

Article

# Crystal Structure and Supramolecular Architecture of Antiallergic Diphenylene Diethyl Dioxalamates

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## Materials and instrumentation

All reagents and solvents were purchased from commercial sources and used as received. <sup>1</sup>H and <sup>13</sup>C NMR spectra were recorded on a Varian Mercury 300 (<sup>1</sup>H, 300.08; <sup>13</sup>C, 75.46 MHz) instrument in CDCl<sub>3</sub> solutions, using SiMe<sub>4</sub> as internal reference, chemical shifts (δ) are in ppm and <sup>n</sup>J(H-H) in Hertz.

IR spectra were recorded neat at 25 °C using a Perkin Elmer Spectrum GX series with FT system spectrophotometer using the ATR device. Melting points were measured on an Electrothermal IA 9100 apparatus and were uncorrected.

## Synthesis and characterization of 1-3

*N,N'*-(4,4'-methanediyl-di-phenyl)-bis-diethyl dioxalamate (**1**) as **1**• $\frac{1}{4}$ (C<sub>2</sub>H<sub>2</sub>O<sub>4</sub>)• $\frac{1}{2}$ H<sub>2</sub>O. To a mixture of 4,4'-diaminodiphenylmethane (3.00 g, 15.1 mmol) in THF (50 mL) and TEA (5.27 mL, 37.8 mmol) were added dropwise ethyl 2-chloro-2-oxoacetate (4.22 mL, 37.8 mmol), under vigorous stirring, at 5–10 °C. After stirring for an additional 24 h at 25 °C, the solid was removed and washed with THF to give **1**• $\frac{1}{4}$ (C<sub>2</sub>H<sub>2</sub>O<sub>4</sub>)• $\frac{1}{2}$ H<sub>2</sub>O (5.26 g, 80.2% yield with respect to **1**) as a white solid. Mp: 144–146 °C, IR  $\nu$ (neat) (cm<sup>-1</sup>): 3367, 3340 (N-H), 2979 (C-H Ar), 1729, 1698 (C=O). <sup>1</sup>H NMR (CDCl<sub>3</sub>), δ: 10.75 (s, 2H, NH), 7.65 (d, 4H, <sup>3</sup>J = 8.2), 7.20 (d, 4H, <sup>3</sup>J = 8.46), 3.81 (s, 2H), 4.29 (q, 4H, O-CH<sub>2</sub>, <sup>3</sup>J = 7.18), 1.31 (t, 6H, CH<sub>2</sub>-CH<sub>3</sub>, <sup>3</sup>J = 7.18). <sup>13</sup>C NMR, δ: 161.3 (C9), 156.0 (C8), 138.4 (C4), 136.1 (C1), 129.5 (C3, C5), 121.3 (C2, C6), 40.6 (C13); 63.0 (C11), 14.5 (C12).

*N,N'*-(4,4'-oxy-di-p-phenylene)-bis-diethyl dioxalamate (**2**). Prepared as described for **1**, starting from 3.00 g of 4,4'-oxydianiline (15.0 mmol), 4.59 mL of TEA (30.0 mmol) and 6.10 mL of ethyl 2-chloro-2-oxoacetate (30.0 mmol). To obtain 4.05 g of **2** (67.5% yield) as a white solid. Mp: 163–165 °C, IR  $\nu$  (neat) (cm<sup>-1</sup>): 3268 (N-H), 2985 (Ar), 1729, 1663 (C=O). <sup>1</sup>H NMR (CDCl<sub>3</sub>), δ: 9.01 (s, 2H, NH), 6.96 (d, 4H, <sup>3</sup>J = 8.9), 7.60 (d, 4H, <sup>3</sup>J = 8.9), 4.36 (q, 4H, O-CH<sub>2</sub>, <sup>3</sup>J = 7.1), 1.37 (t, 6H, CH<sub>2</sub>-CH<sub>3</sub>, <sup>3</sup>J = 8.9). <sup>13</sup>C NMR (CDCl<sub>3</sub>), δ: 161.2 (C9), 154.6 (C8), 154.1 (C4), 132.1 (C1), 121.8 (C3, C5), 119.5 (C2, C6), 63.9 (C11), 14.1 (C12).

*N,N'*-(4,4'-biphenylene)-bis-diethyl dioxalamate (**3**). Prepared as described for **1**, starting from 1.00 g of benzidine (5.4 mmol), 20 mL of THF, 1.89 mL of TEA (10.9 mmol) and 1.51 mL of 2-chloro-2-oxoacetate (10.9 mmol) to obtain 0.94 g of **3** (45.1% yield) as a brown solid. Mp: 117–119 °C. IR  $\nu$

(neat) ( $\text{cm}^{-1}$ ): 3352, 3276 (N-H), 2990 (Ar), 1724, 1686 (C=O).  $^1\text{H}$  NMR ( $\text{CDCl}_3$ ),  $\delta$ : 10.9 (s, 2H, NH), 7.85 (d, 4H,  $^3J = 7.04$ ), 7.69 (d, 4H,  $^3J = 8.5$ ), 4.33 (q, 4H, O- $\text{CH}_2$ ,  $^3J = 5.28$ ), 1.33 (t, 6H,  $\text{CH}_2\text{-CH}_3$ ,  $^3J = 5.28$ ).  $^{13}\text{C}$  NMR,  $\delta$ : 161.2 (C9), 156.1 (C8), 137.4 (C4), 136.3 (C1), 127.3 (C3, C5), 121.5 (C2, C46), 63.1 (C11), 14.4 (C12).

**Table S1.** Selected bond distances ( $\text{\AA}$ ) and angles (deg) for 1-3.

Bond distances ( $\text{\AA}$ )					
	1a	1b	2	3a	3b
C8-C9	1.533(3)		1.522(3)	1.536(4)	
C21-C22	1.536(3)				
C33-C34		1.539(3)			
C46-C47		1.532(3)			
C28-C29					1.533(4)
C48-C49					1.532(3)
Angles ( $^\circ$ )					
C2-C1-N7-C8	-152.1(2)		-16.3(3)	-166.8(3)	
C15-C14-N20-C21	-166.8(2)				
C4-C13-C17-C16	-86.4(2)				
O8-C8-C9-O9	163.5(2)		-165.4(2)	-162.3(3)	
O21-C21-C22-O22	-178.6(2)				
C4-C13-C17	111.1(2)				
C3-C4-O13-C4			-134.1(2)		
C4-O13-C4			118.6(2)		
C27-C26-N32-C33		178.0(2)			
C44-C39-N45-C46		175.4(2)			
C29-C38-C42-C43		43.0(3)			
O33-C33-C34-O34		-170.5(2)			
O46-C46-C47-O47		174.9(2)			
C29-C38-C42		115.9(2)			
C3-C4-C4-C5				2.0(5)	
C22-C21-N27-C28					-157.0(2)
C42-C41-N47-C48					-164.8(2)
O28-C28-C29-O29					177.1(2)
O48-C48-C49-O49					175.6(2)
C23-C24-C44-C45					14.1(3)

**Table S2.** Intramolecular interactions in  $1 \cdot \frac{1}{2}(\text{C}_2\text{H}_2\text{O}_4) \cdot \text{H}_2\text{O}$ , 2 and 3.

Comp.	D—H $\cdots$ A	D—H/ $\text{\AA}$	H $\cdots$ A/ $\text{\AA}$	D $\cdots$ A/ $\text{\AA}$	D—H $\cdots$ A/ $^\circ$	Motif
$1 \cdot \frac{1}{2}(\text{C}_2\text{H}_2\text{O}_4) \cdot \text{H}_2\text{O}$	N7-H7 $\cdots$ O9	0.88	2.31	2.701(2)	106.60	S(5)
	N20-H20 $\cdots$ O22	0.88	2.23	2.666(3)	110.00	S(5)
	C6-H6 $\cdots$ O8	0.95	2.43	2.942(2)	114.00	S(6)
	C19-H19 $\cdots$ O21	0.95	2.34	2.933(3)	120.00	S(6)
	N32-H32 $\cdots$ O34	0.88	2.24	2.679(2)	110.00	S(5)
	N45-H45 $\cdots$ O47	0.88	2.27	2.690(3)	108.90	S(5)
	C31-H31 $\cdots$ O33	0.95	2.31	2.923(3)	122.00	S(6)
	C40-H40 $\cdots$ O46	0.95	2.32	2.925(3)	121.00	S(6)
2	N7-H7 $\cdots$ O9	0.86	2.28	2.689(2)	109.00	S(5)
	C2-H2 $\cdots$ O8	0.93	2.36	2.929(3)	119.00	S(6)
3	N7-H7 $\cdots$ O9	0.86	2.31	2.711(3)	108.24	S(5)
	C6-H6 $\cdots$ O8	0.93	2.33	2.911(4)	120.00	S(6)
	N27-H27 $\cdots$ O29	0.86	2.29	2.698(3)	108.90	S(5)
	N47-H47 $\cdots$ O49	0.86	2.28	2.698(3)	109.50	S(5)
	C26-H26 $\cdots$ O28	0.93	2.40	2.919(3)	115.00	S(6)
	C46-H46 $\cdots$ O48	0.93	2.34	2.910(3)	119.10	S(6)

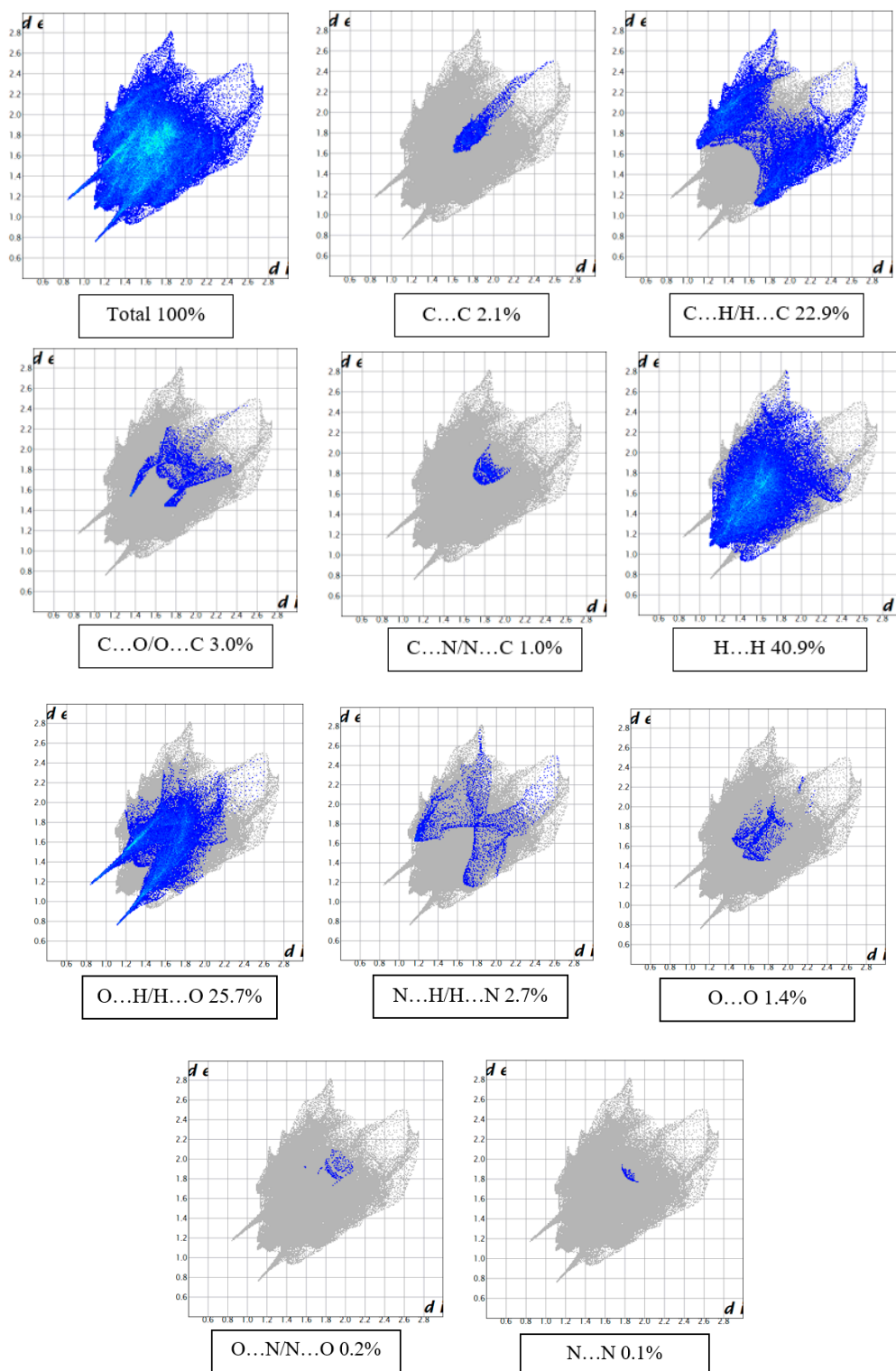


Figure S1. Fingerprints plots of the total and specific intermolecular contacts of 1.

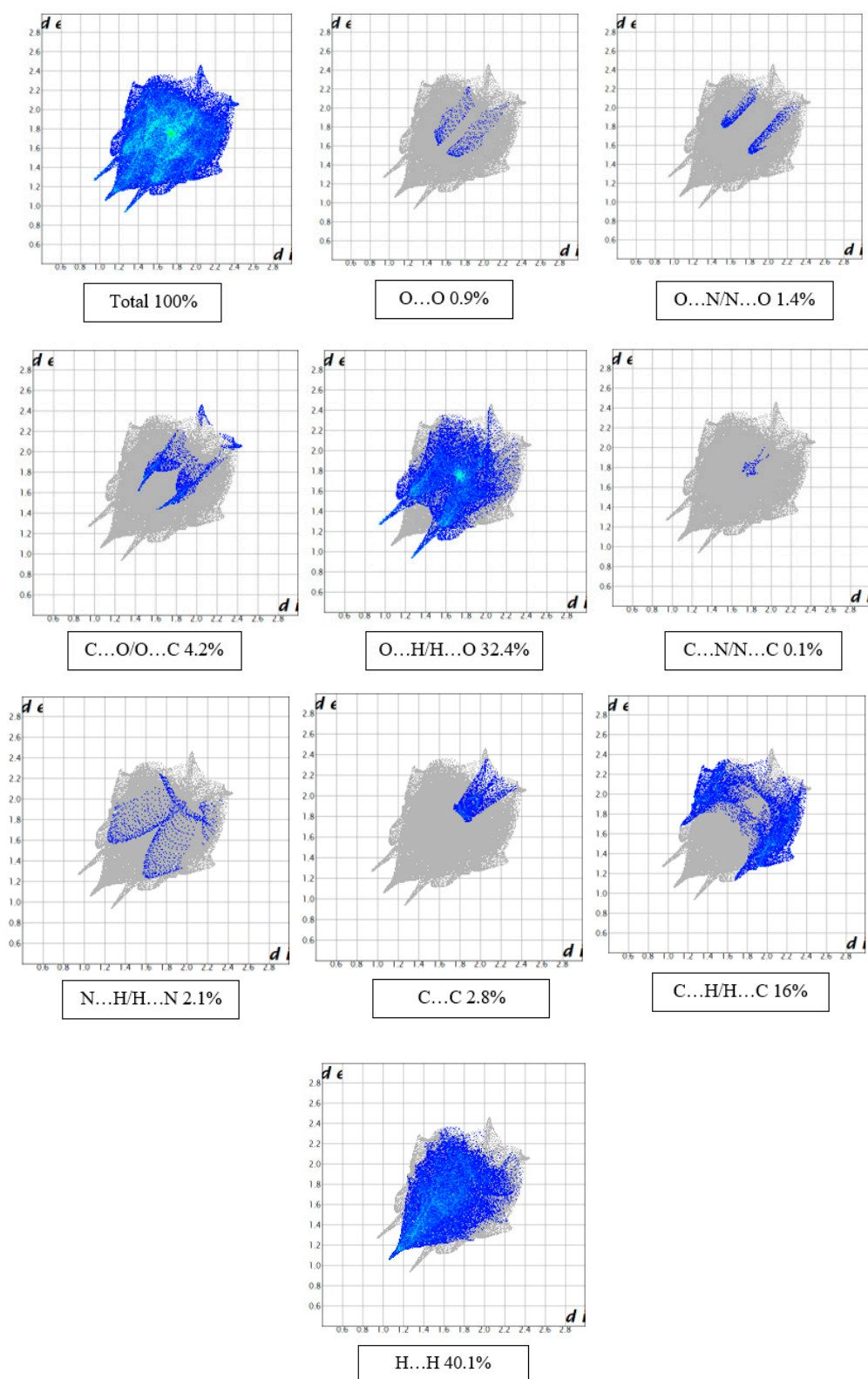
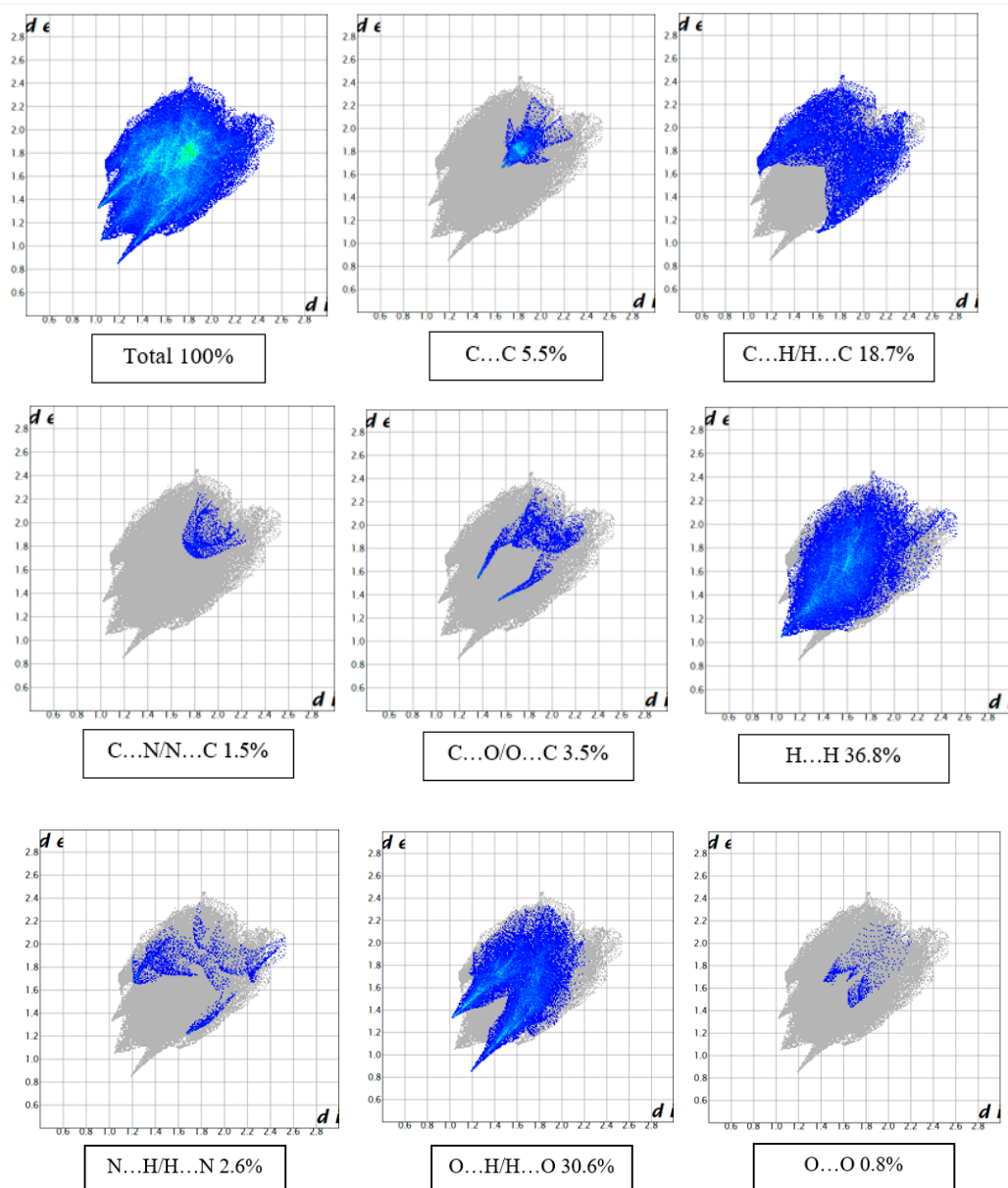


Figure S2. Fingerprints plots of the total and specific intermolecular contacts of 2.



**Figure S3.** Fingerprints plots of the total and specific intermolecular contacts of **3**.



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