

Article

Preparation and Characterization of Electrospun EVOH/Ti₃C₂ Composite Fibers

Xiang Li * and Qiao Xu

Key Laboratory of Functional Fibers and Intelligent Textiles, Yuanpei College, Shaoxing University, Shaoxing 312000, China

* Correspondence: xiangli@usx.edu.cn

Abstract: In this work, the EVOH/Ti₃C₂ composite fibers were prepared via electrospinning and the effect of added Ti₃C₂ on the structure and properties of electrospun EVOH fibrous membranes was further investigated. The morphology, crystal structure, thermal properties, wettability, tensile properties, as well as air permeability and water vapor permeability of as-prepared EVOH/Ti₃C₂ composite fibers were studied. The Ti₃C₂ is uniformly loaded onto the surface and inside the composite fiber and affects the fiber diameters. Furthermore, The Ti₃C₂ self-orient along the fiber axis and does not change the crystal structure of the electrospun EVOH fibers, improving the crystallinity and thermal stability of the electrospun EVOH/Ti₃C₂ fibrous membranes. With the increase in the Ti₃C₂ concentration in the electrospinning polymer solution, the addition of Ti₃C₂ not only rapidly improves the wettability of the fibrous membranes, but also enhances their air permeability, compared with the pristine electrospun EVOH fibrous membranes. The experimental results provide theoretical guidance for the preparation of Ti₃C₂ composite fibers, and also expand the application of electrospun EVOH and EVOH/Ti₃C₂ fibrous membranes.

Keywords: EVOH; Ti₃C₂; electrospinning; structure; properties



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1. Introduction

Electrospinning is a straightforward and prominent technique for the one-step preparation of nanoscale polymeric fibers, which has been extensively applied in drug delivery [1,2], energy storage [3,4], filtration [5,6], tissue engineering [7,8], separation and absorption [9,10]. In view of the wide range of spinnable materials and convenient modifications available within electrospinning, electrospun composite fibers have enormously expanded the versatility of electrospinning in the fields of textile, aerospace, agriculture, biology, defense and security [11,12] due to their high specific surface area, high porosity and interconnected pores [13,14].

Poly(ethylene-co-vinyl alcohol) (EVOH), as a semi-crystalline random copolymer composed of vinyl alcohol and ethylene blocks, is often employed in food packaging owing to its excellent gas resistance, chemical resistance and high transparency [15,16]. However, some completely different applications are found in electrospun EVOH composite fibers, such as filtration and biomedicine [17,18]. Mondragón et al. [19] prepared electrospun poly(ethylene-co-vinyl alcohol)(EVOH)/thermoplastic starch (TPS) blend nanofibers and found that the crosslinked EVOH/TPS fibrous membranes exhibited a superior fluid uptake ability (with 20 wt% of TPS) and superior barrier properties (with 20 and 40 wt% of TPS) for potential dressing materials in comparison to those observed in pristine electrospun EVOH fibrous membranes.

Ding et al. [20] fabricated a novel rechargeable N-halamine antibacterial material by functionalizing electrospun EVOH nanofibers with dimethylol-5,5-dimethylhydantoin (DMDMH). The as-prepared DMDMH functionalized EVOH nanofibers membranes provided rechargeable chlorination capacity, high inactivation efficacy against bacteria, high

filtration efficiency under low air resistance, and robust mechanical properties. Lu et al. [21] designed the superelastic and superhydrophobic thermoplastic polymeric nanofibrous aerogels (NFAs) for the removal of pollutants from water. Silane-coated EVOH NFAs exhibited a superior absorption capacity (40–92 g/g) for a variety of organic pollutants, which could be used in substantial industrial water purification applications. Hence, the application development of electrospun EVOH fibers has been actively underway.

Recently, MXenes have attracted a great deal of attention because of their high electrical conductivity, excellent solvent compatibility and stability, electrochemical behavior, hydrophilicity, and mechanical strength, which are compared to clay, graphene and GO [22,23]. The general formula of MXene is $M_{n+1}X_nT_x$, where M indicates an early transition metal, X signifies carbon and/or nitrogen and T indicates a surface functional group including O, F, and OH [24]. Among them, Ti_3C_2 , as the most commonly studied MXene, was first reported in 2011 [25] and later found to have antibacterial and antifouling properties, like silver [26,27]. Some research has attempted to coat Ti_3C_2 MXene onto fibers for wearable energy storage applications, but the Ti_3C_2 flakes were easily detached from the fibers [28]. It is useful to embed Ti_3C_2 into the fibers to form uniform composite fibers. Levitt et al. [29] produced a free-standing $Ti_3C_2T_x$ MXene/carbon nanofiber electrode by electrospinning $Ti_3C_2T_x$ MXene flakes with polyacrylonitrile (PAN) and carbonizing the fiber networks. Electrospun MXene/carbon nanofibers could suffer from the active material delaminating from the substrate during folding or bending, unlike coated electrodes, which are stable and durable composite electrodes.

In addition, Schauer et al. [30] produced electrospun poly(acrylic acid) (PAA)/ Ti_3C_2 , polyethylene oxide (PEO)/ Ti_3C_2 , and poly(vinyl alcohol) (PVA)/ Ti_3C_2 fibers and compared the effects of 1 wt% Ti_3C_2 addition on the three polymers, and found that Ti_3C_2 affected the solution properties of the polymer, especially the diameter of the fiber. It can be seen that the production of electrospun $Ti_3C_2T_x$ MXene composite fibers can further broaden the applications of $Ti_3C_2T_x$ MXene with the aid of the unique characteristics of electrospun fibers. $Ti_3C_2T_x$ MXene, as a new type of two-dimensional nanomaterial, holds regulable physicochemical properties, good biocompatibility, and outstanding photothermal conversion performance, showing a rapid expansion trend in various application fields such as energy storage, catalysis, electronics, electrochemical, electromagnetics, sensing and biomedical applications [31–33]. Moreover, $Ti_3C_2T_x$ MXene exhibits the antibacterial ability to physically damage bacterial membranes and chemically induce oxidative stress [34], and can be further combined with other materials to obtain multi-functional materials. Thus, in view of the unique properties of EVOH and $Ti_3C_2T_x$ MXene, the multi-functional fibrous membrane can be prepared by adopting electrospinning technology, with potential use in bioprotective materials, dressing materials, etc.

Herein, the electrospun EVOH/ Ti_3C_2 composite fibers were first prepared and the effect of added Ti_3C_2 on the structure and properties of electrospun EVOH fibrous membranes was investigated. This work focuses on the preparation of Ti_3C_2 composite fibrous membranes and characterizations of their structure and conventional properties. The morphologies of electrospun EVOH/ Ti_3C_2 composite fibers with different Ti_3C_2 concentrations in the electrospinning polymer solutions were characterized. Moreover, the crystal structures of the electrospun EVOH/ Ti_3C_2 composite fibers are presented. The properties of the composite fibrous membranes were evaluated, including thermal property, wettability, tensile properties, air permeability and water vapor permeability.

2. Experiments

2.1. Experimental Materials

Poly(ethylene-co-vinyl alcohol) (EVOH) (F171B) was obtained from Kuraray Co., Ltd., Tokyo, Japan. Isopropyl alcohol (IPA) was purchased from Aladdin Chemistry Co. Ltd., Shanghai, China. The 10 wt% EVOH solution was prepared by dissolving EVOH particles in a mixture of IPA/water with a weight ratio of 7/3 at 75 °C for 4 h. Various amounts of Ti_3C_2 dispersion with a known concentration of 25 mg/mL were added to the measured

EVOH solution prepared previously. The Ti_3C_2 concentrations were selected based on the reported research [29,30]. Uniform EVOH/ Ti_3C_2 solutions with 2.5 wt%, 5 wt% and 10 wt% Ti_3C_2 concentrations were obtained by calculating the proportion of polymer solutions.

2.2. Preparation of the EVOH/ Ti_3C_2 Fibers

The EVOH/ Ti_3C_2 polymer solution was transferred into the syringes at a constant flow rate of 3 mL/h. A high-voltage power supply with an applied voltage of 15 kV was supplied to the spinneret and collector. The distance between the needle tip to the collector was 15 cm. The experiments were carried out at a relative humidity of $45 \pm 5\%$ and at room temperature. During electrospinning, the polymer droplets formed Taylor cones under the action of electric field forces. When the electric field forces acting on the droplets overcame the surface tension and viscoelasticity of the polymer droplets, a jet was formed at the top of the droplet cone. Under the action of an external electric field, the jet was highly stretched axially by the electric field force, forming a short distance stable motion. Then, the charges on the surface of the jet repelled each other, causing the jet to produce lateral displacement and enter an unstable motion stage. Finally, the jet flew towards the collector in a spiral whip trajectory, and formed micro–nano fibers with the continuous volatilization of solvent [35,36]. The prepared EVOH/ Ti_3C_2 fibrous membranes were dried and used for the characterization of the structure and properties. Here the as-prepared electrospun EVOH/ Ti_3C_2 fibers with 2.5 wt%, 5 wt%, 10 wt% Ti_3C_2 concentrations in the electrospinning polymer solution are simply named EVOH/ Ti_3C_2 -1, EVOH/ Ti_3C_2 -2, EVOH/ Ti_3C_2 -3, respectively.

2.3. Characterization

Morphology of the electrospun EVOH/ Ti_3C_2 fiber was observed using a scanning electron microscope (SEM, S-4800, Hitachi Ltd., Tokyo, Japan) after gold coating. The SEM morphology was taken by randomly selecting four areas of the fibrous membrane. The average fiber diameters were measured from the SEM images of more than 100 fibers by the image visualization software (Image J v1.43) (National Institutes of Health, Bethesda, MD, USA). The error bars for the fiber diameters were based on the standard deviation of the record. The elemental distributions of electrospun EVOH/ Ti_3C_2 composite fibrous membranes were characterized by energy-dispersive X-ray spectroscopy (EDS, Bruker Quantax 400) (Bruker Instruments, Inc., Billerica, MA, USA).

The crystal structure was characterized by X-ray diffraction (XRD) and Fourier transform infrared spectroscopy (FTIR). XRD measurements were recorded at the beam line BL16B1 of Shanghai Synchrotron Radiation Facility (SSRF, Shanghai, China). The X-ray wavelength is 0.124 nm and a Mar 165 CCD detector with 2048×2048 pixels was used to collect two-dimensional (2D) patterns. The 2D patterns were accumulated over periods of 50 s. Fit 2D software (<https://www.esrf.fr/computing/scientific/FIT2D/>, accessed on 20 February 2024) was applied to transfer 2D patterns into 1D profiles. FTIR spectra were recorded based on the transmission mode using a Thermo Nicolet 6700 FTIR Spectrometer (Thermo Fisher Scientific Inc., Waltham, MA, USA) in the wavenumber range of $400\text{--}4000\text{ cm}^{-1}$ at room temperature. The XRD and FTIR tests were carried out in three randomly selected places of the fiber membrane, respectively.

Thermal properties of electrospun EVOH/ Ti_3C_2 composite fibrous membranes were measured using a differential scanning calorimeter (DSC4000) (PerkinElmer, Inc., Waltham, MA, USA), whose tests were set to heat up from 30° to 240° at a rate of $10^\circ\text{C}/\text{min}$ and then cool down again at the same rate to 30° .

The wettability of electrospun EVOH/ Ti_3C_2 composite fibrous membranes were estimated using a contact angle meter (Kruss DSA100) (Kruss GmbH, Hamburg, Germany) with a droplet of $3\ \mu\text{L}$ at room temperature.

Tensile properties of electrospun EVOH/ Ti_3C_2 composite fibrous membranes were performed using a single fiber tensile testing machine (YG005E) (Wenzhou Fangyuan Instruments Co., Ltd., Wenzhou, China) with a 200 N load cell at room temperature. The

membranes used for testing were cut into strips with a length of 30 mm and a width of 5 mm and extended at a constant cross-head speed of 0.1 mm s^{-1} with a gauge length of 20 mm. Each sample was repeated five times to obtain an average value and further ensure the repeatability.

The air permeability of electrospun EVOH/Ti₃C₂ composite fibrous membrane was tested based on ASTM D 737 standard [37] using a fabric permeability tester (YG461E) (Ningbo Textile Instrument Factory, Ningbo, China) with a test pressure difference of 100 Pa. The samples were cut into a circular shape with a diameter of 10 cm.

The water vapor permeability of electrospun EVOH/Ti₃C₂ composite fibrous membrane was assessed based on ASTM-E96 [38] desiccant method by employing a water vapor transmission tester (YG(B)216-II) (Wenzhou Darong Textile Instrument Co., Ltd., Wenzhou, China). A circular cup containing 33 g of CaCl₂ was placed in a testing chamber at 38 °C and 90% relative humidity with a wind velocity of 1 m s^{-1} . WVTR was calculated based on the following equation [39]:

$$\text{WVTR} = \frac{m_2 - m_1}{s} \times 24 \quad (1)$$

where WVTR indicates the water vapor transmission per square meter every day (24 h) and is expressed in $\text{kg m}^{-2} \text{ d}^{-1}$, $m_2 - m_1$ is the weight change of CaCl₂ in the test assembly in 1 h, s is the effective area of specimen. For each sample, WVTR and air permeability values were measured at least five times.

3. Results and discussion

3.1. Morphologies of Electrospun EVOH/Ti₃C₂ Composite Fibers

The Ti₃C₂ composite fibers were successfully electrospun from 0 wt% to 10 wt% Ti₃C₂ in electrospinning polymer solutions; the morphologies of the electrospun EVOH/Ti₃C₂ fibers are exhibited in Figure 1. Between 0 to 10 wt% Ti₃C₂, the color of the resulting electrospun fibrous membrane gradually changes from white to gray (Figure 1(a₁–d₁)). Specifically, the color of the pristine electrospun EVOH fibrous membrane (0 wt% Ti₃C₂) is white, and the color of the electrospun EVOH/Ti₃C₂ fibrous membrane is gray. As the Ti₃C₂ concentration increases, the composite fibrous membrane gradually changes from light gray to dark gray.

As shown in Figure 1(a₂–d₂, a₃–d₃), it can be clearly seen that the electrospun EVOH or EVOH/Ti₃C₂ fibrous membranes present a three-dimensional non-woven geometry composed of randomly aligned nanofibers. The difference lies in the fact that the pristine electrospun EVOH fibers are smooth and relatively uniform, while the electrospun EVOH/Ti₃C₂ fibers are less uniform with fewer Ti₃C₂ flakes protruding from the fiber surface. Combined with the high-magnification SEM images, we can see that with the increase in the Ti₃C₂ concentration in the electrospinning polymer solution, the Ti₃C₂ flakes on the fiber surface also increase. It has been reported that protruding Ti₃C₂ flakes can serve as sites for destroying bacterial membranes [34]. However, the presence of the Ti₃C₂ flakes results in the inhomogeneity of the fiber (marked by some red dashed circles), which may further affect the mechanical properties of the electrospun EVOH/Ti₃C₂ composite fibrous membranes.

Meanwhile, the control of the fiber diameter plays a pivotal role in the specific material properties that show a clear size dependence [40]. The average diameters of the fibers are estimated in Figure 1(a₄–d₄). The average fiber diameter decreases as a function of the Ti₃C₂ concentration, from $0.82 \pm 0.10 \text{ }\mu\text{m}$ to $0.60 \pm 0.16 \text{ }\mu\text{m}$ for the pristine electrospun EVOH fibers and electrospun EVOH/Ti₃C₂ composite fibers with 10 wt% Ti₃C₂, respectively, suggesting that the addition of the Ti₃C₂ results in a decrease in the fiber diameter and an increase in the fiber uniformity. This is due to the addition of Ti₃C₂, which increases the conductivity of the polymer solution [30]. The increase in the solution conductivity brings with it a higher amount of carried charge, which allows for the further stretching of the polymer jet, resulting in a smaller fiber diameter [41]. Nevertheless, when the Ti₃C₂

increases from 5 wt% to 10 wt%, the average diameter of the fiber decreases only slightly. This phenomenon is due to the fact that when electrospinning a conductive solution, the excess charges in the polymer jet rapidly dissipate to the collector, inducing the collector to produce the opposite charges, resulting in fiber separation, which is not conducive to the formation of fibers [42].

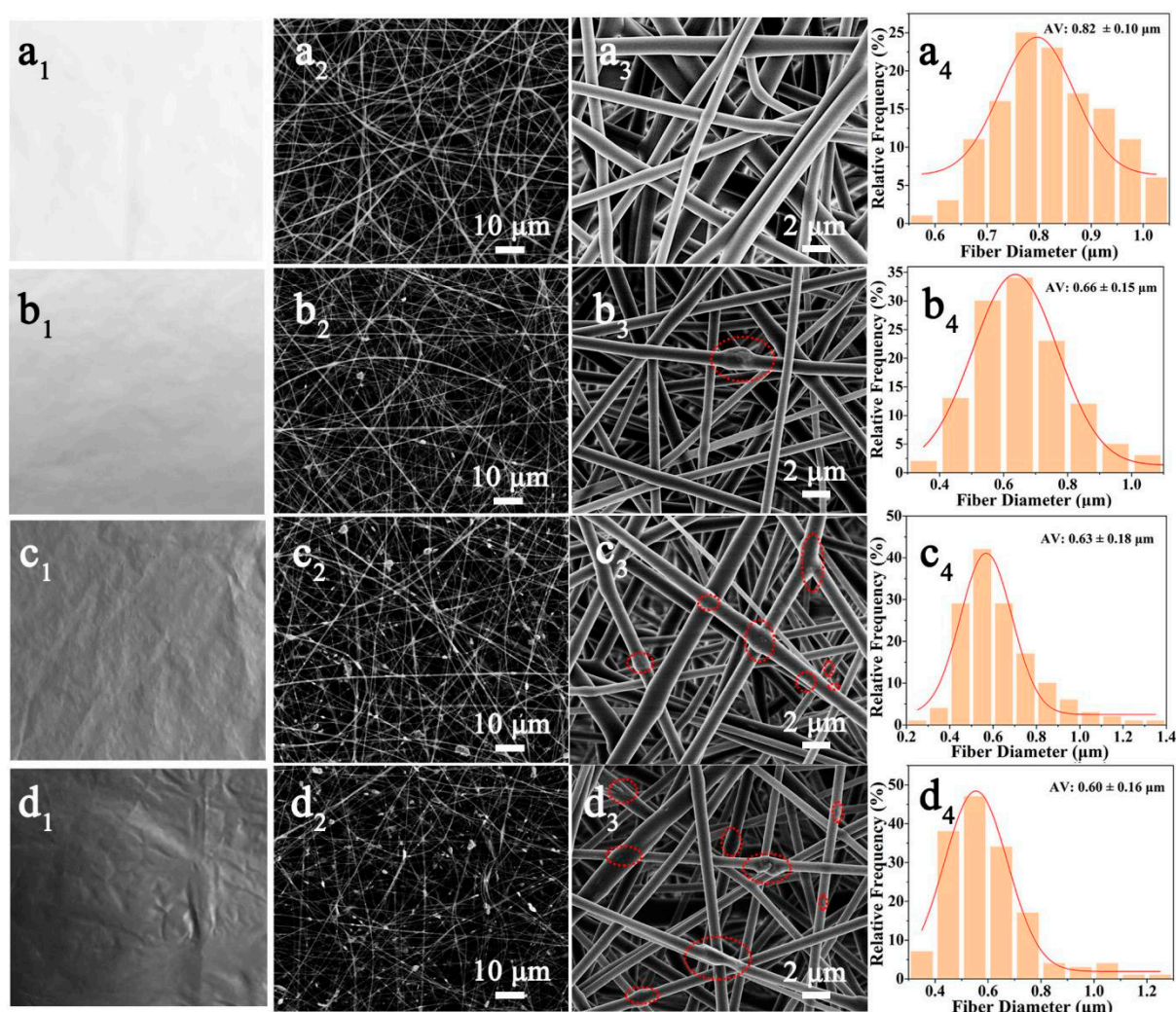


Figure 1. Morphology of electrospun EVOH fiber with increasing concentration of Ti_3C_2 in the electrospinning solution: (a₁–a₄) 0 wt%, (b₁–b₄) 2.5 wt%, (c₁–c₄) 5 wt%, (d₁–d₄) 10 wt%. (a₁, b₁, c₁, d₁) Optical images, (a₂, b₂, c₂, d₂) SEM images at low magnification, (a₃, b₃, c₃, d₃) SEM images at high magnification, and (a₄, b₄, c₄, d₄) fiber diameter distributions of electrospun EVOH/ Ti_3C_2 fibrous membranes.

In order to confirm the load of the Ti_3C_2 in the electrospun EVOH fibers, EDS elemental mapping images were taken, as shown in Figure 2. The pristine electrospun EVOH fibers contain only C and O elements (Figure 1(a₁–a₄)), while the electrospun EVOH/ Ti_3C_2 fibers with 2.5 wt% to 10 wt% Ti_3C_2 in the electrospinning polymer solution show C, O, and Ti elements (Figure 2b–d), which indicates that Ti element has been loaded onto the fiber. With the increase in the Ti_3C_2 concentration, more Ti element is dispersed on the fibrous membrane, especially where the Ti_3C_2 flakes are exposed on the fiber surface, Ti element dispersion is more obvious. In addition, an EDS elemental mapping image of the electrospun EVOH/ Ti_3C_2 composite fiber cross-section with 10 wt% Ti_3C_2 is displayed to further detect the Ti_3C_2 loading within the fibers. It is confirmed that Ti element is uniformly dispersed in the fiber cross-section and Ti_3C_2 flakes are uniformly loaded inside the fibers.

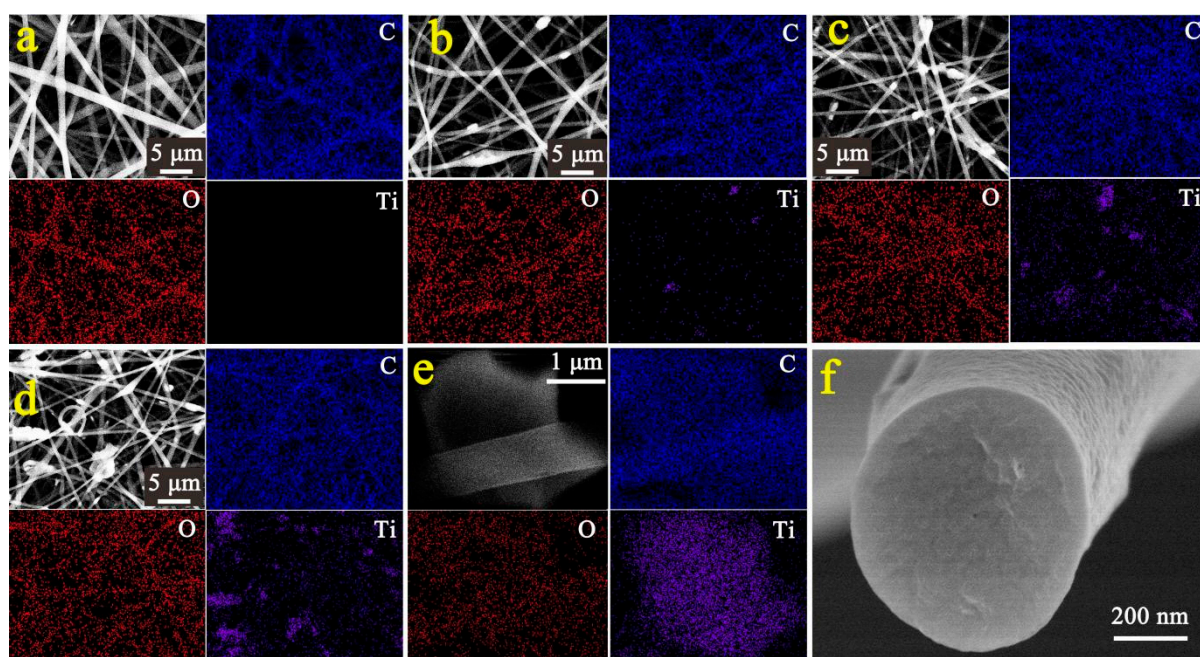


Figure 2. SEM-EDS elemental mapping images of electrospun EVOH/Ti₃C₂ membranes exhibiting the distribution of C, O, and Ti in the same area with increasing concentration of Ti₃C₂ in the electrospinning solution: (a) 0 wt%, (b) 2.5 wt%, (c) 5 wt%, (d) 10 wt%. (e) SEM-EDS elemental mapping images, and (f) SEM image of electrospun EVOH/Ti₃C₂ fiber cross-section with 10 wt% concentration.

3.2. Structure of Electrospun EVOH/Ti₃C₂ Composite Fibers

To investigate the effect of the Ti₃C₂ content on the structure of the electrospun EVOH composite fibers, the XRD patterns and FTIR spectra of the electrospun EVOH/Ti₃C₂ fibrous membranes were analyzed, as shown in Figure 3. A typical XRD pattern of the pristine electrospun EVOH fibrous membrane given by Figure 3a displays a crystal peak at a 2θ of about 16.2° , which has been reported in previous literature [21]. With the increase in the Ti₃C₂ concentration, the intensity and shape of the diffraction peak appearing at $2\theta = 16.2^\circ$ have no obvious change, indicating that the presence of Ti₃C₂ does not affect the crystal structure of the electrospun EVOH fibrous membrane. The 2.5 wt% addition of Ti₃C₂ results in the appearance of new crystal peaks, which appear at $2\theta = 5.4^\circ$, 27.4° , and 34.4° , corresponding to the crystal planes (002), (006), and (010), respectively. As the Ti₃C₂ content increases, the intensities of these diffraction peaks increase. At a concentration of 5 wt% Ti₃C₂, the new diffraction peak appears at 2θ of 31° , corresponding to the (008) crystal plane. When the concentration of Ti₃C₂ increases to 10 wt%, the diffraction peaks at 2θ of 5.4° , 27.4° , 31° , and 34.4° increase, and a new crystal peak appears at $2\theta = 14.8^\circ$, corresponding to the (004) crystal plane. Previous studies have reported that Ti₃C₂ has five diffraction peaks, namely (002), (004), (006), (008), and (010) [29,43]. This manifests that with the increase in the Ti₃C₂ concentration, more Ti₃C₂ is loaded onto the electrospun EVOH fibrous membrane, and the crystal peaks corresponding to Ti₃C₂ are enhanced, meaning that the orientation of Ti₃C₂ within the electrospun fiber increases, which has potential applications in the biological field.

The FTIR spectra in Figure 3b show typical characteristic peaks of electrospun EVOH fibrous membrane with vibrational bands at 2926 cm^{-1} and 2853 cm^{-1} for the C-H stretch, 1431 cm^{-1} and 1327 cm^{-1} for bending, which are in agreement with the literature [17,44]. Meanwhile, the basic stretching vibration of the free hydroxyl group (-OH) corresponds to a sharp peak around 3300 cm^{-1} [45]. It can be seen that there are no peaks broadening or shifting with the participation of Ti₃C₂, which may be attributed to the low loading of Ti₃C₂ in the electrospun EVOH/Ti₃C₂ fibers. As a matter of fact, the Ti₃C₂ characteristic

crystal peaks of the electrospun EVOH/Ti₃C₂ fibrous membranes exhibit a lower intensity in their XRD pattern, especially after the addition of 10 wt% Ti₃C₂; the corresponding diffraction peaks are not significant (Figure 3a), indicating that although Ti₃C₂ has been loaded onto the electrospun EVOH fiber, the low loading in the electrospun EVOH fiber may be the reason for no significant changes in the FTIR. It is worth noting that the typical characteristic peaks of the electrospun EVOH fiber do not change with the increase in the Ti₃C₂ concentration, indicating that the addition of Ti₃C₂ does not affect the structure of the EVOH fiber.

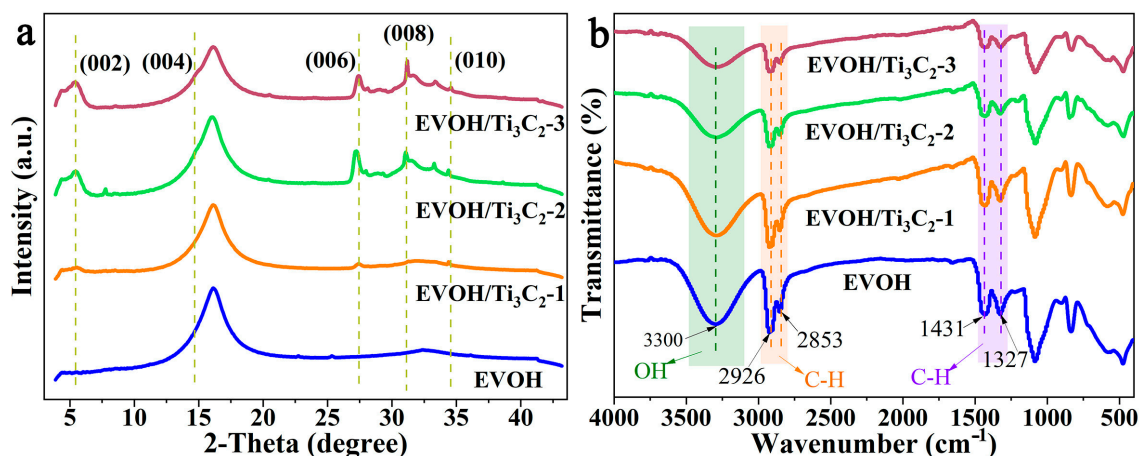


Figure 3. XRD patterns (a) and FTIR spectra (b) of electrospun EVOH/Ti₃C₂ fibers.

3.3. Thermal Properties of Electrospun EVOH/Ti₃C₂ Composite Fibers

DSC analysis was performed to examine the thermal properties of the electrospun EVOH/Ti₃C₂ composite fibrous membranes with increasing concentration of Ti₃C₂ in the electrospinning solution, as illustrated in Figure 4. As can be seen from the figure that the glass transition temperature and the melting temperature both shift slightly to the right due to the addition of Ti₃C₂, that is to say, with the increase in the Ti₃C₂ concentration, the glass transition temperature increases, and the melting temperature also increases slightly. In addition, it can be seen from the melting curve shape that the electrospun EVOH/Ti₃C₂ composite fibrous membrane with 2.5 wt% concentration of Ti₃C₂ can melt more at lower melting temperatures (e.g., 136–149°) in comparison to that of the Ti₃C₂ concentration of 5 wt% and 10 wt%, indicating the presence of thin crystals, which is consistent with the XRD results.

The glass transition temperature (T_g) and melting temperature (T_m) of the electrospun EVOH/Ti₃C₂ fibrous membranes are listed in Table 1. The glass transition temperature and melting temperature of the pristine electrospun EVOH fibrous membrane are 101° and 156°, respectively. With the increasing Ti₃C₂ content, the glass transition temperature of the electrospun EVOH/Ti₃C₂ fibrous membrane rises to 107°, and the melting temperature slightly increases to 160°. This indicates that the participation of Ti₃C₂ increases the thermal stability of the electrospun EVOH fibrous membrane, and the greater the Ti₃C₂ concentration in the electrospinning solution, the better the thermal stability of the fibrous membrane.

Table 1. DSC results (i.e., T_g and T_m) of electrospun EVOH/Ti₃C₂ fibrous membranes.

Samples	T_g (°C)	T_m (°C)
EVOH	101	156
EVOH/Ti ₃ C ₂ -1	104	158
EVOH/Ti ₃ C ₂ -2	105	160
EVOH/Ti ₃ C ₂ -3	107	159

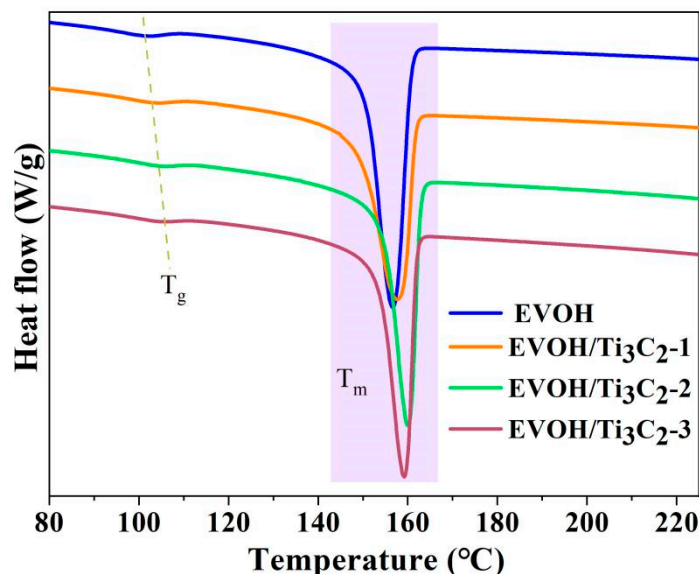


Figure 4. DSC thermograms of electrospun EVOH/Ti₃C₂ composite fibers.

3.4. Wettability of the Fibrous Membranes

The wettability of the electrospun EVOH/Ti₃C₂ fibrous membrane were evaluated by the contact angle and fluid uptake ability. By testing the contact angle of the fibrous membrane within 30 min, it is found that the initial contact angle of the pristine electrospun EVOH fibrous membrane is 127°, which decreases to about 124° at 22 min, and then remains unchanged within the 30 min. For the electrospun EVOH/Ti₃C₂ fibrous membranes, their initial contact angle is almost the same, at about 125°. The difference is the time when the contact angle of the fibrous membrane changes. The greater the Ti₃C₂ content, the faster the contact angle decreases. For example, the electrospun EVOH/Ti₃C₂ fibrous membrane with 10 wt% Ti₃C₂ in the electrospinning solution exhibits the rapid penetration of the water droplet, that is, the water droplet completely penetrates into the fibrous membrane within 18 min, and the corresponding contact angle is 0°, indicating that the presence of Ti₃C₂ improves the hydrophilicity of the electrospun EVOH fibrous membrane.

At the same time, the fluid uptake ability (FUA) of the fibrous membranes was evaluated by immersing the desired fibrous membranes in water to measure their water absorption capacity, according to the following formula [18,19]:

$$\text{FUA}(\%) = \frac{(W_s - W_d) \times 100\%}{W_s} \quad (2)$$

where W_s represents the wet weight of the fibrous membrane, W_d represents the dry weight of the fibrous membrane. In this part of the experimental operation, the dry fibrous membrane was first completely immersed in water for 30 s, then removed and placed on absorbent paper for 30 s, and finally weighed, and the above steps were repeated.

As shown in Figure 5b, the pristine electrospun EVOH fibrous membrane achieves absorption saturation in 150 s, with an absorption rate of 87.6%, and then begins to decline. With the increase in the Ti₃C₂ concentration, the absorption saturation time of the electrospun EVOH/Ti₃C₂ fibrous membrane is shortened, and the corresponding absorption rate is increased. Among them, the absorption saturation time of the EVOH/Ti₃C₂ fibrous membrane with 5 wt% Ti₃C₂ is half of that of the fibrous membrane with 2.5 wt%, while the electrospun EVOH/Ti₃C₂ fibrous membrane with 10 wt% Ti₃C₂ reaches absorption saturation in just 30 s, with an absorption rate as high as 93.7%. Based on the contact angle and fluid uptake ability of the fibrous membranes, it can be observed that the wettability of the electrospun EVOH/Ti₃C₂ fibrous membrane has been significantly improved due to the introduction of Ti₃C₂.

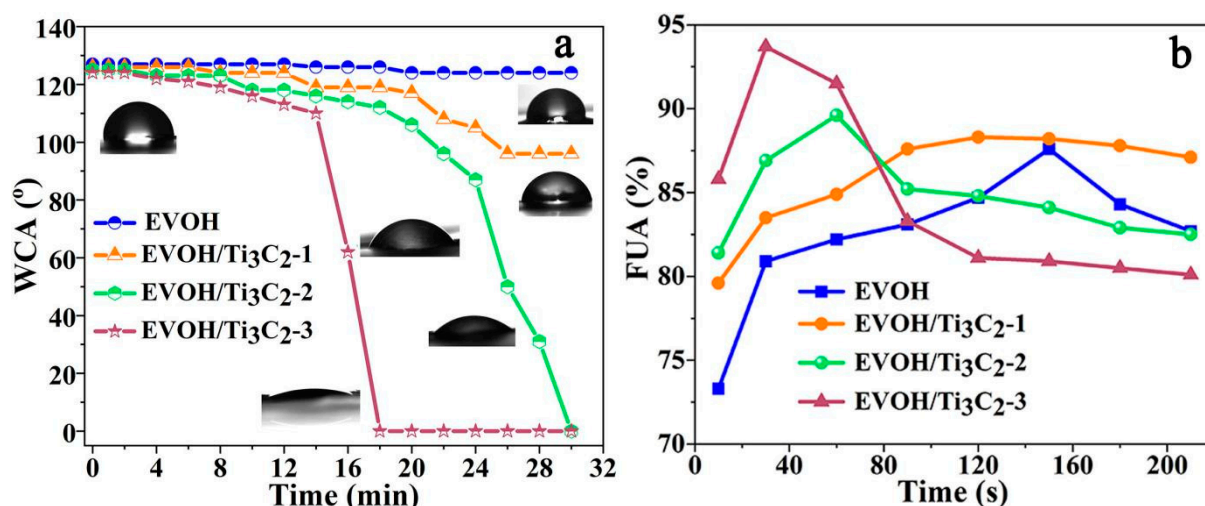


Figure 5. Wettability of electrospun EVOH/Ti₃C₂ fibrous membranes: (a) the variation and digital images of WCA, (b) fluid uptake ability.

3.5. Mechanical Properties of Fibrous Membranes

The mechanical properties of fibrous membranes, which are affected by the morphology and microstructure of fibers, have become an indispensable topic of research for their practical applications [46,47]. Typical stress–strain curves of the electrospun EVOH/Ti₃C₂ fibrous membranes were investigated to study the effect of additions of Ti₃C₂ on the tensile properties of the electrospun EVOH fibrous membranes, as shown in Figure 6. The tensile curves of the fibrous membranes present some general characteristics, starting with an increase in stress (including a clear linear and a nonlinear elastic stage), then showing a decrease in stress followed by a mild strain hardening [48]. The difference lies in the breaking strength and elongation at the break of the fibrous membrane. It can be seen that the pristine electrospun EVOH fibrous membrane presents a higher tensile stress of 5.13 MPa and a higher breaking elongation of 104%. When the concentration of Ti₃C₂ increases from 2.5 wt% to 10 wt%, the tensile stress of the electrospun EVOH/Ti₃C₂ fibrous membrane ranges from 4.77 MPa to 2.95 MPa, and the breaking elongation fluctuates from 87.5% to 58.24%, still showing good strength and elasticity. However, compared to the pristine electrospun EVOH fibrous membranes, the tensile stress of the electrospun EVOH/Ti₃C₂ fibrous membranes with 2.5 wt% and 5 wt% Ti₃C₂ decreased by 7% and 13%, respectively, while the tensile strength of the electrospun EVOH/Ti₃C₂ fibrous membranes with 10 wt% Ti₃C₂ decreased by 43%, and their corresponding strain also decreased, which indicates that the addition of Ti₃C₂ weakens the tensile properties of the fibrous membranes.

Combined with the results in Figure 1, it can be seen that the presence of Ti₃C₂ induces the fiber's thinning, and the smaller the fiber's diameter, the greater the tensile strength of the corresponding fiber membrane [49,50], which is inconsistent with our tensile results. It is worth noting that although the addition of Ti₃C₂ refines the fiber, it causes the fiber to become very uneven. Moreover, the protrusion of some Ti₃C₂ flakes on the fiber surface may become the weak points at which the fibrous membrane stretches, and this also weakens the elasticity of the fibrous membrane. Of course, it has also been shown that when the Ti₃C₂ concentration is high enough (such as 16 wt%), the Ti₃C₂ flakes are interconnected along the fiber axis, which will increase the uniformity of the fibrous membrane and may improve the strength of the fibrous membrane [29].

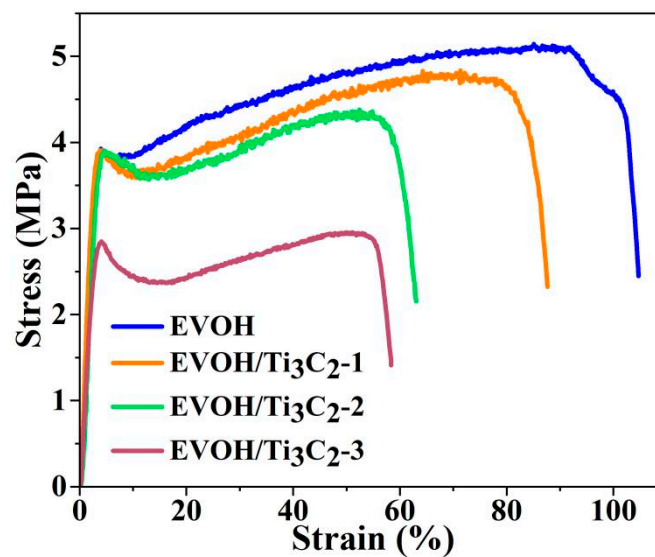


Figure 6. Typical stress–strain curves of electrospun EVOH/Ti₃C₂ fibrous membranes.

3.6. Air Permeability and Water Vapor Permeability of Fibrous Membrane

The air permeability and water vapor permeability of the electrospun EVOH/Ti₃C₂ fibrous membranes were further estimated. As can be seen from Figure 7, the electrospun EVOH/Ti₃C₂ fibrous membranes display excellent air permeability and water vapor permeability. When the Ti₃C₂ concentration increases from 0 wt% to 5 wt%, the air permeability of the fibrous membrane elevates from 87.8 mm s⁻¹ to 97.4 mm s⁻¹, while the WVTR of these membranes declines from 10 kg m⁻² d⁻¹ to 8.6 kg m⁻² d⁻¹, implying that the addition of Ti₃C₂ improves the air permeability of the electrospun EVOH fibrous membranes. It has been reported that the air permeability and water vapor permeability of the electrospun fibrous membranes are linearly positive related to the porosity, that is, the more pores in the fibrous membranes, the more air molecules and water vapor will pass through [51]. Figure 5 shows that Ti₃C₂ can rapidly improve the hydrophilic properties of the electrospun EVOH fibrous membrane, evidencing that it is a hydrophilic material. When the fibrous membranes are tested for water vapor permeability, the Ti₃C₂ flakes will absorb some water vapor, which may result in less water vapor passing through the fibrous membranes.

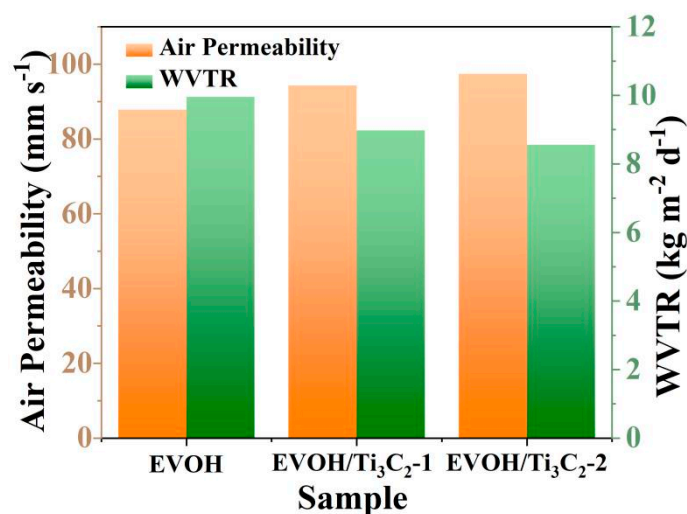


Figure 7. Air permeability and WVTR of electrospun EVOH/Ti₃C₂ fibrous membranes.

4. Conclusions

In summary, we have successfully prepared an EVOH/Ti₃C₂ composite fiber for the first time via a one-step electrospinning method and further investigated the effect of different Ti₃C₂ concentrations on the structure and properties of the electrospun EVOH/Ti₃C₂ fibrous membranes. The added Ti₃C₂ was well loaded into the fibers and induced a reduction in the fiber diameter from 0.82 μm to 0.60 μm. Moreover, Ti₃C₂ could self-orient along the fiber axis but did not change the crystalline structure of the electrospun EVOH fiber, improving the crystallinity and thermal stability of the composite fibers. As the concentration of Ti₃C₂ increased, the contact angles of the electrospun EVOH/Ti₃C₂ fibrous membranes decreased from 125° to 0° within 18 min, and these fibrous membranes reached water absorption saturation within 30 s, with a UFA as high as 93.7%, demonstrating the excellent wettability of electrospun EVOH/Ti₃C₂ fibrous membranes. Meanwhile, the electrospun EVOH/Ti₃C₂ fibrous membranes also showed outstanding air permeability and water vapor permeability, as well as modest mechanical properties. The as-prepared electrospun EVOH/Ti₃C₂ fibrous membranes could serve as a multi-functional textile for the potential candidates in bioprotective materials, dressing materials, etc., laying the foundation for the further development of applications in later work.

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References

1. Cleeton, C.; Keirouz, A.; Chen, X.; Radacsi, N. Electrospun nanofibers for drug delivery and biosensing. *ACS Biomater. Sci. Eng.* **2019**, *5*, 4183–4205. [[CrossRef](#)]
2. Zahra, F.T.; Quick, Q.; Mu, R. Electrospun PVA fibers for drug delivery: A review. *Polymers* **2023**, *15*, 3837. [[CrossRef](#)] [[PubMed](#)]
3. Zhou, T.; Jiang, Q.T.; Wang, L.; Qiu, Z.P.; Liu, Y.Y.; Zhou, J.; Liu, B. Facile preparation of nitrogen-enriched hierarchical porous carbon nanofibers by Mg(OAc)₂-assisted electrospinning for flexible supercapacitors. *Appl. Surf. Sci.* **2018**, *456*, 827–834. [[CrossRef](#)]
4. Rasoolzadeh, M.; Sherafat, Z.; Vahedi, M.; Bagherzadeh, E. Structure dependent piezoelectricity in electrospun PVDF-SiC nanoenergy harvesters. *J. Alloys Compd.* **2022**, *917*, 165505. [[CrossRef](#)]
5. Fan, X.; Wang, Y.; Zheng, M.; Dunne, F.; Liu, T.; Fu, X.W.; Kong, L.; Pan, S.Y.; Zhong, W.H. Morphology engineering of protein fabrics for advanced and sustainable filtration. *J. Mater. Chem. A* **2018**, *6*, 21585–21595. [[CrossRef](#)]
6. Gao, Q.; Agarwal, S.; Greiner, A.; Zhang, T. Electrospun fiber-based flexible electronics: Fiber fabrication, device platform, functionality integration and applications. *Prog. Mater. Sci.* **2023**, *137*, 101139.
7. Sarrami, P.; Karbasi, S.; Farahbakhsh, Z.; Bigham, A.; Rafienia, M. Fabrication and characterization of novel polyhydroxybutyrate-keratin/nanohydroxyapatite electrospun fibers for bone tissue engineering applications. *Int. J. Biol. Macromol.* **2022**, *220*, 1368–1389. [[CrossRef](#)]
8. Liu, Q.X.; Guo, X.P.; Zhang, Y.F.; Wang, X.; Mo, M.; Wu, T. Progress in electrospun fibers for manipulating cell behaviors. *Adv. Fiber Mater.* **2023**, *5*, 1241–1272. [[CrossRef](#)]
9. Zhao, R.; Tian, Y.Y.; Li, S.Y.; Ma, T.T.; Lei, H.T.; Zhu, G.S. An electrospun fiber based metal-organic framework composite membrane for fast, continuous, and simultaneous removal of insoluble and soluble contaminants from water. *J. Mater. Chem. A* **2019**, *7*, 22559–22570. [[CrossRef](#)]
10. Gao, J.F.; Li, B.; Wang, L.; Huang, X.W.; Xue, H.G. Flexible membranes with a hierarchical nanofiber/microsphere structure for oil adsorption and oil/water separation. *J. Ind. Eng. Chem.* **2018**, *68*, 416–424. [[CrossRef](#)]
11. Ramakrishna, S.; Fujihara, K.; Teo, W.E.; Yong, T.; Ma, Z.W.; Ramaseshan, R. Electrospun nanofibers: Solving global issues. *Mater. Today* **2006**, *9*, 40. [[CrossRef](#)]
12. Gao, Z.; Xiao, X.; Carlo, A.D.; Yin, J.; Wang, Y.; Huang, L.; Tang, J.; Chen, J. Advances in wearable strain sensors based on electrospun fibers. *Adv. Funct. Mater.* **2023**, *33*, 2214265. [[CrossRef](#)]

13. Zhao, T.; Xu, Y.; Wu, M.; Li, Y.; Ma, J.; Li, H.; Zheng, Y.; Zeng, Y. Highly efficient fabrication of biomimetic nanoscaled tendrils for high-performance PM_{0.3} air filters. *Nano Lett.* **2024**, *24*, 1385–1391. [[CrossRef](#)] [[PubMed](#)]
14. Zhang, S.; Yan, D.; Zhao, Z.; Lin, L. Composite fibrous membrane comprising PLA and PCL fibers for biomedical application. *Compos. Commun.* **2022**, *34*, 101268. [[CrossRef](#)]
15. Zhang, T.; Jin, Z.; Jia, Z.; Xu, X.; Chen, Y.; Peng, W.; Zhang, J.; Wang, H.; Li, S.; Wen, J. Preparation of antimicrobial PVDF and EVOH membranes by grafting quaternary ammonium compounds. *React. Funct. Polym.* **2022**, *170*, 105117. [[CrossRef](#)]
16. Jing, L.; Jia, Z.; Xu, X.; Chen, Y.; Peng, W.; Zhang, J.; Wang, H.; Li, S.; Wen, J. Preparation of antimicrobial poly (ethylene-co-vinyl alcohol) membrane by grafting with N-halamine. *React. Funct. Polym.* **2022**, *172*, 105187. [[CrossRef](#)]
17. Xu, D.; Zhu, K.; Zheng, X.; Xiao, R. Poly(ethylene-co-vinyl alcohol) functional nanofiber membranes for the removal of Cr(VI) from water. *Ind. Eng. Chem. Res.* **2015**, *54*, 6836–6844. [[CrossRef](#)]
18. Wang, A.; Xu, C.; Zhang, C.; Gan, Y.; Wang, B. Experimental investigation of the properties of electrospun nanofibers for potential medical application. *J. Nanomater.* **2015**, *2015*, 418932. [[CrossRef](#)]
19. Mondragón, M.; López-Villegas, O.; Sánchez-Valdés, S.; Rodríguez-González, F.J. Effect of thermoplastic starch and photocrosslinking on the properties and morphology of electrospun poly(ethylene-co-vinylalcohol) mats. *Polym. Eng. Sci.* **2019**, *60*, 474–480. [[CrossRef](#)]
20. Liang, M.; Wang, F.; Liu, M.; Yu, J.; Si, Y.; Ding, B. N-Halamine functionalized Electrospun Ppoly(Vinyl Alcohol-co-Ethylene) nanofibrous membranes with rechargeable antibacterial activity for bioprotective applications. *Adv. Fiber Mater.* **2019**, *1*, 126–136. [[CrossRef](#)]
21. Lu, J.; Jiang, Y.; Xiao, R.; Jacob, K.; Tao, L.; Li, S.; Guo, L. Chemical vapor deposition based superelastic and superhydrophobic thermoplastic polymeric nanofibrous aerogels for water purification. *J. Inorg. Organomet. Polym. Mater.* **2022**, *32*, 2975–2985. [[CrossRef](#)]
22. Jun, B.M.; Kim, S.; Heo, J.; Park, C.M.; Her, N.; Jang, M.; Huang, Y.; Han, J.; Yoon, Y. Review of MXenes as new nanomaterials for energy storage/delivery and selected environmental applications. *Nano Res.* **2019**, *12*, 471–487. [[CrossRef](#)]
23. Handoko, A.D.; Fredrickson, K.D.; Anasori, B.; Convey, K.W.; Johnson, L.R.; Gogotsi, Y.; Vojvodic, A.; Seh, Z.W. Tuning the basal plane functionalization of two-dimensional metal carbides (MXenes) to control hydrogen evolution activity. *ACS Appl. Energy Mater.* **2017**, *1*, 173–180. [[CrossRef](#)]
24. Riazi, H.; Nemani, S.K.; Grady, M.C.; Anasori, B.; Soroush, M. Ti₃C₂ MXene–polymer nanocomposites and their applications. *J. Mater. Chem. A* **2021**, *9*, 8051. [[CrossRef](#)]
25. Naguib, M.; Kurtoglu, M.; Presser, V.; Lu, J.; Niu, J.; Heon, M.; Hultman, L.; Gogotsi, Y.; Barsoum, M.W. Two-dimensional nanocrystals produced by exfoliation of Ti₃AlC₂. *Adv. Mater.* **2011**, *23*, 4248–4253. [[CrossRef](#)]
26. Pandey, R.P.; Rasool, K.; Madhavan, V.E.; Aissa, B.; Gogotsi, Y.; Mahmoud, K.A. Ultrahigh-flux and fouling-resistant membranes based on layered silver/MXene (Ti₃C₂T_x) nanosheets. *J. Mater. Chem. A* **2018**, *6*, 3522–3533. [[CrossRef](#)]
27. Fan, X.; Ding, Y.; Liu, Y.; Liang, J.; Chen, Y. Plasmonic Ti₃C₂T_x MXene enables highly efficient photothermal conversion for healable and transparent wearable device. *ACS Nano* **2019**, *13*, 8124–8134. [[CrossRef](#)]
28. Zhang, J.; Seyedin, S.; Gu, Z.; Yang, W.; Wang, X.; Razal, J.M. MXene: A potential candidate for yarn supercapacitors. *Nanoscale* **2017**, *9*, 18604–18608. [[CrossRef](#)] [[PubMed](#)]
29. Levitt, A.; Alhabeib, M.; Hatter, C.; Sarycheva, A.; Dionb, G.; Gogotsi, Y. Electrospun MXene/carbon nanofibers as supercapacitor electrodes. *J. Mater. Chem. A* **2019**, *7*, 296. [[CrossRef](#)]
30. Mayerberger, E.A.; Urbanek, O.; McDaniel, R.M.; Street, R.M.; Barsoum, M.W.; Schauer, C.L. Preparation and characterization of polymer-Ti₃C₂T_x (MXene) composite nanofibers produced via electrospinning. *J. Appl. Polym. Sci.* **2017**, *134*, 45295. [[CrossRef](#)]
31. Lei, J.C.; Zhang, X.; Zhou, Z. Recent advances in MXene: Preparation, properties, and applications. *Front. Phys.* **2015**, *10*, 276–286. [[CrossRef](#)]
32. Lin, X.P.; Li, Z.J.; Qiu, J.M.; Wang, Q.; Wang, J.X.; Zhang, H.; Chen, T.K. Fascinating MXene nanomaterials: Emerging opportunities in the biomedical field. *Biomater. Sci.* **2021**, *9*, 5437–5471. [[CrossRef](#)] [[PubMed](#)]
33. George, S.M.; Kandasubramanian, B. Advancements in MXene-polymer composites for various biomedical applications. *Ceram. Int.* **2020**, *46*, 8522–8535. [[CrossRef](#)]
34. Zheng, K.Y.; Li, S.; Jing, L.; Chen, P.; Xie, J.P. Synergistic antimicrobial titanium carbide (MXene) conjugated with gold nanoclusters. *Adv. Healthc. Mater.* **2020**, *9*, 2001007. [[CrossRef](#)]
35. Reneker, D.H.; Yarin, A.L.; Fong, H.; Koombhongse, S. Bending instability of electrically charged liquid jets of polymer solutions in electrospinning. *J. Appl. Phys.* **2000**, *87*, 4531–4547. [[CrossRef](#)]
36. Li, M.; Zheng, Y.; Xin, B.; Xu, Y. Coaxial electrospinning: Jet motion, core-shell fiber morphology, and structure as a function of material parameters. *Ind. Eng. Chem. Res.* **2020**, *59*, 6301–6308. [[CrossRef](#)]
37. ASTM D737; Standard Test Method for Air Permeability of Textile Fabrics, Subcommittee D 13.59 on Fabrics Test Methods. ASTM International: West Conshohocken, PA, USA, 2018.
38. ASTM E96/E96M; Standard Test Methods for Water Vapor Transmission of Materials, Subcommittee C 16.33 on Insulation Finishes and Moisture. ASTM International: West Conshohocken, PA, USA, 2016.
39. Gorji, M.; Jeddi, A.A.A.; Gharehaghaji, A.A. Fabrication and characterization of polyurethane electrospun nanofiber membranes for protective clothing applications. *J. Appl. Polym. Sci.* **2012**, *125*, 4135. [[CrossRef](#)]

40. Wang, M.; Hsieh, A.J.; Rutledge, G.C. Electrospinning of poly(MMA-co-MAA) copolymers and their layered silicate nanocomposites for improved thermal properties. *Polymer* **2005**, *46*, 3407. [[CrossRef](#)]
41. Huang, C.; Chen, S.; Lai, C.; Reneker, D.H.; Qiu, H.; Ye, Y.; Hou, H. Electrospun polymer nanofibres with small diameters. *Nanotechnology* **2006**, *17*, 1558. [[CrossRef](#)]
42. Zhao, W.; Yalcin, B.; Cakmak, M. Dynamic assembly of electrically conductive PEDOT:PSS nanofibers in electrospinning process studied by high speed video. *Synth. Met.* **2015**, *203*, 107–116. [[CrossRef](#)]
43. Gao, G.; Jiang, X.; Liu, C.; Song, S.; Zhang, J.; Shen, J. FeS@LAB-35@Ti₃C₂ as a high-efficiency nanozyme for near infrared light induced photothermal enhanced chemodynamic antibacterial activity and wound healing. *Nano Res.* **2023**, *16*, 2840–2850.
44. Jiang, Y.; Lu, J.; Guo, L. Fabrication of highly carboxylated thermoplastic nanofibrous membranes for efficient absorption and separation of protein. *Colloids Surf. A Physicochem. Eng. Asp.* **2023**, *665*, 131203. [[CrossRef](#)]
45. Dai, L.; Ying, L. Infrared Spectroscopic Investigation of Hydrogen Bonding in EVOH Containing PVA Fibers. *Macromol. Mater. Eng.* **2002**, *287*, 509–514. [[CrossRef](#)]
46. Siva, R.; Mobithis, M.; Ravichandran, R.; Valarmathi, T.; Jeya Jeevahan, J.; Sangeetha, M. Characterization of mechanical, chemical properties and microstructure of untreated and treated Cissus Quadrangularis fiber. *Mater. Today Proc.* **2021**, *47*, 4479–4483. [[CrossRef](#)]
47. Tsai, P.P.; Schreuder-Gibson, H.; Gibson, P. Different electrostatic methods for making electret filter. *J. Electrostat.* **2002**, *54*, 333–341. [[CrossRef](#)]
48. Wang, B.; Chao, X.; Li, Y.; Reid, S.R. Tensile strength of electrospun Poly(ethylene-co-vinyl alcohol) nanofibre sheets. *Key Eng. Mater.* **2013**, *535*, 215–218. [[CrossRef](#)]
49. Richard-Lacroix, M.; Pellerin, C. Orientation and Partial Disentanglement in Individual Electrospun Fibers: Diameter Dependence and Correlation with Mechanical Properties. *Macromolecules* **2015**, *48*, 4511–4519. [[CrossRef](#)]
50. Pai, C.L.; Boyce, M.C.; Rutledge, G.C. Mechanical properties of individual electrospun PA 6(3)T fibers and their variation with fiber diameter. *Polymer* **2011**, *52*, 2295–2301. [[CrossRef](#)]
51. Sheng, J.L.; Xu, Y.; Yu, J.Y.; Ding, B. Robust fluorine-free superhydrophobic amino-silicone Oil/SiO₂ modification of electrospun polyacrylonitrile membranes for waterproof-breathable application. *ACS Appl. Mater. Interfaces* **2017**, *17*, 15139–15147. [[CrossRef](#)] [[PubMed](#)]

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