

## Supporting Information

### Konjac glucomannan induced retarding effects on the early hydration of cement

**Table S1.** OPC composition analysis

Phase	CaO	SiO <sub>2</sub>	Al <sub>2</sub> O <sub>3</sub>	SO <sub>3</sub>	Fe <sub>2</sub> O <sub>3</sub>	MgO	K <sub>2</sub> O	TiO <sub>2</sub>
Content (%)	64.65	20.42	4.68	4.55	2.92	1.08	0.78	0.32
Phase	P <sub>2</sub> O <sub>5</sub>	Na <sub>2</sub> O	SrO	MnO	BaO	Cl	ZnO	ZrO <sub>2</sub>
Content (%)	0.24	0.11	0.06	0.05	0.05	0.04	0.03	0.02

**Table S2.** CSA composition analysis

Phase	CaO	Al <sub>2</sub> O <sub>3</sub>	SO <sub>3</sub>	SiO <sub>2</sub>	Fe <sub>2</sub> O <sub>3</sub>	TiO <sub>2</sub>	MgO	F
Content (%)	60.73	15.26	14.21	4.97	2.52	0.87	0.67	0.22
Phase	Na <sub>2</sub> O	Cl	P <sub>2</sub> O <sub>5</sub>	SrO	MnO	BaO	ZrO <sub>2</sub>	ZnO
Content (%)	0.15	0.15	0.06	0.05	0.05	0.05	0.04	0.01

**Table S3.** CAC composition analysis

Phase	Al <sub>2</sub> O <sub>3</sub>	CaO	SiO <sub>2</sub>	TiO <sub>2</sub>	Fe <sub>2</sub> O <sub>3</sub>	SO <sub>3</sub>	K <sub>2</sub> O	MgO
Content (%)	50.59	36.48	6.11	2.88	2.05	0.64	0.58	0.22
Phase	P <sub>2</sub> O <sub>5</sub>	ZrO <sub>2</sub>	SrO	Cl	Na <sub>2</sub> O	ThO <sub>2</sub>	Ga <sub>2</sub> O <sub>3</sub>	Nb <sub>2</sub> O <sub>5</sub>
Content (%)	0.17	0.10	0.06	0.04	0.04	0.02	0.01	0.01

**Table S4.** Molecular weight and apparent viscosity of KGM

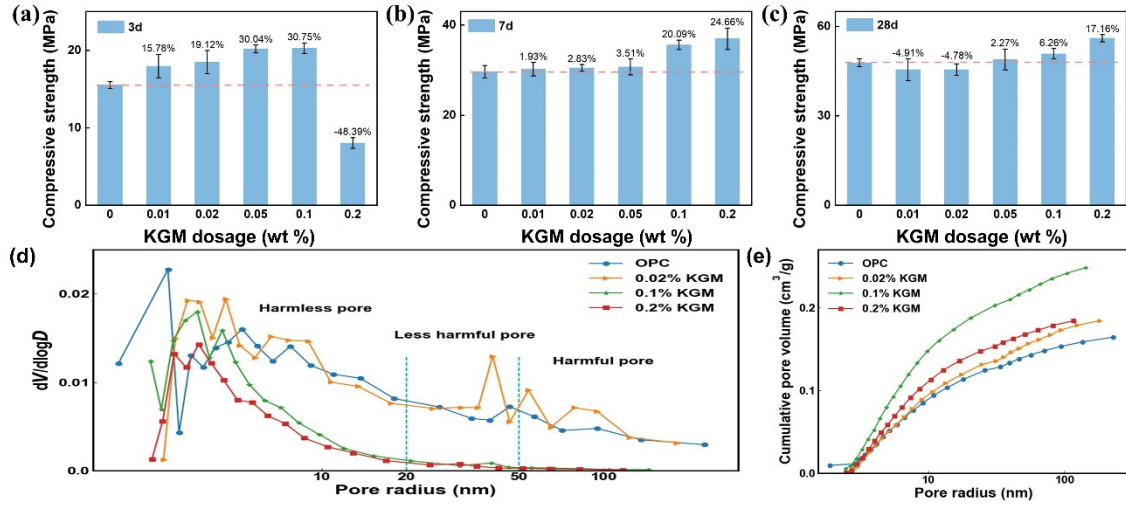
	Treatment time (min)	Viscosity average molecular weight	
		(10 <sup>5</sup> Da)	apparent viscosity (mPa·s)
KGM-1	60	0.8	5800
KGM-2	20	1.1	10600
KGM-3	0	2.8	17650

**Table S5.** The influence of various admixtures on cement pastes' setting time and heat release

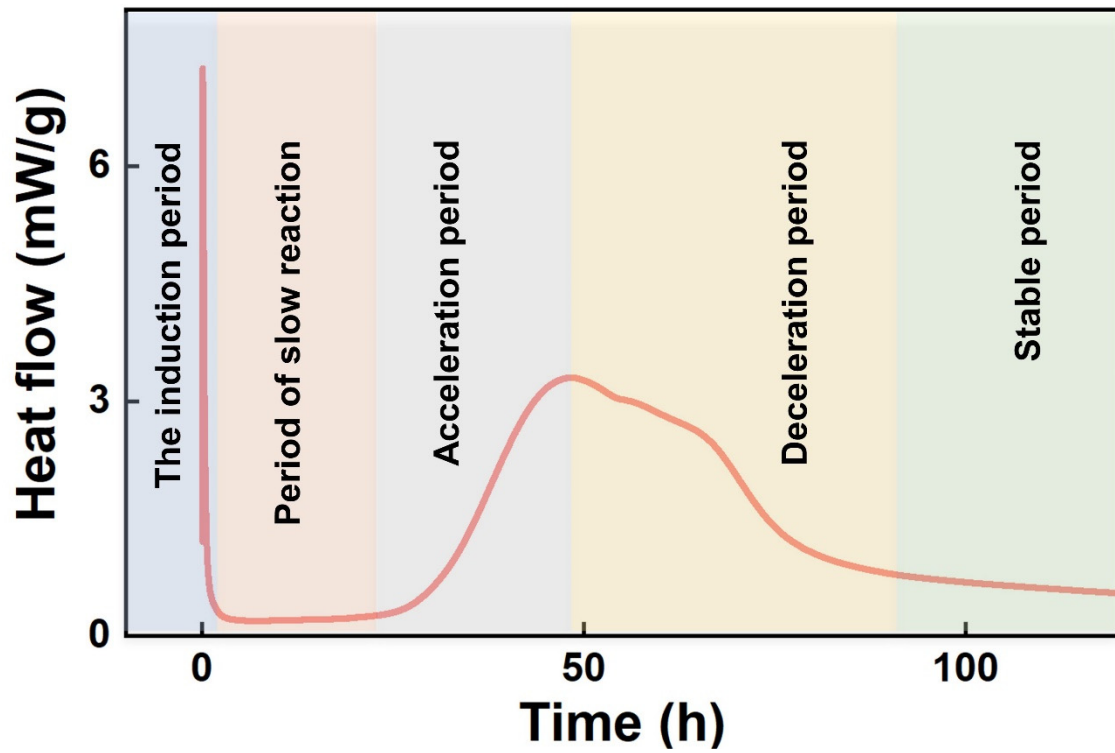
Admixtures	Optimal dosage (wt %)	Retarding time (min)		Temperature rise inhibiting		Refs
		$\Delta t_I$ (= $t_{I(admixture)} - t_{I(OPC)}$ )	$\Delta t_F$ (= $t_{F(admixture)} - t_{F(OPC)}$ )	Heat release delay time (h)	$\Delta_{Heat\ release} = \frac{Q_{OPC} - Q_{admixture}}{Q_{OPC}}$	
Sodium Fluosilicate	0.5	580	576		-	[31]
Threitol	1.5	15	30		-	
Xylitol	0.7	325	355		-	[32]
Sorbitol	0.8	335	375		-	
Sucrose	0.15	310	340		-	
Triisopropanolamine	1		-	0.3	+6%	[33]
Glucose	1		-	5	+13%	
Starch	0.1		-	11	+7%	[34]
Zinc oxide / sucrose	0.5/0.15		-	43	+7%	[35]
Triethanolamine	0.1	40	-	0.3	-	[36]
Sodium gluconate	0.24	145	150	40	-	[37,38]
Maltodextrin	0.2	180	480	18	-12%	[12]
KGM	0.2	385	1150	90	-15%	Our work

**Table S6.** The influence of different KGM content and molecular weight on cement hydration process and heat release.

Fig.	Cement and dosage (wt %) /molecular weight (10 <sup>5</sup> Da)	The induction period deadline (h)	The peak exotherm of acceleration (mW/g)	The acceleration period deadline (h)	The cumulative heat (J)
a, e	OPC (0, 0.02, 0.1, 0.2)	3.5, 4.4, 20.0, 70.0	4.4, 4.1, 3.3, 2.8	17.5, 20.8, 48.2, 129.8	644.5, 658.0, 653.2, 566.9
b, f	CAC (0, 0.05, 0.2)	5.5, 10.0, 20.0	11.3, 9.8, 1.7	9.2, 17.0, 26.5	319.5, 309.5, 233.0
c, g	CSA (0, 0.05, 0.2)	2.0, 3.0, 4.0	5.58, 5.54, 3.43	3.2, 4.2, 6.7	243.4, 235.9, 221.0
d, h	OPC (0, 0.8, 1.1, 2.8)	3.5, 5.2, 5.8, 11.0	2.2, 1.6, 1.5, 0.8	16.5, 33.5, 34.8, 48.2	333.3, 325.9, 321.9, 319.4



**Figure S1** The compressive strength of cement mortar admixed with different KGM dosages with w/c ratio of 0.5 at (a) 3 days, (b) 7 days and (c) 28 days and the porosity of cement mortar admixed with different KGM dosages with w/c ratio of 0.5 at 28 days: (d) the distribution of pore size and (e) cumulative pore volume.



**Figure S2** Schematic diagram of cement hydration process.

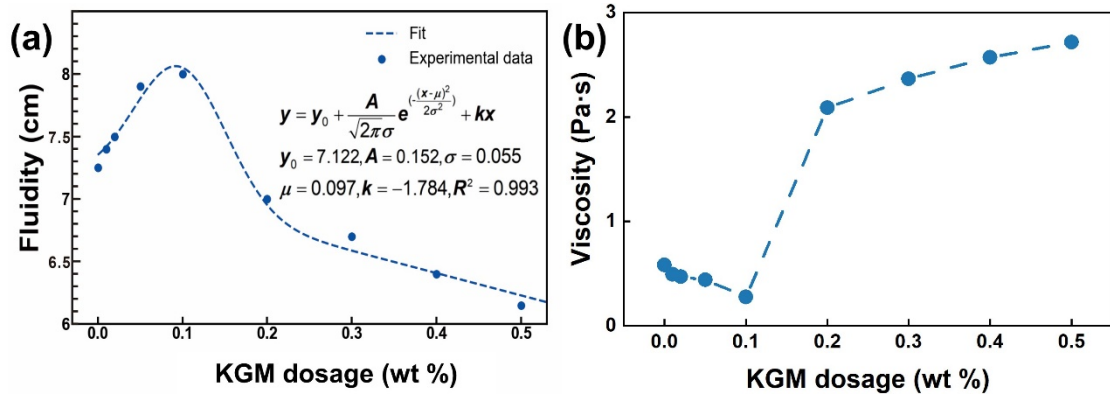


Figure S3 (a) The fluidity of cement pastes with different dosages of KGM with w/c ratio of 0.26 and (b) viscosity of cement paste with different KGM contents.

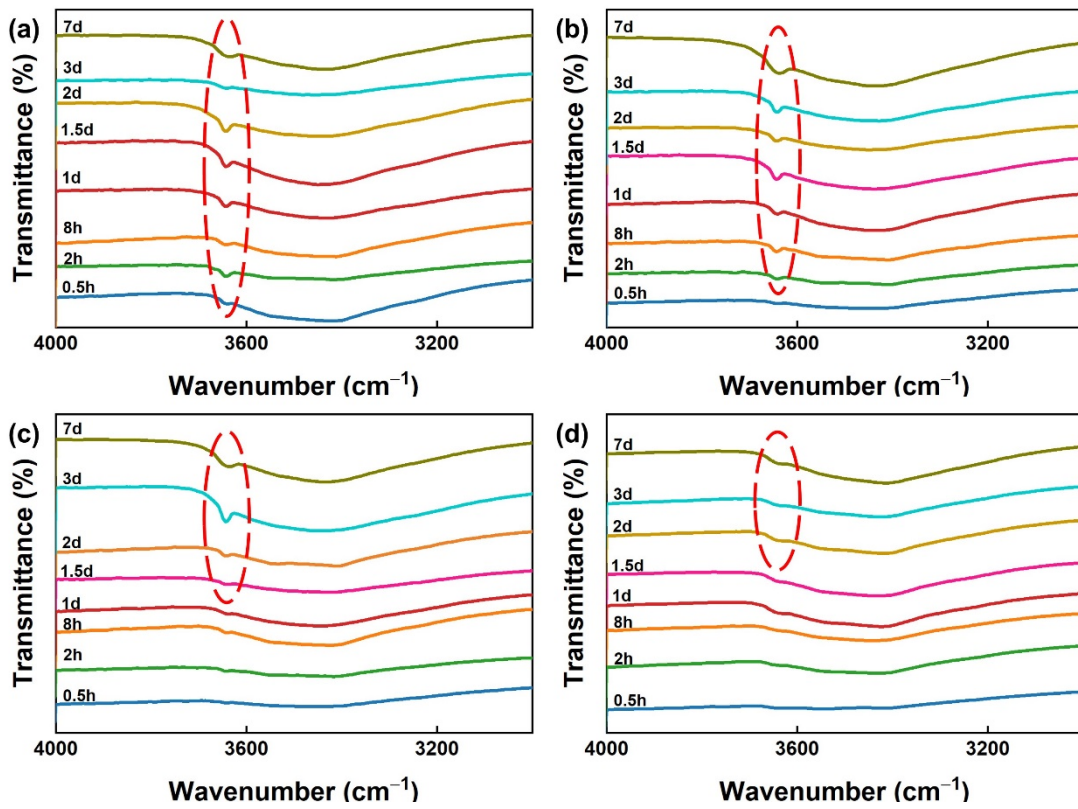
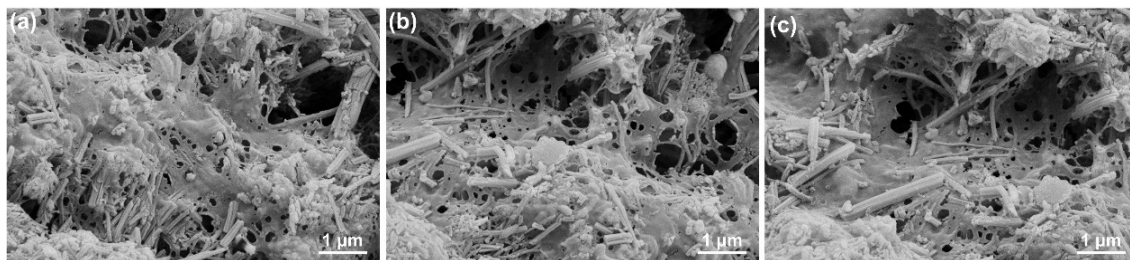


Figure S4 FTIR spectra of the cement pastes admixed with various KGM dosages: (a) pristine OPC and OPC with (b) 0.02 wt %, (c) 0.1 wt %; (d) 0.2 wt % of KGM.



**Figure S5 (a-c)** The morphology of hydration products changes with 0.2 wt % KGM dosages.