



# *Article* **Tribological and Thermo-Mechanical Properties of TiO<sup>2</sup> Nanodot-Decorated Ti3C2/Epoxy Nanocomposites**

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**Abstract:** The micromorphology of fillers plays an important role in tribological and mechanical properties of polymer matrices. In this work, a TiO<sub>2</sub>-decorated Ti<sub>2</sub>C<sub>3</sub> (TiO<sub>2</sub>/Ti<sub>3</sub>C<sub>2</sub>) composite particle with unique micro-nano morphology was engineered to improve the tribological and thermo-mechanical properties of epoxy resin. The  $TiO_2/Ti_3C_2$  were synthesized by hydrothermal growth of  $TiO_2$  nanodots onto the surface of accordion-like  $Ti<sub>3</sub>C<sub>2</sub>$  microparticles, and three different decoration degrees (low, medium, high density) of  $TiO<sub>2</sub>/Ti<sub>3</sub>C<sub>2</sub>$  were prepared by regulating the concentration of  $TiO<sub>2</sub>$ precursor solution. Tribological test results indicated that the incorporation of  $TiO<sub>2</sub>/Ti<sub>3</sub>C<sub>2</sub>$  can effectively improve the wear rate of epoxy resin. Among them, the medium density  $\text{TiO}_2/\text{Ti}_3\text{C}_2/\text{epoxy}$ nanocomposites gained a minimum wear rate. This may be ascribed by the moderate  $TiO<sub>2</sub>$  nanodot protuberances on the  $Ti_3C_2$  surface induced a strong mechanical interlock effect between mediumdensity  $TiO<sub>2</sub>/Ti<sub>3</sub>C<sub>2</sub>$  and the epoxy matrix, which can bear a higher normal shear stress during sliding friction. The morphologies of worn surfaces and wear debris revealed that the wear form was gradually transformed from fatigue wear in neat epoxy to abrasive wear in  $TiO_2/T_{13}C_2$ /epoxy nanocomposites. Moreover, the results of thermo-mechanical property indicated that incorporation of  $TiO_2/Ti_3C_2$  also effectively improved the storage modulus and glass transition temperature of epoxy resin.

**Keywords:** Ti<sub>3</sub>C<sub>2</sub> Mxexe; epoxy; polymer-matrix composites; tribological; thermo-mechanical

### **1. Introduction**

Epoxy resin is a kind of typical thermosetting polymer that possesses high mechanical strength, low shrinkage rate, excellent chemical stability, high adhesive strength, etc., and has been widely used in aerospace, petrochemical engineering, automobile components and other fields  $[1-4]$  $[1-4]$ . However, the intrinsic high brittleness, low resistance to thermal shock, and poor wear resistance of epoxy resin, originating from its highly crosslinked network structure, seriously restrict its practical application and development in many industrial fields, especially in the antifriction and high-frequency vibrating components and parts of machinery [\[5](#page-12-2)[–7\]](#page-13-0). Therefore, it is quite necessary to improve wear resistance and dynamic mechanical properties of epoxy resin.

To date, the physical blending with fillers is considered as one of the most commonly used ways to improve the multifunctional performance of polymer, especially wear resistance  $[8-10]$  $[8-10]$ . In this manner, the high anti-friction feature of fillers can be integrated into



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polymer matrix composites, thereby improving their wear resisting property. Previous studies have demonstrated that many kinds of wear resistant fillers, such as ZnO [\[11\]](#page-13-3), PTFE  $[12]$ , graphite  $[13]$ , graphene oxide  $[14]$ , and WS<sub>2</sub>  $[15]$ , can effectively improve the wear resistance of epoxy resin, and the particle size of fillers could influence significantly the wear rate of epoxy resin [\[16\]](#page-13-8). Shi et al. [\[17\]](#page-13-9) further found that the nano-sized TiO<sub>2</sub> particles imparted a better wear resisting property to the epoxy resin when compared to the micro sized  $TiO<sub>2</sub>$  fillers, mainly ascribing to the stronger interfacial interaction between nano-TiO<sub>2</sub> and epoxy matrix. Therefore, it can be speculated that the interfacial interaction of filler-polymer matrix plays an important role in the wear resisting property of polymer composites, as well as other properties.

The micromorphology of micro/nano particles is one of the key factors that influence the interfacial interaction of particle-polymer. Ramezanzadeh et al. [\[18\]](#page-13-10) reported that the different micromorphology of iron oxide particles could induce the distinguishing interfacial interaction between iron oxide and epoxy, thereby resulting in the different mechanical properties. In recent years, some publications have indicated that the micromorphology of fillers could influence the tribological and/or other properties of polymer [\[19](#page-13-11)[,20\]](#page-13-12). Even, the composite particles with unique micromorphology had been synthesized for improving the tribological property of polymer [\[21,](#page-13-13)[22\]](#page-13-14). However, how to design the micromorphology of fillers for imparting the optimal tribological and mechanical property to polymer matrix still remains a challenge.

Inspired by the micro/nano composite structure from previous reports [\[23,](#page-13-15)[24\]](#page-13-16), we plan to design and synthesize a composite particle with unique micro/nano structure by integrating nano-structured particles into micro-structured particles. Herein,  $Ti<sub>3</sub>C<sub>2</sub>$  MXene, a new kind of 2D transition metal carbides firstly discovered by Gogotsi et al. [\[25\]](#page-13-17) in 2011, was selected as the micro-structure particles, and the  $TiO<sub>2</sub>$  was chosen as the nano-structured particles. It has been widely reported that both  $Ti_3C_2$  and  $TiO_2$  particles can improve the tribological and/or mechanical properties of epoxy and other polymers [\[26–](#page-13-18)[30\]](#page-13-19). In this work, we attempted to synthesize the TiO<sub>2</sub> decorated T<sub>13</sub>C<sub>2</sub> (TiO<sub>2</sub>/T<sub>13</sub>C<sub>2</sub>) composite particles that possessed the unique micro/nano structure through hydrothermally growth of  $TiO<sub>2</sub>$ nanodots on  $Ti_3C_2$  microparticles, to improve the tribological and thermo-mechanical properties of epoxy resin. After removal of Al atom layers of  $Ti<sub>3</sub>AIC<sub>2</sub>$  by HF, the accordion-like  $Ti<sub>3</sub>C<sub>2</sub>$  microparticles were prepared. Subsequently, TiO<sub>2</sub> nanodots were hydrothermally grown on the  $Ti_3C_2$  surface to construct the micro/nano structured  $TiO_2/Ti_3C_2$ , and the  $TiO<sub>2</sub>/Ti<sub>3</sub>C<sub>2</sub>/epoxy nanocomposites were fabricated by ultrasound-assisted blending of$ epoxy with  $TiO<sub>2</sub>/Ti<sub>3</sub>C<sub>2</sub>$  (Scheme [1\)](#page-2-0). The micromorphology of different decoration degrees of  $TiO_2/Ti_3C_2$  composite particles and the fracture surface of  $TiO_2/Ti_3C_2$ /epoxy nanocomposites were investigated, and the tribological and thermo-mechanical properties of  $TiO<sub>2</sub>/Ti<sub>3</sub>C<sub>2</sub>/epoxy nanocomposites were tested. Moreover, the underlying wear$ mechanism of  $\text{TiO}_2/\text{Ti}_3\text{C}_2$ /epoxy nanocomposites was analyzed.

<span id="page-2-0"></span>

**Scheme 1.** Illustration for preparation of TiO<sub>2</sub>/Ti<sub>3</sub>C<sub>2</sub>/epoxy nanocomposites.

## **2. Materials and Methods 2. Materials and Methods**

### *2.1. Materials*

Ti<sub>3</sub>AlC<sub>2</sub> powders (99.5%, 400 mesh) were purchased from Kaifa Special Ceramic Technology Co., Ltd. (Beijing, China). Tetrabutyl titanate (TBOT), hydrofluoric acid (HF, Technology Co., Ltd. (Beijing, China). Tetrabutyl titanate (TBOT), hydrofluoric acid (HF, 40%) and absolute ethyl alcohol (AR) were supplied by Sinopharm Chemical Reagent Co., 40%) and absolute ethyl alcohol (AR) were supplied by Sinopharm Chemical Reagent Co., Ltd. (Shanghai, China). Epoxy resin (E-51), methylhexahydrophthalic anhydride Ltd. (Shanghai, China). Epoxy resin (E-51), methylhexahydrophthalic anhydride (MHHPA) and tetraethyl ammonium bromide (TEAB) were provided by Haining Hailong Chemical Co., Ltd. (Jiaxing, China).

## *2.2. Preparation of Ti3C2 Microparticles 2.2. Preparation of Ti3C<sup>2</sup> Microparticles*

 $Ti<sub>3</sub>AIC<sub>2</sub>$  powders were added into HF solution (40%) at a ratio of 0.1 g per milliliter. After stirring vigorously for 12 h at room temperature (RT), the sediments were centrifugally gally collected at a speed of 4000 rpm, followed by washed with deionized water until the collected at a speed of 4000 rpm, followed by washed with deionized water until the pH value was approximate to neutrality. With the wet powders were dried for 12 h at 60  $\degree$ C in a vacuum drying oven, the  $\text{Ti}_3\text{C}_2$  microparticles were obtained.

### *2.3. Preparation of TiO<sup>2</sup> Decorated Ti3C<sup>2</sup> (TiO2/Ti3C2) Composite Particles*

Ti<sub>3</sub>C<sub>2</sub> microparticles (1 g) were added into a mixed solution of deionized water and absolute ethyl alcohol (30 mL, volume = 1:1) to form a  $Ti<sub>3</sub>C<sub>2</sub>$  dispersion solution. Simultaneously, the desired molar mass of TBOT (0.25, 0.5, and 1.0 mol) were added into 45 mL of absolute ethyl alcohol to prepare the TiO<sub>2</sub> precursor solution. Afterwards, the TiO<sub>2</sub> precursor solution was added dropwise into the Ti<sub>3</sub>C<sub>2</sub> dispersion solution with stirring vigorously, followed by transferred into a 100 mL of Teflon-lined autoclave and kept at 180 °C for 24 h. After cooled to RT, the sediments were centrifugally collected at a speed of 4000 rpm, and dried at 60 °C for 12 h in a vacuum drying oven, the TiO<sub>2</sub>/Ti<sub>3</sub>C<sub>2</sub> composite particles were obtained. According to the molar mass of TBOT, the obtained  $TiO_2/Ti_3C_2$  were named as 0.25 M, 0.5 M, and 1.0 M  $TiO_2/Ti_3C_2$ .

### *2.4. Preparation of TiO2/Ti3C2/Epoxy Nanocomposites*

Firstly, the desired mass fraction of  $TiO<sub>2</sub>/Ti<sub>3</sub>C<sub>2</sub>$  were ultrasonically dispersed into absolute ethyl alcohol, followed by epoxy resin was mixed with  $TiO<sub>2</sub>/Ti<sub>3</sub>C<sub>2</sub>$  dispersion solution for 1 h under ultrasound. With stirring vigorously, the mixed solution of  $TiO<sub>2</sub>/Ti<sub>3</sub>C<sub>2</sub>$ 

and epoxy was reduced pressure distilled at 60  $\degree$ C until the absolute ethyl alcohol was fully evaporated. The MHHPA containing TEAB was then added into the mixture of  $TiO<sub>2</sub>/Ti<sub>3</sub>C<sub>2</sub>$ and epoxy, followed by stirring for 1 h at RT in a vacuum. The  $TiO_2/T_{13}C_2$ /epoxy mixture was subsequently poured into stainless steel mold and cured for 2 h at 120 ◦C, the  $TiO<sub>2</sub>/Ti<sub>3</sub>C<sub>2</sub>$ /epoxy nanocomposites were finally obtained. Herein, the mass ratio of epoxy resin, MHHPA and TEAB was set to 100:80:0.3, and the mass fraction of  $TiO_2/Ti_3C_2$ (relative to nanocomposite) was set to 0.25%, 0.5%, 0.75%, and 1%.

#### *2.5. Characterization and Measurement*

The surface morphologies of the samples were investigated by field-emission scanning electron microscope (FE-SEM, Hitachi S4800, Tokyo, Japan). The phase composition of the samples was analyzed by X-ray diffractometer (XRD, Bruker D8, Karlsruhe, Germany). Fourier transform infrared (FTIR) spectra of  $Ti<sub>3</sub>AlC<sub>2</sub>$ ,  $Ti<sub>3</sub>C<sub>2</sub>$  and  $Ti<sub>0</sub> / Ti<sub>3</sub>C<sub>2</sub>$  were recorded at room temperature by an infrared spectrometer (Nicolet 5700, Madison, WI, USA) with the range from 400 cm<sup>-1</sup> to 4000 cm<sup>-1</sup>. The densities of TiO<sub>2</sub>/Ti<sub>3</sub>C<sub>2</sub>/epoxy nanocomposites were measured with a density tester (GP-300E, Guanwei Instrument, Shanghai, China), and the hardness of  $TiO<sub>2</sub>/Ti<sub>3</sub>C<sub>2</sub>$ /epoxy nanocomposites were examined by Vickers microhardness tester at a load of 0.5 N and loading time of 15 s (HV-1000, Shangcai Instrument, Shanghai, China). For the testing of tribological property, the  $TiO_2/T_{13}C_2$ /epoxy nanocomposites were cut into strip-shaped specimens (16 mm  $\times$  4 mm  $\times$  4 mm), and specimen surfaces were polished with 1200 mesh SiC abrasive paper. After alcohol washed and dried at 60  $\degree$ C, the tribological testing of specimens was performed by a vertical universal friction and wear testing machine (MM-W18, Shijin Co. Ltd., Jinan, China) at a rotating speed of 780 rpm with the load of 4 N and 8 N for 1 h duration under dry friction condition. Six replications were carried out for each nanocomposite. The friction coefficients were evaluated as the computer-calculated average value. The average wear rates were calculated according to the formula:  $W = V/(F \times L)$ , where the W, V, F, and L are the wear rate, wear volume, applied load, and sliding distance, respectively [\[31\]](#page-13-20). Thermo-mechanical properties were measured using DMA 850, (TA Instruments, Newcastle, DE, USA). The  $TiO<sub>2</sub>/Ti<sub>3</sub>C<sub>2</sub>/epoxy$ nanocomposites were cut into strip-shaped specimens (30 mm  $\times$  10 mm  $\times$  4 mm) to match with machine. A temperature scan was conducted from 40 to 190  $\degree$ C with a heating rate of 5 ◦C/min at a frequency of 1 Hz and a strain of 0.05% in a single cantilever mode. The storage modulus and tan δ (the ratio of loss modulus to storage modulus) were obtained from dynamic mechanical analysis, and the glass transition temperature (Tg) was gained from the peak value of tan δ.

#### **3. Results**

### *3.1. Surface Morphology of TiO2/Ti3C<sup>2</sup> Microparticles*

The Ti<sub>3</sub>C<sub>2</sub> microparticles were prepared by HF etching, and the TiO<sub>2</sub>/T<sub>i3</sub>C<sub>2</sub> composite particles were synthesized via hydrothermal method. As shown in Figure [1A](#page-4-0), the original Ti<sub>3</sub>AlC<sub>2</sub> microparticles exhibit irregular morphology, with an approximate size of 10  $\mu$ m, and the selected area magnified image (Figure [1B](#page-4-0)) clearly displays the tightly stacked layer structure of  $Ti<sub>3</sub>AIC<sub>2</sub>$  microparticles. After etched with HF solution, lots of interlayered gaps are produced in etched  $Ti<sub>3</sub>AIC<sub>2</sub>$  microparticles (Figure [1C](#page-4-0)), implying the removal of Al atom layers in T<sub>13</sub>AlC<sub>2</sub> and the possible formation of T<sub>13</sub>C<sub>2</sub> microparticles [\[32\]](#page-13-21). After hydrothermally reacted with TiO<sub>2</sub> precursor solution, the surface of T<sub>i3</sub>C<sub>2</sub> is covered with large amounts of nanodots (Figure [1D](#page-4-0)–F), which maybe suggests that the  $TiO<sub>2</sub>$  are successfully grown on the surface of  $Ti<sub>3</sub>C<sub>2</sub>$ . It can be seen that the density of nanodots on  $Ti<sub>3</sub>C<sub>2</sub>$  surface is gradually increase with increasing the molar mass of TOBT, especially, the obvious aggregated nanodots are found on the surface of  $Ti<sub>3</sub>C<sub>2</sub>$  when the molar mass of TOBT reaches 1.0 mol.

<span id="page-4-0"></span>

Figure 1. SEM images of Ti<sub>3</sub>AlC<sub>2</sub> (A,B), Ti<sub>3</sub>C<sub>2</sub> (C) and TiO<sub>2</sub>/Ti<sub>3</sub>C<sub>2</sub> synthesized by different concentration of TiO<sub>2</sub> precursor solution ((**D**) 0.25 mol; (**E**) 0.5 mol; (**F**) 1.0 mol). solution ((**D**) 0.25 mol; (**E**) 0.5 mol; (**F**) 1.0 mol).

## *3.2. XRD and FTIR Analysis of TiO2/Ti3C2 Composite Particles 3.2. XRD and FTIR Analysis of TiO2/Ti3C<sup>2</sup> Composite Particles*

The phase composition of as-prepared samples was analyzed by XRD pattern. It can be seen from Figure [2A](#page-4-1) that three strong diffraction peaks are found in the pattern of Ti<sub>3</sub>AlC<sub>2</sub>, with the peaks at 9.6°, 39.1° and 41.9° are well indexed to the (002), (104), and (105) lattice plane of hexagonal Ti<sub>3</sub>AlC<sub>2</sub> (JCPDS No. 52-0875), respectively [\[33\]](#page-13-22). After HF etching, the characteristic (104) diffraction peak of Ti<sub>3</sub>AlC<sub>2</sub> has thoroughly disappeared. Additionally, the (002) diffraction peak is broadened, and shift to the lower angles com-Additionally, the (002) diffraction peak is broadened, and shift to the lower angles com-pared to their original location, which indicate the conversion of Ti<sub>3</sub>AlC<sub>2</sub> to Ti<sub>3</sub>C<sub>2</sub> [\[34\]](#page-14-0). After hydrothermal reaction, three new diffraction peaks are observed in the TiO<sub>2</sub>/Ti<sub>3</sub>C<sub>2</sub> composite particles, with peaks at 25.3°, 37.9°, and 48.0° assigned to (101), (004), and (200) lattice plane of anatase (JCPDS no. 21-1272), respectively [35]. M[ore](#page-14-1) importantly, the intensity of (101) diffraction peak at 25.3° markedly enhances with the increase of molar mass of TBOT (Figure 2B), [whi](#page-4-1)ch indicates that the density of  $TiO<sub>2</sub>$  nanodots can be regulated with TiO<sub>2</sub> precursor solution. FTIR spectra show that four obvious adsorption bands are observed in Ti<sub>3</sub>C<sub>2</sub> and TiO<sub>2</sub>/Ti<sub>3</sub>C<sub>2</sub>, with 3411 cm<sup>−1</sup>, 3180 cm<sup>−1</sup>, 1635 cm<sup>−1</sup>, and 1322 cm<sup>-1</sup> corresponding to the stretching vibration of hydroxyl group (–OH) of H<sub>2</sub>O, stretching vibration of Ti–OH, blending vibration of  $-OH$  of  $H_2O$  and stretching vibration of covalent C–F bond of Ti<sub>3</sub>C<sub>2</sub>, respectively [\[27,](#page-13-23)[36\]](#page-14-2). It is worth noting that, compared with  $Ti_3C_2$ , a strong adsorption band at 542 cm<sup>-1</sup> is found in the spectra of  $TiO_2/Ti_3C_2$ , which is assigned to the stretching vibration of Ti–O bond of TiO<sub>2</sub> [\[37\]](#page-14-3).

obvious aggregated nanodots are found on the surface of Ti3C2 when the molar mass of

<span id="page-4-1"></span>

**Figure 2.** XRD patterns (A,B) and FTIR spectra (C) of  $T_{13}$ AlC<sub>2</sub>,  $T_{13}C_2$ , and  $T_{10}C_2/T_{13}C_2$ . Figure 2. XRD patterns  $(A,B)$  and FTIR spectra  $(C)$  of Ti<sub>3</sub>AlC<sub>2</sub>, Ti<sub>3</sub>C<sub>2</sub>, and TiO<sub>2</sub>/Ti<sub>3</sub>C<sub>2</sub>.

## 3.3. Determining the Optimal Mass Fraction of TiO<sub>2</sub>/Ti<sub>3</sub>C<sub>2</sub> through Testing the Tribological *Properties of 0.25 M TiO2/Ti3C2/Epoxy Nanocomposite Properties of 0.25 M TiO2/Ti3C2/Epoxy Nanocomposite*

To determine the optimal mass fraction of TiO<sub>2</sub>/Ti<sub>3</sub>C<sub>2</sub> composite particles, the tribological property of TiO<sub>2</sub>/Ti<sub>3</sub>C<sub>2</sub>/epoxy nanocomposite with different content was investigated. As shown [in](#page-5-0) Figure 3, both the friction coefficient and wear rate of all the TiO<sub>2</sub>/Ti<sub>3</sub>C<sub>2</sub>/epoxy nanocomposites show a trend of first decreasing and then increasing when the mass fraction of TiO<sub>2</sub>/Ti<sub>3</sub>C<sub>2</sub> is increased from 0% to 1%. Among them, the TiO<sub>2</sub>/Ti<sub>3</sub>C<sub>2</sub>/epoxy nanocomposite with 0.5% mass fraction presents the lowest friction coefficient and minimum wear rate. Herein, the friction coefficient of TiO<sub>2</sub>/Ti<sub>3</sub>C<sub>2</sub>/epoxy nanocomposites at 0.5% mass fraction are determined to 0.60 for  $Ti_3C_2$ /epoxy, 0.66 for 0.25 M  $TiO_2/Ti_3C_2$ /epoxy, 0.71 for 0.5 M TiO<sub>2</sub>/Ti<sub>3</sub>C<sub>2</sub>/epoxy, 0.81 for 1.0 M TiO<sub>2</sub>/Ti<sub>3</sub>C<sub>2</sub>/epoxy, and the wear rate of TiO<sub>2</sub>/Ti<sub>3</sub>C<sub>2</sub>/epoxy nanocomposites at 0.5% mass fraction are calculated to 7.66  $\times$ 0.71 for 0.5 M TiO<sub>2</sub>/Ti<sub>3</sub>C<sub>2</sub>/epoxy, 0.81 for 1.0 M TiO<sub>2</sub>/Ti<sub>3</sub>C<sub>2</sub>/epoxy, and the wear rate of TiO<sub>2</sub>/Ti<sub>3</sub>C<sub>2</sub>/epoxy nanocomposites at 0.5% mass fraction are calculated to 7.66 × 10<sup>-14</sup> m<sup>3</sup>/(N·m) for Ti<sub>3</sub>C<sub>2</sub>/epox  $3.37 \times 10^{-14}$  m<sup>3</sup>/(N·m) for 0.5 M TiO<sub>2</sub>/Ti<sub>3</sub>C<sub>2</sub>/epoxy, 9.26 × 10<sup>-14</sup> m<sup>3</sup>/(N·m) for 1.0 M  $3.37 \times 10^{-14}$  m<sup>3</sup>/(N·m) for 0.5 M TiO<sub>2</sub>/Ti<sub>3</sub>C<sub>2</sub>/epoxy, 9.26 × 10<sup>-14</sup> m<sup>3</sup>/(N·m) for 1.0 M TiO<sub>2</sub>/Ti<sub>3</sub>C<sub>2</sub>/epoxy. Therefore, for the TiO<sub>2</sub>/Ti<sub>3</sub>C<sub>2</sub>/epoxy nanocomposites incorporated with different decoration density of TiO<sub>2</sub>/Ti<sub>3</sub>C<sub>2</sub>, the mass fraction of TiO<sub>2</sub>/Ti<sub>3</sub>C<sub>2</sub> composite particles is selected as 0.5% for subsequent studies.

<span id="page-5-0"></span>

Figure 3. (A)Friction coefficient and (B) wear rate of  $TiO_2/T_{13}C_2$ /epoxy nanocomposites with different mass friction under a normal load of 8 N. a normal load of 8 N.

## 3.4. Fracture Surface Analysis of TiO<sub>2</sub>/Ti<sub>3</sub>C<sub>2</sub>/Epoxy Nanocomposites with Different TiO<sub>2</sub>/Ti<sub>3</sub>C<sub>2</sub>

To understand the internal microstructure of  $0.5\%$  TiO<sub>2</sub>/Ti<sub>3</sub>C<sub>2</sub>/epoxy nanocomposites, the morphology of fracture surfaces of  $TiO<sub>2</sub>/Ti<sub>3</sub>C<sub>2</sub>$ /epoxy nanocomposites incorporated with different  $TiO<sub>2</sub>/Ti<sub>3</sub>C<sub>2</sub>$  was investigated. As shown in Figure [4A](#page-6-0), large numbers of riverlike stripes are clearly observed on the fracture surface of neat epoxy, which are typical characteristics of brittle thermosetting polymer [\[38\]](#page-14-4). For Ti<sub>3</sub>C<sub>2</sub>/epoxy nanocomposite (Figure [4B](#page-6-0)), the stripes on the fracture surface decrease, and simultaneously some loose particles are emerged on the fracture surface. After incorporation of 0.25 M  $TiO_2/Ti_3C_2$ composite particles into epoxy matrix, the stripes on the fracture surface of composite greatly decrease, but the loose particles are still found on the fracture surface (Figure [4C](#page-6-0)), which maybe implies the weak interfacial interaction between 0.25 M  $TiO<sub>2</sub>/Ti<sub>3</sub>C<sub>2</sub>$  and the epoxy matrix. When incorporated with  $0.5$  M TiO<sub>2</sub>/Ti<sub>3</sub>C<sub>2</sub>, it seems that some particles are embedded in the epoxy matrix, and the loose particles can hardly be found on the fracture surface (Figure [4D](#page-6-0)), maybe ascribing to the strong interfacial interaction between  $0.5$  M TiO<sub>2</sub>/Ti<sub>3</sub>C<sub>2</sub> and epoxy matrix. However, a large number of particles are observed on the fracture surface of 1.0 M  $TiO<sub>2</sub>/Ti<sub>3</sub>C<sub>2</sub>/epoxy nanocomposite$  (Figure [4E](#page-6-0)), suggesting the poor interfacial interaction between  $TiO<sub>2</sub>/Ti<sub>3</sub>C<sub>2</sub>$  and matrix. In addition, the density

and hardness of TiO<sub>2</sub>/Ti<sub>3</sub>C<sub>2</sub>/epoxy nanocomposites (0.5% mass fraction) are gradually enhanced with the increase of decorated TiO<sub>2</sub> (Figure [4F](#page-6-0)), with the average density of 1.24, 1.22, 1.22, 1.23, 1.23, 1.23, 1.23, 1.21, 1.21, 1.24, 1.24, 1.24, 1.24, 1.21, 1.24, 1.24, 1.24, 1.24, 1.24, 1.24, 1.24, 1.24, 1 1.24, 1.21, 1.23, 1.29, 1.33 g·cm<sup>-3</sup> and average hardness of 135, 139, 144, 153, 156 Hv0.5 for neat epoxy, Ti<sub>3</sub>C<sub>2</sub>/epoxy, 0.25 M TiO<sub>2</sub>/Ti<sub>3</sub>C<sub>2</sub>/epoxy, 0.5 M TiO<sub>2</sub>/Ti<sub>3</sub>C<sub>2</sub>/epoxy, and 1.0 M  $TiO<sub>2</sub>/Ti<sub>3</sub>C<sub>2</sub>/epoxy nanocomposites.$ interfacial interaction between TiO2/Ti3C2 and matrix. In addition, the density and hardand nardness of  $10<sub>2</sub>$ / $1<sub>3</sub><sub>2</sub>$ /epoxy nanocomposites (0.5% mass fraction) are gradually

TiO2/Ti3C2 and epoxy matrix. However, a large number of particles are observed on the

<span id="page-6-0"></span>

Figure 4. SEM images of fracture surfaces of neat epoxy (A),  $Ti_3C_2$ /epoxy (B), 0.25 M  $TiO_2/Ti_3C_2$ /epoxy (C), 0.5 M  $TiO_2/Ti_3C_2$ /epoxy (D), 1.0 M  $TiO_2/Ti_3C_2$ /epoxy (E) nanocomposites at 0.5% mass fraction. Density and hardness (F) of epoxy and TiO2/Ti3C2/epoxy nanocomposites at 0.5% mass fraction. neat epoxy and TiO2/Ti3C2/epoxy nanocomposites at 0.5% mass fraction.

## 3.5. Surface Morphology of TiO<sub>2</sub>/Ti<sub>3</sub>C<sub>2</sub>/Epoxy Nanocomposites with Different TiO<sub>2</sub>/Ti<sub>3</sub>C<sub>2</sub> before *Friction Tests Friction Tests*

The differences in surface irregularity of materials have an important influence on The differences in surface irregularity of materials have an important influence on the results of friction tests. So, the surface morphology of  $TiO<sub>2</sub>/Ti<sub>3</sub>C<sub>2</sub>/epoxy samples$ were investigated before friction tests were carried out. As shown in Figure 5, th[e s](#page-7-0)urface morphologies of all the samples seem to be similar, ascribing to the same treatment conditions. Additionally, the neat epoxy and  $TiO<sub>2</sub>/Ti<sub>3</sub>C<sub>2</sub>/epoxy$  nanocomposites display a homogeneous and regular surface morphology, without apparent protuberant or sunk part, which indicates that the effect of surface irregularity of  $TiO<sub>2</sub>/Ti<sub>3</sub>C<sub>2</sub>/epoxy$  nanocomposites on results of friction tests is negligible.

## *3.6. Tribological Properties of TiO2/Ti3C2/Epoxy Nanocomposites with Different TiO2/Ti3C<sup>2</sup>*

The tribological property of  $TiO<sub>2</sub>/Ti<sub>3</sub>C<sub>2</sub> / epoxy$  nanocomposites was evaluated by measuring their friction coefficient and wear rate. As shown in Figure [6A](#page-8-0), the neat epoxy exhibits the highest friction coefficient of 1.02 under a normal load of 4 N, and the friction coefficient dramatically decreases to 0.67 after incorporation of 0.5%  $Ti<sub>3</sub>C<sub>2</sub>$  into epoxy matrix. When the  $TiO<sub>2</sub>/Ti<sub>3</sub>C<sub>2</sub>$  composite particles were used as the fillers, the friction coefficient of  $\text{TiO}_2/\text{Ti}_3\text{C}_2$ /epoxy nanocomposites increases gradually with increasing the decorated  $TiO<sub>2</sub>$ , and the friction coefficient of 0.25 M  $TiO<sub>2</sub>/Ti<sub>3</sub>C<sub>2</sub>/epoxy, 0.5 M TiO<sub>2</sub>/Ti<sub>3</sub>C<sub>2</sub>/epoxy,$ and 1.0 M  $TiO<sub>2</sub>/Ti<sub>3</sub>C<sub>2</sub>$ /epoxy are determined to 0.74, 0.76, and 0.88, respectively. Furthermore, the friction coefficient of neat epoxy,  $Ti_3C_2$ /epoxy, 0.25 M  $TiO_2/Ti_3C_2$ /epoxy, 0.5 M  $\text{TiO}_2/\text{Ti}_3\text{C}_2/\text{epoxy}$ , and 1.0 M  $\text{TiO}_2/\text{Ti}_3\text{C}_2/\text{epoxy}$  under the normal load of 8 N

are calculated to 0.93, 0.60, 0.68, 0.72, and 0.81, respectively, which is lower than that under the normal load of 4 N. This can be ascribed to the negative exponential relation between applied load and friction coefficient that can be expressed as  $\mu = kN^{n-1}$  [\[39\]](#page-14-5), where the  $\mu$ , k, N are the friction coefficient, constant, and applied load, respectively, and n is a constant between 2/3 and 1, depending on the amount of interaction between elastic and plastic deformation. The wear rate of neat epoxy and  $TiO<sub>2</sub>/Ti<sub>3</sub>C<sub>2</sub>/epoxy nanocom-$ posites is shown in Figure [6B](#page-8-0). It can be seen that the wear rate of neat epoxy is 15.02  $\times$  $10^{-14}$  m<sup>3</sup>/(N·m) under a normal load of 4 N, followed by decreases sharply to a value of 4.87 × 10<sup>-14</sup> m<sup>3</sup>/(N·m) after incorporation of only 0.5% mass fraction of T<sub>13</sub>C<sub>2</sub> microparticles. With the incorporation of  $\overline{0.25 \text{ M}}$  TiO<sub>2</sub>/Ti<sub>3</sub>C<sub>2</sub> into epoxy matrix, the wear rate of TiO<sub>2</sub>/Ti<sub>3</sub>C<sub>2</sub>/epoxy nanocomposite further decreases to 3.7 $\frac{1}{4} \times 10^{-14}$  m<sup>3</sup>/(N·m), and gained the minimum wear rate of  $1.75 \times 10^{-14}$  m<sup>3</sup>/(N·m) in 0.5 M TiO<sub>2</sub>/Ti<sub>3</sub>C<sub>2</sub>/epoxy nanocomposite. When the normal load is enhanced to 8 N, the wear rate of all the samples is obviously increased, and calculated to 21.93  $\times$  10<sup>-14</sup> m<sup>3</sup>/(N·m) for neat epoxy, 7.17  $\times$  $10^{-14}$  m<sup>3</sup>/(N·m) for Ti<sub>3</sub>C<sub>2</sub>/epoxy,  $4.99 \times 10^{-14}$  m<sup>3</sup>/(N·m) for 0.25 M TiO<sub>2</sub>/Ti<sub>3</sub>C<sub>2</sub>/epoxy,  $2.67 \times 10^{-14}$  m<sup>3</sup>/(N·m) for 0.5 M TiO<sub>2</sub>/Ti<sub>3</sub>C<sub>2</sub>/epoxy and  $8.97 \times 10^{-14}$  m<sup>3</sup>/(N·m) for 1.0 M  $TiO<sub>2</sub>/Ti<sub>3</sub>C<sub>2</sub>/epoxy.$  No matter the normal load is 4 N or 8 N, 0.5 M TiO<sub>2</sub>/Ti<sub>3</sub>C<sub>2</sub>/epoxy nanocomposite gains the minimum wear rate, which may be attributed that the medium density TiO<sub>2</sub> nanodots grown on Ti<sub>3</sub>C<sub>2</sub> provide abundant anchors for the strong interaction between TiO<sub>2</sub>/Ti<sub>3</sub>C<sub>2</sub> composite particles and epoxy matrix [\[40\]](#page-14-6). Besides, the wear rate of all the samples under the normal load of  $8 N$  is always higher than that under the normal load of 4 N, mainly ascribing that the enhanced normal load produced a higher shear stress, which resulted in the detachment of more epoxy particles or fillers from the surface of surface is surface it of  $\frac{1}{2}$ epoxy nanocomposites.

<span id="page-7-0"></span>

Figure 5. SEM images of neat epoxy (A),  $Ti_3C_2$ /epoxy (B), 0.25 M  $TiO_2/Ti_3C_2$ /epoxy (C), 0.5 M  $TiO_2/Ti_3C_2$ /epoxy (D), 1.0 M TiO<sub>2</sub>/Ti<sub>3</sub>C<sub>2</sub>/epoxy (E) nanocomposites at 0.5% mass fraction after surface polishing treatment with SiC abrasive paper.

<span id="page-8-0"></span>

**Figure 6.** Friction coefficient (A) and wear rate (B) of neat epoxy,  $Ti_3C_2/epoxy$ , 0.25 M  $TiO_2/Ti_3C_2/epoxy$ , 0.5 M  $\rm TiO_2/Ti_3C_2/epoxy$ , and  $\rm 1.0$  M  $\rm TiO_2/Ti_3C_2/epoxy$  nanocomposites at 0.5% mass fraction under the normal load of 4 N and 8 N.

## 3.7. Worn Surface Analysis of TiO<sub>2</sub>/Ti<sub>3</sub>C<sub>2</sub>/Epoxy Nanocomposites with Different TiO<sub>2</sub>/Ti<sub>3</sub>C<sub>2</sub>

To understand the wear mechanism, the worn surfaces along with wear debris of To understand the wear mechanism, the worn surfaces along with wear debris of neat epoxy and  $\text{TiO}_2/\text{Ti}_3\text{C}_2$ /epoxy nanocomposites under the normal load of 8 N were vestigated. As shown in Figure 7A,B, the neat e[po](#page-9-0)xy displays a relatively rough worn investigated. As shown in Figure 7A,B, the neat epoxy displays a relatively rough worn surface with lots of ploughed furrows, suggesting the fatigue wear of neat epoxy, which surface with lots of ploughed furrows, suggesting the fatigue wear of neat epoxy, which is the typical wear form of brittle polymer [\[41\]](#page-14-7). During the sliding friction, the ploughed furrows can easily induce the detachment of small pieces of epoxy from the bulk epoxy furrows can easily induce the detachment of small pieces of epoxy from the bulk epoxy under the shear stress, so large numbers of small-sized wear debris are observed on the steel counterpart surface (Figure [7C](#page-9-0)), which results in the serious wear behavior of neat epoxy. After incorporation of 0.5%  $Ti_3C_2$ , the amounts of surface ploughed furrows of  $Ti_3C_2$ /epoxy nanocomposite remarkably decrease (Figure [7D](#page-9-0),E), and the worn surface becomes smooth. In addition, the size of wear debris of  $\text{Ti}_3\text{C}_2$ /epoxy nanocomposite is obviously bigger than that of neat epoxy (Figure [7F](#page-9-0)), implying the improved fatigue  $\sum_{i=1}^{n} a_i$ wear in Ti<sub>3</sub>C<sub>2</sub>/epoxy nanocomposite. For 0.25 M TiO<sub>2</sub>/Ti<sub>3</sub>C<sub>2</sub>/epoxy nanocomposite, the sinoodiness of its worn surface is further enhanced, and noticeable ploughed traces have scarcely been found (Figure [7G](#page-9-0),H). Simultaneously, the large amounts of flake-like wear debris are collected from counterpart surface of  $0.25$  M TiO<sub>2</sub>/Ti<sub>3</sub>C<sub>2</sub>/epoxy nanocomposite (Figure [7I](#page-9-0)), which implies the improvement of wear resisting property of epoxy resin. The which implies the improvement of wear resisting property of epoxy resin. The improved improved wear resisting property mainly attributed that the original brittle "epoxy to steel" expected that resisting property mainly attributed that the original britte "epoxy" to steel contact wear is partly substituted with tough " $TiO_2/Ti_3C_2$  to steel" contact wear [\[42\]](#page-14-8). When the 0.5 M  $TiO_2/Ti_3C_2$  composite particles are incorporated into epoxy matrix, the worn  $\frac{1}{2}$  composite particles are interpreted into excellent surface of nanocomposite becomes quite smooth (Figure [7J](#page-9-0),K), indicating the excellent wear resisting property of  $0.5$  M  $TiO<sub>2</sub>/Ti<sub>3</sub>C<sub>2</sub>$ /epoxy nanocomposite. Additionally, some tiny particles are presented on the worn surface, which suggests that the transformation of wear form of 0.5 M TiO<sub>2</sub>/Ti<sub>3</sub>C<sub>2</sub>/epoxy nanocomposite from fatigue wear to abrasive wear may be made. Moreover, the flake-like wear debris with large size are collected from the counterpart surface (Figure [7L](#page-9-0)). This can be explained that, on one hand, the moderate TiO<sub>2</sub> nanodots grown on TiO<sub>2</sub>/Ti<sub>3</sub>C<sub>2</sub> act as the anchors to form strong mechanical interlock between 0.5 M TiO<sub>2</sub>/Ti<sub>3</sub>C<sub>2</sub> and epoxy matrix, which can bear a higher normal smoothness of its worn surface is further enhanced, and noticeable ploughed traces have shear stress during sliding friction; on the other hand, due to the formation of large size of flake-like wear debris, the "composite to flake-like wear debris of composite" contact wear may be established, which can reduce the wear rate of  $0.5$  M TiO<sub>2</sub>/Ti<sub>3</sub>C<sub>2</sub>/epoxy nanocomposite greatly. For the 1.0 M TiO<sub>2</sub>/Ti<sub>3</sub>C<sub>2</sub>/epoxy nanocomposite, a large number of tiny particles are presented on the worn surface (Figure [7M](#page-9-0),N), and the size of wear debris of 1.0 M TiO<sub>2</sub>/Ti<sub>3</sub>C<sub>2</sub>/epoxy nanocomposite are much smaller than that of 0.5 M  $TiO<sub>2</sub>/Ti<sub>3</sub>C<sub>2</sub>$ /epoxy nanocomposite (Figure [7O](#page-9-0)), which reveals that the dominant wear form

of 1.0 M TiO $_2$ /Ti $_3$ C $_2$ /epoxy nanocomposite is abrasive wear, instead of fatigue wear. These results demonstrated that the micromorphology of  $TiO<sub>2</sub>/Ti<sub>3</sub>C<sub>2</sub>$  composite particles can strongly influence the wear behavior of epoxy nanocomposites. influence the wear behavior of epoxy nanocomposites.  $n \geq 7$ , which reveals that the dominant wear form of  $\frac{1}{2}$ .  $T_{10}$  To  $M_{11}$ O<sub>2</sub>/ $T_{13}$ O<sub>2</sub>/epoxy nanocomposite is abrasive wear, instead of fatigue wear. These

sented on the worn surface (Figure 7M,N), and the size of wear debris of 1.0 M

<span id="page-9-0"></span>

Figure 7. SEM images of the worn surfaces of neat epoxy (A,B), Ti<sub>3</sub>C<sub>2</sub>/epoxy (D,E), 0.25 M TiO2/Ti3C2/epoxy (**G**,**H**), 0.5 M TiO2/Ti3C2/epoxy (**J**,**K**), and 1.0 M TiO2/Ti3C2/epoxy (**M**,**N**) nanocomposites at 0.5% mass fraction under a normal load of 8 N. SEM images of wear debris of neat epoxy (**C**), Ti<sub>3</sub>C<sub>2</sub>/epoxy (**F**), 0.25 M TiO<sub>2</sub>/Ti<sub>3</sub>C<sub>2</sub>/epoxy (**I**), 0.5 M TiO<sub>2</sub>/Ti<sub>3</sub>C<sub>2</sub>/epoxy (**L**), and 1.0 M TiO2/Ti3C2/epoxy (**O**) nanocomposites at 0.5% mass fraction under a normal load of 8 N.

### *3.8. Dynamic Mechanical Analysis of Ti3C2/Epoxy Nanocomposites with Different TiO2/Ti3C<sup>2</sup>*

To further analyze the thermo-mechanical properties of neat epoxy and  $\text{TiO}_2/\text{Ti}_3\text{C}_2/$ epoxy nanocomposites, dynamic mechanical analysis was performed. It can be seen from Figure [8A](#page-11-0) that the storage modulus of  $TiO<sub>2</sub>/Ti<sub>3</sub>C<sub>2</sub>$ /epoxy nanocomposites in the glassy state is gradually enhanced with the increase of decorated TiO<sub>2</sub> nanodots of TiO<sub>2</sub>/T<sub>i3</sub>C<sub>2</sub> composite particles, followed by reach to the maximum value at  $0.5$  M  $\text{TiO}_2/\text{Ti}_3\text{C}_2/\text{epoxy}$ nanocomposite and decrease in 1.0 M  $TiO<sub>2</sub>/Ti<sub>3</sub>C<sub>2</sub>/epoxy$  nanocomposite. Herein, the storage modulus of  $Ti_3C_2$ , 0.25 M  $TiO_2/T_{13}C_2$ /epoxy, 0.5 M  $TiO_2/T_{13}C_2$ /epoxy, and 1.0 M  $TiO<sub>2</sub>/Ti<sub>3</sub>C<sub>2</sub>/epoxy at 40 °C are determined to 2481.3 MPa, 2543.4 MPa, 2645.9 MPa,$ 2388.5 MPa, which is increased by 11.0%, 13.8%, 18.4%, and 6.9%, respectively, compared with neat epoxy (2235.2 MPa at 40  $^{\circ}$ C). As shown in Figure [8B](#page-11-0), the Tg of all the samples show a similar trend of raising first and then falling, as observed in their storage modulus, with the Tg of neat epoxy,  $Ti_3C_2$ /epoxy, 0.25 M  $TiO_2/Ti_3C_2$ /epoxy, 0.5 M TiO<sub>2</sub>/Ti<sub>3</sub>C<sub>2</sub>/epoxy, and 1.0 M TiO<sub>2</sub>/Ti<sub>3</sub>C<sub>2</sub>/epoxy are determined to 146.3 °C, 152.5 °C, 153.9 °C, 156.3 °C, and 143.3 °C, respectively. The increased storage modulus and Tg can be attributed that, on one hand, the  $TiO<sub>2</sub>$  nanodot protuberances maybe provided the anchors to form the mechanical interlock between  $TiO<sub>2</sub>/Ti<sub>3</sub>C<sub>2</sub>$  and epoxy matrix, which physically hindered the molecular mobility of epoxy matrix [\[43\]](#page-14-9); on the other hand, the enhanced contact area between  $TiO<sub>2</sub>/Ti<sub>3</sub>C<sub>2</sub>$  and epoxy resulting from  $TiO<sub>2</sub>$  nanodots, as well as abundant functional groups ( $-OH$ ,  $-F$ ) on TiO<sub>2</sub> and Ti<sub>3</sub>C<sub>2</sub> surface induced a strong interaction between  $\text{TiO}_2/\text{Ti}_3\text{C}_2$  and epoxy matrix, which restricted the molecular motion of the local epoxy around  $TiO_2/T_{13}C_2$  [\[44\]](#page-14-10). The minimum Tg in 1.0 M  $TiO_2/T_{13}C_2$ /epoxy nanocomposites maybe ascribed to the poor interaction between  $1.0$  M  $TiO<sub>2</sub>/Ti<sub>3</sub>C<sub>2</sub>$  and epoxy matrix, as well as the reduced degree of cross-linking reaction of epoxy resin resulting from the high-density TiO<sub>2</sub> nanodots of  $1M$  TiO<sub>2</sub>/Ti<sub>3</sub>C<sub>2</sub> [\[45\]](#page-14-11).

### *3.9. A Proposed Mechanism for the Improved Tribological and Thermo-Mechanical Properties of TiO2/Ti3C2/Epoxy Nanocomposites*

On the basis of the abovementioned results, a plausible mechanism is proposed for shedding light on the improved tribological and thermo-mechanical properties of  $TiO<sub>2</sub>/Ti<sub>3</sub>C<sub>2</sub>/$ epoxy nanocomposite, as illustrated in Scheme [2.](#page-11-1) For the low density  $\text{TiO}_2/\text{Ti}_3\text{C}_2$  incorporated epoxy nanocomposite, the sparse  $TiO<sub>2</sub>$  nanodots grown on  $Ti<sub>3</sub>C<sub>2</sub>$  surface only provide a small amount of  $TiO<sub>2</sub>$  nanodot anchors, which induces a weak interfacial interaction and mechanical interlock between  $TiO<sub>2</sub>/Ti<sub>3</sub>C<sub>2</sub>$  and epoxy matrix. As a result, the tribological and thermo-mechanical properties of  $TiO<sub>2</sub>/Ti<sub>3</sub>C<sub>2</sub>/epoxy$  nanocomposite are enhanced to some extent. When the medium density  $TiO<sub>2</sub>/Ti<sub>3</sub>C<sub>2</sub>$  are used as fillers, the moderate  $TiO<sub>2</sub>$  nanodots provide lots of effective anchors (suitable gap between adjacent anchors), supporting the strong interfacial interaction and mechanical interlock between  $Ti_2C_2$  and epoxy matrix. Therefore, the tribological and thermo-mechanical properties of  $\text{TiO}_2/\text{Ti}_3\text{C}_2$ /epoxy nanocomposite are greatly enhanced. However, the dense  $\text{TiO}_2$ nanodots hardly provide effective anchors for the  $TiO<sub>2</sub>/Ti<sub>3</sub>C<sub>2</sub>$ -epoxy interfacial interaction, leads to a poor mechanical interlock between  $TiO<sub>2</sub>/Ti<sub>3</sub>C<sub>2</sub>$  and epoxy matrix. Thus, the worse tribological and thermo-mechanical properties are gained in the high density  $TiO<sub>2</sub>/Ti<sub>3</sub>C<sub>2</sub>$  incorporated epoxy nanocomposite.

<span id="page-11-0"></span>

Figure 8. Storage modulus (A) and tan  $\delta$  (B) curves versus temperature for neat epoxy and TiO2/Ti3C2/epoxy nanocomposites. TiO2/Ti3C2/epoxy nanocomposites.

<span id="page-11-1"></span>

**Scheme 2.** Illustration for the interaction between  $TiO<sub>2</sub>/Ti<sub>3</sub>C<sub>2</sub>$  and epoxy matrix in  $TiO<sub>2</sub>/Ti<sub>3</sub>C<sub>2</sub>/$ nanocomposites. epoxy nanocomposites.

### **4. Conclusions**

In summary, the  $TiO<sub>2</sub>$  decorated  $Ti<sub>3</sub>C<sub>2</sub>$  were synthesized by hydrothermal growth of TiO<sub>2</sub> nanodots onto the surface of accordion-like Ti<sub>3</sub>C<sub>2</sub>, and the different decoration degree (low, medium, high density) of  $TiO<sub>2</sub>/Ti<sub>3</sub>C<sub>2</sub>$  composite particles were prepared by regulating the concentration of  $TiO<sub>2</sub>$  precursor solution. Tribological test results indicated that the incorporation of  $TiO_2/Ti_3C_2$  can effectively improve the wear rate of epoxy resin, and the improvement effect was dependent on the density of  $TiO<sub>2</sub>$  nanodots of  $TiO<sub>2</sub>/Ti<sub>3</sub>C<sub>2</sub>$ . Among, the medium density  $TiO<sub>2</sub>/Ti<sub>3</sub>C<sub>2</sub>$  could improve the wear rate of epoxy resin at the greatest extent, with a wear rate of only  $2.67 \times 10^{-14}$  m<sup>3</sup>/(N·m) was obtained under the normal load of 8 N. This mainly ascribed that the moderate  $TiO<sub>2</sub>$  nanodot protuberances on the  $Ti<sub>3</sub>C<sub>2</sub>$  surface induced a strong mechanical interlocking effect between medium density  $TiO<sub>2</sub>/Ti<sub>3</sub>C<sub>2</sub>$  and epoxy matrix, which can bear a higher normal shear stress during sliding friction. The morphologies of worn surfaces and wear debris revealed that the wear form was gradually transformed from abrasive wear in neat epoxy to adhesive wear in 1.0 M  $\text{TiO}_2/\text{Ti}_3\text{C}_2$ /epoxy nanocomposites. Moreover, the results of thermo-mechanical property indicated that incorporation of  $TiO<sub>2</sub>/Ti<sub>3</sub>C<sub>2</sub>$  also effectively improved the storage modulus and glass transition temperature of epoxy resin. Our results demonstrated that the tribological and thermo-mechanical properties of  $TiO<sub>2</sub>/Ti<sub>3</sub>C<sub>2</sub>/epoxy$  nanocomposites can be strongly influenced by the micromorphology of  $TiO<sub>2</sub>/Ti<sub>3</sub>C<sub>2</sub>$ , which provides a promising strategy for improving the tribological and thermo-mechanical properties of epoxy nanocomposites.

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