

Supporting Information

Article

Continuous-Flow Sunlight-Powered CO₂ Methanation Catalyzed by γ -Al₂O₃-Supported Plasmonic Ru Nanorods

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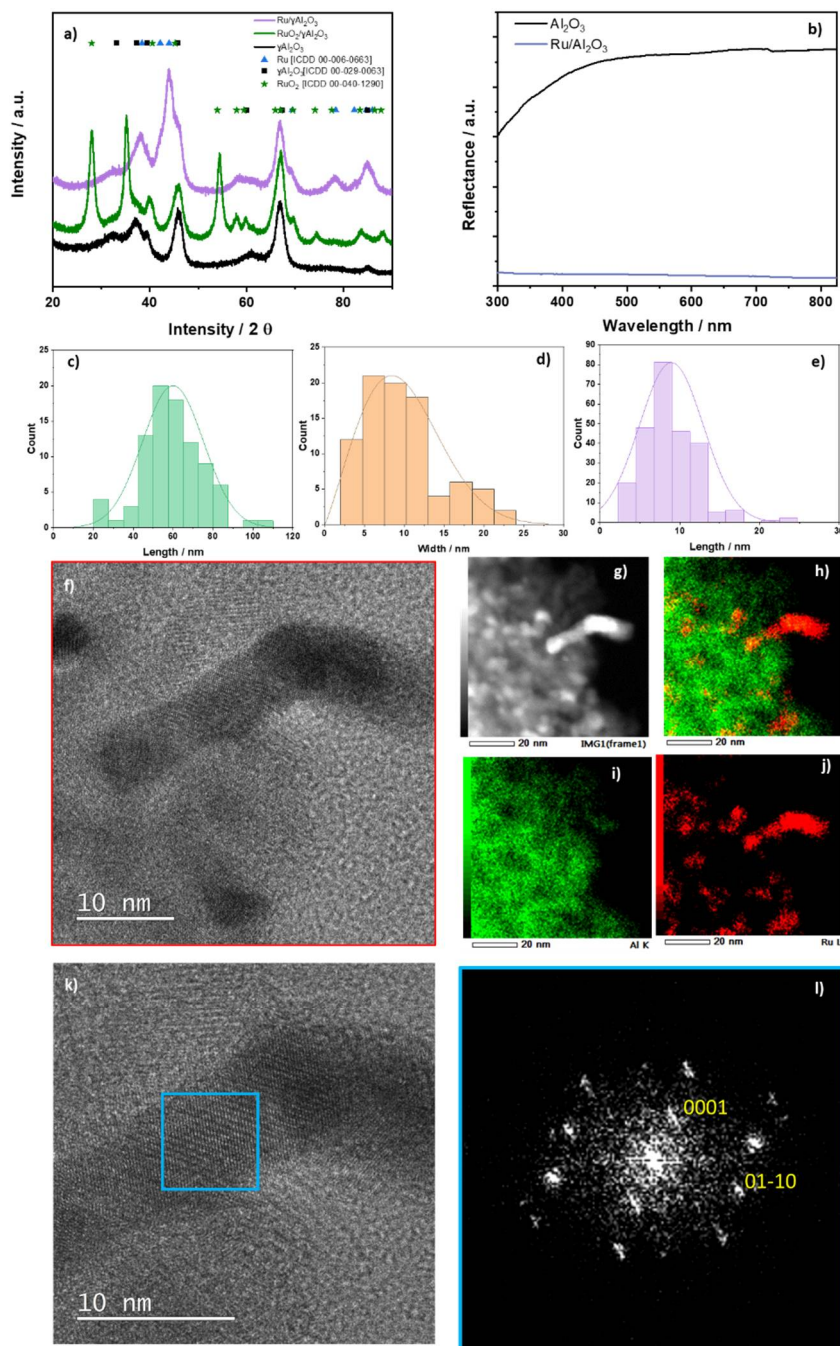


Figure S1. Optical and structural characterization of the Ru/ γ -Al₂O₃ catalyst: **a)** XRD pattern of the Ru/ γ -Al₂O₃ (purple line), RuO₂/ γ -Al₂O₃ (green line) and γ -Al₂O₃ (black line) containing the characteristic peak positions for the corresponding Ru (blue triangles, ICDD 00-006-0663), RuO₂ (green triangles, ICDD 00-040-1290) and γ -Al₂O₃ (black squares, ICDD 00-029-0063) **b)** diffuse reflectance UV-vis spectra **c)** the size distribution histogram of the Ru NRs determined an average NR length of 59.8 nm, **d)** the size distribution histogram an average NR width of 9.9 nm, **e)** the size distribution histogram of the Ru spheres determined an average 8.9 nm, **f)** HRTEM image of Ru on γ -Al₂O₃ **g)** HAADF-STEM image, **h-j)** corresponding EDX elemental maps of Al (green) and Ru (red). **k)** HRTE images of a Ru nanorod. **l)** FFT pattern of the area selected in (k), corresponding to a <2-1-10> zone axis pattern of the hexagonal Ru crystal structure.

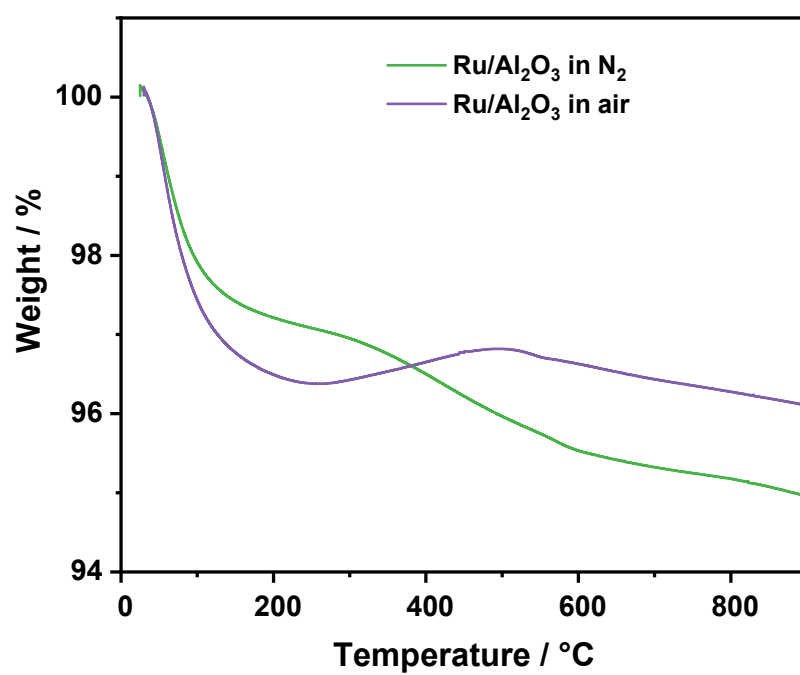


Figure S2. Thermogravimetric analysis for Ru/Al₂O₃ under N₂ or air atmosphere.

Equipment for the CO₂ photomethanation reactions

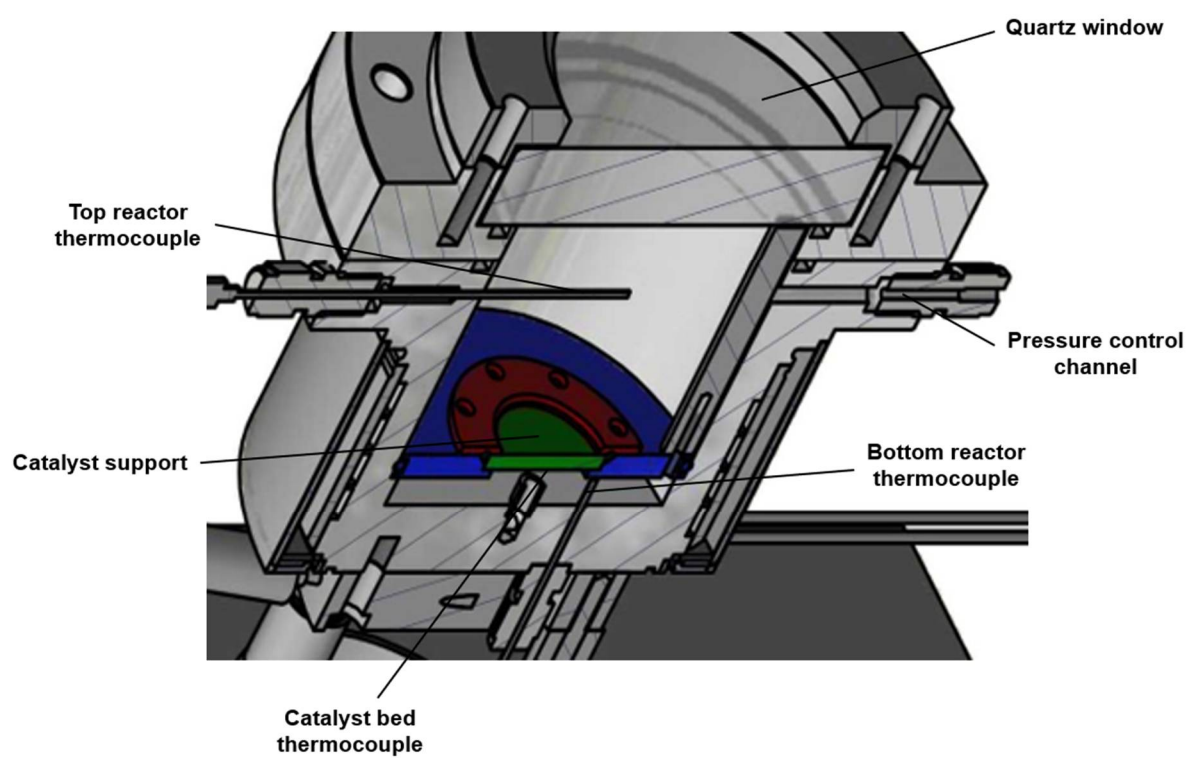


Figure S3. Schematic representation of the photoreactor.

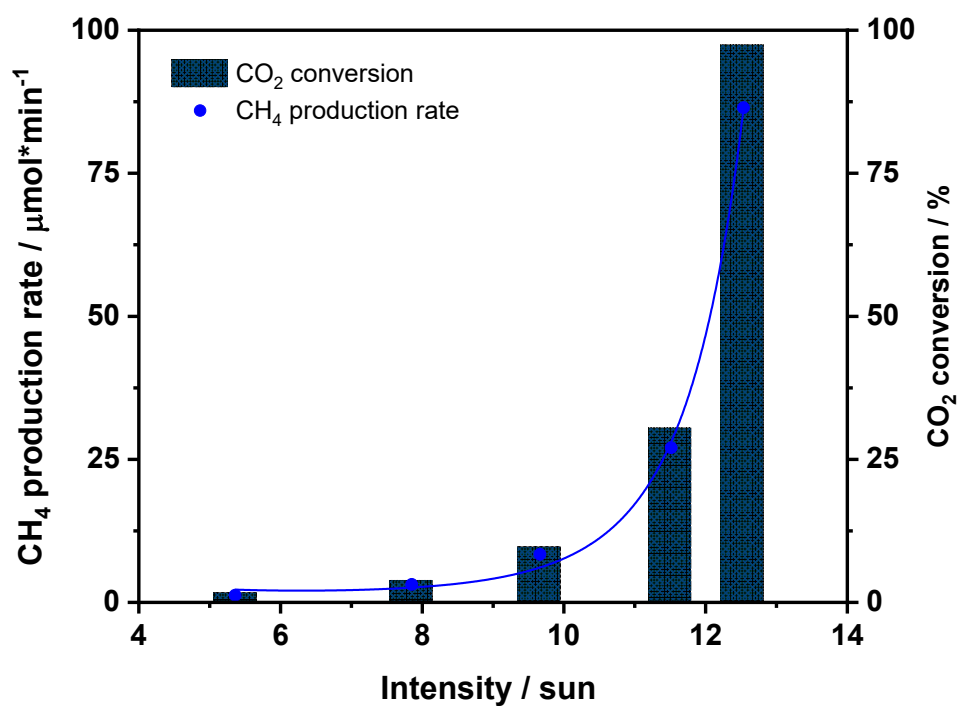


Figure S4. CH₄ production rate and CO₂ conversion as function of light intensity for the CO₂ photomethanation without external heating. Reaction conditions: mixture of H₂:CO₂:N₂ (4.5:1:1) with a flow of (9:2:2) ml·min⁻¹ at 1.5 bar pressure (10wt% Ru/γ-Al₂O₃, 200 mg total catalyst mass, AM1.5 irradiance (1 sun = 1 kW·m⁻²)).

Equation for the (apparent) activation energy calculation

Arrhenius equation:

$$R_{CO} = A \cdot e^{\left(\frac{-E_a}{RT}\right)} \quad (\text{eq. S1})$$

A = Pre-exponential factor.

E_a = Activation Energy.

R = Gas constant (8.314 J·K⁻¹·mol⁻¹).

T = Reaction temperature.

Table S1. Photocatalytic CH₄ production and reaction conditions

Ref	Catalyst	Year	Mode	Lamp		Irradiation		Reactor T (°C)	Catalyst T (°C)	Products	
				Power (W)	Type	Filters	Intensity (mW/cm ²)			CH ₄	CO
[1]	Ru NPs/TiO ₂	1987	Batch	--	--	Solar simulator	80	RT	90	10 μmol·h ⁻¹	--
[2]	Ni/SiO ₂ -Al ₂ O ₃	2014	Batch	--	--	AM 1.5	--	RT	150	54.6 mmol.g _{cat} ⁻¹ .h ⁻¹	94.9%CO ₂ conv. 2.8% Selec.
[3]	Ru/AlO ₃	2014	Batch	300	Xe	--	--	RT	300-400	95.75% CO ₂ conv. 99.22% Selec.	95.75% CO ₂ conv. 0.78% Selec.
[3]	Rh/Al ₂ O ₃	2014	Batch	300	Xe	--	--	RT	300-400	96.25% CO ₂ conv. 99.48% Selec.	96.25% CO ₂ conv. 0.52% Selec.
[3]	Ni/Al ₂ O ₃	2014	Batch	300	Xe	--	--	RT	300-400	93.25% CO ₂ conv. 99.04% Selec.	93.25% CO ₂ conv. 0.25% Selec.
[4]	Rh/Al ₂ O ₃	2017	Batch	--	LED	UV	3000	350	--	7 μmol·s ⁻¹ g ⁻¹	--
[5]	Cu ₂ O/Graphene	2017	Batch	300	Xe	--	200	250	--	14.93 mmol.g _{Cu₂O} ⁻¹ .h ⁻¹	--
[6]	RuO ₂ /i-Si-o	2018	Batch	300	Xe	--	2200	RT	170	4.4 mmol g _{cat} ⁻¹ min ⁻¹	--
[7]	Ru/SiO ₂	2018	Batch	300	Xe	--	2470	RT	150	2.8 mmol g ⁻¹ min ⁻¹	
[8]	Ru NR/Al ₂ O ₃	2019	Batch	300	Xe	AM 1.5	1010	25	221	52 mmol.g _{Ru} ⁻¹ .h ⁻¹	--
[9]	Ru NPs/Al ₂ O ₃	2020	Batch	300	Xe	AM 1.5	6.2	150	220	5.09 mol.g _{Ru} ⁻¹ .h ⁻¹	--
[10]	Ni	2021	Batch	--	--	--	2000	RT	100	0.09 mmol ⁻¹ .h ⁻¹	

[11]	Ni/SiO ₂ -Al ₂ O ₃	2017	Flow	300	Xe	--	232.7	225	--	275 $\mu\text{L}\cdot\text{g}_{\text{cat}}^{-1}\cdot\text{min}^{-1}$ (3.5% CO ₂ conversion)	--
[12]	Ru@LDHs	2017	Flow	300	Xe	--	1000	RT	350	277 $\text{mmol}\cdot\text{g}_{\text{cat}}^{-1}\cdot\text{min}^{-1}$	--
This publication	Ru NR/Al ₂ O ₃		Flow	300	Xe	AM1.5	1450	25	204	792 $\text{mmol}\cdot\text{g}_{\text{Ru}}^{-1}\cdot\text{h}^{-1}$	

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