

## Supporting Information

# Molecular synchronization enhances molecular interactions: An explanatory note of pressure effects

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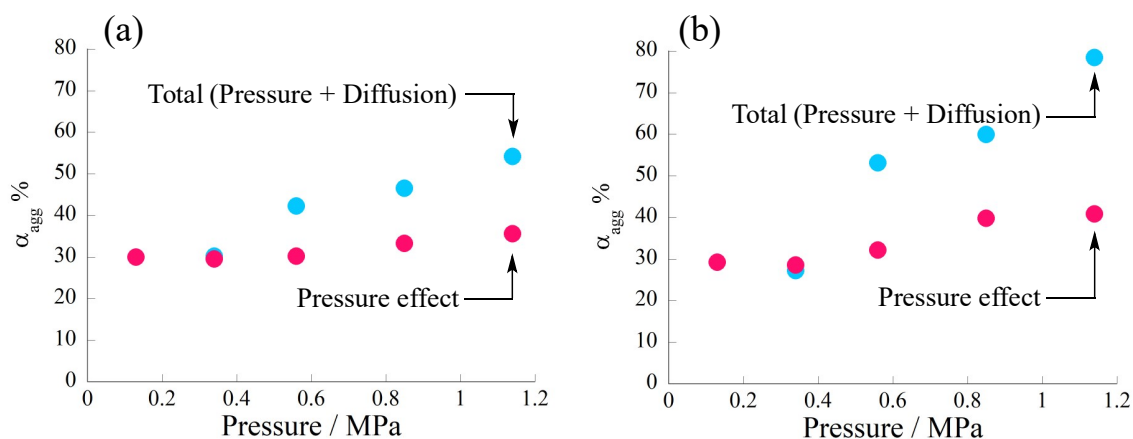
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### 1. General:

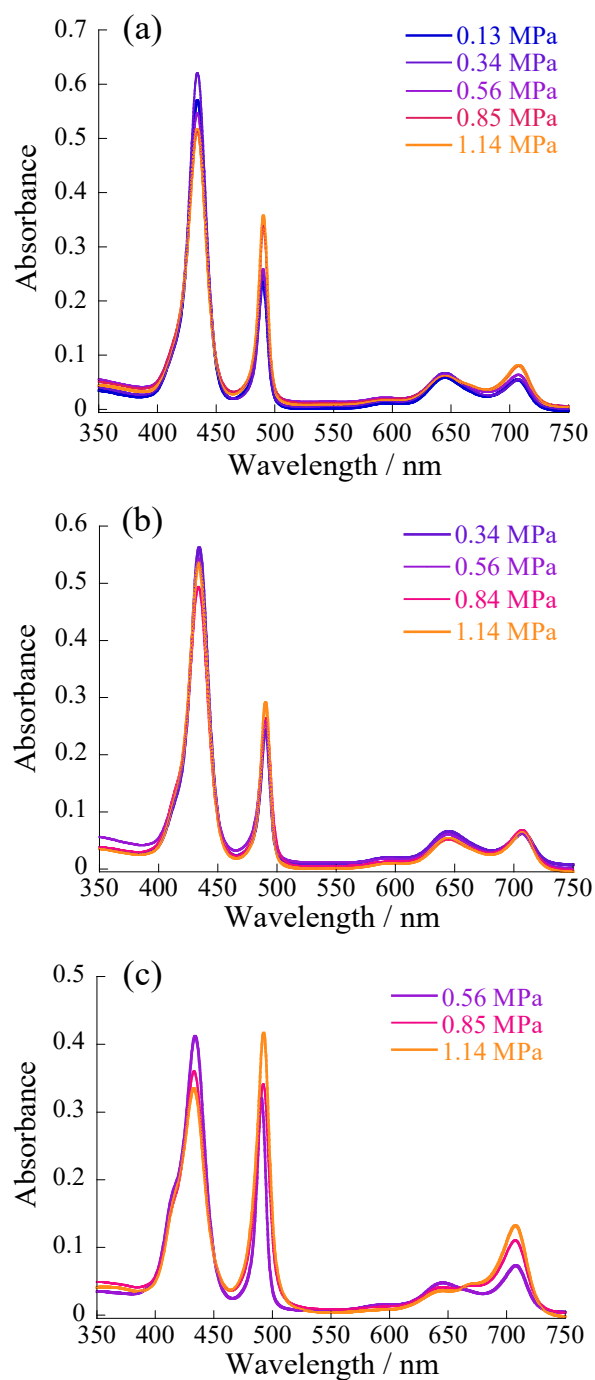
Micro-mixer (a static type reactor made of SUS316) was purchased from YMC (*Kyoto, Japan*) as a ready-made microreactor. The channel depth and width of this reactor were both 200  $\mu\text{m}$ . The mixing volume of the channel was 20  $\mu\text{L}$ .

### 2. Plots of aggregation ratio against pressure:



**Figure S1.** Total aggregation efficiency upon changing the flow rate (including both pressure and diffusion effects; blue dots) and extracted pressure effects under constant proton diffusion effect at 10  $\mu\text{L}/\text{min}$  (red dots); (a) pH 5; (b) pH 3.

### 3. UV-Vis spectral changes at flow rates of 25 and 50 $\mu\text{L}/\text{min}$ (pH 5):



**Figure S2.** UV-Vis spectral changes of TPSS upon changing the pressure inside the channel at pH 5;  $v =$  (a) 10, (b) 25, and (c) 50  $\mu\text{L}/\text{min}$ . 0.1 mm cell, r.t.

#### 4. Summary of aggregation efficiencies at flow rates of 25 and 50 $\mu\text{L}/\text{min}$ :

Pressure (MPa)	0.13	0.29	0.57	0.84	1.14
<i>Flow rate : 25 <math>\mu\text{L}/\text{min}</math></i>					
Solution Width ( $\mu\text{m}$ )	n.d.	Fixed at 25.9 $\mu\text{m}$			
Diffusion Time (ms)	n.d.	Fixed at 14.2 ms			
$\alpha_{\text{agg}}$ (pH = 5)	n.d.	29.7	32.4	34.5	35.2
<i>Flow rate : 50 <math>\mu\text{L}/\text{min}</math></i>					
Solution Width ( $\mu\text{m}$ )	n.d.	n.d.	Fixed at 19.9 $\mu\text{m}$		
Diffusion Time (ms)	n.d.	n.d.	Fixed at 8.4 ms		
$\alpha_{\text{agg}}$ (pH = 5)	n.d.	n.d.	41.3	46.4	52.9

**Figure S3.** Effects of proton diffusion time on aggregation efficiency upon changing the back-pressure inside the channel; summary of extracted pressure effects on aggregation efficiency.

The solution width  $w$  of the central flow containing TPPS molecules was determined from fluorescence microscopy images, using Image-J. The diffusion time of protons was calculated using a two-dimensional model, as follows:

$$t_{\text{diff}} = \frac{w^2}{4D}$$

where  $D$  is the diffusion constant of a proton ( $11.8 \times 10^{-9} \text{ m}^2/\text{sec}$  at 298 K at pH 4.1) and  $w$  is the width of the central flow [1,2].

## 5. Effects of rapid proton diffusion in a micromixer:

Flow rate ( $\mu\text{L min}^{-1}$ )	10	25	50	75	100
Pressure (MPa)	0.01	0.01	0.01	0.01	0.01
$\alpha_{\text{agg}}$ (pH = 5)	41.6	43.7	43.8	41.5	43.9

**Figure S4.** Effects of proton diffusion time on aggregation efficiency in turbulent flow.

## 6. References

1. R. Karnik, F. Gu, P. Basto, C. Cannizzaro, L. Dean, W. Kyei-Manu, R. Langer, O. C. Farokhzad, *Nano Lett.* **2008**, *8*, 2906–2912.
2. J. Choi, N. Hirota, M. Terazima, *J. Phys. Chem. A* **2001**, *105*, 12–18.