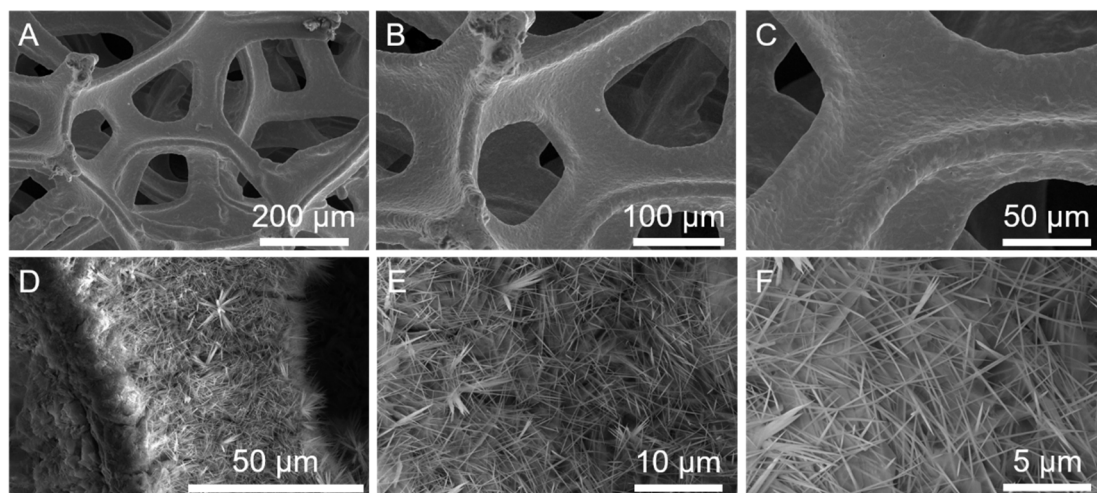
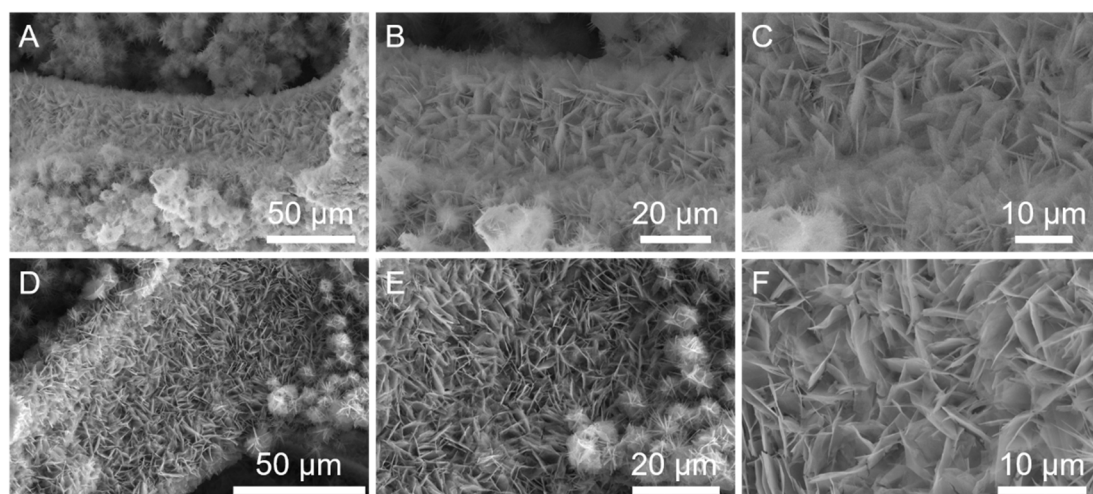


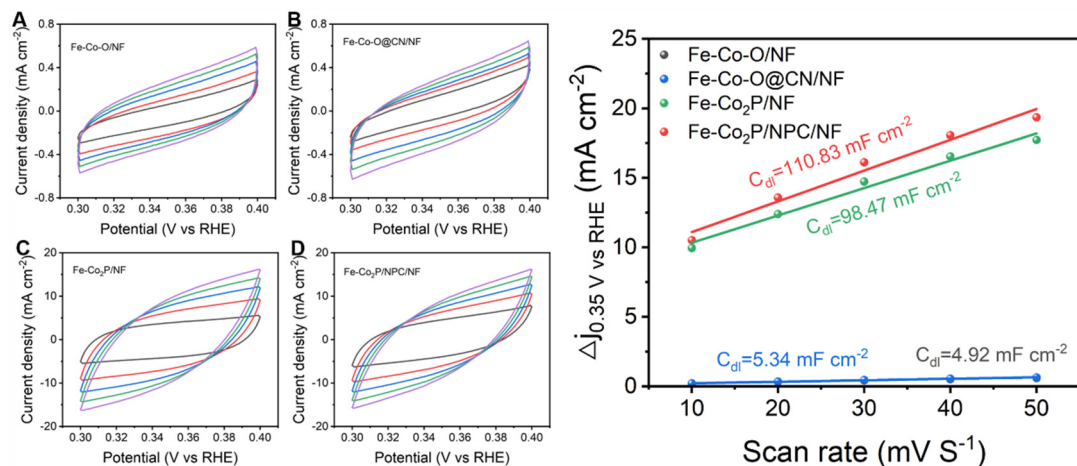
## Supporting Information



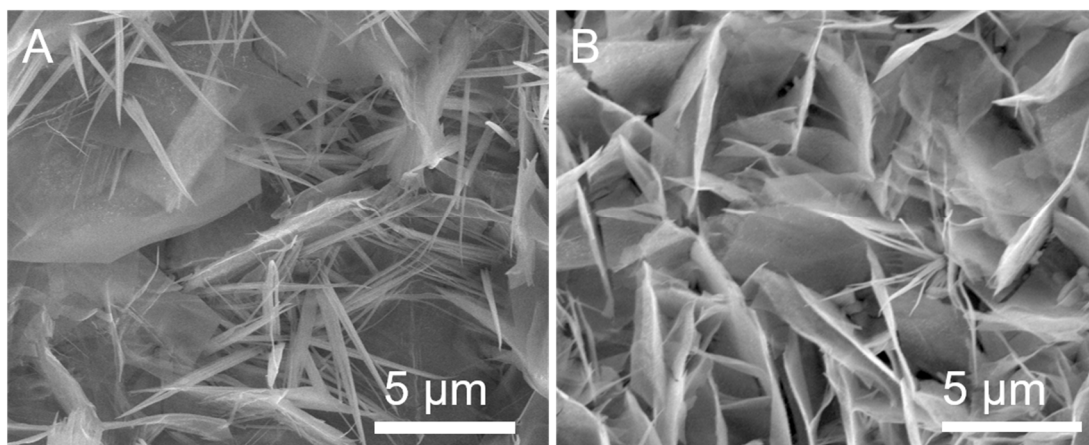
**Figure S1.** SEM images of different catalysts. (A-C) Nickel foam. (D-F) Fe-Co-O precursor/NF.



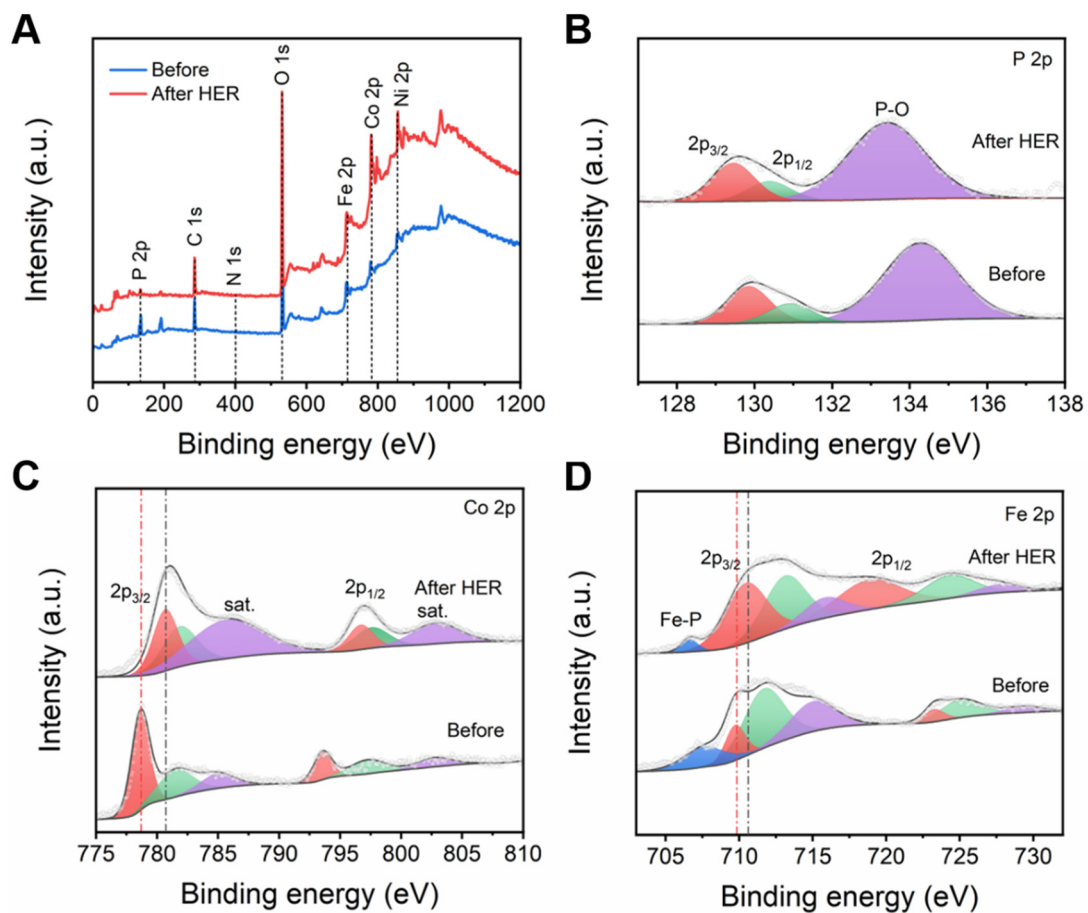
**Figure S2.** SEM images of different catalysts. (A-C) Fe-Co-O@CN/NF. (D-F) Fe-Co<sub>2</sub>P/NPC/NF.



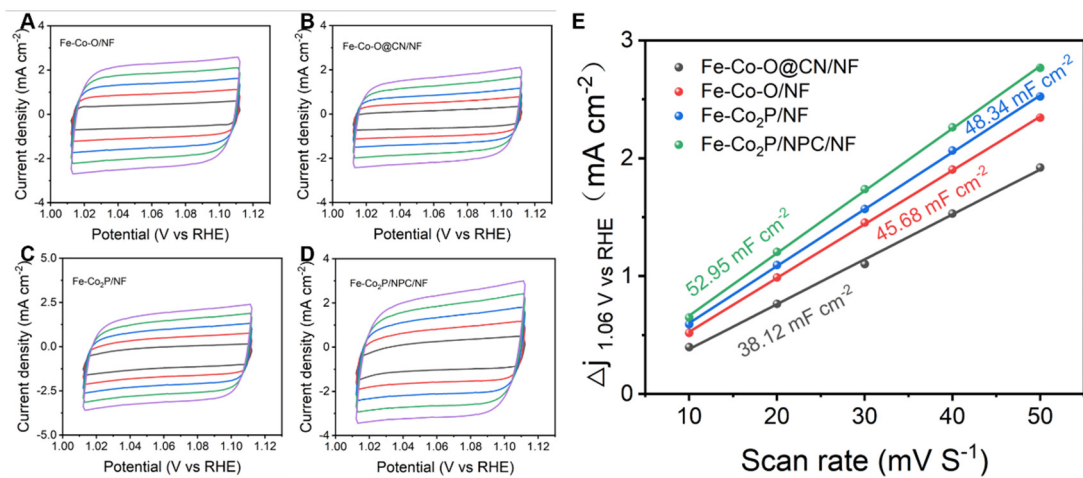
**Figure S3.** Cyclic voltammograms (CV) curves of different catalysts between 0.3 to 0.4 V vs RHE at five different scan rates (10, 20, 30, 40, and 50 mV s<sup>-1</sup>) and electrochemical double layer capacitances ( $C_{dl}$ ). (A) Fe-Co-O/NF. (B) Fe-Co-O@CN/NF. (C) Fe-Co<sub>2</sub>P/NF. (D) Fe-Co<sub>2</sub>P/NPC/NF. (E) The current density variation ( $\Delta j = j_a - j_c$ ) at the potential of 0.35 V vs RHE with varied scan rates fitted to estimate the electrochemical double layer capacitances ( $C_{dl}$ ).



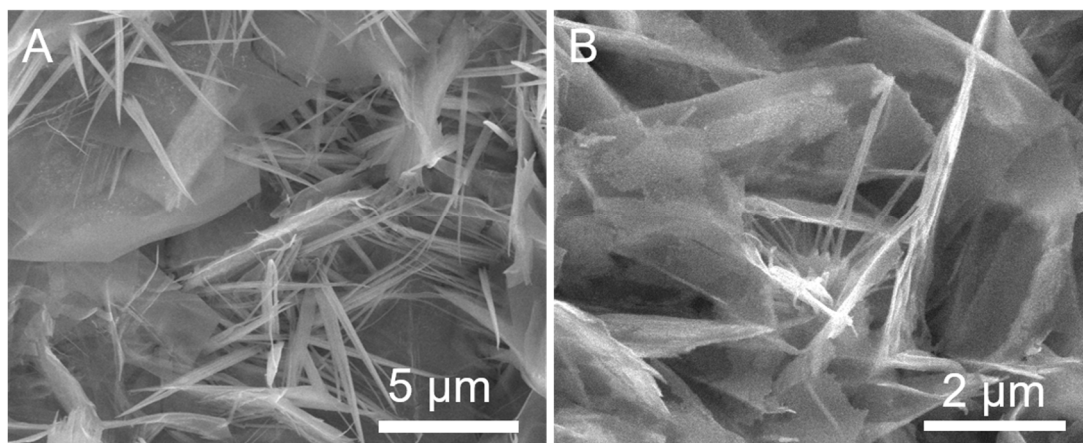
**Figure S4.** (A) SEM patterns of Fe-Co<sub>2</sub>P/NPC/NF before the HER stability test. (B) after the HER stability test.



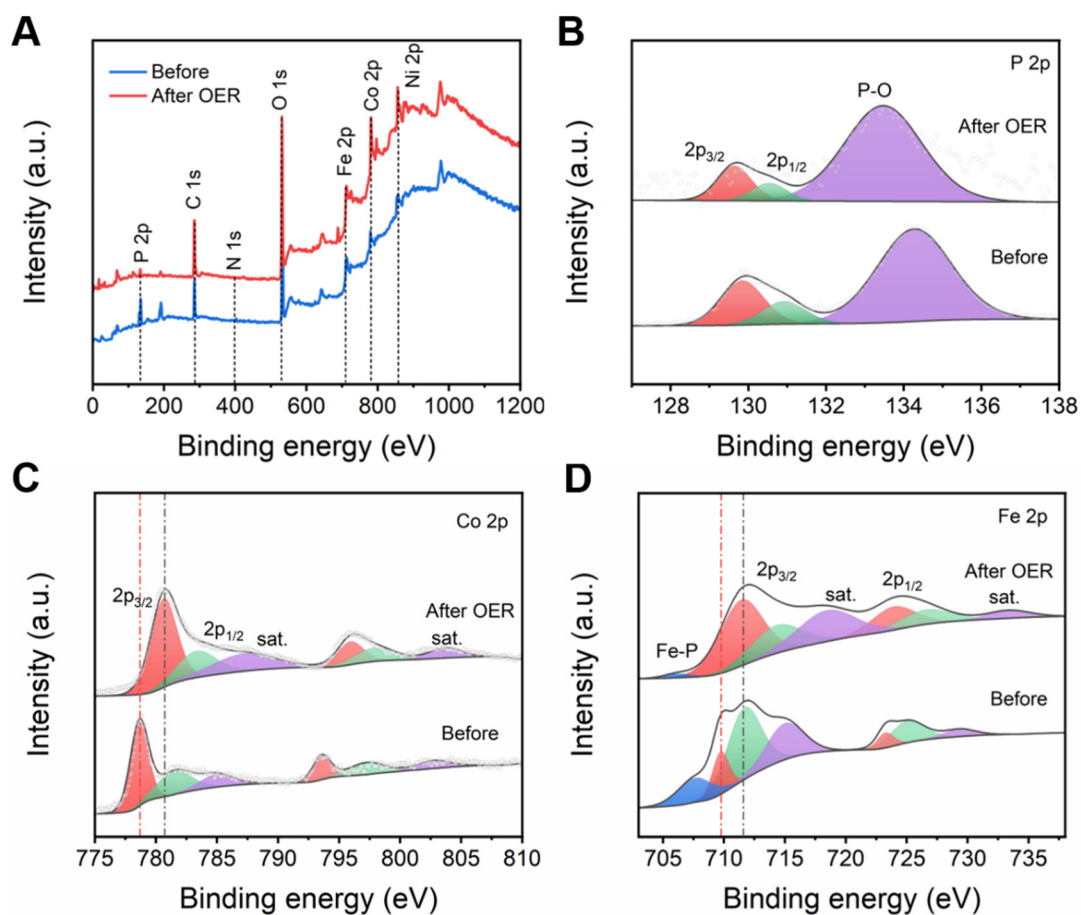
**Figure S5.** (A) XPS scanning full spectrum of Fe-Co<sub>2</sub>P/NPC/NF before and after HER stability test. (B) P 2p, (C) Co 2p and (D) Fe 2p.



**Figure S6.** Cyclic voltammograms (CV) curves of different catalysts between 1.01 to 1.11 V vs RHE at five different scan rates (10, 20, 30, 40, and 50  $\text{mV s}^{-1}$ ) and electrochemical double layer capacitances ( $C_{dl}$ ). (A) Fe-Co-O/NF. (B) Fe-Co-O@CN/NF. (C) Fe-Co<sub>2</sub>P/NF. (D) Fe-Co<sub>2</sub>P/NPC/NF. (E) The current density variation ( $\Delta j = j_a - j_c$ ) at the potential of 1.06 V vs RHE with varied scan rates fitted to estimate the electrochemical double layer capacitances ( $C_{dl}$ ).

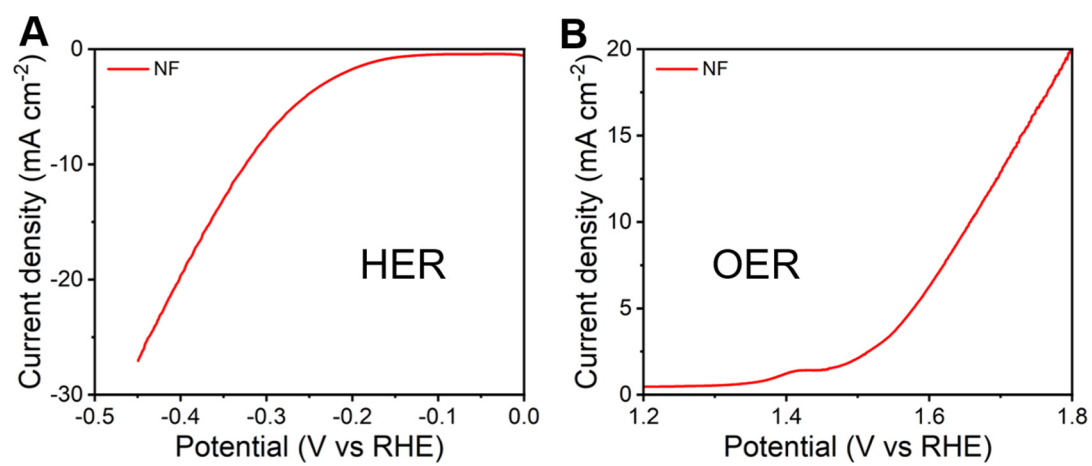


**Figure S7.** (A) SEM patterns of Fe-Co<sub>2</sub>P/NPC/NF before the OER stability test, (B) after the OER stability test.



**Figure S8.** (A) XPS scanning full spectrum of Fe-Co<sub>2</sub>P/NPC/NF before and after OER stability test. (B) P 2p, (C) Co 2p and (D) Fe 2p.





**Figure S9.** polarization curves of NF. (A) HER. (B) OER.

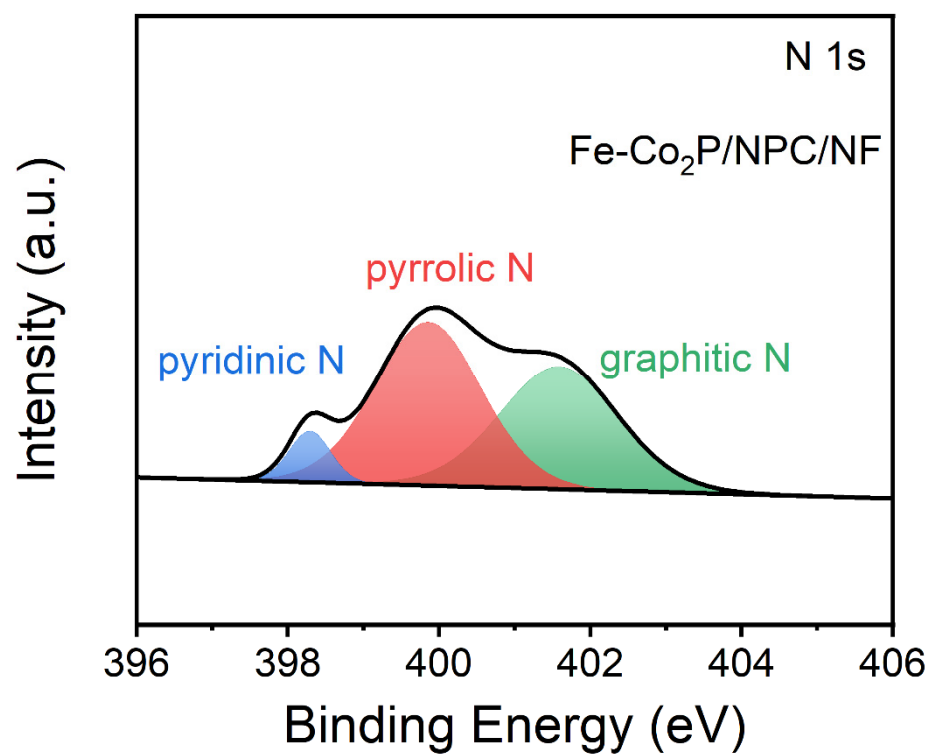
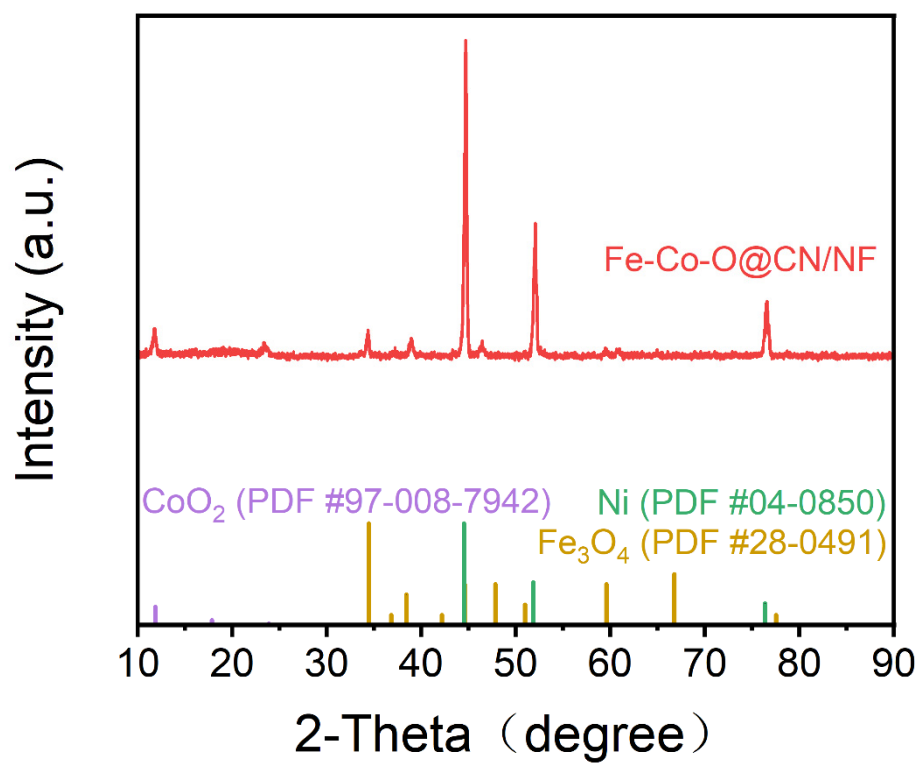


Figure S10. N 1s XPS result of Fe-Co<sub>2</sub>P/NPC/NF.



**Figure S11.** XRD patterns of Fe-Co-O@CN/NF.

**Table S1.** HER performances of different electrocatalysts in alkaline media (1 M KOH). The Fe-Co<sub>2</sub>P/NPC is superior to previously reported low-cost and Co-based HER catalysts that lack the introduction of a noble metal.

Catalyst	$\eta_{10}(\text{mV})$ ( $j=10 \text{ mA cm}^{-2}$ )	$\eta_{100}(\text{mV})$ ( $j=100 \text{ mA cm}^{-2}$ )	Tafel slope ( $\text{mV dec}^{-1}$ )	Ref.
Fe-Co <sub>2</sub> P/NPC	73	185	92.7	<b>This work</b>
Co-P@PC	76	-	49	[46]
Co <sub>x</sub> P/N-doped C	187	-	58.5	[47]
CoP/NCNHP	115	-	66	[48]
CoP <sub>h</sub> /NG	83	-	57	[49]
NiCoFeP/C	149	-	89	[50]
CoFeP/CNT	178	-	71	[51]
NiCo <sub>2</sub> P <sub>x</sub> /CNT	47	-	67.3	[52]
Co <sub>2</sub> P/Co	157	-	59	[53]
Fe <sub>3</sub> O <sub>4</sub> -CoP <sub>x</sub> /TiN	174	-	65	[54]
Co-Pi/CoP/Ti	68	-	-	[55]
N-CoO@CoP	-	201	37	[56]
CoP-FeP	71	-	67	[57]
f-CoP/CoP <sub>2</sub> /Al <sub>2</sub> O <sub>3</sub>	138	-	73	[58]
CoP-Mo <sub>2</sub> C@NC/CC	74	-	79.4	[59]

**Table S2.** OER performances of different electrocatalysts in alkaline media (1 M KOH). The Fe-Co<sub>2</sub>P/NPC is superior to previously reported low-cost and Co-based OER catalysts that lack the introduction of a noble metal.

Catalyst	$\eta_{10}(\text{mV})$ ( $j=10 \text{ mA cm}^{-2}$ )	$\eta_{100}(\text{mV})$ ( $j=100 \text{ mA cm}^{-2}$ )	Tafel slope ( $\text{mVdec}^{-1}$ )	Ref.
Fe-Co <sub>2</sub> P/NPC	217	262	38.4	<b>This work</b>
Co-P@PC	282	-	53	[46]
Co <sub>x</sub> P/N-doped C	380	-	68.1	[47]
CoP/NCNHP	310	-	70	[48]
CoP <sub>h</sub> /NG	262	-	54	[49]
NiCoFeP/C	270	-	65	[50]
CoFeP/CNT	323	-	38	[51]
NiCo <sub>2</sub> P <sub>x</sub> /CNT	284	-	56	[52]
Co <sub>2</sub> P/Co	319	-	79	[53]
Fe <sub>3</sub> O <sub>4</sub> -CoP <sub>x</sub> /TiN	331	-	122	[54]
Co-Pi/CoP/Ti	310	-	58	[55]
N-CoO@CoP	-	332	81.5	[56]
CoP-FeP	250	-	131	[57]
f-CoP/CoP <sub>2</sub> /Al <sub>2</sub> O <sub>3</sub>	300	-	63	[58]
CoP-Mo <sub>2</sub> C@NC/CC	265	-	71.6	[59]

**Table S3.** Overall water splitting performances of different electrocatalysts in alkaline media (1 M KOH). The Fe-Co<sub>2</sub>P/NPC is superior to previously reported low-cost and Co-based water splitting catalysts that lack the introduction of a noble metal.

Catalyst	Cell voltage(V) ( <i>j</i> =10mA cm <sup>-2</sup> )	Cell voltage(V) ( <i>j</i> =100 mA cm <sup>-2</sup> )	Stablity (h)	Ref.
Fe-Co <sub>2</sub> P/NPC	1.56	1.68	20	<b>This work</b>
Co-P@PC	1.6	-	60	[46]
Co <sub>x</sub> P/N-doped C	1.71	-	10	[47]
CoP/NCNHP	1.64	-	36	[48]
CoP <sub>h</sub> /NG	1.58	-	65	[49]
NiCoFeP/C	1.60	-	18	[50]
CoFeP/CNT	1.74	-	20	[51]
NiCo <sub>2</sub> P <sub>x</sub> /CNT	1.61	-	48	[52]
Co <sub>2</sub> P/Co	1.71	-	15	[53]
Fe <sub>3</sub> O <sub>4</sub> -CoP <sub>x</sub> /TiN	1.75	-	11	[54]
Co-Pi/CoP/Ti	1.60	-	24	[55]
N-CoO@CoP	1.79	-	50	[56]
CoP-FeP	1.55	-	24	[57]
f-CoP/CoP <sub>2</sub> /Al <sub>2</sub> O <sub>3</sub>	1.65	-	24	[58]
CoP-Mo <sub>2</sub> C@NC/CC	1.64	-	40	[59]