

Supplementary information

Expanding the Family of Octahedral Chiral-at-Metal Cobalt(III) Catalysts by Introducing Tertiary Amine Moiety Into the Ligand

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Characterization of the complex $\Delta(R,R)$ -2

¹H-NMR (400 MHz, CDCl₃): δ = 7.57 (s, 2H, CH=N), 7.43–7.32 (m, 8H, ArH), 7.31–7.27 (m, 2H, ArH), 7.17 (d, J = 2.5 Hz, 2H, ArH), 7.09 (br.s, 2H, NH), 6.91 (d, J = 2.5 Hz, 2H, ArH), 5.10–4.90 (m, 2H, CH-N), 4.07–3.78 (m, 6H, CH₂-Ph, CH-N), 3.47 (t, J = 9.0 Hz, 2H, CH₂), 3.37 (t, J = 9.01 Hz, 2H, CH₂), 3.31 (t, J = 8.0 Hz, 2H, CH₂), 3.01 (t, J = 8.0 Hz, 2H, CH₂), 1.87 (br.s, 2H, NH), 1.20 (s, 18H, tBu), 0.95 (s, 18H, tBu) ppm.

¹³C-NMR (101 MHz, CDCl₃): δ = 163.4, 161.4, 143.0, 138.9, 137.2, 130.3, 128.6, 128.5, 128.2, 127.4, 117.8, 75.6, 61.6, 60.6, 50.8, 49.9, 35.3, 33.9, 31.4, 29.8 ppm.

Elem. anal. calcd. for C₅₂H₇₂CoN₆O₂Cl x 1.2H₂O (%): C, 67.22; H, 8.07; N, 9.04, found: C, 67.07; H, 8.04; N, 8.48.

$[\alpha]_D^{25} = -2292^\circ$ (c=0.02 in MeOH).

Characterization of the complex $\Delta(S,S)$ -2

¹H-NMR (400 MHz, CDCl₃): δ = 7.57 (s, 2H, CH=N), 7.43–7.32 (m, 8H, ArH), 7.31–7.27 (m, 2H, ArH), 7.17 (d, J = 2.5 Hz, 2H, ArH), 7.09 (br.s, 2H, NH), 6.91 (d, J = 2.5 Hz, 2H, ArH), 5.10–4.90 (m, 2H, CH-N), 4.07–3.78 (m, 6H, CH₂-Ph, CH-N), 3.47 (t, J = 9.0 Hz, 2H, CH₂), 3.37 (t, J = 9.01 Hz, 2H, CH₂), 3.31 (t, J = 8.0 Hz, 2H, CH₂), 3.01 (t, J = 8.0 Hz, 2H, CH₂), 1.87 (br.s, 2H, NH), 1.20 (s, 18H, tBu), 0.95 (s, 18H, tBu) ppm.

¹³C-NMR (101 MHz, CDCl₃): δ = 163.4, 161.4, 143.0, 138.9, 137.2, 130.3, 128.6, 128.5, 128.2, 127.4, 117.8, 75.6, 61.6, 60.6, 50.8, 49.9, 35.3, 33.9, 31.4, 29.8 ppm.

Elem. anal. calcd. for C₅₂H₇₂CoN₆O₂Cl x 0.8H₂O (%): C, 67.74; H, 8.05; N, 9.12, found: C, 67.73; H, 8.13; N, 8.79.

$[\alpha]_D^{25} = +2103^\circ$ (c=0.04 in MeOH).

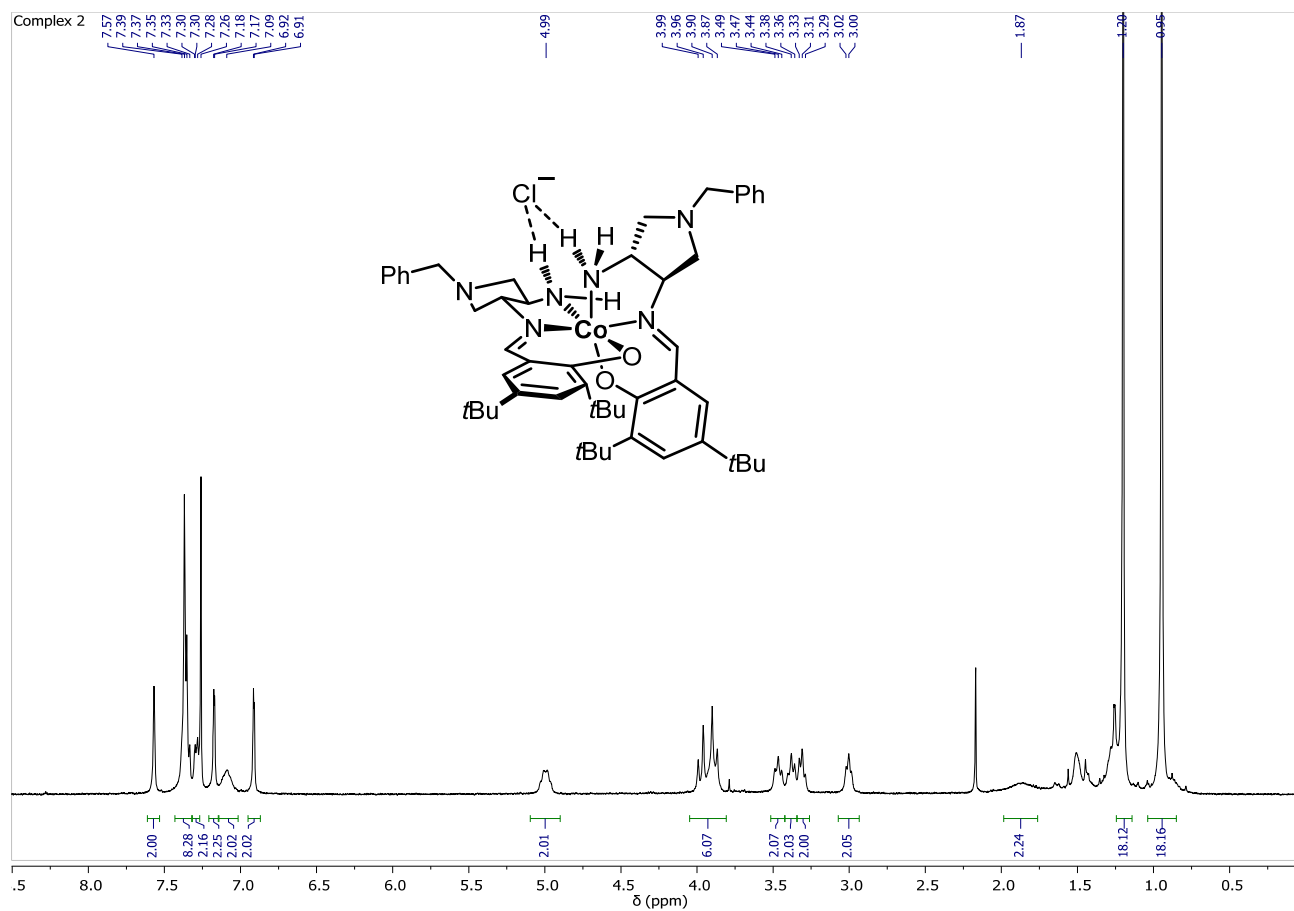


Figure S1. ¹H NMR spectrum of the complex 2.

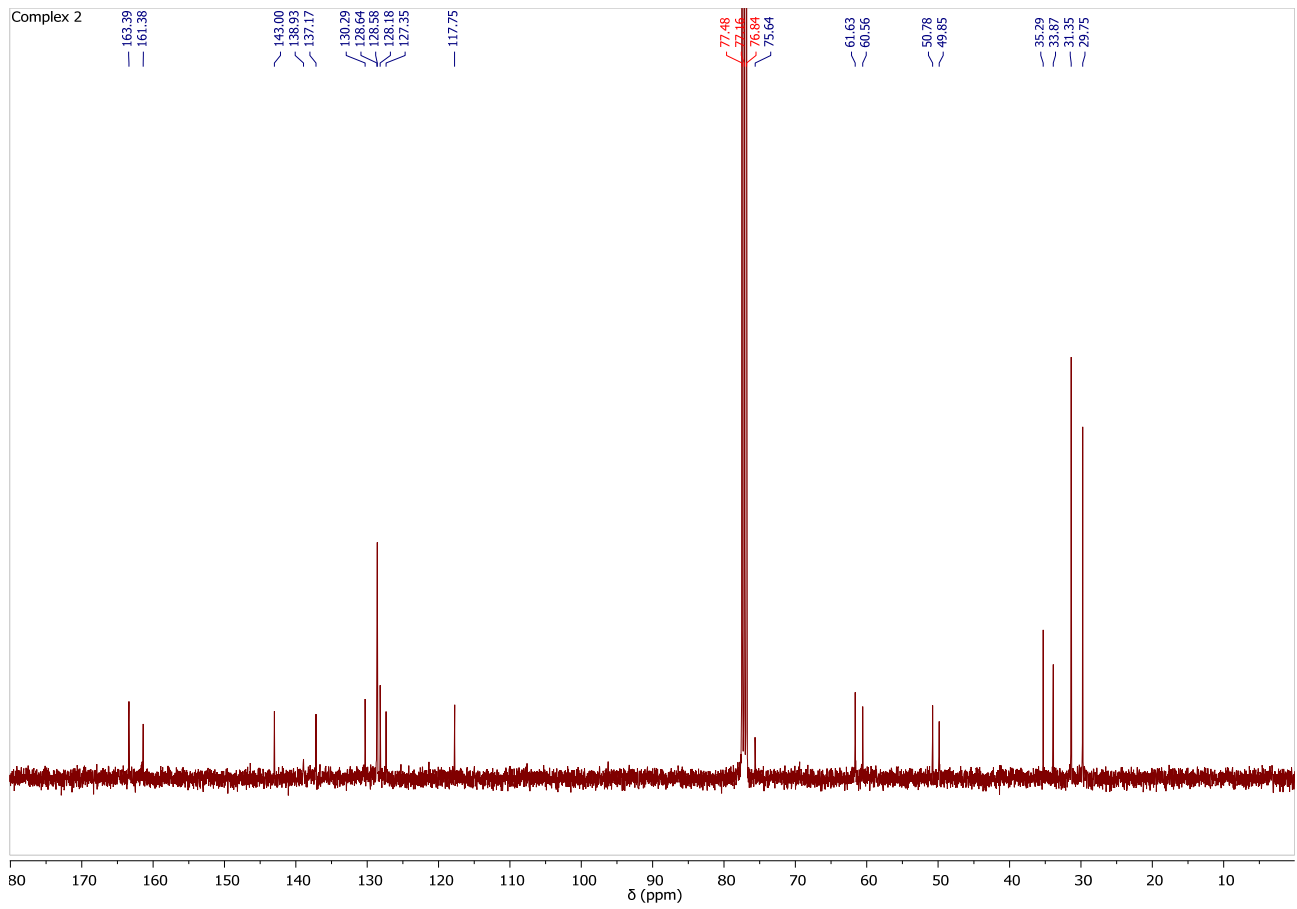


Figure S2. ^{13}C NMR spectrum of the complex 2.

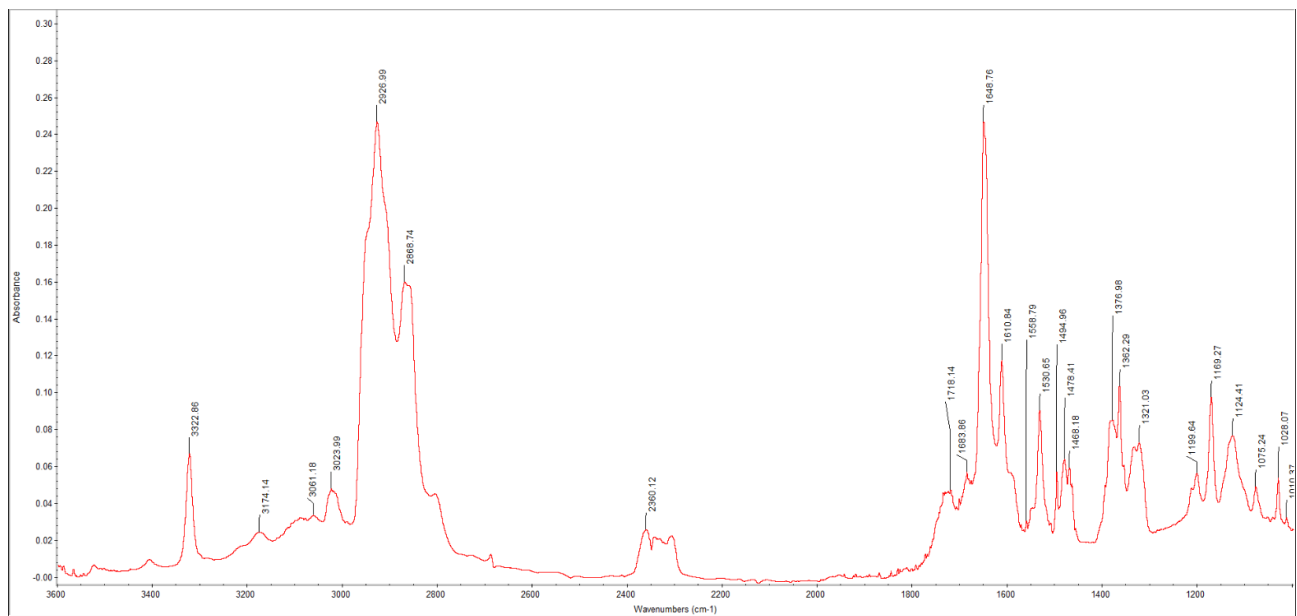


Figure S3. IR spectrum of the complex $\Lambda(R,R)$ -2 recorded in CH_2Cl_2 ($c = 4.4 \times 10^{-3} \text{ M}$).

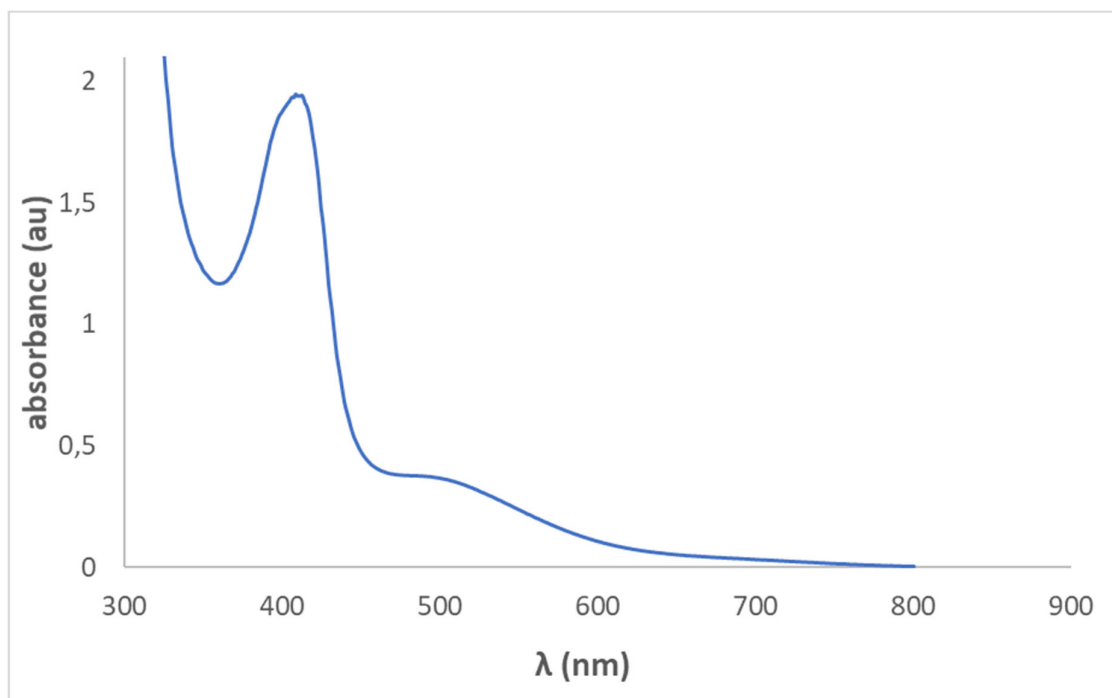


Figure S4. UV-vis spectrum of the complex $\Lambda(R,R)$ -2 recorded in CH_2Cl_2 ($c = 4.4 \times 10^{-3}$ M).

X-ray diffraction study of the Co(III) complex 2

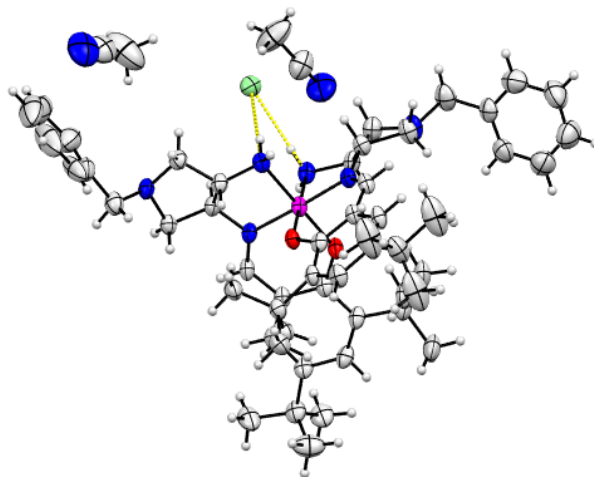


Figure S5. X-ray structure of the Co(III) complex 2 of Λ -configuration with co-crystallized acetonitrile molecules.

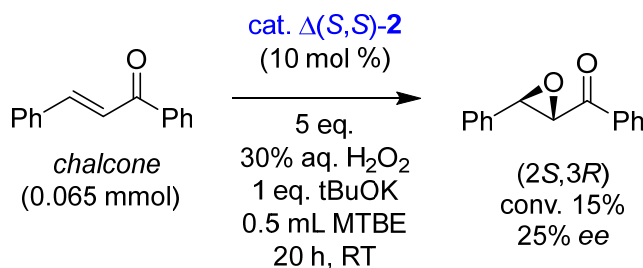
Table S1. Crystallographic data for the complex 2.

Data block	Complex 2
Brutto formula	C ₅₆ H ₇₈ ClCoN ₈ O ₂
Formula weight	989.64
Diffractometer	Bruker SMART APEX CCD area detector
Scan mode	ω and ϕ scans
Anode [Wavelength, Å]	MoK α [0.71073] sealed tube
Crystal Dimensions, mm	0.02 \times 0.18 \times 0.25
Crystal colour	brown
Crystal system	<i>triclinic</i>
a, Å	11.6512(15)
b, Å	14.7315(18)
c, Å	17.205(2)
α , °	90.658(3)
β , °	99.107(3)
γ , °	104.433(2)
Volume, Å ³	2819.7(6)
Density, gcm ⁻³	1.166
Temperature, K	120
T _{min} /T _{max}	0.6437/0.7461
μ , mm ⁻¹	0.397
Space group	<i>P</i> $\bar{1}$
Z	2
F(000)	1060
Reflections collected	27067
Independent reflections	11086
Reflections (I>2 σ (I))	6491
Parameters	628
R _{int}	0.0595
2 θ _{min} - 2 θ _{max} , °	3.634 - 52.000
wR ₂ (all reflections)	0.1858
R ₁ (I> σ (I))	0.0624
GOF	1.024
ρ_{min}/ρ_{max} , eÅ ⁻³	-0.467/0.835

Table S2. Selected bond distances and angles for the complex 2.

Distance	Å	Angle	°	Hydrogen Bond	Å
Co1-O1	1.888(3)	O1-Co1-N2	179.2(1)	H2B...Cl1	2.336
Co1-O2	1.888(2)	O2-Co1-N5	179.4(1)	H5B...Cl1	2.384
Co1-N1	1.903(3)	N1-Co1-N4	177.5(1)		
Co1-N2	1.993(3)	N2-Co1-N5	91.9(1)		
Co1-N4	1.905(3)				
Co1-N5	1.987(3)				

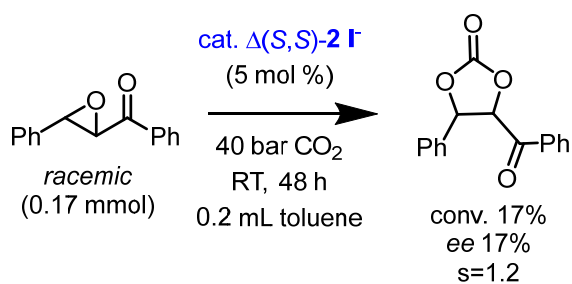
Evaluation of catalytic activity of the enantiopure complexes 2



Scheme S1. Epoxidation of chalcone.

The reaction was performed under our previously published conditions (ref. [13]).

(2*S*,3*R*)-Phenyl-3-phenyloxiran-2-yl)methanone: $^1\text{H NMR}$ (400 MHz, CDCl_3): δ = 8.01 (d, $J=7.5$ Hz, 2H), 7.62 (t, $J=7.4$ Hz, 1H), 7.49 (t, $J=7.7$ Hz, 2H), 7.46–7.31 (m, 5H), 4.31 (d, $J=1.8$ Hz, 1H), 4.08 (d, $J=1.6$ Hz, 1H) ppm. HPLC Kromasil 3-AmyCoat column; 25 °C, hexane/isopropanol 95:5, λ = 254 nm, 1 mL/min, t_R = 10.1 (major), 10.9 min.

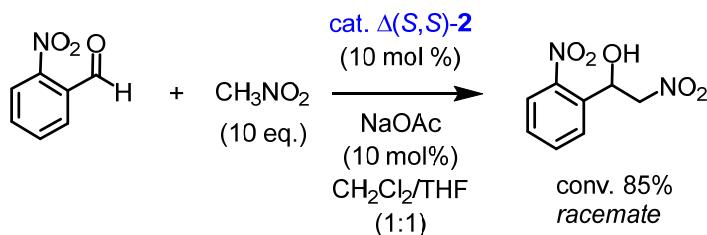


Scheme S2. Kinetic resolution of chalcone epoxide with CO_2 .

The reaction was performed under our previously published conditions (ref. [13]).

4-Benzoyl-5-phenyl-1,3-dioxolan-2-one: $^1\text{H NMR}$ (400 MHz, CDCl_3): δ = 7.96 (d, $J=7.5$ Hz, 2H), 7.66 (t, $J=7.5$ Hz, 1H), 7.55–7.43 (m, 7H), 5.98 (d, $J=6.2$ Hz, 1H), 5.60 (d, $J=6.2$ Hz, 1H) ppm. HPLC CHIRALCEL OD-H column; 25 °C, heptane/isopropanol 90:10, λ = 220 nm, 1 mL/min, t_R = 18.0, 33.1 (major) min. The iodide complex **2** was obtained according to our published procedure (*ChemSusChem* 2016, 9, 216) and used without further purification and characterization.

S factor calculated as $(\ln[1-c(1+ee)])/(\ln[1-c(1-ee)])$ where c is the conversion and ee is the enantiomeric excess of the cyclic carbonate product.

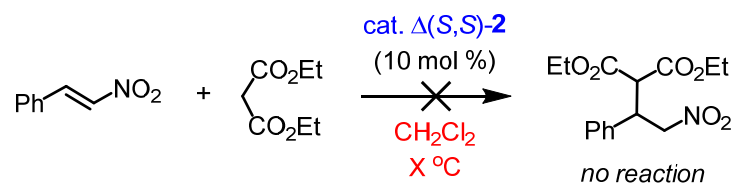


Scheme S3. Asymmetric Henry reaction of the *o*-nitrobenzaldehyde with the nitromethane.

The reaction was performed under our previously published conditions (*Inorg. Chem.* 2019, 58, 11051).

2-Nitro-1-(2-nitrophenyl)ethanol: $^1\text{H NMR}$ (400 MHz, CDCl_3): δ = 8.11–8.08 (m, 1H), 7.98–7.96 (m, 1H), 7.78–7.74 (m, 1H), 7.59–7.55 (m, 1H), 6.07 (ddd, $J=8.8, 4.2, 2.2$ Hz, 1H), 4.89 (dd, $J=13.9, 2.2$ Hz, 1H), 4.57 (dd, $J=13.9, 8.8$ Hz, 1H), 3.15 (d, $J=4.2$ Hz, 1H) ppm. HPLC

Kromasil 3-AmyCoat column; 25 °C, heptane/isopropanol 90:10, λ = 254 nm, 1 mL/min, t_R = 11.0, 12.7 min.



Scheme S4. Michael addition of dimethyl malonate to *trans*- β -nitrostyrene.

The reaction was performed under published conditions (ref. [24]).