

# Aromatic diamines as halogen bond acceptors

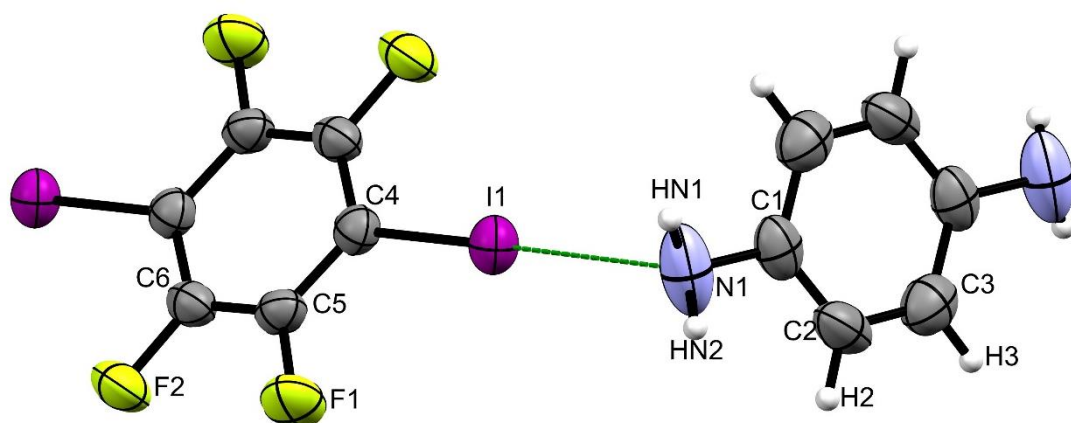
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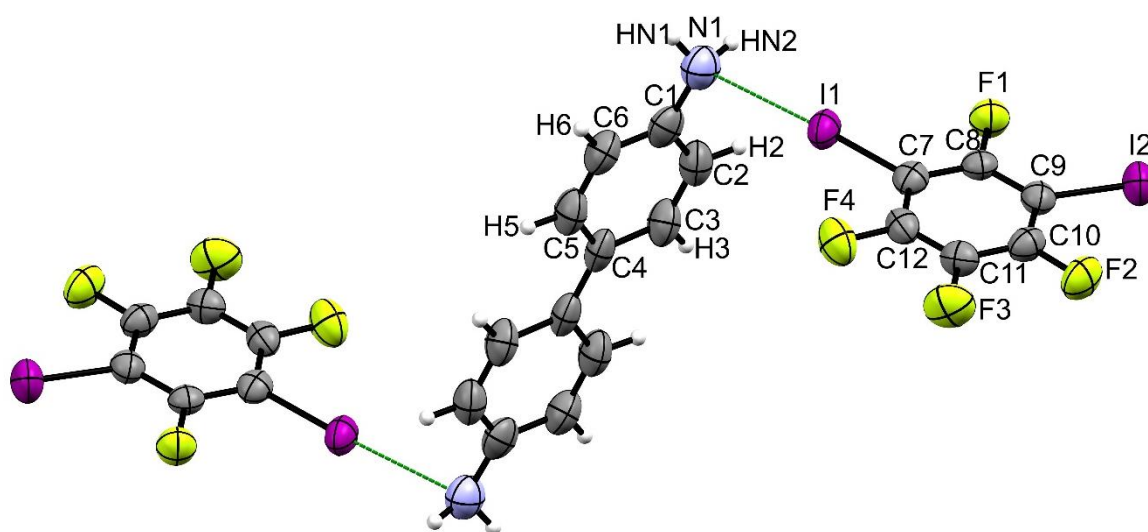
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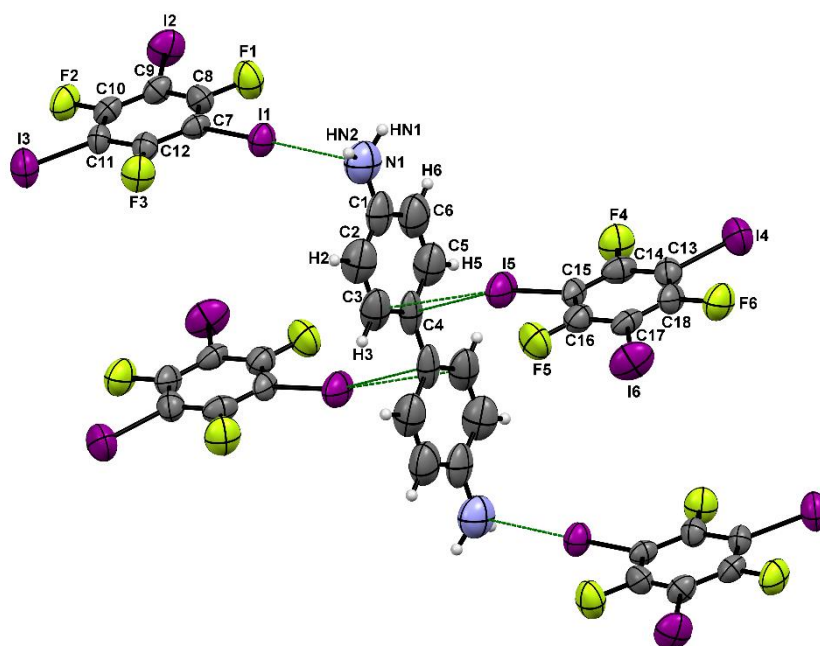
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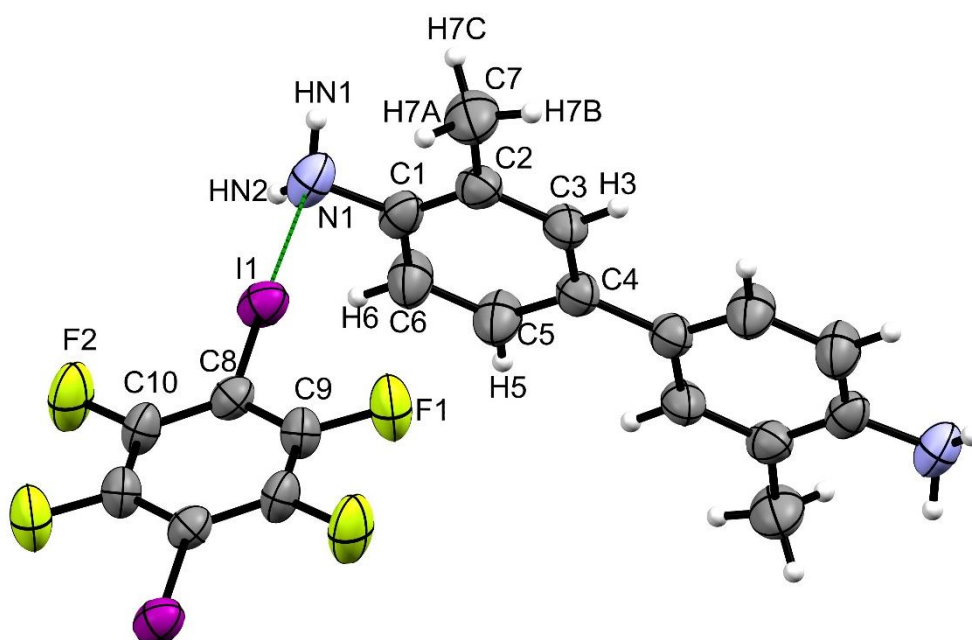
**Figure S1.** Molecular structure of (pphda)(14tfib) showing the atom-labelling scheme. Displacement ellipsoids are drawn at the 50 % probability level, and H atoms are shown as small spheres of arbitrary radius.



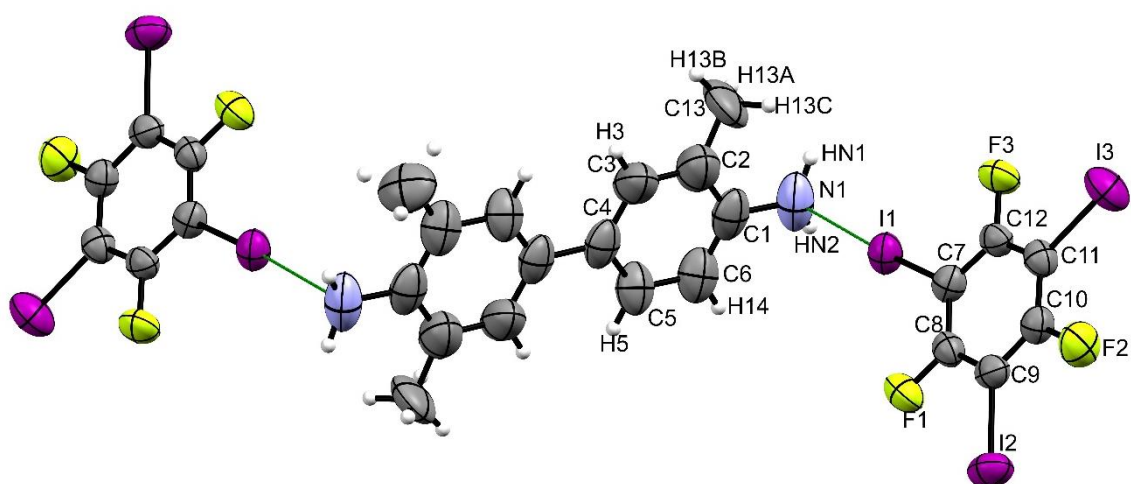
**Figure S2.** Molecular structure of (bnzd)(13tfib)<sub>2</sub> showing the atom-labelling scheme. Displacement ellipsoids are drawn at the 50 % probability level, and H atoms are shown as small spheres of arbitrary radius.



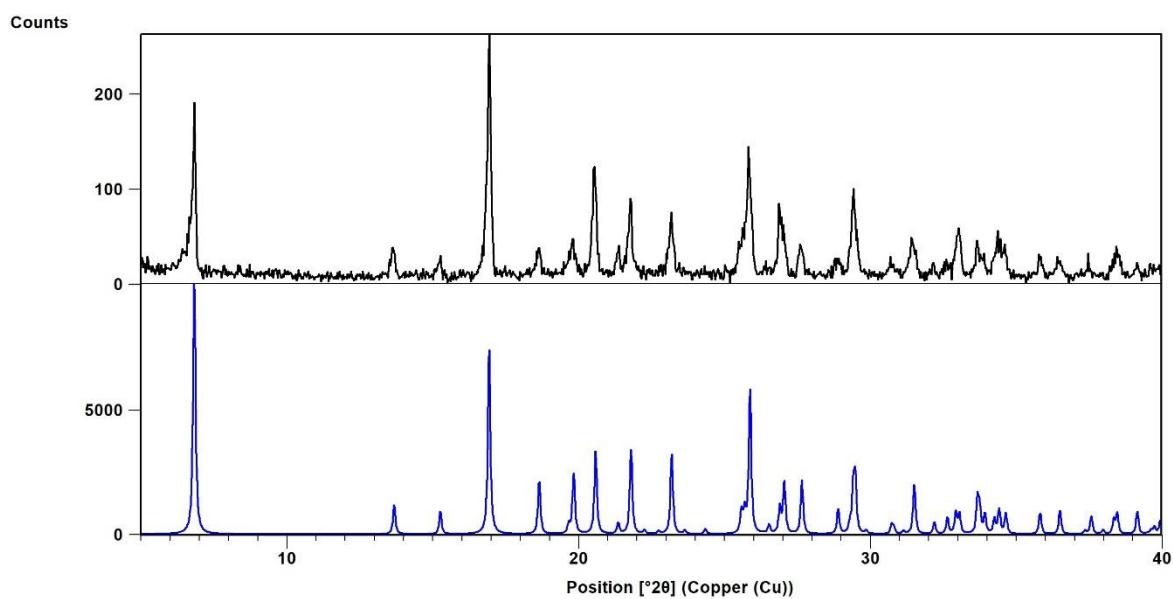
**Figure S3.** Molecular structure of **(bnzd)(135tfib)<sub>4</sub>** showing the atom-labelling scheme. Displacement ellipsoids are drawn at the 50 % probability level, and H atoms are shown as small spheres of arbitrary radius.



**Figure S4.** Molecular structure of **(otol)(14tfib)** showing the atom-labelling scheme. Displacement ellipsoids are drawn at the 50 % probability level, and H atoms are shown as small spheres of arbitrary radius.



**Figure S5.** Molecular structure of **(otol)(135tfib)<sub>2</sub>** showing the atom-labelling scheme. Displacement ellipsoids are drawn at the 50 % probability level, and H atoms are shown as small spheres of arbitrary radius.



**Figure S6.** Measured (black) and calculated (blue) PXRD patterns of **(pphda)(14tfib)**.

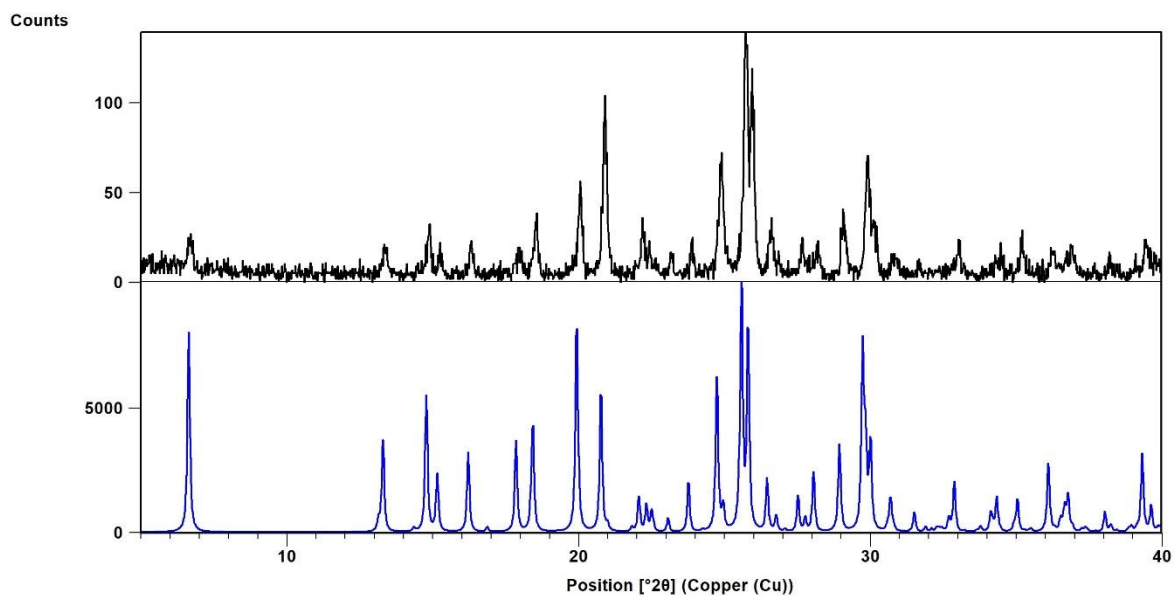


Figure S7. Measured (black) and calculated (blue) PXRD patterns of **(bnzd)(13tfib)<sub>2</sub>**.

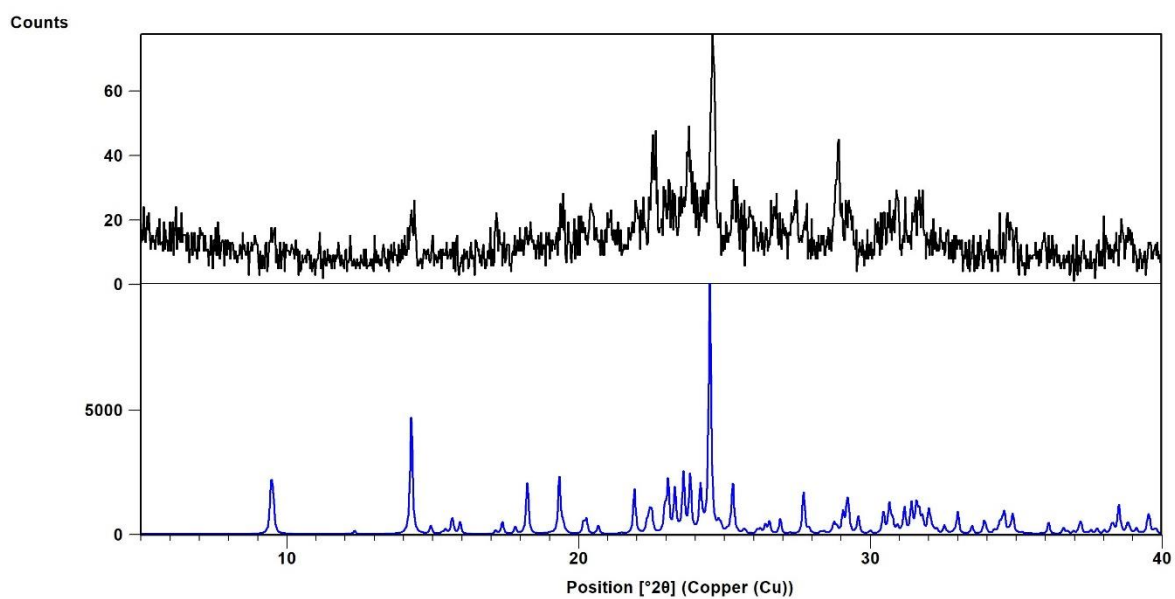
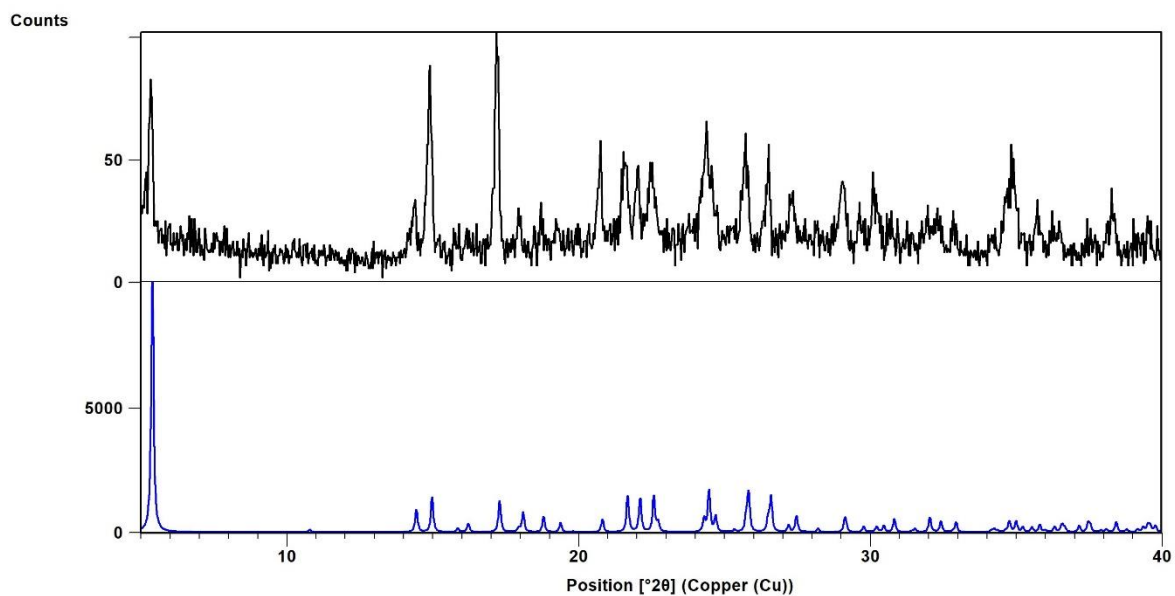
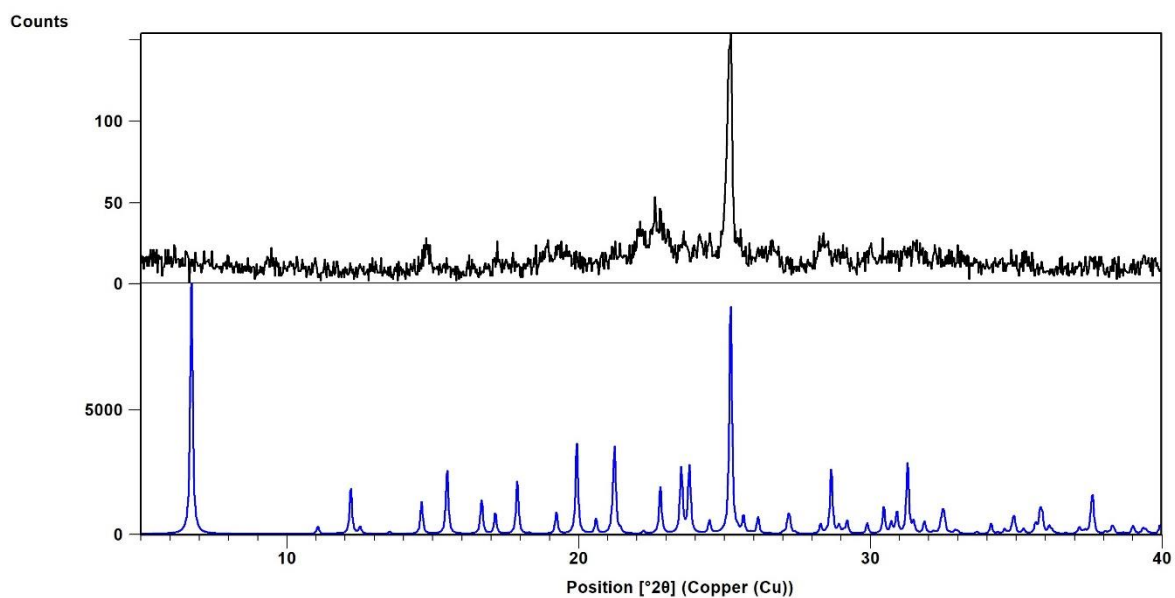


Figure S8. Measured (black) and calculated (blue) PXRD patterns of **(bnzd)(135tfib)<sub>4</sub>**.



**Figure S9.** Measured (black) and calculated (blue) PXRd patterns of (otol)(14tfib).



**Figure S10.** Measured (black) and calculated (blue) PXRd patterns of (otol)(135tfib)<sub>2</sub>.

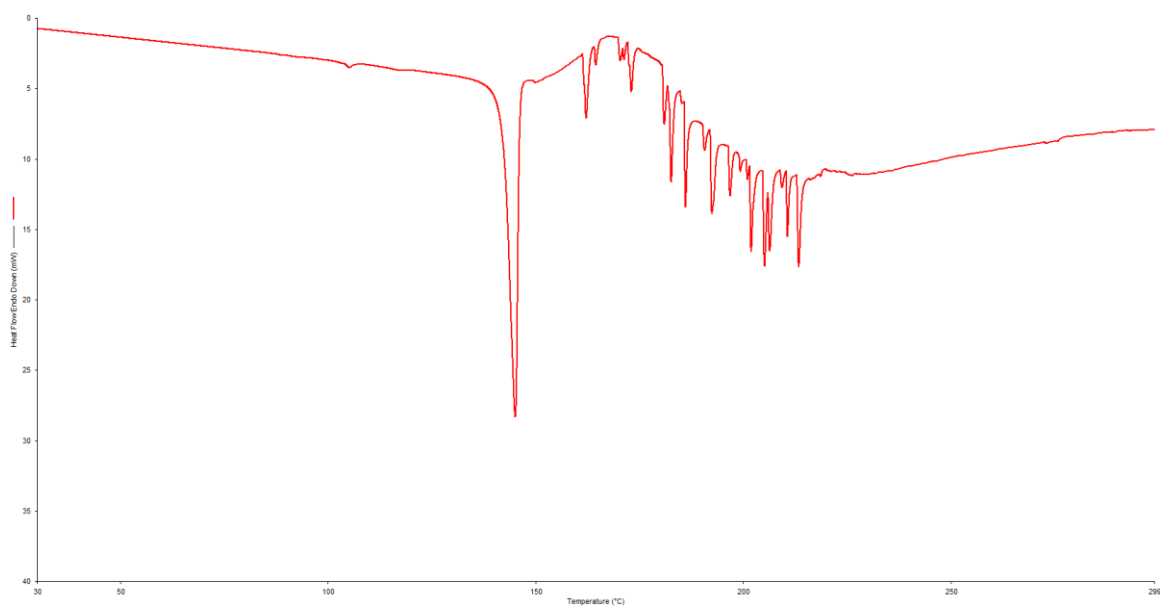


Figure S11. DSC thermogram of product of solution synthesis of (pphda)(14tfib).

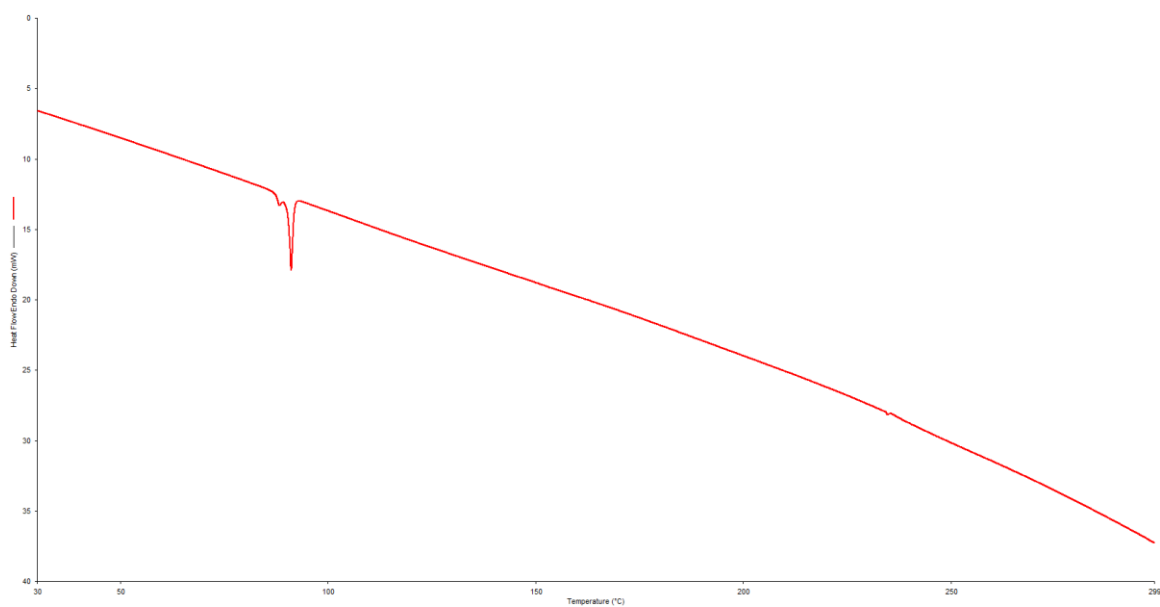


Figure S12. DSC thermogram of product of solution synthesis of (bnzd)(13tfib)<sub>2</sub>.

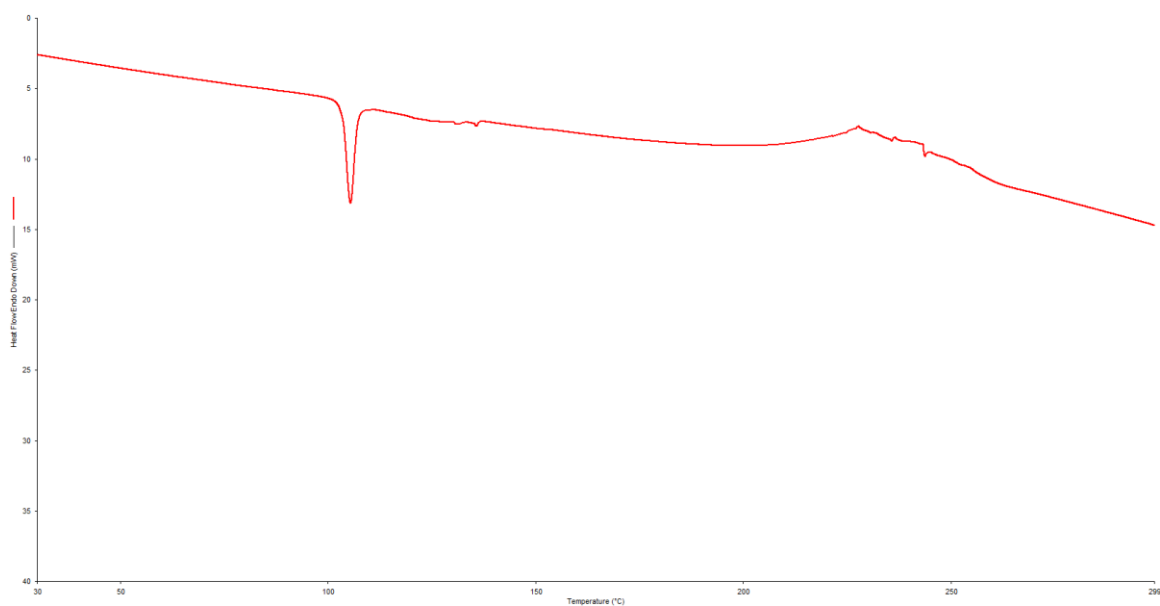


Figure S13. DSC thermogram of product of solution synthesis of **(bnzd)(135tfib)<sub>4</sub>**.

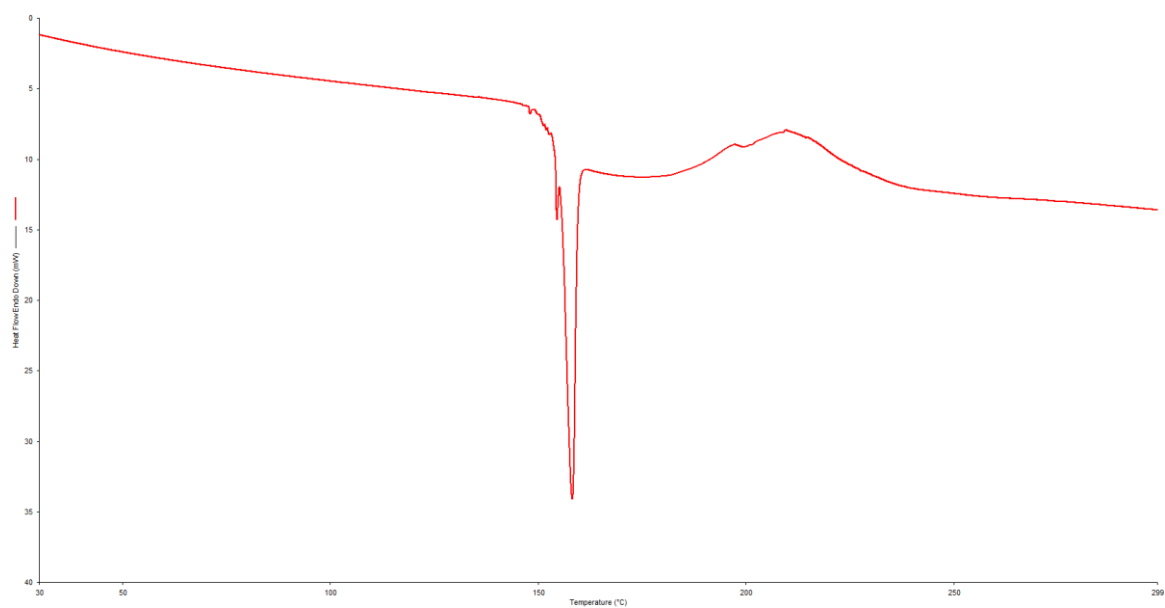
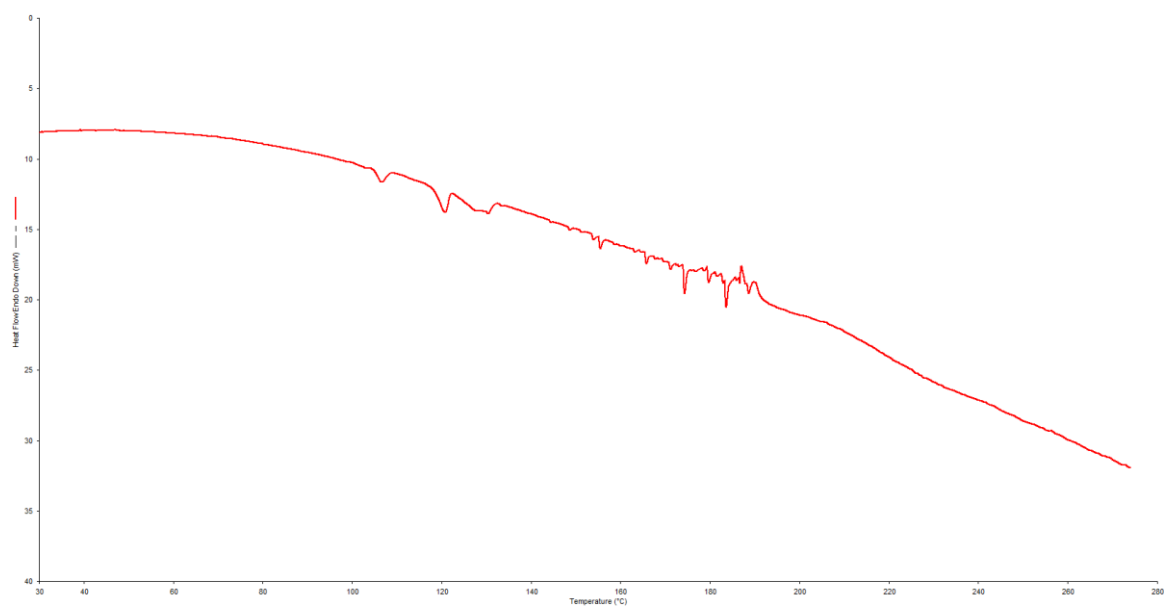


Figure S14. DSC thermogram of product of solution synthesis of **(otol)(14tfib)**.





**Figure S15.** DSC thermogram of (otol)(135tfib)<sub>2</sub>.

**Table S1.** Masses of reactants and volumes of solvents for successful crystallization experiments.

Reactants		Molar ratio A : D	<i>m</i> (A) / mg	<i>m</i> (D) / mg	Solvent (L)	<i>V</i> (L) / mL
A	D					
<b>otol</b>	<b>14tfib</b>	1 : 1	10	20	EtOH	2
<b>otol</b>	<b>135tfib</b>	1 : 2	10	40	EtOH	10
<b>bnzd</b>	<b>13tfib</b>	1 : 2	10	40	EtOH	3
<b>bnzd</b>	<b>135tfib</b>	1 : 2	10	50	EtOH	3
<b>pphda</b>	<b>14tfib</b>	1 : 1	10	35	EtOH/Me <sub>2</sub> CO	3

**Table S2.** An overview and crystallographic data of the prepared compounds.

	(pphda)(14tfib)	(bnzd)(13tfib) <sub>2</sub>	(bnzd)(135tfib) <sub>4</sub>	(otol)(14tfib)	(otol)(135tfib) <sub>2</sub>
Molecular formula	C <sub>12</sub> H <sub>8</sub> F <sub>4</sub> I <sub>2</sub> N <sub>2</sub>	C <sub>24</sub> H <sub>12</sub> F <sub>8</sub> I <sub>4</sub> N <sub>2</sub>	C <sub>30</sub> H <sub>20</sub> F <sub>12</sub> I <sub>12</sub> N <sub>2</sub>	C <sub>20</sub> H <sub>20</sub> F <sub>4</sub> I <sub>2</sub> N <sub>2</sub>	C <sub>26</sub> H <sub>16</sub> F <sub>6</sub> I <sub>6</sub> N <sub>2</sub>
$M_r$	510.00	987.96	2159.28	618.18	1231.81
Crystal system	triclinic	orthorhombic	triclinic	triclinic	monoclinic
Space group	$P-1$	$P bca$	$P-1$	$P-1$	$P 2_1/c$
Crystal data:					
$a / \text{\AA}$	4.6046(3)	8.5466(5)	9.4338(12)	5.0079(3)	7.5646(3)
$b / \text{\AA}$	6.1794(5)	11.9758(7)	11.7698(15)	6.1910(3)	26.2421(13)
$c / \text{\AA}$	13.2376(11)	26.6168(9)	11.9237(11)	16.5464(8)	8.4074(4)
$\alpha / ^\circ$	81.193(7)	90	75.793(9)	94.147(4)	90
$\beta / ^\circ$	80.366(6)	90	86.652(9)	96.507(5)	94.025(4)
$\gamma / ^\circ$	82.247(6)	90	82.035(8)	96.482(5)	90
$V / \text{\AA}^3$	364.65(5)	2722.3(2)	1270.7(3)	504.55(5)	1664.84(13)
$Z$	1	4	1	1	2
$D_{\text{calc}} / \text{g cm}^{-3}$	2.322	2.409	2.905	2.021	2.457
$\lambda(\text{MoK}\alpha) / \text{\AA}$	0.71073	0.71073	0.71073	0.71073	0.71073
$T / \text{K}$	295	295	295	295	295
Crystal size / mm <sup>3</sup>	0.28 x 0.24 x 0.16	0.26 x 0.15 x 0.08	0.30 x 0.12 x 0.10	0.22 x 0.22 x 0.18	0.18 x 0.18 x 0.14
$\mu / \text{mm}^{-1}$	4.346	4.648	7.385	3.160	5.651
$F(000)$	236	1816	986	292	1116
Refl. collected/unique	3031 / 1576	16542 / 2657	7917 / 4425	4809 / 2188	10315 / 3620
Data/restraints / parameters	99	179	286	134	200
$\Delta\rho_{\text{max}}, \Delta\rho_{\text{min}} / \text{e \AA}^{-3}$	0.391; -1.074	0.565; -0.747	2.168; -1.542	0.501; -0.705	0.710; -0.968
$R[F^2 > 4\sigma(F^2)]$	0.0284	0.0372	0.0597	0.0278	0.0413
$wR(F^2)$	0.0661	0.0846	0.1710	0.0724	0.1050
Goodness-of-fit, $S$	1.083	1.095	1.025	1.035	1.043

Total electron energies and Cartesian coordinates for optimised structures:

### Aniline

$E = -287.78568793$  a.u.

1	C	0	0.935599	0.000005	-0.007480
2	C	0	0.219775	1.203289	-0.004196
3	C	0	-1.167913	1.197851	0.003296
4	C	0	-1.875425	0.000000	0.006870
5	C	0	-1.167905	-1.197856	0.003292
6	C	0	0.219776	-1.203289	-0.004195
7	H	0	0.758374	2.144371	-0.011304
8	H	0	-1.700290	2.141251	0.007302
9	H	0	-2.957291	-0.000010	0.013265
10	H	0	-1.700289	-2.141252	0.007294
11	H	0	0.758387	-2.144364	-0.011292
12	N	0	2.328248	0.000000	-0.073409
13	H	0	2.779966	0.836168	0.261529
14	H	0	2.779961	-0.836167	0.261542

### Methylamine

$E = -95.90249789$  a.u.

1	N	0	-0.749282	-0.000001	-0.120150
2	H	0	-1.153826	-0.813506	0.327764
3	H	0	-1.153822	0.813506	0.327765
4	C	0	0.706511	-0.000001	0.017425
5	H	0	1.116606	-0.878107	-0.485304
6	H	0	1.116590	0.878184	-0.485179
7	H	0	1.080360	-0.000070	1.051459

### Iodopentafluorobenzene

$E = -1025.91139651$  a.u.

1	C	0	0.209838	-0.000111	-0.000074
2	C	0	-0.503142	1.194287	-0.000110
3	C	0	-1.891350	1.201319	0.000048
4	C	0	-2.587057	-0.000091	0.000134
5	C	0	-1.891563	-1.201511	0.000027
6	C	0	-0.503205	-1.194260	-0.000146
7	F	0	0.129251	2.368821	-0.000256
8	F	0	-2.559275	2.354530	0.000047
9	F	0	-3.917117	0.000390	0.000074
10	F	0	-2.559186	-2.354740	0.000079
11	F	0	0.129154	-2.368926	-0.000268
12	I	0	2.301763	0.000029	0.000069

### Aniline dimer

$E = -575.46220950$  a.u.

1	N	0	1.047720	2.003497	0.205657
2	C	0	1.819415	0.858136	0.033235
3	C	0	3.212068	0.907220	0.182737
4	C	0	3.978702	-0.243629	0.064498
5	C	0	3.381502	-1.471361	-0.202532
6	C	0	1.998591	-1.524972	-0.350323
7	C	0	1.222004	-0.380967	-0.236147
8	H	0	1.536113	2.874535	0.072773
9	H	0	0.128993	2.008278	-0.217181
10	H	0	3.691254	1.856522	0.396362
11	H	0	5.053743	-0.177913	0.182068
12	H	0	3.981362	-2.367235	-0.294495
13	H	0	1.511881	-2.470285	-0.558453
14	H	0	0.148020	-0.445366	-0.347800
15	N	0	-2.117948	2.188747	-0.425500
16	C	0	-2.297486	0.827506	-0.134999
17	C	0	-2.555684	-0.089371	-1.157454
18	C	0	-2.628373	-1.447803	-0.879026
19	C	0	-2.446675	-1.916428	0.417654
20	C	0	-2.192463	-1.004078	1.436376
21	C	0	-2.117617	0.355084	1.167855
22	H	0	-2.566670	2.502905	-1.272844
23	H	0	-2.311102	2.811784	0.344687
24	H	0	-2.682223	0.263547	-2.174608
25	H	0	-2.823186	-2.143906	-1.685263
26	H	0	-2.497063	-2.975895	0.630025
27	H	0	-2.037915	-1.352485	2.449656
28	H	0	-1.885516	1.053779	1.962718

**Aniline\_iodopentafluorobenzene** $E = -1313.64758258$  a.u.

1	I	0	-0.497474	-0.619114	-0.580661
2	F	0	2.048077	-2.373829	0.383395
3	F	0	1.147372	2.157433	-0.670610
4	C	0	1.514481	-0.128781	-0.161208
5	C	0	1.970397	1.179127	-0.280061
6	C	0	2.422896	-1.097387	0.249313
7	F	0	3.703613	2.779290	-0.119772
8	F	0	4.598875	-1.725289	0.928464
9	C	0	4.177129	0.533531	0.409949
10	C	0	3.743689	-0.778980	0.534709
11	C	0	3.287335	1.516852	0.000615
12	C	0	-5.136237	1.778769	0.192053
13	H	0	-5.389350	2.778474	-0.138002
14	C	0	-4.471511	-0.785931	1.017986
15	H	0	-4.202814	-1.783436	1.346302
16	C	0	-5.487813	1.356819	1.469786
17	N	0	-3.378658	-1.204678	-1.114890
18	H	0	-3.492401	-1.010087	-2.099282
19	H	0	-3.491259	-2.190859	-0.928955
20	C	0	-4.458037	0.929541	-0.671509
21	H	0	-4.177573	1.271239	-1.661184
22	C	0	-4.123569	-0.363890	-0.266385
23	H	0	-6.016140	2.021391	2.140257
24	C	0	-5.149582	0.070175	1.875045
25	H	0	-5.413859	-0.273154	2.867382
26	F	0	5.442624	0.848302	0.681963

### Methylamine dimer

$E = -191.81295444$  a.u.

1	N	0	1.405064	-0.612568	-0.427370
2	H	0	1.982661	-1.444117	-0.396004
3	H	0	0.517481	-0.837375	0.017212
4	N	0	-1.659380	-0.621527	0.334857
5	H	0	-2.218699	-0.675898	1.177404
6	H	0	-2.019829	-1.317211	-0.307565
7	C	0	-1.728531	0.717291	-0.259777
8	H	0	-1.355969	1.446985	0.460905
9	H	0	-1.069223	0.751787	-1.127880
10	H	0	-2.733298	1.030072	-0.571390
11	C	0	2.058180	0.492085	0.271553
12	H	0	3.000944	0.737816	-0.222630
13	H	0	1.423845	1.379264	0.211933
14	H	0	2.274403	0.311091	1.334953

### Methylamine\_iodopentafluorobenzene

$E = -1121.82761655$  a.u.

1	I	0	-1.666467	-0.146556	-0.131432
2	F	0	0.661680	-2.394498	0.034305
3	F	0	0.425028	2.336359	-0.104682
4	C	0	0.453909	-0.033698	-0.038484
5	C	0	1.112589	1.189651	-0.044090
6	C	0	1.231564	-1.183436	0.025872
7	F	0	3.108095	2.459843	0.005089
8	F	0	3.343395	-2.243536	0.143837
9	C	0	3.252487	0.109592	0.075736
10	C	0	2.617311	-1.124311	0.082878
11	C	0	2.497258	1.272154	0.012070
12	N	0	-4.517990	-0.285436	-0.228251

13	H	0	-4.804661	-0.093458	-1.180734
14	H	0	-4.781914	-1.240855	-0.018907
15	F	0	4.582425	0.177906	0.129570
16	C	0	-5.125490	0.658554	0.715505
17	H	0	-4.769327	0.434813	1.721309
18	H	0	-4.797708	1.668693	0.469257
19	H	0	-6.221063	0.640620	0.722749