

## Supplementary material

### **Oxidative Thermal Conversion of Hydrothermal Derived Precursors toward the Mixed-Metal Cobaltite Spinel Oxides ( $\text{ZnCo}_2\text{O}_4$ and $\text{NiCo}_2\text{O}_4$ ): In-Situ Investigation by Synchrotron-Radiation XRD and XAS techniques**

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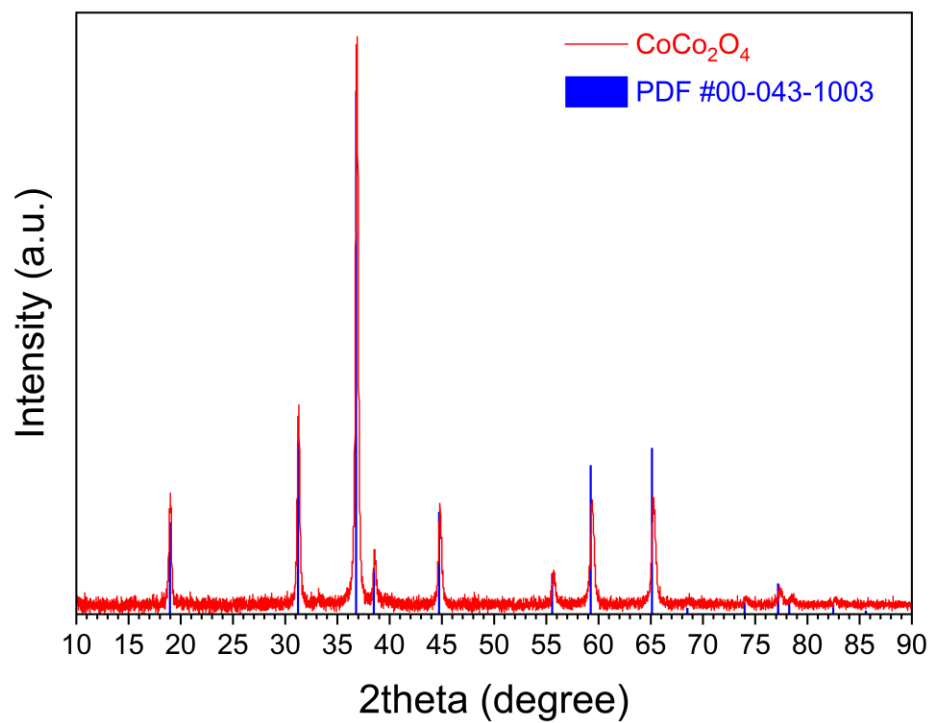
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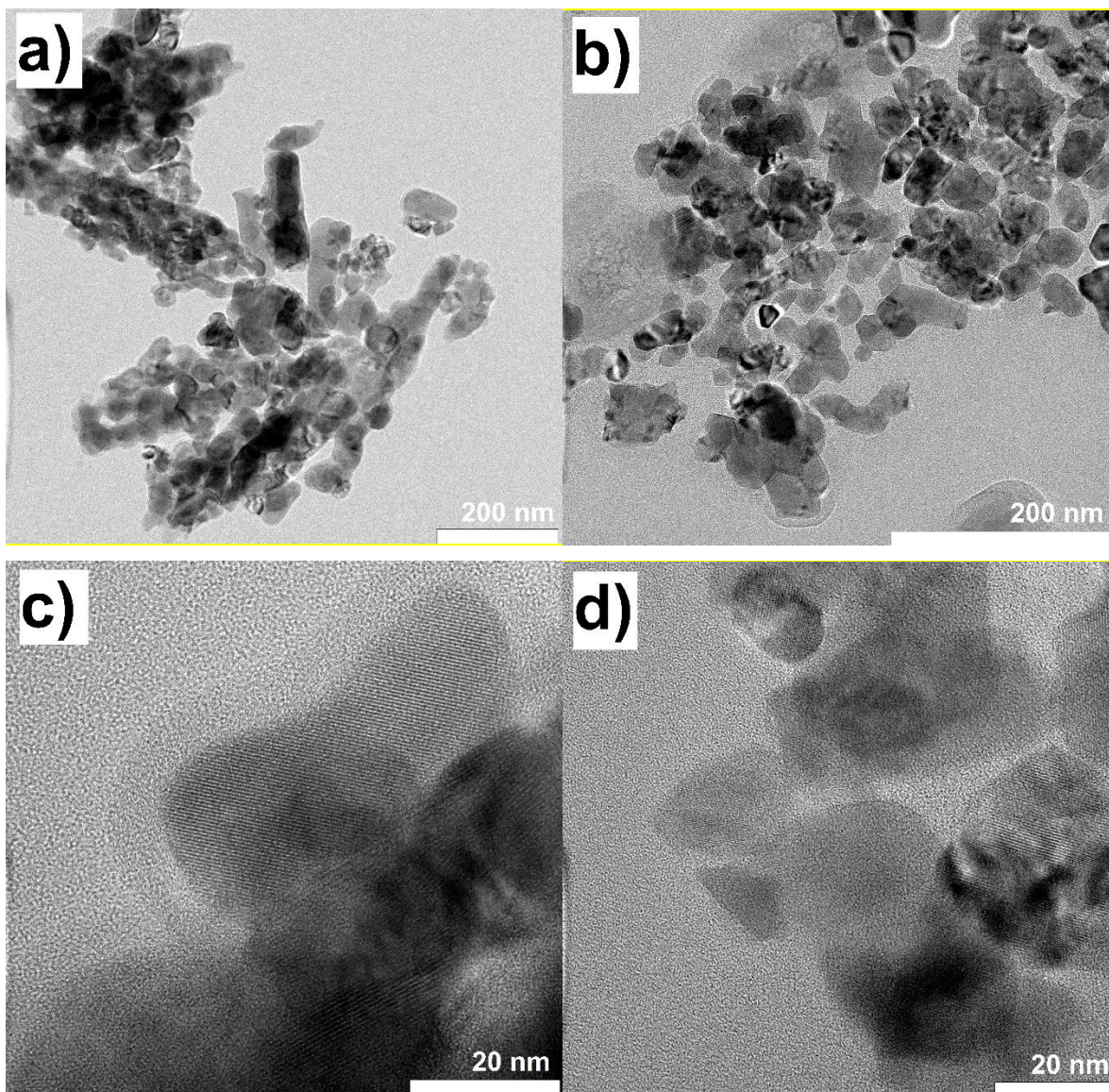
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**Figure S1.** XRD profile of  $\text{CoCo}_2\text{O}_4$  spinel oxide sample prepared by calcination of the CCO precursor at 480 °C, 10 hours, under air atmosphere (heating rate = 1 °C·min<sup>-1</sup>).



**Figure S2.** TEM images of the (a, c)  $\text{ZnCo}_2\text{O}_4$  and (b, d)  $\text{NiCo}_2\text{O}_4$  spinel oxides obtained by field-emission transmission electron microscope (FETEM, JEOL-JEM-3100F, operating voltage = 300 kV). The spinel oxide sample was prepared by calcining the prepared hydrothermal derived ZCO and NCO precursors at 480 °C in muffle furnace, under air flow for 10 hours (heating rate = 1 °C·min<sup>-1</sup>).

**Table S1.** XRD peak positions of the obtained product (at room temperature) after the thermal conversion of the CCO precursor.

$2\theta$ (°)	Lattice spacing (Å)	h	k	l	Space group	Crystalline phase
12.71	4.6661	1	1	1	Fd-3m	<sup>a</sup> CoCo <sub>2</sub> O <sub>4</sub>
20.84	2.8569	2	2	0	Fd-3m	<sup>a</sup> CoCo <sub>2</sub> O <sub>4</sub>
24.29	2.4551	1	1	1	Fm-3m	<sup>b</sup> CoO
24.50	2.4349	3	1	1	Fd-3m	<sup>a</sup> CoCo <sub>2</sub> O <sub>4</sub>
25.61	2.3311	2	2	2	Fd-3m	<sup>a</sup> CoCo <sub>2</sub> O <sub>4</sub>
28.10	2.1283	2	0	0	Fm-3m	<sup>b</sup> CoO
29.66	2.0186	4	0	0	Fd-3m	<sup>a</sup> CoCo <sub>2</sub> O <sub>4</sub>
36.56	1.6470	4	2	2	Fd-3m	<sup>a</sup> CoCo <sub>2</sub> O <sub>4</sub>
38.85	1.5535	5	1	1	Fd-3m	<sup>a</sup> CoCo <sub>2</sub> O <sub>4</sub>
40.17	1.5042	2	2	0	Fm-3m	<sup>b</sup> CoO
42.46	1.4267	4	4	0	Fd-3m	<sup>a</sup> CoCo <sub>2</sub> O <sub>4</sub>

<sup>a</sup>PDF #00-043-1003, <sup>b</sup>PDF #00-048-1719

**Table S2.** XRD peak positions of the obtained product (at room temperature) after the thermal conversion of the ZCO precursor.

$2\theta$ (°)	Lattice spacing (Å)	h	k	l	Space group	Crystalline phase
12.71	4.6661	1	1	1	Fd-3m	$\text{ZnCo}_2\text{O}_4$
20.84	2.8563	2	2	0	Fd-3m	$\text{ZnCo}_2\text{O}_4$
24.49	2.4361	3	1	1	Fd-3m	$\text{ZnCo}_2\text{O}_4$
25.59	2.3331	2	2	2	Fd-3m	$\text{ZnCo}_2\text{O}_4$
29.64	2.0200	4	0	0	Fd-3m	$\text{ZnCo}_2\text{O}_4$
36.56	1.6471	4	2	2	Fd-3m	$\text{ZnCo}_2\text{O}_4$
38.84	1.5538	5	1	1	Fd-3m	$\text{ZnCo}_2\text{O}_4$
42.45	1.4269	4	4	0	Fd-3m	$\text{ZnCo}_2\text{O}_4$

$\text{PDF 00-023-1390}$

**Table S3.** XRD peak positions of the obtained product (at room temperature) after the thermal conversion of the NCO precursor.

$2\theta$ (°)	Lattice spacing (Å)	h	k	l	Space group	Crystalline phase
12.72	4.6635	1	1	1	Fd-3m	<sup>d</sup> NiCo <sub>2</sub> O <sub>4</sub>
20.83	2.8583	2	2	0	Fd-3m	<sup>d</sup> NiCo <sub>2</sub> O <sub>4</sub>
24.46	2.4389	3	1	1	Fd-3m	<sup>d</sup> NiCo <sub>2</sub> O <sub>4</sub>
25.61	2.3310	2	2	2	Fd-3m	<sup>d</sup> NiCo <sub>2</sub> O <sub>4</sub>
29.63	2.0204	4	0	0	Fd-3m	<sup>d</sup> NiCo <sub>2</sub> O <sub>4</sub>
36.50	1.6497	4	2	2	Fd-3m	<sup>d</sup> NiCo <sub>2</sub> O <sub>4</sub>
38.84	1.5538	5	1	1	Fd-3m	<sup>d</sup> NiCo <sub>2</sub> O <sub>4</sub>
42.38	1.4292	4	4	0	Fd-3m	<sup>d</sup> NiCo <sub>2</sub> O <sub>4</sub>

<sup>d</sup>PDF 01-073-1702