

1 Article

2 **The Roles of Graphene and Ag in the Hybrid**
 3 **Ag@Ag₂O-Graphene for Sulfamethoxazole**
 4 **Degradation**

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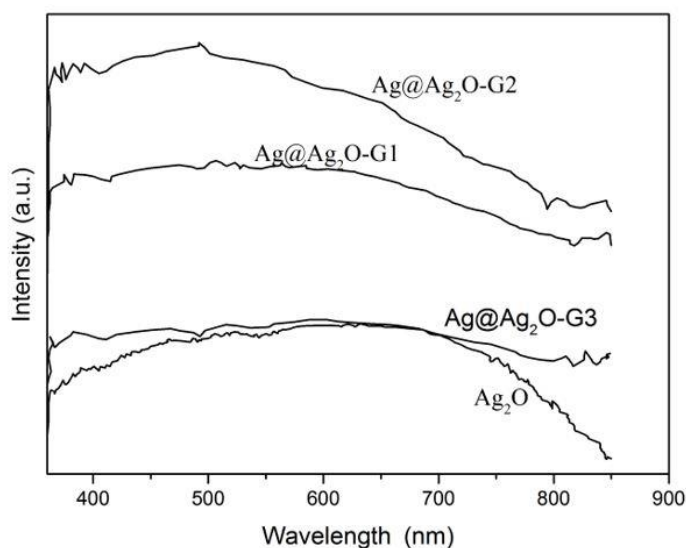
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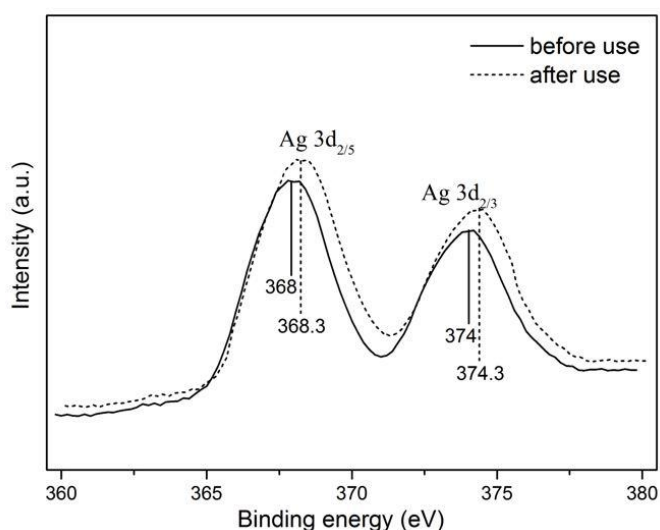
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15 **Figure S1.** UV-vis spectra of Ag₂O, Ag@Ag₂O-G1, Ag@Ag₂O-G2 and Ag@Ag₂O-G3.

16 The UV-vis diffuse reflectance spectra of the series photocatalysts are illustrated in Figure S1.
 17 The pure Ag₂O showed strong absorption in ultraviolet and visible-light region that should be due
 18 to its narrow band gap [1]. After loading different amount of graphene and Ag⁰ on the Ag₂O, there are
 19 apparent increase in the light absorption for Ag@Ag₂O-G1 and Ag@Ag₂O-G2, while the Ag@Ag₂O-G3
 20 showed slight increase in the light absorption, which may be owing to the too much graphene loading [2].
 21 It should be noted that the above light absorption order was consistent with the photocatalytic
 22 activities of photocatalysts, suggested that the increase light response was helpful for the
 23 enhancement of the photocatalytic performance.

24 **Table S1.** The rate constants and R² of first order kinetic reaction for Ag₂O, Ag@Ag₂O-G1, Ag@Ag₂O-
 25 G2 and Ag@Ag₂O-G3 to degrade SMX.

Light Source	Simulated Solar Light				Visible Light			
	Ag ₂ O	Ag@Ag ₂ O-G1	Ag@Ag ₂ O-G2	Ag@Ag ₂ O-G3	Ag ₂ O	Ag@Ag ₂ O-G1	Ag@Ag ₂ O-G2	Ag@Ag ₂ O-G3
Photocatalysts K×10 ⁻² (min ⁻¹)	1.57	3.53	4.51	2.69	1.22	3.40	3.79	2.78
R ²	0.965	0.982	0.955	0.978	0.984	0.989	0.953	0.973



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Figure S2. XPS spectra of Ag3d in Ag@Ag₂O-G2 before and after recycle use.

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Compared with the XPS spectrum of Ag 3d in the original Ag@Ag₂O-G2 (368 eV and 374 eV for Ag 3d_{2/5} and Ag 3d_{2/3}, respectively), the Ag 3d of used Ag@Ag₂O-G2 shifted to a higher binding energy (368.3 eV and 374.3 eV for Ag 3d_{2/5} and Ag 3d_{2/3}, respectively, as shown in Figure S2). Because of the higher binding energy of Ag 3d electrons of Ag⁰ than that of Ag⁺, the shift of the Ag 3d peak may indicate the formation of more Ag⁰ during the photocatalytic process [3], which could explain the change of adsorption ability and photocatalytic activities of Ag@Ag₂O-G2.

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1. Xu, Y.; Schoonen, M.A.A. The absolute energy positions of conduction and valence bands of selected semiconducting minerals. *Am. Mineral.* **2000**, *85*, 543–556, doi:10.2138/am-2000-0416.
2. Ma, Y.; Wang, J.; Xu, S.; Feng, S.; Wang, J. Ag₂O/sodium alginate-reduced graphene oxide aerogel beads for efficient visible light driven photocatalysis. *Appl. Surf. Sci.* **2018**, *430*, 155–164, doi:10.1016/j.apsusc.2017.03.299.
3. Yu, J.; Dai, G.; Huang, B. Fabrication and Characterization of Visible-Light-Driven Plasmonic Photocatalyst Ag/AgCl/TiO₂ TiO₂ Nanotube Arrays. *J. Phys. Chem. C* **2009**, *113*, 16394–16401, doi:10.1021/jp905247j.