Supporting information

Tandem Synthesis of High Yield MoS₂ Nanosheets and Enzyme Peroxidase Mimicking Properties

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Figure S1. Energy dispersive spectroscopy (EDS) elemental analysis of B1-MoS₂ NSs.



Figure S2. DLS size distribution analysis of bulk, R1 and R2-MoS₂.



Figure S3. FTIR spectra of bulk MoS₂ and as synthesized B1, B2 and B3-MoS₂ NSs



Figure S4. XRD spectra of as synthesized B1, B2 and B3-MoS₂ NSs



Figure S5. TMB oxidation assay for evaluating the peroxidase activity of B2 and B3-MoS₂ NSs at identical conditions (i.e. same concentration of NSs, 100 μ L of MoS₂ NSs (~500 μ g mL⁻¹)).

Reaction conditions	Sonication method and	Remarks	Ref.
	reaction time (h)		
Bulk MoS ₂ , liquid nitrogen,	ultrasonication at a power	Toxic solvents, complex	S 1
Isopropanol (IPA), sodium	of 180 W for 5 h	process	
borohydride (NaBH4)			
Bulk MoS ₂ , BSA	ultra-sonication for 48 h	Longer times	S2
Bulk MoS ₂ , house-hold	solid probe sonicator, 3 h	Pretreatment at 40–	S 3
detergent		50 °C, stability of NSs	
		with detergent unknown	
Bulk MoS ₂ , ethanol/water	sonication for 8 h	Organic solvents	S4
Bulk MoS ₂ , N-methyl 2-	probe sonicator, for 2h	Toxic solvents	S5
pyrrolidone (NMP)			
Bulk MoS ₂ , polyacrylic	Probe sonication, tandem	Green solvent water	Present
acid	process		work

Table S1. Literature summary of methods for synthesizing the water soluble MoS2 nanosheets via sonication assisted methods by using bulk MoS₂.

Supporting references

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