

Article

Eff**ect of the Zr-Substitution on the Structural and Electrical Properties of LaFeO3: XRD, Raman Scattering, SEM, and Impedance Spectroscopy Study**

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Received: 8 April 2020; Accepted: 12 May 2020; Published: 15 May 2020

Abstract: The LaFe_{1-x}Zr_xO₃ (x = 0.01, 0.05) ceramics were prepared by sol-gel and annealing method and studied by XRD, Raman scattering analysis, SEM, and impedance spectroscopy method. The crystal structure and phonon characteristics analysis revealed that the crystal structure tends to preserve its ideal orthorhombic structure, following the increase in driving force of the $Fe/ZrO₆$ octahedral tilting. The frequency-dependent dielectric parameters at each temperature decreased with increasing Zr content. The temperature dependence dielectric relaxation and dc conduction mechanism satisfied the Arrhenius law and increased with increasing Zr content. The activation energy ranged from 0.30 to 0.50 eV and was similar in the relaxation and conduction mechanisms, indicating that both transport mechanisms were based on a similar mechanism.

Keywords: LaFe_{1-x}Zr_xO₃; crystal structure; phonon characteristics; relaxation; conduction

1. Introduction

The interest in $LaFeO₃$ among perovskite inorganic materials increased tremendously due to their applicability ranging as chemical sensors, electrode materials, etc. [\[1](#page-9-0)[,2\]](#page-9-1). Doping the La-site with divalent metals and/or Fe-site with transition metals has been reported to increase the structural and thermal stability, magnetic, and electrical properties as compared with the parent compound [\[1,](#page-9-0)[3\]](#page-9-2). Specifically, doping in Fe-site with varying metal cations and concentrations has been reported in many studies [\[4–](#page-9-3)[7\]](#page-10-0).

Ti-doped on Fe-site showed weak ferromagnetic behavior, followed by the increase of magnetization with increasing Ti concentration [\[4\]](#page-9-3). Doping with Mg^{2+} and Zn^{2+} induces changes in the band structure-property and enhances the oxygen reduction's efficiency when photocathodes are built from this material [\[5\]](#page-9-4). Another study reported that Mg substitution at the Fe-site with 10% of concentration reduces the resistance of $LaFeO₃$ and produces better response and selectivity to ethanol gas [\[6\]](#page-10-1). Zn-doped on Fe-site exhibited the colossal dielectric constant following by the Debye-like dipolar relaxation mechanism [\[7\]](#page-10-0).

In spite of the fact that there are numerous works committed to studying the physical properties in LaFeO₃ containing valence cations as substituents $[1–7]$ $[1–7]$, research investigations on the effect of Zr cations on the Fe-site of LaFeO₃ are still lacking information. The question of the effect of Zr, as tetravalent cations occupying the Fe-site of LaFe O_3 on the structural parameters and electrical transport mechanism, has not been systematically studied. Therefore, this present work concentrates on the investigation of the effect of the Zr content on the structure and electrical properties of LaFeO₃. In this work, we prepared the Zr-doped LaFeO₃ with different concentration of Zr. The XRD was employed to determine the crystal structure and structural parameters for the prepared samples, while the phonon

characteristics were studied by Raman scattering spectroscopy. The morphology and grain distribution were investigated using SEM. The impedance spectroscopy was applied to get a better understanding of the electrical mechanism working on this material.

2. Experimental Details

The LaFe_{1−x}Zr_xO₃ (x = 0.01, 0.05) powdered ceramics were prepared by sol-gel method using La₂O₃, Fe(NO₃)₃.9H₂O, Cl₂H₁₂O₉Zr, and citric acid monohydrate as precursors (all analytically pure) [\[8\]](#page-10-2).

$$
La_2O_3 + (1 - x) Fe(NO_3)_3.9H_2O + x Cl_2H_{12}O_9Zr \to LaFe_{1-x}Zr_xO_3
$$

Distilled water and citric acid monohydrate were used to solve the precursors with certain stoichiometry. The solvent obtained was stirred and heated under certain condition to form the homogeneous gel. The gel-obtained was dried to form the black powder and then calcined at 900 ◦C for 6 h. This powder-calcined was pressed into a thin pellet with 1 cm of diameter under 3 kN of pressure for 3 min. Finally, the pellet was sintered and then annealed at 800 ◦C for 6 h to form the polycrystalline ceramics.

The crystal structure was investigated by X-ray diffraction (X'Pert PRO PANalytical diffractometer, Almelo, The Netherlands) using Cu K α (λ = 1.5418Å) radiation source in the 20 range of 20° \leq 20 ≤ 90◦ . The structural analysis obtaining from XRD data were analyzed using Fullprof.2k (Institut Laue-Langevin, Grenoble, France) and VESTA (JP-Mineral.org, Ibaraki, Japan) software. The Raman scattering spectra were recorded in the range from $70-2000$ cm⁻¹ using 532 nm laser source by Thermo Scientific DXR2 Raman Microscope (Thermo Electron Scientific Instruments, Fitchburg, WI, USA). The morphology of the ceramic samples was analyzed by SEM (FEI QUANTA 650 FEG, Thermo Electron Scientific Instruments, Fitchburg, WI, USA). The sample's surface was gold-coated before surface observation. Electrical properties were carried out using LCR meter (FLUKE-PM 6303, Glottertal, Germany) with single parallel resistance-capacitance model in the temperature range of 125–300 °C under the frequency range of 100 Hz to 1 MHz.

3. Results and Discussion

*3.1. X-ray Di*ff*raction Analysis*

Figure [1](#page-2-0) denotes the XRD patterns of LaFe_{1−x}Zr_xO₃, annealed at 800 °C for 6 h. The analysis results confirm the formation of the single-crystalline perovskite. Peaks corresponding to lanthanum oxide, iron hydrate, and zirconium hydrate did not appear, indicating the incorporation of Zr with Fe in perovskite lattice. The indexation was performed according to the proposed orthorhombic structure within the Pbnm space group with four formula units $(Z = 4)$. The obtained crystallographic parameters are tabulated in Table [1.](#page-1-0) The lattice constant increased with increasing Zr content because the ionic radii of Fe³⁺ (64 pm) are smaller than that of Zr^{4+} (86 pm) and might be due to the changes in cation degree of Zr as compensation of electric charge.

Table 1. Crystallographic parameters of the LaFe_{1−x}Zr_xO₃ annealed at 800 °C for 6 h, obtained by the FullProf 2k programme.

x	Parameters					
	a(A)	b(A)	c(A)	Volume (Å3)	Crystallite Size (nm)	
0.01	5.5556(8)	5.5631(9)	7.8569(9)	242.83(9)	68.7(1)	
0.05	5.5615(3)	5.5720(4)	7.8646(0)	243.71(7)	59.9(2)	

Figure 1. XRD pattern of LaFe1−xZrxO3 annealed at 800 °C for 6 h. **Figure 1.** XRD pattern of LaFe1−xZrxO³ annealed at 800 ◦C for 6 h.

The average crystallite size was calculated based on Scherrer's method [\[9\]](#page-10-3):

$$
D = \frac{K\lambda}{\beta_{1/2}\cos\theta_{hkl}}
$$

where λ (1.5405 A) is the wavelength of the X-rays, θ is the Bragg angle, $\beta_{1/2}$ is the FWHM, and *K* $=$ 0.89. We calculated the average crystallite size using XRD data. We found that the crystallite size decreases with increasing Zr content, indicating the decrease in crystallinity degree [\[10\]](#page-10-4).

Now, we turn to illustrate the geometrical parameters for each sample. The obtained geometrical vibrational modes of LaFe_{1-x}Zr_xO₃ at room temperature [\[11–](#page-10-5)[13\]](#page-10-6). The Brillouin zone center normal refinement analysis was used to estimate the selection rules, symmetry, and normal modes of the modes of LaFe_{1−x}Zr_xO₃ with *Pbnm* symmetry (x = 0.01, 0.05) are represented in Table [2](#page-2-1) [\[12\]](#page-10-7). The results predict 24 Raman active phonon modes, 28 infrared modes, and 8 silent modes for the room temperature LaFe_{1-x}Zr_xO₃ structure (belonging to the *Pbnm* symmetry) [12,13]. This prediction was used to analyze the vibrational modes characterized by Raman scattering spectra. To determine the distortion degree for each sample, the calculated tolerance factor and Fe/ZrO₆ tilting degree have been estimated from the bond angle and bond length data [10,13]. The obtained tolerance factors indicat[e th](#page-10-6)at the crystal distortion decreased with increasing Zr content. In other words, the crystal structure tends to preserve the ideal orthorhombic structure with increasing Zr content. The decreasing Fe/ZrO₆ tilting angle (see Table 3) indicates an increasing driving force of the octahedral tilting.

Table 2. Atoms, Wickoff position, and modes representation of LaFe_{1−x}Zr_xO₃ (x = 0.01, 0.05) with Pbnm symmetry.

Atom	Wickoff Position	Modes Representation
La	4c	$2A_g + B_{1g} + 2B_{2g} + B_{3g} + A_u + 2B_{1u} + B_{2u} + 2B_{3u}$
Fe/Zr	4a	$3A_{11} + 3B_{111} + 3B_{211} + 3B_{311}$
O1	4c	$2A_g + B_{1g} + 2B_{2g} + B_{3g} + A_u + 2B_{1u} + B_{2u} + 2B_{3u}$
O ₂	8d	$3A_g + 3B_{1g} + 3B_{2g} + 3B_{3g} + 3A_u + 3B_{1u} + 3B_{2u} + 3B_{3u}$

Modes classification: Raman modes: $7A_g + 5B_{1g} + 7B_{2g} + 5B_{3g}$; IR modes: $10B_{1u} + 8B_{2u} + 10B_{3u}$; IR-silent modes: $8A_u$.

Parameters		$x = 0.01$	$x = 0.05$
Atomic Position			
	X	0.998	0.993
La	y	0.027	0.030
	z	0.25	0.25
	X	$\overline{0}$	Ω
Fe/Zr	y	0.5	0.5
	z	θ	θ
	x	0.719	0.719
O1	y	0.302	0.302
	z	0.029	0.029
	$\mathbf x$	0.023	0.08
O2	y	0.496	0.485
	z	0.25	0.25
Bond Angle (°)			
Fe/Zr-O2-Fe/Zr		154.03(8)	154.08(2)
Fe/Zr-O1-Fe/Zr		157.08(1)	157.07(3)
Bond Length (Å)			
$La-O1(m)$		2.4508(8)	2.4536(4)
La $-O1$ (l)		2.7602(1)	2.7629(1)
La $-O2(s)$		2.3812(2)	2.3859(2)
$\langle La-O \rangle$		2.5307(3)	2.5341(2)
$Fe/Zr-O1$ (m)		2.0125(2)	2.0166(6)
$Fe/Zr-O1$ (s)		1.9234(4)	1.9254(3)
$Fe/Zr-O2$ (l)		2.0859(1)	2.0890(9)
$\langle Fe/Zr-O \rangle$		2.0073(2)	2.0104(6)
Tolerance Factor		0.8914(9)	0.8909(1)
Average Tilt Angle $\langle \varphi \rangle$ (°)		14.923(9)	14.915(2)
R-Factors			
R_p		5.35	5.51
R_{wp}		6.99	7.44
R _{exp}		5.28	5.53
χ^2		1.76	1.81

Table 3. Geometrical parameters characterizing the crystal structure of LaFe_{1-x}Zr_xO₃ annealed at 800 °C for 6 h.

3.2. Raman Scattering Spectra

For complex perovskite, normally all the Raman modes cannot be experimentally observed. However, there are some dominating modes assigned in Figure [2](#page-4-0) and tabulated in Table [4.](#page-3-1) Due to the similar crystal structure and symmetry, the Raman scattering spectra of LaFe_{1-x}Zr_xFeO₃ ceramics is similar to a previously reported Raman scattering spectrum of pure-LaFeO₃ and LaFe_{1-x}Mo_xO₃ [\[12](#page-10-7)[,13\]](#page-10-6).

Table 4. Frequency and symmetry assignment of some specific Raman-active phonon modes (ω) observed for LaFe_{1-x}Zr_xFeO₃ (x = 0.01, 0.05) at room temperature. All units are cm⁻¹.

	ω_1	w٠	ω_3	ω_4	ω_{5}	ω_{6}	ω_7
$x = 0.0$	150.2	173.9	288.4	428.5	637.5	1128	1304
$x = 0.5$	147.9	171.4	286.5	427.2	637.7	1133	1303
Symmetry	A_{σ}	$A_{\mathfrak{C}}$	$A_{\rm g}$	A_{σ}	B_{1g}	Second-order	Second-order

Figure 2. Raman scattering spectra of LaFe1−xZrxFeO3 ceramics with (**a**) x = 0.01, and (**b**) x = 0.05 at **Figure 2.** Raman scattering spectra of LaFe_{1−x}Zr_xFeO₃ ceramics with (**a**) $x = 0.01$, and (**b**) $x = 0.05$ at room temperature.

As confirmed in previous reports [\[12](#page-10-7)[,13\]](#page-10-6), the ω_1 and ω_2 modes are related to vibrations of the La-site cations with A_g symmetry. The ω_3 mode is attributed to the Fe/ZrO₆ octahedra displacement with A_g symmetry. The ω_4 mode is considered by the asymmetric stretching vibrations, to be related **ω1 ω² ω³ ω⁴ ω5 ω6 ω⁷** vibrations of Fe/ZrO⁶ octahedra with B1g symmetry. Another phonon mode over 700 cm−¹ could be indicated with multiphonon processes $\left[13,14\right]$ $\left[13,14\right]$ $\left[13,14\right]$. The Raman phonon modes tend to shift to the lower wavenumber reflecting the lattice constant and bond length increased with Zr content which is fitted with refinement results (Table [1\)](#page-1-0). The Raman intensity for all phonon modes is increased and their *3.3. SEM* linewidth is compressed with increasing Zr content. This reflects the decreasing of lattice disorder and analysis (Table [2\)](#page-2-1). with Jahn–Teller distortions with A_g symmetry. The ω_5 mode is assigned to the stretching symmetric crystal distortion which is consistent with the decreased tolerance factor observed by X-ray diffraction

3. SEM and 127 nm for α = 0.05, respectively. In the typical energy dispersively. In the typical energy disper *3.3. SEM*

 Γ ray spectroscopy pattern of the elements of the elements Γ Figure [3](#page-5-0) shows the SEM micrographs of the ceramics samples. The typical polycrystalline microstructure is shown with a larger non-uniform grain distribution. The average grain size was
example in the average state in the average grain size was estimated to be 159 and 127 nm for $x = 0.01$ and $x = 0.05$, respectively. In the typical energy dispersive X-ray spectroscopy pattern of each ceramic, all of the elements' (La, Fe, Zr, and O) peaks are presented in their respective molar concentrations during the annealing process, indicating the purity and phase formation of LaFe $_{1-x}Zr_xO_3$ ceramics.

3.4. Electrical Properties

3.4.1. Impedance Spectroscopy Analysis

Figure [4](#page-5-1) shows the Nyquist plot of LaFe_{1-x}Zr_xO₃ annealed at 800 °C for 6 h at different temperatures. All plots are suppressed into two semicircular arcs centering in Z' axis; the first one is observed at high frequencies while the second circle is observed at low frequencies, which are represented by the grain and grain boundary contribution working on electrical transport mechanism in these materials, respectively [\[13\]](#page-10-6). The semicircle decreases with increasing temperature, in other words, the decreasing of impedance value with temperature, indicates the increase in electrical conductivity with temperature and displays the polycrystalline nature of the material. [\[15](#page-10-9)[,16\]](#page-10-10). The impedance value is also shown to be increasing with increasing Zr content.

Figure 3. SEM-EDX of $\text{LaFe}_{1-x}\text{Zr}_x\text{O}_3$ annealed at 800 °C for 6 h. (a) and (c) represent SEM images of $x = 0.01$ and 0.05, respectively. (b) and (d) represent EDX spectrum of $x = 0.01$ and 0.05, respectively.

Figure 4. Nyquist plots of $\text{LaFe}_{1-x}\text{Zr}_x\text{O}_3$ with (a) $x = 0.01$ and (b) $x = 0.05$ annealed at 800 °C for 6 h at different temperatures. different temperatures.

Figure 5 displays the real part of the impedance (Z') at different temperatures as a function of the frequency for LaFe_{1-x}Zr_xO₃ ceramics. The Z' decreases with increasing frequency in the low-frequency region and becomes independent in the high-frequency region. The Z' independent parts are merged at higher frequency regions for all temperatures. The Z' decreases with increasing temperature, indicating a negative thermal coefficient of resistance (NTCR) behavior at lower frequency regions, together with the increasing of electrical conductivity with temperature [\[17\]](#page-10-11). The increase in electrical conductivity could be due to the release of space charges which is causing the reduction of potential conductivity could be due to the release of space charges which is causing the reduction of potential barrier strength [\[16\]](#page-10-10). increase in electrical conductivity could be due to the release of space charges which is causing the relation of pow

Figure 5. Bode plots for Z'(f) of LaFe_{1-x}Zr_xO₃ with (a) $x = 0.01$ and (b) $x = 0.05$ annealed at 800 °C for h at different temperatures. 6 h at different temperatures.

Figure [6](#page-6-1) shows the imaginary part of the impedance (Z") of LaFe_{1-x}Zr_xO₃ ceramics as a function ϵ the frequency at different temperatures. All the plots means at the higher frequency region in dia of the frequency at different temperatures. All the plots merge at the higher frequency region indicating
of the frequency at different temperatures. All the plots merge at the higher frequency region indicating the charge carrier through grains at higher frequency [\[16](#page-10-10)[,18\]](#page-10-12). The increase in frequency is accompanied by an increase of $Z^{\prime\prime}$ until a maximum value is reached at a particular frequency, and then decreases with an increasing frequency, indicating the existence of a relaxation mechanism in the ceramics. The relaxation peaks shift to the high frequencies and become more broad with a rise in the temperature, indicating that the relaxation is thermally activated [\[18\]](#page-10-12).

Figure 6. Bode plots for $Z''(f)$ of LaFe_{1-x} Zr_xO_3 with (a) $x = 0.01$ and (b) $x = 0.05$ annealed at 800 °C for 6 h at different temperatures. 6 h at different temperatures.

By using the Arrhenius law [\[19\]](#page-10-13), the activation energy of the relaxation mechanism for each sample was calculated and tabulated in Table [5.](#page-7-0) The activation energy increased with increasing Zr content, indicating the decreasing of charge carrier concentration in materials [\[20\]](#page-10-14). The normalized spectra of the imaginary impedance in each sample are shown in the insets of Figure [6.](#page-6-1) All normalized spectra were non-overlapping and scaled to the multiple master curve, suggesting the temperature-dependent and localized relaxation dynamics [\[21\]](#page-10-15).

Table 5. Activation energy of the LaFe_{1−x}Zr_xFeO₃ (x = 0.01, 0.05) ceramics.

x	E_a (eV)
0.01	0.43
0.05	0.51

3.4.2. Dielectric Analysis

Figure [7](#page-7-1) shows the dielectric constant of LaFe_{1−x}Zr_xO₃ ceramics as a function of frequency at different temperatures. The dielectric constant decreases with increasing frequency. The dielectric constant at lower frequency is higher because all polarizations appear in this region which comes from the grain boundary effect, which is a characteristic of dielectric material [\[22\]](#page-10-16). A decrease in the dielectric constant value is due to the reduction in ionic and dipolar polarizations at higher frequency regions. Moreover, the dielectric constant decreases with increasing Zr content.

Figure 7. Dielectric constant, $\varepsilon'(f)$, of LaFe_{1-x}Zr_xO₃ with (a) x = 0.01 and (b) x = 0.05 annealed at 800 °C for 6 h at different temperatures.

The evolution of dielectric loss as a function of frequency at different temperatures is shown in The evolution of dielectric loss as a function of frequency at different temperatures is shown in Figure 8. The dielectric loss spectra have a similar trend with the dielectric constant. The dielectric Figure [8.](#page-8-0) The dielectric loss spectra have a similar trend with the dielectric constant. The dielectric loss decreases with increasing Zr content. This convinces the usefulness of the Zr substitution which has the significance of reducing the loss dielectric. Overall, the dielectric spectra analysis does not exhibit any specific anomaly associated with the phase transition in this temperature range.

 \mathbf{a} consists of two different regions: Ac conductivity, increasing with the frequency approximating to **x = 0.01 1110**
. **150 o C 4** complex conductivity followed the Jonscher's power law [\[23\]](#page-10-17): **x = 0.05** linear function, and dc conductivity, dominating in low frequencies. The frequency dependence of the Figure [9](#page-8-1) shows the complex conductivity for LaFe_{1−x}Zr_xO₃ ceramics. It is clear that the plot $\sigma(\omega)$

$$
\sigma(\omega) = \sigma_{DC} + A\omega^s
$$

where σ_{DC} is the DC conductivity, ω is the (measured) angular frequency, the exponent *s* is a temperature-dependent constant, and *A* is a constant which strongly depends on the temperature and composition of the material. From this, σ_{DC} as a function of temperature could be obtained. let σ_{DC} is the DC contactivity, ω is the uncastical angular irequency, the exponent s is

Figure 8. Dielectric loss, $\varepsilon''(f)$, of LaFe_{1-x}Zr_xO₃ with (a) x = 0.01 and (b) x = 0.05 annealed at 800 °C for 6 h at different temperatures.

Figure 9. Complex conductivity, $\sigma(f)$, of LaFe_{1-x}Zr_xO₃ with (a) x = 0.01 and (b) x = 0.05 annealed at 800 °C for 6 h at different temperatures. 800 ◦C for 6 h at different temperatures.

1
1 conductionmecha with increasing Zr content. The activation energy is in the range of $0.20 < E_a < 1.0$ eV, suggesting the the slope of log σ_{DC} vs reciprocal temperature (10^3 /T). It was found that the activation energy increased the Arrhenius law [24]. The activation e **18** function of temperature on a semi-logarithmic scale. The dc conductivity vs reciprocal temperature **10** fulfils the Arrhenius law [\[24\]](#page-10-18). The activation energy of the dc conductivity process was calculated from **18** Figure [10](#page-9-5) shows the dc conductivity versus 1000/T, including the inset shows dc conductivity as a **b**
3
1

1 dc conduction mechanism is dominated by polaron hopping [\[25\]](#page-10-19).

Figure 10. DC conductivity, σ_{DC} , of LaFe_{1-x}Zr_xO₃ with (a) x = 0.01 and (b) x = 0.05 annealed at 800 °C for 6 h at different temperatures. for 6 h at different temperatures.

4. Conclusion 4. Conclusions

The effect of Zr substitution on LaFeO₃ has been studied in this paper. The X-ray diffraction revealed that Zr substitution, until 5% molar concentration, still preserves the single phase of revealed that Zr substitution, until 5% molar concentration, still preserves the single phase of orthorhombic perovskite structure with *Pbnm* symmetry, followed by the increase of lattice constant with increasing Zr content. The average crystallite and grain size decrease with Zr content indicating the reduced degree of crystallinity. The phonon characteristics from Raman scattering analysis reflect the decrease in lattice disorder and crystal distortion with Zr content which is consistent with the decreased calculated tolerance factor. The dielectric parameters decrease with increasing Zr content. The activation energy calculated in relaxation and dc conduction mechanism is found to be increased with increasing Zr content.

Author Contributions: Conceptualization, D.T.; methodology, D.T., I.P., and R.A.R.; software, I.P. and R.A.R.; validation, D.T., I.P. and R.A.R.; formal analysis, D.T.; investigation, D.T. and I.P.; resources, I.P. and R.A.R.; data curation, I.P. and R.A.R.; writing—original draft preparation, D.T.; writing—review and editing, D.T.; visualization, I.P. and R.A.R.; supervision, D.T.; funding acquisition, D.T. and R.A.R. All authors have read and agreed to the published version of the manuscript.

Funding: The authors acknowledge funding HIBAH PUTI Q2—2020 from DRPM Universitas Indonesia.

Conflicts of Interest: The authors declare no conflict of interest.

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