

Supplementary Materials

New Rod-like H-bonded Assembly Systems: Mesomorphic and Geometrical Aspects

Laila A. Al-Mutabagani ¹, Latifah Abdullah Alshabanah ¹, Hoda A. Ahmed ^{2,3,*},
Khulood A. Abu Al-Ola ⁴ and Mohamed Hagar ^{3,5,*}

¹ Chemistry Department, College of Science, Princess Nourah bint Abdulrahman University, Riyadh 11671, Saudi Arabia; laalmutbagani@pnu.edu.sa (L.A.A.-M.); laalsabanah@pnu.edu.sa (L.A.A.)

² Department of Chemistry, Faculty of Science, Cairo University, Cairo 12613, Egypt

³ Chemistry Department, College of Sciences, Yanbu, Taibah University, Yanbu 30799, Saudi Arabia

⁴ Chemistry Department, College of Sciences, Al-Madina Al-Munawarah, Taibah University, Al-Madina 30002, Saudi Arabia; kabualola@taibahu.edu.sa

⁵ Chemistry Department, Faculty of Science, Alexandria University, Alexandria 21321, Egypt

* Correspondence: ahoda@sci.cu.edu.eg (H.A.A.); mhagar@taibahu.edu.sa (M.H.)

Characterizations

Supramolecular complex formation was confirmed by TA Instruments Co. Q20 Differential Scanning Calorimeter (DSC; USA), polarized-optical microscopy (POM, Wild, Germany) and FT-IR (Nicolet iS 10 Thermo scientific) spectroscopic analysis.

Calorimetric measurements were carried out using a PL-DSC of Polymer Laboratories, England. The instrument was calibrated for temperature, heat and heat flow according to the method recommended by Cammenga, et. al. [1]. Measurements were carried out for small samples (2–3 mg) placed in sealed aluminum pans. All measurements were conducted at a heating rate of (10 °C/min) in an inert atmosphere of nitrogen gas (10 mL/min). For DSC, the sample was heated from room temperature to 280 °C, at heating rate of 10 °C/min under nitrogen atmosphere and then cooled in the cell to 0 °C. All weighed samples were made using an ultra-microbalance, England, with accuracy ± 0.0001 .

Transition temperature for the 1:1 supramolecular H-bonded complexes (**An/Bm**) was investigated by DSC in heating and cooling cycles. The type of the mesophase was identified using a standard polarized-optical microscopy POM (Wild, Germany), attached with Mettler FP82HT hot stage. Measurements were made twice, and the results were found to have accuracy in transition temperature and enthalpy within ± 0.2 .

Computational methods and calculations

The theoretical calculations for the investigated compounds were carried out by Gaussian 09 software [2]. DFT/B3LYP methods using 6-31G (d, p) basis set was selected for the calculations. The geometries were optimized by minimizing the energies with respect to all geometrical parameters without imposing any molecular symmetry constraints. The structures of the optimized geometries had been drawn with Gauss View [3]. Moreover, the calculated frequencies were carried out using

the same level of theory. The frequency calculations showed that all structures were stationary points in the geometry optimization method with none imaginary frequency.

References:

1. Cammenga H.K.; Eysel W.; Gmelin E.; Hemminger W.; Höhne G.W.; Sarge S.M.; The temperature calibration of scanning calorimeters: Part 2. Calibration substances, *Thermochimica Acta*, **1993**, 219, 333–342.
2. Frisch, M.; Trucks, G.; Schlegel, H.B.; Scuseria, G.; Robb, M.; Cheeseman, J.; Scalmani, G.; Barone, V.; Mennucci, B.; Petersson, G. Gaussian 09, Revision A. 02; Gaussian Inc.: Wallingford, CT, USA, 2009; p. 200.
3. Dennington, R.; Keith, T.; Millam, J. GaussView, Version 5; Semichem Inc.: Shawnee Mission, KS, USA, 2009.