

Article

# Pr<sub>9.33</sub>(SiO<sub>4</sub>)<sub>6</sub>O<sub>2</sub> Crystals: Czochralski Growth and Near UV-Visible FR Performance

Xin Chen and Jianzhong Chen \*

College of Chemistry, Fuzhou University, Fuzhou 350116, China; xin.chen@fzu.edu.cn

\* Correspondence: j.z.chen@fzu.edu.cn; Tel.: +86-591-22866243

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**Abstract:** Pr<sub>9.33</sub>(SiO<sub>4</sub>)<sub>6</sub>O<sub>2</sub> crystals have been grown by the Cz-method for the first time for near UV–visible Faraday rotation applications. Rietveld refinement of XRD data demonstrates that the compound crystallizes in the oxyapatite structure with space group *P*6<sub>3</sub>/*m*, with oxyapatite structure. In contrast to Tb<sub>3</sub>Ga<sub>5</sub>O<sub>12</sub>, Pr<sub>9.33</sub>(SiO<sub>4</sub>)<sub>6</sub>O<sub>2</sub> crystal shows a higher transparency in the UV–visible wavelength region, and a shorter cutoff at 270 nm. The Faraday rotation performance and the temperature-dependence of the field-cooled (FC) and zero-field-cooled (ZFC) magnetic susceptibility have been investigated, which indicate that the Pr<sub>9.33</sub>(SiO<sub>4</sub>)<sub>6</sub>O<sub>2</sub> crystal exhibits paramagnetic behavior in the experimental temperature range from 2 to 300 K and yields a larger FR angle which rapidly increases towards the cutoff. Pr<sub>9.33</sub>(SiO<sub>4</sub>)<sub>6</sub>O<sub>2</sub> crystal is therefore a promising magneto-optical crystal in particular for potential FR applications in the near UV–visible spectral region.

**Keywords:** magneto-optic material; Czochralski technique; magnetism; Faraday rotation

## 1. Introduction

The Faraday isolators (FIs) using the non-reciprocity principle are poised to be a significant player in a modern optical information processing system to eliminate optical noise and feedback [1–5]. Yttrium-iron garnet (Y<sub>3</sub>Fe<sub>5</sub>O<sub>12</sub>, YIG), the most commonly used FI material in the near IR spectral region so far, is transparent in the 1.2–5.0 μm wavelength regions and exhibits an extremely large Faraday rotation (FR) angle [6,7]. Nevertheless, with the rapid expansion of visible solid-state lasers and laser-diodes, as well as excimer lasers in the UV, the demand of FIs operated at these spectral regions is rapidly promoting [8–10], while the conventional Yttrium-iron garnet crystals are not practical because of their poor transparency in this wavelength region [11].

The most widely employed magneto-optical (MO) materials for the visible and near IR spectral regions are terbium-gallium garnet (Tb<sub>3</sub>Ga<sub>5</sub>O<sub>12</sub>, TGG) crystals [1,12] and Tb-doped glasses [13–15]. Nevertheless, due to poor relative stability and low thermal conductivity, Tb-doped glasses could not be employed in high-power lasers [16]. For another, TGG crystal, as a crystalline MO material, is a congruently melting material and therefore can be synthesized by the Czochralski (Cz) method. However, because of the evaporation and decomposition of Ga<sub>2</sub>O<sub>3</sub> from the congruent melt during the Czochralski growth process, its growth technique is not exempt of difficulties [17,18]. Furthermore, owing to the electric-dipole 4*f*–4*f* transitions of the MO active ion Tb<sup>3+</sup>, TGG crystal presents two physical drawbacks: the increasing transmission loss in the visible spectral region, and the strong absorption bands in the near UV [19]. Therefore, with the continuous development of fiber-lasers and laser-diodes operating at shorter spectral regions, the demand of FIs operating in the UV–visible wavelength region is rapidly increasing.

Among the rare earth ions, Tb<sup>3+</sup>, Ce<sup>3+</sup>, Pr<sup>3+</sup>, and Dy<sup>3+</sup> have been pointed to exhibit the best MO performance [20,21], and Pr<sup>3+</sup> is more transparent in the near UV. Taking into account the requirements

for a new generation of MO materials employed in the near UV–visible spectral region,  $\text{Pr}_{9.33}(\text{SiO}_4)_6\text{O}_2$  (PSO), a member of the oxyapatite-type silicate [22–24], which crystallizes in the hexagonal system and demonstrates great potential for application in phosphors [25] and high-power lasers [26,27], has attracted growing attention from our research team in recent years.

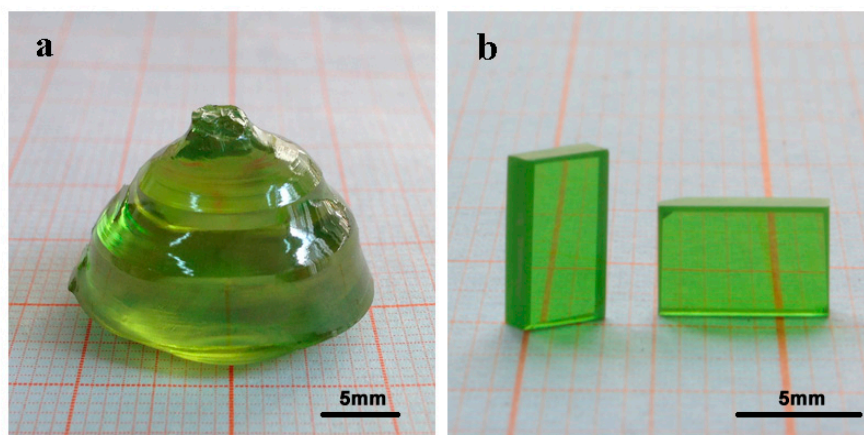
PSO crystal melts congruently and presents strong crystallizability [28], hence, in the present investigation, we try to adopt the common RF-heating Cz-method to grow PSO crystals. The optical transmission, magnetic, and magneto-optical performances have also been discussed.

## 2. Experimental

PSO single crystals were synthesized by the common Cz-technique with RF-heating [29]. The polycrystalline materials used for PSO crystals growth were synthesized by the solid-phase reaction method using high-purity initial reagents  $\text{Pr}_6\text{O}_{11}$ , and  $\text{SiO}_2$  powders. The stoichiometric amounts of starting materials were weighed accurately with the overweight 1 wt% of  $\text{SiO}_2$  in order to make up for its potential volatilization losses during the PSO crystals growth. The raw materials mixed homogeneously in a corundum mortar with the ethanol were pressed into tablets and then sintered at 1470 K in air for 23 h.

The polycrystalline tablets with single apatite phase were added into an iridium crucible with dimensions of  $\text{Ø}56 \times 43 \text{ mm}^2$ , and then heated by a radio frequency induction furnace under a protective atmosphere of  $\text{N}_2$ . After the PSO polycrystalline material was melted completely, the  $\langle 001 \rangle$  oriented PSO crystal, obtained through growth on the iridium wire, was used as seeds and dipped into the PSO melt with 15–20 rpm rotating rate and  $1.0\text{--}1.5 \text{ mm h}^{-1}$  pulling rate. After growth, the PSO crystal was pulled out of the melt and then annealed at a rate of  $15\text{--}60 \text{ K h}^{-1}$  to room temperature.

The photograph of the as-grown PSO single crystal is shown in Figure 1a. The PSO crystal is crack free and transparent with a light green coloration. Owing to the optically uniaxial crystal, PSO single crystal should be oriented along the crystallographic *c*-axis to avoid the effects of birefringence in the FR measurements. The PSO crystal was oriented by X-ray diffraction, and the wafers, cut along the (001) crystal plane, are shown in Figure 1b.



**Figure 1.** Photographs of as-grown PSO single crystal (a), and polished wafers (b).

The X-ray diffraction measurement for Rietveld refinement were carried out on a Rigaku-Ultima III X-ray diffractometer (Rigaku, Tokyo, Japan) using the  $\text{Cu-K}\alpha$  line ( $\lambda = 1.54051 \text{ \AA}$ ) between  $10^\circ < 2\theta < 80^\circ$  with a scanning step of  $0.01^\circ$  and a scanning rate of  $0.12^\circ/\text{min}$  at room temperature. Structural parameter refinement of PSO crystal was carried out by analyzing the X-ray diffraction data using the DBWS-9411 PC program [30]. The absorption spectrum of as-grown PSO crystal was tested by using a Lambda900 UV-Vis-NIR spectrophotometer (Perkin-Elmer, Waltham, MA, USA), over the spectral range of 200–800 nm at room temperature with a 2 nm resolution. The temperature-dependence

of the field-cooled (FC) and zero-field-cooled (ZFC) magnetic susceptibilities were tested by using a superconducting quantum interference device magnetometer (SQUID, Quantum Design MPMS-XL, San Diego, CA, USA) in the temperature range from 2 to 300 K with an applied field of 0.1 T. Faraday rotation angles of the as-grown PSO single crystal along the crystallographic *c*-axis were measured by using the photo-extinction technique [31] at room temperature with the average values of six data for interpretation. A commercial TGG single crystal, supplied by Physcience Opto-electronic Co., Ltd. Beijing, was used as a reference material and has also been evaluated under the same test conditions.

### 3. Results and Discussion

#### 3.1. Structure Refinement

The powder X-ray diffraction pattern of the as-grown PSO single crystal are displayed in Figure 2. The Rietveld refinement unambiguously indicates the existence of the single hexagonal oxyapatite phase with the space group  $P6_3/m$ . The atomic coordinates, unit cell parameters, and refinement results for the as-grown PSO crystal are compiled in Table 1, which indicate that each cation occupies the expected cationic site according to the atomic radius, that is,  $\text{Pr}_{9.33}(\text{SiO}_4)_6\text{O}_2$ . The  $6h$  sites are fully occupied by  $\text{Pr}^{3+}$  and the cation vacancy are all present at the  $4f$  sites with concentration of 6.7%. It is noteworthy that the oxide ions coordinated by Pr (1) ions at the  $4f$  site are all belong to  $(\text{SiO}_4)_4$  group. However, the other oxide ions, at the  $2a$  site, have three-fold coordination with the Pr (2) ions at the  $6h$  site and do not belong to any  $(\text{SiO}_4)_4$  group. Therefore, placing Pr ions on the  $6h$  site can satisfy electrostatic charge balance more efficiently. On the other hand, the  $4f$  site is an ionic site, while the  $6h$  site is a covalent site on the basis of the bond valence sums [32]. So it seems to be that the cation vacancies are preferred at the  $4f$  site. These conclusions from our Rietveld refinement present good agreement with previous ones found in other oxyapatite-type silicates [22–24,28].

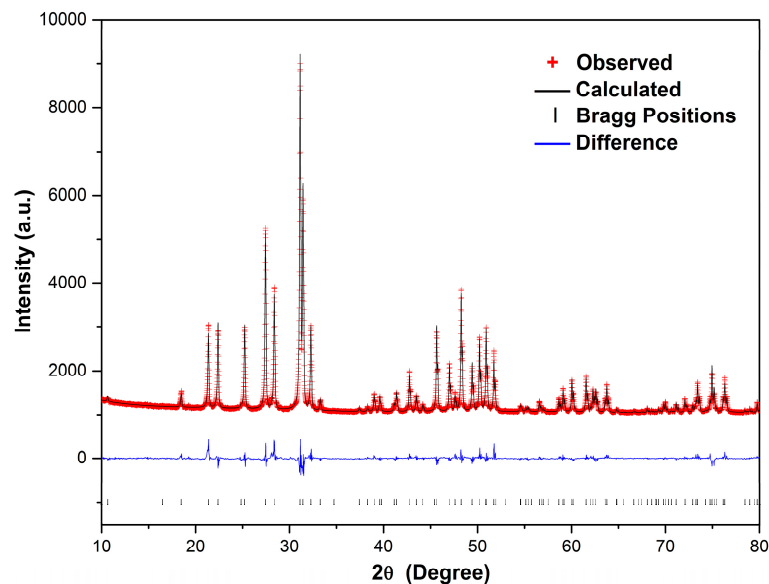


Figure 2. Fitted diffraction profiles of the as-grown PSO single crystal at room temperature.

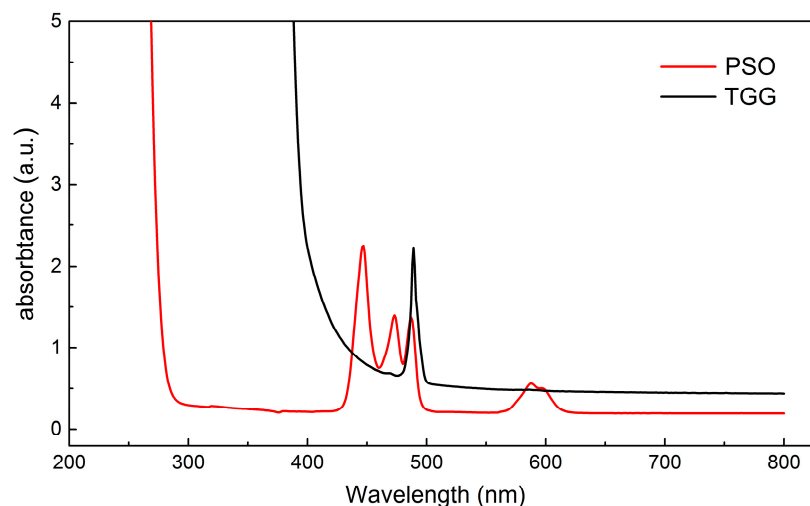
**Table 1.** Refined structural parameters for the as-grown PSO crystal according to  $P6_3/m$  model.

Atom	Site	Site Occ.	$x$	$y$	$z$
Pr1	4f	0.849	1/3	2/3	0.0014
Pr2	6h	1.0	0.0124	0.2400	1/4
Si	6h	1.0	0.3935	0.3702	1/4
O1	6h	1.0	0.3165	0.4693	1/4
O2	6h	1.0	0.5813	0.4818	1/4
O3	12i	1.0	0.3432	0.2570	0.0735
O4	2a	1.0	0	0	1/4

Hexagonal, space group  $P6_3/m$ ;  $a = b = 9.61229(3)$  Å,  $c = 7.06561(2)$  Å,  $R_{wp} = 6.52\%$ .

### 3.2. Optical Transmission

The absorption spectrum of as-grown crystal PSO in the near UV–visible spectral region is given in Figure 3, in contrast to the commercial TGG crystal. As can be seen, the absorbance of TGG continuously increases in the visible spectral region till the beginning of the  ${}^7F_6 \rightarrow {}^5D_3$ ,  ${}^5D_2$   $Tb^{3+}$  absorption bands below 390 nm, which seriously restricts the application of TGG FIs in the UV–visible spectral region. In comparison with TGG, PSO crystal presents a smaller dispersion, a higher transparency in the near UV–visible spectral region, and a shorter cutoff at 265 nm, which enables the use of PSO single crystals also at spectral region where TGG crystals could not be employed. However, a strong absorption band, attributed to the energy level transition  ${}^3H_4 \rightarrow {}^1S_0$  reduces the transparency below 270 nm. In addition, PSO presents two absorption bands in the 430–497 nm and the 570–612 nm wavelength regions, which are mainly related to the energy level transitions  ${}^3H_4 \rightarrow {}^3P_j$  and  ${}^3H_4 \rightarrow {}^1D_2$  of  $Pr^{3+}$  in the oxyapatite structure, and responsible for the typical light green coloration of PSO crystal.

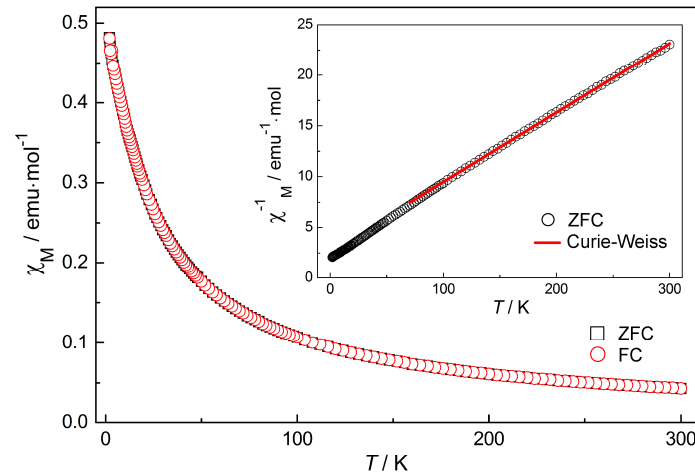


**Figure 3.** Absorption spectra of as-grown PSO crystal in comparison with that of commercial TGG crystal.

### 3.3. Paramagnetic Behavior

The temperature-dependence of the magnetic susceptibility ( $\chi_M$ ) for as-grown PSO crystal measured over the temperature range from 2 to 300 K are exhibited in Figure 4. Paramagnetic behavior was observed in the experimental temperature range, that is, no magnetic interaction was found down to 2.0 K. The reciprocal magnetic susceptibility as a function of the temperature is shown in the inset of Figure 4, and the solid line indicated a Curie–Weiss behavior above 70 K. The deviation from the Curie–Weiss law below 70 K may related to the influence of the crystal field on the magnetic

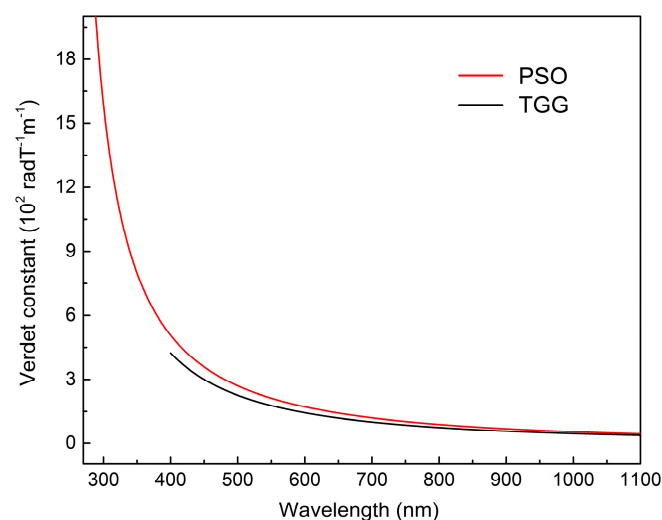
ions  $\text{Pr}^{3+}$  in the PSO compound [33,34]. A fit of the  $\chi_M^{-1} - T$  curve to the Curie–Weiss law from 200 to 300 K gives the effective magnetic moment ( $\mu_{\text{eff}}$ ) of PSO to be  $3.54(3) \mu_B$ , which is in good agreement with the theoretical effective magnetic moment for a free  $\text{Pr}^{3+}$  ion, i.e.,  $\mu_{\text{cal}} = 3.58 \mu_B$  [35].



**Figure 4.** Temperature-dependence of the magnetic susceptibility for as-grown crystal PSO. The inset shows the reciprocal magnetic susceptibility vs. temperature curve. The solid line is the Curie–Weiss fitting.

### 3.4. Magneto-Optical Performance

The magneto-optical response of as-grown PSO single crystal along the crystallographic  $c$ -axis at room temperature is exhibited in Figure 5 as a function of the wavelength. It is seen that, although PSO crystal just has a slightly larger Faraday rotation angle than that of the commercial TGG crystal in the visible spectral region, the Verdet constant increment of PSO in the near UV spectral region is very remarkable:  $V_{\text{PSO}}$  varies from  $511$  to  $1590 \text{ rad T}^{-1} \text{ m}^{-1}$  between  $400$  and  $300 \text{ nm}$ .



**Figure 5.** Verdet constant of as-grown PSO crystal vs. wavelength  $\lambda$ . TGG was tested as reference, and the experimental data were fitted by an equation  $A/(\lambda^2 - \lambda_0^2)$ , with  $\lambda_0$  and  $A$  constants.

On the basis of FR standard theory, which considers a single electronic transition frequency, the Verdet constants ( $V$ ), are proportional to the inverse of the wavelength square ( $\lambda^2$ ) and to the concentration of magnetic rare-earth ions ( $N_{RE}$ ) as follows

$$V = \frac{E}{\lambda^2 - \lambda_0^2} \quad (1)$$

where  $\lambda_0$  corresponds to the wavelength of  $RE^{3+}$  ions  $4f-4f5d$  transition, and the factor  $E$  includes all the proportionality factor, e.g., the Lande splitting factor, the concentration of magnetic ions  $N_{RE}$ , and the transition probability [36]. Fitting the experimental data to Equation (1) yields  $E$  and  $\lambda_0$ , listed in Table 2. As magneto-optic effect of PSO crystal comes mainly from Pr ions, we can assume that the concentration of magnetic ions and Pr ions is approximately the same. The concentration of Pr ions can be obtained according to the refined structural parameters of PSO. As shown in Table 2, the transition wavelengths of  $Pr^{3+}$  in PSO is slightly smaller than that of  $Tb^{3+}$  in TGG at 239 and 260 nm, respectively. Moreover, the rotation efficiency of  $Tb^{3+}$  in TGG clearly overcomes that of  $Pr^{3+}$  in PSO, i.e.,  $(E/N_{Tb})_{TGG} > (E/N_{Pr})_{PSO}$ . Hence, it can be deduced that the larger Verdet constant observed in PSO is merely owing to a higher concentration of the MO active  $RE^{3+}$  ions. Moreover, in comparison with  $PrF_3$ , PSO crystal not only presents a larger transition wavelength, i.e.,  $(\lambda_0)_{PSO} > (\lambda_0)_{PrF_3}$ , but also a higher rotation efficiency, i.e.,  $(E/N_{Pr})_{PSO} \gg (E/N_{Pr})_{PrF_3}$ , which makes it have a larger Verdet constant than that of  $PrF_3$  [2].

**Table 2.** The fitted results of PSO, TGG, and  $PrF_3$  single crystals according to Equation (1), and the respective  $RE^{3+}$  ion concentrations  $N_{RE}$  in the crystals.

Crystal	$\lambda_0$ (nm)	$E$ ( $10^7$ rad nm <sup>2</sup> T <sup>-1</sup> m <sup>-1</sup> )	$N_{RE}$ ( $10^{21}$ ions cm <sup>-3</sup> )
PSO	239	5.2755	16.71
TGG	260	4.2595	12.74
$PrF_3$ [2]	184	4.5965	19.10

#### 4. Conclusions

In summary, the performance of oxyapatite-type PSO crystals for magneto-optical applications in the near UV–visible spectral region was investigated. Crack free PSO single crystals with high optical-quality and large size were successfully fabricated by the Cz-method. The Rietveld refinement indicates that the as-grown PSO crystallizes in  $P6_3/m$ , with oxyapatite structure. As expected, PSO crystal exhibits paramagnetic behavior down to 2 K and presents a higher transparency in the near UV–visible spectral region, as well as a shorter cutoff at 265 nm. In comparison with Pr-based crystal  $PrF_3$  and commercially available TGG, we demonstrate that the as-grown PSO crystal yields a larger FR which rapidly increases towards the cutoff. Presented results therefore indicate that PSO crystals are promising magneto-optical crystal in particular for potential magneto-optical applications in the near UV–visible spectral region.

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**Author Contributions:** Xin Chen and Jianzhong Chen conceived and designed the experiments; Xin Chen performed the experiments, analyzed the data, and wrote the paper; Jianzhong Chen modified the article.

**Conflicts of Interest:** The authors declare no conflict of interest.

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