



**Figure S1.** Microphotographs of the synthetic Alm-Pyr crystals.

**Table S1.** Details of the structural refinement of synthesis single garnet crystals.

**(1) Alm<sub>100</sub>**

Wavelength (Å)	0.4340
Pressure (GPa)	0.0001
Temperature (K)	298
Composition	Fe <sub>3.13</sub> Al <sub>1.9</sub> Ti <sub>0.1</sub> Si <sub>2.91</sub> O <sub>12</sub>
Symmetry	cubic, $Ia\bar{3}d$
Lattice parameters	a=b=c=11.53025(9)Å, $\alpha=\beta=\gamma=90^\circ$
Volume (Å <sup>3</sup> )	1532.91(4)

$R_{\text{int}}$	0.0536				
All reflections	5981				
Great reflections	133				
Refined parameters	17				
$R_1$	0.0197				
$wR_2$	0.0558				
S	1.1704				
Atom	Occupancy	x	y	z	$U_{\text{iso}}$
Fe	1	0.875	0.5	0.25	0.0062(3)
Si	1	0.125	0.5	0.25	0.0021(4)
Al	1	0	0.5	0.5	0.0017(4)
O	1	0.03415(8)	0.54921(8)	0.34725(8)	0.0035(4)

[AlO<sub>6</sub>] octahedron

Average bond length = 1.8918(10) Å

Octahedron volume = 9.0206 Å<sup>3</sup>

Effective coordination number = 6

[FeO<sub>8</sub>] dodecahedron

Average bond length = 2.2971(10) Å

bond I length = 2.2241(9) Å

bond II length = 2.3701(10) Å

Polyhedral volume = 20.8362 Å<sup>3</sup>

Effective coordination number = 7.6815

[SiO<sub>4</sub>] tetrahedron

Average bond length = 1.6360(10) Å

Polyhedral volume = 2.2058 Å<sup>3</sup>

Effective coordination number = 4

## (2) Pyr<sub>9</sub>Alm<sub>91</sub>

Wavelength (Å)	0.4340				
Pressure (GPa)	0.0001				
Temperature (K)	298				
Composition	(Fe <sub>2.88</sub> Mg <sub>0.27</sub> ) Al <sub>1.89</sub> Ti <sub>0.05</sub> Si <sub>2.94</sub> O <sub>12</sub>				
Symmetry	cubic, $Ia\bar{3}d$				
Lattice parameters	a=b=c=11.52701(11)Å, $\alpha=\beta=\gamma=90^\circ$				
Volume (Å <sup>3</sup> )	1531.62(4)				
$R_{\text{int}}$	0.0369				
All reflections	5985				
Great reflections	123				
Refined parameters	19				
$R_1$	0.0181				
$wR_2$	0.0511				
S	1.0883				
Atom	Occupancy	x	y	z	$U_{\text{iso}}$
Mg	0.11(2)	0.875	0.5	0.25	0.0063(4)
Fe	0.899(11)	0.875	0.5	0.25	0.0063(4)
Si	1	0.125	0.5	0.25	0.0023(5)
Al	1	0	0.5	0.5	0.0019(5)
O	1	0.03402(8)	0.54910(9)	0.34724(8)	0.0048(5)

[AlO<sub>6</sub>] octahedron

Average bond length = 1.8907(11) Å

Octahedron volume = 9.0044 Å<sup>3</sup>

Effective coordination number = 6

[FeO<sub>8</sub>] / [MgO<sub>8</sub>] dodecahedron

Average bond length = 2.2962(10) Å

bond I length = 2.2218(10) Å  
 bond II length = 2.3704(10) Å  
 Polyhedral volume = 20.8197 Å<sup>3</sup>  
 Effective coordination number = 7.6691  
 [SiO<sub>4</sub>] tetrahedron  
 Average bond length = 1.6360(10) Å  
 Polyhedral volume = 2.2047 Å<sup>3</sup>  
 Effective coordination number = 4

### (3) Pyr<sub>14</sub>Alm<sub>86</sub>

Wavelength (Å)	0.4340				
Pressure (GPa)	0.0001				
Temperature (K)	298				
Composition	(Fe <sub>2.71</sub> Mg <sub>0.43</sub> ) Al <sub>1.92</sub> Ti <sub>0.07</sub> Si <sub>2.92</sub> O <sub>12</sub>				
Symmetry	cubic, <i>Ia</i> $\bar{3}d$				
Lattice parameters	a=b=c=11.5241(3)Å, $\alpha=\beta=\gamma=90^\circ$				
Volume (Å <sup>3</sup> )	1530.46(12)				
<i>R</i> <sub>int</sub>	0.0632				
All reflections	5656				
Great reflections	108				
Refined parameters	19				
<i>R</i> <sub>1</sub>	0.0225				
<i>wR</i> <sub>2</sub>	0.0568				
S	1.1715				
Atom	Occupancy	x	y	z	<i>U</i> <sub>iso</sub>
Mg	0.16(2)	0.875	0.5	0.25	0.0057(5)
Fe	0.842(13)	0.875	0.5	0.25	0.0057(5)
Si	1	0.125	0.5	0.25	0.0020(7)
Al	1	0	0.5	0.5	0.0011(7)
O	1	0.0339(11)	0.54928(10)	0.34724(10)	0.0019(6)

[AlO<sub>6</sub>] octahedron  
 Average bond length = 1.8931(13) Å  
 Octahedron volume = 9.0381 Å<sup>3</sup>  
 Effective coordination number = 6  
 [FeO<sub>8</sub>] / [MgO<sub>8</sub>] dodecahedron  
 Average bond length = 2.2936(12) Å  
 bond I length = 2.2193(12) Å  
 bond II length = 2.3680(12) Å  
 Polyhedral volume = 20.7538 Å<sup>3</sup>  
 Effective coordination number = 7.6679  
 [SiO<sub>4</sub>] tetrahedron  
 Average bond length = 1.6353(13) Å  
 Polyhedral volume = 2.2006 Å<sup>3</sup>  
 Effective coordination number = 4

### (4) Pyr<sub>23</sub>Alm<sub>76</sub>

Wavelength (Å)	0.4340				
Pressure (GPa)	0.0001				
Temperature (K)	298				
Composition	(Fe <sub>2.55</sub> Mg <sub>0.78</sub> ) Al <sub>1.87</sub> Ca <sub>0.01</sub> Ti <sub>0.05</sub> Si <sub>2.88</sub> O <sub>12</sub>				
Symmetry	cubic, <i>Ia</i> $\bar{3}d$				
Lattice parameters	a=b=c=11.52759(16)Å, $\alpha=\beta=\gamma=90^\circ$				
Volume (Å <sup>3</sup> )	1531.85(6)				
<i>R</i> <sub>int</sub>	0.0382				

All reflections	5797
Great reflections	116
Refined parameters	19
$R_1$	0.0171
$wR_2$	0.0484
S	1.1512

Atom	Occupancy	x	y	z	$U_{iso}$
Mg	0.28(2)	0.875	0.5	0.25	0.0110(4)
Fe	0.729(10)	0.875	0.5	0.25	0.0110(4)
Si	1	0.125	0.5	0.25	0.0075(4)
Al	1	0	0.5	0.5	0.0043(5)
O	1	0.03391(8)	0.54951(8)	0.34692(8)	0.0100(4)

[AlO<sub>6</sub>] octahedron

Average bond length = 1.8954(10) Å

Octahedron volume = 9.0697 Å<sup>3</sup>

Effective coordination number = 6

[FeO<sub>8</sub>] / [MgO<sub>8</sub>] dodecahedron

Average bond length = 2.2933(9) Å

bond I length = 2.2203(9) Å

bond II length = 2.3662(9) Å

Polyhedral volume = 20.7357 Å<sup>3</sup>

Effective coordination number = 7.6806

[SiO<sub>4</sub>] tetrahedron

Average bond length = 1.6360(10) Å

Polyhedral volume = 2.2037 Å<sup>3</sup>

Effective coordination number = 4

### (5) Pyr<sub>31</sub>Alm<sub>68</sub>

Wavelength (Å)	0.4340
Pressure (GPa)	0.0001
Temperature (K)	298
Composition	(Fe <sub>2.16</sub> Mg <sub>0.99</sub> ) Al <sub>1.9</sub> Ca <sub>0.01</sub> Ti <sub>0.08</sub> Si <sub>2.92</sub> O <sub>12</sub>
Symmetry	cubic, $Im\bar{3}d$
Lattice parameters	a=b=c=11.51055(6)Å, $\alpha=\beta=\gamma=90^\circ$
Volume (Å <sup>3</sup> )	1525.06(2)
$R_{int}$	0.0635
All reflections	5851
Great reflections	106
Refined parameters	19
$R_1$	0.0233
$wR_2$	0.0607
S	1.1479

Atom	Occupancy	x	y	z	$U_{iso}$
Mg	0.41(2)	0.875	0.5	0.25	0.0080(5)
Fe	0.600(10)	0.875	0.5	0.25	0.0080(5)
Si	1	0.125	0.5	0.25	0.0046(5)
Al	1	0	0.5	0.5	0.0033(6)
O	1	0.0337(8)	0.54967(8)	0.34694(8)	0.0054(5)

[AlO<sub>6</sub>] octahedron

Average bond length = 1.8925(10) Å

Octahedron volume = 9.0269 Å<sup>3</sup>

Effective coordination number = 6

[FeO<sub>8</sub>] / [MgO<sub>8</sub>] dodecahedron

Average bond length = 2.2882(9) Å

bond I length = 2.2159(9) Å  
 bond II length = 2.3606(9) Å  
 Polyhedral volume = 20.5985 Å<sup>3</sup>  
 Effective coordination number = 7.685  
 [SiO<sub>4</sub>] tetrahedron  
 Average bond length = 1.6357(10) Å  
 Polyhedral volume = 2.2019 Å<sup>3</sup>  
 Effective coordination number = 4

### (6) Pyr<sub>48</sub>Alm<sub>52</sub>

Wavelength (Å)	0.4340				
Pressure (GPa)	0.0001				
Temperature (K)	298				
Composition	(Fe <sub>1.52</sub> Mg <sub>1.41</sub> ) Al <sub>1.99</sub> Ca <sub>0.01</sub> Si <sub>3.04</sub> O <sub>12</sub>				
Symmetry	cubic, <i>Ia</i> $\bar{3}d$				
Lattice parameters	a=b=c=11.50955(6)Å, $\alpha=\beta=\gamma=90^\circ$				
Volume (Å <sup>3</sup> )	1524.67(2)				
<i>R</i> <sub>int</sub>	0.0391				
All reflections	5808				
Great reflections	115				
Refined parameters	19				
<i>R</i> <sub>1</sub>	0.0175				
<i>wR</i> <sub>2</sub>	0.0505				
S	1.0767				
Atom	Occupancy	x	y	z	<i>U</i> <sub>iso</sub>
Mg	0.451(18)	0.875	0.5	0.25	0.0076(4)
Fe	0.551(9)	0.875	0.5	0.25	0.0076(4)
Si	1	0.125	0.5	0.25	0.0037(5)
Al	1	0	0.5	0.5	0.0019(5)
O	1	0.03364(6)	0.54969(6)	0.34702(6)	0.0057(4)

[AlO<sub>6</sub>] octahedron  
 Average bond length = 1.8913(7) Å  
 Octahedron volume = 9.0097 Å<sup>3</sup>  
 Effective coordination number = 6  
 [FeO<sub>8</sub>] / [MgO<sub>8</sub>] dodecahedron  
 Average bond length = 2.2876(7) Å  
 bond I length = 2.2153(6) Å  
 bond II length = 2.3598(7) Å  
 Polyhedral volume = 20.5820 Å<sup>3</sup>  
 Effective coordination number = 7.6858  
 [SiO<sub>4</sub>] tetrahedron  
 Average bond length = 1.6370(7) Å  
 Polyhedral volume = 2.2068 Å<sup>3</sup>  
 Effective coordination number = 4

### (7) Pyr<sub>57</sub>Alm<sub>43</sub>

Wavelength (Å)	0.4340				
Pressure (GPa)	0.0001				
Temperature (K)	298				
Composition	(Fe <sub>1.36</sub> Mg <sub>1.82</sub> ) Al <sub>1.93</sub> Ca <sub>0.01</sub> Ti <sub>0.01</sub> Si <sub>2.95</sub> O <sub>12</sub>				
Symmetry	cubic, <i>Ia</i> $\bar{3}d$				
Lattice parameters	a=b=c=11.5013(4), $\alpha=\beta=\gamma=90^\circ$				
Volume (Å <sup>3</sup> )	1521.39(16)				
<i>R</i> <sub>int</sub>	0.0336				

All reflections	5802
Great reflections	114
Refined parameters	19
$R_1$	0.0204
$wR_2$	0.0544
S	1.0958

Atom	Occupancy	x	y	z	$U_{iso}$
Mg	0.52(2)	0.875	0.5	0.25	0.0095(5)
Fe	0.484(12)	0.875	0.5	0.25	0.0095(5)
Si	1	0.125	0.5	0.25	0.0058(5)
Al	1	0	0.5	0.5	0.0042(5)
O	1	0.03365(8)	0.54984(8)	0.34696(8)	0.0082(4)

[AlO<sub>6</sub>] octahedron

Average bond length = 1.8912(10) Å

Octahedron volume = 9.0068 Å<sup>3</sup>

Effective coordination number = 6

[FeO<sub>8</sub>] / [MgO<sub>8</sub>] dodecahedron

Average bond length = 2.2853(9) Å

bond I length = 2.2140(9) Å

bond II length = 2.3599(9) Å

Polyhedral volume = 20.514 Å<sup>3</sup>

Effective coordination number = 7.6937

[SiO<sub>4</sub>] tetrahedron

Average bond length = 1.6359(10) Å

Polyhedral volume = 2.2024 Å<sup>3</sup>

Effective coordination number = 4

### (8) **Pyr<sub>67</sub>Alm<sub>32</sub>**

Wavelength (Å)	0.4340
Pressure (GPa)	0.0001
Temperature (K)	298
Composition	(Fe <sub>1.03</sub> Mg <sub>2.13</sub> ) Al <sub>1.92</sub> Ca <sub>0.01</sub> Ti <sub>0.03</sub> Si <sub>2.95</sub> O <sub>12</sub>
Symmetry	cubic, $Im\bar{3}d$
Lattice parameters	a=b=c=11.4887(3)Å, $\alpha=\beta=\gamma=90^\circ$
Volume (Å <sup>3</sup> )	1516.40(12)
$R_{int}$	0.0411
All reflections	5824
Great reflections	120
Refined parameters	19
$R_1$	0.0213
$wR_2$	0.0523
S	1.1687

Atom	Occupancy	x	y	z	$U_{iso}$
Mg	0.666(19)	0.875	0.5	0.25	0.0075(4)
Fe	0.334(9)	0.875	0.5	0.25	0.0075(4)
Si	1	0.125	0.5	0.25	0.0033(4)
Al	1	0	0.5	0.5	0.0025(4)
O	1	0.03342(7)	0.54989(7)	0.34683(6)	0.0050(4)

[AlO<sub>6</sub>] octahedron

Average bond length = 1.8901(8)Å

Octahedron volume = 8.9903 Å<sup>3</sup>

Effective coordination number = 6

[FeO<sub>8</sub>] / [MgO<sub>8</sub>] dodecahedron

Average bond length = 2.2810(8) Å

bond I length = 2.2088(8) Å  
 bond II length = 2.3532(8) Å  
 Polyhedral volume = 20.4075 Å<sup>3</sup>  
 Effective coordination number = 7.684  
 [SiO<sub>4</sub>] tetrahedron  
 Average bond length = 1.6349(8) Å  
 Polyhedral volume = 2.1970 Å<sup>3</sup>  
 Effective coordination number = 4

### (9) Pyr<sub>78</sub>Alm<sub>22</sub>

Wavelength (Å)	0.4340
Pressure (GPa)	0.0001
Temperature (K)	298
Composition	(Fe <sub>0.71</sub> Mg <sub>2.49</sub> ) Al <sub>1.85</sub> Ca <sub>0.01</sub> Si <sub>3.01</sub> O <sub>12</sub>
Symmetry	cubic, <i>Ia</i> $\bar{3}d$
Lattice parameters	a=b=c=11.48023(8)Å, $\alpha=\beta=\gamma=90^\circ$
Volume (Å <sup>3</sup> )	1513.04(3)
<i>R</i> <sub>int</sub>	0.0368
All reflections	5900
Great reflections	116
Refined parameters	19
<i>R</i> <sub>1</sub>	0.0181
<i>wR</i> <sub>2</sub>	0.0504
S	1.1291

Atom	Occupancy	x	y	z	<i>U</i> <sub>iso</sub>
Mg	0.753(19)	0.875	0.5	0.25	0.0087(4)
Fe	0.244(9)	0.875	0.5	0.25	0.0087(4)
Si	1	0.125	0.5	0.25	0.0038(4)
Al	1	0	0.5	0.5	0.0036(4)
O	1	0.03331(6)	0.55001(6)	0.34670(6)	0.0058(4)

[AlO<sub>6</sub>] octahedron  
 Average bond length = 1.8903(7) Å  
 Octahedron volume = 8.9914 Å<sup>3</sup>  
 Effective coordination number = 6  
 [FeO<sub>8</sub>] / [MgO<sub>8</sub>] dodecahedron  
 Average bond length = 2.2779(7) Å  
 bond I length = 2.2057(7) Å  
 bond II length = 2.3501(7) Å  
 Polyhedral volume = 20.3269 Å<sup>3</sup>  
 Effective coordination number = 7.6832  
 [SiO<sub>4</sub>] tetrahedron  
 Average bond length = 1.6340(7) Å  
 Polyhedral volume = 2.1923 Å<sup>3</sup>  
 Effective coordination number = 4

### (10) Pyr<sub>100</sub>

Wavelength (Å)	0.4340
Pressure (GPa)	0.0001
Temperature (K)	298
Composition	Mg <sub>3.12</sub> Al <sub>1.96</sub> Ti <sub>0.01</sub> Si <sub>2.96</sub> O <sub>12</sub>
Symmetry	cubic, <i>Ia</i> $\bar{3}d$
Lattice parameters	a=b=c= 11.45522(10)Å, $\alpha=\beta=\gamma=90^\circ$
Volume (Å <sup>3</sup> )	1513.18(4)
<i>R</i> <sub>int</sub>	0.0579

All reflections	6012
Great reflections	134
Refined parameters	17
$R_1$	0.0228
$wR_2$	0.0624
S	1.0938

Atom	Occupancy	x	y	z	$U_{iso}$
Mg	1	0.875	0.5	0.25	0.0086(4)
Si	1	0.125	0.5	0.25	0.0027(4)
Al	1	0	0.5	0.5	0.0032(4)
O	1	0.03298(7)	0.55039(6)	0.34677(6)	0.0045(4)

[AlO<sub>6</sub>] octahedron

Average bond length = 1.8860(8) Å

Octahedron volume = 8.9264 Å<sup>3</sup>

Effective coordination number =

[MgO<sub>8</sub>] dodecahedron

Average bond length = 2.2779(8) Å

Average bond I length = 2.1993(8) Å

Average bond II length = 2.3400(7) Å

Polyhedral volume = 20.1007 Å<sup>3</sup>

Effective coordination number = 7.6980

[SiO<sub>4</sub>] tetrahedron

Average bond length = 1.6350(8) Å

Polyhedral volume = 2.1954 Å<sup>3</sup>

Effective coordination number = 4