

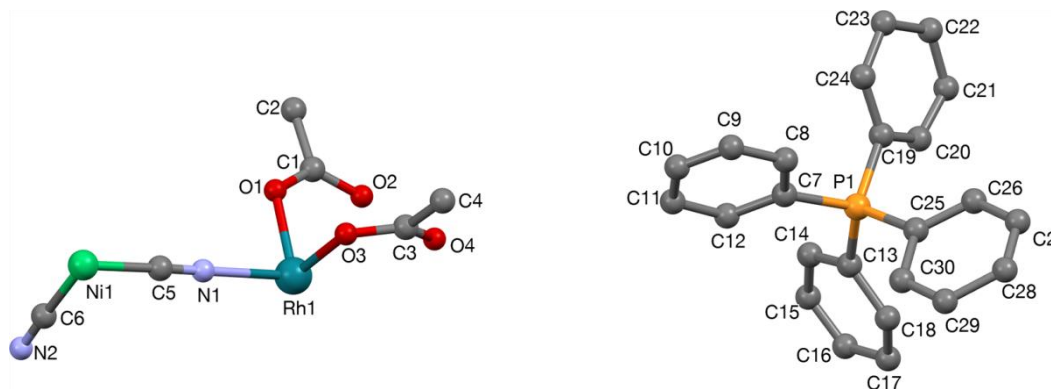
Supplementary Information
S1. Crystal Structure Solution for Compound 1


Figure S1. Asymmetric unit of compound 1. Hydrogen atoms have been omitted for clarity.

Table S1. Sample and crystal data for 1.

CCDC Number	1954793
Chemical formula	C ₆₀ H ₅₂ N ₄ NiO ₈ P ₂ Rh ₂
Formula weight	1283.52 g/mol
Temperature	293(2) K
Wavelength	0.71073 Å
Crystal system	triclinic
Space group	<i>P</i> -1
Unit cell dimensions	<i>a</i> = 12.669(2) Å α = 84.405(5)° <i>b</i> = 12.8404(8) Å β = 67.765(7)° <i>c</i> = 14.640(1) Å γ = 70.795(7)°
Volume	2080.7(3) Å ³
Z	1
Density (calculated)	1.024 g/cm ³
Absorption coefficient	0.694 mm ⁻¹
F(000)	652

Table S2. Data collection and structure refinement for 1.

Theta range for data collection	3.27 to 25.00°
Index ranges	-15 ≤ <i>h</i> ≤ 15, -14 ≤ <i>k</i> ≤ 15, -17 ≤ <i>l</i> ≤ 17
Reflections collected	15,447
Independent reflections	7322 [R(int) = 0.0550]
Refinement method	Full-matrix least-squares on F ²
Refinement program	SHELXL-2014/7 (Sheldrick, 2014)
Function minimized	$\Sigma w(F_o^2 - F_c^2)^2$
Data/restraints/parameters	7322/0/349
Goodness-of-fit on F²	1.017
Final R indices	4505 data; I > 2σ(I) R ₁ = 0.0740, wR ₂ = 0.2276

	all data	$R_1 = 0.1104$, $wR_2 = 0.2500$
Weighting scheme	$w=1/[\sigma^2(F_o^2)+(0.1420P)^2]$ where $P=(F_o^2+2F_c^2)/3$	
Largest diff. peak and hole	0.815 and $-0.752 \text{ e}\text{\AA}^{-3}$	
R.M.S. deviation from mean	1.125 $\text{e}\text{\AA}^{-3}$	

S2. Crystal Structure Solution for Compound 2

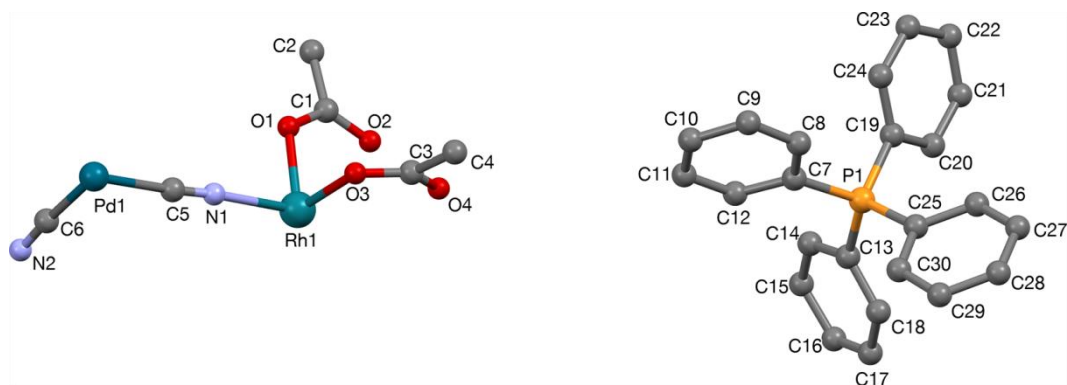


Figure S2. Asymmetric unit of compound 2. Hydrogen atoms have been omitted for clarity.

Table S3. Sample and crystal data for 2.

CCDC Number	1954794
Chemical formula	$\text{C}_{60}\text{H}_{52}\text{N}_4\text{O}_8\text{P}_2\text{PdRh}_2$
Formula weight	1331.21 g/mol
Temperature	293(2) K
Wavelength	0.71073 \AA
Crystal system	triclinic
Space group	$P-1$
Unit cell dimensions	$a = 12.636(1) \text{ \AA}$ $\alpha = 85.137(9)^\circ$ $b = 13.095(2) \text{ \AA}$ $\beta = 68.479(9)^\circ$ $c = 14.524(2) \text{ \AA}$ $\gamma = 70.436(9)^\circ$
Volume	2104.7(4) \AA^3
Z	1
Density (calculated)	1.050 g/cm^3
Absorption coefficient	0.676 mm^{-1}
F(000)	670

Table S4. Data collection and structure refinement for 2.

Theta range for data collection	3.21 to 26.00 $^\circ$
Index ranges	$-15 \leq h \leq 15$, $-15 \leq k \leq 16$, $-17 \leq l \leq 17$
Reflections collected	17,912
Independent reflections	8262 [R(int) = 0.0526]
Refinement method	Full-matrix least-squares on F^2
Refinement program	SHELXL-2014/7 (Sheldrick, 2014)
Function minimized	$\sum w(F_o^2 - F_c^2)^2$
Data/restraints/parameters	8262/0/343

Goodness-of-fit on F²	1.005
Final R indices	4648 data; I>2σ(I) R ₁ = 0.0662, wR ₂ = 0.1699 all data R ₁ = 0.1057, wR ₂ = 0.1854
Weighting scheme	w=1/[σ ² (F _o ²)+(0.0846P) ²]where P=(F _o ² +2F _c ²)/3
Largest diff. peak and hole	1.279 and −0.772 eÅ ^{−3}
R.M.S. deviation from mean	0.117 eÅ ^{−3}

S3. Crystal Structure Solution for Compound 3

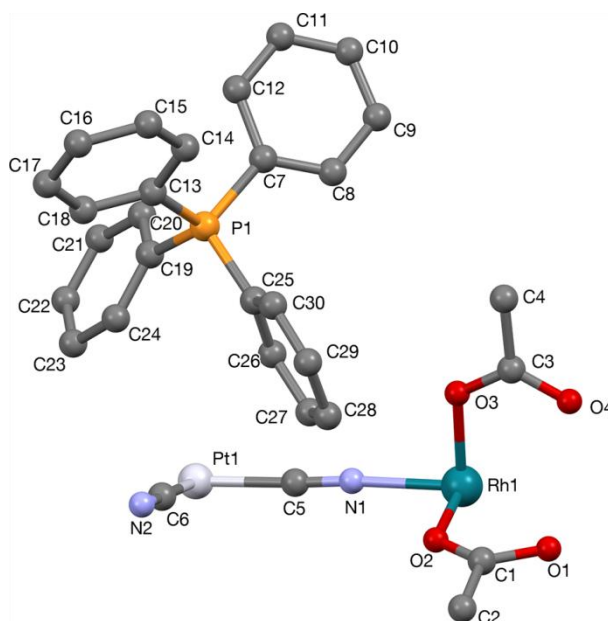


Figure S3. Asymmetric unit of compound 3. Hydrogen atoms have been omitted for clarity.

Table S5. Sample and crystal data for 3.

CCDC Number	1954794
Chemical formula	C ₆₀ H ₅₂ N ₄ O ₈ P ₂ PtRh ₂
Formula weight	1419.90 g/mol
Temperature	293(2) K
Wavelength	0.71073 Å
Crystal system	monoclinic
Space group	<i>P2₁/n</i>
Unit cell dimensions	<i>a</i> = 13.0959(4) Å α = 90° <i>b</i> = 11.7615(4) Å β = 101.613(4)° <i>c</i> = 25.1321(11) Å γ = 90°
Volume	3791.8(2) Å ³
Z	2
Density (calculated)	1.244 g/cm ³
Absorption coefficient	2.356 mm ^{−1}
F(000)	1404

Table S6. Data collection and structure refinement for 3.

Theta range for data collection	3.27 to 26.00°
Index ranges	-15<=h<=16, -13<=k<=14, -22<=l<=30
Reflections collected	20,720
Independent reflections	7438 [R(int) = 0.0435]
Refinement method	Full-matrix least-squares on F ²
Refinement program	SHELXL-2014/7 (Sheldrick, 2014)
Function minimized	$\Sigma w(F_o^2 - F_c^2)^2$
Data/restraints/parameters	7438/0/351
Goodness-of-fit on F ²	1.008
Final R indices	4025 data; I>2 σ (I) R ₁ = 0.0395, wR ₂ = 0.1046 all data R ₁ = 0.0788, wR ₂ = 0.1179
Weighting scheme	w=1/[$\sigma^2(F_o^2)+(0.0430P)^2$] where P=(F _o ² +2F _c ²)/3
Largest diff. peak and hole	0.616 and -0.565 eÅ ⁻³
R.M.S. deviation from mean	0.110 eÅ ⁻³

S4. Coordination Environment Parameters

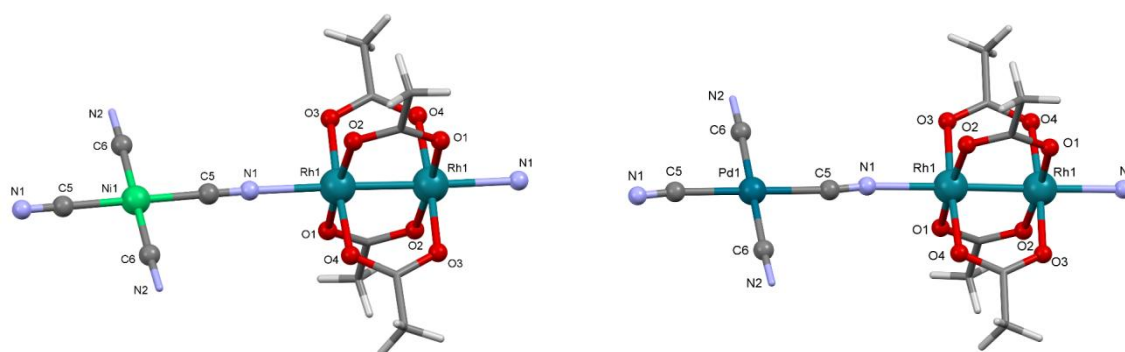


Figure S4. Coordination environments of metal atoms in 1 (left) and 2 (right).

Table S7. Coordination environment distances (Å) in 1 and 2.

	1 (M = Ni)	2 (M = Pd)
Rh1-O1	2.035(5)	2.020(4)
Rh1-O2 ⁱ	2.024(5)	2.040(4)
Rh1-O3	2.019(5)	2.031(4)
Rh1-O4 ⁱ	2.041(5)	2.052(4)
Rh1-N1	2.275(6)	2.246(5)
Rh1-Rh1 ⁱ	2.394(1)	2.3968(9)
M1-C5	1.888(8)	2.080(6)
M1-C6	1.850(8)	1.960(8)

(i) = -x + 2, -y + 1, -z

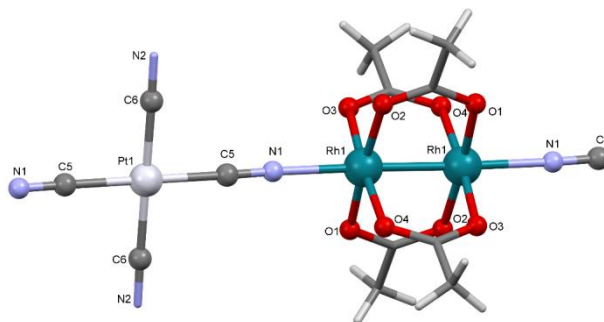


Figure S5. Coordination environments of metal atoms in **3**.

Table S8. Coordination environment distances (Å) in **3**.

3	
Rh1-O1 ⁱ	2.026(4)
Rh1-O2	2.029(4)
Rh1-O3	2.042(4)
Rh1-O4 ⁱ	2.035(4)
Rh1-N1	2.209(5)
Rh1-Rh1 ⁱ	2.4007(8)
Pt1-C5	1.998(6)
Pt2-C6	1.979(8)
(i) $-x+2, -y+1, -z+1$	

Table S9. Coordination environment angles (°) in **1** and **2**.

	1 (M = Ni)	2 (M = Pd)
O1-Rh1-O2 ⁱ	175.5(2)	175.26(2)
O1-Rh1-O3	90.7(2)	90.62(2)
O1-Rh1-O4 ⁱ	88.5(2)	88.8(2)
O1-Rh1-N1	91.6(2)	92.0(2)
O1-Rh1-Rh1 ⁱ	87.5(1)	87.3(1)
O2 ⁱ -Rh1-O3	88.2(2)	88.7(2)
O2 ⁱ -Rh1-O4	92.3(2)	91.6(2)
O2 ⁱ -Rh1-N1	92.7(2)	92.7(2)
O2 ⁱ -Rh1-Rh1	88.2(1)	88.0(1)
O3-Rh1-O4 ⁱ	175.9 (2)	175.7(2)
O3-Rh1-N1	93.4(2)	93.1(2)
O3-Rh1-Rh1 ⁱ	88.0(1)	87.5(1)
O4 ⁱ -Rh1-N1	90.7(2)	91.1(2)
O4 ⁱ -Rh1-Rh1 ⁱ	88.0(1)	88.2(1)
N1-Rh1-Rh1 ⁱ	178.4(2)	179.1(1)
C5-M1-C5	180	180
C5-M1-C6	91.2(3)	91.1(2)
C6-M1-C5 ⁱⁱ	88.8(3)	88.9(2)
C6-M1-C6 ⁱⁱ	180	180

(i) $-x+2, -y+1, -z$; (ii) $-x+2, -y+2, -z$

Table S10. Coordination environment angles ($^{\circ}$) in **3**.

3	
O1 ⁱ -Rh1-O2	175.6(2)
O1 ⁱ -Rh1-O3	88.5(2)
O1 ⁱ -Rh1-O4 ⁱ	91.4(2)
O1 ⁱ -Rh1-N1	91.4(2)
O1 ⁱ -Rh1-Rh1 ⁱ	88.2(1)
O2-Rh1-O3	90.6(2)
O2-Rh1-O4 ⁱ	89.2(2)
O2-Rh1-N1	93.1(2)
O2-Rh1-Rh1 ⁱ	87.4(1)
O3-Rh1-O4 ⁱ	175.4 (2)
O3-Rh1-N1	94.5(2)
O3-Rh1-Rh1 ⁱ	87.3(1)
O4-Rh1-N1	90.1(2)
O4 ⁱ -Rh1-Rh1 ⁱ	88.2(1)
N1-Rh1-Rh1 ⁱ	178.2(1)
C5-Pt1-C5 ⁱⁱ	180
C5-Pt1-C6	90.1(3)
C6-Pt1-C5 ⁱⁱ	89.9(3)
C6-Pt1-C6 ⁱⁱ	180

(i) $-x + 2, -y + 1, -z + 1$; (ii) $-x + 1, -y + 1, -z + 1$

S5. Supramolecular Interactions

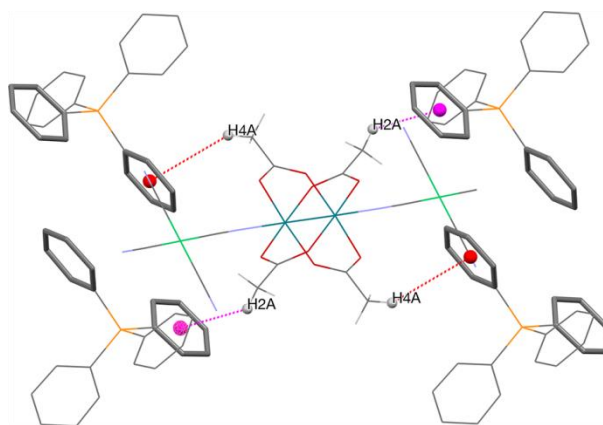


Figure S6. C-H... π interactions in **1** and **2**; the rings involved in these interactions are represented with thicker lines, the C7- C17 ring centroid is depicted in magenta and C13-C18 in red.

Table S11. Supramolecular interactions in **1** and **2**.

	1 (Å)	2 (Å)
C2-H2A...centroid C13-C18	3.280	3.208
C4-H4A...centroid C7-C12	3.310	3.385

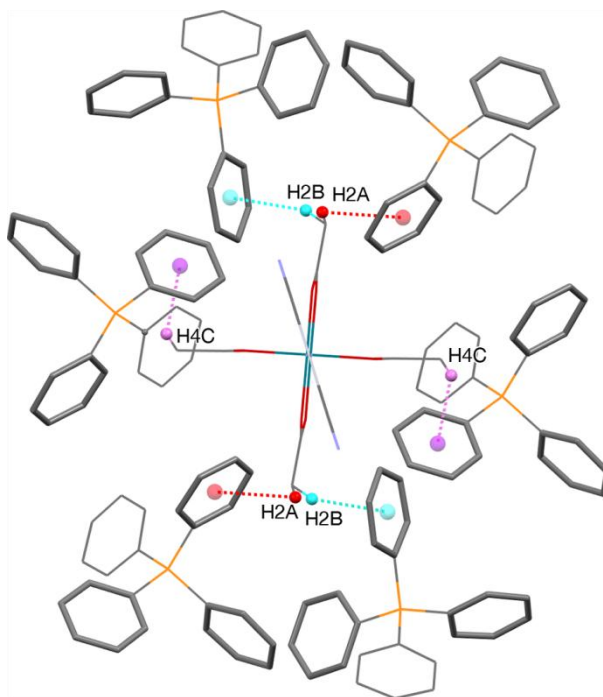


Figure S7. C-H... π interactions in **3**; the rings involved in these interactions are represented with thicker lines, the C7- C17 ring centroid is depicted in red, C13-C18 in cyan, and C25-C30 in pink.

Table S12. Supramolecular interactions in **3**.

	3 (Å)
C2-H2A...centroid C7-C12	3.836
C2-H2B...centroid C13-C18	2.796
C4-H4C...centroid C25-C30	3.482



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