catalysts

Supporting information

Photocatalytic Hydrogen Evolution Using Bi-Metallic (Ni/Pt) $Na_2Ti_3O_7$ Whiskers: Effect of the Deposition Order

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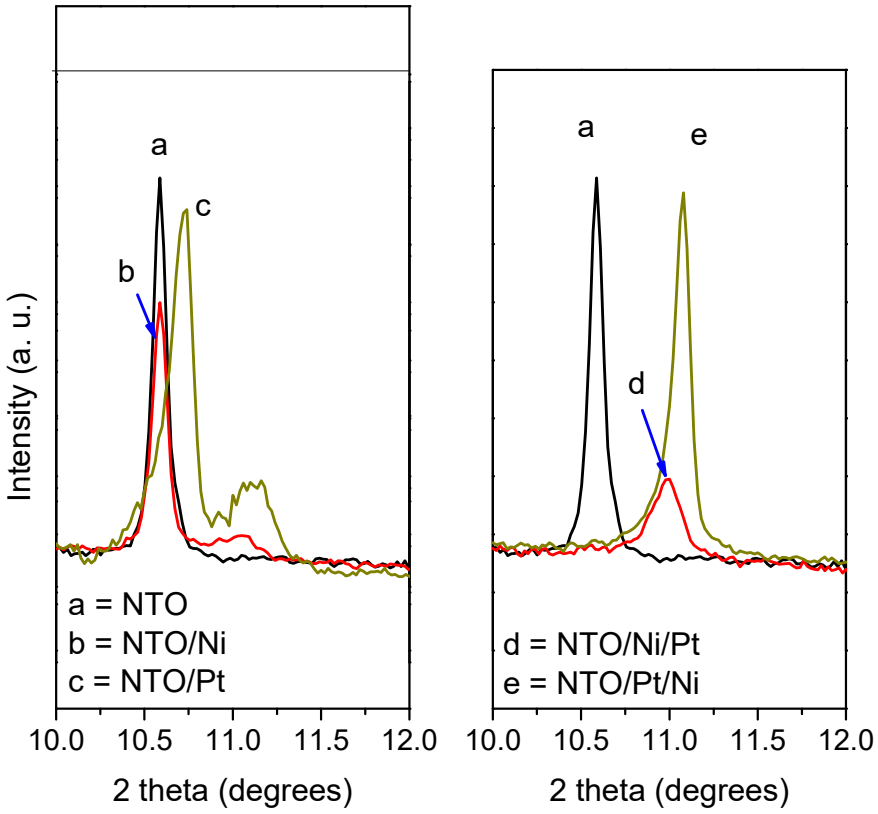

Figure S1. Close view of (0 0 1) main reflection on the prepared catalysts.

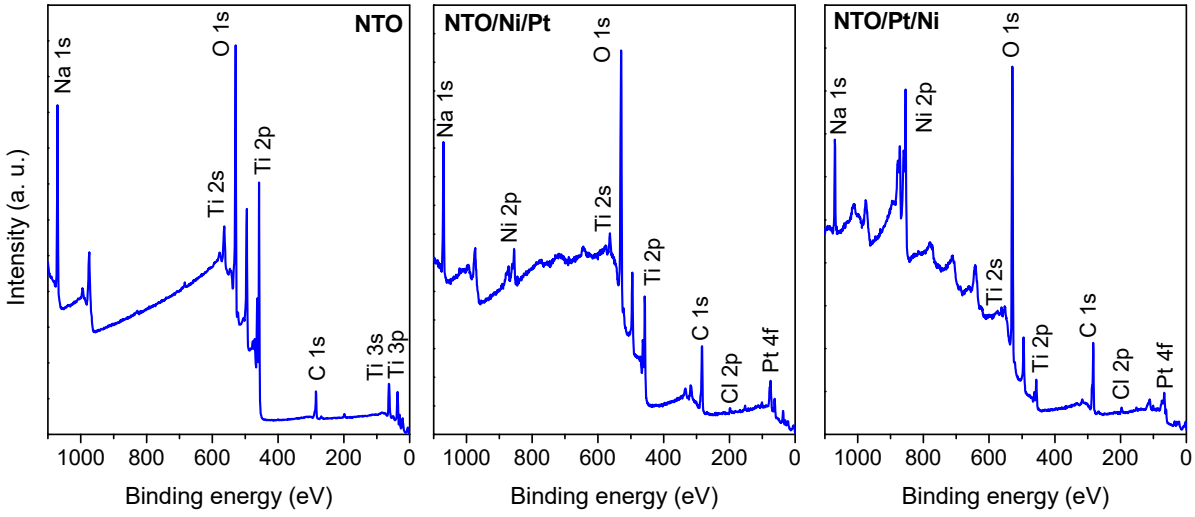

Figure S2. XPS survey spectra of NTO, NTO/Ni/Pt, and NTO/Pt/Ni samples.

Table S1. Semi-quantitative analysis of XPS peaks of bi-metallic samples.

Sample	Peak area						
	Na 1s	**Ti $^{4+}$**	**Ti $^{3+}$**	**O 1s**	**Ni $^{2+}$**	**Pt^{2+}**	**Pt $^{4+}$**
NTO/Ni/Pt	10,926.3	8540.5	3094.2	19,980.2	2418.7	680.4	2102.3
NTO/Pt/Ni	9431.5	2730.6	1258.7	28,170.0	15,121.5	449.4	308.2
NTO/Pt/Ni post-reaction	11,401.9	328.46	291.1	21,072.7	2415.4	-	-

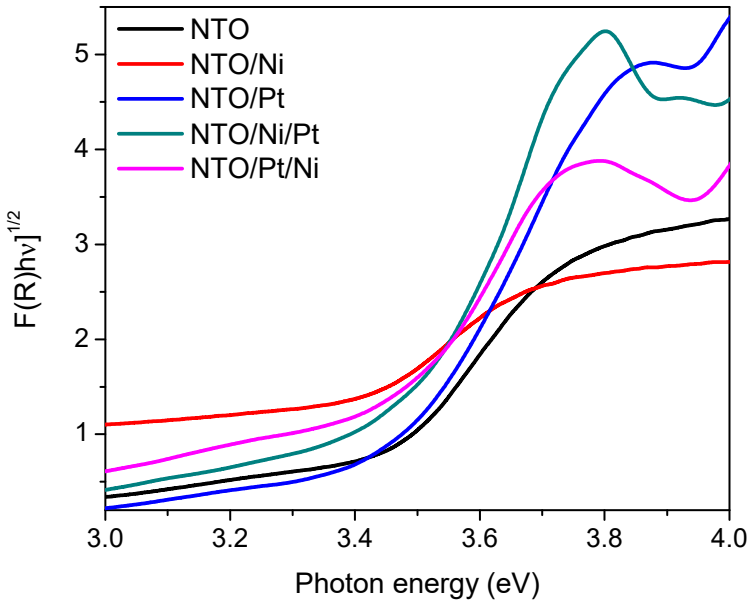

Figure S3. Kubelka.Munk spectra of Na2Ti3O7 bare and deposited with Ni and Pt metals.

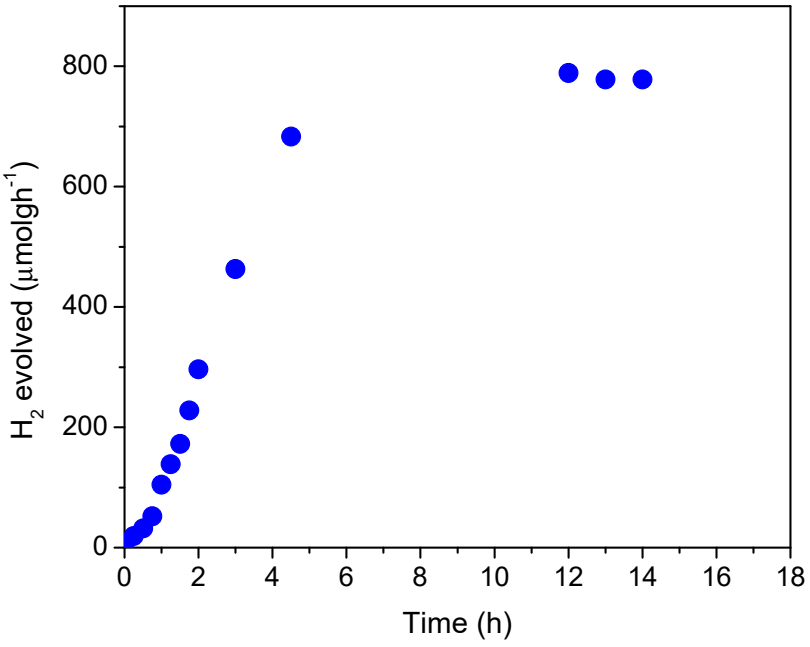

Figure S4. Time-dependent hydrogen evolution over the NTO/Pt/Ni sample.

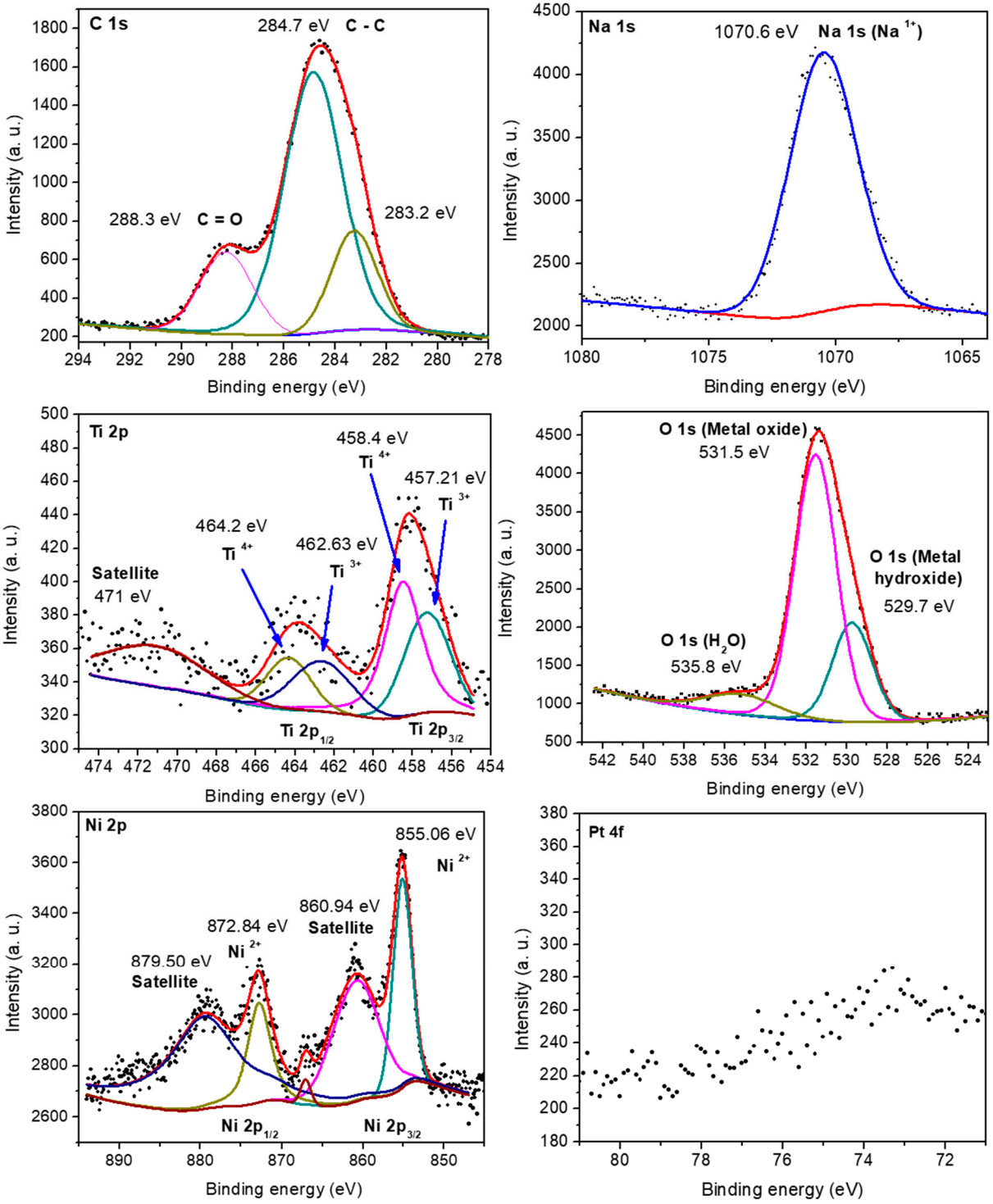

Figure S5. XPS characterization of the sample with the best performance (NTO/Pt/Ni) post reaction.

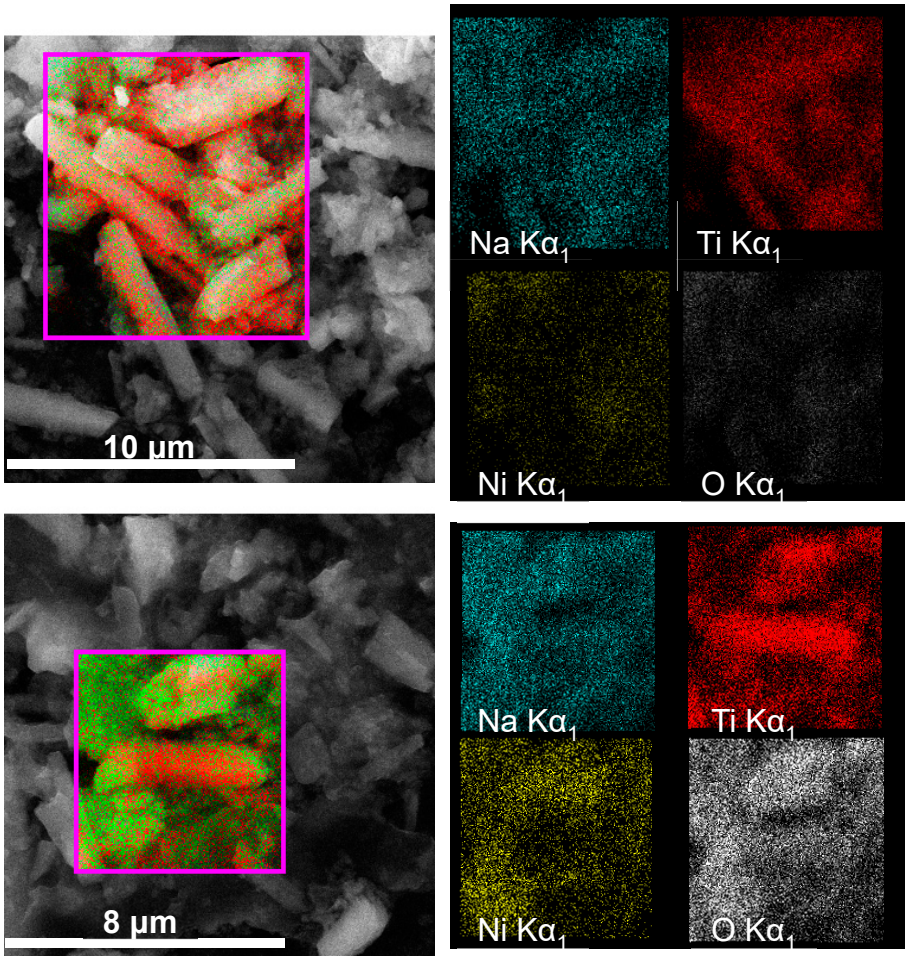

Figure S6. EDS mapping analysis of the NTO/Pt/Ni sample after photocatalytic testing.