Table S1. GC-MS analysis of reaction products.

Reaction time [min] / Chromatogram	Retention time	Retention time	Retention time	Retention time	Retention time [min]
	[min]	[min]	[min]	[min]	1
	1	/	/	/	chemical compound
	chemical compound	chemical compound	chemical compound	chemical compound	
15 min 100 8.72	3.81 2-Methyl-4 5-	-	8.72 H	9.42 OH	-
9.42	dihydrofuran		Furfural	Furfuryl alcohol	
30 min 8.72 9.47 21,83	2-Methyl-4 5-dihydrofuran	-	8.72 H O Furfural	9.47 OH Furfuryl alcohol	_
3.82 21.57 23.82					

60 min 9.59 21.83 21.57 4.69 18.95 23.83	- 4.69 8.71 2-Methyl-tetrahydrofuran (2-MTHF) Furfural	9.59 OH Tetrahydrofurfuryl alcohol HO CH ₃ 5-Hydroxy-2- pentanone HO CH ₃ 2-Methyloxolan-2-
120 min 9.55 21.85 4.70 9.04 19.46 20.27	- 4.70 - 2-Methyl-tetrahydrofuran (2-MTHF)	ol 9.56 OH Tetrahydrofurfuryl alcohol HO CH ₃ 5-Hydroxy-2- pentanone HO CH ₃ 2-Methyloxolan-2- ol

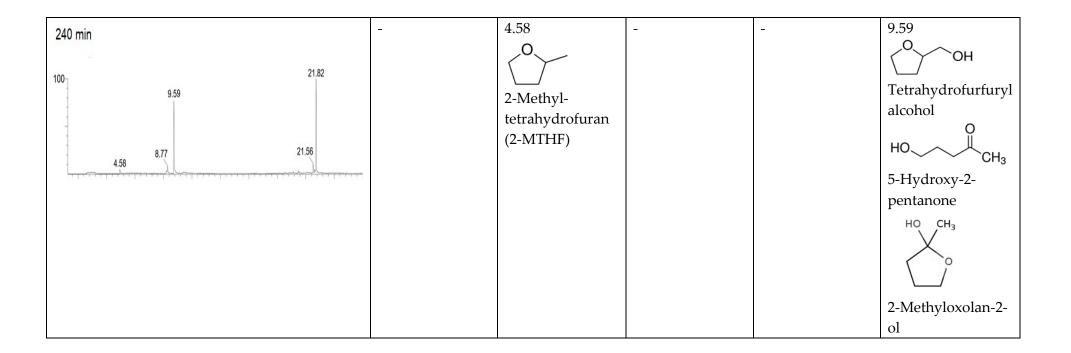
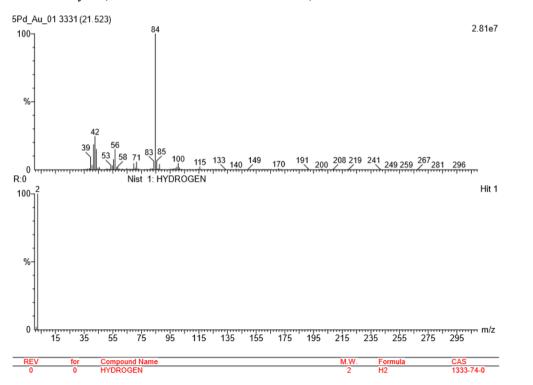


Table S2. Mass spectra of products with retention times 21.5(6) and 21.8(2) in the reduction of furfural in water phase over 5%Pd-1%Au/SiO₂ catalyst (after 240 min of reaction run).



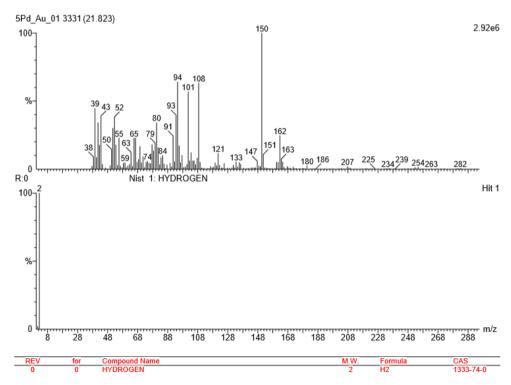
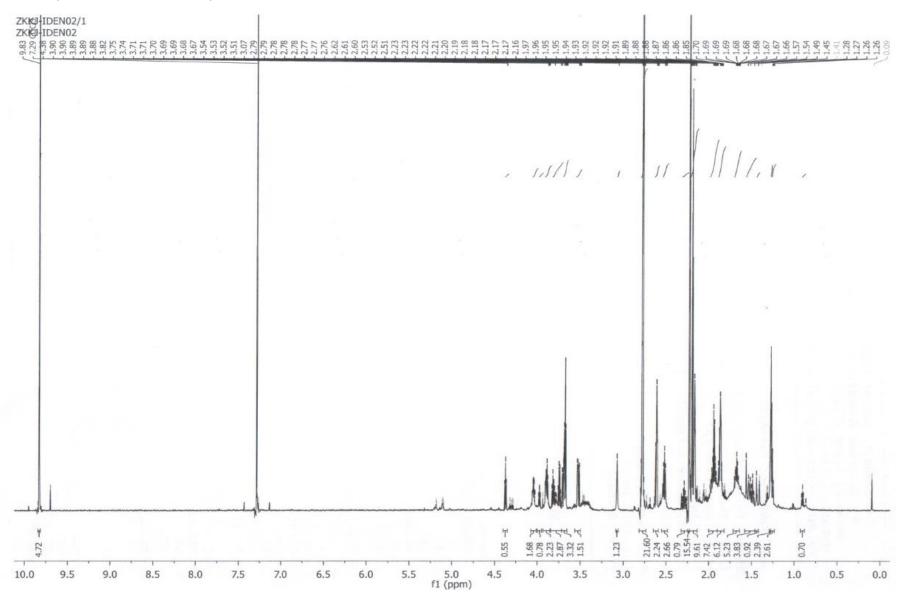
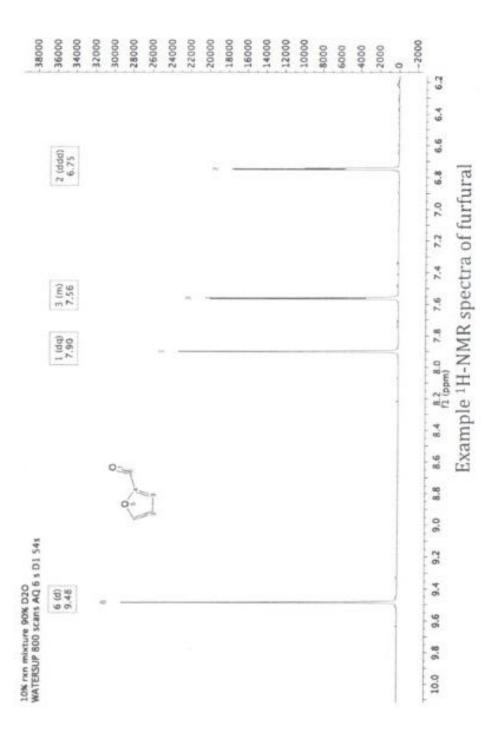


Fig.S1. ¹H-NMR spectrum of the sample of reaction mixture extracted to chloroform after 120 min hydrogenation of furfural in water over 5%Pd-1%Au/SiO₂ catalyst.(Obtained on Bruker Spektrometer AVANCE DPX 250 MHz)





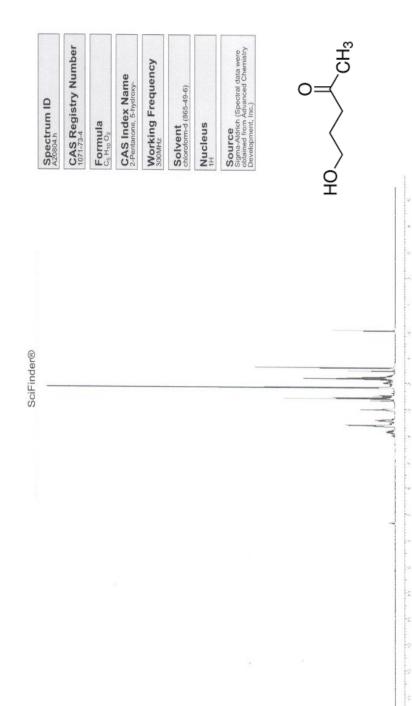


Fig.S2. ¹³C-NMR spectrum of the sample of reaction mixture extracted to chloroform after 120 min hydrogenation of furfural in water over 5%Pd-1%Au/SiO₂ catalyst.(Obtained on Bruker Spektrometer AVANCE DPX 250 MHz)

