

Advantageous microwave-assisted Suzuki polycondensation for the synthesis of aniline-fluorene alternate copolymers as molecular model with solvent sensing properties

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Supplementary Data, Summary:

- Pag. 2-7: NMR spectra of PAF, PAFBr and PAFAm.
- Pag. 8-11: GPC Chromatograms of batches PAF.
- Pag. 12: FTIR and TGA
- Pag. 13-15: Proposed mechanism and details.
- Pag. 16: Modes of μ W-assisted polymerization and electronic spectra of PAFBr and PAFAm.
- Pag. 17-18: Tables of μ W -assisted of PAF.

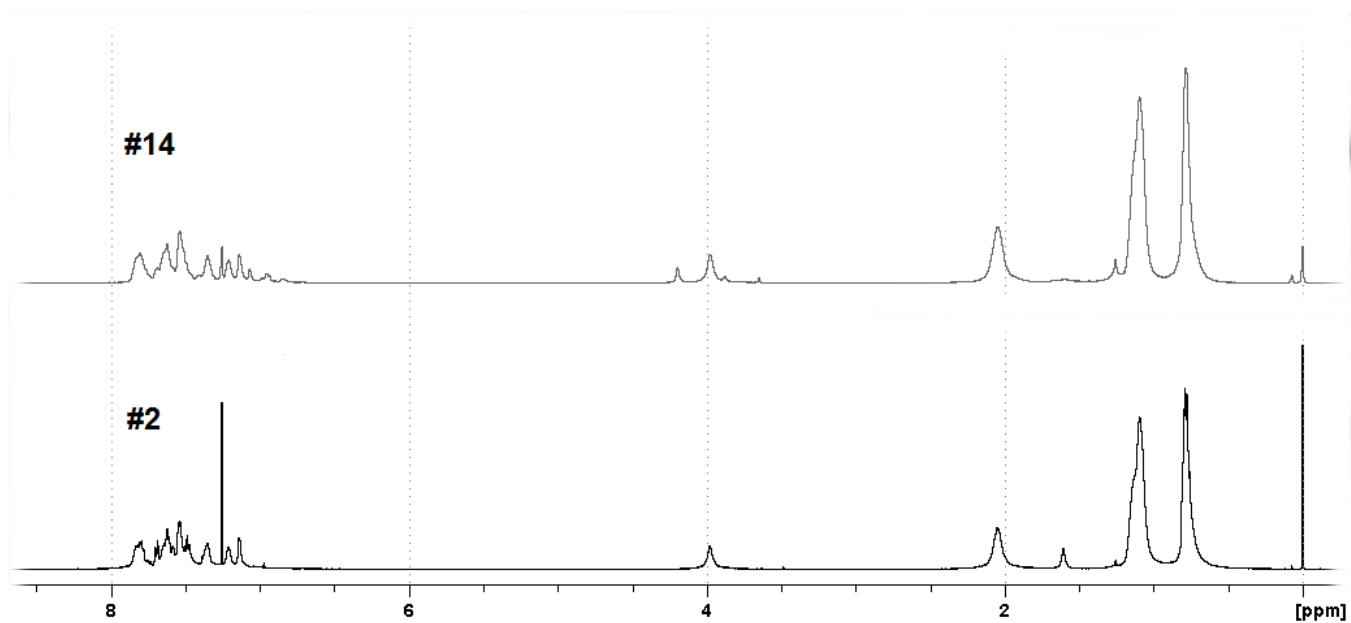


Figure S1. ¹H-NMR spectra comparatives of PAF batches (#14, %RR=69, Mw: 6450kg/mol; and #2, %RR=87%, Mw: 20775kg/mol) in CDCl₃, 500MHz.

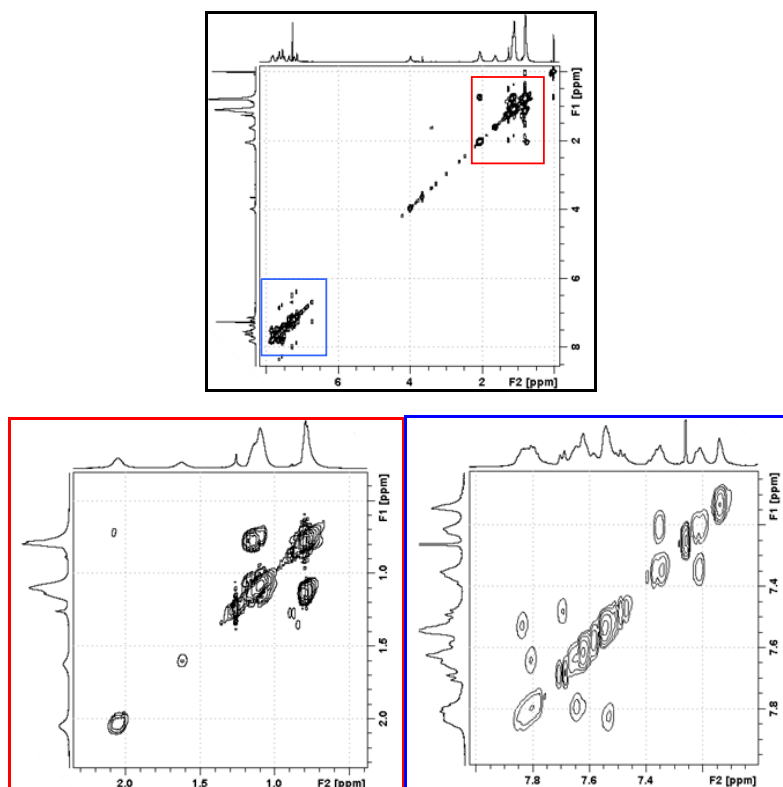


Figure S2. 2D ^1H - ^1H COSY NMR spectra of **PAF** (top). Aliphatic (red) and aromatic (blue) zones.

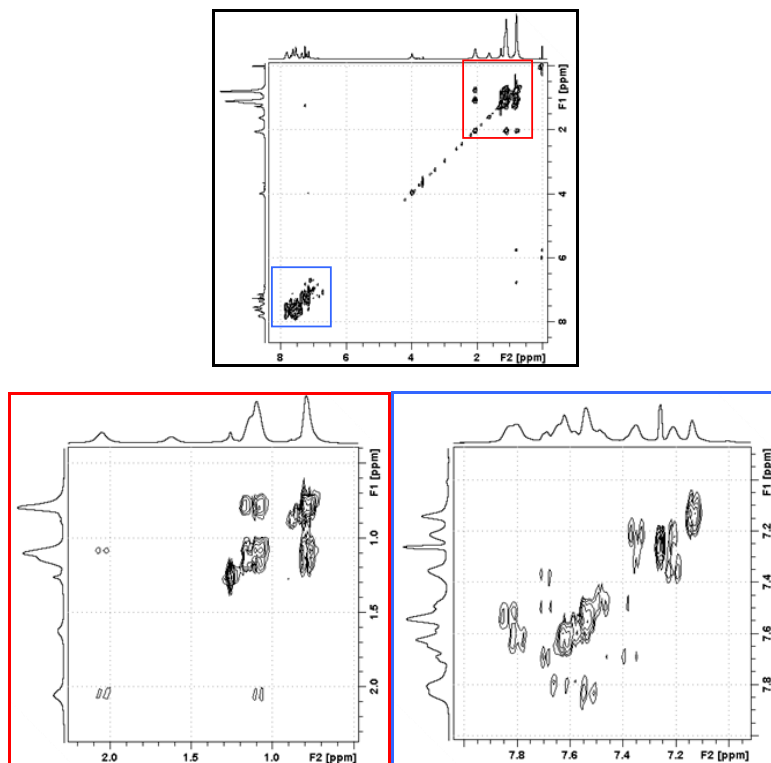


Figure S3. 2D ^1H - ^1H TOCSY NMR spectra of **PAF** (top). Aliphatic (red) and aromatic (blue) zones.

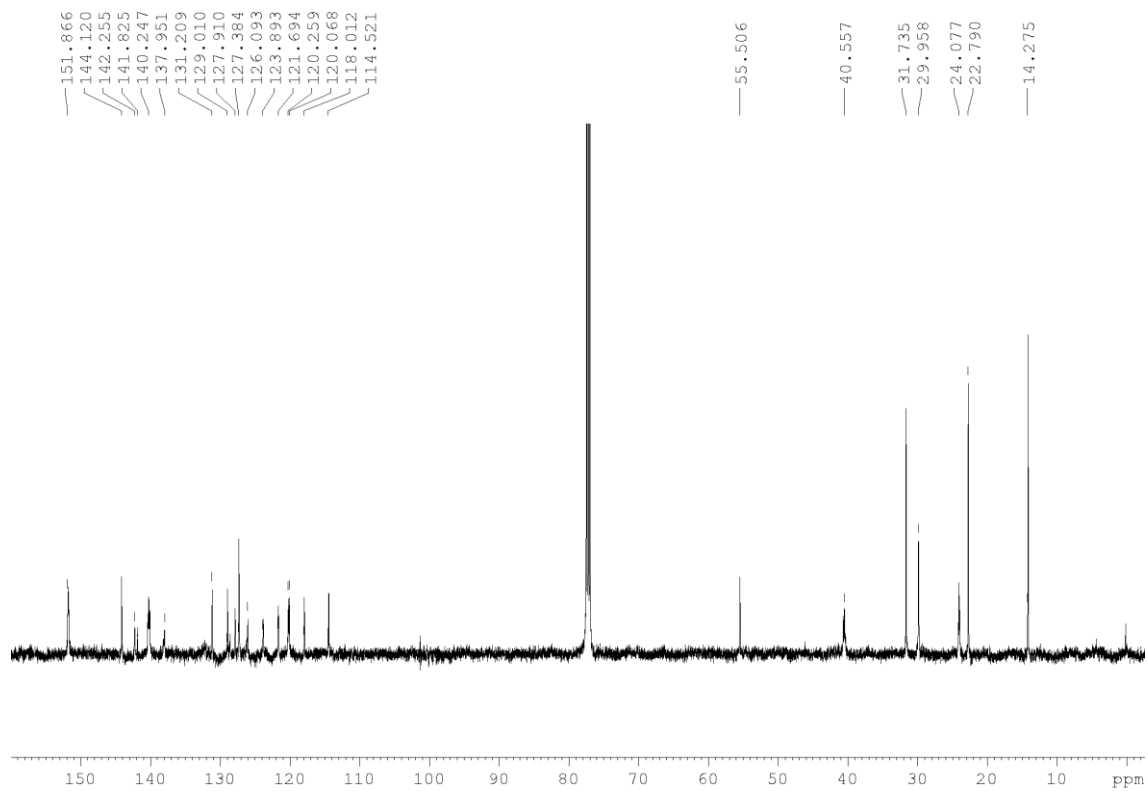


Figure S4. ^{13}C -NMR spectrum of **PAF** in CDCl_3 , 125 MHz.

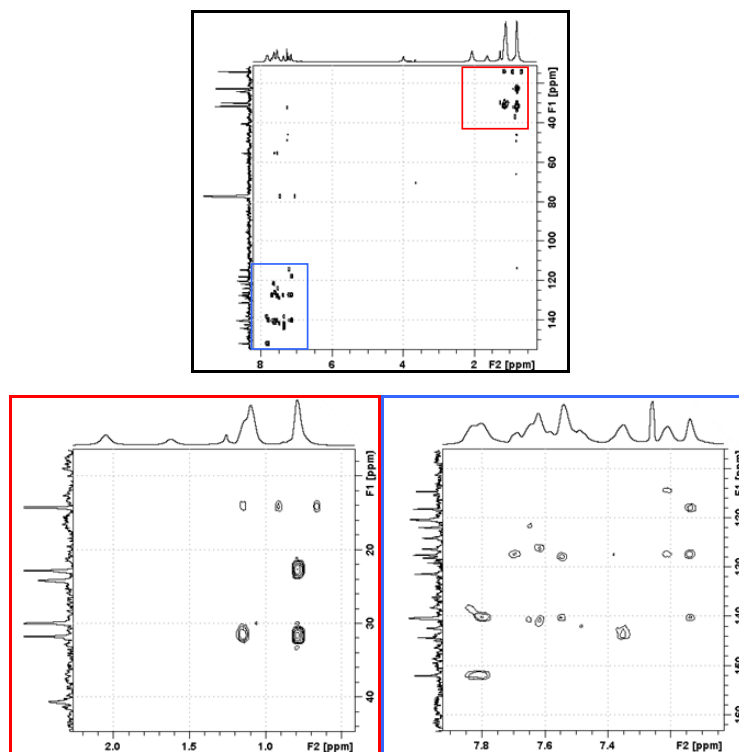


Figure S5. 2D ^1H - ^{13}C HMBC NMR spectra of **PAF** (top). Aliphatic (red) and aromatic (blue) zones.

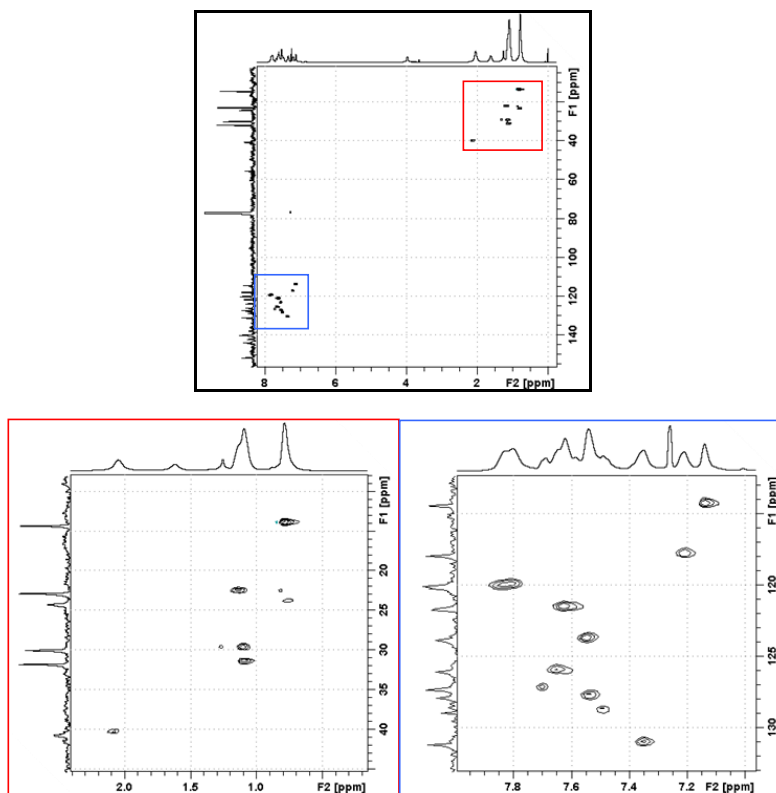


Figure S6. 2D ^1H - ^{13}C HMQC NMR spectra of **PAF** (top). Aliphatic (red) and aromatic (blue) zones.

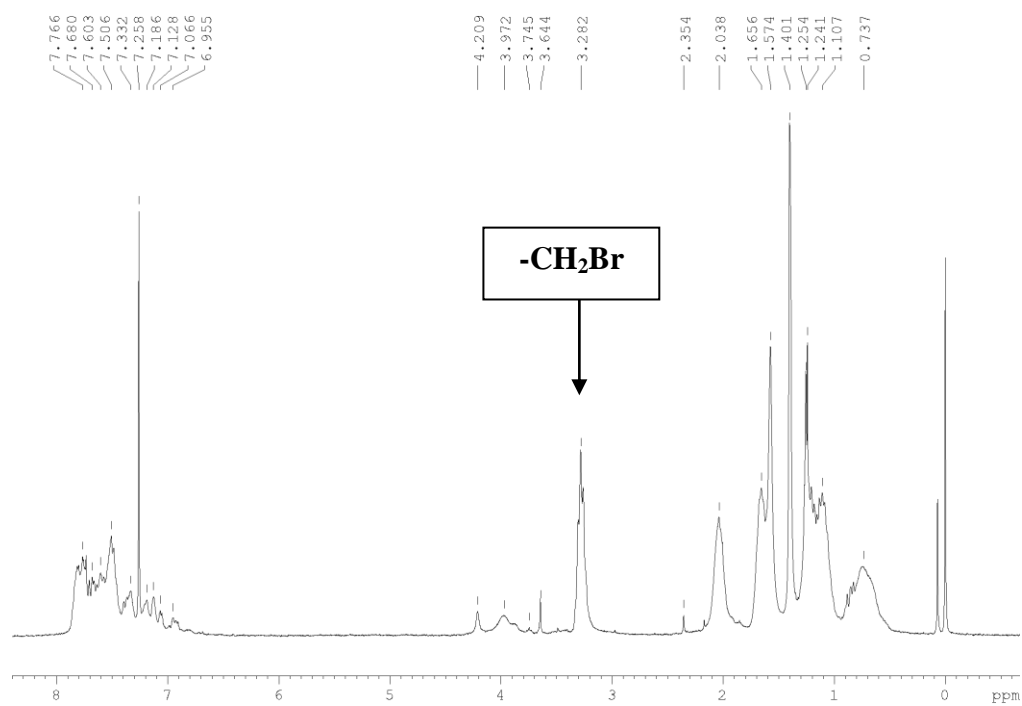


Figure S7. ^1H -NMR spectrum of PAFBr in CDCl_3 .

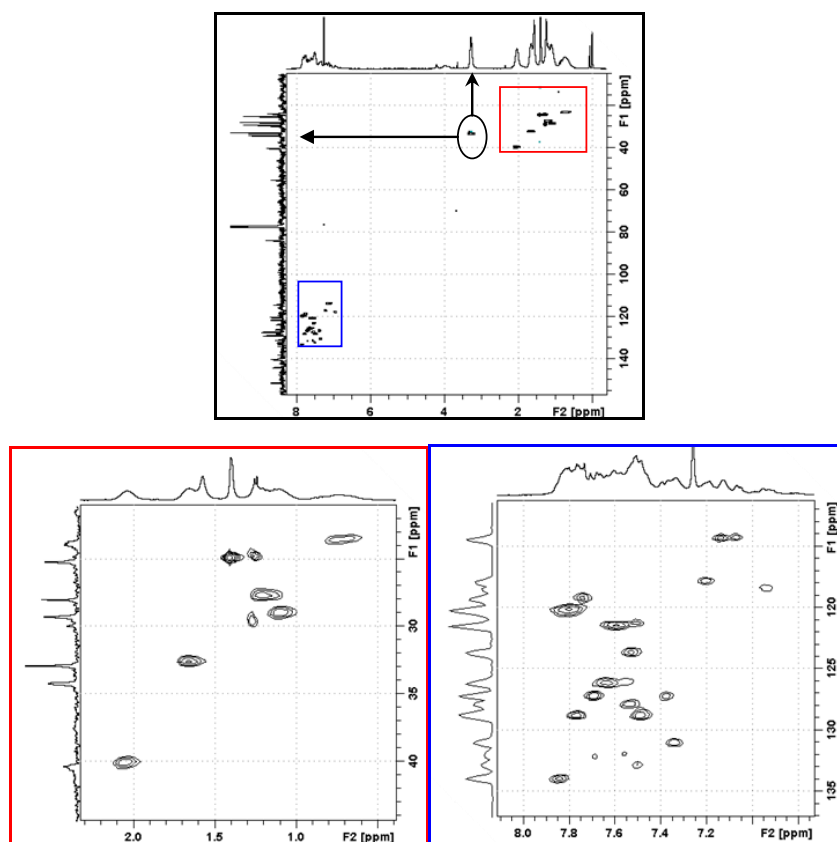


Figure S8. 2D ^1H - ^{13}C HMQC NMR spectra of PAFBr in CDCl_3 .

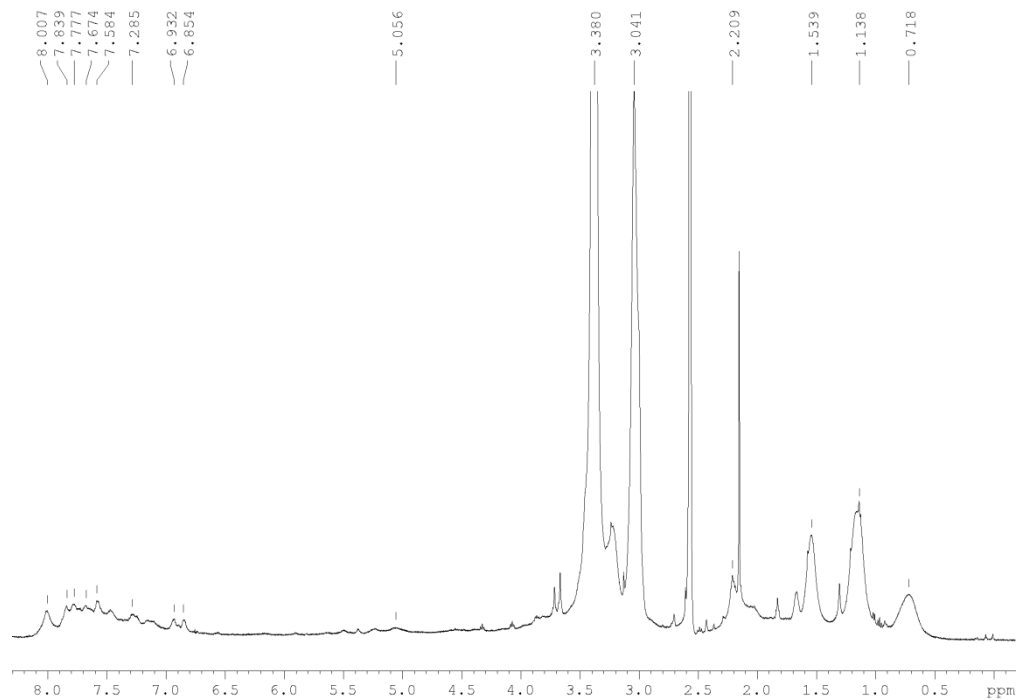


Figure S9. ^1H -NMR spectrum of PAFam in DMSO- d_6 .

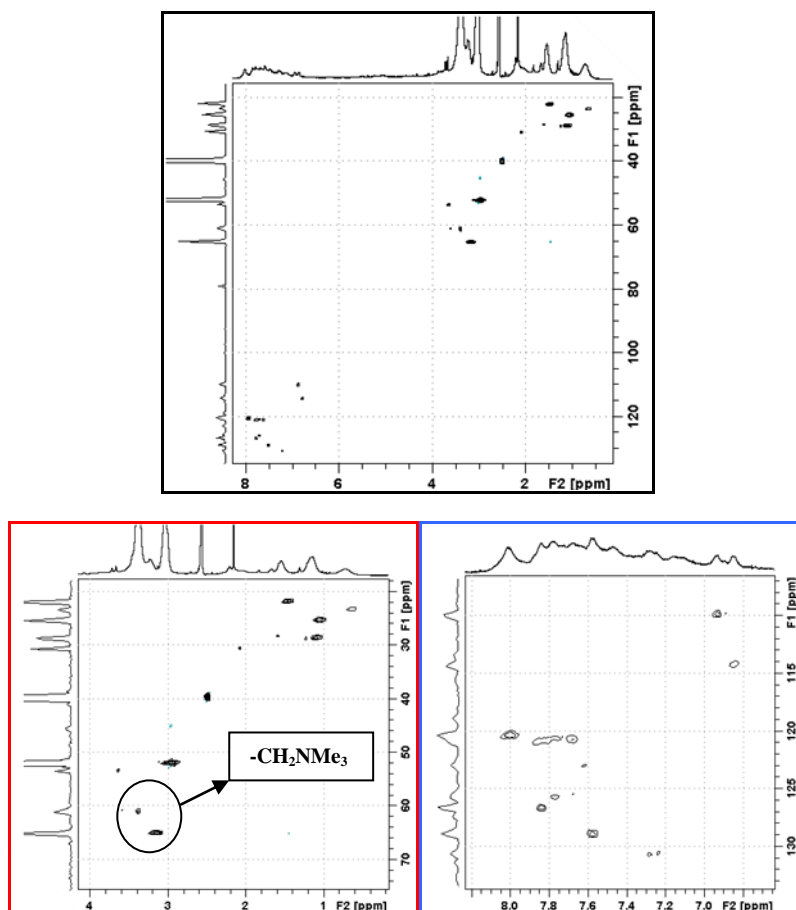
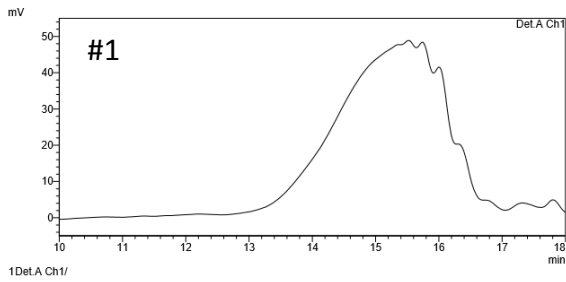
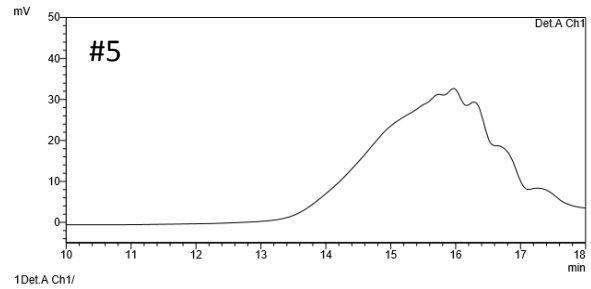


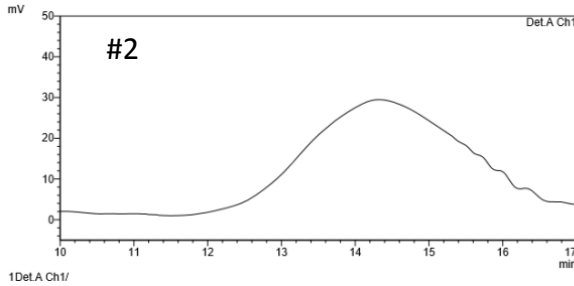
Figure S10. 2D ^1H - ^{13}C HMQC NMR spectra of PAFam in DMSO- d_6 .



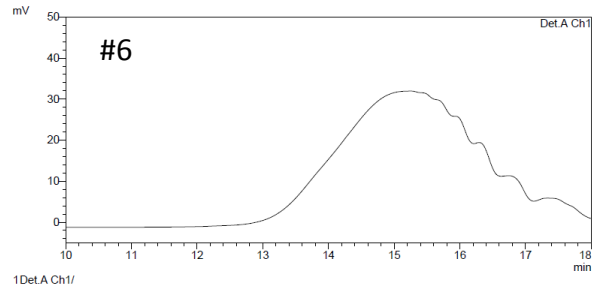
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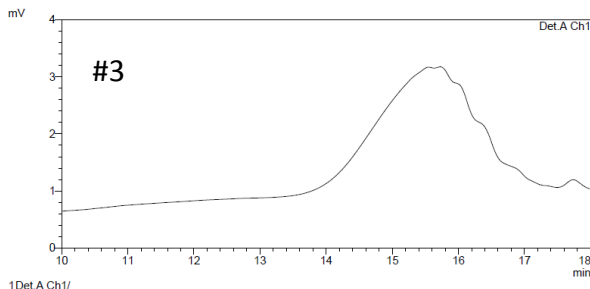
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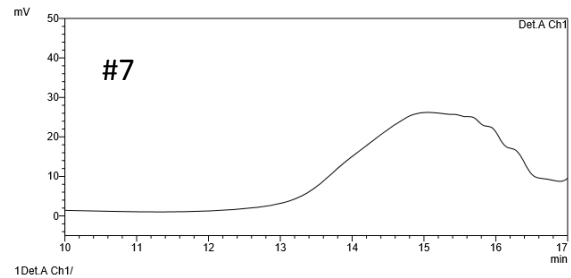
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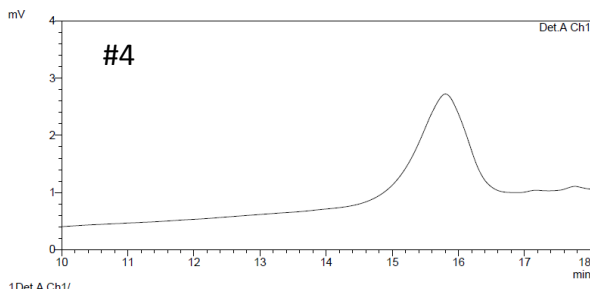
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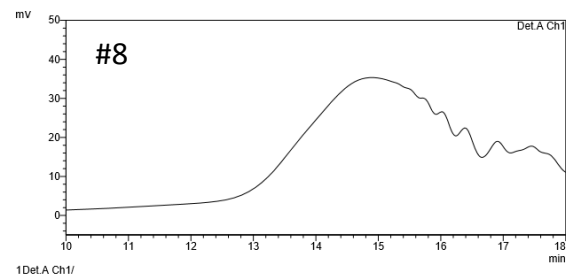
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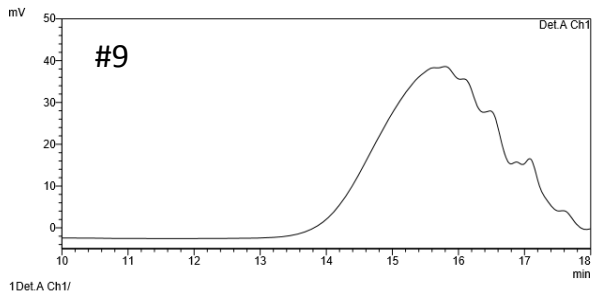
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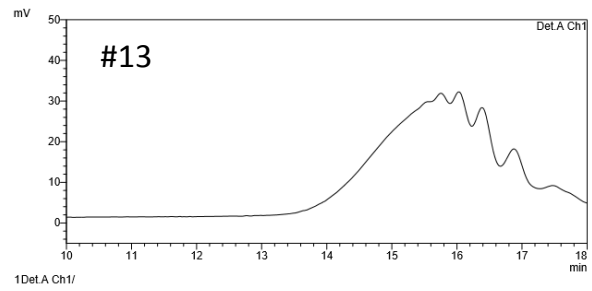
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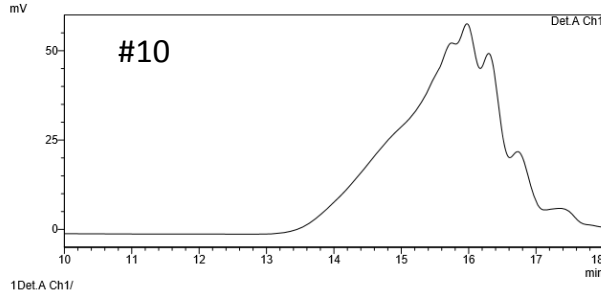
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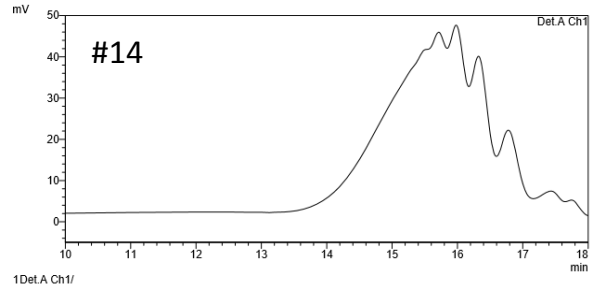
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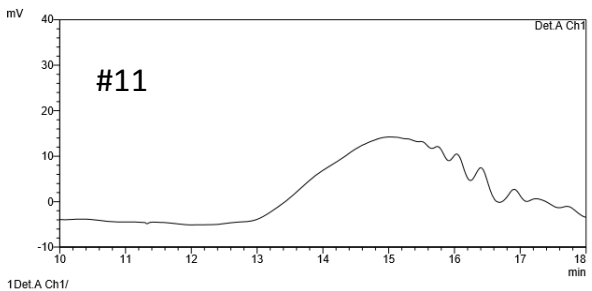
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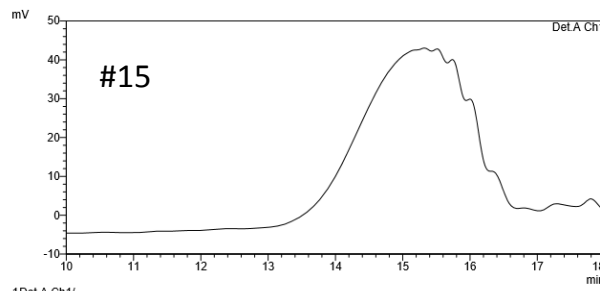
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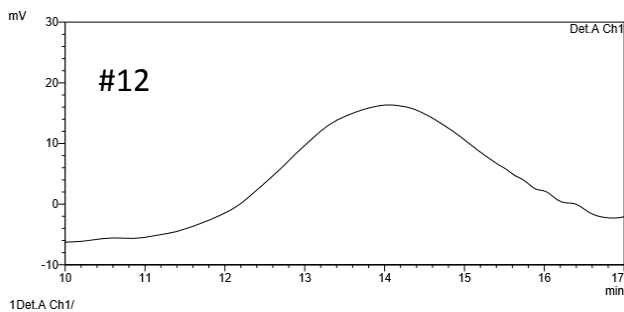
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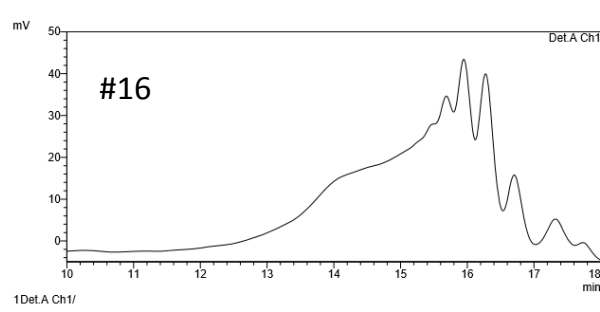
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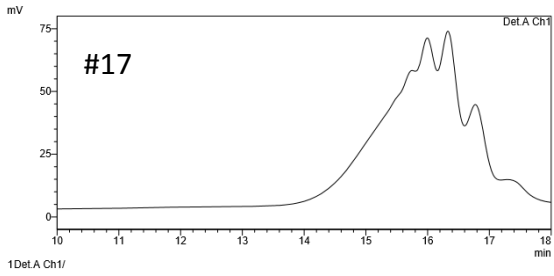
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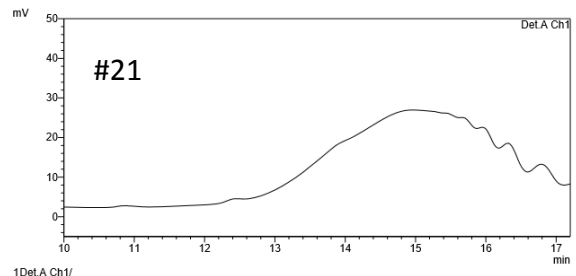
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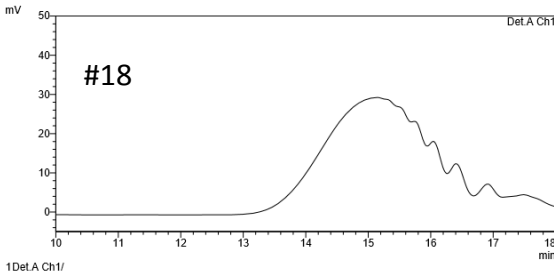
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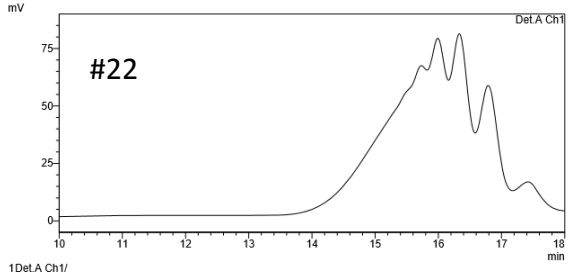
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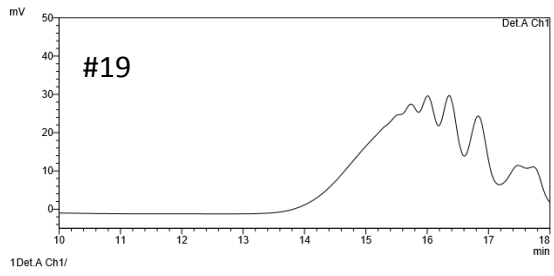
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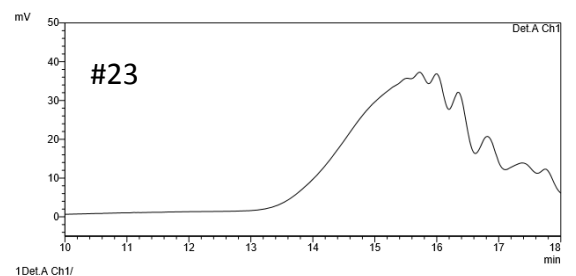
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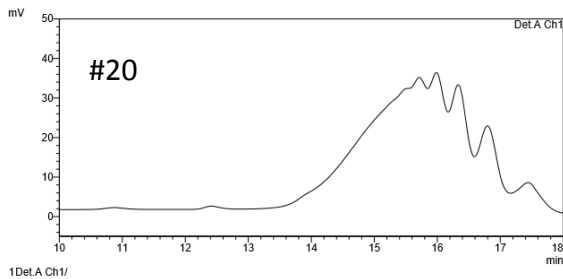
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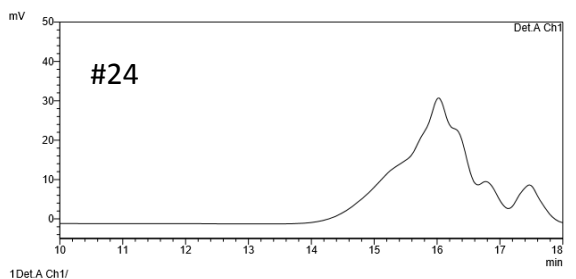
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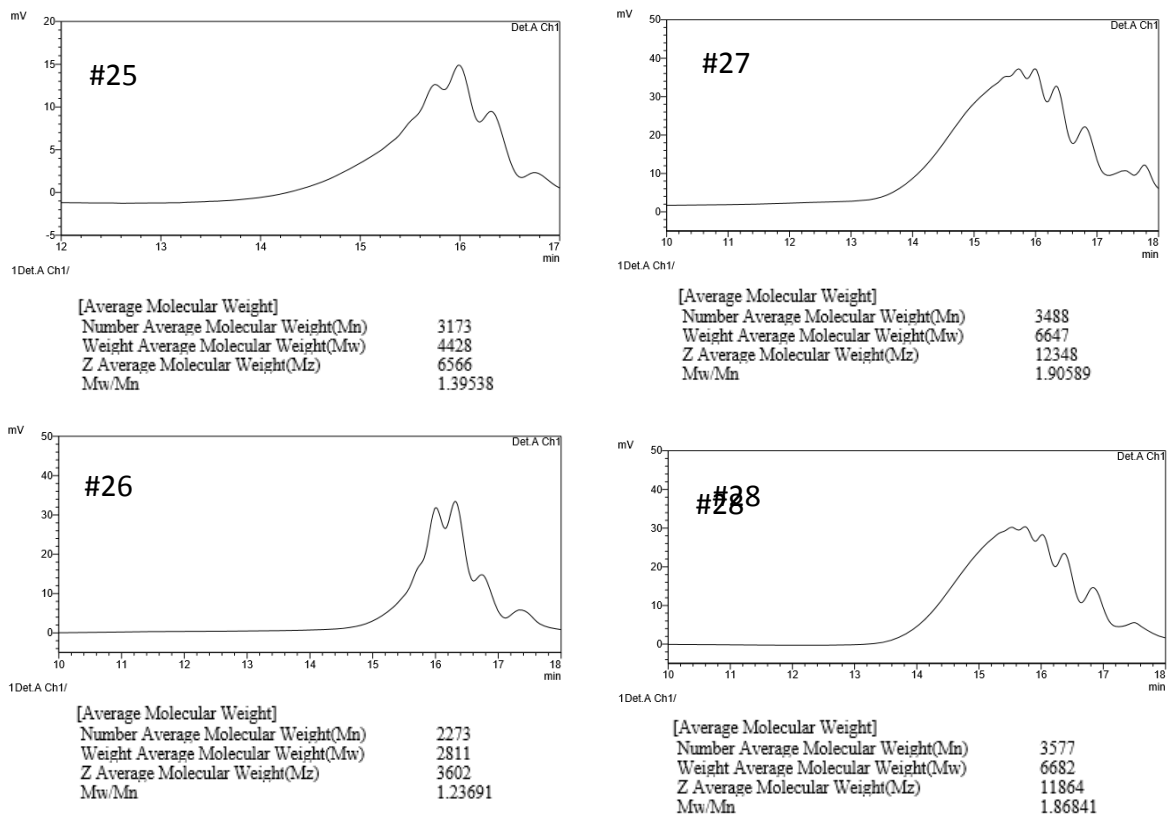


Figure S11. GPC Chromatograms of PAF batches.

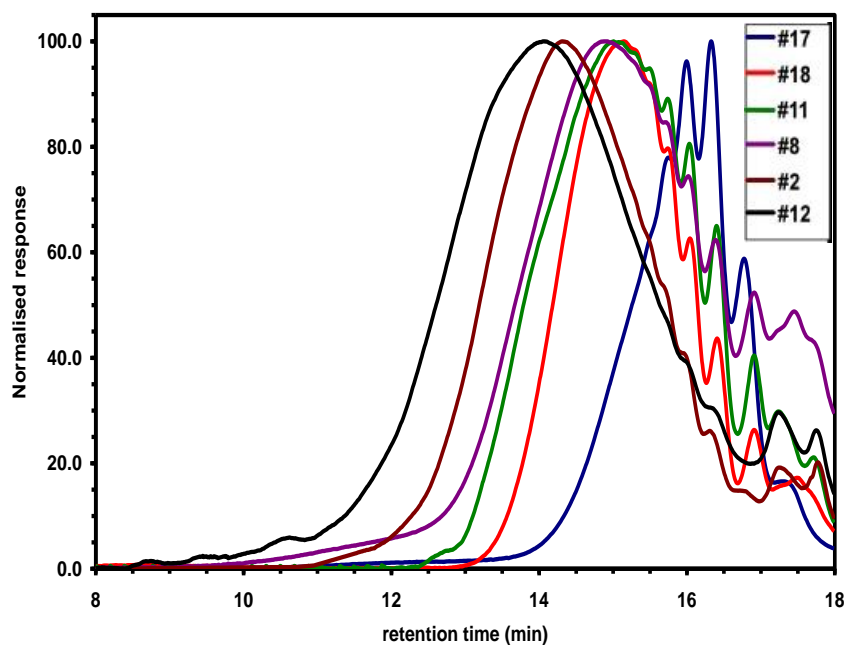


Figure S12. GPC Chromatograms for selected batches of PAF (Table S1).

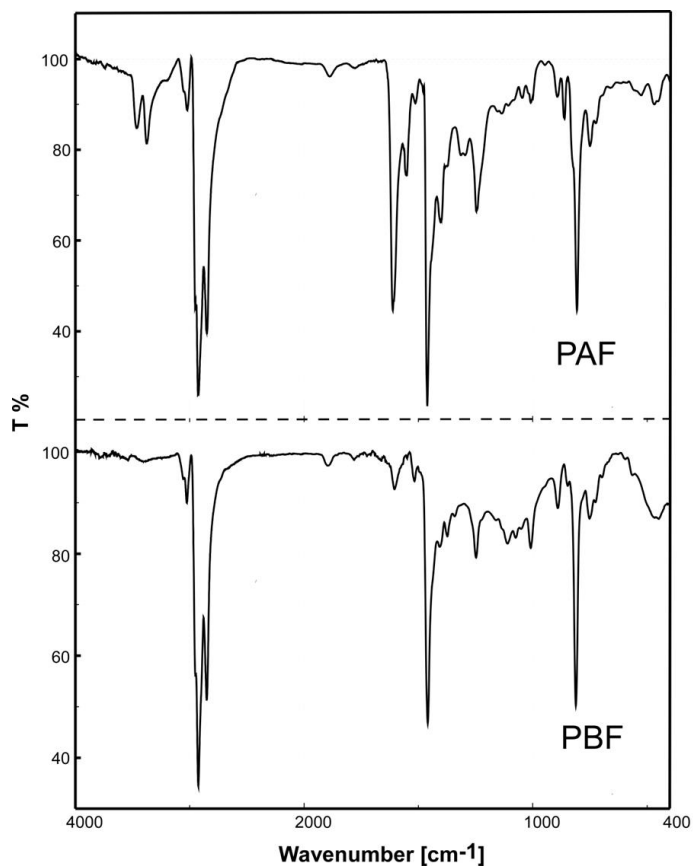


Figure S13. Comparative FT-IR spectra of **PAF** (top) and **PBF** (bottom) in BrK pellets.

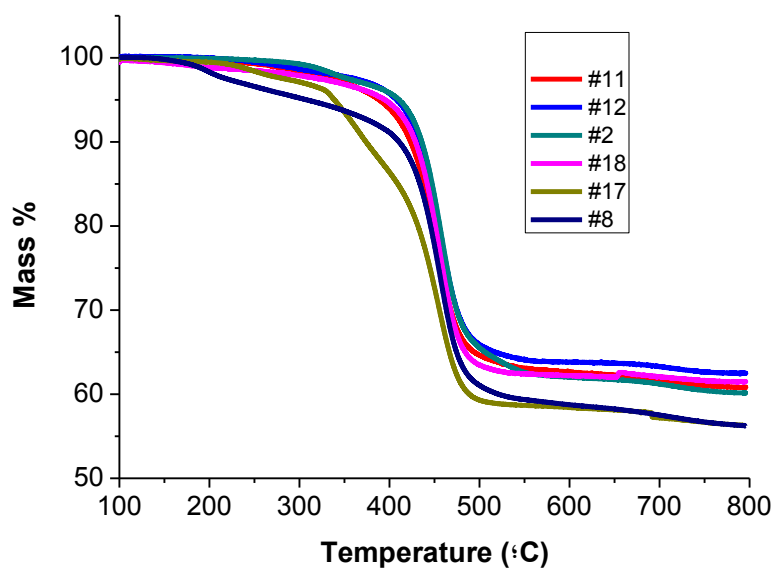


Figure S14. TGA curves for selected batches of **PAF** (Table S1).

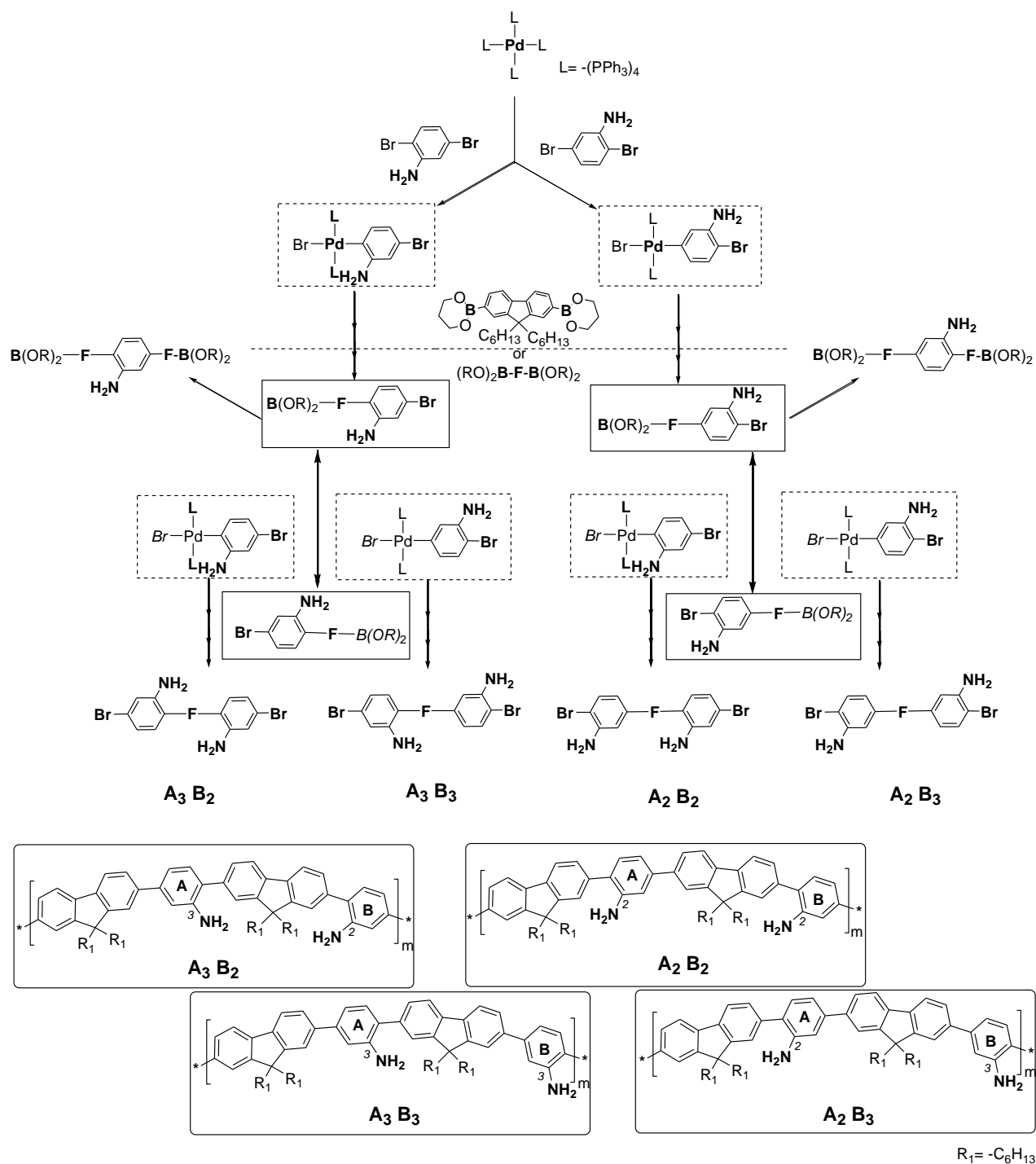
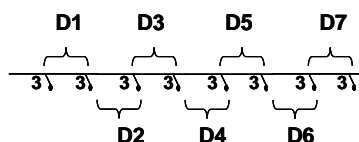


Figure S15. Proposed mechanism in the polymerization catalyzed by active species of palladium.

n= 6.....256..... 2:14:42:70:70:42:14:2
 n= 9.....512..... 2:16:56:112:140:112:56:16:2
 n= 10.....1024..... 2:18:72:168:252:252:168:72:18:2

n= 8.....256..... 2:14:42:70:70:42:14:2 NUMBER DYADS: n-1

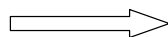


Analysis of probability for polymer sequence n=8, (2⁸=256)

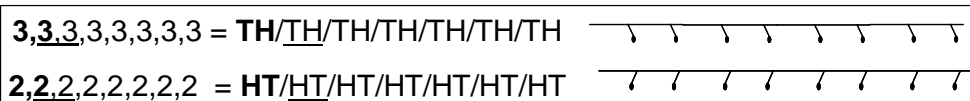
A. Two extreme cases of dyads (with two samples each):

➤ **Unidirectional Polymer**

-Total number case: 2

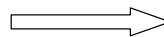


Prediction NMR:
 One peak higher 4.10ppm
 Areas (I:II); 100:0

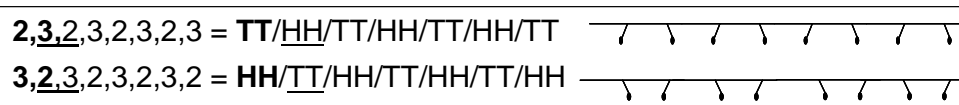


➤ **Alternating polymer**

-Total number case: 2



Prediction NMR:
 One peak lower 4.10ppm
 Areas (I:II); 0:100



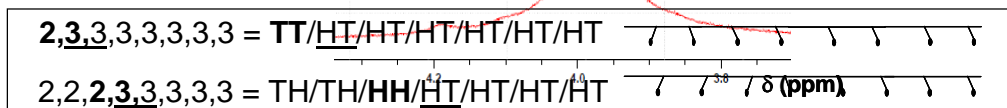
B. Two proximity extreme cases of dyads (with two samples each):

➤ **Block copolymer**

-Total number case: 14

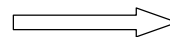
Area (I) Area (II)

Prediction NMR: Two peaks
 Areas (I:II) 6:1; %RR= 14%



➤ **“Quasi” alternating polymer.**

-Total number case: 14



Prediction NMR: Two peaks
 Areas (I:II) 1:6; %RR= 86%

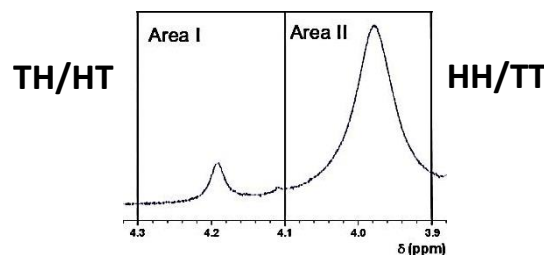
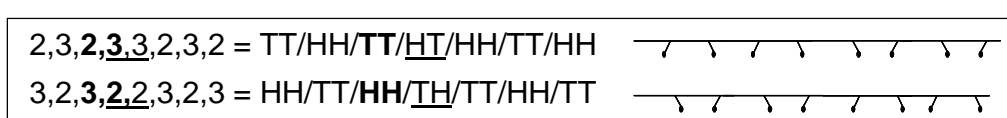


Figure S16. Analysis of probability for polymer sequence (n= number of chains) and correspondence of number of dyad cases, as normal distribution (number of cases: 2ⁿ, for n= Mw/Munity).

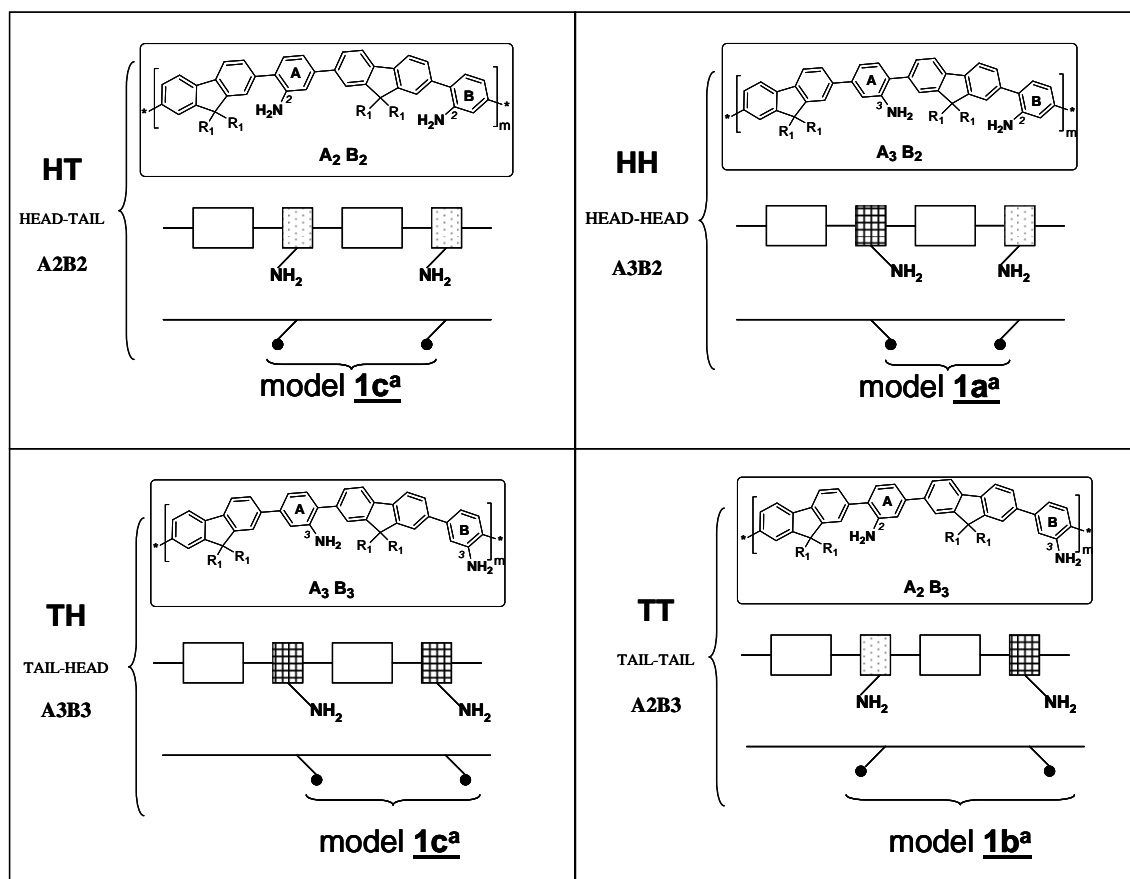


Figure S17. Symbols and structures used in this manuscript.

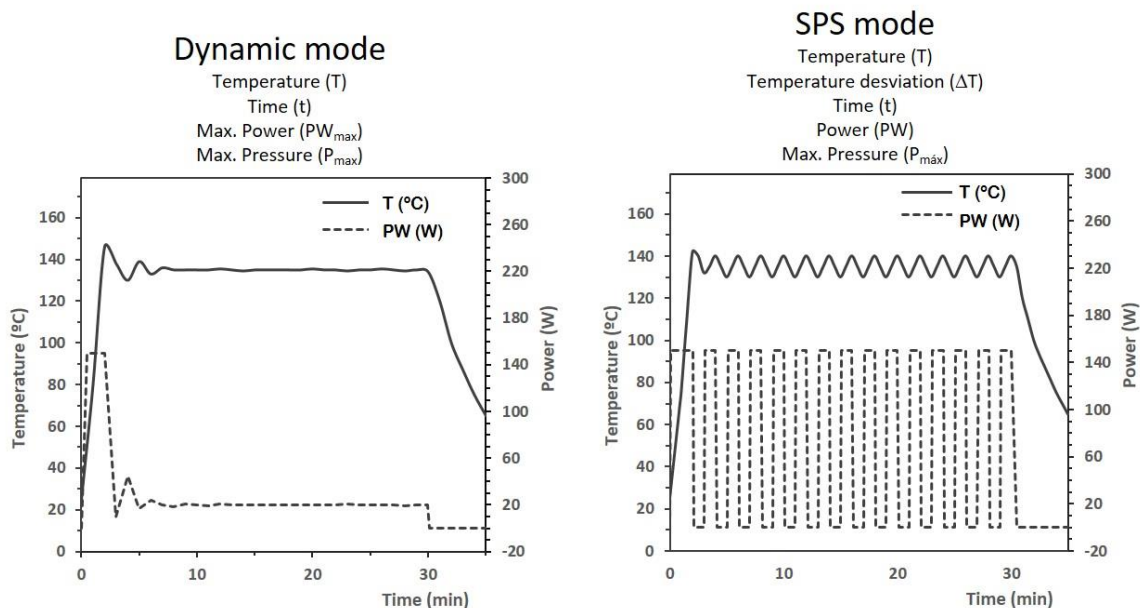


Figure S18. Temperature and power *versus* time for Suzuki Coupling microwave-assisted in Dynamic mode (left) and SPS mode, with temperature variation 5°C (right).

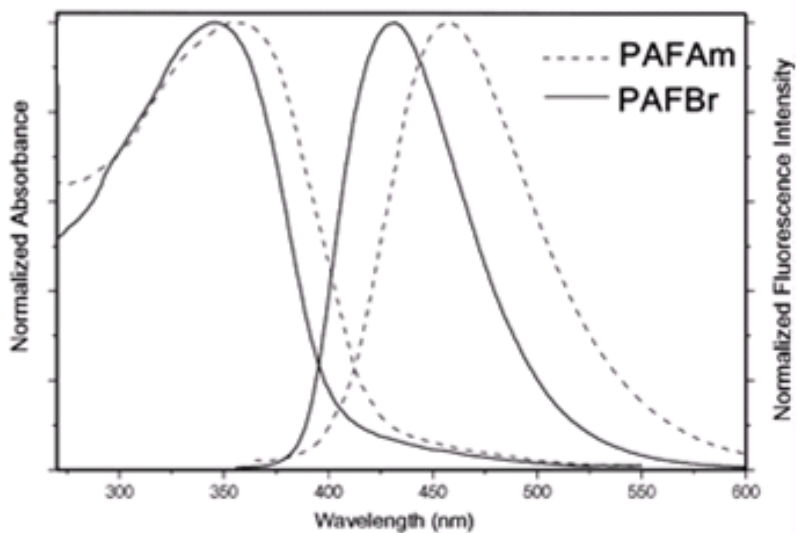


Figure S19. Normalized absorbance and emission spectra of PAFBr (solid line) and PAFAm (dashed line) in chloroform and water solutions, respectively.

Table S1. Optical data as function of RR of dyads in selected batches of PAF.

N #	Heating mode ^a	n ^b	% RR by NMR ^c	$\lambda_{\text{abs}}^{\text{max}}$ (nm)	$\lambda_{\text{em}}^{\text{max}}$ (nm)	$\Delta\lambda$ (nm)
17	MW-SPS	10	68	352	421	69
18	MW-SPS	23	95	360	419	59
11	Oil bath	24	93	360	421	61
8	MW-SPS	34	93	358	421	63
2	Oil bath	47	86	362	419	57
12	Oil bath	83	68	362	419	57
<i>Ref^d</i>	Oil bath	82	-	361	422	61

a See experimental conditions of each batch.

b n (number of monomer unities) = M_w/M_u ; where M_u = molecular weight unity (441.7 g/mol).

c. Percentages of regioregularity (%RR) were calculated in based on ratio NMR to integrated the Area II ($3.9 > \delta > 4.1$ ppm) over Total Area ($3.9 > \delta > 4.3$ ppm).

d. Yamaguchi, I.; Mizoguchi, N.; Sato, M. *Macromolecules* 2009, 42, (13), 4416-4425.

Table S2. Effect of polymerization time.

N #	Heating mode ^b	Time	Yield (%)	M_w^a (kg/mol)	PDI ^b	n ^b	% RR by NMR ^c
1	Oil bath	1 d	62	9.90	1.9	22	75
15		2 d	55	8.86	1.6	20	75
2		3 d	86	20.77	1.6	47	86
16		4 d	58	10.70	1.9	24	61
17	SPS	7 min	37	4.21	1.6	10	68
7		14 min	90	11.60	2.0	26	72
8		22 min	99	15.02	2.1	34	93
18		30 min	93	10.11	1.7	23	95

a. M_w = weight-average molecular weight, estimated by GPC in THF on the basis of polystyrene calibration.

b. PDI (Polydispersity index) = M_w/M_n ; where M_n = number-average molecular weight, and n (number of monomer unities) = M_w/M_u ; where M_u = molecular weight unity (441.7 g/mol).

c. Percentages of regioregularity (%RR) were calculated in based on ratio NMR to integrated the Area II ($3.9 > \delta > 4.1$ ppm) over Total Area ($3.9 > \delta > 4.3$ ppm).

Table S3. Effect of microwave parameter: solvent and volume on the polymerization.^a

N #	Solvent (v/v)	Volume (mL)	Yield (%)	M _w (kg/mol)	PDI	n	% RR by NMR ^c
19	THF/H ₂ O (2:1)	6	96	4.88	1.7	11	60
7	Toluene/H ₂ O (2:1)	6	90	11.60	2.0	26	72
20		4.5	73	6.08	1.9	14	77
21		3	86	14.14	2.6	32	82

^a Same description on table S1.

Table S4. Effect of power microwave on the polymerization.^a

N #	Power (W)	Yield (%)	M _w (kg/mol)	PDI	n	% RR by NMR
22	100	56	4.39	1.7	10	70
7	150	90	11.60	2.0	26	72
23	200	84	7.42	1.9	17	81

^a Same description on table S1.

Table S5. Effect of reaction temperature and the temperature fluctuation on the polymerization.^a

N #	Temp. (°C)	ΔT (°C)	Yield (%)	M _w (kg/mol)	PDI	n	% RR by NMR
24	80	1	3	3.96	1.4	9	38
25		5	5	4.43	1.4	10	64
26	100	5	11	2.81	1.3	6	51
27	135	1	72	6.65	1.9	15	78
7		5	90	11.60	2.0	26	72
28	150	5	86	6.68	1.9	15	84

^a Same description on table 1.