

Article

# Tailoring the Size and Shape—New Path for Ammonium Metavanadate Synthesis

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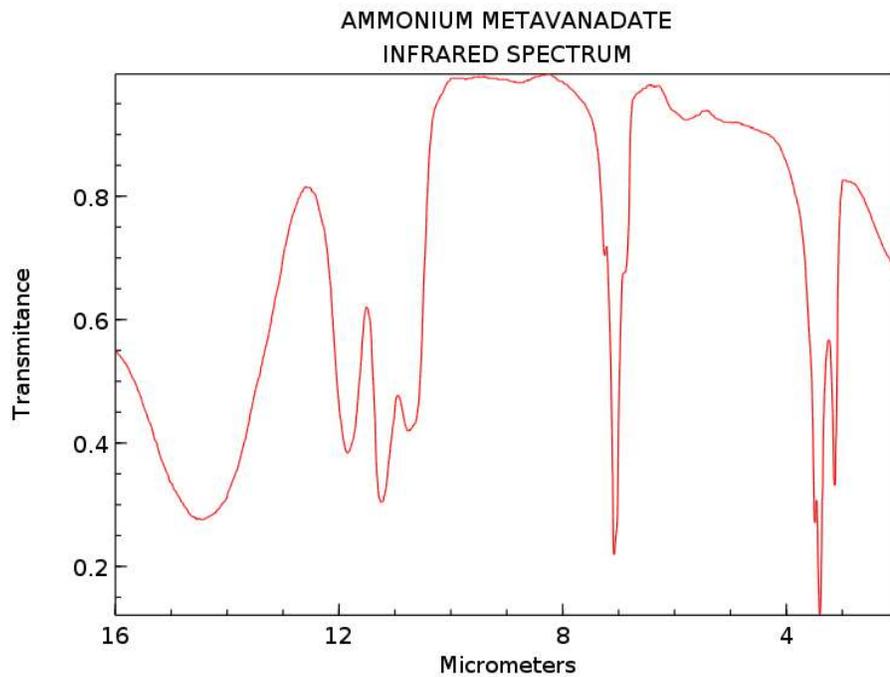


Figure S1. FTIR spectrum of ammonium metavanadate [1].

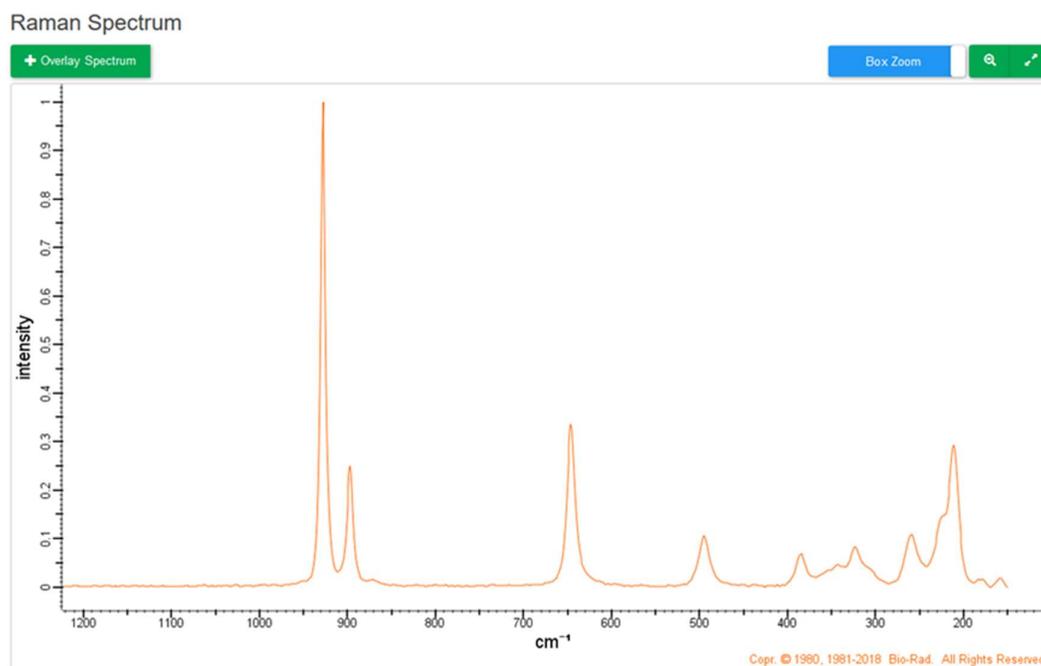
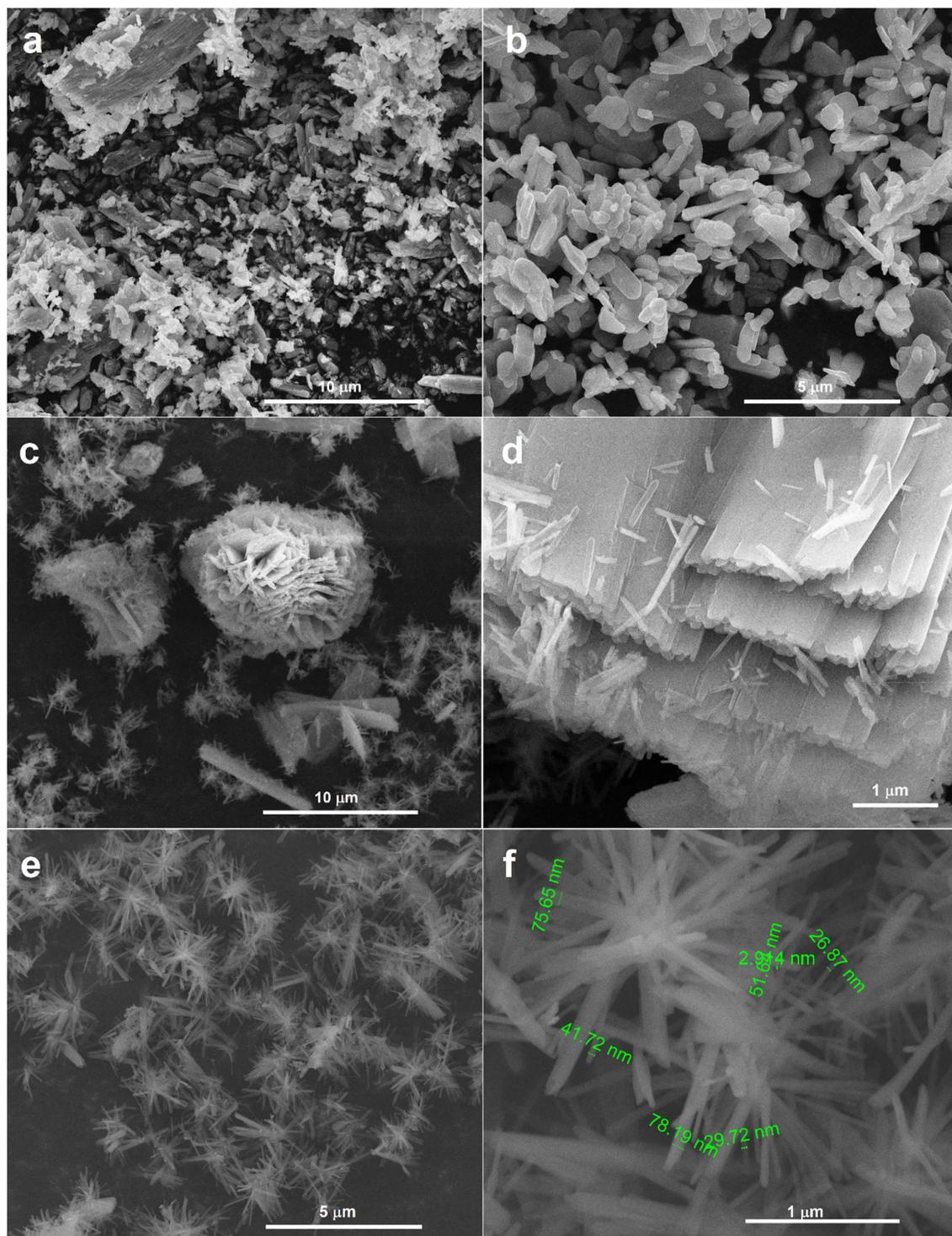
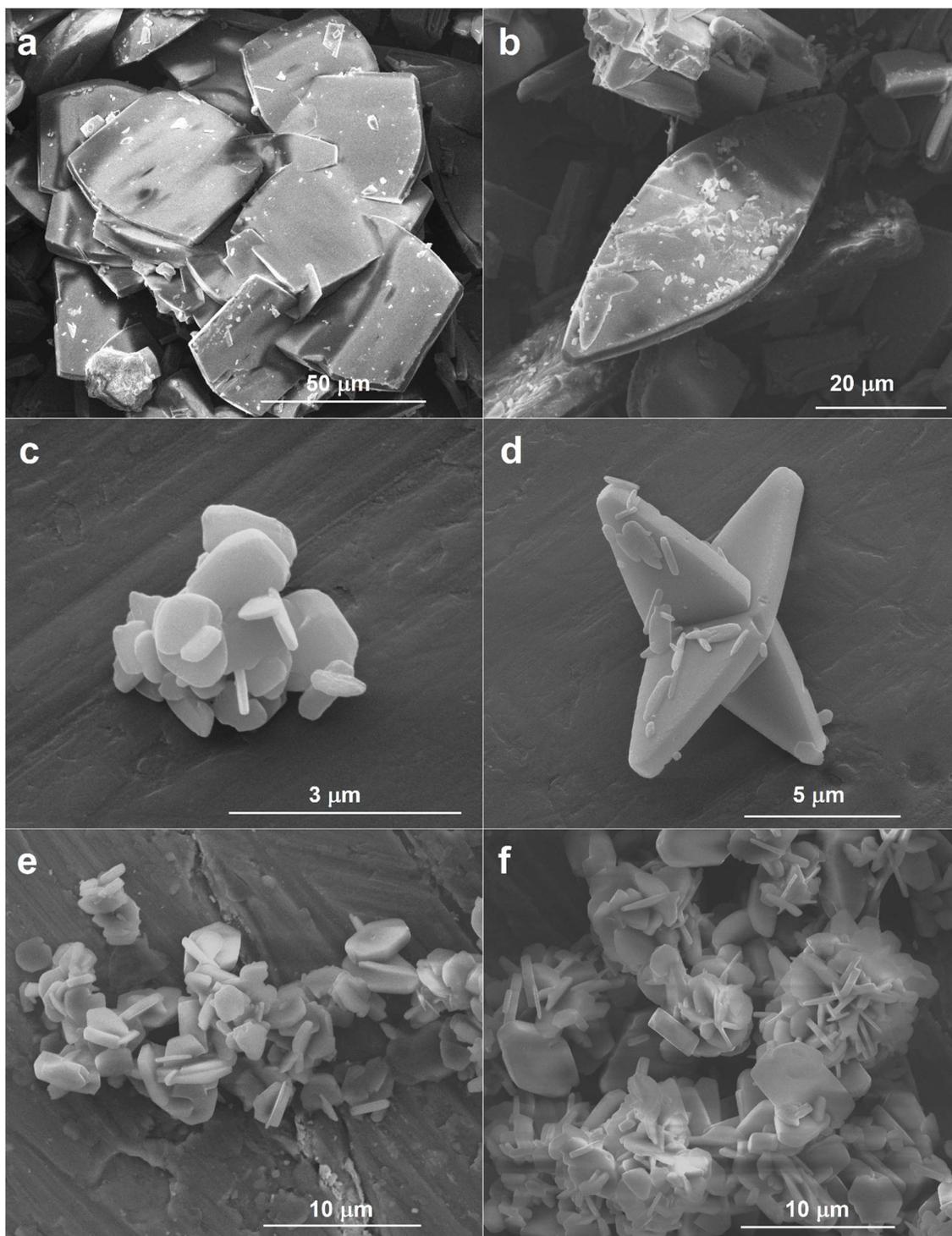


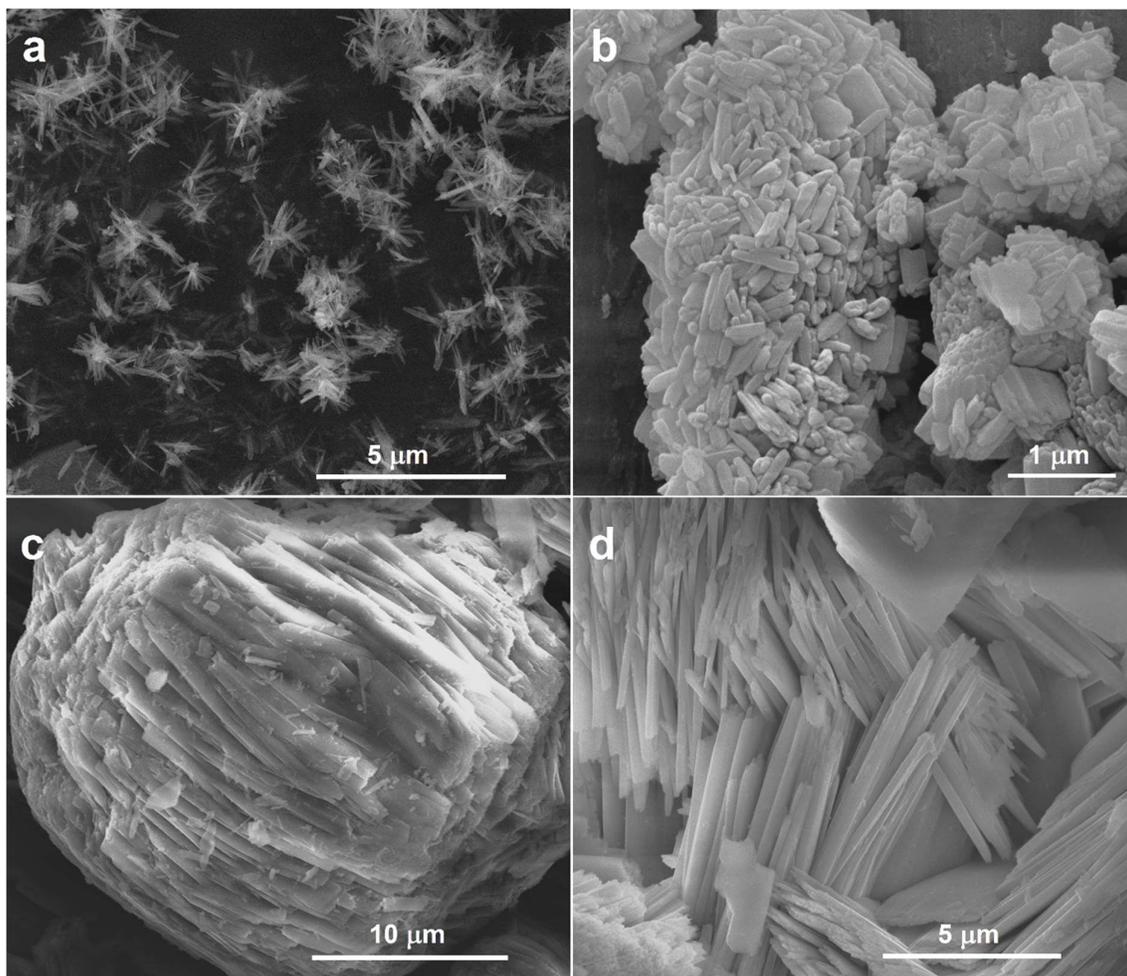
Figure S2. Raman spectrum of ammonium metavanadate.



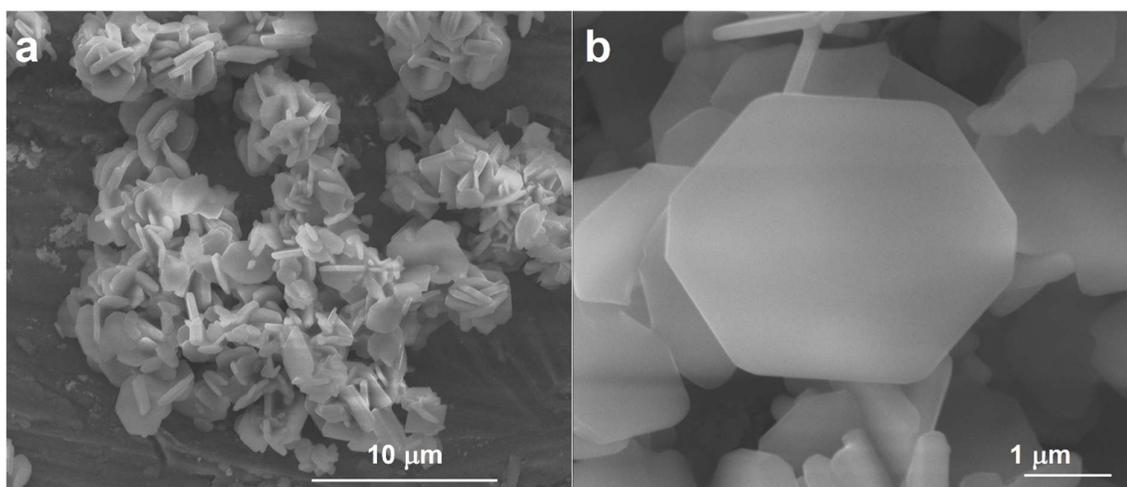
**Figure S3.** SEM images presenting the structure of V<sub>2</sub>O<sub>5</sub> used as a precursor (a,b) and intermediates formed in the reaction of V<sub>2</sub>O<sub>5</sub> with ammonium formate in formamide (c,d), leading to flower-like nanostructural crystals of NH<sub>4</sub>VO<sub>3</sub> (e,f).



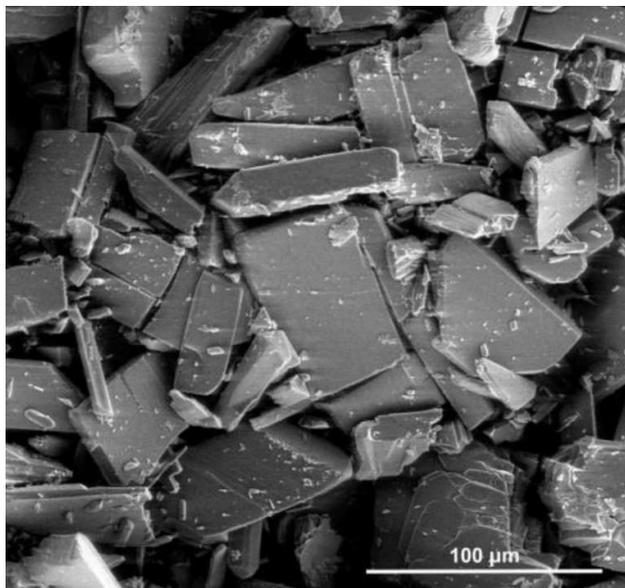
**Figure S4.** SEM images presenting structures of  $\text{NH}_4\text{VO}_3$  obtained from  $\text{V}_2\text{O}_5$  in water using ammonium formate (a,b), ammonium acetate (c,d), and equimolar mixture of ammonium formate and ammonium acetate (e,f).



**Figure S5.** SEM images presenting structures of  $\text{NH}_4\text{VO}_3$  obtained from  $\text{V}_2\text{O}_5$  in formamide using ammonium formate (a), ammonium acetate (b), and equimolar mixture of ammonium formate and ammonium acetate (c,d).



**Figure S6.** SEM images presenting structures of  $\text{NH}_4\text{VO}_3$  obtained in the reaction of  $\text{V}_2\text{O}_5$  and ammonium formate carried out in the 1:1 (v/v) mixture of water and formamide.



**Figure S7.** SEM image of commercial  $\text{NH}_4\text{VO}_3$  used in a hydrothermal synthesis.

#### References:

1. NIST Chemistry WebBook; Available online: <https://webbook.nist.gov/chemistry> (accessed on 27 July 2019).



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