

Article

Fabrication of Ultra-High Aspect Ratio (>420:1) Al₂O₃ Nanotube Arrays by Sidewall Transfer Metal Assistant Chemical Etching

Hailiang Li and Changqing Xie *

Key Laboratory of Microelectronic Devices & Integrated Technology, Institute of Microelectronics of Chinese Academy of Sciences, Beijing 100029, China; lihailiang@ime.ac.cn

* Correspondence: xiechangqing@ime.ac.cn; Tel.: +86-01-82995581

Received: 28 February 2020; Accepted: 30 March 2020; Published: 3 April 2020



Abstract: We report a robust, sidewall transfer metal assistant chemical etching scheme for fabricating Al₂O₃ nanotube arrays with an ultra-high aspect ratio. Electron beam lithography followed by low-temperature Au metal assisted chemical etching (MacEtch) is used to pattern high resolution, high aspect ratio, and vertical silicon nanostructures, used as a template. This template is subsequently transferred by an atomic layer deposition of the Al₂O₃ layer, followed by an annealing process, anisotropic dry etching of the Al₂O₃ layer, and a sacrificial silicon template. The process and characterization of the Al₂O₃ nanotube arrays are discussed in detail. Vertical Al₂O₃ nanotube arrays with line widths as small as 50 nm, heights of up to 21 μm, and aspect ratios up to 420:1 are fabricated on top of a silicon substrate. More importantly, such a sidewall transfer MacEtch approach is compatible with well-established silicon planar processes, and has the benefits of having a fully controllable linewidth and height, high reproducibility, and flexible design, making it attractive for a broad range of practical applications.

Keywords: Al₂O₃ nanotube; ultra-high aspect ratio; gold (Au) metal assisted chemical etching; atomic layer deposition; anisotropic dry etching

1. Introduction

In recent years, Al₂O₃ nanotube arrays possessing high surface-to-volume ratios have attracted much attention, owing to their potential applications in optoelectronics [1,2], biotechnology, and photocatalysis [3,4]. For example, several recent studies have indicated that Al₂O₃ nanotube arrays exhibit excellent dielectric properties [5] and good flexibility [6], as compared with other oxide nanotubes. Thus, Al₂O₃ nanotube arrays would be more advantageous for use as optical transportation media in optoelectronics. They can also be utilized as a new biomineralization nanoreactor in biotechnology, and so on. Various template-based strategies for the fabrication of Al₂O₃ nanotubes with a high aspect ratio have been proposed. These include hydrothermal reaction methods [7], coating of carbon nanotubes with aluminum isopropoxide [8], anodization of Si-based Al film [9,10], and etching of a porous anodic alumina (PAA) template in a NaOH solution and ultrasonic treatment of PAA membrane [11,12]. In addition to the Al₂O₃ nanotubes, atomic layer deposition (ALD) coatings of silicon oxide (SiO₂) [13], zinc oxide (ZnO) [14,15], zirconium dioxide (ZrO₂) [16], and titanium oxide (TiO₂) [16,17] have also been successfully applied to the preparation of nanostructures. Using these methods, much progress has been made on achieving highly ordered nanotube arrays with a controllable size and short length. However, due to the poor consistency of the pillar size in the templates, it is not easy to confine and control the profile of the fabricated nanostructures. Moreover, separating hollow nanostructures from a template to obtain individual nanostructures remains a

challenge. Many kinds of nanotubes with high aspect ratios will lose their tubular structure after extraction from the template. With the continuous request for high performance nanotube-based devices, the aspect ratio of the fabricated nanotube arrays as obtained by tailoring electrochemical conditions is far from satisfactory.

Metal assisted chemical etching (MacEtch), first reported by Li and Bohn in 2001 [18], offers a promising wet etching solution for generating silicon nanostructures [19–22]. In addition, the MacEtch of germanium (Ge) [23] and III–V compound semiconductors, such as GaN [24,25], GaAs [26–28], GaP [29], InP [30], AlGaAs [31], InGaAs [32], and InGaP [27], have also been demonstrated. This method has the benefits of inherent simplicity, a low cost, high versatility, and high reproducibility, making it attractive for preparing silicon nanowire arrays. The conventional method for generating micrometer, submicrometer, and nanosized silicon structures with high aspect ratios [19–22] is by submerging a metal-coated Si sample into an etchant solution composed of hydrofluoric acid, hydrogen peroxide solution, and a diluting agent. Furthermore, large-area uniform micro-gratings with well controlled morphological features and depths as large as 80 μm have also been successfully produced by optimizing the MacEtch method [33]. Highly dense Si/TiO₂ core/shell nanowire arrays have also been synthesized using a nanostructured Si template obtained by the MacEtch of the Si substrates and a layer of TiO₂ deposited by ALD [34,35]. In the previous works, using Ti/Au nanostructures patterned with electron-beam lithography followed by ion beam etching, we fabricated vertical silicon nanopillar arrays with a period of 250 nm and an aspect ratio of 160:1 using MacEtch [36]. A 20 nm minimum feature size was also realized by MacEtch-based nanoimprinting [37]. Unfortunately, the MacEtch method cannot be directly adopted to generate oxide nanotubes.

Sidewall transfer lithography has been widely recognized as a promising technology that can fabricate deca-nanometer metal-oxide-semiconductor field-effect transistors (MOSFETs) and silicon nanostructures [38–40]. The key idea is to combine three well-established techniques (lithography, sidewall process, and dry etching) to create silicon nanostructures. In this work, we present a reliable means, called sidewall transfer metal assistant chemical etching, to fabricate Al₂O₃ nanotube arrays with an ultra-high aspect ratio. The key idea is to combine a low-temperature MacEtch and sidewall transfer process based on ALD and dry etching to simultaneously achieve their respective advantages. The former is used to generate vertically oriented silicon nanostructures that serve as a sacrificial layer, while the latter is used to transfer the silicon nanostructures to Al₂O₃ nanotube arrays with a higher aspect ratio and smaller feature size. The ultimate aspect ratio and feature size of the fabricated Al₂O₃ nanotube arrays can be controlled by modifying the thickness of the Al₂O₃ film deposited by ALD. The influence of the deposition temperature and annealing temperature on the structure and optical properties of Al₂O₃ thin films deposited by ALD are also examined.

2. Materials and Methods

The process flow of the proposed sidewall transfer MacEtch is shown in Figure 1. In our experiments, 4-in. p-Type <100> Cz silicon wafers (Silicon Quest International, San Jose, CA, USA) with a resistivity of 2–10 $\Omega\cdot\text{cm}$ and thickness of 500 μm were used. After the samples (Figure 1a) were cleaned in sulfuric acid and hydrogen peroxide to remove the surface native oxide layers, a 3-nm Ti/20 nm Au thin film was deposited on the silicon substrates by a magnetron sputtering system (ACS-4000, ULVAC Company, Chigasaki, Japan). The working pressure was maintained at 4.5×10^{-6} Torr, and the temperature of the chamber was kept at 25 $^{\circ}\text{C}$ over the entire deposition process. Then, highly sensitive chlorinated electron beam resist ZEP520A was spin-coated on the Ti/Au layer to a thickness of about 400 nm, and was baked on a hotplate at 180 $^{\circ}\text{C}$ for 2 min (Figure 1b). The resist was exposed with an electron beam lithography system (JBX-6300FS, JEOL Company, Tokyo, Japan) for patterning definition (Figure 1c), and ion beam etching (IBE) was performed to transfer ZEP520A resist pattern onto the Ti/Au layer, forming Ti/Au nanostructures that serve as a local cathode (Figure 1d). Then, the MacEtch process was carried out to generate silicon nanostructures with a high aspect ratio (Figure 1e). The ALD process was used to deposit the Al₂O₃ film on the sidewalls and on top of the generated silicon

nanostructures (Figure 1g). Finally, the exposed tops of the Al_2O_3 film and its silicon nanostructures underneath were removed by dry etching, forming Al_2O_3 nanotube arrays with a higher aspect ratio and smaller feature size.

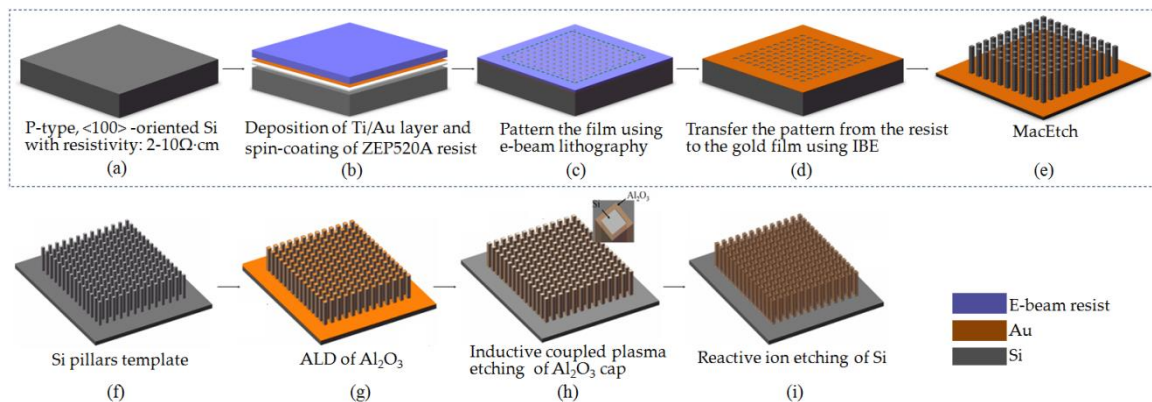


Figure 1. Schematic diagram of the sidewall transfer MacEtch process for the realization of the Al_2O_3 nanotube arrays with ultra-high aspect ratios: (a) silicon substrate; (b) deposition of the Ti/Au layer and ZEP520A resist; (c) electron beam lithography; (d) ion beam etching; (e) metal assisted chemical etching (MacEtch); (f) Si pillars template; (g) atomic layer deposition (ALD) Al_2O_3 film; (h) dry etching; (i) reactive ion etching of Si.

The morphology and structure of the fabricated Al_2O_3 nanotube arrays were characterized by a scanning electron microscope (SEM; SUPRA-55, Zeiss, Oberkochen, Germany). The information on the surface chemistry of the Al_2O_3 layer deposited by the atomic layer deposition (ALD) technique was investigated using X-ray photo emission spectroscopy (XPS) with a monochromatic Al $K\alpha$ X-ray source. Both the Al_2O_3 film thickness and its refractive index profile were determined by a spectroscopic ellipsometry (Horiba Uvisel FUV, Kyoto, Japan) over the spectral range of 150 to 900 nm using the Cauchy model.

2.1. Fabrication of Ti/Au Nanostructures with Low Aspect Ratio

To obtain high-resolution resist patterns, electron beam lithography was performed at an accelerating voltage of 100 kV, with a beam current of 200 pA and an exposure dose of $400 \mu\text{C}/\text{cm}^2$. After electron beam exposure, the ZEP520A resist was developed using a standard developer N50D (ZEON) for 1 min at 18°C , and rinsed with isopropanol (IPA) for 40 s to stop development. Then, the sample was dried with a steady stream of N_2 .

Next, a home-made argon (Ar) IBE system was used to transfer the resist patterns into a Ti/Au thin film deposited onto the silicon wafer. The working pressure was maintained at 1.0×10^{-4} Torr, and the substrate temperature was lower than 100°C over the entire IBE process. The Ar^+ ion energy and the beam current density were 500 eV and $1 \text{ mA}/\text{cm}^2$, respectively. The corresponding etching rates of Au and the resist were 120 nm/min and 20 nm/min, respectively, resulting in a selectivity ratio of 6:1. The etching time in our experiments was 12 s. After completion of the IBE process, the residual ZEP520A resist was removed by washing with a resist removal solution (ZDMAC, ZEON Company, Tokyo, Japan), followed by oxygen plasma ashing. The patterned Ti/Au nanostructures served as a catalyst in the subsequent low-temperature MacEtch process.

2.2. Fabrication of Silicon Nanostructures with High Aspect Ratio

A low-temperature Au MacEtch process was performed to generate silicon nanostructures with a high aspect ratio. It should be noted that samples must be kept clean and tidy before the Au MacEtch process. The Au MacEtch process was carried out in an etchant solution composed of hydrofluoric acid, hydrogen peroxide solution, and deionized water ($4.1 \text{ M HF}/0.15 \text{ M H}_2\text{O}_2/45 \text{ M H}_2\text{O}$) at 2°C . The

etching was conducted for 15 min. It is well known that the collapse of large aspect ratio nanostructures often occurs during the drying step. In our drying scheme, we used isopropanol with low surface tension, instead of deionized water (DI) water as a rinse solution. This reduced the capillary force acting on the silicon nanostructures. After the sample was rinsed with isopropanol, it was evaporated naturally and dried at room temperature.

2.3. Al₂O₃ Film Deposition by ALD

The ALD process allows one to precisely deposit highly uniform and conformal thin films onto complex three-dimensional topographies. Here, an Al₂O₃ film was deposited using Al(CH₃)₃ (TMA) and deionized water with a hot-wall atomic layer deposition system (Picosun R200, Espoo, Finland). The deposition temperature was 300 °C. An Al₂O₃ film was grown on the sidewalls and on top of the fabricated silicon nanostructures with an aspect ratio of 160:1, using TMA and deionized water. TMA (Sigma Aldrich, St. Louis, MO, USA) was used as the precursor and deionized water was used as an oxidant source during the ALD process. The TMA reactant exposure time, N₂ purge time following TMA reactant exposure, water exposure time, and N₂ purge time following the H₂O reactant exposure were 0.5 s, 2 s, 0.5 s, and 2 s, respectively. The growth rate of the Al₂O₃ film was 0.089 nm/cycle, as inferred by spectroscopic ellipsometry. The sample was processed with 400 ALD cycles, and the corresponding Al₂O₃ film thickness was 50 nm.

2.4. Annealing Process and Characterization of the Al₂O₃ Film

The thermal expansion coefficient of Al₂O₃ film is $8.2 \times 10^{-6} \text{ }^\circ\text{C}^{-1}$, which is higher than that of silicon ($2.6 \times 10^{-6} \text{ }^\circ\text{C}^{-1}$). To compensate the lattice mismatch between the Al₂O₃ ALD film and silicon nanostructures, four samples were annealed at 700 °C, 800 °C, 900 °C, and 1000 °C under a vacuum for 90 min, separately. The specifications of the Al₂O₃ film was investigated using X-ray photoelectron spectra (XPS). Characterization of the optical properties of our Al₂O₃ film from 150 to 900 nm wavelengths was also performed using spectroscopic ellipsometry.

2.5. Formation of Al₂O₃ Nanotube Arrays with an Ultra-High Aspect Ratio

First, the exposed tops of the Al₂O₃ film on the silicon nanostructures were removed by an inductive coupled plasma (ICP) etching system (ULVAC, Japan), with a mixture of BCl₃ and Cl₂ reactive gas. Secondly, the silicon nanostructures underneath the exposed tops of the Al₂O₃ film were removed by the same ICP etching system with SF₆ reactive gas and an isotropic reactive ion etching mode. The etching parameters of the Al₂O₃ film and silicon nanostructures are summarized in Table 1.

Table 1. Recipes for Al₂O₃ and Si in etch system.

Process Parameters	Al ₂ O ₃ Etch	Si Etch
Cl ₂ (sccm)	1.2	–
BCl ₃ (sccm)	6.8	–
SF ₆ (sccm)	–	90
Pressure (mtorr)	3	4
Coil power (W)	900	400
Platen power (W)	200	3

Process temperature is 25 °C for all of the processes.

The etch rates of the Al₂O₃ film and silicon nanostructures were 0.89 nm/s and 9 nm/s, respectively. The etching selectivity between the silicon nanostructures and Al₂O₃ film was as high as 66,000:1 for the Si etching recipe [41].

3. Results and Discussion

Figure 2 shows the dependence of the Au MacEtch process parameters (etchant solutions and etching temperature) on the quality of the fabricated silicon nanostructures. When the hydrogen peroxide concentration in the etchant solution was relatively high, the lateral etching rate increased and defects began to occur on the sidewalls, as shown in Figure 2a. This is because as the concentration of H_2O_2 increased, the number of the generated holes also increased, resulting in an increase in the silicon etching rate. In other words, when the generated holes between the Si and Ti/Au interfaces could not be completely consumed, the excess holes spread laterally, leading to lateral corrosion and the formation of defects in the sidewalls. Using an optimized H_2O_2 concentration, the transverse etching rate could be limited by the availability of the generated hole. Thus, vertical silicon nanostructures could be obtained, as shown in Figure 2b.

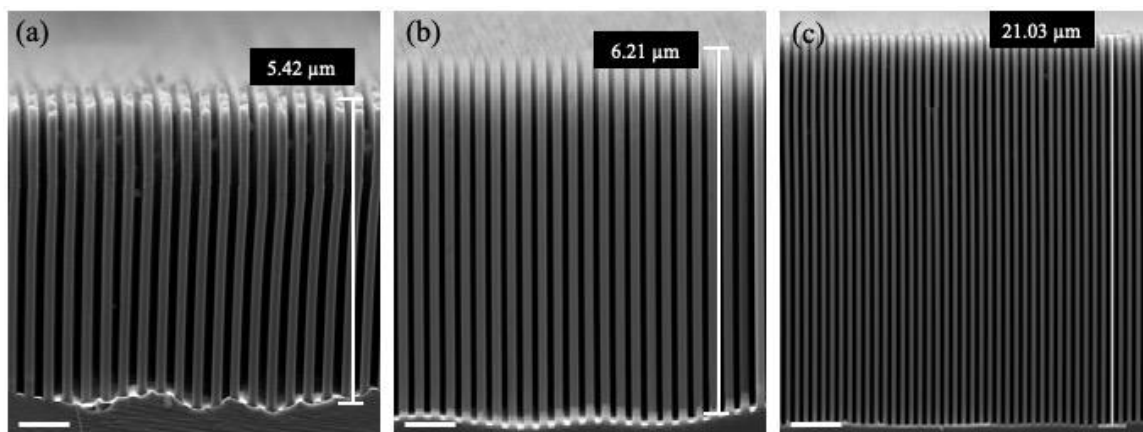


Figure 2. Cross-sectional scanning electron microscope (SEM) micrographs of silicon nanostructures using various etchant solutions composed of hydrofluoric acid, hydrogen peroxide solution, and deionised (DI) water with molar ratios of (a) 4.8:0.35:50, (b) 4.8:0.15:50, and (c) 4.8:0.15:50. The scale bars in (a–c) are 850 nm, 900 nm, and 1.9 μm , respectively. The etching temperatures in (a,b) are 22 $^{\circ}\text{C}$, and the etching temperature in (c) is 2 $^{\circ}\text{C}$.

Etching temperature also plays an essential role in the Au MacEtch process. When the Au MacEtch process was performed at 2 $^{\circ}\text{C}$, a better morphology could be obtained than that at room temperature, at the expense of a much lower etching rate. In this work, a low-temperature Au MacEtch process could result in silicon nanostructures with an aspect ratio of up to 160:1, as shown in Figure 2c. Silicon nanostructure bottoms that are especially clean and flat can be obtained.

Figure 3 shows the binding energies of the Al_2O_3 film deposited at 300 $^{\circ}\text{C}$. Two signatures of orbital, 74.4 eV for Al (2p) and 531.5 eV for O (1s), can be observed from an XPS wide scan. The O/Al ratio of the Al_2O_3 film is close to the expected value of 1.5, corresponding to the lattice oxygen of Al_2O_3 [42]. The difference between these two elemental peaks is close to the standard values in the literature for different forms of aluminum oxide [43,44]. A peak of binding energy of 1s carbon in the Al_2O_3 film could also be observed, indicating either an incomplete reaction or an insufficient N_2 purge time.

Figure 4 shows the Al_2O_3 ALD film thickness and refractive index as a function of the annealing temperature. The refractive index of the Al_2O_3 ALD film initially increased with an increase of annealing temperature from 700 $^{\circ}\text{C}$ to 800 $^{\circ}\text{C}$, and later saturated with the increasing annealing temperature. When the annealing temperature reached 1150 $^{\circ}\text{C}$, the refractive index approached a maximized value of 1.724, which is slightly smaller than that of crystalline sapphire (1.76). The measured data indicate that the amorphous Al_2O_3 ALD film in this experiment would densify further upon crystallization. This is consistent with the existing result [45]. Meanwhile, there was a 10% decrease in the Al_2O_3 ALD film thickness with increasing the annealing temperature from 700 $^{\circ}\text{C}$ to 800 $^{\circ}\text{C}$, which was mainly

due to an increase in the density and purity levels of the Al₂O₃ films deposited by ALD. The measured data also indicate the occurrence of Al₂O₃ crystallization after high-temperature annealing, which appears to be particularly advantageous for preventing the collapse of ultra-high aspect ratio Al₂O₃ nanotube arrays.

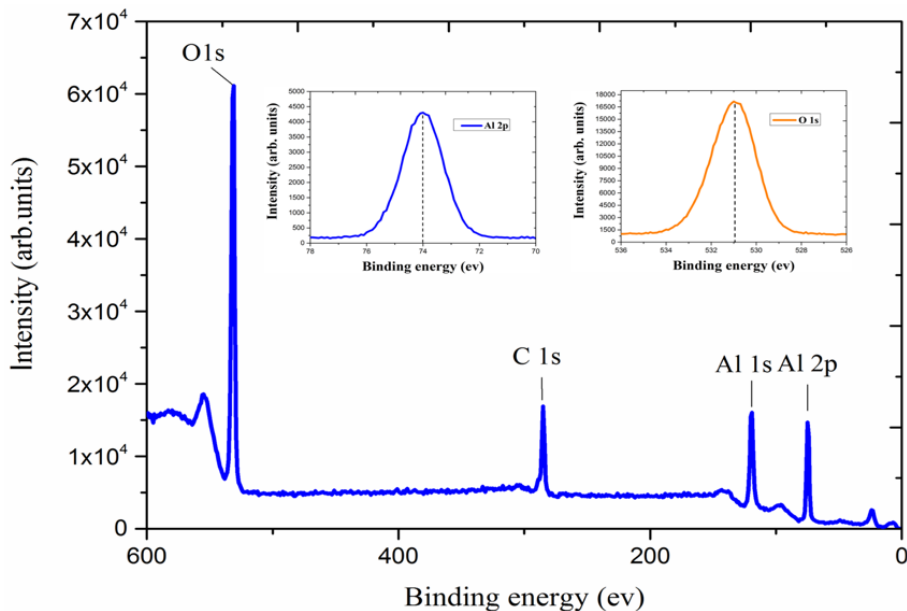


Figure 3. X-ray photoelectron spectra of the Al₂O₃ layer deposited at 300 °C. The two insets show the Al (2p) and O (1s) peaks, respectively.

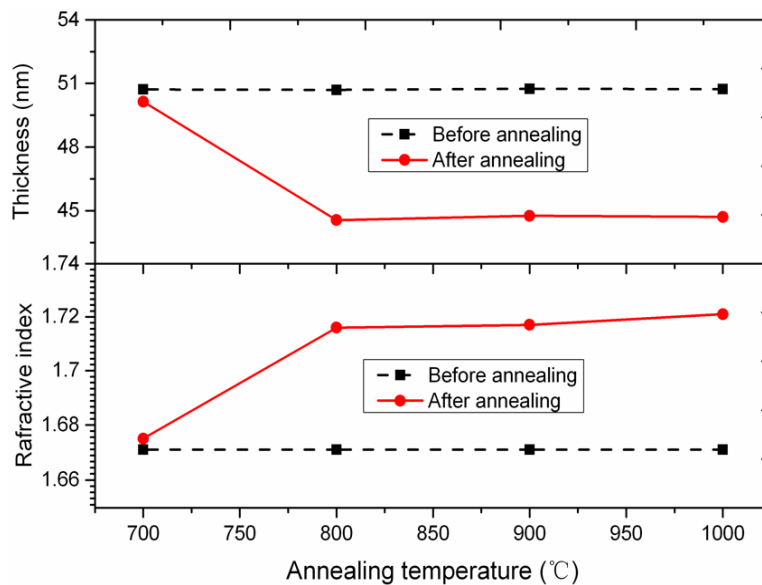


Figure 4. Variation of film thickness and refractive index versus annealing temperature.

Figure 5a shows the SEM image of the top view of the generated Si nanostructures, followed by the Al₂O₃ film deposition by ALD and annealed at 1150 °C for 90 min. The results of the generated Al₂O₃ nanotube arrays without the annealing process are also given in Figure 5b. One can clearly see that the Al₂O₃ nanotube arrays bend and lean against each other. The bending is probably caused by a mismatch in internal stress between the Al₂O₃ layer and Si pillar. The thermal expansion coefficient of the Al₂O₃ film is $8.2 \times 10^{-6} \text{ }^\circ\text{C}^{-1}$, while the thermal expansion coefficient of silicon is $2.6 \times 10^{-6} \text{ }^\circ\text{C}^{-1}$. It should be noted that the Al₂O₃ ALD film was deposited by the temperature gradient method over a temperature range from room temperature to 300 °C, and the Si cores were removed afterwards using

a plasma reactive ion etching at 25 °C. By comparison, Figure 5c,d shows the SEM images of the top and cross section views, respectively, of the generated Al₂O₃ nanotube arrays after the whole process was finished. These uniform structures had a line width of 50 nm and height of 21 μm, corresponding to an aspect ratio as high as 420:1. The Al₂O₃ nanotube arrays were perfectly preserved from collapsing after the annealing process, confirming the successful pattern transfer of the sidewall transfer MacEtch process. The shape of the Al₂O₃ nanotube arrays exhibited very little deformation relative to that of the original Si nanostructures. Figure 5e shows a tilted SEM view of the Al₂O₃ nanotube arrays after the whole process. One can see that the silicon template was almost completely removed by plasma reactive ion etching at 25 °C.

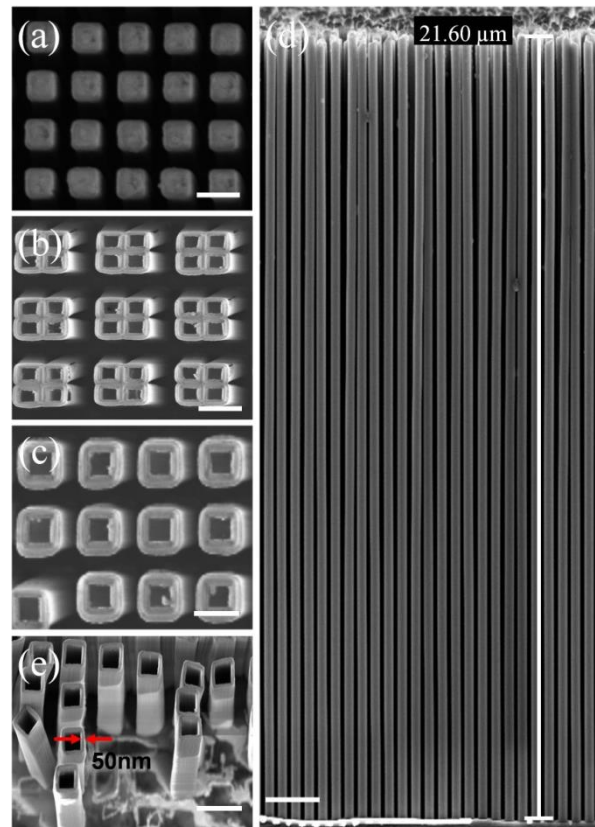


Figure 5. A series of images demonstrating sidewall transfer MacEtch for the preparation of ultra-high aspect ratio Al₂O₃ nanotube arrays. (a) Top view SEM of the Si nanostructures after Al₂O₃ film deposition by ALD. The scale bar is 300 nm. (b) Top view SEM of the generated Al₂O₃ nanotube arrays without annealing process. The scale bar is 400 nm. (c,d) SEM images of the top and cross section views of the generated Al₂O₃ nanotube arrays resulting from the sidewall transfer MacEtch process. The scale bars in (c,d) are 250 nm and 1.6 μm, respectively. (e) Tilted SEM view of Al₂O₃ nano-tube structures after the whole process. The scale bar is 400 nm. The tilting angle is 35°.

4. Conclusions

In summary, we have demonstrated a reliable sidewall transfer MacEtch process to fabricate ultra-high aspect ratio Al₂O₃ nanotube arrays with linewidths as small as 50 nm, heights of up to 21 μm, and an aspect ratio of up to 420:1. This technique combines the advantages of the high aspect ratio nanostructure capabilities of the low-temperature Au MacEtch with the sidewall transfer process. The use of the sidewall transfer process has two advantages. First, it leads to higher-resolution and higher aspect ratio Al₂O₃ nanotube patterns than with MacEtch, and second, the line-width of the Al₂O₃ nanotubes can be precisely controlled by the cycle number of ALD for Al₂O₃. The sidewall transfer MacEtch provides a promising route to the scalable manufacturing of ultra-high aspect ratio

Al₂O₃ nanotube arrays, and is applicable to other kinds of oxide nanotubes, as long as the oxide can be deposited by ALD.

Author Contributions: H.L. and C.X. conceived the idea for the experiment; H.L. performed the experiments; H.L. and C.X. wrote the manuscript. All authors have read and agreed to the published version of the manuscript.

Funding: This work was supported by the National Key Research and Development Program of China (grant no. 2017YFA0206002) and the National Natural Science Foundation of China (grant no. U1832217, no. 61804169, and no. 61821091).

Conflicts of Interest: The authors declare no conflict of interest.

References

1. Shanmugham, S.; Hendrick, M.; Richards, N.; Oljaca, M. Candidate oxidation resistant coatings via combustion chemical vapor deposition. *J. Mater. Sci.* **2004**, *39*, 377–381. [[CrossRef](#)]
2. Gutiérrez, M.; Lloret, F.; Pham, T.T.; Cañas, J.; Reyes, D.F.D.L.; Eon, D.; Pernot, J.; Araujo, D. Control of the Alumina Microstructure to Reduce Gate Leaks in Diamond MOSFETs. *Nanomater* **2018**, *8*, 584. [[CrossRef](#)] [[PubMed](#)]
3. Thamaraiselvi, T.V.; Rajewari, S. Biological evaluation of bioceramic materials—A review. *Trends Biomater. Artif. Organs* **2004**, *81*, 9–17.
4. Xifré-Pérez, E.; Ferre-Borull, J.; Pallarès, J.; Marsal, L.F. Mesoporous alumina as a biomaterial for biomedical applications. *Open Mater. Sci.* **2015**, *2*, 13–32. [[CrossRef](#)]
5. De Wit, H.; Crevecoeur, C. The dielectric breakdown of anodic aluminum oxide. *Phys. Lett. A* **1974**, *50*, 365–366. [[CrossRef](#)]
6. Mei, Y.; Wu, X.; Shao, X.F.; Siu, G.G.; Bao, X.M. Formation of an array of isolated alumina nanotubes. *EPL Europhys. Lett.* **2003**, *62*, 595–599. [[CrossRef](#)]
7. Lee, H.C.; Kim, H.J.; Chung, S.H.; Lee, K.H.; Lee, H.C.; Lee, J.S. Synthesis of unidirectional alumina nanostructures without added organics solvents. *J. Am. Chem. Soc.* **2003**, *125*, 2882–2883. [[CrossRef](#)] [[PubMed](#)]
8. Zhang, Y.; Liu, J.; He, R.; Zhang, Q.; Zhang, X.; Zhu, J. Synthesis of alumina nanotubes using carbon nanotubes as templates. *Chem. Phys. Lett.* **2002**, *360*, 579–584. [[CrossRef](#)]
9. Mei, Y.; Wu, X.; Shao, X.; Huang, G.; Siu, G. Formation mechanism of alumina nanotube array. *Phys. Lett. A* **2003**, *309*, 109–113. [[CrossRef](#)]
10. Pu, L.; Bao, X.M.; Zou, J.P.; Feng, D. Individual alumina nanotubes. *Angew. Chem. Int. Ed.* **2001**, *113*, 1538. [[CrossRef](#)]
11. Xiao, Z.-L.; Han, C.; Welp, U.; Wang, H.H.; Vlasko-Vlasov, V.; Kwok, W.K.; Miller, D.J.; Hiller, J.; Cook, R.E.; Willing, G.A.; et al. Nickel antidot arrays on anodic alumina substrates. *Appl. Phys. Lett.* **2002**, *81*, 2869–2871. [[CrossRef](#)]
12. Huang, G.S.; Wu, X.L.; Xie, Y.; Shao, X.F.; Wang, S.H. Light emission from silicon-based porous anodic alumina form ed in 0.5 M oxalic acid. *J. Appl. Phys.* **2003**, *94*, 2407–2410. [[CrossRef](#)]
13. Jensen, D.S.; Kanyal, S.S.; Madaan, N.; Miles, A.J.; Davis, R.C.; Vanfleet, R.; Vail, M.A.; Dadson, A.E.; Linford, M.R. Ozone priming of patterned carbon nanotube forests for subsequent atomic layer deposition-like deposition of SiO₂ for the preparation of microfabricated thin layer chromatography plates. *J. Vac. Sci. Technol. B* **2013**, *31*, 31803. [[CrossRef](#)]
14. Zhang, Y.; Liu, M.; Ren, W.; Ye, Z.-G. Well-ordered ZnO nanotube arrays and networks grown by atomic layer deposition. *Appl. Surf. Sci.* **2015**, *340*, 120–125. [[CrossRef](#)]
15. Elam, J.W.; Routkevitch, D.; Mardilovich, P.P.; George, S.M. Conformal coating on ultra high-aspect-ratio nanopores of anodic alumina by atomic layer deposition. *Chem. Mater.* **2003**, *15*, 3507–3517. [[CrossRef](#)]
16. Zazpe, R.; Knaut, M.; Sopha, H.; Hromadko, L.; Albert, M.; Prikryl, J.; Gärtnerová, V.; Bartha, J.W.; Macak, J.M. Atomic Layer Deposition for Coating of High Aspect Ratio TiO₂ Nanotube Layers. *Langmuir* **2016**, *32*, 10551–10558. [[CrossRef](#)]
17. Shin, H.; Jeong, D.-K.; Lee, J.; Sung, M.M.; Kim, J. Formation of TiO₂ and ZrO₂ Nanotubes Using Atomic Layer Deposition with Ultraprecise Control of the Wall Thickness. *Adv. Mater.* **2004**, *16*, 1197–1200. [[CrossRef](#)]

18. Li, X.; Bohn, P.W. Metal-assisted chemical etching in HF/H₂O₂ produces porous silicon. *Appl. Phys. Lett.* **2000**, *77*, 2572–2574. [[CrossRef](#)]
19. Bai, S.; Du, Y.; Wang, C.; Wu, J.; Sugioka, K. Reusable Surface-Enhanced Raman Spectroscopy Substrates Made of Silicon Nanowire Array Coated with Silver Nanoparticles Fabricated by Metal-Assisted Chemical Etching and Photonic Reduction. *Nanomater* **2019**, *9*, 1531. [[CrossRef](#)]
20. Chang, C.; Sakdinawat, A. Ultra-high aspect ratio high-resolution nanofabrication for hard X-ray diffractive optics. *Nat. Commun.* **2014**, *5*, 4243. [[CrossRef](#)] [[PubMed](#)]
21. Chen, X.; Bi, Q.; Sajjad, M.; Wang, X.; Ren, Y.; Zhou, X.; Xu, W.; Liu, Z. One-Dimensional Porous Silicon Nanowires with Large Surface Area for Fast Charge–Discharge Lithium-Ion Batteries. *Nanomaterials* **2018**, *8*, 285. [[CrossRef](#)] [[PubMed](#)]
22. Hoshian, S.; Gaspar, C.; Vasara, T.; Jahangiri, F.; Jokinen, V.; Franssila, S. Non-Lithographic Silicon Micromachining Using Inkjet and Chemical Etching. *Micromachines* **2016**, *7*, 222. [[CrossRef](#)]
23. Rezvani, S.J.; Pinto, N.; Boarino, L. Rapid formation of single crystalline Ge nanowires by anodic metal assisted etching. *CrystEngComm* **2016**, *18*, 7843–7848. [[CrossRef](#)]
24. Geng, X.; Duan, B.K.; Grismer, D.A.; Zhao, L.; Bohn, P.W. Monodisperse GaN nanowires prepared by metal-assisted chemical etching with in situ catalyst deposition. *Electrochem. Commun.* **2012**, *19*, 39–42. [[CrossRef](#)]
25. Duan, B.K.; Bohn, P.W. High sensitivity hydrogen sensing with Pt-decorated porous gallium nitride prepared by metal-assisted electroless etching. *Analyst* **2010**, *135*, 902. [[CrossRef](#)] [[PubMed](#)]
26. DeJarld, M.; Shin, J.C.; Chern, W.; Chanda, D.; Balasundaram, K.; Rogers, J.A.; Li, X. Formation of High Aspect Ratio GaAs Nanostructures with Metal-Assisted Chemical Etching. *Nano Lett.* **2011**, *11*, 5259–5263. [[CrossRef](#)]
27. Wilhelm, T.S.; Soule, C.W.; Baboli, M.A.; O’Connell, C.J.; Mohseni, P.K. Fabrication of Suspended III–V Nanofoils by Inverse Metal-Assisted Chemical Etching of In_{0.49}Ga_{0.51}P/GaAs Heteroepitaxial Films. *ACS Appl. Mater. Interfaces* **2018**, *10*, 2058–2066. [[CrossRef](#)]
28. Lova, P.; Robbiano, V.; Cacialli, F.; Comoretto, D.; Soci, C. Black GaAs by Metal-Assisted Chemical Etching. *ACS Appl. Mater. Interfaces* **2018**, *10*, 33434–33440. [[CrossRef](#)]
29. Kim, J.; Oh, J. Formation of GaP Nanocones and Micro-mesas by Metal-assisted Chemical Etching. *Phys. Chem. Chem. Phys.* **2016**, *18*, 3402–3408. [[CrossRef](#)]
30. Kim, S.H.; Mohseni, P.K.; Song, Y.; Ishihara, T.; Li, X. Inverse Metal-Assisted Chemical Etching Produces Smooth High Aspect Ratio InP Nanostructures. *Nano Lett.* **2014**, *15*, 641–648. [[CrossRef](#)]
31. Wilhelm, T.S.; Wang, Z.-H.; Baboli, M.A.; Yan, J.; Preble, S.F.; Mohseni, P.K. Ordered Al_xGa_{1-x}As Nanopillar Arrays via Inverse Metal-Assisted Chemical Etching. *ACS Appl. Mater. Interfaces* **2018**, *10*, 27488–27497. [[CrossRef](#)] [[PubMed](#)]
32. Kong, L.; Song, Y.; Kim, J.D.; Yu, L.; Wasserman, D.; Chim, W.K.; Chiam, S.Y.; Li, X. Damage-Free Smooth-Sidewall InGaAs Nanopillar Array by Metal-Assisted Chemical Etching. *ACS Nano* **2017**, *11*, 10193–10205. [[CrossRef](#)]
33. Romano, L.; Vila-Comamala, J.; Jefimovs, K.; Stampanoni, M. Effect of isopropanol on gold assisted chemical etching of silicon microstructures. *Microelectron. Eng.* **2017**, *177*, 59–65. [[CrossRef](#)]
34. Scuderi, V.; Impellizzeri, G.; Romano, L.; Scuderi, M.; Nicotra, G.; Bergum, K.; Irrera, A.; Svensson, B.; Privitera, V. TiO₂-coated nanostructures for dye photo-degradation in water. *Nanoscale Res. Lett.* **2014**, *9*, 458. [[CrossRef](#)]
35. Hwang, Y.J.; Boukai, A.; Yang, P. High Density n-Si/n-TiO₂Core/Shell Nanowire Arrays with Enhanced Photoactivity. *Nano Lett.* **2009**, *9*, 410–415. [[CrossRef](#)] [[PubMed](#)]
36. Li, H.; Ye, T.; Shi, L.; Xie, C. Fabrication of ultra-high aspect ratio (>160:1) silicon nanostructures by using Au metal assisted chemical etching. *J. Micromech. Microeng.* **2017**, *27*, 124002. [[CrossRef](#)]
37. Li, H.; Niu, J.; Wang, G.; Wang, E.; Xie, C. Direct Production of Silicon Nanostructures with Electrochemical Nanoimprinting. *ACS Appl. Electron. Mater.* **2019**, *1*, 1070–1075. [[CrossRef](#)]
38. Kim, D.H.; Sung, S.-K.; Sim, J.S.; Kim, K.R.; Lee, J.D.; Park, B.-G.; Choi, B.H.; Hwang, S.; Ahn, D. Single-electron transistor based on a silicon-on-insulator quantum wire fabricated by a side-wall patterning method. *Appl. Phys. Lett.* **2001**, *79*, 3812–3814. [[CrossRef](#)]
39. Liu, D.; Syms, R.R.A. NEMS by Sidewall Transfer Lithography. *J. Microelectromech. Syst.* **2014**, *23*, 1366–1373. [[CrossRef](#)]

40. Hållstedt, J.; Hellstrom, P.-E.; Radamson, H. Sidewall transfer lithography for reliable fabrication of nanowires and deca-nanometer MOSFETs. *Thin Solid Films* **2008**, *517*, 117–120. [[CrossRef](#)]
41. Sainiemi, L.; Franssila, S. Mask material effects in cryogenic deep reactive ion etching. *J. Vac. Sci. Technol. B Microelectron. Nanometer. Struct.* **2007**, *25*, 801. [[CrossRef](#)]
42. Moulder, J.F. *Hand Book of X-ray Photoelectron Spectroscopy*; Perkin Elmer Corporation: Eden Prairie, MN, USA, 1992; pp. 182–187.
43. Ghiraldelli, E.; Pelosi, C.; Gombia, E.; Chiavarotti, G.; Vanzetti, L. ALD growth, thermal treatments and characterisation of Al₂O₃ layers. *Thin Solid Films* **2008**, *517*, 434–436. [[CrossRef](#)]
44. Böse, O.; Kemnitz, E.; Lippitz, A.; Unger, W.E.S. C 1s and Au 4f/2 referenced XPS binding energy data obtained with different aluminium oxides,-hydroxides and-fluorides. *Fresenius' J. Anal. Chem.* **1997**, *358*, 175–179.
45. Gehr, R.J.; Boyd, R.W. Optical Properties of Nanostructured Optical Materials. *Chem. Mater.* **1996**, *8*, 1807–1819. [[CrossRef](#)]



© 2020 by the authors. Licensee MDPI, Basel, Switzerland. This article is an open access article distributed under the terms and conditions of the Creative Commons Attribution (CC BY) license (<http://creativecommons.org/licenses/by/4.0/>).