

Supplementary Materials: Lewis Acid Properties of Tetrel Tetrafluorides—The Coincidence of the σ -Hole Concept with the QTAIM Approach

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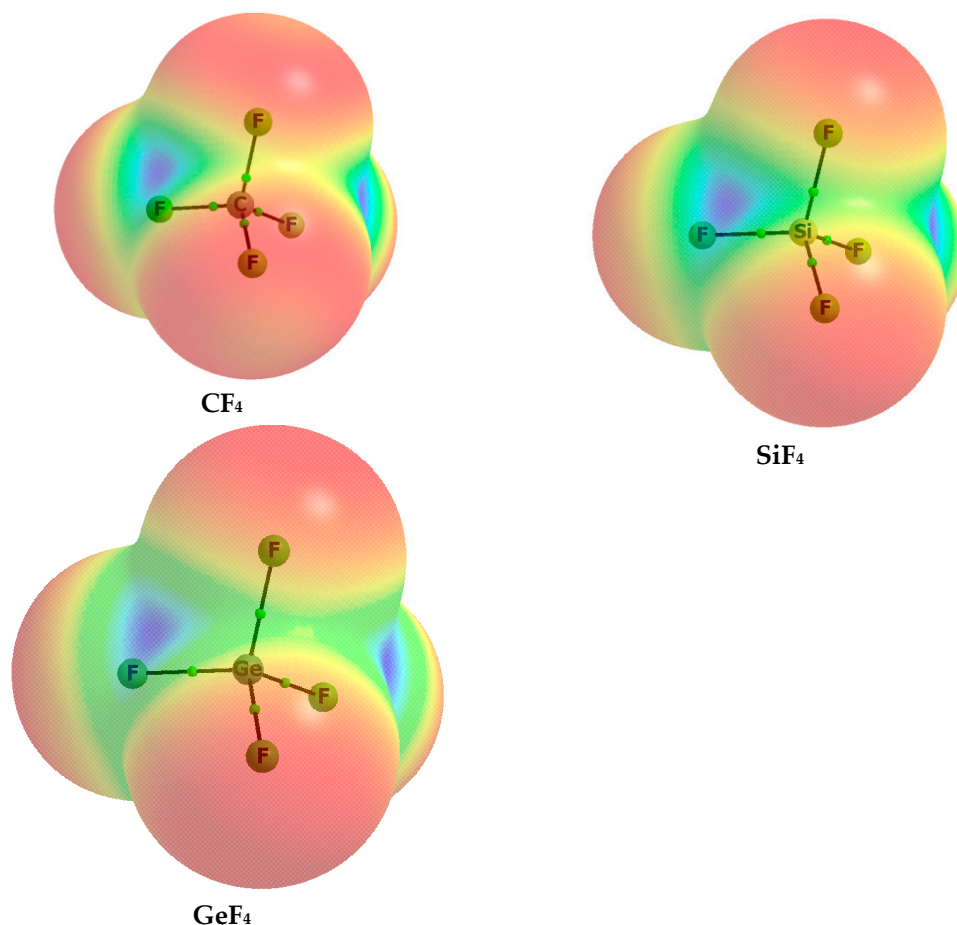


Figure S1. The maps of the electrostatic potential (EP) calculated at the 0.001 au molecular electron density surfaces for ZF₄; red and blue colors correspond to negative and positive EP, respectively; MP2/aug-cc-pVTZ level of calculations.

Table S1. The energetic parameters (in kcal/mol); the interaction energy, E_{int} , and the binding energy, E_{bin} , corrected for BSSE, the deformation energy, E_{def} . The tetrel–Lewis base centre distances are included, Z...B (i.e., Z...N or Z...As; Z = C, Si, Ge, in Å).

Complex	Z...B	E_{int}	E_{bin}	BSSE	E_{def}
CF ₄ ...NH ₃	3.658	−1.1	−1.1	0.3	0.0
CF ₄ ...(NH ₃) ₂	1.658	−29.1	90.1	3.0	119.2
CF ₄ ...(NH ₃) ₂ *	1.658	−76.8	91.9	6.5	168.7
SiF ₄ ...NH ₃	2.072	−29.6	−8.5	2.5	21.1
SiF ₄ ...(NH ₃) ₂	1.940	−44.4	−14.7	2.4	29.7
SiF ₄ ...(NH ₃) ₂ *	1.940	−93.9	−23.7	4.4	70.2
GeF ₄ ...NH ₃ ^a	2.084	−34.2	−16.7	4.7	17.5
GeF ₄ ...NH ₃	2.080	−33.8	−16.2	4.7	17.6
GeF ₄ ...(NH ₃) ₂ ^a	2.001	−43.1	−19.8	4.7	23.3
GeF ₄ ...(NH ₃) ₂	1.997	−42.4	−19.2	4.9	23.2
GeF ₄ ...(NH ₃) ₂ ^{*,a}	2.001	−91.9	−35.1	10.8	56.8

GeF ₄ ...(NH ₃) ₂ *	1.997	-90.5	-34.2	10.9	56.3
GeF ₄ ...AsH ₃ ^a	3.381	-3.1	-1.9	2.1	1.2
GeF ₄ ...AsH ₃	3.399	-2.8	-1.8	1.9	1.0
GeF ₄ ...(AsH ₃) ₂ ^a	2.550	-23.3	8.6	5.3	31.9
GeF ₄ ...(AsH ₃) ₂	2.543	-22.3	10.3	5.1	32.6
GeF ₄ ...(AsH ₃) ₂ ^{*,a}	2.550	-50.3	9.8	10.5	60.1
GeF ₄ ...(AsH ₃) ₂ *	2.543	-48.1	11.5	10.1	59.6

* Interaction and binding energies are related to three components (see text for explanation).

^a Relativistic effects not taken into account (MP2/aug-cc-pVTZ level).

Table S2. Electrostatic potential local maxima and minima for ZF₄, ZF₄...NH₃ and GeF₄...AsH₃ species (in au); for the species containing Ge and As centres these values are the same if relativistic effects are taken into account; in a case of ZF₄...NH₃ complexes the local maximum is indicated since the global EP maximum corresponds to H-centres, in a case of the GeF₄...AsH₃ complex that is the global EP maximum; the EP results for the surfaces of the electron density of 0.001 au are presented.

ZF ₄ or ZF ₄ -BH ₃	Max	Max Localization	Min	Min Localization
CF ₄	+0.034	F-C extension	-0.002	F-centre
SiF ₄	+0.072	F-Si extension	-0.012	F-centre
GeF ₄	+0.086	F-Ge extension	-0.012	F-centre
CF ₄ ...NH ₃	+0.025	F-C extension	-0.031	F...N area
SiF ₄ ...NH ₃	+0.018	F-Si extension	-0.053	F-centre
GeF ₄ ...NH ₃	+0.021	F-Ge extension	-0.053	F-centre
GeF ₄ ...AsH ₃	+0.069	F-Ge extension	-0.023	F-centre

Table S3. The QTAIM characteristics (in au) of the Z-F bond critical point; electron density at BCP, ρ_{BCP} , its laplacian, $\nabla^2\rho_{\text{BCP}}$, the total electron energy density at BCP, H_{BCP} , the potential electron energy density at BCP, V_{BCP} , the kinetic electron energy density at BCP, G_{BCP} .

Complex	ρ_{BCP}	V_{BCP}	G_{BCP}	H_{BCP}	$\nabla^2\rho_{\text{BCP}}$
CF ₄ ...NH ₃	0.305	-0.908	0.410	-0.498	-0.354
CF ₄ ...NH ₃ *	0.299	-0.879	0.393	-0.486	-0.369
CF ₄ ...(NH ₃) ₂	0.177	-0.274	0.110	-0.164	-0.217
SiF ₄ ...NH ₃	0.134	-0.356	0.316	-0.040	1.103
SiF ₄ ...NH ₃ *	0.135	-0.361	0.317	-0.045	1.087
SiF ₄ ...(NH ₃) ₂	0.114	-0.278	0.244	-0.033	0.844
GeF ₄ ...NH ₃	0.149	-0.354	0.283	-0.071	0.846
GeF ₄ ...NH ₃ *	0.153	-0.365	0.289	-0.076	0.853
GeF ₄ ...(NH ₃) ₂	0.131	-0.290	0.231	-0.058	0.692
GeF ₄ ...AsH ₃	0.163	-0.405	0.321	-0.084	0.949
GeF ₄ ...AsH ₃ *	0.162	-0.400	0.318	-0.083	0.940
GeF ₄ ...(AsH ₃) ₂	0.134	-0.297	0.236	-0.060	0.704

* for the Z-F bond located at the opposite side of the NH₃ or AsH₃ ligand.