

Modeling and Optimization of Pollutants Removal during Simultaneous Adsorption onto Activated Carbon with Advanced Oxidation in Aqueous Environment

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Table S1. Physicochemical characteristics of activated carbons selected for investigations.

Carbon	S	V _p	D _{ehl}	IN	MBN	DetN	Cont.s.sol.	Cont. ash	pH
WDex	1050	0.95	5	990	36	20.6	1.07	20.64	8.6
WG-12	980	0.89	3	1230	32	30.2	2.48	8.85	9.6
F-200S	720	0.72	7	710	24	12.7	1.16	7.28	4.5
F-200R	800	0.86	6	770	26	15.7	1.74	7.84	7.2

S—specific surface area (m²/g), V_p—pore volume (cm³/g), D_{ehl}—dechlorination half-length (cm), IN—iodine number (mg/g), MBN—methylene number (cm³), Cont.s.sol.—soluble substance content (%), DetN.—detergent number, Cont. ash—ash content (%), pH—pH of the aqueous extract.

In Table S2, Table S3 and Table S4, respectively, determined values of the test probability are given for the correlation between the selected characteristics of the activated carbons, average dye concentration (c_{av}) and the reaction rate constants in the pseudo-first and pseudo-second order equations.

Table S2. Probability (p) value (characteristics of activated carbons).

Variables	D _{ehl}	MBN	IN	DetN	Cont. ash	Cont.s.sol.	S	V _p	pH
D _{ehl}	0.00000	0.03124	0.00433	0.04436	0.02665	0.07567	0.02064	0.02755	0.00493
MBN	-	0.00000	0.01664	0.01982	0.02230	0.08108	0.02064	0.02147	0.02315
IN	-	-	0.00000	0.02755	0.01821	0.06912	0.00873	0.01436	0.00259
DetN	-	-	-	0.00000	0.08385	0.02665	0.03031	0.04763	0.02401
Cont. ash	-	-	-	-	0.00000	0.03609	0.01077	0.00616	0.02938
Cont.s.sol.	-	-	-	-	-	0.00000	0.08808	0.07567	0.05099
S	-	-	-	-	-	-	0.00000	0.00374	0.01218
V _p	-	-	-	-	-	-	-	0.00000	0.01742
pH	-	-	-	-	-	-	-	-	0.00000

where p ≤ 0.05—statistically significant correlations (bold in table).

Table S3. Probability value (average dye concentration).

Variables	c_{av}	
	(Crystal Violet)	(Phenol Red)
H ₂ O ₂	0.026	0.018
D _{eh1}	0.006	0.008
MBN	0.047	0.051
IN	0.006	0.013
S	0.008	0.032

c_{av} —average dye concentration (mg/L). where: $p \leq 0.05$ —statistically significant correlations (bold in table).

Table S4. Probability value (reaction rate constants in the pseudo-first and pseudo-second order equation)

Variables	Crystal Violet		Phenol Red	
	k_1	k_2	k_1	k_2
S	0.02	0.031	0.004	0.003
Con.s.sol.	0.067	0.051	0.074	0.064
MBN	0.023	0.031	0.004	0.003
D _{eh1}	0.022	0.031	0.004	0.003
H ₂ O ₂	0.017	0.029	0.038	0.051

where: $p \leq 0.05$ —statistically significant correlations (bold in table).



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