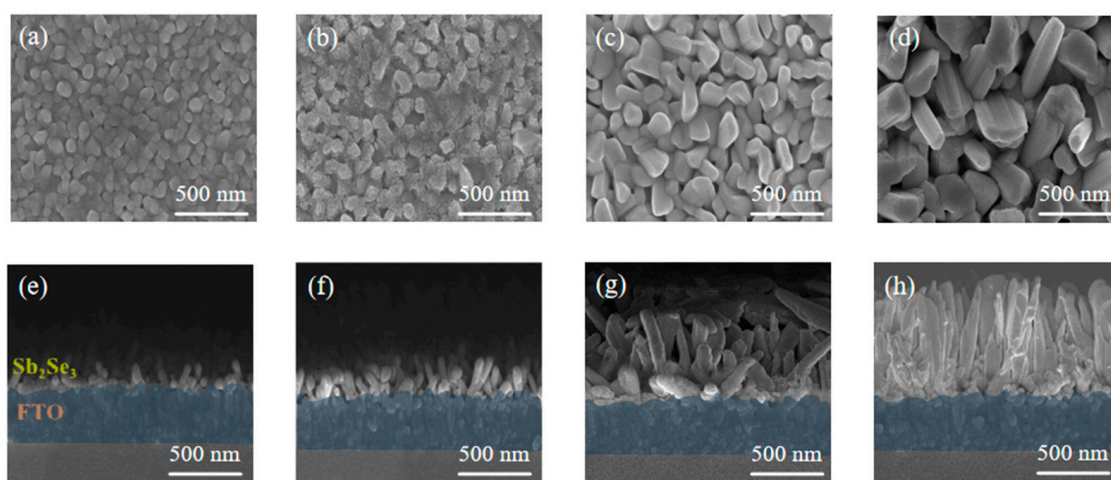
**Figure 2.** Comparative analysis between the experimental and simulated results of the highest efficiency  $Sb_2Se_3$  devices: (a) J-V curves; (b) EQE curves.

The photocurrent-voltage (J-V) characteristics and external quantum efficiency (EQE) of the experimentally prepared, most efficient  $Sb_2Se_3$  solar cell samples with a deposition time of 60 s were simulated, and these simulations were compared in detail with the corresponding experimental data. The evident correlation between them affirmed the validity of the model. Considering the increased thickness of the  $Sb_2Se_3$  absorption layer, alongside the more pronounced growth of  $Sb_2Se_3$  crystals in a direction perpendicular to the substrate, the absorption coefficient of the  $Sb_2Se_3$  layer has been appropriately enhanced. This adjustment ensures consistency between the values of the  $J_{sc}$  and the EQE. Then, the defect density of the  $Sb_2Se_3$  absorber layer was fine-tuned to match the FF, and adjustments were made to the defect density at the interface as well as the electron and hole capture rates to align with the  $V_{oc}$ . Table 1 provides a comprehensive list of all simulation parameters utilized in this research. Through comparison with experimental data, it was observed that the simulation results of the optimized device showed a high degree of consistency with the experimental observations. This not only proved the effectiveness of the optimization strategies employed but also provided a solid theoretical foundation and methodological guidance for further enhancing the performance of  $Sb_2Se_3$  solar cells.

**Table 1.** Parameters used in the simulation of  $Sb_2Se_3$  solar cells [38–44].

Parameter	$SnO_2$	$Sb_2Se_3$	P3HT
Thickness (nm)	50	500	50
Relative permittivity ( $\epsilon_r$ )	9	10	10
Electron affinity, $X$ (eV)	4.5	4.04	4
Electron mobility, $\mu_n$ ( $cm^2/Vs$ )	0.2	4	0.006
Hole mobility, $\mu_p$ ( $cm^2/Vs$ )	0.2	0.1	0.006
$N_a$ ( $1/cm^3$ )	0	$1 \times 10^{13}$	$5 \times 10^{19}$
$N_d$ ( $1/cm^3$ )	$1 \times 10^{18}$	0	0
$N_t$ ( $1/cm^3$ )	$1 \times 10^{18}$	$2 \times 10^{16}$	$1 \times 10^{17}$
$E_g$ (eV)	3.6	1.2	3.2
Density of state of the conduction band, $N_c$ ( $1/cm^3$ )	$2.2 \times 10^{18}$	$2.2 \times 10^{18}$	$1.0 \times 10^{21}$
Density of state of the valence band, $N_v$ ( $1/cm^3$ )	$1.8 \times 10^{19}$	$1.8 \times 10^{19}$	$2.0 \times 10^{20}$

Figure 3 shows images captured through scanning electron microscopy (SEM) in both planar view and cross-section. In the evaporation process, the  $\text{Sb}_2\text{Se}_3$  source material underwent heating and vaporization, subsequently depositing onto a slide as a film through surface diffusion and rearrangement. The images provide valuable insights into both the microstructural features and the deposition mechanisms involved in forming the  $\text{Sb}_2\text{Se}_3$  thin films. In the SEM cross-sectional images, the FTO substrate and the  $\text{Sb}_2\text{Se}_3$  crystalline layer were distinguished via color coding. It was observed that the thicknesses of the  $\text{Sb}_2\text{Se}_3$  films obtained under different preparation conditions were 160 nm, 250 nm, 500 nm, and 735 nm, respectively. As depicted in Figure 3a,e,  $\text{Sb}_2\text{Se}_3$  grains deposited for 40 s appear small in size and are densely dispersed on the film surface. Shorter deposition times may have implied faster nucleation rates and slower grain growth, leading to smaller and more densely distributed grains. Grain size is directly related to light absorption efficiency and carrier recombination rates, thereby affecting the conversion efficiency of solar cells.



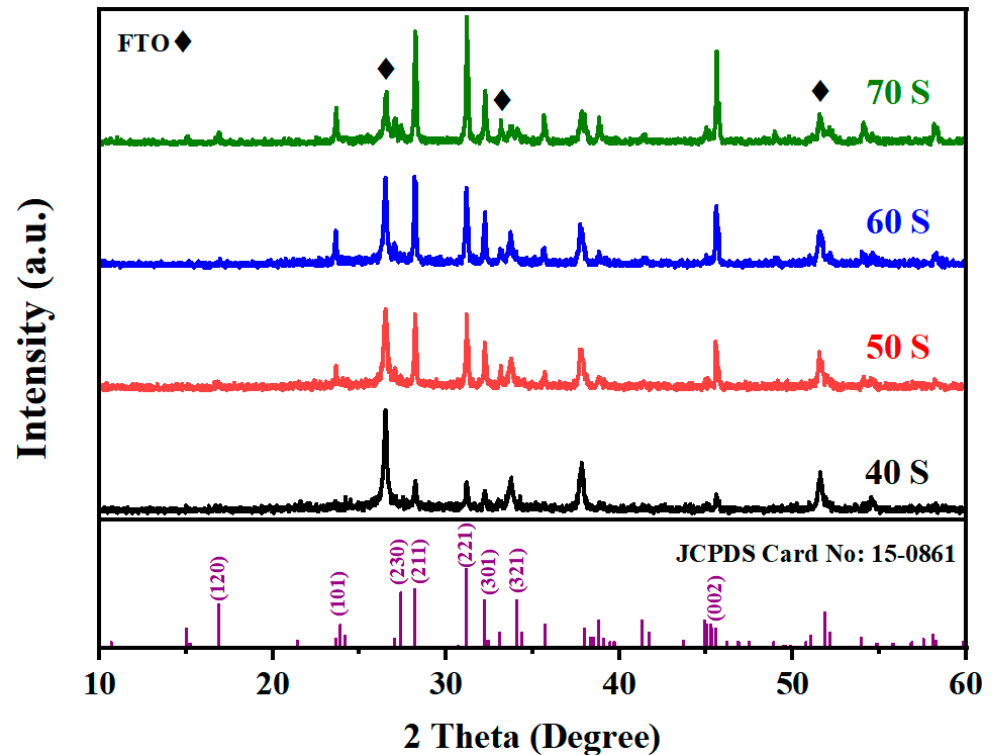
**Figure 3.** SEM images of  $\text{Sb}_2\text{Se}_3$  thin films prepared with optimized deposition times under the conditions of (a,e) 40 s, (b,f) 50 s, (c,g) 60 s, and (d,h) 70 s.

Extending the deposition time to 60 s (Figure 3c,g) enhances the contact between the  $\text{Sb}_2\text{Se}_3$  grains and the substrate, resulting in clearer microscopic details and a significant increase in grain size. Additionally, a more stable trend of longitudinal growth of the  $(\text{Sb}_4\text{Se}_6)_n$  band is observed. Samples grown for 70 s (Figure 3d,h) exhibit a distinct feature of vertically oriented grain distribution in the morphology of the  $\text{Sb}_2\text{Se}_3$  films. This phenomenon could be attributed to the extension of the evaporation time, which provided more opportunities for crystal growth, thereby enhancing the crystallinity and orderliness of the crystals. The extended evaporation time allowed for a more stable deposition rate, enabling atoms or molecules in the vapor phase to have more opportunities to find optimal sites for lattice incorporation. High crystallinity and orderliness are typically associated with better charge transport and fewer defects.

In Figure 4, we have investigated the impact of deposition time on the crystal quality of  $\text{Sb}_2\text{Se}_3$  thin films through XRD analysis. All discernible diffraction peaks aligned with the orthorhombic  $\text{Sb}_2\text{Se}_3$  index, consistent with the standard card (JCPDS 15-0861). This confirmed the crystalline nature of the investigated material.

At a deposition time of 40 s, distinct peaks corresponding to the (211) and (221) planes are observed in the  $\text{Sb}_2\text{Se}_3$  films, indicating a clear preferential crystallographic orientation of the present  $(\text{Sb}_4\text{Se}_6)_n$  nanoribbon along the  $[\text{hk}1]$  direction. With the change of evaporation time to 60 s, the enhancement trend of the diffraction peaks on the (211), (221) crystal surface of  $\text{Sb}_2\text{Se}_3$  was obvious, and it could be inferred that the orientation and crystallization properties of the  $\text{Sb}_2\text{Se}_3$  crystal structure had been improved. If the overall intensity of the  $(\text{hk}1)$  diffraction peak surpasses that of  $(\text{hk}0)$ , the  $\text{Sb}_2\text{Se}_3$  film will exhibit a columnar growth pattern, demonstrating improved carrier transport properties [45].

Upon extending the deposition time to 70 s, a significant enhancement was observed in the diffraction peak corresponding to the (002) plane, indicating further optimization and orderly arrangement of the  $\text{Sb}_2\text{Se}_3$  crystal structure under this condition. This result is consistent with previous SEM observations, which revealed a noticeable improvement in the surface morphology and growth of the  $\text{Sb}_2\text{Se}_3$  thin films. It further confirms that adjusting the deposition time can effectively control and enhance the crystalline quality of the  $\text{Sb}_2\text{Se}_3$  thin films.



**Figure 4.** The XRD patterns of the  $\text{Sb}_2\text{Se}_3$  films synthesized at different deposition times.

In order to further investigate the effect of deposition time on the texture and preferred orientation of  $\text{Sb}_2\text{Se}_3$  films, the study also investigated the texture coefficient (TC) values for different crystalline surfaces ( $hkl$ ). Exploring the TC of various crystallographic planes in  $\text{Sb}_2\text{Se}_3$  thin films uncovered trends in crystal orientation changes during the film growth process. This investigation provided critical theoretical groundwork for optimizing the optoelectronic properties and application development of  $\text{Sb}_2\text{Se}_3$  thin films. The texture coefficients (TCs) of the (120), (230), (321), (301), (221), (211), (101) and (002) peaks were calculated using the following formula:

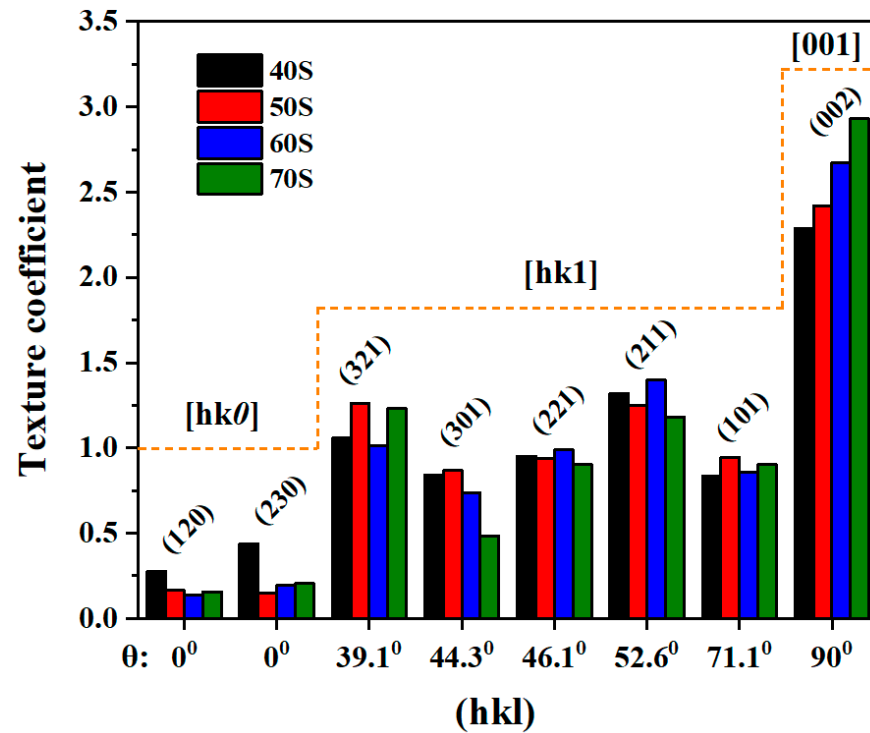
$$TC(hkl) = \frac{I(hkl)/I_0(hkl)}{\sum_n I_i(hkl)/I_0(hkl)} \times 100\% \quad (1)$$

where  $I_{(hkl)}$  denotes the intensity of the ( $hkl$ ) grain,  $I_0(hkl)$  denotes the corresponding intensity of the ( $hkl$ ) plane in the standard powder diffraction data (JCPDS 15-0861),  $N$  denotes all number of diffraction peaks taken into consideration in the formula [46].

Figure 5 illustrates that as the deposition time of  $\text{Sb}_2\text{Se}_3$  extends from 40 to 60 s, the orientations of TC(221) and TC(211) exhibit an upward trend. Upon reaching 70 s, the TC(002) orientation of the  $\text{Sb}_2\text{Se}_3$  film peaks, accompanied by a decline in the TC(221) and TC(211) orientations. Additionally, it is observed that the TC(101) orientation gradually surpasses 1, indicating a potential reorientation of the  $(\text{Sb}_4\text{Se}_6)_n$  nanoribbons within the film as crystallinity improves. A portion of the ( $h0l$ ) crystallographic plane undergoes a shift toward the preferred orientation, while the predominance of the (002) crystallographic



plane persists. However, the diffraction peak weaving coefficients of the (211) and (221) crystallographic planes diminish. These findings suggest that optimizing the deposition time enhances the crystallinity and grain size of the films, fostering a more pronounced (hk1) preferred orientation. Consequently,  $\text{Sb}_2\text{Se}_3$  films exhibit superior properties when deposited for 60 s.



**Figure 5.** Texture coefficient of the  $\text{Sb}_2\text{Se}_3$  films synthesized at different deposition times.

According to the above evaporation time study, the  $\text{Sb}_2\text{Se}_3$  films were characterized by high crystallinity and preferred (hk1) orientation at an evaporation time of 60 s. As a result,  $\text{Sb}_2\text{Se}_3$  solar cells with FTO/ $\text{SnO}_2$ / $\text{Sb}_2\text{Se}_3$ /P3HT/C device structure were prepared, and the J-V curves of the devices are shown in Figure 6a. As the crystallinity and structural integrity of the  $\text{Sb}_2\text{Se}_3$  thin films reach an ideal state at 60 s, the device performance gains significant improvement in terms of charge transport as well as interface matching. The (hk1) optimally oriented grains result in a more ordered and stable structure in the grain boundary region, which leads to a lower surface energy of the grain boundaries, and the obtained  $\text{Sb}_2\text{Se}_3$  grains grown on the substrate along the perpendicular direction are conducive to the reduction of interfacial compounding and absorption of transport losses. Table 2 shows the photovoltaic parameters of  $\text{Sb}_2\text{Se}_3$  thin-film solar cells with different deposition times.

**Table 2.** Photovoltaic parameters of  $\text{Sb}_2\text{Se}_3$  thin-film solar cells at different deposition times.

Deposition Time (s)	$V_{oc}$ (V)	$J_{sc}$ ( $\text{mA}/\text{cm}^2$ )	FF (%)	PCE (%)	$R_s$ ( $\Omega$ )	$R_{sh}$ ( $\Omega$ )
40	0.284	20.05	47.43	2.704	79.56	2379.4
50	0.318	22.01	54.19	3.794	63.55	4465.1
60	0.352	26.79	53.76	5.067	54.72	4556.2
70	0.295	23.51	44.00	3.047	81.85	829.52

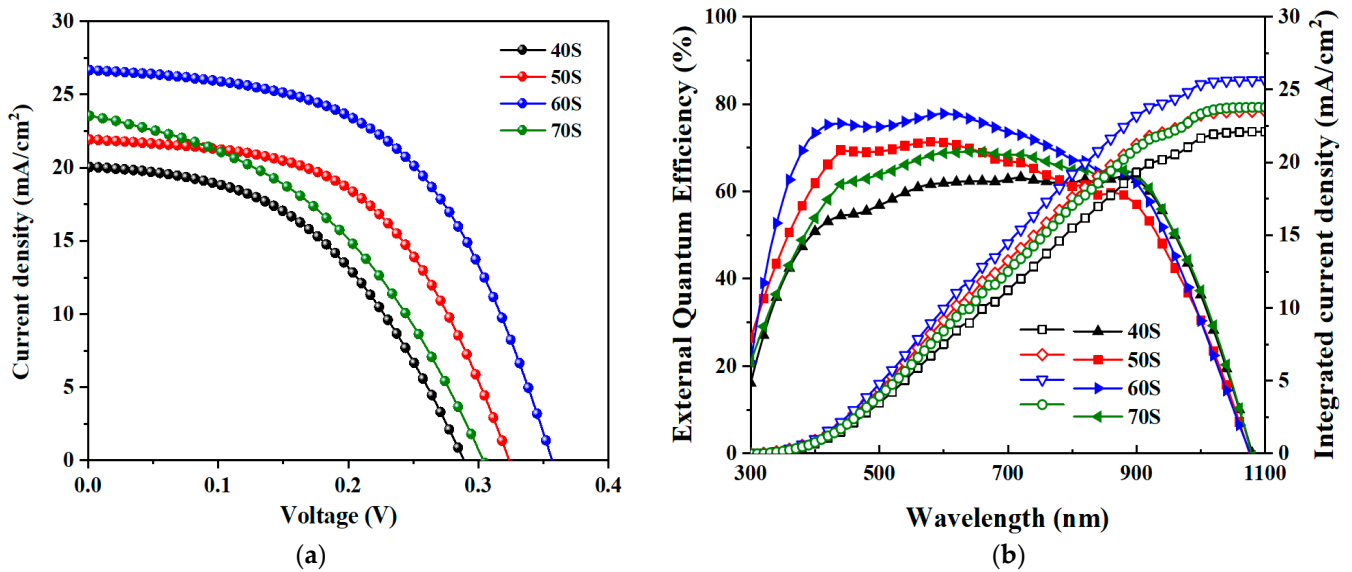


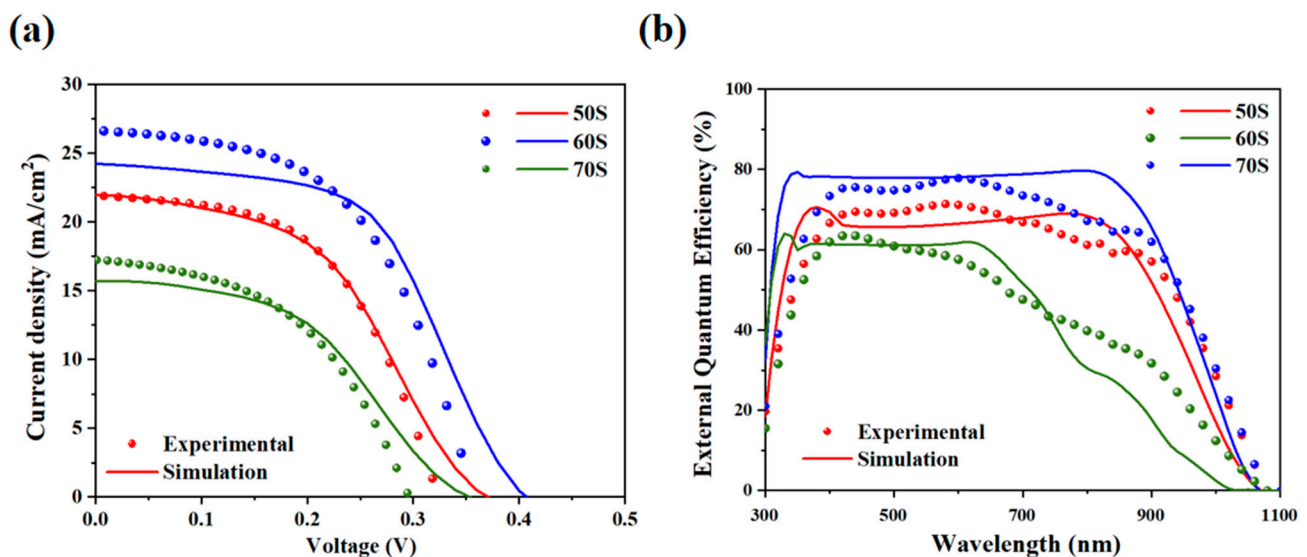
Figure 6.  $\text{Sb}_2\text{Se}_3$  thin-film solar cells at different deposition times: (a) J-V curves; (b) EQE curves.

In particular, the increase in evaporation time from 40 s to 60 s, the decrease in series resistance ( $R_s$ ) from  $79.56 \Omega$  to  $54.72 \Omega$ , and the increase in parallel resistance ( $R_{sh}$ ) from  $2379.4 \Omega$  to  $4556.2 \Omega$  can be attributed to the fact that an increase in the (hk1)-orientation ratio of the  $\text{Sb}_2\text{Se}_3$  photoreactive layer enhances the longitudinal conductivity, which explains the corresponding enhancement in FF. The optimum photoelectric conversion efficiency of 5.06% was obtained for the device under this condition. Whereas the performance degradation of the devices with an evaporation time greater than 60 s may be attributed to the shift of the (101) crystallographic facets of the  $\text{Sb}_2\text{Se}_3$  films to selective growth, the additional defects introduced by the nonessential crystallographic facets introduce energy at the surface of the grain boundaries, leading to an increase in the surface energy. The device  $V_{oc}$  decreases to 0.244 V and the  $J_{sc}$  decreases to  $19.86 \text{ mA}/\text{cm}^2$ . In addition, (002) peak strength selectivity implies that the longitudinal grains perpendicular to the substrate have the largest percentage of grains, and the larger intergranular voids and surface roughness give the films high interfacial defects. Furthermore, the increased surface roughness was likely detrimental to the interface characteristics between  $\text{Sb}_2\text{Se}_3$  and P3HT. All these factors contributed to an increase in  $J_0$  (reverse saturation current), which in turn led to a reduction in the FF from 53.76% to 44.00%.

The broad response of all devices in the range of 350–1050 nm can be seen from the EQE spectra in Figure 6b, with the 60s device obtaining an EQE maximum of 78% at 610 nm. The EQE integral currents measured in the AM1.5 spectra are in very good agreement with  $J_{sc}$ . In addition, the effect of longitudinal optimization on the device is reflected in the significantly improved short-wave response, whereas the EQE mapping shows that the integration current of the  $\text{Sb}_2\text{Se}_3$  thin-film device decreases sharply in the wavelength range of 400~800 nm. This suggests that as the evaporation time increases, when carriers encounter voids during their movement, they may be blocked or trapped in the voids. The carriers are conducted through the jumps between the lattices, which leads to the restriction of the carrier flow, causing the gradual degradation of the various device performances. This defect was primarily concentrated at the rear end of the absorber layer, leading to significant recombination of photogenerated carriers in this area. Consequently, the spectral response in the mid and long wavelengths was reduced, resulting in a decrease in both the FF and  $J_{sc}$  of the device, thereby diminishing the device's performance.

Further, we simulated and fitted the results of J-V curves and EQE curves for  $\text{Sb}_2\text{Se}_3$  devices with different deposition times (50 s, 60 s, 70 s) in Figure 6. For all fitted samples, the simulated J-V curves are slightly higher than the experimental data, which may indicate a slight idealization of the parameters assumed by the model. The simulated short-circuit

current density was slightly lower than the experimental data, while the simulated open-circuit voltage was marginally higher than the experimental results. This indicated that there were discrepancies in the absorption coefficient, the rate of electron-hole pair generation, or transport properties considered in the simulation compared to the experimental conditions. Consequently, the photovoltaic performance of the device post-simulation fitting was slightly optimistic relative to the experimental outcomes. For the champion device with a deposition time of 60 s that was mentioned earlier, the simulation curves exhibited good agreement with the experimental data, suggesting that the mechanisms for photogenerated carrier capture and recombination considered in the simulation might have closely mirrored the experimental conditions, with carrier recombination losses at interfaces being minimized. Furthermore, as observed from Figure 7b, the consistency between experimental data and simulation was better in the short-wavelength region, aligning the model's assumptions with the actual absorption and conversion efficiency of that spectral part. However, in the long-wavelength region, particularly beyond 800 nm, the divergence became more pronounced. The increase in the thickness of the  $\text{Sb}_2\text{Se}_3$  absorption layer, accompanied by an increase in defects within the  $\text{Sb}_2\text{Se}_3$  film, likely led to exacerbated carrier recombination, as reflected by the reduced EQE in the 400~750 nm wavelength range, suggesting significant carrier recombination within the absorber layer. This deviation corresponded to the previous analysis of  $\text{Sb}_2\text{Se}_3$  crystal growth in experimental preparation.



**Figure 7.** Experimental and simulation comparison of  $\text{Sb}_2\text{Se}_3$  thin-film solar cells with different deposition times: (a) J-V curves; (b) EQE curves.

The experimental results of J-V and EQE curves align with theoretical simulations, confirming that an optimal thickness of the absorptive layer ensures both high absorbance and the establishment of a robust built-in electric field with low defect density, thus yielding high photovoltaic conversion efficiency. Consequently, meticulously controlling the thickness of  $\text{Sb}_2\text{Se}_3$  films via the CSS technique produces high-quality films with a preferred (hk1) orientation, enhancing the photovoltaic performance of  $\text{Sb}_2\text{Se}_3$  solar cells.

#### 4. Conclusions

In this study,  $\text{Sb}_2\text{Se}_3$  thin films were successfully prepared using the close-space sublimation method, and precise control of film thickness and crystallization quality was achieved, based on which solar cell devices with FTO/ $\text{SnO}_2$ / $\text{Sb}_2\text{Se}_3$ /P3HT/C structures were prepared. The effects of deposition time on the crystal structure, morphology and optical properties of  $\text{Sb}_2\text{Se}_3$  films were systematically investigated, and the formation mechanism of (hk1)-oriented  $\text{Sb}_2\text{Se}_3$  films was thoroughly discussed. The results show

that a reasonable deposition time can regulate the growth process of  $(\text{Sb}_4\text{Se}_6)_n$  nanoribbons and make them grow in an orderly manner along the (hk1) crystallographic direction. The efficiency of photovoltaic devices made of high-quality  $\text{Sb}_2\text{Se}_3$  films based on the (hk1) orientation is significantly improved, and an optimal device with a PCE of 5.06% is finally realized. However, prolonged deposition time leads to additional defects in grains with non-desired orientation, which affect carrier transport and lifetime, thus degrading device performance. In addition, we simulated and validated the experimental data of  $\text{Sb}_2\text{Se}_3$  solar cells using wxAMPS software to further confirm the scientific validity and credibility of the conclusions drawn from the experiments, thus providing theoretical support for the related research. This work highlights the significant potential of the CSS technique in fabricating photovoltaic thin films with precise crystal orientations. It offers implications not only for advancing high-performance photovoltaic devices through close-space sublimation methods but also for inspiring novel ideas and directions in future photovoltaic material and device design.

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