

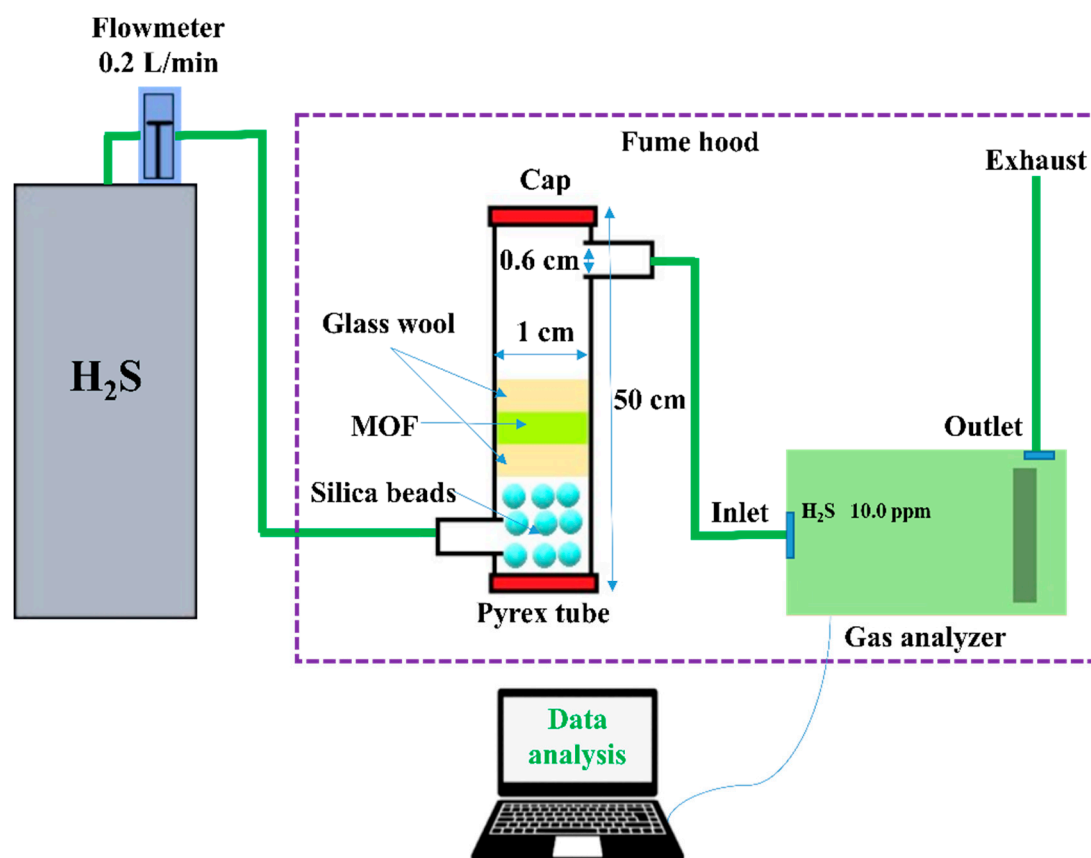
Supplementary material

# Role of bimetallic solutions in the growth and functionality of Cu-BTC metal-organic framework

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**Figure S1.** Schematic representation of H<sub>2</sub>S adsorption system used in the study.

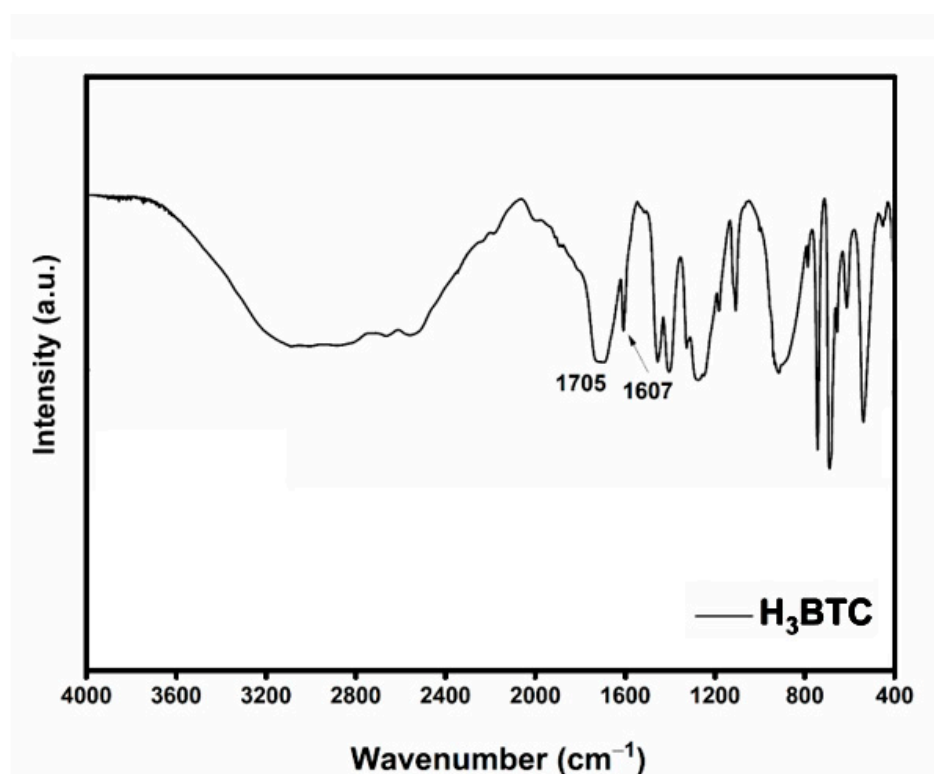
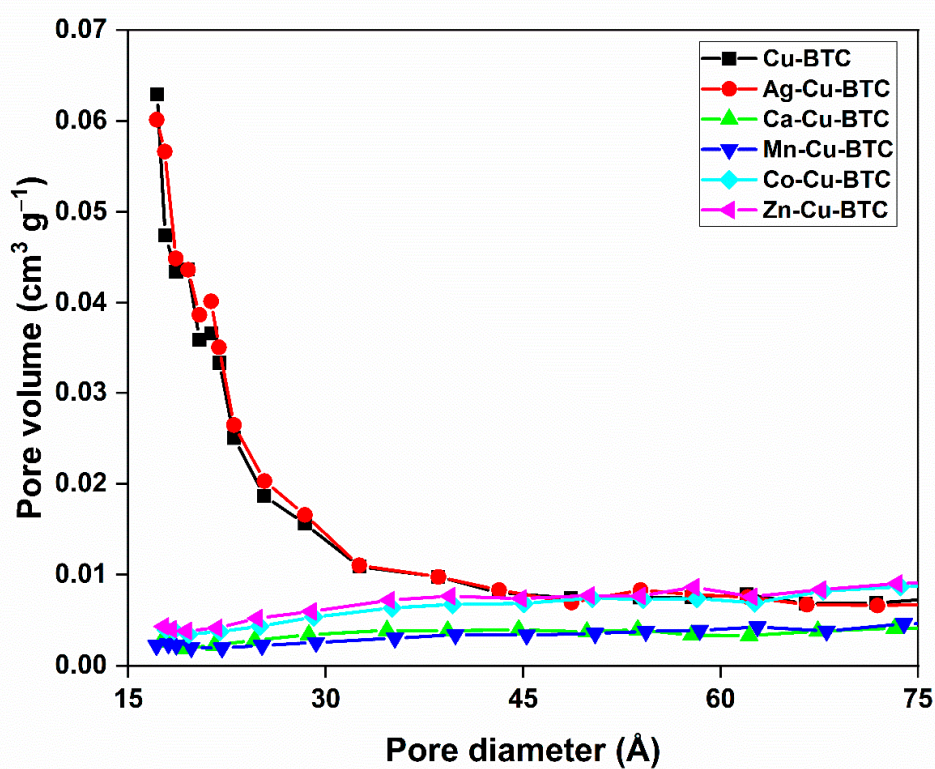
Figure S2. FTIR spectrum of H<sub>3</sub>BTC ligand.

Figure S3. Pore size distribution curves of M-Cu-BTC MOFs.

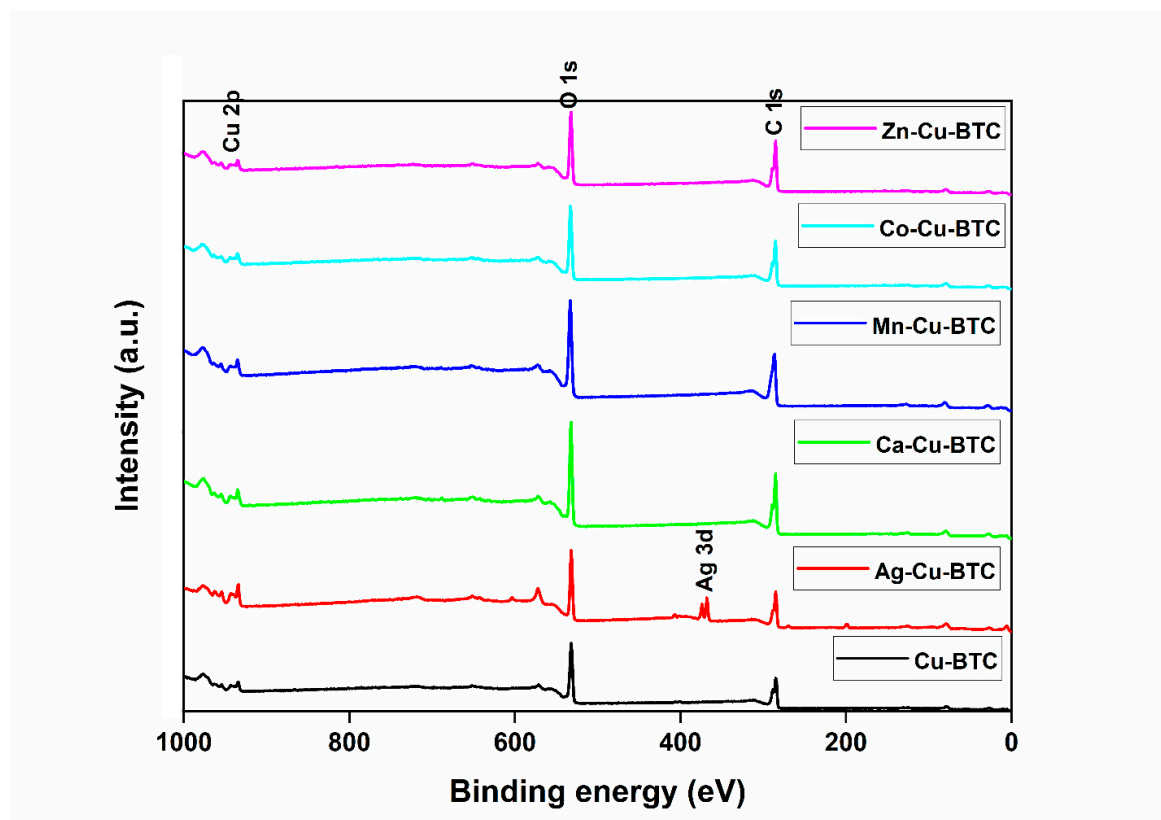


Figure S4. XPS surveys of M-Cu-BTC MOFs.

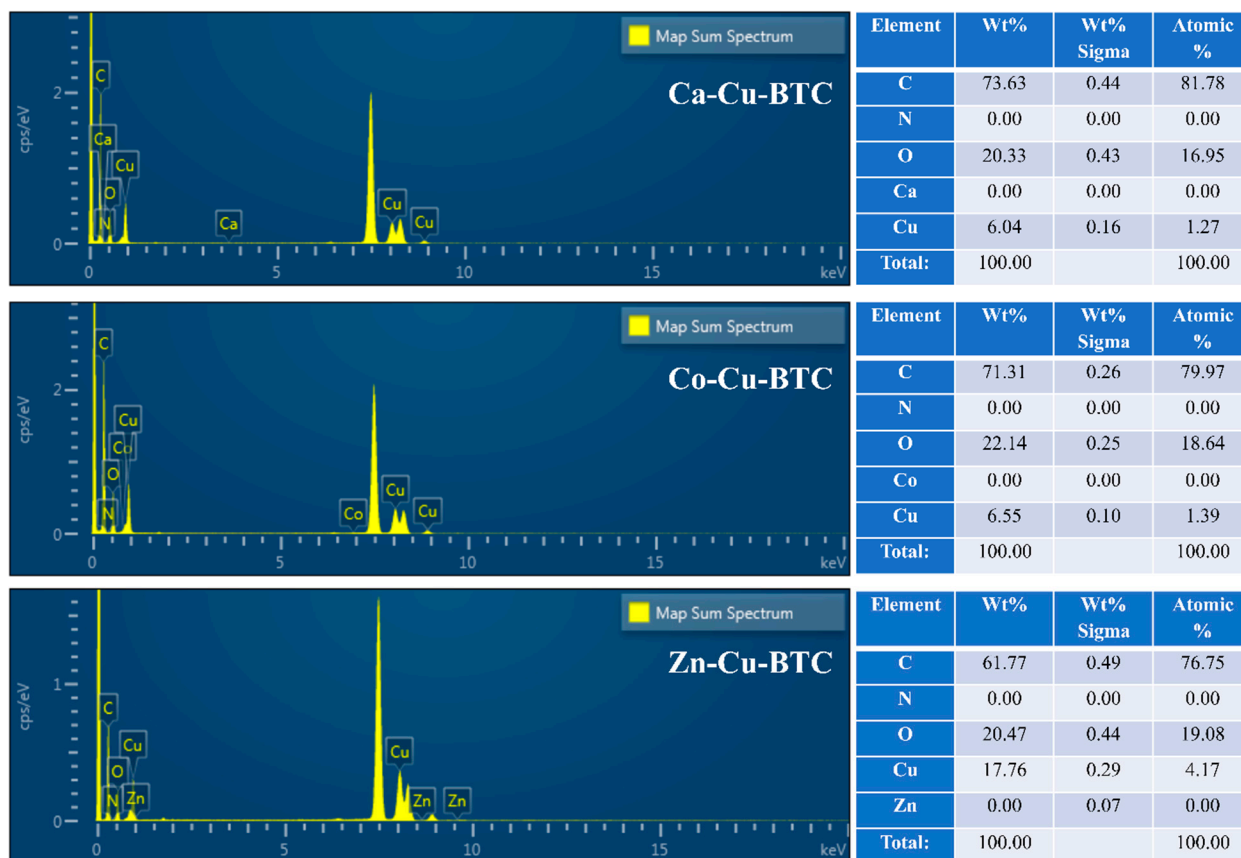


Figure S5. TEM-EDAX analysis of Ca-Cu-BTC, Co-Cu-BTC, and Zn-Cu-BTC.

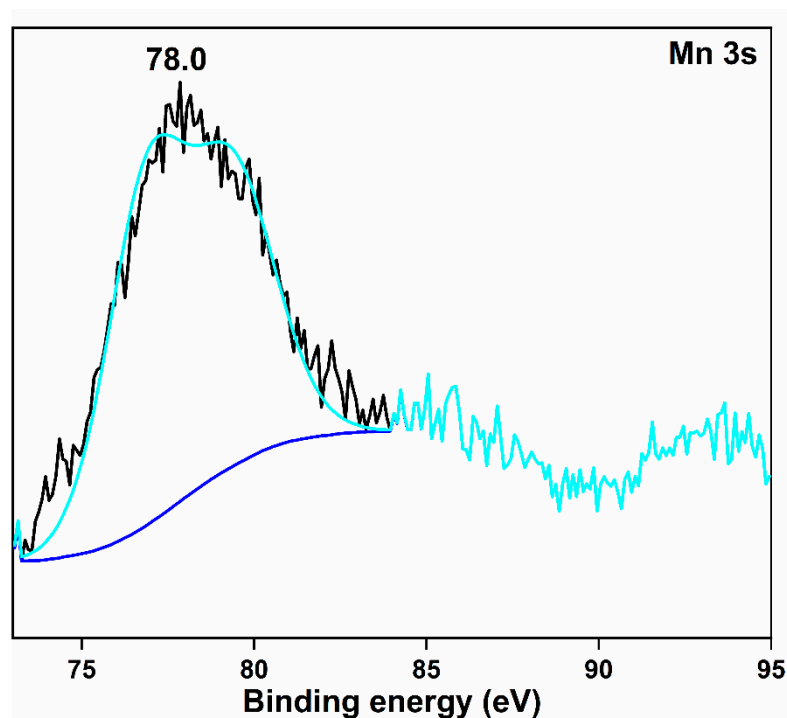


Figure S6. HRXPS spectrum of Mn 3s in Mn-Cu-BTC.

Table S1. Surface elemental analysis using XPS analysis.

MOF	Elements			
	C	O	Cu	Secondary Metal
Cu-BTC	64.8	33.7	1.5	-
Ag-Cu-BTC	65.0	31.2	2.4	1.4
Ca-Cu-BTC	65.8	32.7	1.5	No signal
Mn-Cu-BTC	65.6	32.2	1.4	0.8
Co-Cu-BTC	65.9	32.6	1.5	No signal
Zn-Cu-BTC	68.0	30.5	1.5	No signal

Table S2. HRXPS Cu 2p signals of MOFs.

Samples	Assignment	E <sub>B</sub> (eV)	FWHM (eV)	At. %
Cu-MOF	Cu 2p <sub>3/2</sub> Cu <sup>+</sup>	932.5	1.6	15.7
	Cu 2p <sub>3/2</sub> Cu <sup>2+</sup>	934.7	2.1	84.3
	Satellite Cu <sup>2+</sup>	939.2	4.0	-
	Satellite Cu <sup>2+</sup>	943.7	3.5	-
Ag-Cu-MOF	Cu 2p <sub>3/2</sub> Cu <sup>+</sup>	931.9	1.5	5.4
	Cu 2p <sub>3/2</sub> Cu <sup>2+</sup>	934.3	2.5	94.6
	Satellite Cu <sup>2+</sup>	939.2	4.0	-
	Satellite Cu <sup>2+</sup>	943.4	3.3	-
Ca-Cu-MOF	Cu 2p <sub>3/2</sub> Cu <sup>+</sup>	932.4	1.7	23.0
	Cu 2p <sub>3/2</sub> Cu <sup>2+</sup>	934.6	2.2	77
	Satellite Cu <sup>2+</sup>	938.8	4.0	-
	Satellite Cu <sup>2+</sup>	943.6	3.5	-
Mn-Cu-MOF	Cu 2p <sub>3/2</sub> Cu <sup>+</sup>	932.1	1.7	20.1
	Cu 2p <sub>3/2</sub> Cu <sup>2+</sup>	934.4	2.5	79.9
	Satellite Cu <sup>2+</sup>	938.9	4.0	-
	Satellite Cu <sup>2+</sup>	943.6	3.5	-
Co-Cu-MOF	Cu 2p <sub>3/2</sub> Cu <sup>+</sup>	932.5	1.7	23.2
	Cu 2p <sub>3/2</sub> Cu <sup>2+</sup>	934.7	2.5	76.8
	Satellite Cu <sup>2+</sup>	939.1	4.0	-

<b>Zn-Cu-MOF</b>	<b>Satellite</b> $\text{Cu}^{2+}$	943.8	3.5	-
	<b>Cu 2p<sub>3/2</sub></b> $\text{Cu}^{+}$	932.3	1.7	27.0
	<b>Cu 2p<sub>3/2</sub></b> $\text{Cu}^{2+}$	934.6	2.5	73.0
	<b>Satellite</b> $\text{Cu}^{2+}$	938.9	4.0	-
	<b>Satellite</b> $\text{Cu}^{2+}$	943.6	3.5	-

**Table S3.** Room temperature  $\text{H}_2\text{S}$  adsorption capacity comparison with the reported Cu-BTC.

Adsorbent	$[\text{H}_2\text{S}]$ (ppm)	BTP (%)	$q_e$ ( $\text{mg g}^{-1}$ )
<b>MOF-199</b> [1]	500	100	77.1
<b>MOF-199</b> [2]	423	1	56.8
<b>MOF-199</b> [3]	10	100	69.0
<b>MOF-199</b> [4]	282	1	57.2
<b>CuBTC</b> [5]	1000	100	115.6
<b>CuBTC</b> [6]	500	80	27.1
<b>HKUST-1</b> [7]	10000	100	41.1
<b>HKUST-1</b> [8]	1000	10	92.0
<b>HKUST-1</b> [9]	99.6	5	23.8
<b>Ag-Cu-BTC</b> [10]	500	20	69.7
<b>Cu-BTC</b>	500	2	35.3
<b>Ag-Cu-BTC</b> [This study]			59.1

<sup>BTP</sup> Breakthrough point; <sup>N.A.</sup> Not available

**Table S4.** Kinetic parameters for the IPD model.

MOFs	$k_{ip}$ ( $\text{mg g}^{-1} \text{min}^{-1/2}$ )	$C_i$ ( $\text{mg g}^{-1}$ )	$R^2$
<b>Cu-BTC</b>	2.11	-2.71	0.99
<b>Ag-Cu-BTC</b>	1.46	-1.58	0.99
<b>Mn-Cu-BTC</b>	2.83	-0.29	0.99
<b>Co-Cu-BTC</b>	2.71	-0.52	0.99
<b>Ca-Cu-BTC</b>	2.69	-0.21	1.00
<b>Zn-Cu-BTC</b>	2.89	0.90	0.99

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