

# Supplementary Materials: Wastewater Contaminated with Hydrazine as Scavenger Agent for Hydrogen Production by Cu/Ti Nanostructures

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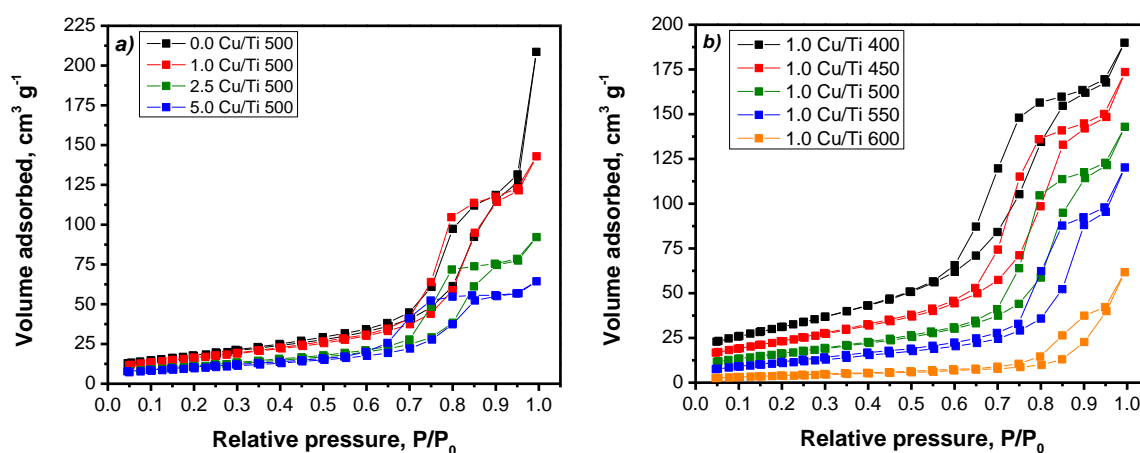
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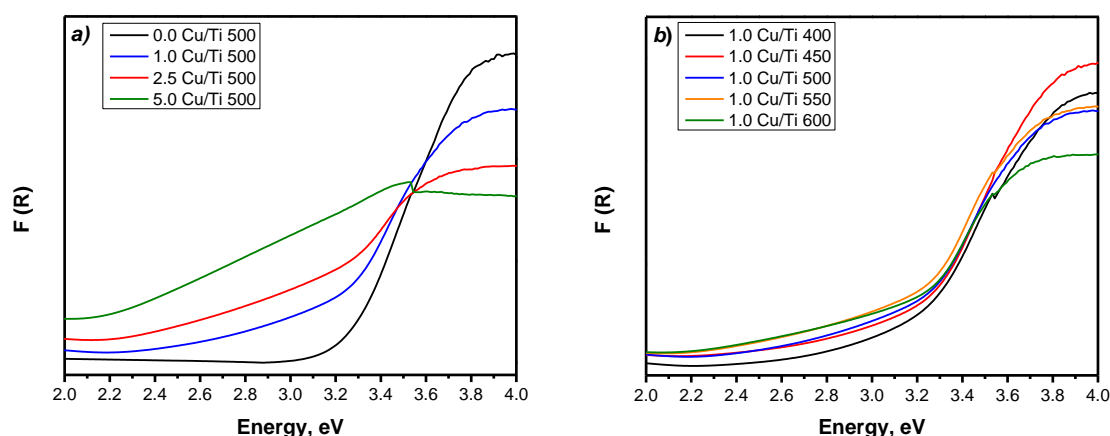
The band gap of the photocatalysts was calculated by linearization of the slope to the X axis (wavelength, nm) with the Y axis (absorbance) equal to zero. For practical purposes, the band gap energy for the different samples was calculated using the follow equation.

$$E_g = (1239 \times m) / (-b) \quad (1)$$

Where m and b are obtained by the linear fit ( $y = mx + b$ ) of the flat section of the UV-Vis spectrum [1].



**Figure S1.** Adsorption – desorption nitrogen isotherms of a) different copper loading Cu/Ti calcined at 500 °C, b) 1.0 Cu/Ti calcined at different temperatures.



**Figure S2.** UV-Vis spectra of a) different copper loading Cu/Ti calcined at 500 °C, b) 1.0 Cu/Ti calcined at different temperatures.

In the follow table S1 are presented some comparative examples of hydrogen production. It is important to mention that the  $577.9 \mu\text{mol g}^{-1} \text{h}^{-1}$  of hydrogen produced are in correspondence with the literature. Other copper materials produce  $427.8 \mu\text{mol g}^{-1} \text{h}^{-1}$  but using ethanol as scavenger, or other materials with precious metals as Pt produces a similar quantity of  $525.7 \mu\text{mol g}^{-1} \text{h}^{-1}$ . On the other hand, gold catalysts produces a grant quantity of hydrogen of  $2488 \mu\text{mol g}^{-1} \text{h}^{-1}$  by using similar experimental conditions. In this sense the results here obtained represent a great strategy to produce hydrogen by a hydrazine-wastewater.

**Table S1.** Some comparative examples of hydrogen production from current literature using several transition metals.

Material	Scavenger	Light source	H <sub>2</sub> produced ( $\mu\text{mol g}^{-1} \text{h}^{-1}$ )	Reference
0.05 Pt/TiO <sub>2</sub>	Pure water	250W high-pressure Hg lamp	525.7	[2]
Cu <sub>2</sub> O/TiO <sub>2</sub>	Methanol	300 W Xe lamp	985	[3]
Ag/SrTiO <sub>3</sub>	Ethanol	UV lamp (254 nm)	400	[4]
Cu@NT	Ethanol	UV lamp (254 nm)	427.8	[5]
Au/TiO <sub>2</sub>	Methanol	UV lamp 348/395 nm	2488	[6]
1-Cu/TiO <sub>2</sub>	Hydrazine	UV lamp (254 nm)	577.8	This work

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